PHYSICAL GEOLOGY



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23.1: About the Author

Steven Earle was born in the Okanagan Valley of British Columbia. He earned a BSc in geology from the University of British Columbia in 1975 and a PhD in geochemistry from Imperial College (London University) in 1982. He worked as a geologist and geochemist in the mineral exploration industry in western Canada from 1978 to 2000. For 20 years he developed and taught a wide range of earth science courses at Vancouver Island University. He currently designs and teaches distance courses for Thompson Rivers University (Open Learning), and also helps to grow food and drive the Community Bus on Gabriola Island. He maintains that the best way to see rocks is from a kayak.

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Physical Geology

Steven Earle

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1: INTRODUCTION TO GEOLOGY

1.1: What is Geology?
1.2: Why Study Earth?
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1.1: What is Geology?

In its broadest sense, geology is the study of Earth—its interior and its exterior surface, the minerals, rocks and other materials that are around us, the processes that have resulted in the formation of those materials, the water that flows over the surface and through the ground, the changes that have taken place over the vastness of geological time, and the changes that we can anticipate will take place in the near future. Geology is a science, meaning that we use deductive reasoning and scientific methods to understand geological problems. It is, arguably, the most integrated of all of the sciences because it involves the understanding and application of all of the other sciences: physics, chemistry, biology, mathematics, astronomy, and others. But unlike most of the other sciences, geology has an extra dimension, that of time—deep time—billions of years of it. Geologists study the evidence that they see around them, but in most cases, they are observing the results of processes that happened thousands, millions, and even billions of years in the past. Those were processes that took place at incredibly slow rates—millimeters per year to centimeters per year—but because of the amount of time available, they produced massive results.

Geology is displayed on a grand scale in mountainous regions, perhaps nowhere better than the Rocky Mountains in Canada (Figure 1.1.1). The peak on the right is Rearguard Mountain, which is a few kilometers northeast of Mount Robson, the tallest peak in the Canadian Rockies (3,954 meters). The large glacier in the middle of the photo is the Robson Glacier. The river flowing from Robson Glacier drains into Berg Lake in the bottom right. There are many geological features portrayed here. The sedimentary rock that these mountains are made of formed in ocean water over 500 million years ago. A few hundred million years later, these beds were pushed east for tens to hundreds of kilometers by tectonic plate convergence and also pushed up to thousands of meters above sea level. Over the past two million years this area—like most of the rest of Canada—has been repeatedly glaciated, and the erosional effects of those glaciations are obvious.

The Robson Glacier is now only a small remnant of its size during the Little Ice Age of the 15th to 18th centuries, and even a lot smaller that it was just over a century ago in 1908. The distinctive line on the slope on the left side of both photos shows the elevation of the edge of the glacier a few hundred years ago. Like almost all other glaciers in the world, it receded after the 18th century because of natural climate change, is now receding even more rapidly because of human-caused climate change.



Figure 1.1.1 Rearguard Mountain and Robson Glacier in Mount Robson Provincial Park, BC. Left: Robson Glacier in 2012. Right: Robson Glacier circa 1908.

Geology is also about understanding the evolution of life on Earth; about discovering resources such as water, metals and energy; about recognizing and minimizing the environmental implications of our use of those resources; and about learning how to mitigate the hazards related to earthquakes, volcanic eruptions, and slope failures. All of these aspects of geology, and many more, are covered in this textbook.



Figure 1.1.2

There is no single method of inquiry that is specifically the “scientific method”; furthermore, scientific inquiry is not necessarily different from serious research in other disciplines. The most important thing that those involved in any type of inquiry must do is to be skeptical. As the physicist Richard Feynman once said: the first principle of science is that “you must not fool yourself—and you are the easiest person to fool.” A key feature of serious inquiry is the creation of a hypothesis (a tentative explanation) that could explain the observations that have been made, and then the formulation and testing (by experimentation) of one or more predictions that follow from that hypothesis.

For example, we might observe that most of the cobbles in a stream bed are well rounded (see photo above), and then derive the hypothesis that the rocks are rounded by transportation along the stream bed. A prediction that follows from this hypothesis is that cobbles present in a stream will become increasingly rounded as they are transported downstream. An experiment to test this prediction would be to place some angular cobbles in a stream, label them so that we can be sure to find them again later, and then return at various time intervals (over a period of years) to carefully measure their locations and roundness.

A critical feature of a good hypothesis and any resulting predictions is that they must be testable. For example, an alternative hypothesis to the one above is that an extraterrestrial organization creates rounded cobbles and places them in streams when nobody is looking. This may indeed be the case, but there is no practical way to test this hypothesis. Most importantly, there is no way to prove that it is false, because if we aren’t able to catch the aliens at work, we still won’t know if they did it!

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1.2: Why Study Earth?

The simple answer to this question is that Earth is our home—our only home for the foreseeable future—and in order to ensure that it continues to be a great place to live, we need to understand how it works. Another answer is that some of us can't help but study it because it's fascinating. But there is more to it than that:

- We rely on Earth for valuable resources such as soil, water, metals, industrial minerals, and energy, and we need to know how to find these resources and exploit them sustainably.
- We can study rocks and the fossils they contain to understand the evolution of our environment and the life within it.
- We can learn to minimize our risks from earthquakes, volcanoes, slope failures, and damaging storms.
- We can learn how and why Earth's climate has changed naturally in the past, and use that knowledge to understand both natural and human-caused climate change.
- We can recognize how our activities have altered the environment in many ways and the climate in increasingly serious ways, and how to avoid more severe changes in the future.
- We can use our knowledge of Earth to understand other planets in our solar system, as well as those around distant stars.

An example of the importance of geological studies for minimizing risks to the public is illustrated in Figure 1.2.1. This is a slope failure that took place in January 2005 in the Riverside Drive area of North Vancouver. The steep bank beneath the house shown gave way, and a slurry of mud and sand flowed down, destroying another house below and killing one person. This event took place following a heavy rainfall, which is a common occurrence in southwestern B.C. in the winter.



Figure 1.2.1 The aftermath of a deadly debris flow in the Riverside Drive area of North Vancouver in January 2005.

The irony of the 2005 slope failure is that the District of North Vancouver had been warned in a geological report written in 1980 that this area was prone to slope failure and that steps should be taken to minimize the risk to residents. Very little was done in the intervening 25 years, and the consequences of that were deadly.

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1.3: What Do Geologists Do?

Geologists are involved in a range of widely varying occupations with one thing in common: the privilege and responsibility of studying this fascinating planet. In Canada, many geologists work in the resource industries, including mineral exploration and mining and energy exploration and extraction. Other major areas where geologists work include hazard assessment and mitigation (e.g., assessment of risks from slope failures, earthquakes, and volcanic eruptions); water supply planning, development, and management; waste management; and assessment of geological issues in the forest industry, and on construction projects such as highways, tunnels, and bridges. Most geologists are employed in the private sector, but many work for government-funded geological organizations, such as the Geological Survey of Canada or one of the provincial geological surveys. And of course, many geologists are involved in education at the secondary and the post-secondary levels.

Some people are attracted to geology because they like to be outdoors, and it is true that many geological opportunities involve fieldwork in places that are as amazing to see as they are interesting to study. But a lot of geological work is also done in offices or laboratories. Geological work tends to be varied and challenging, and for these reasons and many others, geologists are among those who are the most satisfied with their employment.



Figure 1.3.1 Geologists examining ash-layer deposits at Kilauea Volcano, Hawaii.

In Canada, most working geologists are required to be registered with an association of professional geoscientists. This typically involves meeting specific post-secondary educational standards and gaining several years of relevant professional experience under the supervision of a registered geoscientist. More information can be found at Engineers and Geoscientists British Columbia.

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1.4: Minerals and Rocks

The rest of this chapter is devoted to a brief overview of a few of the important aspects of physical geology, starting with minerals and rocks. This is followed by a review of Earth's internal structure and the processes of plate tectonics, and an explanation of geological time.

The Earth is made up of varying proportions of the 90 naturally occurring elements—hydrogen, carbon, oxygen, magnesium, silicon, iron, and so on. In most geological materials, these combine in various ways to make minerals. Minerals will be covered in some detail in Chapter 2, but here we will briefly touch on what minerals are, and how they are related to rocks.

A mineral is a naturally occurring combination of specific elements that are arranged in a particular repeating three-dimensional structure or **lattice**.[1] The mineral **halite** is shown as an example in Figure 1.4.1.

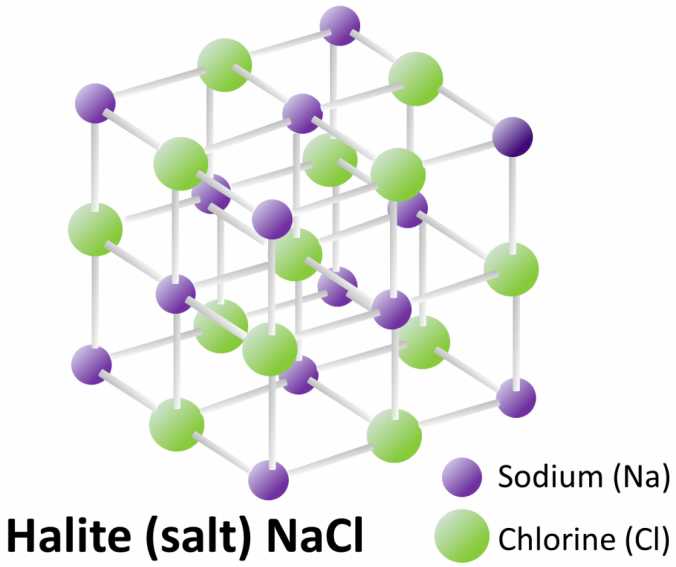


Figure 1.4.1 The lattice structure and composition of the mineral halite (common table salt).

In this case, atoms of sodium (Na: purple) alternate with atoms of chlorine (Cl: green) in all three dimensions, and the angles between the bonds are all 90°. Even in a tiny crystal, like the ones in your salt shaker, the lattices extend in all three directions for thousands of repetitions. Halite always has this composition and this structure.

Note: Element symbols (e.g., Na and Cl) are used extensively in this book. In Appendix 1, you will find a list of the symbols and names of the elements common in minerals and a copy of the periodic table. Please use those resources if you are not familiar with the element symbols.

There are thousands of minerals, although only a few dozen are mentioned in this book. In nature, minerals are found in rocks, and the vast majority of rocks are composed of at least a few different minerals. A close-up view of **granite**, a common rock, is shown in Figure 1.4.2. Although a hand-sized piece of granite may have thousands of individual mineral crystals in it, there are typically only a few different minerals, as shown here.

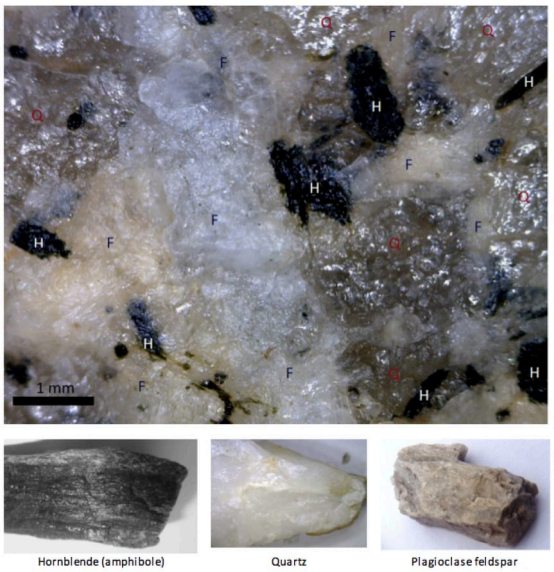


Figure 1.4.2 A close-up view of the rock granite and some of the minerals that it typically contains (H = hornblende (amphibole), Q = quartz and F = feldspar). The crystals range from about 0.1 to 3 millimeters (mm) in diameter. Most are irregular in outline, but some are rectangular.

Rocks can form in a variety of ways. Igneous rocks form from **magma** (molten rock) that has either cooled slowly underground (e.g., to produce granite) or cooled quickly at the surface after a volcanic eruption (e.g., **basalt**). Sedimentary rocks, such as **sandstone**, form when the weathered products of other rocks accumulate at the surface and are then buried by other sediments. Metamorphic rocks form when either igneous or sedimentary rocks are heated and squeezed to the point where some of their minerals are unstable and new minerals form to create a different type of rock. An example is **schist**.

A critical point to remember is the difference between a mineral and a rock. A mineral is a pure substance with a specific composition and structure, while a rock is typically a mixture of several different minerals (although a few types of rock may include only one type of mineral). Examples of minerals are feldspar, quartz, mica, halite, calcite, and amphibole. Examples of rocks are granite, basalt, sandstone, limestone, and schist.

> If you are currently taking a geology course, you'll likely be asked more than once to name a mineral or a rock that has specific characteristics or composition, or was formed in a specific environment. Please make sure that if you're asked for a **rock name** that you don't respond with a **mineral name**, and *vice versa*. Confusing minerals and rocks is one of the most common mistakes that geology students make.

Exercise 1.1 Find a piece of granite

The rock granite is very common in most parts of North America, and unless everything is currently covered with snow where you live, you should have no trouble finding a sample of it near you. The best places to look are pebbly ocean or lake beaches, a gravel bar of a creek or river, a gravel driveway, or somewhere where gravel has been used in landscaping. In Figure 1.4.3 , taken on a beach, the granitic pebbles are the ones that are predominantly light-colored with dark specks. The one in the very centre is a good example.



Figure 1.4.3

Select a sample of granite and, referring to Figure 1.4.2, see if you can identify some of the minerals in it. It may help to break it in half with a hammer to see a fresh surface, but be careful to protect your eyes if you do so. You should be able to see glassy-looking quartz, dull white plagioclase feldspar (and maybe pink potassium feldspar), and black hornblende or, in some cases, flaky black biotite mica (or both).

In addition to identifying the minerals in your granite, you might also try to describe the texture in terms of the range sizes of the mineral crystals (in millimeters) and the shapes of the crystals (some may be rectangular in outline, most will be irregular). Think about where your granite might have come from and how it got to where you found it.

See Appendix 3 for Exercise 1.1 answers.

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1. Terms in bold are defined in the glossary at the end of the book. ↵

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1.5: Fundamentals of Plate Tectonics

Plate tectonics is the model or theory that has been used for the past 60 years to understand and explain how the Earth works—more specifically the origins of continents and oceans, of folded rocks and mountain ranges, of earthquakes and volcanoes, and of continental drift. Plate tectonics is explained in some detail in Chapter 10, but is introduced here because it includes concepts that are important to many of the topics covered in the next few chapters.

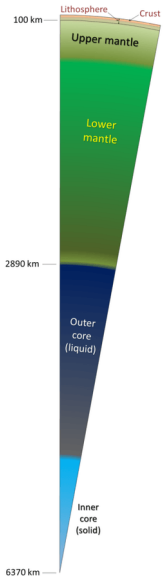


Figure 1.5.1 The components of the interior of the Earth (click on the image to see a full-size version).

Key to understanding plate tectonics is an understanding of Earth's internal structure, which is illustrated in Figure 1.5.1. Earth's **core** consists mostly of iron. The outer core is hot enough for the iron to be liquid. The inner core—although even hotter—is under so much pressure that it is solid. The **mantle** is made up of iron and magnesium **silicate** minerals. The bulk of the mantle surrounding the outer core is solid rock, but is plastic enough to be able to flow slowly. The outermost part of the mantle is rigid. The **crust**—composed mostly of granite on the continents and mostly of basalt beneath the oceans—is also rigid. The crust and outermost rigid mantle together make up the **lithosphere**. The lithosphere is divided into about 20 **tectonic plates** that move in different directions on Earth's surface.

An important property of Earth (and other planets) is that the temperature increases with depth, from close to 0°C at the surface to about 7000°C at the centre of the core. In the crust, the rate of temperature increase is about 30°C every kilometre. This is known as the **geothermal gradient**.

Heat is continuously flowing outward from Earth's interior, and the transfer of heat from the core to the mantle causes convection in the mantle (Figure 1.5.2). This convection is the primary driving force for the movement of tectonic plates. At places where convection currents in the mantle are moving upward, new lithosphere forms (at ocean ridges), and the plates move apart (diverge). Where two plates are converging (and the convective flow is downward), one plate will be **subducted** (pushed down) into the mantle beneath the other. Many of Earth's major earthquakes and volcanoes are associated with convergent boundaries.

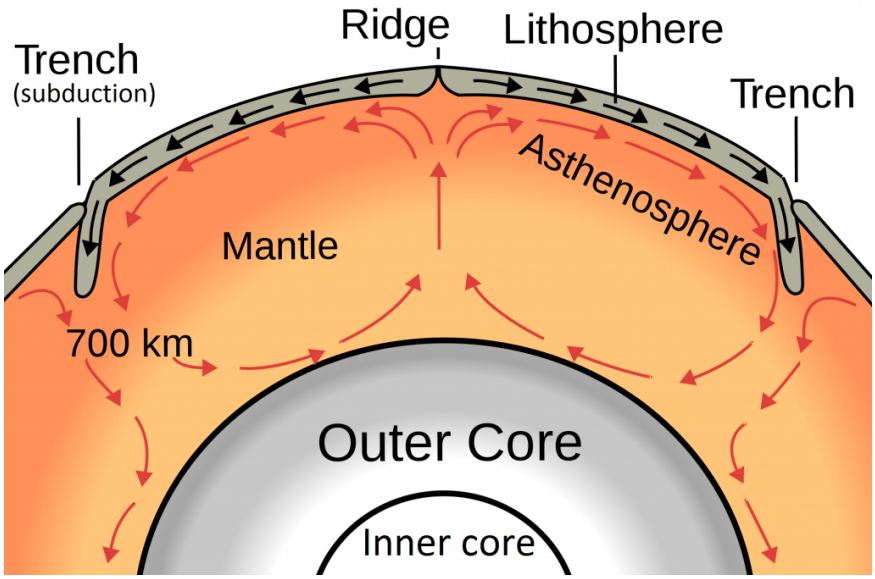


Figure 1.5.2 Depiction of the convection in the mantle and it's relationship to plate motion

Earth's major tectonic plates and the directions and rates at which they are diverging at sea-floor ridges, are shown in Figure 1.5.3 .

Exercise 1.2 Plate

Using either a map of the tectonic plates from the Internet or Figure 1.5.3 determine which tectonic plate you are on right now, approximately how fast it is moving, and in what direction. How far has that plate moved relative to Earth's core since you were born?

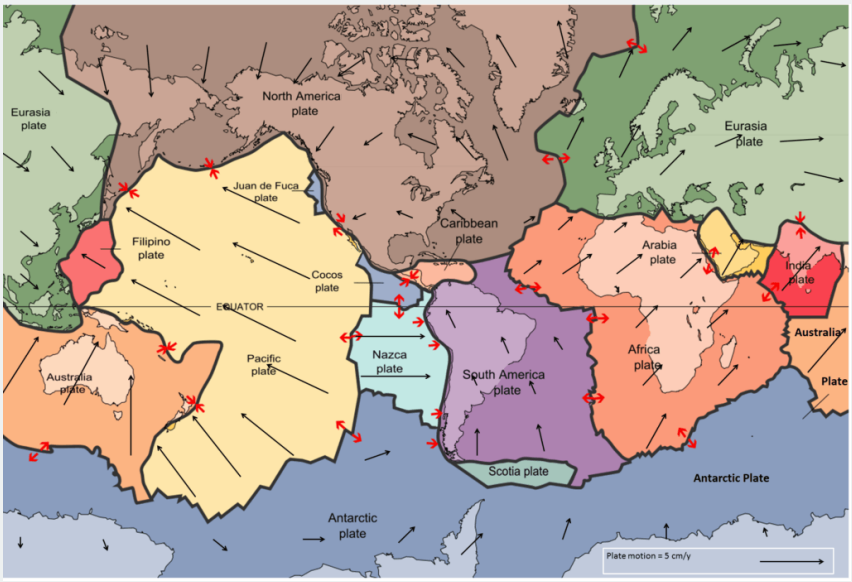


Figure 1.5.3 A map showing 15 of the Earth's tectonic plates and the approximate rates and directions of plate motions.

See Appendix 3 for Exercise 1.2 answers.

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1.6: Geological Time

In 1788, after many years of geological study, James Hutton, one of the great pioneers of geology, wrote the following about the age of Earth: *The result, therefore, of our present enquiry is, that we find no vestige of a beginning — no prospect of an end.*[1] Of course he wasn't exactly correct, there was a beginning and there will be an end to Earth, but what he was trying to express is that geological time is so vast that we humans, who typically live for less than a century, have no means of appreciating how much geological time there is. Hutton didn't even try to assign an age to Earth, but we now know that it is approximately 4,570 million years old. Using the scientific notation for geological time, that is 4,570 **Ma** (for *mega annum* or "millions of years") or 4.57 **Ga** (for *gigaannum* or billions of years). More recent dates can be expressed in **ka** (*kiloannum*); for example, the last cycle of glaciation ended at approximately 11.7 ka or 11,700 years ago. This notation will be used for geological dates throughout this book.

Exercise 1.3 Using geological time notation

To help you understand the scientific notation for geological time—which is used extensively in this book—write the following out in numbers (for example, 3.23 Ma = 3,230,000 years).

1. 2.75 ka
2. 0.93 Ga
3. 14.2 Ma

We use this notation to describe geological events in the same way that we might say "they arrived at 2 pm." For example, we can say "this rock formed at 45 Ma." But this notation is not used to express elapsed time. We don't say: "I studied for 4 pm for that test." And we don't say: "The dinosaurs lived for 160 Ma." Instead, we could say: "The dinosaurs lived from 225 Ma to 65 Ma, which is 160 million years."

See Appendix 3 for Exercise 1.3 answers.

Unfortunately, knowing how to express geological time doesn't really help us understand or appreciate its extent. A version of the geological time scale is included as Figure 1.6.1. Unlike time scales you'll see in other places, or even later in this book, this time scale is linear throughout its length, meaning that 50 Ma during the **Cenozoic** is the same thickness as 50 Ma during the **Hadean**—in each case about the height of the "M" in Ma. The Pleistocene glacial epoch began at about 2.6 Ma, which is equivalent to half the thickness of the thin grey line at the top of the yellow bar marked "Cenozoic." Most other time scales have earlier parts of Earth's history compressed so that more detail can be shown for the more recent parts. That makes it difficult to appreciate the extent of geological time.

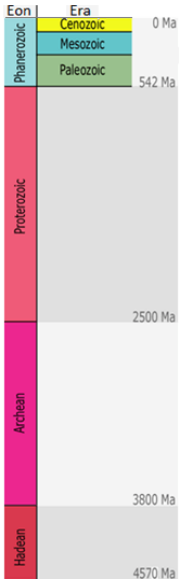


Figure 1.6.1 The geological time scale. 1.06: Geological Time#fig1.10

To create some context, the **Phanerozoic** Eon (the last 542 million years) is named for the time during which visible (*phaneros*) life (*zoi*) is present in the geological record. In fact, large organisms—those that leave fossils visible to the naked eye—have existed for a little longer than that, first appearing around 600 Ma, or a span of just over 13% of geological time. Animals have been on land for 360 million years, or 8% of geological time. Mammals have dominated since the demise of the dinosaurs around 65 Ma, or 1.5% of geological time, and the genus *Homo* has existed since approximately 2.8 Ma, or 0.06% (1/1,600th) of geological time.

Geologists (and geology students) need to understand geological time. That doesn't mean memorizing the geological time scale; instead, it means getting your mind around the concept that although most geological processes are extremely slow, very large and important things can happen if such processes continue for enough time.

For example, the Atlantic Ocean between Nova Scotia and northwestern Africa has been getting wider at a rate of about 2.5 centimeters (cm) per year. Imagine yourself taking a journey at that rate—it would be impossibly and ridiculously slow. And yet, since it started to form at around 200 Ma (just 4% of geological time), the Atlantic Ocean has grown to a width of over 5,000 kilometers (km)!

A useful mechanism for understanding geological time is to scale it all down into one year. The origin of the solar system and Earth at 4.57 Ga would be represented by January 1, and the present year would be represented by the last tiny fraction of a second on New Year's Eve. At this scale, each day of the year represents 12.5 million years; each hour represents about 500,000 years; each minute represents 8,694 years; and each second represents 145 years. Some significant events in Earth's history, as expressed on this time scale, are summarized on Table 1.1.

Table 1.1 A summary of some important geological dates expressed as if all of geological time was condensed into one year.

[Skip Table]

Event	Approximate Date	Calendar Equivalent
Formation of oceans and continents	4.5 to 4.4 Ga	January
Evolution of the first primitive life forms	3.8 Ga	early March
Formation of British Columbia's oldest rocks	2.0 Ga	July
Evolution of the first multi-celled animals	0.6 Ga or 600 Ma	November 15
Animals first crawled onto land	360 Ma	December 1
Vancouver Island reached North America and the Rocky Mountains were formed	90 Ma	December 25
Extinction of the non-avian dinosaurs	65 Ma	December 26
Beginning of the Pleistocene ice age	2 Ma or 2000 ka	8 p.m., December 31
Retreat of the most recent glacial ice from southern Canada	14 ka	11:58 p.m., December 31
Arrival of the first people in British Columbia	10 ka	11:59 p.m., December 31
Arrival of the first Europeans on the west coast of what is now Canada	250 years ago	2 seconds before midnight, December 31

Exercise 1.4 Take a trip through geological time

We're going on a road trip! Pack some snacks and grab some of your favourite music. We'll start in Tofino on Vancouver Island and head for the Royal Tyrrell Museum just outside of Drumheller, Alberta, 1,500 km away. Along the way, we'll talk about some important geological sites that we pass by, and we'll use the distance as a way of visualizing the extent of geological time. Of course it's just a "virtual" road trip, but it will be fun anyway. To join in, go to: Virtual Road Trip.

Once you've had a chance to do the road trip, answer these questions:

1. We need oxygen to survive, and yet the first presence of free oxygen (O_2 gas) in the atmosphere and the oceans was a "catastrophe" for some organisms. When did this happen and why was it a catastrophe?
2. Approximately how much time elapsed between the colonization of land by plants and animals?
3. Explain why the evolution of land plants was such a critical step in the evolution of life on Earth.

See Appendix 3 for Exercise 1.4 answers.

Image descriptions

Figure 1.6.1 image description: The Hadean eon (3800 Ma to 4570 Ma), Archean eon (2500 Ma to 3800 Ma), and Proterozoic eon (542 Ma to 2500 Ma) make up 88% of geological time. The Phanerozoic eon makes up the last 12% of geological time. The Phanerozoic eon (0 Ma to 542 Ma) contains the Paleozoic, Mesozoic, and Cenozoic eras. [Return to Figure 1.6.1]

Media Attributions

- Figure 1.6.1: © Steven Earle. CC BY.

1. Hutton, J, 1788. Theory of the Earth; or an investigation of the laws observable in the composition, dissolution, and restoration

of land upon the Globe. Transactions of the Royal Society of Edinburgh. ↵

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1.7: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
1.1 What is Geology?	Geology is the study of Earth. It is an integrated science that involves the application of many of the other sciences, but geologists also have to consider geological time because most of the geological features that we see today formed thousands, millions, or even billions of years ago.
1.2 Why Study Earth?	Geologists study Earth out of curiosity and for other more practical reasons, including understanding the evolution of life on Earth, searching for resources, understanding risks from geological events such as earthquakes, volcanoes, and slope failures, and documenting past environmental and climate changes so that we can understand how human activities are affecting Earth.
1.3 What Do Geologists Do?	Geologists work in the resource industries and in efforts to protect our natural resources and the environment in general. They are involved in ensuring that risks from geological events (e.g., earthquakes) are minimized and that the public understands what the risks are. Geologists are also engaged in fundamental research about Earth and in teaching.
1.4 Minerals and Rocks	Minerals are naturally occurring, specific combinations of elements that have particular three-dimensional structures. Rocks are made up of mixtures of minerals and can form though igneous, sedimentary, or metamorphic processes.
1.5 Fundamentals of Plate Tectonics	The Earth's mantle is convecting because it is being heated from below by the hot core. Those convection currents contribute to the movement of tectonic plates (which are composed of the crust and the uppermost rigid mantle). Plates are formed at divergent boundaries and consumed (subducted) at convergent boundaries. Many important geological processes take place at plate boundaries.
1.6 Geological Time	Earth is approximately 4,570,000,000 years old; that is, 4.57 billion years or 4.57 Ga or 4,570 Ma. It's such a huge amount of time that even extremely slow geological processes can have an enormous impact.

Questions for Review

Answers to Review Questions at the end of each chapter are provided in Appendix 2.

1. In what way is geology different from the other sciences, such as chemistry and physics?
2. How would some familiarity with biology be helpful to a geologist?
3. List three ways in which geologists can contribute to society.
4. Describe the lattice structure and elemental composition of the mineral halite.
5. In what way is a mineral different from a rock?
6. What is the main component of Earth's core?
7. What process leads to convection in the mantle?
8. How does mantle convection contribute to plate tectonics?
9. What are some of the processes that take place at a divergent plate boundary?
10. Dinosaurs first appear in the geological record in rocks at 225 Ma and then disappear at 65 Ma. For what proportion (%) of geological time did dinosaurs exist?
11. If a typical rate for the accumulation of sediments is 1 mm/year, what thickness (meters) of sedimentary rock could accumulate over a period of 30 million years?

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2: MINERALS

2.1: Electrons, Protons, Neutrons, and Atoms
2.2: Bonding and Lattices
2.3: Mineral Groups
2.4: Silicate Minerals
2.5: Formation of Minerals
2.6: Mineral Properties
2.7: Summary

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2.1: Electrons, Protons, Neutrons, and Atoms

All matter that we are familiar with, including mineral crystals, is made up of atoms, and all atoms are made up of three main particles: **protons**, **neutrons**, and **electrons**. As summarized in Table 2.1, protons are positively charged, neutrons are uncharged and electrons are negatively charged. The –1 charge of one electron balances the +1 charge of one proton. Both protons and neutrons have a mass of 1, while electrons have almost no mass.

Table 2.1 Charges and masses of the particles within atoms

Elementary Particle	Charge	Mass
Proton	+1	1
Neutron	0	1
Electron	–1	~0

The element hydrogen has the simplest atoms, each with just one proton and one electron. The proton forms the nucleus, while the electron orbits around it. All other elements have neutrons as well as protons in their nucleus, such as helium, which is depicted in Figure 2.1.1. The positively charged protons tend to repel each other, but the neutrons help to hold the nucleus together. The number of protons is the **atomic number**, and the number of protons plus neutrons is the **atomic mass**. For hydrogen, the atomic mass is 1 because there is one proton and no neutrons. For helium, it is 4: two protons and two neutrons.

For most of the 16 lightest elements (up to oxygen) the number of neutrons is equal to the number of protons. For most of the remaining elements there are more neutrons than protons because extra neutrons are needed to keep the nucleus together by overcoming the mutual repulsion of the increasing numbers of protons concentrated in a very small space. For example, silicon has 14 protons and 14 neutrons. Its atomic number is 14 and its atomic mass is 28. The most common isotope of uranium has 92 protons and 146 neutrons. Its atomic number is 92 and its atomic mass is 238 (92 + 146).

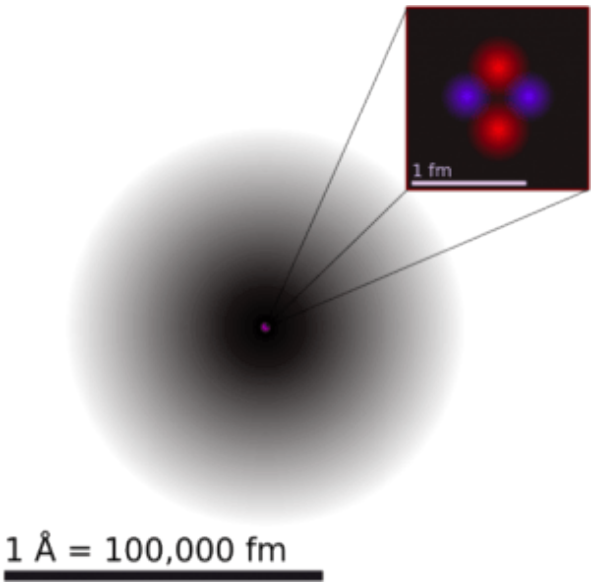


Figure 2.1.1 A depiction of a helium atom.

A helium atom is depicted on Figure 2.1.1. The dot in the middle is the nucleus, and the surrounding cloud represents where the two electrons might be at any time. The darker the shade, the more likely that an electron will be there. The helium atom is about 1 angstrom across. An angstrom (Å) is 10^{-10} meters (m). The helium nucleus is about 1 femtometre across. A femtometre (fm) is 10^{-15} m. In other words, a helium atom's electron cloud is about 100,000 times bigger than its nucleus. Stanley Park in Vancouver is about 2 km across. If Stanley Park was a helium atom, the nucleus would be the size of a walnut.

Electrons orbiting around the nucleus of an atom are arranged in shells—also known as "energy levels." The first shell can hold

only two electrons, while the next shell holds up to eight electrons. Subsequent shells can hold more electrons, but the outermost shell of any atom holds no more than eight electrons. As we'll see, the electrons in the outermost shell play an important role in bonding between atoms. The electron shell configurations for 29 of the first 36 elements are listed in Table 2.2.

*Table 2.2 Electron shell configurations up to element 36. (The inert elements, with filled outer shells, have a *.)*

[Skip Table]

Element	Symbol	Atomic No.	Number of Electrons in the 1st Shell	Number of Electrons in the 2nd Shell	Number of Electrons in the 3rd Shell	Number of Electrons in the 4th Shell
Hydrogen	H	1	1	0	0	0
Helium *	He	2	2	0	0	0
Lithium	Li	3	2	1	0	0
Beryllium	Be	4	2	2	0	0
Boron	B	5	2	3	0	0
Carbon	C	6	2	4	0	0
Nitrogen	N	7	2	5	0	0
Oxygen	O	8	2	6	0	0
Fluorine	F	9	2	7	0	0
Neon *	Ne	10	2	8	0	0
Sodium	Na	11	2	8	1	0
Magnesium	Mg	12	2	8	2	0
Aluminum	Al	13	2	8	3	0
Silicon	Si	14	2	8	4	0
Phosphorus	P	15	2	8	5	0
Sulphur	S	16	2	8	6	0
Chlorine	Cl	17	2	8	7	0
Argon *	Ar	18	2	8	8	0
Potassium	K	19	2	8	8	1
Calcium	Ca	20	2	8	8	2
Scandium	Sc	21	2	8	9	2
Titanium	Ti	22	2	8	10	2
Vanadium	V	23	2	8	11	2
Chromium	Cr	24	2	8	13	1
Manganese	Mn	25	2	8	13	2
Iron	Fe	26	2	8	14	2
.	.	.	.	.	.	.
Selenium	Se	34	2	8	18	6
Bromine	Br	35	2	8	18	7
Krypton *	Kr	36	2	8	18	8

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Media Attributions

- Figure 2.1.1: Helium Atom QM. © Yzmo. CC BY-SA.

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2.2: Bonding and Lattices

As we've just seen, an atom seeks to have a full outer shell (i.e., eight electrons for most elements, or two electrons for hydrogen and helium) to be atomically stable. This is accomplished by transferring or sharing electrons with other atoms.

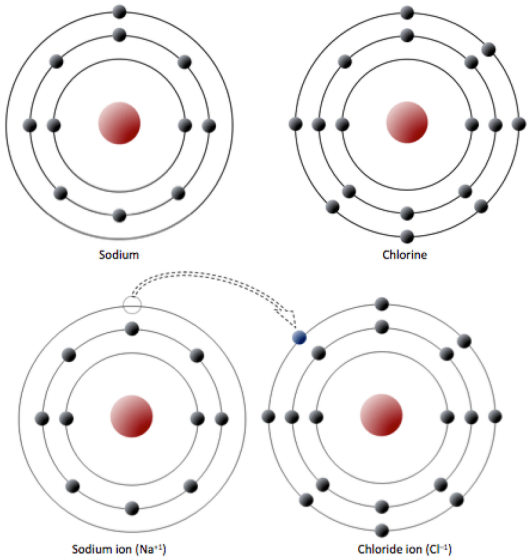


Figure 2.2.1 A very simplified electron configuration of sodium and chlorine atoms (top). Sodium gives up an electron to become a cation (bottom left) and chlorine accepts an electron to become an anion (bottom right).

Sodium has 11 electrons: two in the first shell, eight in the second, and one in the third (Figure 2.2.1). Sodium readily gives up that single third-shell electron, and when it loses this one negative charge, it becomes positively charged (because it now has 11 protons and only 10 electrons). By giving up its lone third-shell electron, sodium ends up with a full outer shell. Chlorine, on the other hand, has 17 electrons: two in the first shell, eight in the second, and seven in the third. Chlorine readily accepts an eighth electron to fill its third shell, and therefore becomes negatively charged because it has 17 protons and 18 electrons. In changing their number of electrons, these atoms become **ions**—the sodium loses an electron to become a positive ion or **cation**, and the chlorine gains an electron to become a negative ion or **anion** (Figure 2.2.1).

Since negative and positive charges attract, sodium and chlorine ions can stick together, creating an **ionic bond**. Electrons can be thought of as being *transferred* from one atom to another in an ionic bond. Common table salt (NaCl) is a mineral composed of chlorine and sodium linked together by ionic bonds (Figure 2.2.1). The mineral name for NaCl is halite.

An element like chlorine can also form bonds without forming ions. For example, two chlorine atoms, which each seek an eighth electron in their outer shell, can share an electron in what is known as a **covalent bond** to form chlorine gas (Cl_2) (Figure 2.2.2). Electrons are *shared* in a covalent bond.

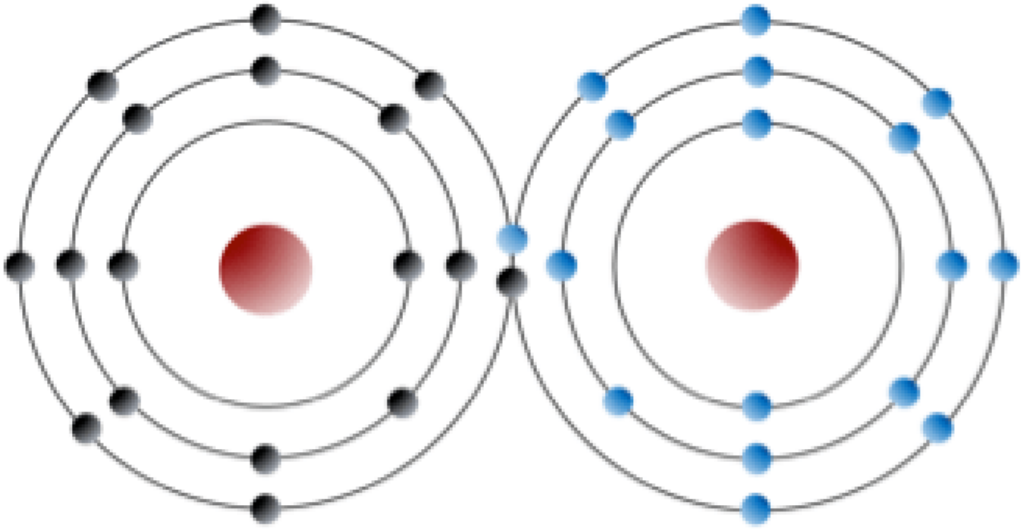


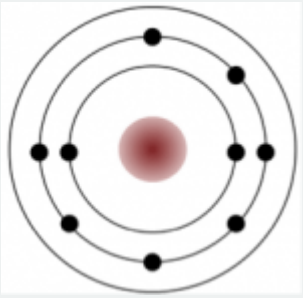
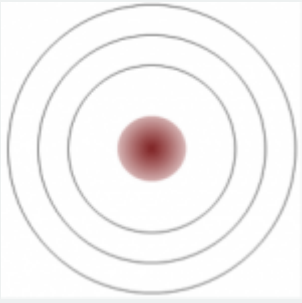
Figure 2.2.2 Depiction of a covalent bond between two chlorine atoms. The electrons are black in the left atom and blue in the right atom. Two electrons are shared (one black and one blue) so that each atom "appears" to have a full outer shell.

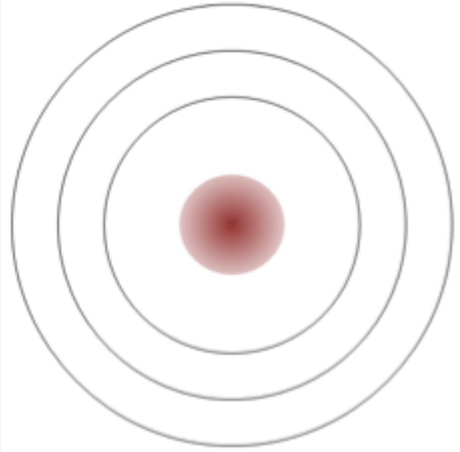
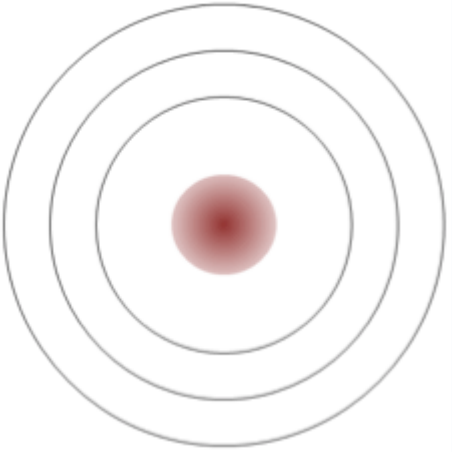
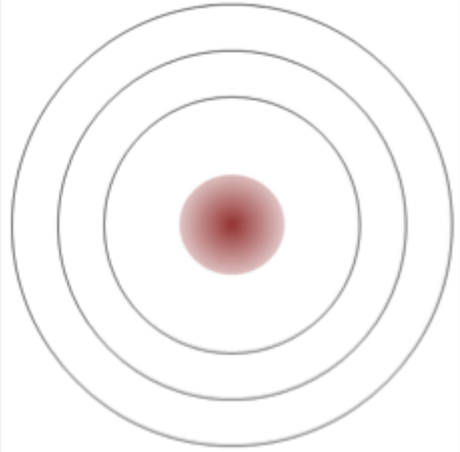
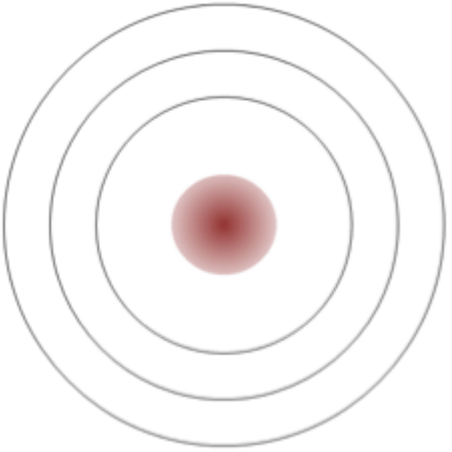
Exercise 2.1 Cations, anions, and ionic bonding

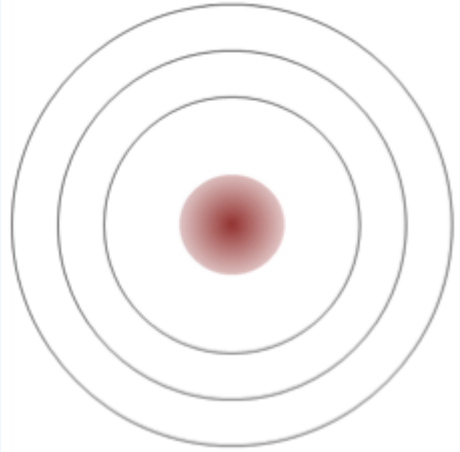
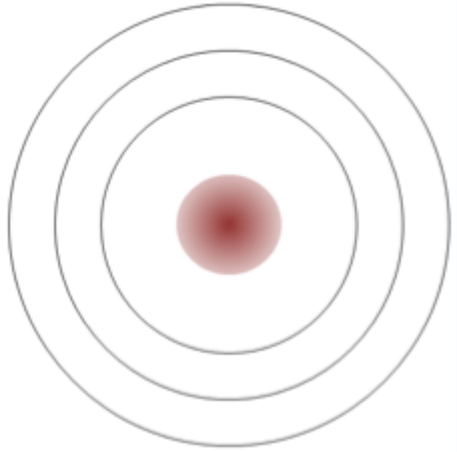
A number of elements are listed below along with their atomic numbers. Assuming that the first electron shell can hold two electrons and subsequent electron shells can hold eight electrons, sketch in the electron configurations for these elements. Predict whether the element is likely to form a cation (+) or an anion (–) when electron transfer takes place, and what charge it would have (e.g., +1, +2, –1).

The first one is done for you. Fluorine needed an extra electron to have 8 in its outermost shell, and in gaining that electron it became negatively charged.

See Appendix 3 for Exercise 2.1 answers.

Fluorine (9)	**Lithium (3)**
	
anion (–1)	_____

Magnesium (12)  _____	**Argon (18)**  _____
Chlorine (17)  _____	**Beryllium (3)**  _____

Oxygen (8)	Sodium (11)
	
_____	_____

A carbon atom has six protons and six electrons; two of the electrons are in the inner shell and four in the outer shell (Figure 2.2.3). Carbon would need to gain or lose four electrons to have a filled outer shell, and this would create too great a charge imbalance for the ion to be stable. On the other hand, carbon can share electrons to create covalent bonds. In the mineral diamond, the carbon atoms are linked together in a three-dimensional framework, where each carbon atom is bonded to four other carbon atoms and every bond is a very strong covalent bond. In the mineral graphite, the carbon atoms are linked together in sheets or layers (Figure 2.2.3), and each carbon atom is covalently bonded to three others. Graphite-based compounds, which are strong because of the strong intra-layer covalent bonding, are used in high-end sports equipment such as ultralight racing bicycles. Graphite itself is soft because the bonding between these layers is relatively weak, and it is used in a variety of applications, including lubricants and pencils.

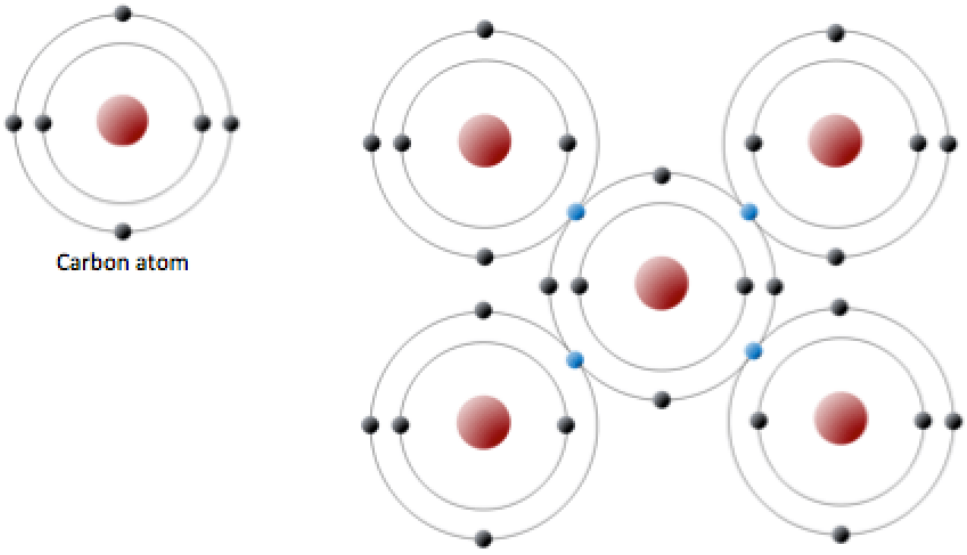


Figure 2.2.3 The electron configuration of carbon (left) and the sharing of electrons in covalent bonding of diamond (right). The electrons shown in blue are shared between adjacent Carbon atoms. Although shown here in only two dimensions, diamond has a three-dimensional structure as shown on Figure 2.2.5. [Image description]

Silicon and oxygen bond together to create a **silica tetrahedron**, which is a four-sided pyramid shape with O at each corner and Si in the middle (Figure 2.2.4). This structure is the building block of the silicate minerals (which are described in Section 2.4). The bonds in a silica tetrahedron have some of the properties of covalent bonds and some of the properties of ionic bonds. As a result

of the ionic character, silicon becomes a cation (with a charge of +4) and oxygen becomes an anion (with a charge of –2). The net charge of a silica tetrahedron (SiO_4) is: 4 + 4(–2) = 4 – 8 = –4. As we will see later, silica tetrahedra (plural of *tetrahedron*) link together in a variety of ways to form most of the common minerals of the crust.

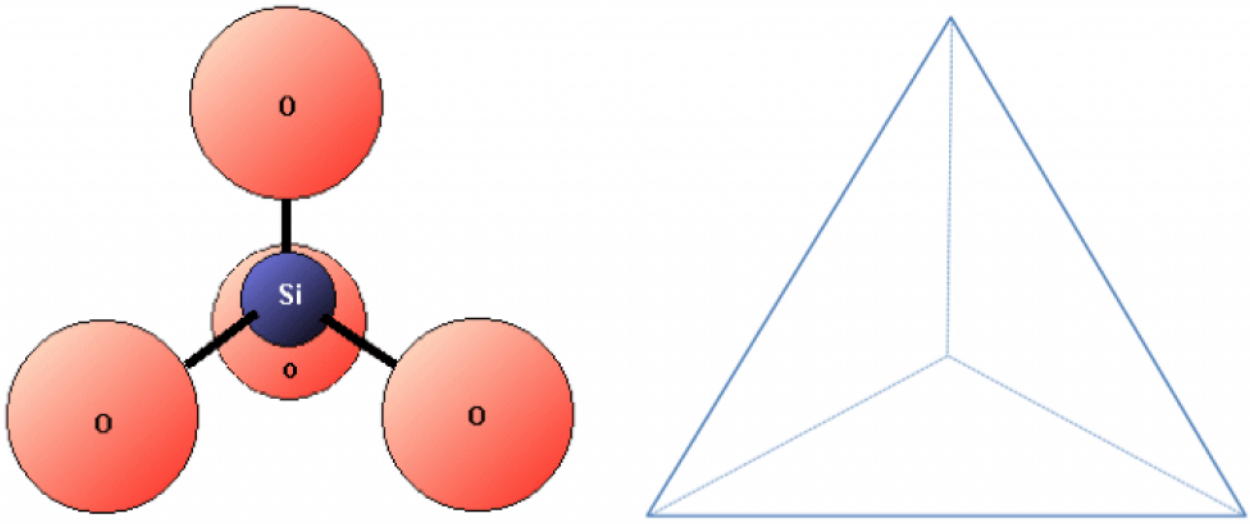


Figure 2.2.4 The silica tetrahedron, the building block of all silicate minerals. (Because the silicon ion has a charge of +4 and the four oxygen ions each have a charge of –2, the silica tetrahedron has a net charge of –4.)

Most minerals are characterized by ionic bonds, covalent bonds, or a combination of the two, but there are other types of bonds that are important in minerals, including metallic bonds and weaker electrostatic forces (hydrogen or Van der Waals bonds). Metallic elements have outer electrons that are relatively loosely held. (The metals are highlighted on the periodic table in Appendix 1.) When bonds between such atoms are formed, these electrons can move freely from one atom to another. A metal can thus be thought of as an array of positively charged atomic nuclei immersed in a sea of mobile electrons. This feature accounts for two very important properties of metals: their electrical conductivity and their malleability (they can be deformed and shaped).

Molecules that are bonded ionically or covalently can also have other weaker electrostatic forces holding them together. Examples of this are the force holding graphite sheets together and the attraction between water molecules.

The element **silicon** is one of the most important geological elements and is the second-most abundant element in Earth's crust (after oxygen). Silicon bonds readily with oxygen to form a **silica** tetrahedron (Figure 2.2.4). Pure silicon crystals (created in a lab) are used to make semi-conductive media for electronic devices. A **silicate** mineral is one in which silicon and oxygen are present as silica tetrahedra. Silica also refers to a chemical component of a rock and is expressed as % SiO_2. The mineral quartz is made up entirely of silica tetrahedra, and some forms of quartz are also known as "silica". **Silicone** is a synthetic product (e.g., silicone rubber, resin, or caulking) made from silicon-oxygen chains and various organic molecules. To help you keep the "sili" names straight, here is a summary table:

Table 2.3 Summary of "Sili" names

"Sili" name	Definition
Silicon	The 14th element
Silicon wafer	A crystal of pure silicon sliced very thinly and used for electronics
Silica tetrahedron	A combination of one silicon atom and four oxygen atoms that form a tetrahedron
% silica	The proportion of a rock that is composed of the component SiO_2
Silica	A solid made out of SiO_2 (but not necessarily a mineral – e.g., opal)

"Sili" name	Definition
Silicate	A mineral that contains silica tetrahedra (e.g., quartz, feldspar, mica, olivine)
Silicone	A flexible synthetic material made up of Si–O chains with attached organic molecules

Elements that have a full outer shell are described as **inert** because they do not tend to react with other elements to form compounds. That's because they don't need to lose or gain any electrons to become stable, and so they don't become ions. They all appear in the far-right column of the periodic table. Examples are: helium, neon, argon, etc.

As described in Chapter 1, all minerals are characterized by a specific three-dimensional pattern known as a lattice or crystal structure. These structures range from the simple cubic pattern of halite (NaCl) (Figure 2.2.1), to the very complex patterns of some silicate minerals. Two minerals may have the same composition, but very different crystal structures and properties. Graphite and diamond, for example, are both composed only of carbon, but while diamond is the hardest substance known, graphite is softer than paper. Their lattice structures are compared in Figure 2.2.5.

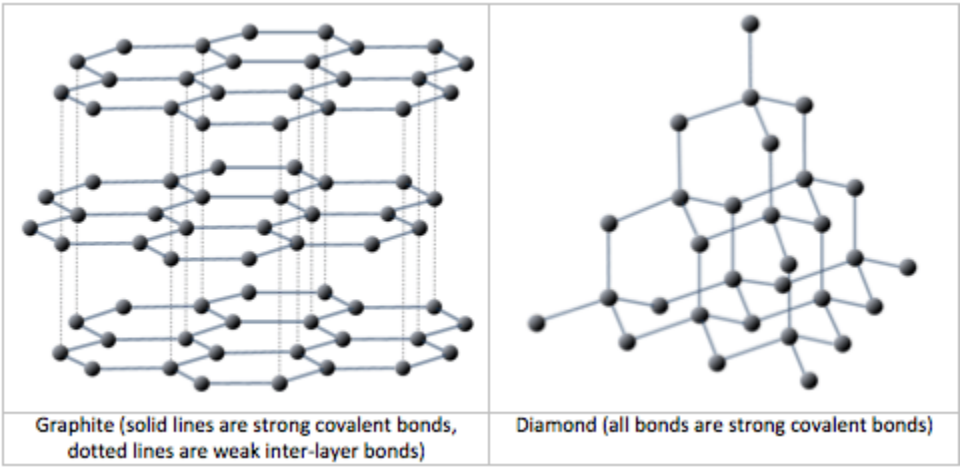


Figure 2.2.5 A depiction of the lattices of graphite and diamond.

Mineral lattices have important implications for mineral properties, as exemplified by the hardness of diamond and the softness of graphite. Lattices also determine the shape that mineral crystals grow in and how they break. For example, the right angles in the lattice of the mineral halite (Figure 2.2.1) influence both the shape of its crystals (cubic), and the way those crystals break (Figure 2.2.6).



Figure 2.2.6 Cubic crystals (left) and right-angle cleavage planes (right) of the mineral halite. If you look closely at the cleavage fragment on the right, you can see where it would break again (cleave) along a plane parallel to an existing surface. In most minerals, cleavage planes do not align with crystal surfaces.

Image Descriptions

Figure 2.2.1 image description: Sodium has one electron in its outer shell and chlorine has 7 electrons in it its outer shell. Sodium's one outer electron goes to chlorine which makes Chlorine slightly negative and Sodium slightly positive. They attract each other and together they form Sodium Chloride. [Return to Figure 2.2.1]

Figure 2.2.3 image description: (Left) A carbon atom has two electrons in its inner shell and four electrons in its outer shell. (Right) One Carbon atom shares electrons with four other carbon atoms to form a complete outer shell. [Return to Figure 2.2.3]

Media Attributions

- Figures 2.2.1, 2.2.2, 2.2.3, 2.2.4, 2.2.5 and 2.2.6 (right): © Steven Earle. CC BY.
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2.3: Mineral Groups

Most minerals are made up of a cation (a positively charged ion) or several cations, plus an anion (a negatively charged ion (e.g., S^{2-})) or an anion complex (e.g., SO_4^{2-}). For example, in the mineral hematite (Fe_2O_3), the cation is Fe^{3+} (iron) and the anion is O^{2-} (oxygen). The two Fe^{3+} ions have an overall charge of +6 and that balances the overall charge of −6 from the three O^{2-} ions.

We group minerals into classes on the basis of their predominant anion or anion complex. These include oxides, sulfides, carbonates, silicates, and others. Silicates are by far the predominant group in terms of their abundance within the crust and mantle. (They will be discussed in Section 2.4). Some examples of minerals from the different mineral groups are given in Table 2.4.

Table 2.4 The main mineral groups and some examples of minerals in each group.

[Skip Table]

Group	Examples
Oxides	Hematite (iron oxide Fe_2O_3), corundum (aluminum oxide Al_2O_3), water ice (H_2O)
Sulphides	Galena (lead sulfide PbS), pyrite (iron sulfide FeS_2), chalcopyrite (copper-iron sulfide $CuFeS_2$)
Sulphates	Gypsum (calcium sulphate $CaSO_4{\cdot}H_2O$), barite (barium sulphate $BaSO_4$) *(Note that sulphates are different from sulfides. Sulphates have the SO_4^{-2} ion while sulfides have the S^{-2} ion)*
Halides	Fluorite (calcium flouride CaF_2), halite (sodium chloride NaCl) *(Halide minerals have halogen elements as their anion — the minerals in the second last column on the right side of the periodic table, including F, Cl, Br, etc. — see the periodic table in Appendix 1: List of Geologically Important Elements and the Periodic Table.)*
Carbonates	Calcite (calcium carbonate $CaCO_3$), dolomite (calcium-magnesium carbonate $(Ca,Mg)CO_3$)
Phosphates	Apatite ($Ca_5(PO_4)3(OH)$), Turquoise ($CuAl_6(PO_4)4(OH)_8{\cdot}5H_2O$)
Silicates	Quartz (SiO_2), feldspar (sodium-aluminum silicate $NaAlSi_3O_8$), olivine (iron or magnesium silicate $(Mg,Fe)_2SiO_4$) *(Note that in quartz the anion is oxygen, and while it could be argued, therefore, that quartz is an oxide, it is always classed with the silicates.)*
Native minerals	Gold (Au), diamond (C), graphite (C), sulfur (S), copper (Cu)

Oxide minerals have oxygen (O^{2-}) as their anion, but they exclude those with oxygen complexes such as carbonate (CO_3^{2-}), sulphate (SO_4^{2-}), and silicate (SiO_4^{4-}). The most important oxides are the iron oxides hematite and magnetite (Fe_2O_3 and Fe_3O_4, respectively). Both of these are common ores of iron. Corundum (Al_2O_3) is used as an abrasive, but can also be a gemstone in its ruby and sapphire varieties. If the oxygen is also combined with hydrogen to form the hydroxyl anion (OH^-) the mineral is known as a **hydroxide**. Some important hydroxides are limonite and bauxite, which are ores of iron and aluminium respectively. Frozen water (H_2O) is a mineral (an oxide), but liquid water is not because it doesn't have a regular lattice.

Sulphides are minerals with the S^{-2} anion, and they include galena (PbS), sphalerite (ZnS), chalcopyrite ($CuFeS_2$), and molybdenite (MoS_2), which are the most important ores of lead, zinc, copper, and molybdenum respectively. Some other sulfide minerals are pyrite (FeS_2), bornite (Cu_5FeS_4), stibnite (Sb_2S_3), and arsenopyrite (FeAsS).

Sulphates are minerals with the SO_4^{-2} anion, and these include anhydrite ($CaSO_4$) and its cousin gypsum ($CaSO_4.2H_2O$) and the sulphates of barium and strontium: barite ($BaSO_4$) and celestite ($SrSO_4$). In all of these minerals, the cation has a +2 charge, which balances the −2 charge on the sulphate ion.

The **halides** are so named because the anions include the **halogen** elements chlorine, fluorine, bromine, etc. Examples are halite (NaCl), cryolite (Na_3AlF_6), and fluorite (CaF_2).

The **carbonates** include minerals in which the anion is the CO_3^{-2} complex. The carbonate combines with +2 cations to form minerals such as calcite ($CaCO_3$), magnesite ($MgCO_3$), dolomite ($(Ca,Mg)CO_3$)[1], and siderite ($FeCO_3$). The copper minerals malachite and azurite are also carbonates.

In **phosphate** minerals, the anion is the PO_4^{-3} complex. An important phosphate mineral is apatite ($Ca_5(PO_4)_3(OH)$), which is what your teeth are made of. Note that it is called a phosphate, not a hydroxide, even though it has a hydroxyl ion.

The **silicate** minerals include the elements silicon and oxygen in varying proportions ranging from Si : O_2 to Si : O_4. These are discussed at length in Section 2.4.

Native minerals are single-element minerals, such as gold, copper, sulfur, and graphite.

Exercise 2.2 Mineral groups

We classify minerals according to the anion part of the mineral formula, and mineral formulas are always written with the anion part on the right. For example, for pyrite (FeS_2), Fe^{2+} is the cation and S^- is the anion. This helps us to know that it's a sulfide, but it is not always that obvious. Hematite (Fe_2O_3) is an oxide; that's easy, but anhydrite ($CaSO_4$) is a sulphate because SO_4^{2-} is the anion, not O. Along the same lines, calcite ($CaCO_3$) is a carbonate, and olivine (Mg_2SiO_4) is a silicate. Minerals with only one element (such as S) are native minerals, while those with an anion from the halogen column of the periodic table (Cl, F, Br, etc.) are halides. Provide group names for the following minerals:

Table 2.5 Provide group names for the following minerals

[Skip Table]

Name	Formula	Group
sphalerite	ZnS	
magnetite	Fe_3O_4	
pyroxene	$MgSiO_3$	
anglesite	$PbSO_4$	
sylvite	KCl	
silver	Ag	
fluorite	CaF_2	
ilmenite	$FeTiO_3$	
siderite	$FeCO_3$	
feldspar	$KAlSi_3O_8$	
sulfur	S	
xenotime	YPO_4	

See Appendix 3 for Exercise 2.2 answers.

1. The notations of two (or more) elements enclosed in parentheses with a comma between them: (Ca,Mg), indicates that both can be present, in varying proportions, but that there is still only one of them for each anion present. ↵

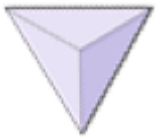
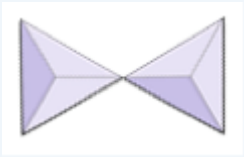
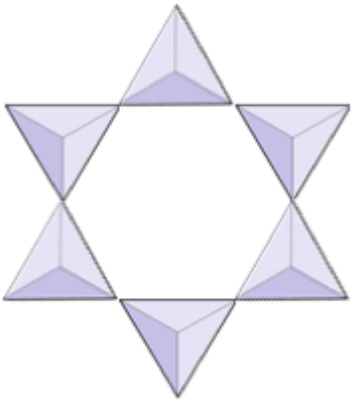
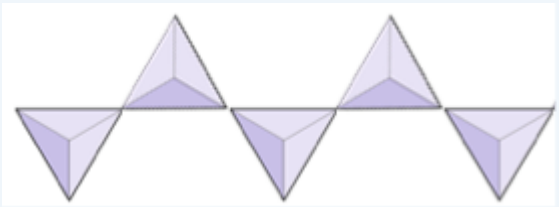
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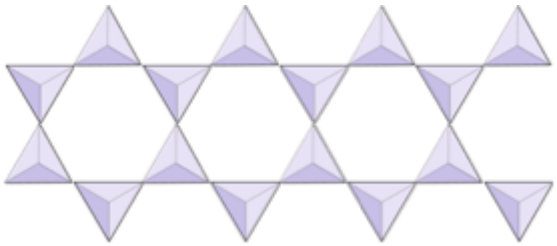
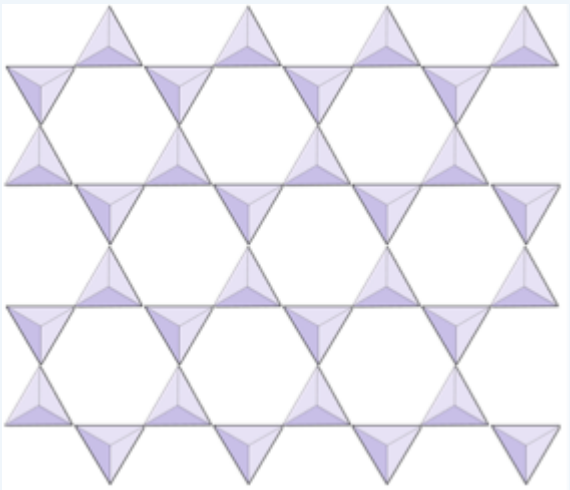
2.4: Silicate Minerals

The vast majority of the minerals that make up the rocks of Earth's crust are silicate minerals. These include minerals such as quartz, feldspar, mica, amphibole, pyroxene, olivine, and a variety of clay minerals. The building block of all of these minerals is the **silica tetrahedron**, a combination of four oxygen atoms and one silicon atom. As we've seen, it's called a tetrahedron because planes drawn through the oxygen atoms form a shape with 4 surfaces (Figure 2.4.4). Since the silicon ion has a charge of 4 and each of the four oxygen ions has a charge of –2, the silica tetrahedron has a net charge of –4.

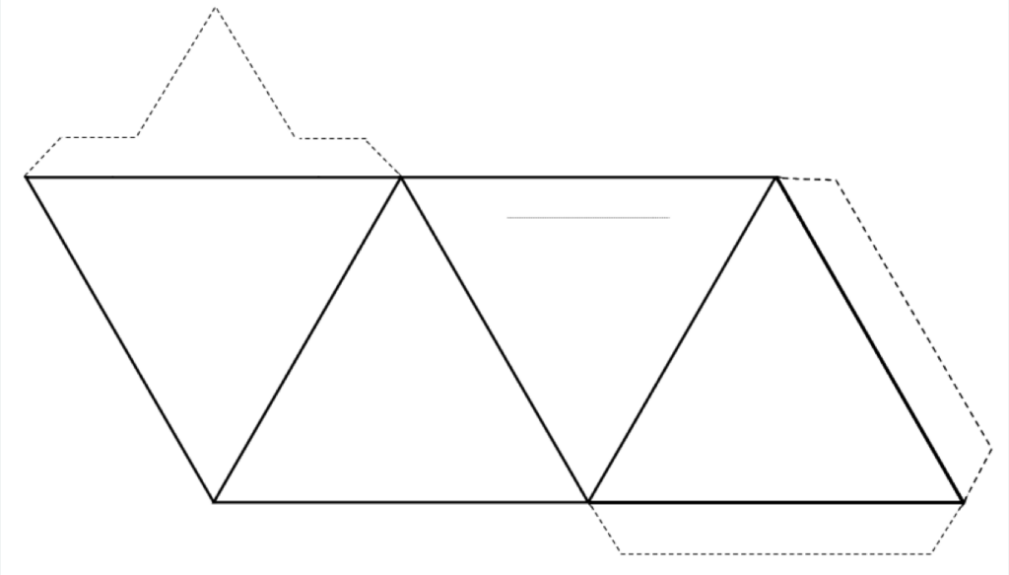
In silicate minerals, these tetrahedra are arranged and linked together in a variety of ways, from single units to complex frameworks (Table 2.6). The simplest silicate structure, that of the mineral **olivine**, is composed of isolated tetrahedra bonded to iron and/or magnesium ions. In olivine, the –4 charge of each silica tetrahedron is balanced by two **divalent** (i.e., +2) iron or magnesium cations. Olivine can be either Mg_2SiO_4 or Fe_2SiO_4, or some combination of the two $(Mg,Fe)_2SiO_4$. The divalent cations of magnesium and iron are quite close in radius (0.73 versus 0.62 angstroms[1]). Because of this size similarity, and because they are both divalent cations (both can have a charge of +2), iron and magnesium can readily substitute for each other in olivine and in many other minerals.

Table 2.6 Silicate mineral configurations. The triangles represent silica tetrahedra.

Tetrahedron Configuration Picture	Tetrahedron Configuration Name	Example Minerals
	Isolated (nesosilicates)	Olivine, garnet, zircon, kyanite
	Pairs (sorosilicates)	Epidote, zoisite
	Rings (cyclosilicates)	Tourmaline
	Single chains (inosilicates)	Pyroxenes, wollastonite

Tetrahedron Configuration Picture	Tetrahedron Configuration Name	Example Minerals
	Double chains (inosilicates)	Amphiboles
	Sheets (phyllosilicates)	Micas, clay minerals, serpentine, chlorite
3-dimensional structure	Framework (tectosilicates)	Feldspars, quartz, zeolite

Exercise 2.3 Make a Tetrahedron



Cut around the outside of the shape (solid lines and dotted lines), and then fold along the solid lines to form a tetrahedron. If you have glue or tape, secure the tabs to the tetrahedron to hold it together. If you don't have glue or tape, make a slice along the thin grey line and insert the pointed tab into the slit.

If you are doing this in a classroom, try joining your tetrahedron with others into pairs, rings, single and double chains, sheets, and even three-dimensional frameworks.

See Appendix 3 for Exercise 2.3 answers.

In olivine, unlike most other silicate minerals, the silica tetrahedra are not bonded to each other. Instead they are bonded to the iron and/or magnesium ions, in the configuration shown on Figure 2.4.1.

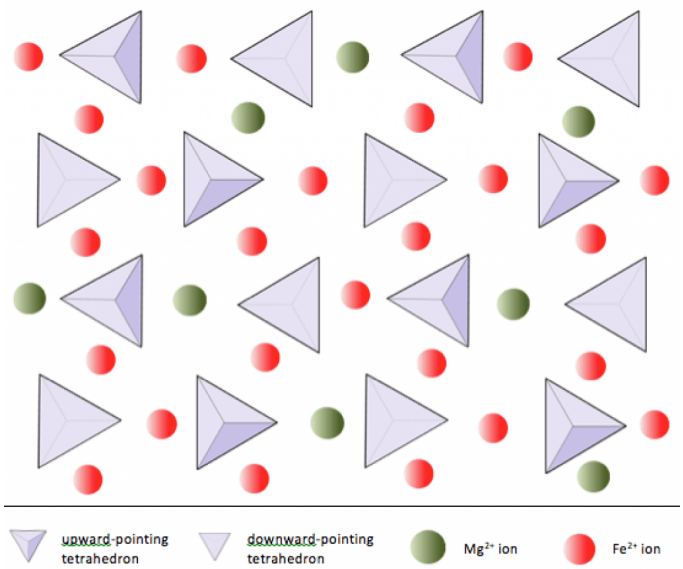


Figure 2.4.1 A depiction of the structure of olivine as seen from above. The formula for this particular olivine, which has three Fe ions for each Mg ion, could be written: $Mg_{0.5}Fe_{1.5}SiO_4$.

As already noted, the 2 ions of iron and magnesium are similar in size (although not quite the same). This allows them to substitute for each other in some silicate minerals. In fact, the ions that are common in silicate minerals have a wide range of sizes, as depicted in Figure 2.4.2. All of the ions shown are cations, except for oxygen. Note that iron can exist as both a +2 ion (if it loses two electrons during ionization) or a +3 ion (if it loses three). Fe^{2+} is known as **ferrous** iron. Fe^{3+} is known as **ferric** iron. Ionic radii are critical to the composition of silicate minerals, so we'll be referring to this diagram again.

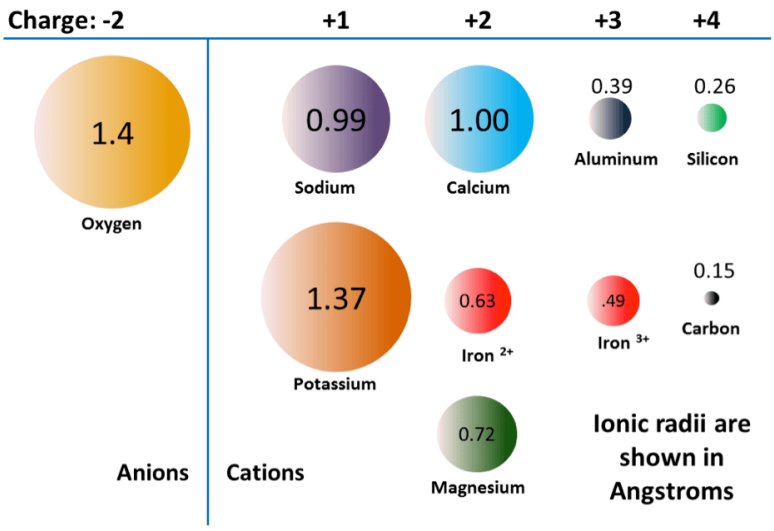


Figure 2.4.2 The ionic radii (effective sizes) in angstroms, of some of the common ions in silicate minerals.

The structure of the single-chain silicate pyroxene is shown on Figures 2.4.3 and 2.4.4. In **pyroxene**, silica tetrahedra are linked

together in a single chain, where one oxygen ion from each tetrahedron is shared with the adjacent tetrahedron, hence there are fewer oxygens in the structure. The result is that the oxygen-to-silicon ratio is lower than in olivine (3:1 instead of 4:1), and the net charge per silicon atom is less (–2 instead of –4). Therefore, fewer cations are necessary to balance that charge. Pyroxene compositions are of the type $MgSiO_3$, $FeSiO_3$, and $CaSiO_3$, or some combination of these. Pyroxene can also be written as $(Mg,Fe,Ca)SiO_3$, where the elements in the brackets can be present in any proportion. In other words, pyroxene has one cation for each silica tetrahedron (e.g., $MgSiO_3$) while olivine has two (e.g., Mg_2SiO_4). Because each silicon ion is +4 and each oxygen ion is –2, the three oxygens (–6) and the one silicon (+4) give a net charge of –2 for the single chain of silica tetrahedra. In pyroxene, the one divalent cation (2) per tetrahedron balances that –2 charge. In olivine, it takes two divalent cations to balance the –4 charge of an isolated tetrahedron.The structure of pyroxene is more “permissive” than that of olivine—meaning that cations with a wider range of ionic radii can fit into it. That’s why pyroxenes can have iron (radius 0.63 Å) or magnesium (radius 0.72 Å) or calcium (radius 1.00 Å) cations (see Figure 2.4.2 above).

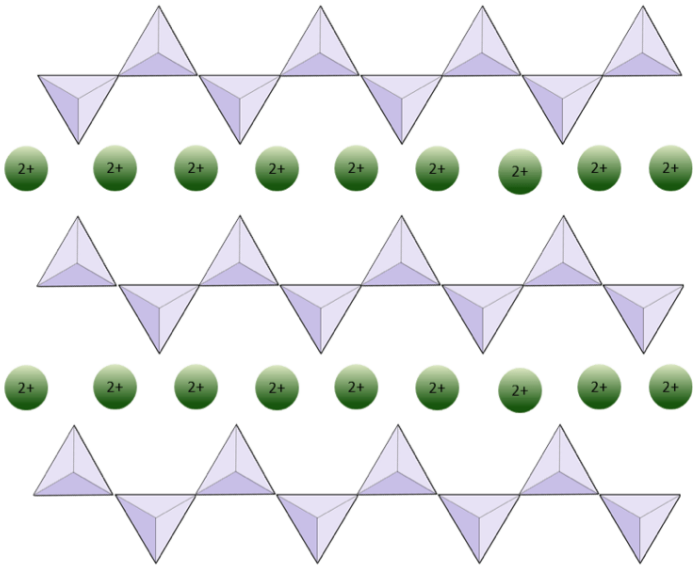


Figure 2.4.3 A depiction of the structure of pyroxene. The tetrahedral chains continue to left and right and each is interspersed with a series of divalent cations. If these are Mg ions, then the formula is $MgSiO_3$.

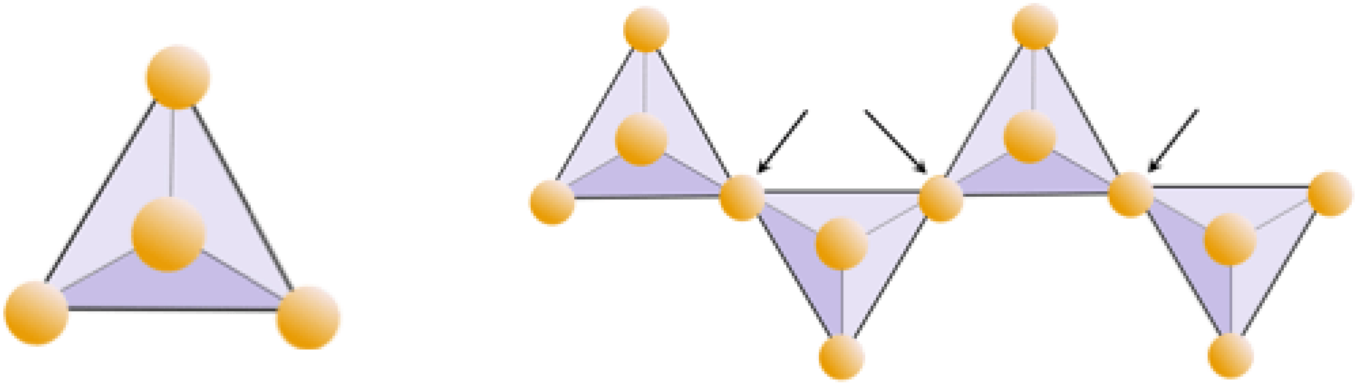
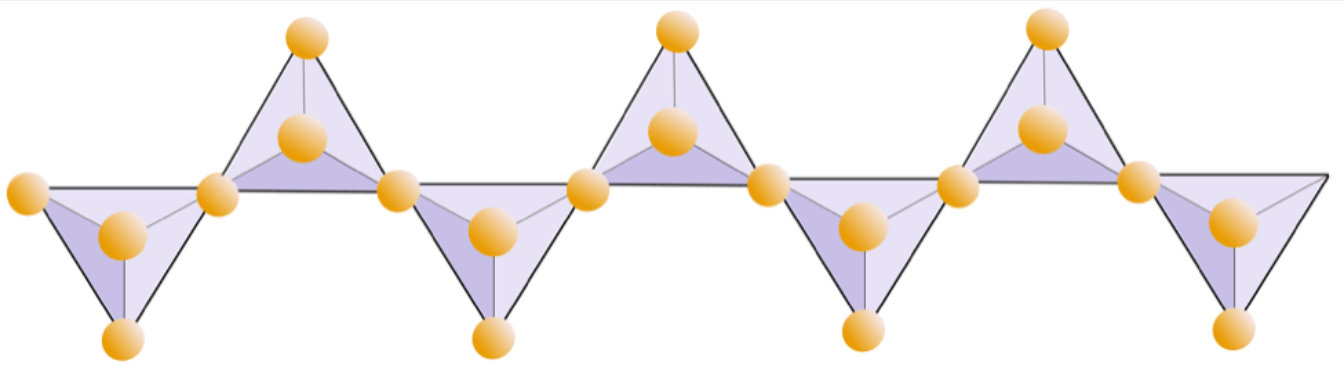


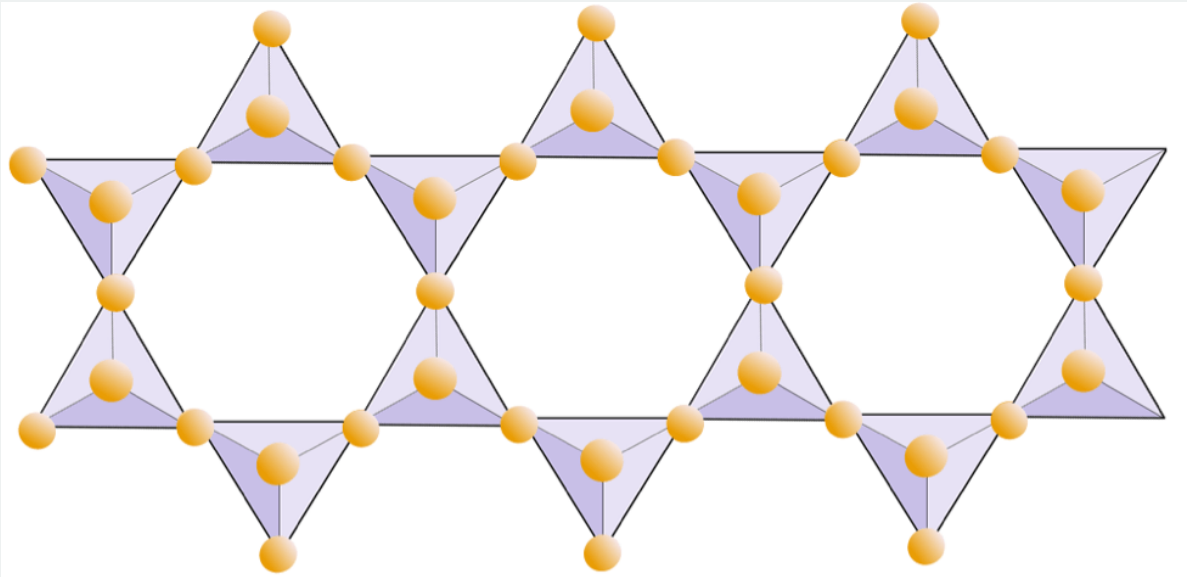
Figure 2.4.4 A single silica tetrahedron (left) with four oxygen ions per silicon ion (SiO_4). Part of a single chain of tetrahedra (right), where the oxygen atoms at the adjoining corners are shared between two tetrahedra (arrows). For a very long chain the resulting ratio of silicon to oxygen is 1 to 3 (SiO_3).

Exercise 2.4 Oxygen deprivation

The diagram below represents a single chain in a silicate mineral. Count the number of tetrahedra versus the number of oxygen ions (yellow spheres). Each tetrahedron has one silicon ion so this should give you the ratio of Si to O in single-chain silicates (e.g., pyroxene).



The diagram below represents a double chain in a silicate mineral. Again, count the number of tetrahedra versus the number of oxygen ions. This should give you the ratio of Si to O in double-chain silicates (e.g., amphibole).



See Appendix 3 for Exercise 2.4 answers.

In **amphibole** structures, the silica tetrahedra are linked in a double chain that has an oxygen-to-silicon ratio lower than that of pyroxene, and hence still fewer cations are necessary to balance the charge. Amphibole is even more permissive than pyroxene and its compositions can be very complex. Hornblende, for example, can include sodium, potassium, calcium, magnesium, iron, aluminum, silicon, oxygen, fluorine, and the hydroxyl ion (OH^-).

In **mica** structures, the silica tetrahedra are arranged in continuous sheets, where each tetrahedron shares three oxygen anions with adjacent tetrahedra. There is even more sharing of oxygens between adjacent tetrahedra and hence fewer cations are needed to balance the charge of the silica-tetrahedra structure in sheet silicate minerals. Bonding between sheets is relatively weak, and this accounts for the well-developed one-directional cleavage in micas (Figure 2.4.5). **Biotite** mica can have iron and/or magnesium in it and that makes it a **ferromagnesian** silicate mineral (like olivine, pyroxene, and amphibole). **Chlorite** is another similar mineral that commonly includes magnesium. In **muscovite** mica, the only cations present are aluminum and potassium; hence it is a non-ferromagnesian silicate mineral.

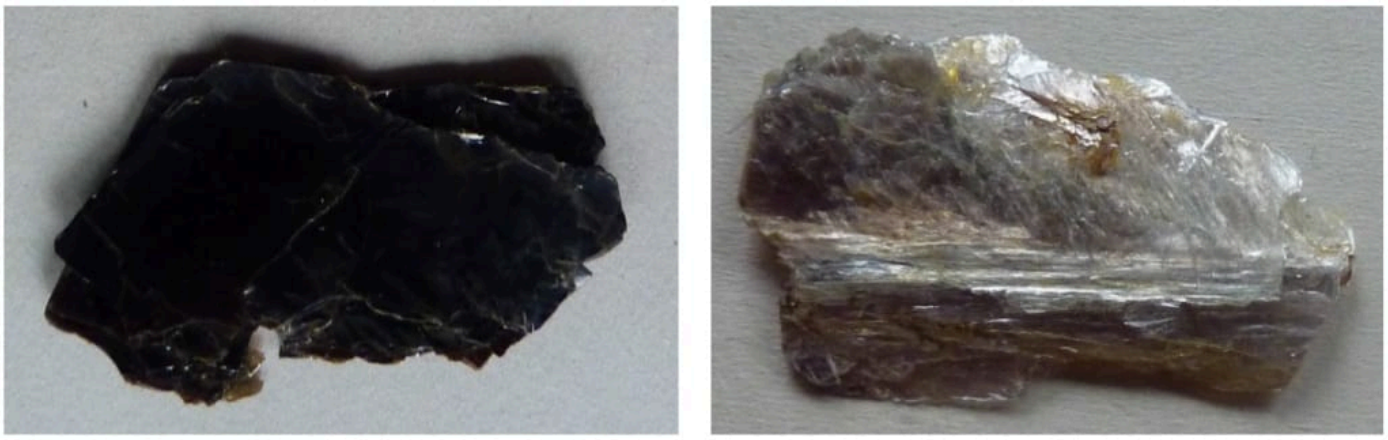


Figure 2.4.5 Biotite mica (left) and muscovite mica (right). Both are sheet silicates and split easily into thin layers along planes parallel to the sheets. Biotite is dark like the other iron- and/or magnesium-bearing silicates (e.g., olivine, pyroxene, and amphibole), while muscovite is light colored. (Each sample is about 3 cm across.)

Apart from muscovite, biotite, and chlorite, there are many other **sheet silicates** (a.k.a. **phyllosilicates**), many of which exist as clay-sized fragments (i.e., less than 0.004 millimeters). These include the clay minerals **kaolinite**, **illite**, and **smectite**, and although they are difficult to study because of their very small size, they are extremely important components of rocks and especially of soils.

All of the sheet silicate minerals also have water molecules within their structure.

Silica tetrahedra are bonded in three-dimensional frameworks in both the **feldspars** and **quartz**. These are **non-ferromagnesian minerals**—they don't contain any iron or magnesium. In addition to silica tetrahedra, feldspars include the cations aluminum, potassium, sodium, and calcium in various combinations. Quartz contains only silica tetrahedra.

The three main **feldspar** minerals are **potassium feldspar**, (a.k.a. K-feldspar or K-spar) and two types of plagioclase feldspar: **albite** (sodium only) and **anorthite** (calcium only). As is the case for iron and magnesium in olivine, there is a continuous range of compositions (solid solution series) between albite and anorthite in plagioclase. Because the calcium and sodium ions are almost identical in size (1.00 Å versus 0.99 Å) any intermediate compositions between $CaAl_2Si_3O_8$ and $NaAlSi_3O_8$ can exist (Figure 2.4.6). This is a little bit surprising because, although they are very similar in size, calcium and sodium ions don't have the same charge (Ca^{2+} versus Na^{+}). This problem is accounted for by the corresponding substitution of Al^{+3} for Si^{+4} . Therefore, albite is $NaAlSi_3O_8$ (1 Al and 3 Si) while anorthite is $CaAl_2Si_2O_8$ (2 Al and 2 Si), and plagioclase feldspars of intermediate composition have intermediate proportions of Al and Si. This is called a "coupled-substitution."

The intermediate-composition plagioclase feldspars are oligoclase (10% to 30% Ca), andesine (30% to 50% Ca), labradorite (50% to 70% Ca), and bytownite (70% to 90% Ca). **K-feldspar** ($KAlSi_3O_8$) has a slightly different structure than that of plagioclase, owing to the larger size of the potassium ion (1.37 Å) and because of this large size, potassium and sodium do not readily substitute for each other, except at high temperatures. These high-temperature feldspars are likely to be found only in volcanic rocks because intrusive igneous rocks cool slowly enough to low temperatures for the feldspars to change into one of the lower-temperature forms.

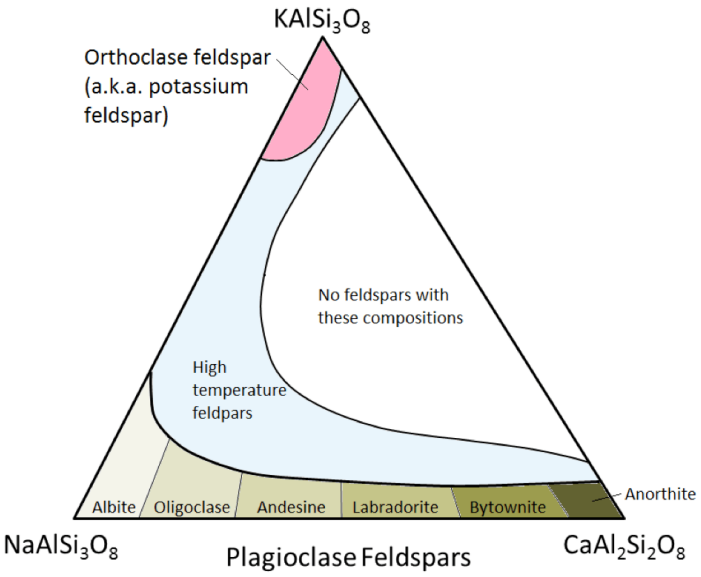


Figure 2.4.6 Compositions of the feldspar minerals.

In **quartz** (SiO_2)**,** the silica tetrahedra are bonded in a "perfect" three-dimensional framework. Each tetrahedron is bonded to four other tetrahedra (with an oxygen shared at every corner of each tetrahedron), and as a result, the ratio of silicon to oxygen is 1:2. Since the one silicon cation has a +4 charge and the two oxygen anions each have a –2 charge, the charge is balanced. There is no need for aluminum or any of the other cations such as sodium or potassium. The hardness and lack of cleavage in quartz result from the strong covalent/ionic bonds characteristic of the silica tetrahedron.

Exercise 2.5 Ferromagnesian silicates?

Silicate minerals are classified as being either ferromagnesian or non-ferromagnesian depending on whether or not they have iron (Fe) and/or magnesium (Mg) in their formula. A number of minerals and their formulas are listed below. For each one, indicate whether or not it is a *ferromagnesian silicate*.

Mineral	**Formula**	**Ferromagnesian silicate?**
olivine	$(Mg,Fe)_2SiO_4$	.
pyrite	FeS_2	.
plagioclase feldspar	$CaAl_2Si_2O_8$	.
pyroxene	$MgSiO_3$	.
hematite	Fe_2O_3	.
orthoclase feldspar	$KAlSi_3O_8$	.
quartz	SiO_2	.
amphibole	$Fe_7Si_8O_{22}(OH)_2$	.
muscovite	$K_2Al_4Si_6Al_2O_{20}(OH)_4$	.
magnetite	Fe_3O_4	.
biotite	$K_2Fe_4Al_2Si_6Al_4O_{20}(OH)_4$	.
dolomite	$(Ca,Mg)CO_3$	.
garnet	$Fe_2Al_2Si_3O_{12}$	.
serpentine	$Mg_3Si_2O_5(OH)_4$	.

See Appendix 3 for Exercise 2.5 answers.*Some of the formulas, especially the more complicated ones, have been simplified.

Image Descriptions

Figure 2.4.2 image description: The ionic radii of elements in angstroms and their charges.

Element	Ionic Radii (in angstroms)	Charge
Oxygen	1.4	–2 (Anion)
Potassium	1.37	1 (Cation)
Calcium	1.00	2 (Cation)
Sodium	0.99	1 (Cation)
Magnesium	0.72	2 (Cation)
Iron	0.63	2 (Cation)
	0.49	3 (Cation)
Aluminum	0.39	3 (Cation)
Silicon	0.26	4 (Cation)
Carbon	0.15	4 (Cation)

[Return to Figure 2.4.2]

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1. An angstrom is the unit commonly used for the expression of atomic-scale dimensions. One angstrom is 10^{-10} meters or 0.0000000001 meters. The symbol for an angstrom is Å. ↵

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2.5: Formation of Minerals

In order for a mineral crystal to grow, the elements needed to make it must be present in the appropriate proportions, the physical and chemical conditions must be favourable, and there must be sufficient time for the atoms to become arranged.

Physical and chemical conditions include factors such as temperature, pressure, presence of water, pH, and amount of oxygen available. Time is one of the most important factors because it takes time for atoms to become ordered. If time is limited, the mineral grains will remain very small. The presence of water enhances the mobility of ions and can lead to the formation of larger crystals over shorter time periods.

Most of the minerals that make up tehe rocks around us formed through the cooling of molten rock, known as **magma**. At the high temperatures that exist deep within Earth, some geological materials are liquid. As magma rises up through the crust, either by volcanic eruption or by more gradual processes, it cools and minerals crystallize. If the cooling process is rapid (minutes, hours, days, or years), the components of the minerals will not have time to become ordered and only small crystals can form before the rock becomes solid. The resulting rock will be fine-grained (i.e., with crystals less than 1 mm). If the cooling is slow (from decades to millions of years), the degree of ordering will be higher and relatively large crystals will form. In some cases, the cooling will be so fast (seconds) that the texture will be glassy, which means that no crystals at all form. **Volcanic glass** is not composed of minerals because the magma has cooled too rapidly for crystals to grow, although over time (millions of years) the volcanic glass may crystallize into various silicate minerals.

Minerals can also form in several other ways:

- Precipitation from aqueous solution (i.e., from hot water flowing underground, from evaporation of a lake or inland sea, or in some cases, directly from seawater) (see Exercise 2.7 below),
- Precipitation from gaseous emanations,
- Metamorphism: formation of new minerals directly from the elements within existing minerals under conditions of elevated temperature and pressure,
- Weathering: during which minerals unstable at Earth's surface may be altered to other minerals,
- Organic formation: formation of minerals within shells (primarily calcite) and teeth and bones (primarily apatite) by organisms (these organically formed minerals are still called minerals because they can also form inorganically).

Opal is a mineraloid (i.e., not an actual mineral) because although it has all of the other properties of a mineral, it does not have a specific structure. Pearl is not a mineral because it can *only* be produced by organic processes.

Exercise 2.7 Making crystals from solution



Figure 2.5.1

Place about ½ teaspoon (~2.5 cm^3) of any kind of table salt into a small bowl. Add about 2 teaspoons (~10 mL) of very hot water

and swirl it around for a few minutes until all or almost all of the salt has dissolved. (Be careful not to splash yourself with the hot water.)

Place the bowl in a safe place (windowsill, bookshelf), and check back every 24 hours to see what has happened. Depending on the level of humidity in the room, you should see crystals forming within 24 hours, and all of the water should be gone, with reasonably large crystals formed, within about 3 days. They should look a little like those shown here. In other words, they should be cubes.

Now try the same experiment again, but this time put the salt and water into a small sauce pan on the stove top at the lowest heat possible. Within 10 to 20 minutes all of the water should be gone and you should be left with some very small salt crystals—too small to even see their shapes. It takes time for mineral crystals to form.

The global demand for lithium has increased dramatically in the past decade, and will increase even more in the future as long as there is increasing demand for lithium-ion batteries in electronic devices, electric vehicles and for storage of solar- and wind-generated energy. Most of the world's lithium supply comes from salt lakes (*salars* in Spanish) like the one shown below in southwestern Bolivia.



Figure 2.5.2

The salty water of this and other lakes in the region has enough lithium in to make it a viable source of the metal, especially because, in the dry climate, that concentration can be increased by more evaporation. When this water is evaporated lithium crystallizes out as the mineral lithium carbonate (Li_2CO_3). For use in batteries the lithium is converted to other mineral forms—such as lithium cobalt oxide or lithium iron phosphate.

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2.6: Mineral Properties

Minerals are universal. A crystal of hematite on Mars will have the same properties as one on Earth, and the same as one on a planet orbiting another star. That's good news for geology students who are planning interplanetary travel since we can use those properties to help us identify minerals anywhere. That doesn't mean that it's easy, however; identification of minerals takes a lot of practice. Some of the mineral properties that are useful for identification are as follows: color, streak, lustre, hardness, crystal habit, cleavage/fracture, density and a few others.

Colour

For most of us, color is one of our key ways of identifying objects. While some minerals have particularly distinctive colors that make good diagnostic properties, many do not, and for many, color is simply unreliable. The mineral sulfur (2.6.1 left) is always a distinctive and unique yellow. Hematite, on the other hand, is an example of a mineral for which color is not diagnostic. In some forms hematite is deep dull red, but in others it is black and shiny metallic (Figure 2.6.2). Many other minerals can have a wide range of colors (e.g., quartz, feldspar, amphibole, fluorite, and calcite). In most cases, the variations in colors are a result of varying proportions of trace elements within the mineral. In the case of quartz, for example, yellow quartz (citrine) has trace amounts of ferric iron (Fe^{3+}), rose quartz has trace amounts of manganese, purple quartz (amethyst) has trace amounts of iron, and milky quartz, which is very common, has millions of fluid inclusions (tiny cavities, each filled with water).

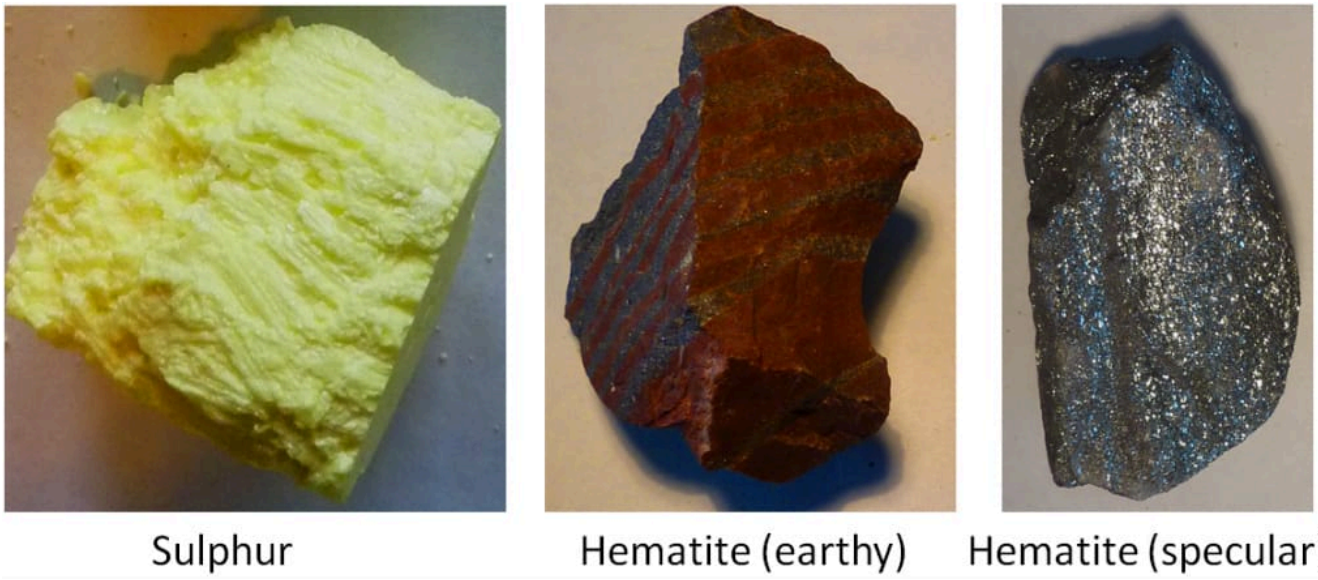


Figure 2.6.1 Examples of the colors of the minerals sulfur and hematite.

Streak

In the context of minerals, "color" is what you see when light reflects off the surface of the sample. One reason that color can be so variable is that the type of surface is variable. It may be a crystal face or a fracture surface or a cleavage plane, and the crystals may be large or small depending on the nature of the rock. If we grind a small amount of the sample to a powder we get a much better indication of its actual color. This can easily be done by scraping a corner of the sample across a streak plate (a piece of unglazed porcelain) to make a **streak**. The result is that some of the mineral gets ground to a powder and we can get a better impression of its "true" color (Figure 2.6.2).

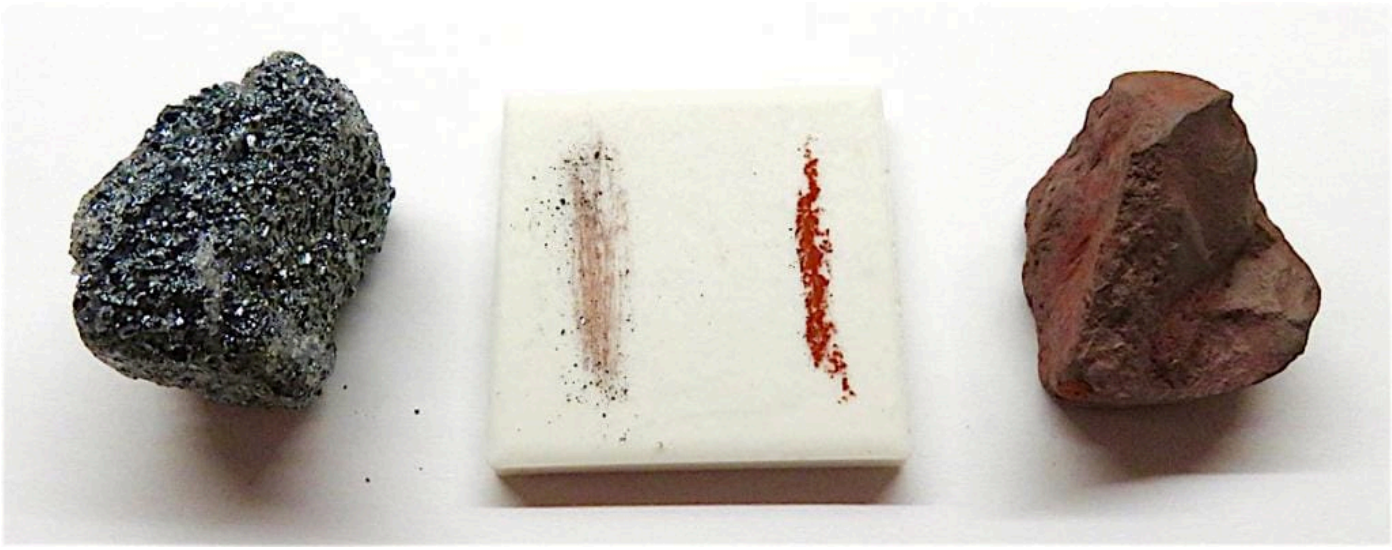


Figure 2.6.2 The streak colors of specular (metallic) hematite (left) and earthy hematite (right). Hematite leaves a distinctive reddish-brown streak whether the sample is metallic or earthy.

Lustre

Lustre is the way light reflects off the surface of a mineral, and the degree to which it penetrates into the interior. The key distinction is between **metallic** and **non-metallic lustres**. Light does not pass through metals, and that is the main reason they look "metallic." Even a thin sheet of metal—such as aluminum foil—will not allow light to pass through it. Many non-metallic minerals may look as if light will not pass through them, but if you take a closer look at a thin edge of the mineral you can see that it does. If a non-metallic mineral has a shiny, reflective surface, then it is called "glassy." If it is dull and non-reflective, it is "earthy." Other types of non-metallic lustres are "silky," "pearly," and "resinous." Lustre is a good diagnostic property since most minerals will always appear either metallic or non-metallic. There are a few exceptions to this (e.g., hematite in Figure 2.6.1).

Hardness

One of the most important diagnostic properties of a mineral is its hardness. In 1812 German mineralogist Friedrich Mohs came up with a list of 10 reasonably common minerals that had a wide range of hardnesses. These minerals are shown in Figure 2.6.3 , with the Mohs scale of hardness along the bottom axis. In fact, while each mineral on the list is harder than the one before it, the relative measured hardnesses (vertical axis) are not linear. For example apatite is about three times harder than fluorite and diamond is three times harder than corundum. Some commonly available reference materials are also shown on this diagram, including a typical fingernail (2.5), a piece of copper wire (3.5), a knife blade or a piece of window glass (5.5), a hardened steel file (6.5), and a porcelain streak plate (7). These are tools that a geologist can use to measure the hardness of unknown minerals. For example, if you have a mineral that you can't scratch with your fingernail, but you can scratch with a copper wire, then its hardness is between 2.5 and 3.5. And of course the minerals themselves can be used to test other minerals.

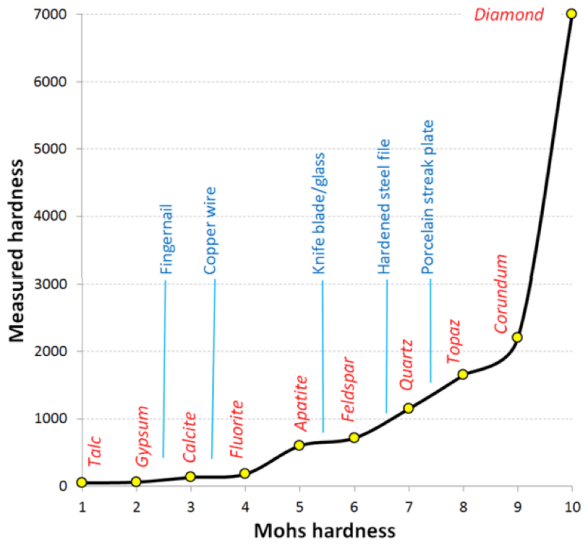


Figure 2.6.3 Minerals and reference materials in the Mohs scale of hardness. The “measured hardness” values are Vickers Hardness numbers.

Crystal Habit

When minerals form within rocks, there is a possibility that they will form in distinctive crystal shapes if they formed slowly and if they are not crowded out by other pre-existing minerals. Every mineral has one or more distinctive crystal **habits**, but it is not that common, in ordinary rocks, for the shapes to be obvious. Quartz, for example, will form six-sided prisms with pointed ends (Figure 2.6.4a), but this typically happens only when it crystallizes from a hot water solution within a cavity in an existing rock. Pyrite can form cubic crystals (Figure 2.6.4b), but can also form crystals with 12 faces, known as **dodecahedra** (“dodeca” means 12). The mineral garnet also forms dodecahedral crystals (Figure 2.6.4c).



Figure 2.6.4a Hexagonal prisms of quartz.



Figure 2.6.4b Cubic crystals of pyrite.



Figure 2.6.4c A dodecahedral crystal of garnet.

Because well-formed crystals are rare in ordinary rocks, habit isn't as useful a diagnostic feature as one might think. However, there are several minerals for which it is important. One is garnet, which is common in some metamorphic rocks and typically displays the dodecahedral shape. Another is amphibole, which forms long thin crystals, and is common in igneous rocks like granite (Figure 2.6.2).

Mineral habit is often related to the regular arrangement of the molecules that make up the mineral. Some of the terms that are used to describe habit include bladed, botryoidal (grape-like), dendritic (branched), drusy (an encrustation of minerals), equant (similar in all dimensions), fibrous, platy, prismatic (long and thin), and stubby.

Cleavage and Fracture

Crystal habit is a reflection of how a mineral grows, while cleavage and fracture describe how it breaks. Cleavage and fracture are the most important diagnostic features of many minerals, and often the most difficult to understand and identify. **Cleavage** is what we see when a mineral breaks along a specific plane or planes, while **fracture** is an irregular break. Some minerals tend to cleave along planes at various fixed orientations, some do not cleave at all (they only fracture). Minerals that have cleavage can also fracture along surfaces that are not parallel to their cleavage planes.

As we've already discussed, the way that minerals break is determined by their atomic arrangement and specifically by the orientation of weaknesses within the lattice. Graphite and the micas, for example, have cleavage planes parallel to their sheets (Figures 2.2.5 and 2.4.5), and halite has three cleavage planes parallel to the lattice directions (Figure 2.6.6).

Quartz has no cleavage because it has equally strong Si–O bonds in all directions, and feldspar has two cleavages at 90° to each other (Figure 2.6.5).

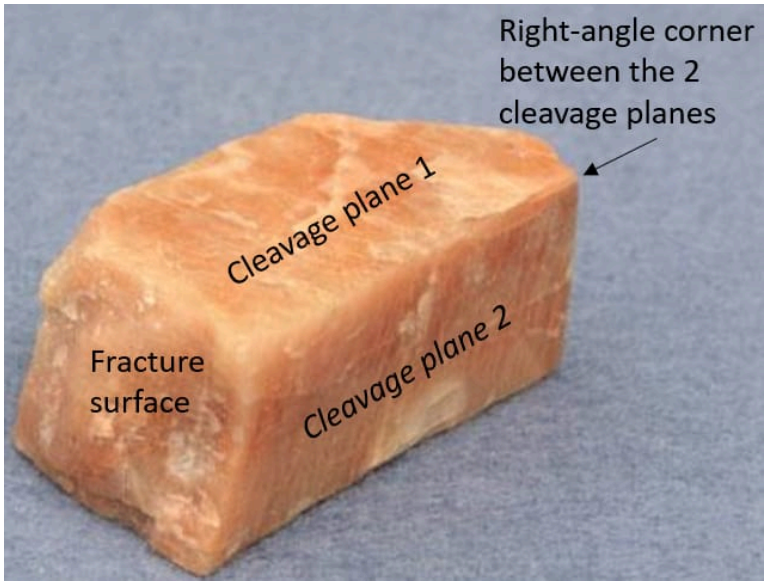


Figure 2.6.5 Cleavage and fracture in potassium feldspar

One of the main difficulties with recognizing and describing cleavage is that it is visible only in individual crystals. Most rocks have small crystals and it's very difficult to see the cleavage within those crystals. Geology students have to work hard to understand and recognize cleavage, but it's worth the effort since it is a reliable diagnostic property for most minerals.

One last thing: it is important to recognize the difference between cleavage planes and crystal surfaces. As already noted, crystal surfaces are related to how a mineral grows while cleavage planes are related to how it breaks. In most minerals cleavage planes and crystal surfaces do not align with one-another. An exception is halite, which grows in cubic crystals and has cleavage along those same planes (Figure 2.6.1 and 2.2.6). But this doesn't hold for most minerals. Quartz has crystal surfaces but no cleavage at all. Fluorite forms cubic crystals like those of halite, but it cleaves along planes that differ in orientation from the crystal surfaces. This is illustrated in Figure 2.6.6.

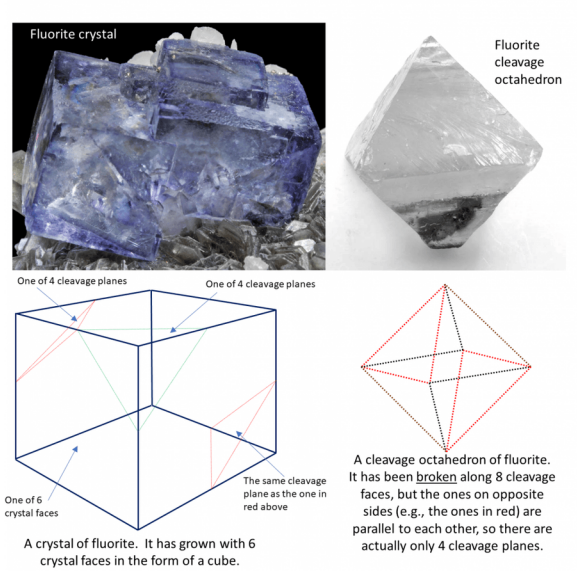


Figure 2.6.6 Crystal faces and cleavage planes in the mineral fluorite. The top-left photo shows a natural crystal of fluorite. It has crystal surfaces but you can see some future cleavage planes inside the crystal. The top-right photo shows what you can create if you take a crystal like the one on the left and carefully break it along its cleavage planes.

Density

Density is a measure of the mass of a mineral per unit volume, and it is a useful diagnostic tool in some cases. Most common minerals, such as quartz, feldspar, calcite, amphibole, and mica, have what we call “average density” (2.6 to 3.0 grams per cubic centimetre (g/cm^3)), and it would be difficult to tell them apart on the basis of their density. On the other hand, many of the metallic minerals, such as pyrite, hematite, and magnetite, have densities over 5 g/cm^3. They can easily be distinguished from the lighter minerals on the basis of density, but not necessarily from each other. A limitation of using density as a diagnostic tool is that one cannot assess it in minerals that are a small part of a rock that is mostly made up of other minerals.

Other Properties

Several other properties are also useful for identification of some minerals. For example, calcite is soluble in dilute acid and will give off bubbles of carbon dioxide. Magnetite is magnetic, so will affect a magnet. A few other minerals are weakly magnetic.

Image Descriptions

Figure 2.6.3 image description

	Talc	Gypsum	Calcite	Fluorine	Apatite	Feldspar	Quartz	Topaz	Corundum	Diamond
Measured Hardness	50	60	105	200	659	700	1100	1648	2085	7000
Mohs Hardness	1	2	3	4	5	6	7	8	9	10

[Return to Figure 2.6.3]

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- Figure 2.6.4a: Quartz Bresil © Didier Descouens. CC BY.
- Figure 2.6.4b: Pyrite cubic crystals on marlstone © Carles Millan. CC BY-SA.
- Figure 2.6.4c: Almandine garnet © Eurico Zimbres (FGEL/UERJ) and Tom Epaminondas (mineral collector). CC BY-SA.

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2.7: Summary

The topics covered in this chapter can be summarized as follows.

Section	Summary
2.1 Electrons, Protons, Neutrons, and Atoms	An atom is made up of protons and neutrons in the nucleus and electrons arranged in energy shells around the nucleus. The first shell holds two electrons, and outer shells hold more, but atoms strive to have eight electrons in their outermost shell (or two for H and He). They either gain or lose electrons (or share) to achieve this, and in so doing become either cations (if they lose electrons) or anions (if they gain them).
2.2 Bonding and Lattices	The main types of bonding in minerals are ionic bonding (electrons transferred) and covalent bonding (electrons shared). Some minerals have metallic bonding or other forms of weak bonding. Minerals form in specific three-dimensional lattices, and the nature of the lattices and the type of bonding within them have important implications for mineral properties.
2.3 Mineral Groups	Minerals are grouped according to the anion part of their formula, with some common types being oxides, sulfides, sulphates, halides, carbonates, phosphates, silicates, and native minerals.
2.4 Silicate Minerals	Silicate minerals are, by far, the most important minerals in Earth's crust. They all include silica tetrahedra (four oxygens surrounding a single silicon atom) arranged in different structures (chains, sheets, etc.). Some silicate minerals include iron or magnesium and are called ferromagnesian silicates.
2.5 Formation of Minerals	Most minerals in the crust form from the cooling and crystallization of magma. Some form from hot water solutions, during metamorphism or weathering, or through organic processes.
2.6 Mineral Properties	Some of the important properties for mineral identification include hardness, cleavage/fracture, density, lustre, color, and streak color. It's critical to be able to recognize these properties in order to be able to identify minerals.

Questions for Review

Answers to Review Questions can be found in Appendix 2.

1. What is the electrical charge on a proton? A neutron? An electron? What are their relative masses?
2. Explain how the need for an atom's outer shell to be filled with electrons contributes to bonding.
3. Why are helium and neon non-reactive?
4. What is the difference in the role of electrons in an ionic bond compared to a covalent bond?
5. What is the electrical charge on an anion? A cation?
6. What chemical feature is used in the classification of minerals into groups?
7. Name the mineral group for the following minerals:
    - calcite
    - gypsum
    - hematite
    - quartz
    - biotite
    - galena
    - graphite
    - fluorite
    - pyrite
    - orthoclase
    - magnetite
    - olivine

8. What is the net charge on an unbonded silica tetrahedron?
9. What allows magnesium to substitute freely for iron in olivine?
10. How are the silica tetrahedra structured differently in pyroxene and amphibole?
11. Why is biotite called a ferromagnesian mineral, while muscovite is not?
12. What are the names and compositions of the two end-members of the plagioclase series?
13. Why does quartz have no additional cations (other than Si^{+4})?
14. Why is color not necessarily a useful guide to mineral identification?
15. You have an unknown mineral that can scratch glass but cannot scratch a porcelain streak plate. What is its approximate hardness?

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3: INTRUSIVE IGNEOUS ROCKS

3.1: The Rock Cycle
3.2: Magma and Magma Formation
3.3: Crystallization of Magma
3.4: Classification of Igneous Rock
3.5: Intrusive Igneous Bodies
3.6: Summary

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3.1: The Rock Cycle

The rock components of the crust are slowly but constantly being changed from one form to another and the processes involved are summarized in the **rock cycle** (Figure 3.1.1). The rock cycle is driven by two forces: (1) Earth's internal heat engine, which moves material around in the core and the mantle and leads to slow but significant changes within the crust, and (2) the hydrological cycle, which is the movement of water, ice, and air at the surface, and is powered by the sun.

The rock cycle is still active on Earth because our core is hot enough to keep the mantle moving, our atmosphere is relatively thick, and we have liquid water. On some other planets or their satellites, such as the Moon, the rock cycle is virtually dead because the core is no longer hot enough to drive mantle convection and there is no atmosphere or liquid water.

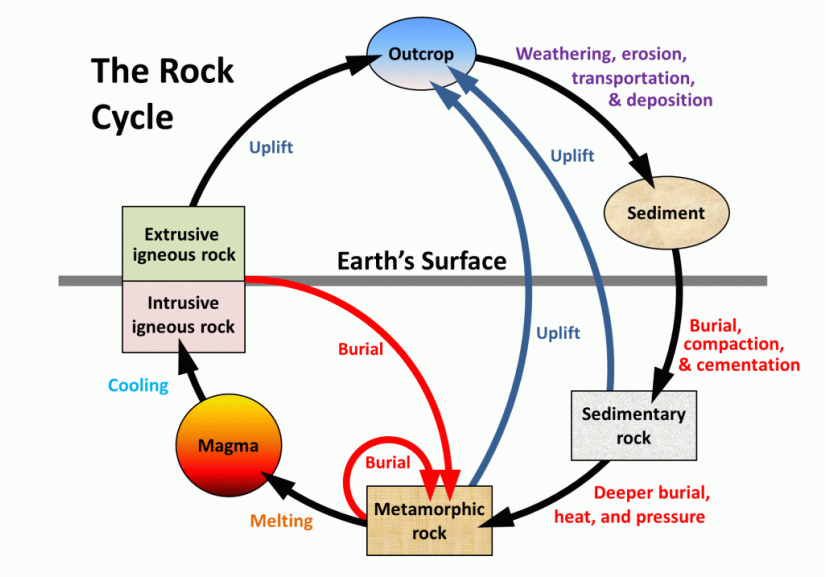


Figure 3.1.1 A schematic view of the rock cycle. The rock cycle takes place both above and below the Earth's surface. The rock deepest beneath the earth's surface, and under extreme heat and pressure, is metamorphic rock. This metamorphic rock can melt and become magma. When magma cools below the earth's surface, it becomes "intrusive igneous rock." If magma cools above the earth's surface, it is "extrusive igneous rock" and becomes part of the outcrop. The outcrop is subject to weathering and erosion, and can be moved and redeposited around the earth by forces such as water and wind. As the outcrop is eroded, it becomes sediment which can be buried, compacted, and cemented beneath the Earth's surface to become sedimentary rock. As sedimentary rock gets buried deeper and comes under increased heat and pressure, it returns to its original state as metamorphic rock. Rocks in the rock cycle do not always make a complete loop. It is possible for sedimentary rock to be uplifted back above the Earth's surface and for intrusive and extrusive igneous rock to be reburied and become metamorphic rock.

In describing the rock cycle, we can start anywhere we like, although it's convenient to start with magma. As we'll see in more detail below, magma is rock that is hot to the point of being entirely molten, with a temperature of between about 800° and 1300°C, depending on the composition and the pressure.



Figure 3.1.2 Magma forming pahoehoe basalt at Kilauea Volcano, Hawaii.

Magma can either cool slowly within the crust (over centuries to millions of years)—forming **intrusive igneous rock**, or erupt onto the surface and cool quickly (within seconds to years)—forming **extrusive igneous rock** (volcanic rock) (Figure 3.1.2). Intrusive igneous rock typically crystallizes at depths of hundreds of meters to tens of kilometers below the surface. To change its position in the rock cycle, intrusive igneous rock has to be uplifted and then exposed by the erosion of the overlying rocks.

Through the various plate-tectonics-related processes of mountain building, all types of rocks are uplifted and exposed at the surface. Once exposed, they are weathered, both physically (by mechanical breaking of the rock) and chemically (by weathering of the minerals), and the weathering products—mostly small rock and mineral fragments—are eroded, transported, and then deposited as **sediments**. Transportation and deposition occur through the action of glaciers, streams, waves, wind, and other agents, and sediments are deposited in rivers, lakes, deserts, and the ocean.

Exercise 3.1 Rock around the rock-cycle clock

Referring to the rock cycle (Figure 3.1.1), list the steps that are necessary to cycle some geological material starting with a sedimentary rock, which then gets converted into a metamorphic rock, and eventually a new sedimentary rock.

A *conservative* estimate is that each of these steps would take approximately 20 million years (some may be less, others would be more, and some could be much more). How long might it take for this entire process to be completed?

See Appendix 3 for Exercise 3.1 Answers.



Figure 3.1.3 Cretaceous-aged marine sandstone overlying marine mudstone, Gabriola Island, B.C.

Unless they are re-eroded and moved along, sediments will eventually be buried by more sediments. At depths of hundreds of meters or more, they become compressed and cemented into **sedimentary rock** (See Figure 3.1.3 for example). Again through various means, largely resulting from plate-tectonic forces, different kinds of rocks are either uplifted, to be re-eroded, or buried deeper within the crust where they are heated up, squeezed, and changed into **metamorphic rock** (Figure 3.1.4)



Figure 3.1.4 Metamorphosed and folded Triassic-aged limestone, Quadra Island, B.C.

Images Attributions

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3.2: Magma and Magma Formation

Magmas can vary widely in composition, but in general they are made up of only eight elements; in order of importance: oxygen, silicon, aluminum, iron, calcium, sodium, magnesium, and potassium (Figure 3.2.1). Oxygen, the most abundant element in magma, comprises a little less than half the total, followed by silicon at just over one-quarter. The remaining elements make up the other one-quarter. Magmas derived from crustal material are dominated by oxygen, silicon, aluminum, sodium, and potassium.

The composition of magma depends on the rock it was formed from (by melting), and the conditions of that melting. Magmas derived from the mantle have higher levels of iron, magnesium, and calcium, but they are still likely to be dominated by oxygen and silicon. All magmas have varying proportions of elements such as hydrogen, carbon, and sulfur, which are converted into gases like water vapor, carbon dioxide, and hydrogen sulfide as the magma cools.

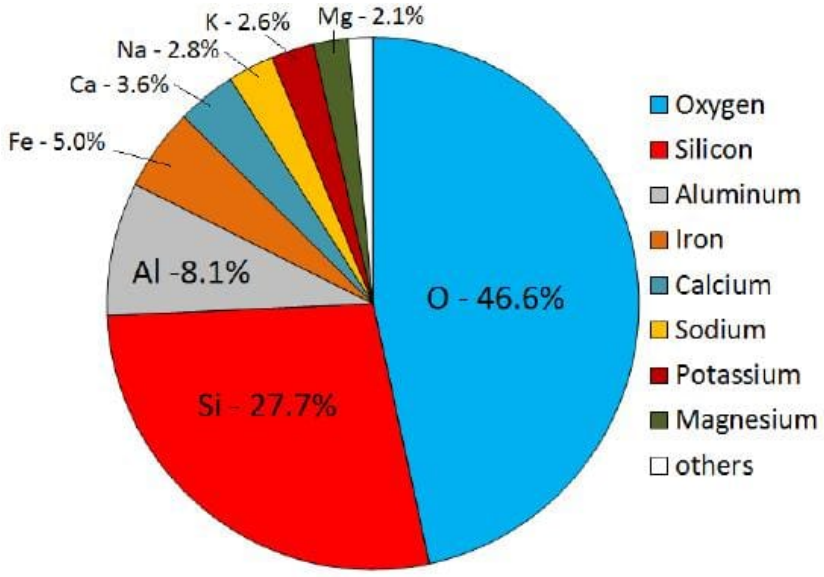


Figure 3.2.1 Average elemental proportions in Earth's crust, which is close to the average composition of magmas within the crust. The average elemental proportions in the Earth's crust from the largest amount to the smallest amount. Oxygen (46.6%), Silicon (27.7%), Aluminum (8.1%), Iron (5.0%), Calcium (3.6%), Sodium (2.8%), Potassium (2.6%), Magnesium (2.1%), Others (1.5%).

Virtually all of the igneous rocks that we see on Earth are derived from magmas that formed from **partial melting** of existing rock, either in the upper mantle or the crust. Partial melting is what happens when only some parts of a rock melt; it takes place because rocks are not pure materials. Most rocks are made up of several minerals, each of which has a different melting temperature. The wax in a candle is a pure material. If you put some wax into a warm oven (50°C will do as the melting temperature of most wax is about 40°C) and leave it there for a while, it will soon start to melt. That's complete melting, not partial melting. If instead you took a mixture of wax, plastic, aluminum, and glass and put it into the same warm oven, the wax would soon start to melt, but the plastic, aluminum, and glass would not melt (Figure 3.2.2a). That's partial melting and the result would be solid plastic, aluminum, and glass surrounded by liquid wax (Figure 3.2.2b). If we heat the oven up to around 120°C, the plastic would melt too and mix with the liquid wax, but the aluminum and glass would remain solid (Figure 3.2.2c). Again this is partial melting. If we separated the wax/plastic "magma" from the other components and let it cool, it would eventually harden. As you can see from Figure 3.2.2 d, the liquid wax and plastic have mixed, and on cooling, have formed what looks like a single solid substance. It is most likely that this is a very fine-grained mixture of solid wax and solid plastic, but it could also be some other substance that has formed from the combination of the two.

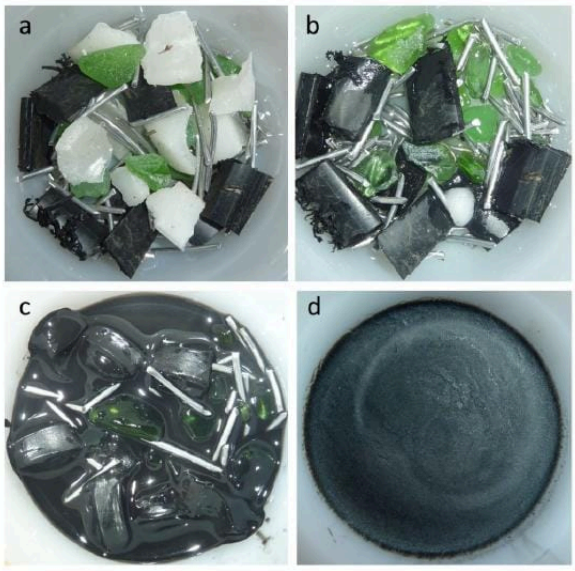


Figure 3.2.2 Partial melting of “pretend rock”: (a) the original components of white candle wax, black plastic pipe, green beach glass, and aluminum wire, (b) after heating to 50˚C for 30 minutes only the wax has melted, (c) after heating to 120˚C for 60 minutes much of the plastic has melted and the two liquids have mixed, (d) the liquid has been separated from the solids and allowed to cool to make a “pretend rock” with a different overall composition.

In this example, we partially melted some pretend rock to create some pretend magma. We then separated the magma from the source and allowed it to cool to make a new pretend rock with a composition quite different from the original material (it lacks glass and aluminum).

Of course partial melting in the real world isn’t exactly the same as in our pretend-rock example. The main differences are that rocks are much more complex than the four-component system we used, and the mineral components of most rocks have more similar melting temperatures, so two or more minerals are likely to melt at the same time to varying degrees. Another important difference is that when rocks melt, the process takes thousands to millions of years, not the 90 minutes it took in the pretend-rock example.

Contrary to what one might expect, and contrary to what we did to make our pretend rock, most partial melting of real rock does not involve heating the rock up. The two main mechanisms through which rocks melt are **decompression melting** and **flux melting**. Decompression melting takes place within Earth when a body of rock is held at approximately the same temperature but the pressure is reduced. This happens because the rock is being moved toward the surface, either at a **mantle plume** (a.k.a., hot spot), or in the upwelling part of a mantle convection cell.[1] The mechanism of decompression melting is shown in Figure 3.2.3a . If a rock that is hot enough to be close to its melting point is moved toward the surface, the pressure is reduced, and the rock can pass to the liquid side of its melting curve. At this point, *partial* melting starts to take place. The process of flux melting is shown in Figure 3.2.3b. If a rock is close to its melting point and some water (a flux that promotes melting) is added to the rock, the melting temperature is reduced (solid line versus dotted line), and partial melting starts.

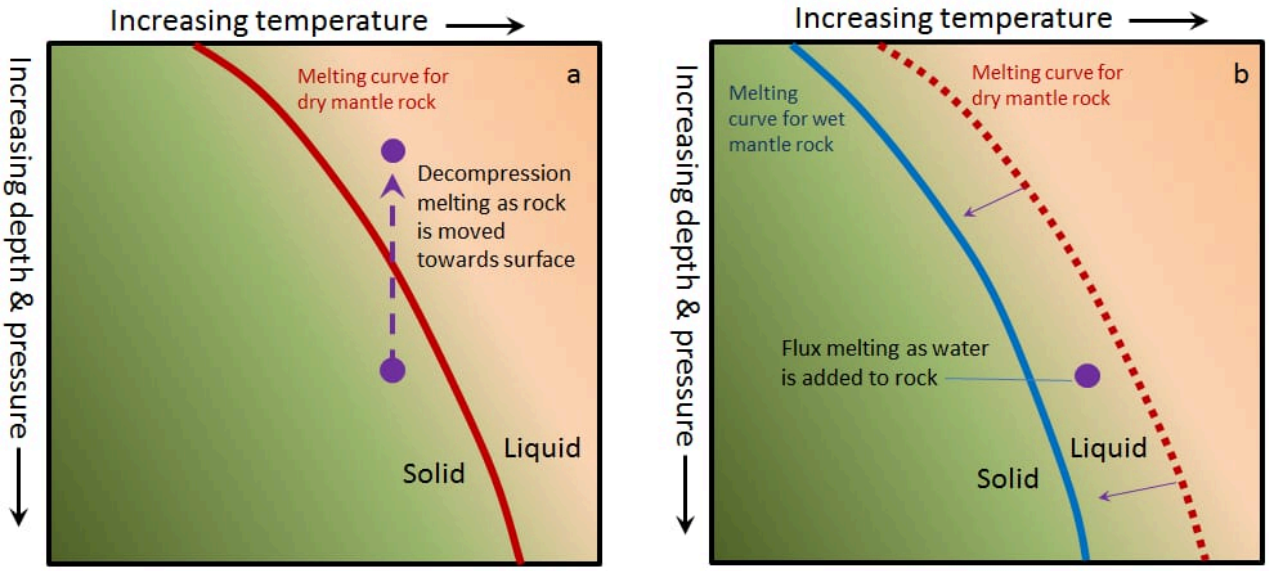


Figure 3.2.3 Mechanisms for (a) decompression melting (the rock is moved toward the surface) and (b) flux melting (water is added to the rock) and the melting curve is displaced. (a) Dry mantle rock is predominately solid. However, its melting point is dependent on the temperature and pressure the rock is under. The higher the pressure (meaning the farther the rock is from the Earth's surface), the more likely dry mantle rock is going to be solid. Dry mantle rock under extreme pressure requires a much higher temperature to melt than dry mantle rock under less pressure. As pressure drops (meaning as the rock rises towards the Earth's surface), the required temperature to melt the mantle rock drops as well. (b) In comparison to dry mantle rock, wet mantle rock under the same amount of pressure (at the same distance from the earth's surface) requires a lower temperature to melt. When liquid is added to dry mantel rock at a pressure and temperature point in which wet mantle rock would be melted, flux melting occurs.

The partial melting of rock happens in a wide range of situations, most of which are related to plate tectonics. The more important of these are shown in Figure 3.2.3. At both mantle plumes and in the upward parts of convection systems, rock is being moved toward the surface, the pressure is dropping, and at some point, the rock crosses to the liquid side of its melting curve. At subduction zones, water from the wet, subducting oceanic crust is transferred into the overlying hot mantle. This provides the flux needed to lower the melting temperature. In both of these cases, only partial melting takes place—typically only about 10% of the rock melts—and it is always the most silica-rich components of the rock that melt, creating a magma that is more silica-rich than the rock from which it is derived. (By analogy, the melt from our pretend rock is richer in wax and plastic than the "rock" from which it was derived.) The magma produced, being less dense than the surrounding rock, moves up through the mantle, and eventually into the crust.

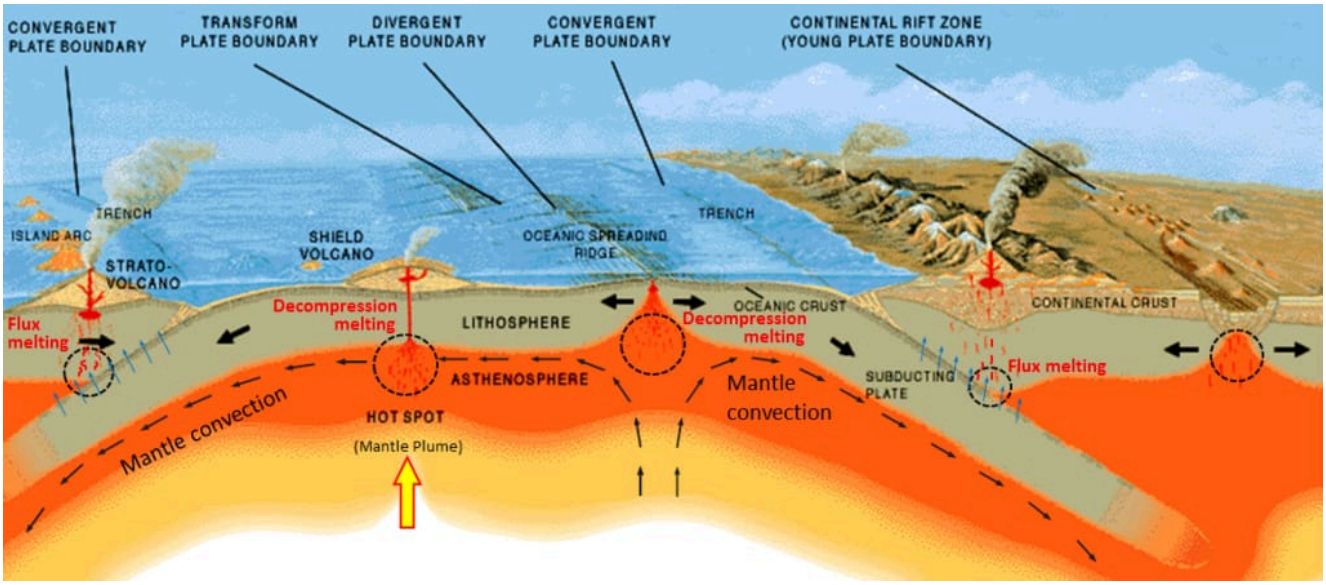


Figure 3.2.4 Common sites of magma formation in the upper mantle. The black circles are regions of partial melting. The blue arrows represent water being transferred from the subducting plates into the overlying mantle.

As it moves toward the surface, and especially when it moves from the mantle into the lower crust, the hot magma interacts with the surrounding rock. This typically leads to partial melting of the surrounding rock because most such magmas are hotter than the melting temperature of crustal rock. (In this case, melting is caused by an increase in temperature.) Again, the more silica-rich parts of the surrounding rock are preferentially melted, and this contributes to an increase in the silica content of the magma.

At very high temperatures (over 1300°C), most magma is entirely liquid because there is too much energy for the atoms to bond together. As the temperature drops, usually because the magma is slowly moving upward, things start to change. Silicon and oxygen combine to form silica tetrahedra, and then, as cooling continues, the tetrahedra start to link together to make chains (**polymerize**). These silica chains have the important effect of making the magma more viscous (less runny), and as we'll see in Chapter 4, magma viscosity has significant implications for volcanic eruptions. As the magma continues to cool, crystals start to form.

Exercise 3.2 Making magma viscous

This is an experiment that you can do at home to help you understand the properties of magma. It will only take about 15 minutes, and all you need is half a cup of water and a few tablespoons of flour.

If you've ever made gravy, white sauce, or roux, you'll know how this works.

Place about 1/2 cup (125 mL) of water in a saucepan over medium heat. Add 2 teaspoons (10 mL) of white flour (this represents silica) and stir while the mixture comes close to boiling. It should thicken like gravy because the gluten in the flour becomes polymerized into chains during this process.

Now you're going to add more "silica" to see how this changes the viscosity of your magma. Take another 4 teaspoons (20 mL) of flour and mix it thoroughly with about 4 teaspoons (20 mL) of water in a cup and then add all of that mixture to the rest of the water and flour in the saucepan. Stir while bringing it back up to nearly boiling temperature, and then allow it to cool. This mixture should slowly become much thicker — something like porridge — because there is more gluten and more chains have been formed (see the photo).



Figure 3.2.5 Flour-and-water magma experiment.

This is analogous to magma, of course. As we'll see below, magmas have quite variable contents of silica and therefore have widely varying viscosities ("thicknesses") during cooling.

See Appendix 3 for Exercise 3.2 answers.

Media Attributions

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- Figure 3.2.4: "Cross section" by José F. Vigil from *This Dynamic Planet* — a wall map produced jointly by the U.S. Geological Survey, the Smithsonian Institution, and the U.S. Naval Research Laboratory. Adapted by Steven Earle. Public domain.

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1. Mantle plumes are described in Chapter 4 and mantle convection in Chapter 9.

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3.3: Crystallization of Magma

The minerals that make up igneous rocks crystallize at a range of different temperatures. This explains why a cooling magma can have some crystals within it and yet remain predominantly liquid. The sequence in which minerals crystallize from a magma is known as the **Bowen reaction series** (Figure 3.3.1 and Figure 3.3.3).

Of the common silicate minerals, olivine normally crystallizes first, at between 1200° and 1300°C. As the temperature drops, and assuming that some silica remains in the magma, the olivine crystals will react (combine) with some of the silica in the magma to form pyroxene. As long as there is silica remaining and the rate of cooling is slow, this process continues down the discontinuous branch: olivine to pyroxene, pyroxene to amphibole, and (under the right conditions) amphibole to biotite.

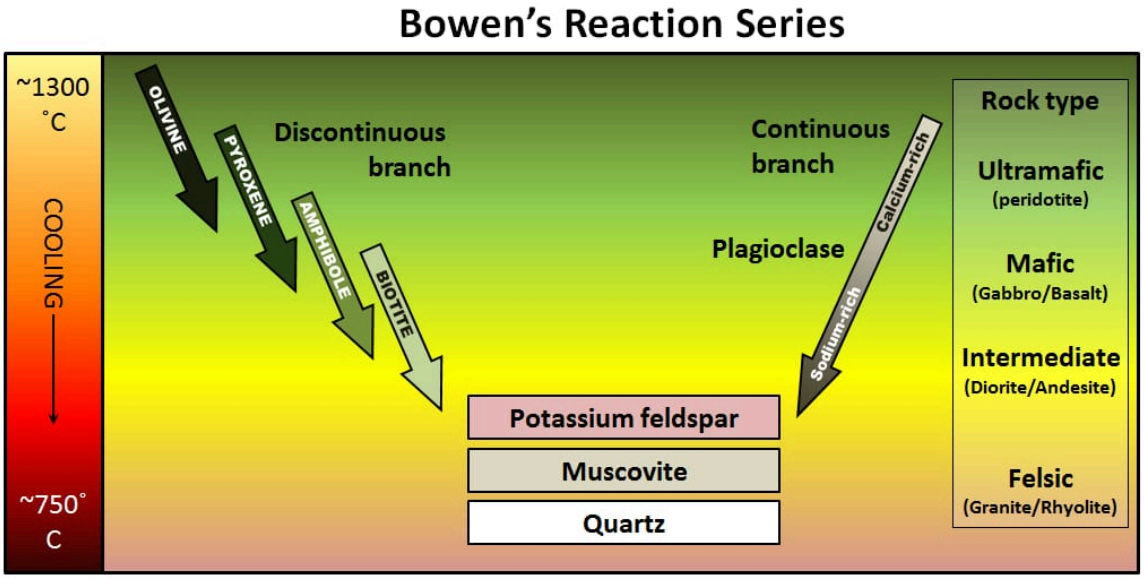


Figure 3.3.1 The Bowen reaction series describes the process of magma crystallization.

At about the point where pyroxene begins to crystallize, plagioclase feldspar also begins to crystallize. At that temperature, the plagioclase is calcium-rich (anorthite) (see Figure 3.3.1). As the temperature drops, and providing that there is sodium left in the magma, the plagioclase that forms is a more sodium-rich variety.

In some cases, individual plagioclase crystals can be zoned from calcium-rich in the centre to more sodium-rich around the outside. This occurs when calcium-rich early-forming plagioclase crystals become coated with progressively more sodium-rich plagioclase as the magma cools. Figure 3.3.2 shows a zoned plagioclase under a microscope.



Figure 3.3.2 A zoned plagioclase crystal. The central part is calcium-rich and the darker outside part is sodium-rich.

Finally, if the magma is quite silica-rich to begin with, there will still be some left at around 750° to 800°C, and from this last magma, potassium feldspar, quartz, and maybe muscovite mica will form.

Who was Bowen, and what is a reaction series?

Norman Levi Bowen, born in Kingston Ontario, studied geology at Queen's University and then at MIT in Boston. In 1912, Norman Levi Bowen joined the Carnegie Institution in Washington, D.C., where he carried out groundbreaking experimental research into the processes of cooling magmas. Working mostly with basaltic magmas, he determined the order of crystallization of minerals as the temperature drops. The method, in brief, was to melt the rock to a magma in a specially-made kiln, allow it to cool slowly to a specific temperature (allowing some minerals to form), and then quench it (cool it quickly) so that no new minerals form (only glass). The results were studied under the microscope and by chemical analysis. This was done over and over, each time allowing the magma to cool to a lower temperature before quenching.



Figure 3.3.3

The Bowen reaction series is one of the results of his work, and even a century later, it is an important basis for our understanding of igneous rocks. The word *reaction* is critical. In the discontinuous branch, olivine is typically the first mineral to form (at just below 1300°C). As the temperature continues to drop, olivine becomes unstable while pyroxene becomes stable. The early-forming olivine crystals *react* with silica in the remaining liquid magma and are converted into pyroxene, something like this:

$$Mg_2SiO_4 + SiO_2 \text{ (olivine) becomes } 2MgSiO_3 \text{ (proxene)}$$

This continues down the chain, as long as there is still silica left in the liquid.

The composition of the original magma is critical to magma crystallization because it determines how far the reaction process

can continue before all of the silica is used up. The compositions of typical **mafic**, intermediate, and **felsic** magmas are shown in Figure 3.3.4. Note that, unlike Figure 3.3.1, these compositions are expressed in terms of "oxides" (e.g., Al_2O_3 rather than just Al). There are two reasons for this: one is that in the early analytical procedures, the results were always expressed that way, and the other is that all of these elements combine readily with oxygen to form oxides.

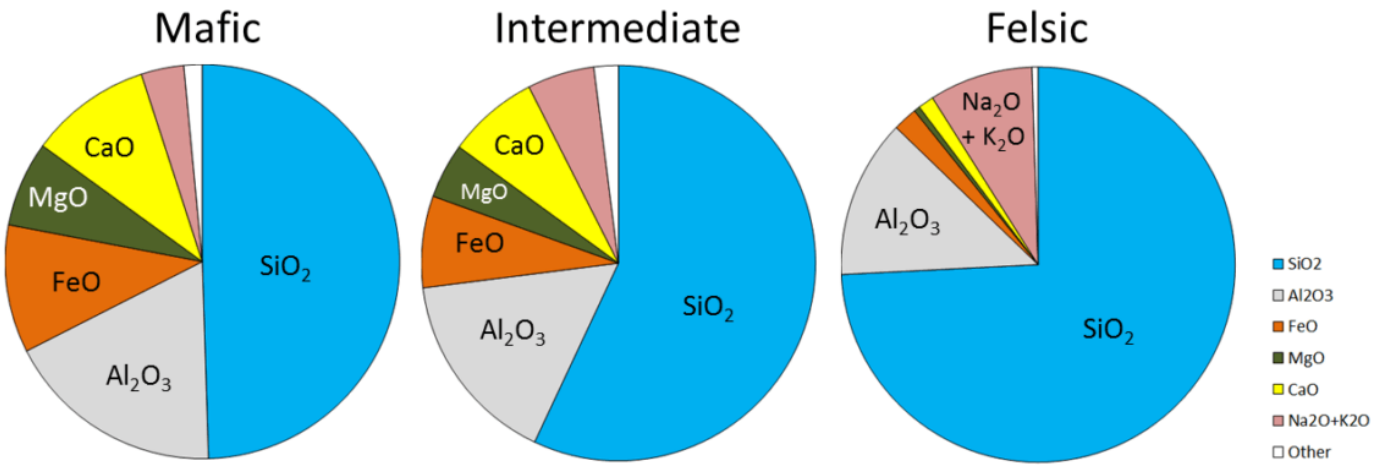


Figure 3.3.4 The chemical compositions of typical mafic, intermediate, and felsic magmas and the types of rocks that form from them.

Mafic magmas have 45% to 55% SiO_2, about 25% total of FeO and MgO plus CaO, and about 5% Na_2O + K_2O. Felsic magmas, on the other hand, have much more SiO_2 (65% to 75%) and Na_2O + K_2O (around 10%) and much less FeO and MgO plus CaO (about 5%).

Exercise 3.3 Determining rock types based on magma composition

The proportions of the main chemical components of felsic, intermediate, and mafic magmas are listed in the table below. (The values are similar to those shown in Figure 3.3.4.)

Table 3.1 Proportions of the main chemical components in felsic, intermediate, and mafic magma

Oxide	Felsic Magma	Intermediate Magma	Mafic Magma
SiO_2	65% to 75%	55% to 65%	45% to 55%
Al_2O_3	12% to 16%	14% to 18%	14% to 18%
FeO	2% to 4%	4% to 8%	8% to 12%
CaO	1% to 4%	4% to 7%	7% to 11%
MgO	0% to 3%	2% to 6%	5% to 9%
Na_2O	2% to 6%	3% to 7%	1% to 3%
K_2O	3% to 5%	2% to 4%	0.5% to 3%

Chemical data for four rock samples are shown in the following table. Compare these with those in the table above to determine whether each of these samples is felsic, intermediate, or mafic.

Table 3.2 Chemical Data for Four Unidentified Rock Samples

Rock Sample	SiO_2	Al_2O_3	FeO	CaO	MgO	Na_2O	K_2O	What type of magma is it?
Rock 1	55%	17%	5%	6%	3%	4%	3%	
Rock 2	74%	14%	3%	3%	0.5%	5%	4%	
Rock 3	47%	14%	8%	10%	8%	1%	2%	

Rock Sample	SiO_2	Al_2O_3	FeO	CaO	MgO	Na_2O	K_2O	What type of magma is it?
Rock 4	65%	14%	4%	5%	4%	3%	3%	

See Appendix 3 for Exercise 3.3 answers.

As a *mafic* magma starts to cool, some of the silica combines with iron and magnesium to make olivine. As it cools further, much of the remaining silica goes into calcium-rich plagioclase, and any silica left may be used to convert some of the olivine to pyroxene. Soon after that, all of the magma is used up and no further changes takes place. The minerals present will be olivine, pyroxene, and calcium-rich plagioclase. If the magma cools slowly underground, the product will be **gabbro**; if it cools quickly at the surface, the product will be **basalt** (Figure 3.3.5).

Felsic magmas tend to be cooler than mafic magmas when crystallization begins (because they don't have to be as hot to remain liquid), and so they may start out crystallizing pyroxene (not olivine) and plagioclase. As cooling continues, the various reactions on the discontinuous branch will proceed because silica is abundant, the plagioclase will become increasingly sodium-rich, and eventually potassium feldspar and quartz will form. Commonly even very felsic rocks will not have biotite or muscovite because they may not have enough aluminum or enough hydrogen to make the OH complexes that are necessary for mica minerals. Typical felsic rocks are **granite** and **rhyolite** (Figure 3.3.5).

The cooling behaviour of intermediate magmas lie somewhere between those of mafic and felsic magmas. Typical mafic rocks are gabbro (intrusive) and basalt (extrusive). Typical intermediate rocks are **diorite** and **andesite**. Typical felsic rocks are granite and rhyolite (Figure 3.3.5).

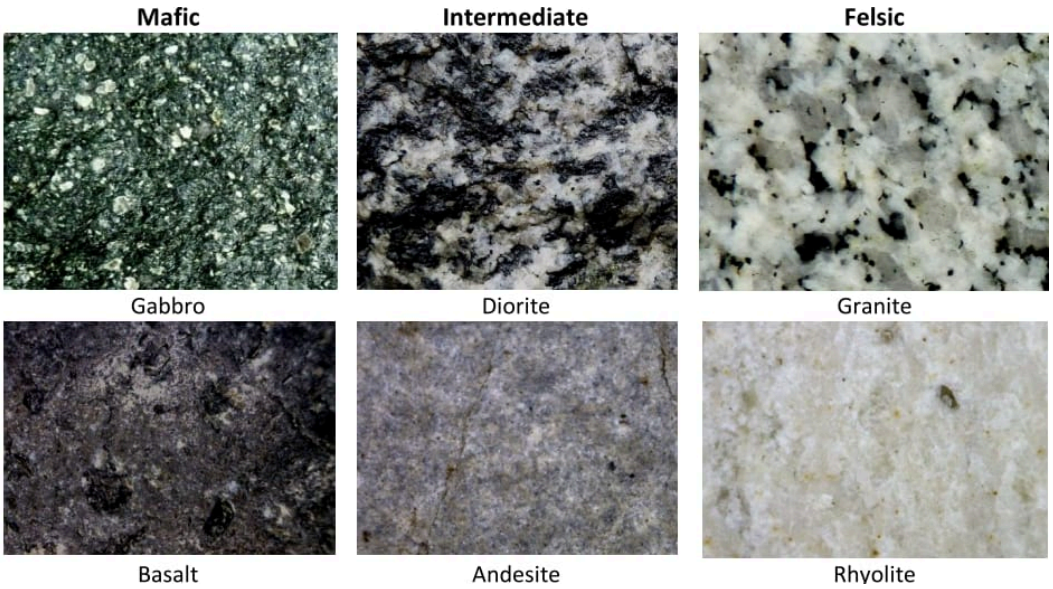


Figure 3.3.5 Examples of the igneous rocks that form from mafic, intermediate, and felsic magmas.

A number of processes that take place within a magma chamber can affect the types of rocks produced in the end. If the magma has a low viscosity (i.e., it's runny)—which is likely if it is mafic—the crystals that form early, such as olivine (Figure 3.3.6a), may slowly settle toward the bottom of the magma chamber (Figure 3.3.6b). The means that the overall composition of the magma near the top of the magma chamber will become more felsic, as it is losing some iron- and magnesium-rich components. This process is known as **fractional crystallization**. The crystals that settle might either form an olivine-rich layer near the bottom of the magma chamber, or they might remelt because the lower part is likely to be hotter than the upper part (remember, from Chapter 1, that temperatures increase steadily with depth in Earth because of the geothermal gradient). If any melting takes place, crystal settling will make the magma at the bottom of the chamber more mafic than it was to begin with (Figure 3.3.6c).

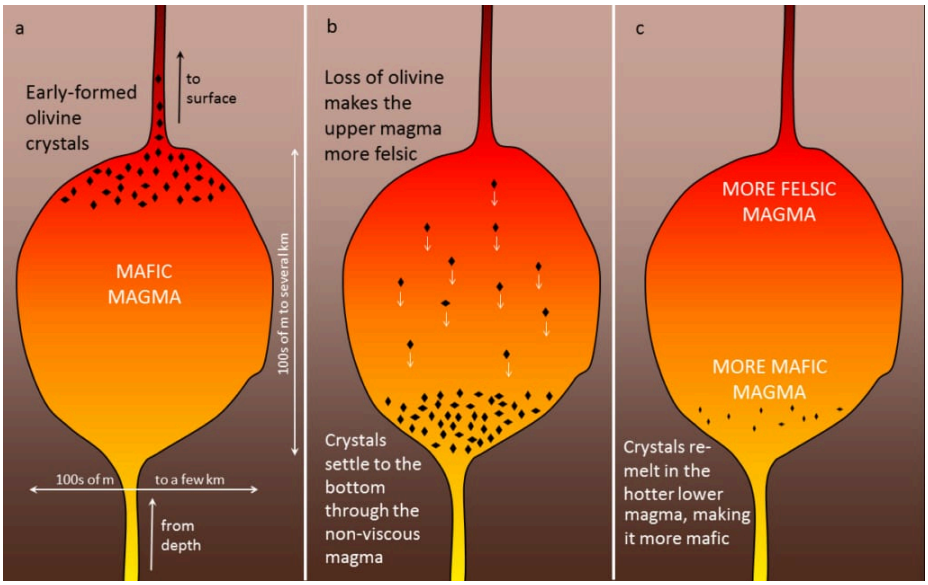


Figure 3.3.6 An example of crystal settling and the formation of a zoned magma chamber.

If crystal settling does not take place, because the magma is too viscous, then the process of cooling will continue as predicted by the Bowen reaction series. In some cases, however, partially cooled but still liquid magma, with crystals in it, will either move farther up into a cooler part of the crust, or all the way to the surface during a volcanic eruption. In either of these situations, the magma that has moved toward the surface is likely to cool much faster than it did within the magma chamber, and the rest of the rock will have a finer crystalline texture. An igneous rock with large crystals embedded in a matrix of much finer crystals is indicative of a two-stage cooling process, and the texture is **porphyritic** (Figure 3.3.7). For the rock to be called "porphyritic" there has to be a significant difference in crystal size, where the larger crystals are at least 10 times larger than the average size of the smaller crystals.

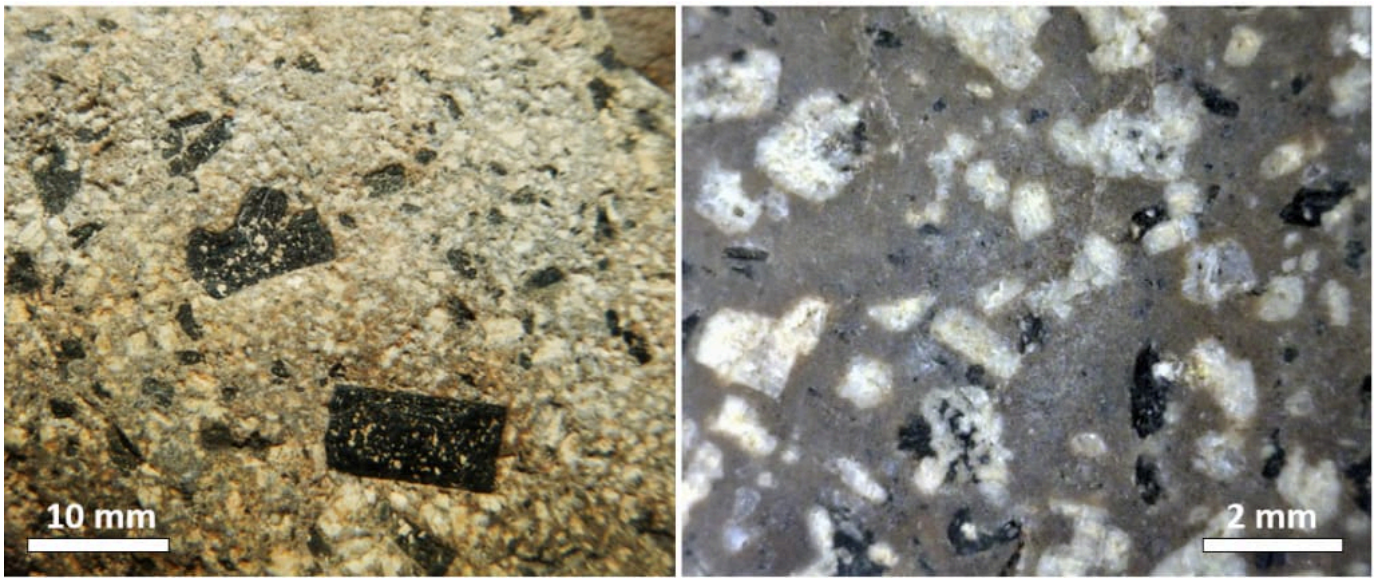


Figure 3.3.7 Porphyritic textures, left: 1.3 cm long amphibole crystals in an intrusive igneous rock in which most of the crystals are less than 1 mm, right: 1 to 2 mm long feldspar crystals and 1 mm long amphibole crystals in a volcanic rock where most of the crystals are less than 0.1 mm.

Exercise 3.4 Porphyritic minerals

As a magma cools below 1300°C, minerals start to crystallize within it. If that magma is then involved in a volcanic eruption, the rest of the liquid will cool quickly to form a **porphyritic** texture. The rock will have some relatively large crystals (**phenocrysts**) of the minerals that crystallized early, and the rest will be very fine grained or even glassy. Using Figure 3.3.8, predict what phenocrysts might be present where the magma cooled as far as line **a** in one case, and line **b** in another.

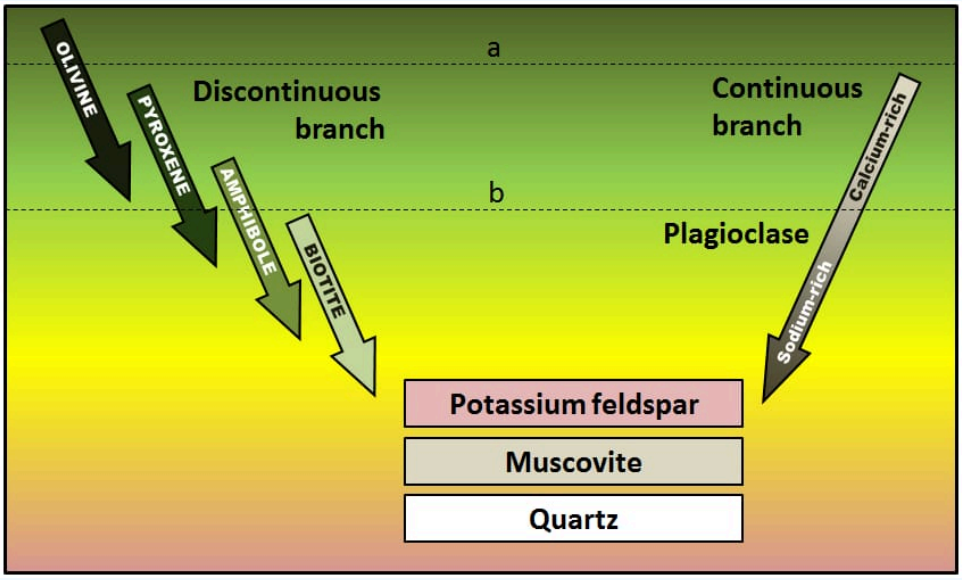


Figure 3.3.8 Bowen reaction series. Line a – at high temperature – intersects olivine, Line b – at a lower temperature – intersects pyroxene and amphibole on the left, and plagioclase feldspar on the right

See Appendix 3 for Exercise 3.4 answers.

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3.4: Classification of Igneous Rock

As has already been described, igneous rocks are classified into four categories: felsic, intermediate, mafic, and ultramafic, based on either their chemistry or their mineral composition. The diagram in Figure 3.4.1 can be used to help classify igneous rocks by their mineral composition. An important feature to note on this diagram is the red line separating the non-ferromagnesian silicates in the lower left (K-feldspar, quartz, and plagioclase feldspar) from the ferromagnesian silicates in the upper right (biotite, amphibole, pyroxene, and olivine). In classifying intrusive igneous rocks, the first thing to consider is the percentage of ferromagnesian silicates. In most igneous rocks the ferromagnesian silicate minerals are clearly darker than the others, but it is still quite difficult to estimate the proportions of minerals in a rock.

Based on the position of the red line in Figure 3.4.1, it is evident that felsic rocks can have between 1% and 20% ferromagnesian silicates (the red line intersects the left side of the felsic zone 1% of the distance from the top of the diagram, and it intersects the right side of the felsic zone 20% of the distance from the top). Intermediate rocks have between 20% and 50% ferromagnesian silicates, and mafic rocks have 50% to 100% ferromagnesian silicates. To be more specific, felsic rocks typically have biotite and/or amphibole; intermediate rocks have amphibole and, in some cases, pyroxene; and mafic rocks have pyroxene and, in some cases, olivine.

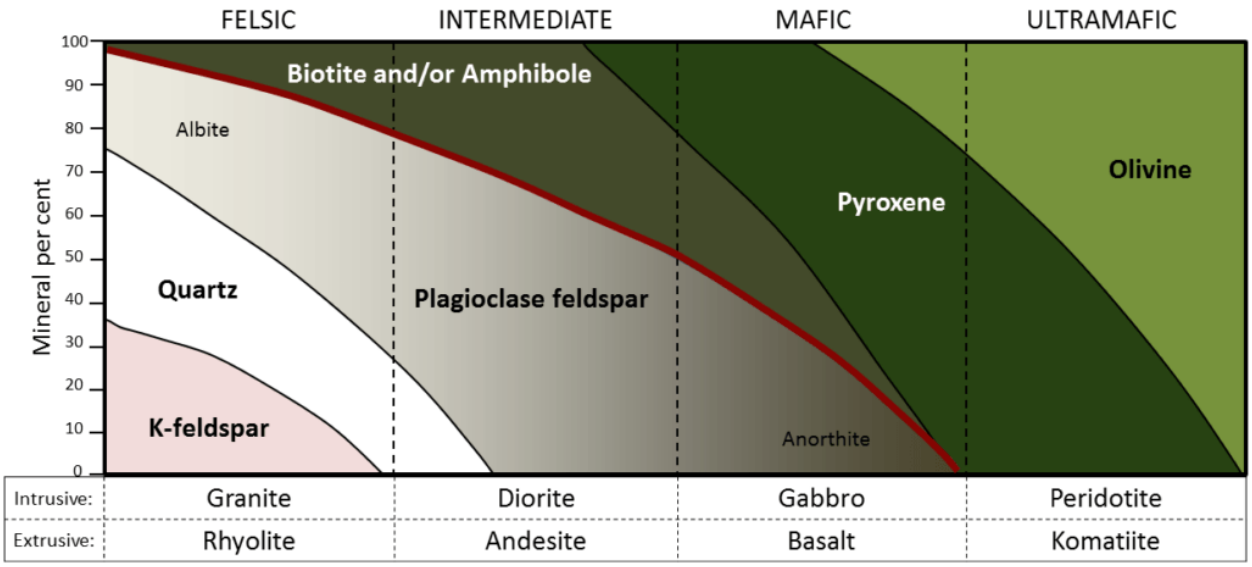


Figure 3.4.1 A simplified classification diagram for igneous rocks based on their mineral compositions.

If we focus on the non-ferromagnesian silicates, it is evident that felsic rocks can have from 0% to 35% K-feldspar, from 25% to 35% quartz (the vertical thickness of the quartz field varies from 25% to 35%), and from 25% to 50% plagioclase (and that plagioclase will be sodium-rich, or albitic). Intermediate rocks can have up to 25% quartz and 50% to 75% plagioclase. Mafic rocks only have plagioclase (up to 50%), and that plagioclase will be calcium-rich, or anorthitic.

Exercise 3.5 Mineral proportions in igneous rocks

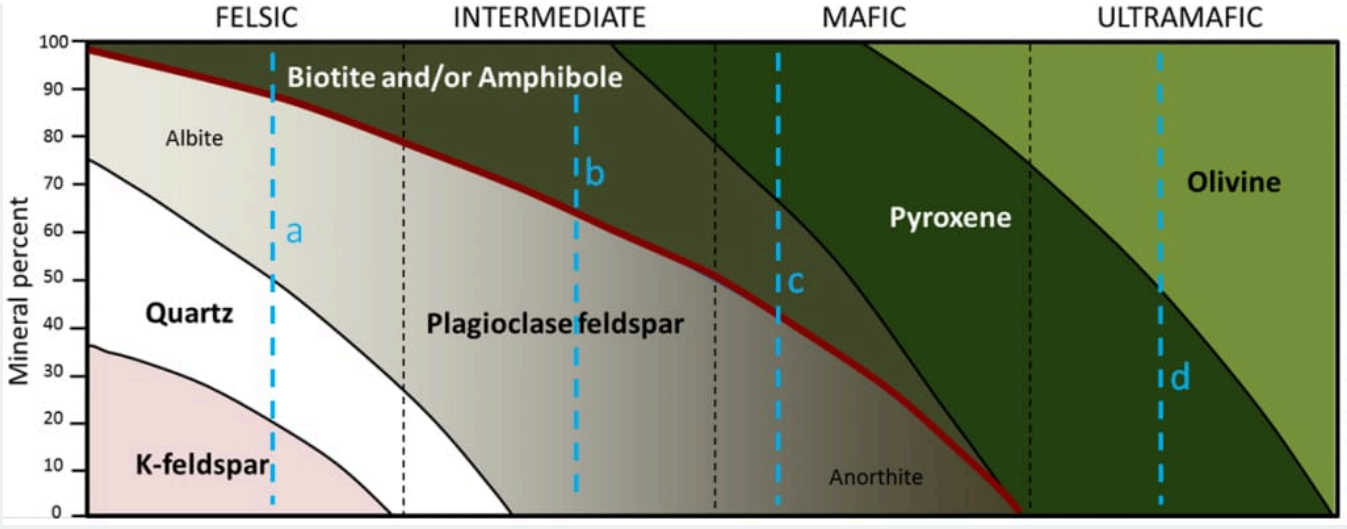


Figure 3.4.2

The dashed blue lines (labelled a, b, c, d) in Figure 3.4.2 represent four igneous rocks. Complete the table by estimating the mineral proportions (percent) of the four rocks (to the nearest 10%).

Hint: Rocks **b** and **d** are the easiest; start with those.

Rock	Biotite/amphibole	Pyroxene	Olivine	Plagioclase	Quartz	K-feldspar
a						
b						
c						
d						

See Appendix 3 for Exercise 3.5 answers.

Figure 3.4.3 provides a diagrammatic representation of the proportions of dark minerals in light-colored rocks. You can use that when trying to estimate the ferromagnesian mineral content of actual rocks, and you can get some practice doing that by completing Exercise 3.6. Be warned! Geology students almost universally over-estimate the proportion of dark minerals.

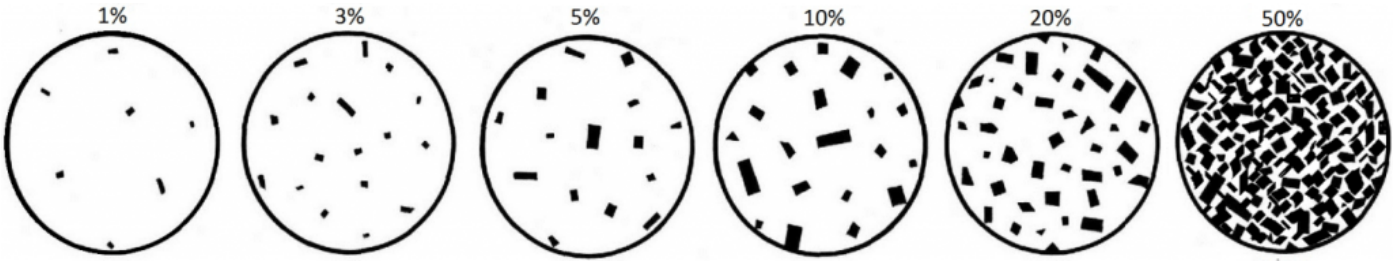
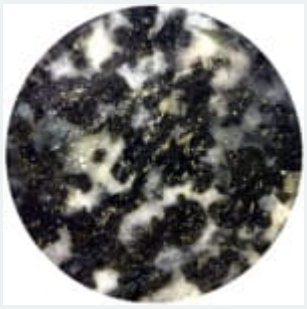
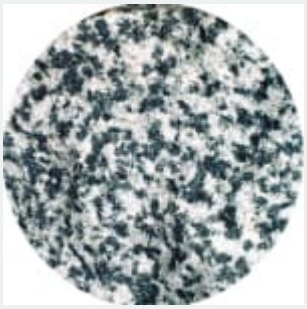


Figure 3.4.3 A guide to estimating the proportions of dark minerals in light-colored rocks.

Exercise 3.6 Proportions of ferromagnesian silicates

The four igneous rocks shown below have differing proportions of ferromagnesian silicates. Estimate those proportions using the diagrams in Figure 3.4.3, and then use Figure 3.4.1 to determine the likely rock name for each one.

			
___%	___%	___%	___%
_____	_____	_____	_____

See Appendix 3 for Exercise 3.6 answers.

Igneous rocks are also classified according to their textures. The textures of volcanic rocks will be discussed in Chapter 4, so here we'll only look at the different textures of intrusive igneous rocks. Almost all intrusive igneous rocks have crystals that are large enough to see with the naked eye, and we use the term **phaneritic** (from the Greek word phaneros meaning visible) to describe that. Typically that means they are larger than about 0.5 millitmeres (mm) — the thickness of a strong line made with a ballpoint pen. (If the crystals are too small to distinguish, which is typical of most volcanic rocks, we use the term **aphanitic** (from the Greek word *aphanos* – unseen) The intrusive rocks shown in Figure 3.4.5 are all phaneritic, as are those shown in Exercise 3.6.

In general, the size of crystals is proportional to the rate of cooling. The longer it takes for a body of magma to cool, the larger the crystals can grow. It is not uncommon to see an intrusive igneous rock with crystals up to 1 centimetre (cm) long. In some situations, especially toward the end of the cooling stage, the magma can become water rich. The presence of liquid water (still liquid at high temperatures because it is under pressure) promotes the relatively easy movement of ions, and this allows crystals to grow large, sometimes to several centimeters (Figure 3.4.4). Finally, as already described, if an igneous rock goes through a two-stage cooling process, its texture will be porphyritic (Figure 3.4.7).



Figure 3.4.4 A pegmatitic rock with large crystals

Image Descriptions

Figure 3.4.1 image description: Mineral composition of igneous rocks

Igneous Rocks	Felsic	Intermediate	Mafic	Ultramafic
K-feldspar	0 to 35%	0%	0%	0%
Quartz	25 to 35%	0 to 25%	0%	0%
Plagioclase feldspar	25 to 50%	50 to 70%	0 to 50%	0%
Biotite and/or Amphibole	0 to 20%	20 to 40%	0 to 30%	0%
Pyroxene	0%	0 to 20%	20 to 75%	0% to 75%
Olivine	0%	0%	0 to 25 %	25% to 100%
Intrusive	Granite	Diorite	Gabbro	Peridotite
Extrusive	Rhyolite	Andesite	Basalt	Komatiite

[Return to Figure 3.4.1]

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3.5: Intrusive Igneous Bodies

In most cases, a body of hot magma is less dense than the rock surrounding it, so it has a tendency to move very slowly up toward the surface. It does so in a few different ways, including filling and widening existing cracks, melting the surrounding rock (called **country rock**[1]**)**, pushing the rock aside (where it is somewhat plastic), and breaking the rock. Where some of the country rock is broken off, it may fall into the magma, a process called **stoping**. The resulting fragments, illustrated in Figure 3.5.1, are known as **xenoliths** (Greek for “strange rocks”).



Figure 3.5.1 Xenoliths of mafic rock in granite, Victoria, B.C. The fragments of dark rock have been broken off and incorporated into the light-colored granite.

Some upward-moving magma reaches the surface, resulting in volcanic eruptions, but most cools within the crust. The resulting body of rock is known as a **pluton**. Plutons can have various different shapes and relationships to the surrounding country rock as shown in Figure 3.5.2.

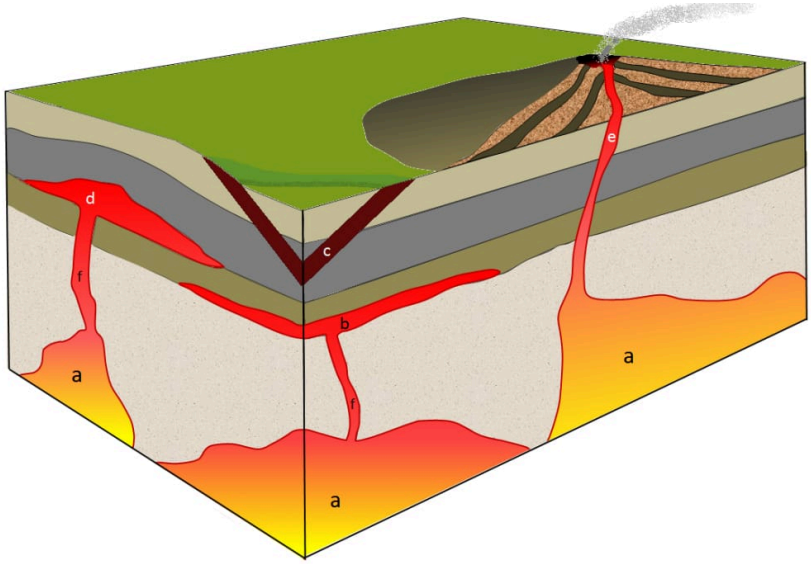


Figure 3.5.2 Depiction of some of the types of plutons. a: stocks (if they coalesce at depth then they might become large enough to be called a batholith), b: sill (a tabular body, in this case parallel to bedding), c: dyke (cross-cuts bedding), d: laccolith (a sill that has pushed up the overlying rock layers), e: pipe (a cylindrical conduit feeding a volcano). The two features labelled f could be pipes or dykes, but from this perspective it's not possible to determine if they are cylindrical or tabular.

Large irregular-shaped plutons are called either **stocks** or **batholiths**. The distinction between the two is made on the basis of the area that is exposed at the surface: if the body has an exposed surface area greater than 100 square kilometers (km^2), then it's a batholith; smaller than 100 km^2 and it's a stock. Batholiths are typically formed only when a number of stocks coalesce beneath the surface to create one large body. One of the largest batholiths in the world is the Coast Range Plutonic Complex, which extends all the way from the Vancouver region to southeastern Alaska (Figure 3.5.3). More accurately, it's many batholiths.

Tabular (sheet-like) plutons are distinguished on the basis of whether or not they are **concordant** with (i.e., parallel to) existing layering (e.g., sedimentary bedding or metamorphic foliation) in the country rock. A **sill** is concordant with existing layering, and a **dyke** is **discordant**. If the country rock has no bedding or foliation, then any tabular body within it is a dyke. Note that the sill-versus-dyke designation is not determined simply by the orientation of the feature. A dyke can be horizontal and a sill can be vertical (if the bedding is vertical). A large dyke can be seen in Figure 3.5.3.

A **laccolith** is a sill-like body that has expanded upward by deforming the overlying rock.

Finally, a **pipe** is a cylindrical body (with a circular, ellipitical, or even irregular cross-section) that served as a conduit for the movement of magma from one location to another. Most known pipes fed volcanoes, although pipes can also connect plutons. It is also possible for a dyke to feed a volcano.



Figure 3.5.3 The Stawamus Chief, part of the Coast Range Plutonic Complex, near to Squamish, B.C. The cliff is about 600 meters (m) high. Most of the dark stripes are a result of algae and lichen growth where the surface is frequently wet, but there is a large (about 10 m across) vertical dyke that extends from bottom to top.

As discussed already, plutons can interact with the rocks into which they are intruded, sometimes leading to partial melting of the country rock or to stoping and formation of xenoliths. And, as we'll see in Chapter 7, the heat of a body of magma can lead to metamorphism of the country rock. The country rock can also have an effect on the magma within a pluton. The most obvious such effect is the formation of a chilled margin along the edges of the pluton, where it came in contact with country rock that was significantly colder than the magma. Within the chilled margin, the magma cooled more quickly than in the centre of the dyke, so the texture is finer and the color may be different. An example is shown in Figure 3.5.4.



Figure 3.5.4 A mafic dyke with chilled margins within basalt at Nanoose, B.C. The coin is 24 mm in diameter. The dyke is about 25 centimeters (cm) across and the chilled margins are 2 cm wide.

Exercise 3.7 Pluton Problems

Figure 3.5.5 shows a cross-section through part of the crust showing a variety of intrusive igneous rocks. Except for the granite (a), all of these rocks are mafic in composition. Indicate whether each of the plutons labelled **a** to **e** on the diagram below is a **dyke**, a **sill,** a **stock**, or a **batholith**.

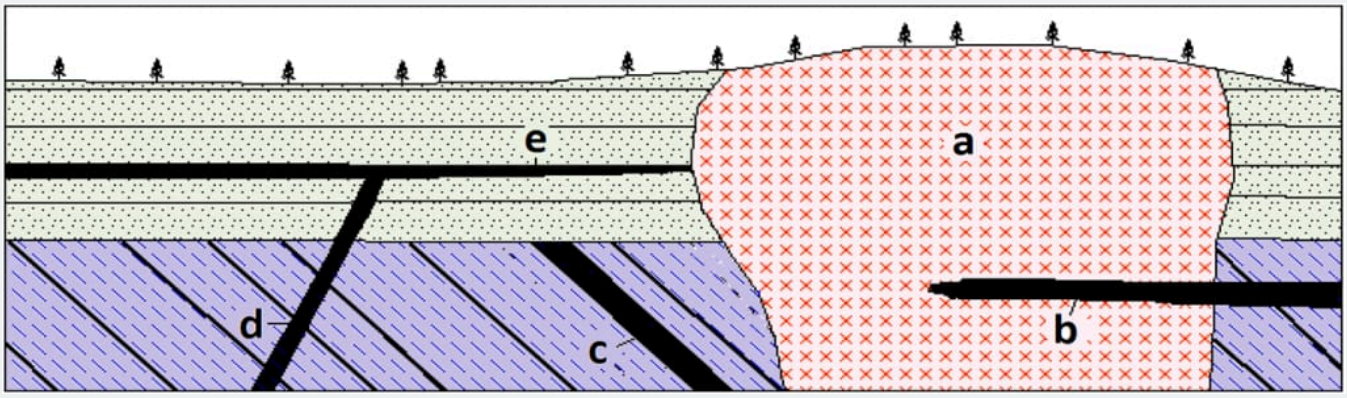


Figure 3.5.5

See Appendix 3 for Exercise 3.7 answers.

Media Attributions

- Figures 3.5.1, 3.5.2, 3.5.3, 3.5.4, 3.5.5: © Steven Earle. CC BY.

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1. "Country rock" is not necessarily music to a geologist's ears. The term refers to the original "rock of the country" or region, and hence the rock into which the magma intruded to form a pluton. ↵

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3.6: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
3.1 The Rock Cycle	The three types of rocks are igneous: formed from magma; sedimentary: formed from fragments of other rocks or precipitation from solution; and metamorphic: formed when existing rocks are altered by heat, pressure, and/or chemical action. The rock cycle summarizes the processes that contribute to cycling of rock material among these three types. The rock cycle is driven by Earth's internal heat, and by processes happening at the surface, which are driven by solar energy.
3.2 Magma and Magma Formation	Magma is molten rock, and in most cases, it forms from partial melting of existing rock. The two main processes of magma formation are decompression melting and flux melting. Magmas range in composition from ultramafic to felsic. Mafic rocks are rich in iron, magnesium, and calcium, and have around 50% silica. Felsic rocks are rich in silica (~75%) and have lower levels of iron, magnesium, and calcium and higher levels of sodium and potassium than mafic rocks. Intermediate rocks have compositions between felsic and mafic.
3.3 Crystallization of Magma	As a body of magma starts to cool, the first process to take place is the polymerization of silica tetrahedra into chains. This increases the magma's viscosity (makes it thicker) and because felsic magmas have more silica than mafic magmas, they tend to be more viscous. The Bowen reaction series allows us to predict the order of crystallization of magma as it cools. Magma can be modified by fractional crystallization (separation of early-forming crystals) and by incorporation of material from the surrounding rocks by partial melting.
3.4 Classification of Igneous Rock	Igneous rocks are classified based on their mineral composition and texture. Felsic igneous rocks have less than 20% ferromagnesian silicates (amphibole and/or biotite) plus varying amounts of quartz and both potassium and plagioclase feldspars. Mafic igneous rocks have more than 50% ferromagnesian silicates (primarily pyroxene) plus plagioclase feldspar. Most intrusive igneous rocks are phaneritic (crystals are visible to the naked eye). If there were two stages of cooling (slow then fast), the texture may be porphyritic (large crystals in a matrix of smaller crystals). If the cooling was extremely slow, or if water was present during cooling, the texture may be pegmatitic (very large crystals).
3.5 Intrusive Igneous Bodies	Magma intrudes into country rock by pushing it aside or melting through it. Intrusive igneous bodies tend to be either irregular (stocks and batholiths), tabular (dykes and sills), or pipe-like. Batholiths have exposed areas of greater than 100 km^2, while stocks are smaller. Sills are parallel to existing layering in the country rock, while dykes cut across layering. A pluton that intruded into cold rock it is likely to have a chilled margin.

Questions for Review

Answers to Review Questions at the end of each chapter are provided in Appendix 2.

1. What processes must take place to transform rocks into sediment?
2. What processes normally take place in the transformation of sediments to sedimentary rock?
3. What are the processes that lead to the formation of a metamorphic rock?
4. What is the significance of the term *reaction* in the name of the Bowen reaction series?
5. Why is it common for plagioclase crystals to be zoned from relatively calcium-rich in the middle to more sodium-rich on the outside?
6. What must happen within a magma chamber for fractional crystallization to take place?
7. Explain the difference between aphanitic and phaneritic textures.
8. Explain the difference between porphyritic and pegmatitic textures.
9. Name the following rocks:
    1. An extrusive rock with 40% Ca-rich plagioclase and 60% pyroxene
    2. An intrusive rock with 65% plagioclase, 25% amphibole, and 10% pyroxene
    3. An intrusive rock with 25% quartz, 20% orthoclase, 50% feldspar, and minor amounts of biotit
10. With respect to tabular intrusive bodies, what is the difference between a concordant body and a discordant body?
11. Why does a dyke commonly have a fine-grained margin?

12. What is the difference between a batholith and a stock?
13. Describe two ways in which batholiths intrude into existing rock.
14. Why is compositional layering a common feature of mafic plutons but not of felsic plutons?

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4: VOLCANISM

A volcano is any location where magma comes to the surface, or has done so within the past several million years. This can include eruptions on the ocean floor (or even under the water of lake), where they are called **subaqueous eruptions**, or on land, where they are called **subaerial eruptions**. Not all volcanic eruptions produce the volcanic mountains with which we are familiar; in fact most of Earth's volcanism takes place along the spreading ridges on the sea floor and does not produce volcanic mountains at all — not even sea-floor mountains.

4.1: Plate Tectonics and Volcanism

4.2: Magma Composition and Eruption Style

4.3: Types of Volcanoes

4.4: Volcanic Hazards

4.5: Monitoring Volcanoes and Predicting Eruptions

4.6: Volcanoes in British Columbia

4.7: Summary

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4.1: Plate Tectonics and Volcanism

The relationships between plate tectonics and volcanism are shown on Figure 4.1.1. As summarized in Chapter 3, magma is formed at three main plate-tectonic settings: divergent boundaries (decompression melting), convergent boundaries (flux melting), and mantle plumes (decompression melting).

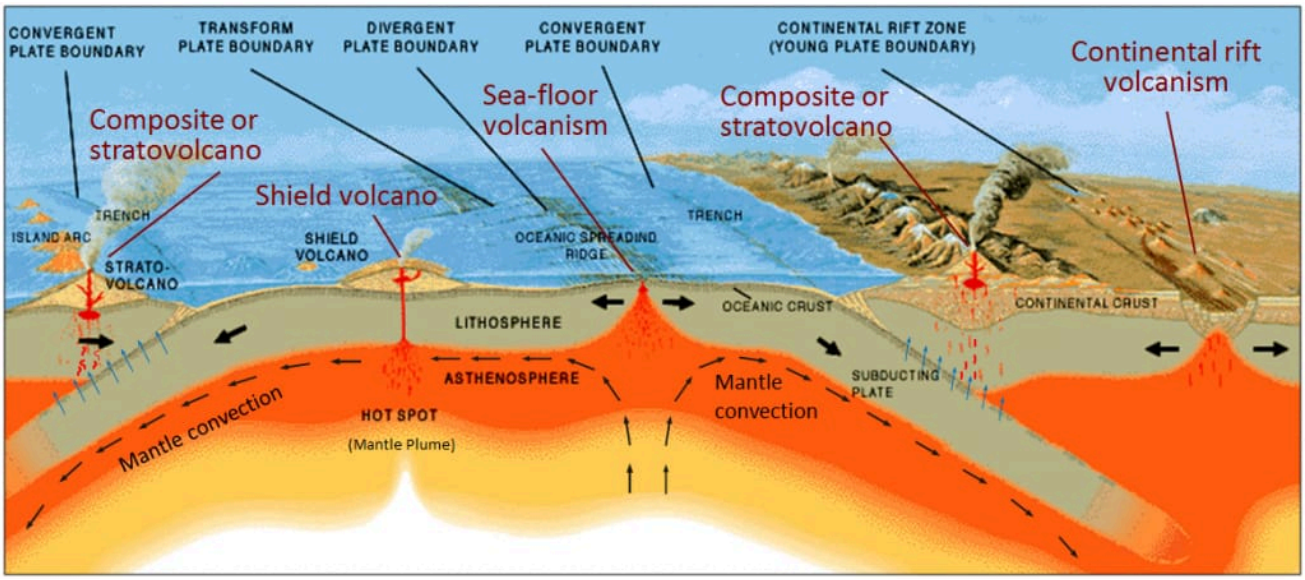


Figure 4.1.1 The plate-tectonic settings of common types of volcanism. Composite volcanoes form at subduction zones, either on ocean-ocean convergent boundaries (left) or ocean-continent convergent boundaries (right). Both shield volcanoes and cinder cones form in areas of continental rifting. Shield volcanoes form above mantle plumes, but can also form at other tectonic settings. Sea-floor volcanism can take place at divergent boundaries, mantle plumes and ocean-ocean-convergent boundaries.

The mantle and crustal processes that take place in areas of volcanism are illustrated in Figure 4.1.2. At a spreading ridge, hot mantle rock moves slowly upward by convection (centimetre/year), and within about 60 kilometers (km) of the surface, partial melting starts because of decompression. Over the triangular area shown in Figure 4.1.2a, about 10% of the ultramafic mantle rock melts, producing mafic magma that moves upward toward the axis of spreading (where the two plates are moving away from each other). The magma fills vertical fractures produced by the spreading and spills out onto the sea floor to form basaltic **pillows** (more on that later) and lava flows. There is spreading-ridge volcanism taking place about 200 km offshore from the west coast of Vancouver Island.

Exercise 4.1 How thick is the oceanic crust?

Figure 4.1.2a shows a triangular zone about 60 km thick; within this zone, approximately 10% of the mantle rock melts to form oceanic crust. Based on this information, approximately how thick do you think the resulting oceanic crust should be?

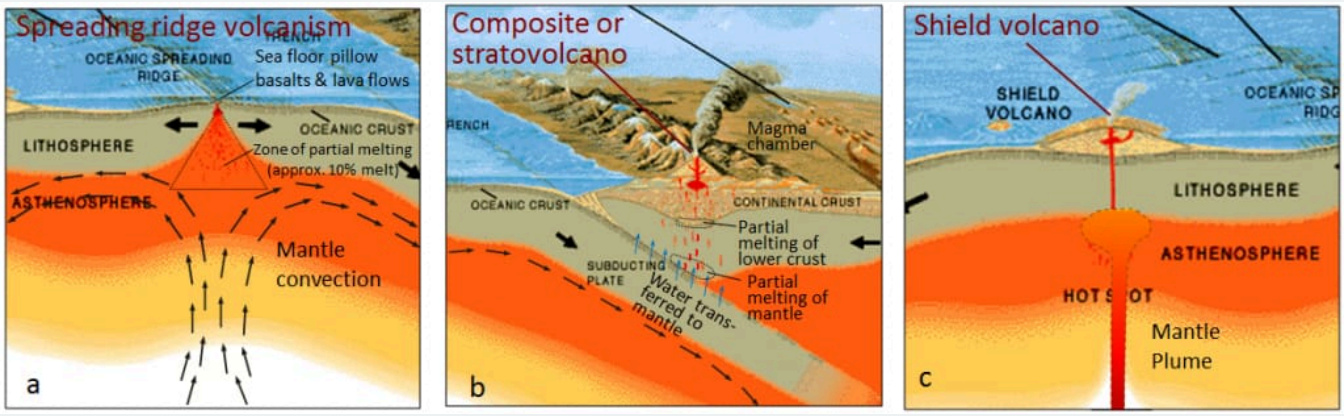


Figure 4.1.2 The processes that lead to volcanism in the three main volcanic settings on Earth: (a) volcanism related to plate divergence, (b) volcanism at an ocean-continent boundary (Similar processes take place at an ocean-ocean convergent boundary), and (c) volcanism related to a mantle plume.

See Appendix 3 for Exercise 4.1 answers.

At an ocean-continent convergent boundary, part of a plate that is made up of oceanic crust is subducting beneath part of another plate made up of continental crust. At an ocean-ocean convergent boundary, oceanic crust is being subducted beneath another oceanic-crust plate.[/footnote] (Figure 4.1.2b). In both situations the oceanic crust is heated up, and while there isn't enough heat to melt the subducting crust, there is enough heat to force the water out of some of its minerals. This released water rises into the overlying mantle where it contributes to flux melting of the mantle rock. The mafic magma produced rises through the mantle to the base of the crust. There it contributes to partial melting of crustal rock, and thus it assimilates much more felsic material. That magma, now likely intermediate in composition, continues to rise and assimilate crustal material. In the upper part of the crust, it accumulates into plutons. From time to time, the magma from the plutons rises toward surface, leading to volcanic eruptions. Mount Garibaldi (Figures 4.0.1 and 4.0.2) is an example of subduction-related volcanism.

A mantle plume is an ascending column of hot rock (not magma) that originates deep in the mantle, possibly just above the core-mantle boundary. Mantle plumes are thought to rise approximately 10 times faster than the rate of mantle convection. The ascending column may be on the order of kilometers to tens of kilometers across, but near the surface it spreads out to create a mushroom-style head that is several tens to over 100 km across. Near the base of the lithosphere (the rigid part of the mantle), the mantle plume (and possibly some of the surrounding mantle material) partially melts to form mafic magma that rises to feed volcanoes. Since most mantle plumes are beneath the oceans, the early stages of volcanism typically take place on the sea floor. Over time, islands may form like those in Hawaii.

Volcanism in northwestern B.C. (Figures 4.1.3 and 4.1.4) is related to continental rifting. This area is not at a divergent or convergent boundary, and there is no evidence of an underlying mantle plume. A likely explanation is that the crust of northwestern B.C. is being stressed by the northward movement of the Pacific Plate against the North America Plate, and the resulting crustal fracturing provides a conduit for the flow of magma from the mantle. This may, or may not, be an early stage of continental rifting, such as that found in eastern Africa.

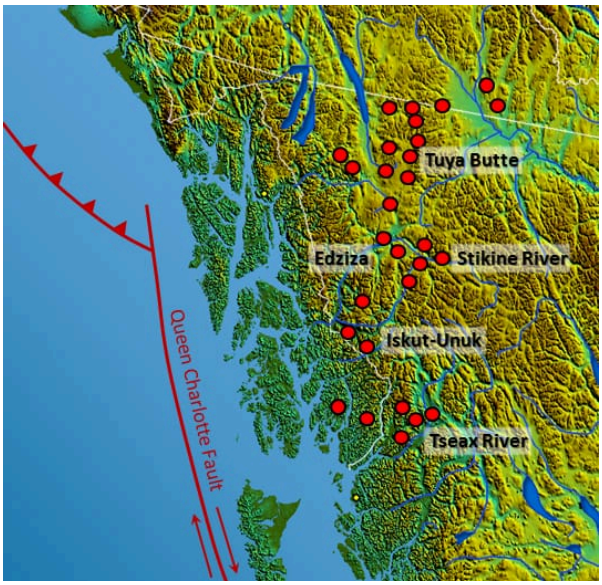


Figure 4.1.3 Volcanoes and volcanic fields in the Northern Cordillera Volcanic Province, B.C.



Figure 4.1.4 Volcanic rock at the Tseax River area, northwestern B.C.

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4.2: Magma Composition and Eruption Style

As noted in the previous section, the types of magma produced in the various volcanic settings can differ significantly. At divergent boundaries and oceanic mantle plumes, where there is little interaction with crustal materials the magma tends to be consistently mafic. At subduction zones, where the magma ascends through significant thicknesses of crust, interaction between the magma and the crustal rock—some of which is quite felsic—leads to increases in the felsic character of the magma.

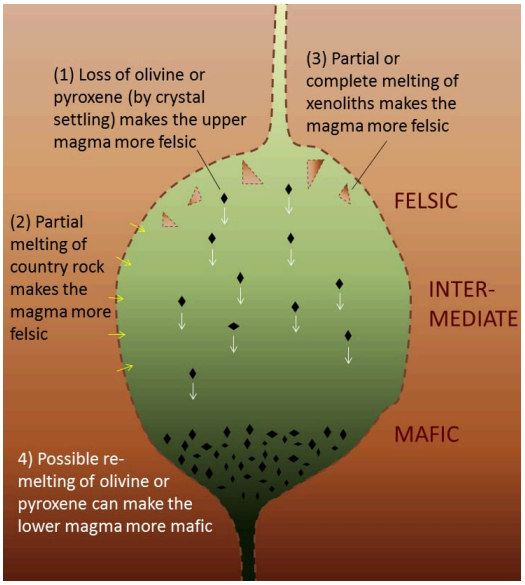


Figure 4.2.1 The important processes that lead to changes in the composition of magmas stored within magma chambers within relatively felsic rocks of the crust.

As illustrated in Figure 4.2.1, several processes can make magma that is stored in a chamber within the crust more felsic than it was to begin with, and can also contribute to development of vertical zonation from more mafic at the bottom to more felsic at the top. Partial melting of country rock and country-rock xenoliths increases the overall felsic character of the magma; first, because the country rocks tends to be more felsic than the magma, and second, because the more felsic components of the country rock melt preferentially. Settling of ferromagnesian crystals from the upper part of the magma, and possible remelting of those crystals in the lower part can both contribute to the vertical zonation from relatively mafic at the bottom to more felsic at the top.

From the perspective of volcanism there are some important differences between felsic and mafic magmas. First, as we've already discussed, felsic magmas tend to be more viscous because they have more silica, and hence more polymerization. Second, felsic magmas tend to have higher levels of volatiles; that is, components that behave as gases during volcanic eruptions. The most abundant volatile in magma is water (H_2O), followed typically by carbon dioxide (CO_2), and then by sulfur dioxide (SO_2).

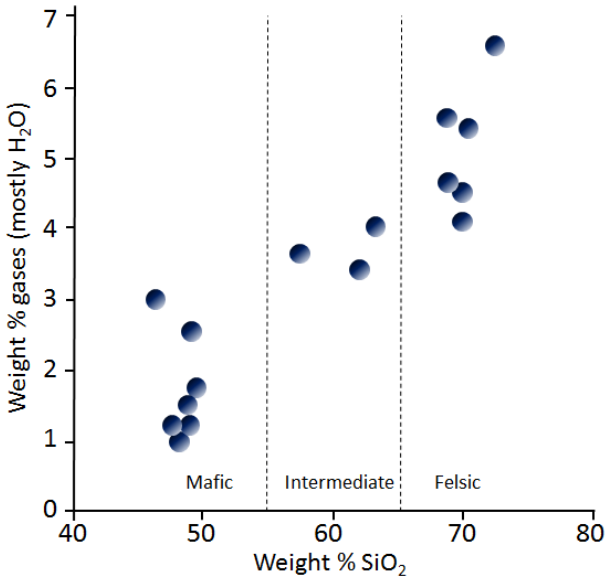


Figure 4.2.2 Variations in the volatile compositions of magmas as a function of silica content.

The general relationship between the SiO_2 content of magma and the amount of volatiles is shown in Figure 4.2.2. Although there are many exceptions to this trend, mafic magmas typically have 1% to 3% volatiles, intermediate magmas have 3% to 4% volatiles, and felsic magmas have 4% to 7% volatiles.

Differences in viscosity and volatile levels have significant implications for the nature of volcanic eruptions. When magma is deep beneath the surface and under high pressure from the surrounding rocks, the gases remain dissolved. As magma approaches the surface, the pressure exerted on it decreases. Gas bubbles start to form, and the more gas there is in the magma, the more bubbles form. If the magma is runny enough for gases to rise up through it and escape to surface, the pressure will not become excessive. Assuming that it can break through to the surface, the magma will flow out relatively gently. An eruption that involves a steady non-violent flow of magma is called **effusive**.

Exercise 4.2 Under pressure!



Figure 4.2.3

A good analogy for a magma chamber in the upper crust is a plastic bottle of pop on the supermarket shelf. Go to a supermarket and pick one up off the shelf (something not too dark). You'll find that the bottle is hard because it was bottled under pressure, and you should be able to see that there are no gas bubbles inside.

Buy a small bottle of pop (you don't have to drink it!) and open it. The bottle will become soft because the pressure is released, and small bubbles will start forming. If you put the lid back on and shake the bottle (best to do this outside!), you'll enhance the processes of bubble formation, and when you open the lid, the pop will come gushing out, just like an explosive volcanic eruption.

A pop bottle is a better analogue for a volcano than the old baking soda and vinegar experiment that you did in elementary school, because pop bottles—like volcanoes—come pre-charged with gas pressure. All we need to do is release the confining pressure and the gases come bubbling out, forcing the pop with them.

If the magma is felsic, and therefore too viscous for gases to escape easily, or if it has a particularly high gas content, it is likely to be under high pressure. Viscous magma doesn't flow easily, so even if there is a conduit for it to move towards surface, it may not flow out. Under these circumstances pressure will continue to build as more magma moves up from beneath and gases continue to exsolve. Eventually some part of the volcano will break and then all of that pent-up pressure will lead to an explosive eruption.

Mantle plume and spreading-ridge magmas tend to be consistently mafic, so effusive eruptions are the norm. At subduction zones, the average magma composition is likely to be close to intermediate, but as we've seen, magma chambers can become zoned and so compositions ranging from felsic to mafic are possible—even likely. Eruption styles can be correspondingly variable.

Image Descriptions

Figure 4.2.1 **image description:** Four processes that can affect the composition of magma stored in chambers.

1. Loss of olivine or pyroxene (by crystal settling) makes the upper magma more felsic.
2. Partial melting of country rock makes the magma more felsic.
3. Partial or complete melting of xenoliths makes the magma more felsic.
4. Possible re-melting of olivine or pyroxene can make the lower magma more mafic.

[Return to Figure 4.2.1]

Figure 4.2.2 *image description: The following table describes the range of data points adapted from the original scatter plot graph.*

Type of Magma	Weight % of gasses (mostly H_2O)	Weight % of SiO_2
Mafic	1 to 3%	47 to 50%
Intermediate	3 to 4%	57 to 64%
Felsic	4 to 7%	69 to 72%

[Return to Figure 4.2.2]

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- Figure 4.2.2: Original image by Schminke, 2004. (Schminke, H-U., 2004, Volcanism, Springer-Verlag, Heidelberg). Modified by Steven Earle.
- Figure 4.2.3: Champagne uncorking photographed with a high speed air-gap flash © Niels Noordhoek. CC BY-SA.

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4.3: Types of Volcanoes

There are numerous types of volcanoes or volcanic sources; some of the more common ones are summarized in Table 4.1.

Table 4.1 A summary of the important types of volcanism

[Skip Table]

Type	Tectonic Setting	Size and Shape	Magma and Eruption Characteristics	Example
Cinder cone	Various; some form on the flanks of larger volcanoes	Small (10s to 100s of meters) and steep (Greater than 20°)	Most are mafic and form from the gas-rich early stages of a shield- or rift-associated eruption	Eve Cone, northern B.C.
Composite volcano	Almost all are at subduction zones	Medium size (1000s of meters high and up to 20 km across) and moderate steepness (10° to 30°)	Magma composition varies from felsic to mafic, and from explosive to effusive	Mount St. Helens
Shield volcano	Most are at mantle plumes; some are on spreading ridges	Large (up to several 1,000 meters high and up to 200 kilometers across), not steep (typically 2° to 10°)	Magma is almost always mafic, and eruptions are typically effusive, although cinder cones are common on the flanks of shield volcanoes	Kilauea, Hawaii
Large igneous provinces	Associated with “super” mantle plumes	Enormous (up to millions of square kiometers) and 100s of meters thick	Magma is always mafic and individual flows can be 10s of meters thick	Columbia River basalts
Sea-floor volcanism	Generally associated with spreading ridges but also with mantle plumes	Large areas of the sea floor associated with spreading ridges	Pillows form at typical eruption rates; lava flows develop if the rare of flow is faster	Juan de Fuca ridge
Kimberlite	Upper-mantle sourced	The remnants are typically 10s to 100s of meters across	Most appear to have had explosive eruptions forming cinder cones; the youngest one is dated at about 10 ka, and all others are at least 30 Ma	Lac de Gras Kimberlite Field, N.W.T.

The sizes and shapes of typical shield, composite, and cinder-cone volcanoes are compared in Figure 4.3.1, although, to be fair, Mauna Loa is the largest shield volcano on Earth; all others are smaller. Mauna Loa rises from the surrounding flat sea floor, and its diameter is in the order of 200 km. Its elevation is 4,169 m above sea level. Mount St. Helens, a composite volcano of average size, rises above the surrounding hills of the Cascade Range. Its diameter is about 6 km, and its height is 2,550 m above sea level. Cinder cones are much smaller. On this drawing, even a large cinder cone is just a dot.

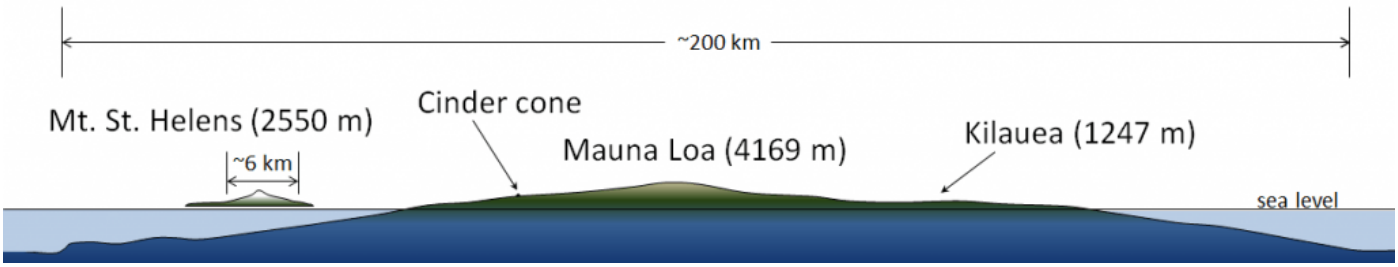


Figure 4.3.1 Profiles of Mauna Loa shield volcano, Mount St. Helens composite volcano, and a large cinder cone.

Cinder Cones

Cinder cones, like Eve Cone in northern B.C. (Figure 4.3.2), are typically only a few hundred meters in diameter, and few are more

than 200 m high. Most are made up of fragments of **vesicular** mafic rock (scoria) that were expelled as the magma boiled when it approached the surface, creating fire fountains. In many cases, these later became the sites of effusive lava flows when the gases were depleted. Most cinder cones are **monogenetic**, meaning that they formed during a single eruptive phase that might have lasted weeks or months. Because cinder cones are made up almost exclusively of loose fragments, they have very little strength. They can be easily, and relatively quickly, eroded away.

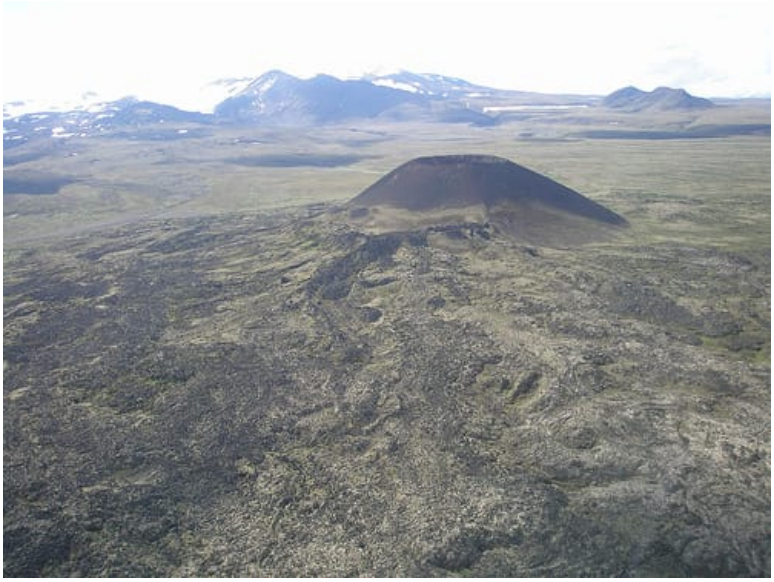


Figure 4.3.2 Eve Cone, situated near to Mount Edziza in northern B.C., formed approximately 700 years ago.

Composite Volcanoes

Composite volcanoes, like Mount St. Helens in Washington State (Figure 4.3.3), are almost all associated with subduction at convergent plate boundaries—either ocean-continent or ocean-ocean boundaries (Figure 4.3.2b). They can extend up to several thousand meters from the surrounding terrain, and, with slopes ranging up to 30˚ They can be up to about 20 km across. At many such volcanoes, magma is stored in a magma chamber in the upper part of the crust. For example, at Mount St. Helens, there is evidence of a magma chamber that is approximately 1 kilometre wide and extends from about 6 km to 14 km below the surface (Figure 4.3.4). Systematic variations in the composition of volcanism over the past several thousand years at Mount St. Helens imply that the magma chamber is zoned, from more felsic at the top to more mafic at the bottom.



Figure 4.3.3 The north side of Mount St. Helens in southwestern Washington State, 2003. The large 1980 eruption reduced the height of the volcano by 400 m, and a sector collapse removed a large part of the northern flank. Between 1980 and 1986 the slow eruption of more mafic and less viscous lava led to construction of a dome inside the crater.

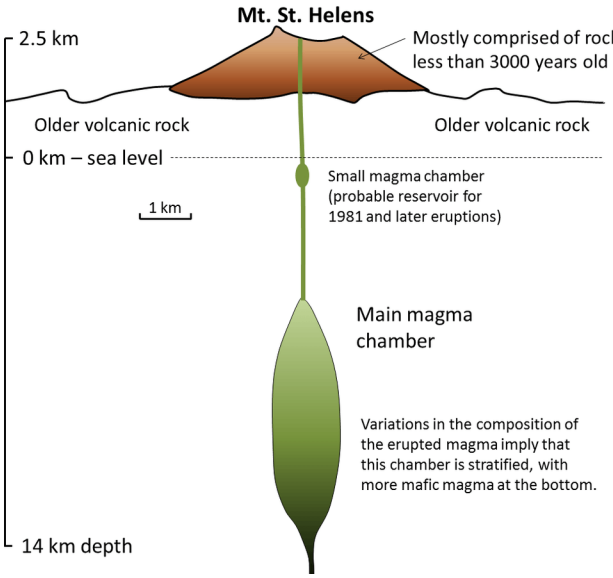


Figure 4.3.4 A cross-section through the upper part of the crust at Mount St. Helens showing the zoned magma chamber.

Mafic eruptions (and some intermediate eruptions), on the other hand, produce lava flows; the one shown in Figure 4.3.5b is thick enough (about 10 m in total) to have cooled in a **columnar jointing** pattern (Figure 4.3.7). Lava flows both flatten the profile of the volcano (because the lava typically flows farther than pyroclastic debris falls) and protect the fragmental deposits from erosion. Even so, composite volcanoes tend to erode quickly. Patrick Pringle, a volcanologist with the Washington State Department of Natural Resources, describes Mount St. Helens as a "pile of junk." The rock that makes up Mount St. Helens ranges in composition from rhyolite (Figure 4.3.5a) to basalt (Figure 4.3.5b); this implies that the types of past eruptions have varied widely in character. As already noted, felsic magma doesn't flow easily and doesn't allow gases to escape easily. Under these circumstances, pressure builds up until a conduit opens, and then an explosive eruption results from the gas-rich upper part of the magma chamber, producing **pyroclastic** debris, as shown on Figure 4.3.5a. This type of eruption can also lead to rapid melting of ice and snow on a volcano, which typically triggers large mudflows known as **lahars** (Figure 4.3.5a). Hot, fast-moving pyroclastic flows and lahars are the two main causes of casualties in volcanic eruptions. Pyroclastic flows killed approximately 30,000 people during the 1902 eruption of Mount. Pelée on the Caribbean island of Martinique. Most were incinerated in their homes. In 1985 a massive lahar, triggered by the eruption of Nevado del Ruiz, killed 23,000 people in the Colombian town of Armero, about 50 km from the volcano.

In a geological context, composite volcanoes tend to form relatively quickly and do not last very long. Mount St. Helens, for example, is made up of rock that is all younger than 40,000 years; most of it is younger than 3,000 years. If its volcanic activity ceases, it might erode away within a few tens of thousands of years. This is largely because of the presence of pyroclastic eruptive material, which is not strong.



a b

Figure 4.3.5 Mount St. Helens volcanic deposits: (a) lahar deposits (L) and felsic pyroclastic deposits (P) and (b) a columnar basalt lava flow. The two photos were taken at locations only about 500 m apart.

Exercise 4.3 Volcanoes and Subduction

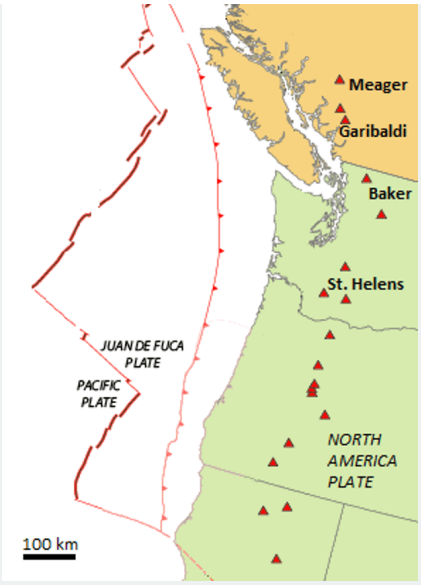


Figure 4.3.6

The map shown here illustrates the interactions between the North America, Juan de Fuca, and Pacific Plates off the west coast of Canada and the United States. The Juan de Fuca Plate is forming along the Juan de Fuca ridge, and is then subducted beneath the North America Plate along the red line with teeth on it ("Subduction boundary").

1. Using the scale bar in the lower left of the map, estimate the average distance between the subduction boundary and the Cascadia composite volcanoes.
2. If the subducting Juan de Fuca Plate descends 40 km for every 100 km that it moves inland, what is its likely depth in the area where volcanoes are forming?

See Appendix 3 for Exercise 4.3 answers.

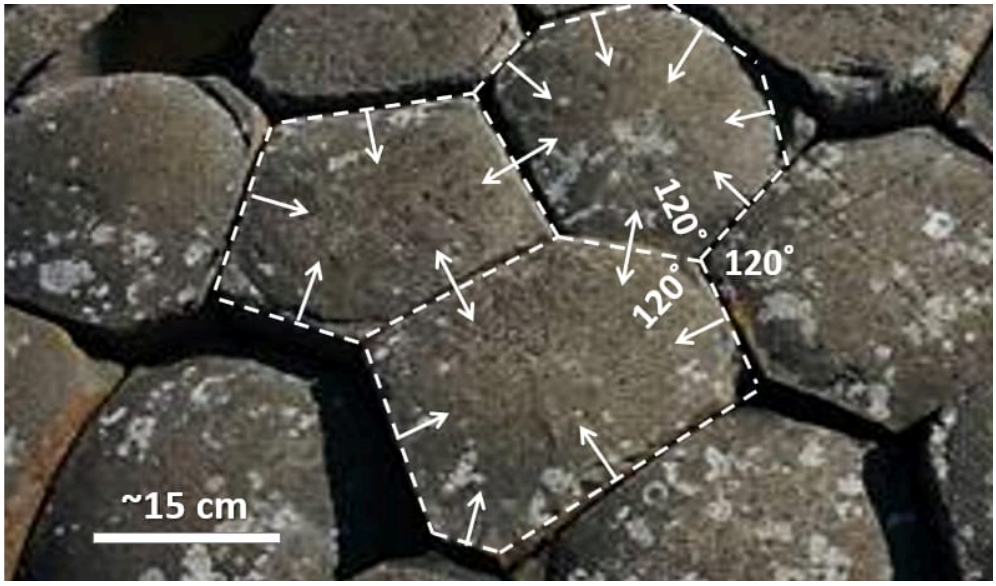


Figure 4.3.7 The development of columnar jointing in basalt, here seen from the top looking down. As the rock cools it shrinks, and because it is very homogenous it shrinks in a systematic way. When the rock breaks it does so with approximately 120˚ angles between the fracture planes. The resulting columns tend to be 6-sided but 5- and 7-sided columns also form.

Shield Volcanoes

Most **shield volcanoes** are associated with mantle plumes, although some form at divergent boundaries, either on land or on the sea floor. Because of their non-viscous mafic magma they tend to have relatively gentle slopes (2 to 10˚) and the larger ones can be over 100 km in diameter. The best-known shield volcanoes are those that make up the Hawaiian Islands, and of these, the only active ones are on the big island of Hawaii. Mauna Loa, the world's largest volcano and the world's largest mountain (by volume) last erupted in 1984. Kilauea, arguably the world's most active volcano, has been erupting, virtually without interruption, since 1983. Loihi is an underwater volcano on the southeastern side of Hawaii. It is last known to have erupted in 1996, but may have erupted since then without being detected.

All of the Hawaiian volcanoes are related to the mantle plume that currently lies beneath Mauna Loa, Kilauea, and Loihi (Figure 4.3.8). In this area, the Pacific Plate is moving northwest at a rate of about 7 centimeters (cm) per year. This means that the earlier formed — and now extinct — volcanoes have now moved well away from the mantle plume. As shown on Figure 4.3.8, there is evidence of crustal magma chambers beneath all three active Hawaiian volcanoes. At Kilauea, the magma chamber appears to be several kilometers in diameter, and is situated between 8 km and 11 km below surface.[1]

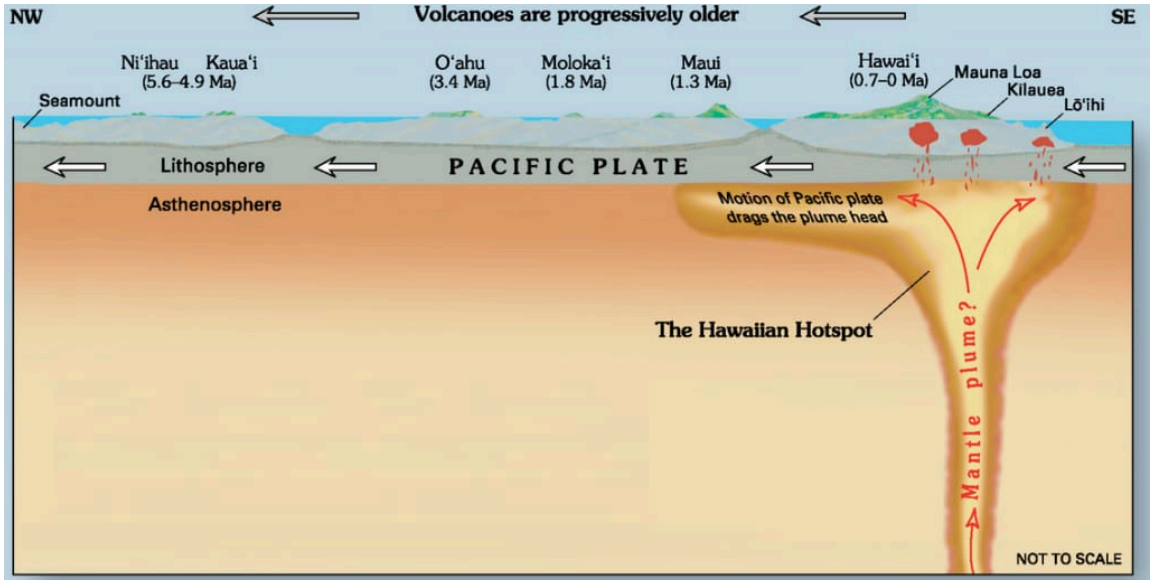


Figure 4.3.8 The mantle plume beneath the volcanoes of the island of Hawaii

Although it is not a prominent mountain (Figure 4.3.2), Kilauea volcano has a large **caldera** in its summit area (Figure 4.3.9). A caldera is a volcanic **crater** that is more than 2 km in diameter; this one is 4 km long and 3 km wide. It contains a smaller feature called Halema'uma'u crater, which has a total depth of over 200 m below the surrounding area. Most volcanic craters and calderas are formed above magma chambers, and the level of the crater floor is influenced by the amount of pressure exerted by the magma body. During historical times, the floors of both Kilauea caldera and Halema'uma'u crater have moved up during expansion of the magma chamber and down during deflation of the chamber.



Figure 4.3.9 Aerial view of the Kilauea caldera. The caldera is about 4 km across, and up to 120 m deep. It encloses a smaller and deeper crater known as Halema'uma'u.

One of the conspicuous features of Kilauea caldera is rising water vapor (the white cloud in Figure 4.3.9) and a strong smell of sulfur (Figure 4.3.10). As is typical in magmatic regions, water is the main volatile component, followed by carbon dioxide and sulfur dioxide. These, and some minor gases, originate from the magma chamber at depth and rise up through cracks in the overlying rock. This degassing of the magma is critical to the style of eruption at Kilauea, which, for most of the past 35 years, has been effusive, not explosive.



Figure 4.3.10 A gas-composition monitoring station (left) within the Kilauea caldera close to the edge of Halema'uma'u crater. The rising clouds are mostly composed of water vapor, but also include carbon dioxide and sulfur dioxide. Sulphur crystals (right) have formed around a gas vent in the caldera.

The Kilauea eruption that began in 1983 started with the formation of a cinder cone at Pu'u 'O'o, approximately 15 km east of the caldera (Figure 4.3.11). The magma feeding this eruption flowed along a major conduit system known as the East Rift, which extends for about 20 km from the caldera, first southeast and then east. Lava fountaining and construction of the Pu'u 'O'o cinder cone (Figure 4.3.12a) continued until 1986 at which time the flow became effusive. From 1986 to 2014, lava flowed from a gap in the southern flank of Pu'u 'O'o down the slope of Kilauea through a **lava tube** (Figure 4.3.12d), emerging at or near the ocean. During 2014 and 2015, the lava flowed northeast toward the community of Pahoa (see Exercise 4.4). In May of 2018 a new eruption started another 15 km east of the 2014/15 flow in the area known as Leilani Estates. The Lower East Rift Flow was active for 6 months. During that time, 35 km^2 of existing land was covered in lava and 3.5 km^2 of new land was created (Figure 4.3.11), about 48 km of road were covered in lava and 716 dwellings were destroyed (see USGS Overview of Kilauea Volcanoe's 2018 eruption[PDF]). Volcanic activity on the East Rift ceased in August 2018, and there has been no activity on Kilauea since then. This appears to mark the end of the eruption cycle that lasted—with only a few short interruptions—for 35 years. Kilauea will almost certainly erupt again within years or decades.

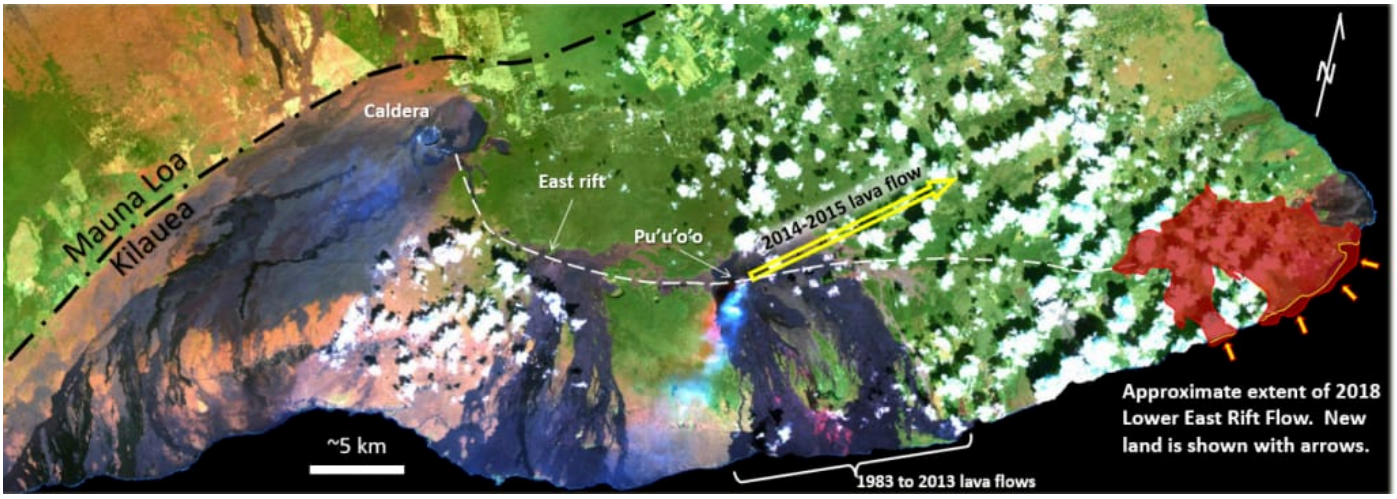


Figure 4.3.11 Satellite image of Kilauea volcano showing the East rift and Pu'u 'O'o, the site of the eruption that started in 1983 and continues to the present day. The 2014-2015 lava flow described in Exercise 4.4 is shown with a yellow arrow, and the area of the 2018 lava flow is shown in red. The puffy white blobs are clouds.

The two main types of textures created during effusive subaerial eruptions are pahoehoe and aa. **Pahoehoe**, ropy lava that forms as non-viscous lava, flows gently, forming a skin that gels and then wrinkles because of ongoing flow of the lava below the surface (Figure 4.3.12b, and "lava flow video"). **Aa**, or blocky lava, forms when magma is forced to flow faster than it is able to (down a slope

for example) (Figure 4.3.12c). **Tephra** (lava fragments) is produced during explosive eruptions, and accumulates in the vicinity of cinder cones.

Figure 4.3.12d is a view into an active lava tube on the southern edge of Kilauea. The red glow is from a stream of very hot lava (~1200°C) that has flowed underground for most of the 8 km from the Pu'u 'O'o vent. Lava tubes form naturally and readily on both shield and composite volcanoes because flowing mafic lava preferentially cools near its margins, forming solid **lava levées** that eventually close over the top of the flow. The magma within a lava tube is not exposed to the air, so it remains hot and fluid and can flow for tens of km, thus contributing to the large size and low slopes of shield volcanoes. The Hawaiian volcanoes are riddled with thousands of old lava tubes, some as long as 50 km.



Figure 4.3.12 Images of Kilauea volcano. (a) Pu'u'O'o cinder cone in the background with tephra in the foreground and aa lava in the middle, (b) Formation of pahoehoe on the southern edge of Kilauea, (c) Formation of aa on a steep slope on Kilauea, (d) Skylight in an active lava tube, Kilauea. Photos B & C taken in 2002 photos A & D taken in 2007.

Kilauea started forming at approximately 300 ka, while neighbouring Mauna Loa dates back to 700 ka and nearby Mauna Kea ito around 1 Ma. If volcanism continues above the Hawaii mantle plume in the same manner that it has since 85 Ma, it is likely that Kilauea will continue to erupt for at least another 500,000 years. By that time, its neighbour, Loihi, will have emerged from the sea floor, and its other neighbours, Mauna Loa and Mauna Kea, will have become significantly eroded, like their cousins, the islands to the northwest (Figure 4.3.8).

Exercise 4.4 Kilauea's 2014 lava flow

The U.S. Geological Survey Hawaii Volcano Observatory (HVO) map shown here, dated January 29, 2015, shows the outline of lava that started flowing northeast from Pu'u 'O'o on June 27, 2014 (the "June 27th Lava flow," a.k.a. the "East Rift Lava Flow"). The flow reached the nearest settlement, Pahoa, on October 29, 124 days later. After damaging some infrastructure west of Pahoa, the flow stopped advancing. A new outbreak occurred November 1, branching out to the north from the main flow.

What is the average rate of advance of the flow front from June 27 to October 29, 2014, in meters per day and meters per hour?

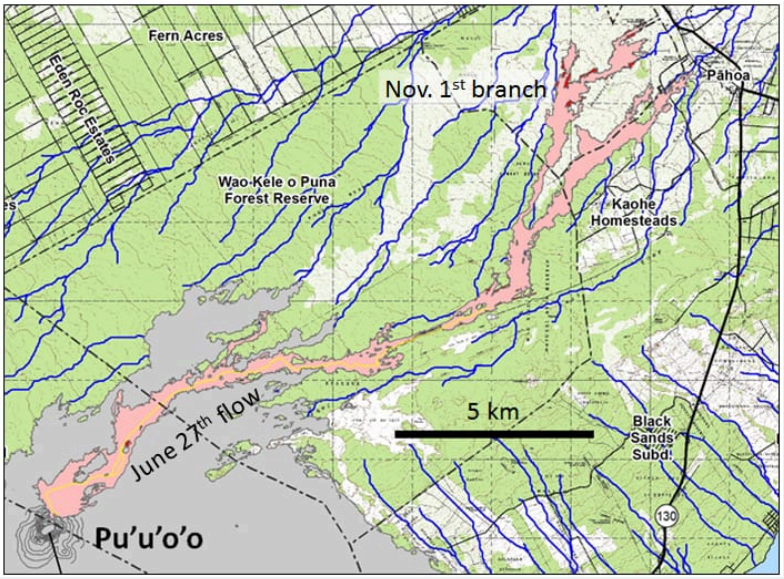


Figure 4.3.13

See Appendix 3 for Exercise 4.4 Answers.

Large Igneous Provinces

While the Hawaii mantle plume has produced a relatively low volume of magma for a very long time (~85 Ma), other mantle plumes are less consistent, and some generate massive volumes of magma over relatively short time periods. Although their origin is still controversial, it is thought that the volcanism leading to **large igneous provinces** (LIP) is related to very high volume but relatively short duration bursts of magma from mantle plumes. An example of an LIP is the Columbia River Basalt Group (CRGB), which extends across Washington, Oregon, and Idaho (Figure 4.3.14). This volcanism, which covered an area of about 160,000 square kilometers (km^2) with basaltic rock up to several hundred meters thick, took place between 17 and 14 Ma.

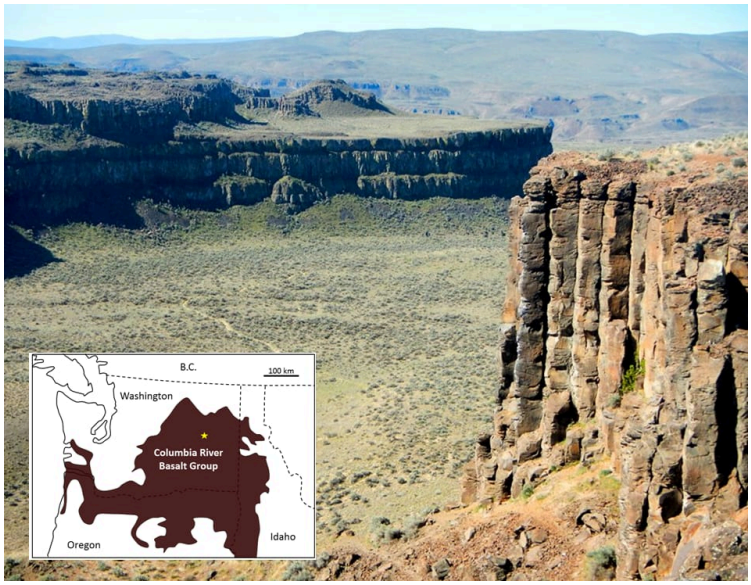


Figure 4.3.14 A part of the Columbia River Basalt Group at Frenchman Coulee, eastern Washington. All of the flows visible here have formed large (up to two m in diameter) columnar basalts, a result of relatively slow cooling of flows that are tens of meters thick. The inset map shows the approximate extent of the 17 to 14 Ma Columbia River Basalts, with the location of the photo shown as a star.

Most other LIP eruptions are much bigger. The Siberian Traps (also basalt), which erupted at the end of the Permian period at 250 Ma, are estimated to have produced approximately 40 times as much lava as the CRBG.

The mantle plume that is assumed to be responsible for the CRBG is now situated beneath the Yellowstone area, where it leads to felsic volcanism. Over the past 2 Ma three very large explosive eruptions at Yellowstone have yielded approximately 900 cubic kilometers (km^3) of felsic magma, about 900 times the volume of the 1980 eruption of Mount St. Helens, but only 5% of the volume of mafic magma in the CRBG.

Sea-Floor Volcanism

Some LIP eruptions occur on the sea floor, the largest known being the one that created the Ontong Java plateau in the western Pacific Ocean at around 122 Ma. But most sea-floor volcanism originates at divergent boundaries and involves relatively low-volume eruptions. Under these conditions, hot lava that oozes out into the cold seawater quickly cools on the outside and then behaves a little like toothpaste. The resulting blobs of lava are known as **pillows**, and they tend to form piles around a sea-floor lava vent (Figure 4.3.15). In terms of area, there is very likely more pillow basalt on the sea floor than any other type of rock on Earth.

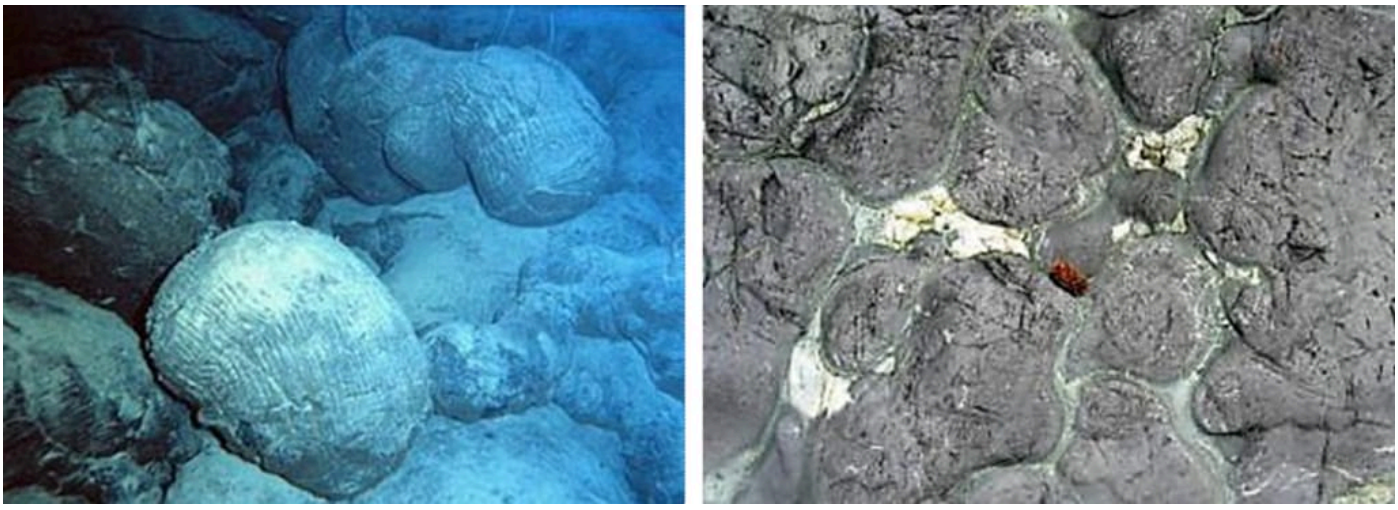


Figure 4.3.15 (Left) Modern sea-floor pillows in the south Pacific. (Right) Ancient sea-floor pillow basalts. Eroded 40 to 50 Ma pillows on the shore of Vancouver Island, near to Sooke. The pillows are 30 to 40 cm in diameter.

Kimberlites

While all of the volcanism discussed so far is thought to originate from partial melting in the upper mantle or within the crust, there is a special class of volcanoes called **kimberlites** that have their origins much deeper in the mantle, at depths of 150 km to 450 km. During a kimberlite eruption, material from this depth may make its way to surface quickly (hours to days) with little interaction with the surrounding rocks. As a result, kimberlite eruptive material is representative of mantle compositions: it is ultramafic.

Kimberlite eruptions that originate at depths greater than 200 km, within areas beneath old thick crust (**shields**), traverse the region of stability of diamond in the mantle, and in some cases, bring diamond-bearing material to the surface. All of the diamond deposits on Earth are assumed to have formed in this way; an example is the rich Ekati Mine in the Northwest Territories (Figure 4.3.16).



Figure 4.3.16 Ekati diamond mine, Northwest Territories, part of the Lac de Gras kimberlite field

The kimberlites at Ekati erupted between 45 and 60 Ma. Many kimberlites are older, some much older. There have been no kimberlite eruptions in historic times. The youngest known kimberlites are in the Igwisi Hills in Tanzania and are only about 10,000 years old. The next youngest known are dated to about 30 Ma.

The Smithsonian Institution maintains a comprehensive catalogue of the world's volcanoes, with information and eruptive history for nearly 2700 volcanic sites. If you spend some time looking around that site you'll discover the frequency of eruptions at different volcanoes is enormously variable, although we can make some generalizations. Focusing just on shield volcanoes and composite volcanoes some of the data are as follows:

Table 4.2 Eruptions of composite and shield volcanoes

Composite volcanoes	**Shield volcanoes**
Avachinsky (Russia): 5 eruptions over the past 7000 years	Fernandina (Galapagos): 31 eruptions over the past 1000 years
Pinatubo (Philippines): 4 eruptions over the past 9000 years	Kilauea (Hawaii): 62 eruptions over the past 250 years
Adams (Oregon, USA): 6 eruptions over the past 7000 years	Nyamuragira (Congo): 48 eruptions over the past 154 years

Based only on these numbers it is evident that, in general, shield volcanoes are much more active than composite volcanoes, but there are many exceptions to this trend. Some composite volcanoes are nearly as active as the shield volcanoes listed here, and some shield volcanoes that are still considered to be "active" are almost as inactive as the composite volcanoes listed here.

Image Descriptions

Figure 4.3.4 image description: Mount St. Helens rises over 2.5 kilometers above sea lever and consists mostly of rock less than 3,000 years old. Underneath the mountain is older volcanic rock. Just below sea level is a small magma chamber, which is a probable reservoir for 1981 and later eruptions. Down 5 to 14 kilometers below sea level is the main magma chamber. Variations in the composition of the erupted magma imply this chamber is stratified, with more magma at the bottom. [Return to Figure 4.3.4]

Figure 4.3.5 image description: Image (A) shows a cliff wall with grey/brown and orange horizontal layers. The sides look soft like they would be easily worn away. The grey/brown layers are lahar deposits and the orange layers are felsic pyroclastic deposits.

Image (B) shows a columnar basalt lava flow which looks like a rocky, stone cliff with vertical layers. [Return to Figure 4.3.5]

Figure 4.3.13 image description: The U.S. Geological Survey Hawaii Volcano Observatory (HVO) map, dated January 29, 2015, shows the outline of lava that started flowing northeast from Pu'u 'O'o on June 27, 2004 (the "June 27th Lava flow," a.k.a. the "East Rift Lava Flow"). The flow reached the nearest settlement, Pahoa, on October 29, after covering a distance of 20 km in 124 days. After damaging some infrastructure west of Pahoa, the flow stopped advancing. A new outbreak occurred November 1, branching out to the north from the main flow about 6 km southwest of Pahoa. [Return to Figure 4.3.13]

Figure 4.3.14 image description: The Columbia River Basalt Group covers most of south eastern Washington state and stretches along the borders between Washington, Idaho, and Oregon. The columnar basalts shown in the photo are in in eastern Washington. They rise up out of a flat valley as tall cliffs. [Return to Figure 4.3.14]

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1. Lin, G, Amelung, F, Lavallee, Y, and Okubo, P, 2014, Seismic evidence for a crustal magma reservoir beneath the upper east rift zone of Kilauea volcano, Hawaii. Geology. V. ↩

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4.4: Volcanic Hazards

There are two classes of volcanic hazards, direct and indirect. Direct hazards are forces that directly kill or injure people, or destroy property or wildlife habitat. Indirect hazards are volcanism-induced environmental changes that lead to distress, famine, or habitat degradation. It is estimated that indirect effects of volcanism have accounted for approximately 8 million deaths during historical times, while direct effects have accounted for fewer than 200,000, or 2.5% of the total. Some of the more important types of volcanic hazards are summarized in Table 4.3.

Table 4.3 A summary of the important volcanic hazards

[Skip Table]

Type	Description	Risk
Tephra emissions	Small particles of volcanic rock emitted into the atmosphere	• Respiration problems for some individuals • Significant climate cooling leading to crop failure and famine • Damage to aircraft
Gas emissions	The emission of gases before, during, and after an eruption	• Climate cooling leading to crop failure and famine • In some cases, widespread poisoning
Pyroclastic density current	A very hot (several 100°C) mixture of gases and volcanic fragments (tephra) that flows rapidly (up to 100s of kilometers per hour (km/h)) down the side of a volcano	Extreme hazard — destroys anything in the way
Pyroclastic fall	Vertical fall of tephra in the area surrounding an eruption	• Thick tephra coverage of areas close to the eruption (1 km to 10s of kilometers) • Collapsed roofs
Lahar	A flow of mud and debris down a channel leading away from a volcano, triggered either by an eruption or a severe rain event	Severe risk of destruction for anything within the channel—lahar mud flows can move at 10s of km/h
Sector collapse/ debris avalanche	The failure of part of a volcano, either due to an eruption or for some other reason, leading to the failure of a large portion of the volcano	Severe risk of destruction for anything in the path of the debris avalanche
Lava flow	The flow of lava away from a volcanic vent	People and infrastructure at risk, but lava flows tend to be slow (less than km/h) and are relatively easy to avoid

Volcanic Gas and Tephra Emissions

Large volumes of **tephra** (rock fragments, mostly **pumice**) and gases are emitted during major **plinian eruptions** (large explosive eruptions with hot gas and tephra columns extending into the stratosphere) at composite volcanoes, and a large volume of gas is released during some very high-volume effusive eruptions. One of the major effects is cooling of the climate by 1° to 2°C for several months to a few years because the dust particles and tiny droplets and particles of sulfur compounds block the sun. The last significant event of this type was in 1991 and 1992 following the large eruption of Mount Pinatubo in the Philippines. A

temperature decrease of 1° to 2°C may not seem like very much, but that is the global average amount of cooling, and cooling was much more severe in some regions and at some times.

Over an eight-month period in 1783 and 1784, a massive effusive eruption took place at the Laki volcano in Iceland. Although there was relatively little volcanic ash involved, a massive amount of sulfur dioxide was released into the atmosphere, along with a significant volume of hydrofluoric acid (HF). The sulphate **aerosols** that formed in the atmosphere led to dramatic cooling in the northern hemisphere. There were serious crop failures in Europe and North America, and a total of 6 million people are estimated to have died from famine and respiratory complications. In Iceland, poisoning from the HF resulted in the death of 80% of sheep, 50% of cattle, and the ensuing famine, along with HF poisoning, resulted in more than 10,000 human deaths—about 25% of the population.

Volcanic ash can also have serious implications for aircraft because it can destroy jet engines. For example, over 5 million airline passengers had their travel disrupted by the 2010 Eyjafjallajökull volcanic eruption in Iceland.

Pyroclastic Density Currents

In a typical explosive eruption at a composite volcano, the tephra and gases are ejected with explosive force and are hot enough to be forced high up into the atmosphere. As the eruption proceeds, and the amount of gas in the rising magma starts to decrease, parts will become heavier than air, and they can then flow downward along the flanks of the volcano (Figure 4.4.1). As they descend, they cool more and flow faster, reaching speeds up to several hundred kilometers per hour. A **pyroclastic density current** (PDC) consists of tephra ranging in size from boulders to microscopic shards of glass (made up of the edges and junctions of the bubbles of shattered pumice), plus gases (dominated by water vapor, but also including other gases). The temperature of this material can be as high as 1000°C. Among the most famous PDCs are the one that destroyed Pompeii in the year 79 CE, killing an estimated 18,000 people, and the one that destroyed the town of St. Pierre, Martinique, in 1902, killing an estimated 30,000.

The buoyant upper parts of pyroclastic density currents can flow over water, in some cases for several kilometers. The 1902 St. Pierre PDC flowed out into the city's harbour and destroyed several wooden ships anchored there.



Figure 4.4.1 The plinian eruption of Mount Mayon, Philippines. in 1984. Although most of the eruption column is ascending into the atmosphere, there are pyroclastic density currents flowing down the sides of the volcano in several places.

Pyroclastic Fall

Most of the tephra from an explosive eruption ascends high into the atmosphere, and some of it is distributed around Earth by high-altitude winds. The larger components (larger than 0.1 mm) tend to fall relatively close to the volcano, and the amount produced by large eruptions can cause serious damage and casualties. The large 1991 eruption of Mount Pinatubo in the Philippines resulted in the accumulation of tens of centimeters of ash in fields and on rooftops in the surrounding populated region. Heavy typhoon rains that hit the island at the same time added to the weight of the tephra, leading to the collapse of

thousands of roofs and to at least 300 of the 700 deaths attributed to the eruption.

Lahar

A **lahar** is any mudflow or **debris flow** that is related to a volcano. Most are caused by melting snow and ice during an eruption, as was the case with the lahar that destroyed the Colombian town of Armero in 1985 (described earlier). Lahars can also happen when there is no volcanic eruption, and one of the reasons is that, as we've seen, composite volcanoes tend to be weak and easily eroded.

In October 1998, category 5 hurricane Mitch slammed into the coast of central America. Damage was extensive and 19,000 people died, not so much because of high winds but because of intense rainfall—some regions received almost 2 m of rain over a few days! Mudflows and debris flows occurred in many areas, especially in Honduras and Nicaragua. An example is at the Casita Volcano in Nicaragua, where the heavy rains weakened rock and volcanic debris on the upper slopes, resulting in a debris flow that rapidly built in volume as it raced down the steep slope, and then ripped through the towns of El Porvenir and Rolando Rodriguez killing more than 2,000 people (Figure 4.4.2). El Porvenir and Rolando Rodriguez were new towns that had been built without planning approval in an area that was known to be at risk of lahars.



Figure 4.4.2 Part of the path of the lahar from Casita Volcano, October 30, 1998.

Sector Collapse and Debris Avalanche

In the context of volcanoes, **sector collapse** or flank collapse is the catastrophic failure of a significant part of an existing volcano, creating a large debris avalanche. This hazard was first recognized with the failure of the north side of Mount St. Helens immediately prior to the large eruption on May 18, 1980. In the weeks before the eruption a large bulge had formed on the side of the volcano, the result of magma transfer from depth into a satellite magma body within the mountain itself. Early on the morning of May 18, a moderate earthquake struck nearby; this is thought to have destabilized the bulge, leading to Earth's largest ever observed slope failure. The failure of this part of the volcano exposed the underlying satellite magma chamber, causing it to explode sideways, which then exposed the conduit leading to the magma chamber below. The resulting plinian eruption—with a 24 kilometre high eruption column—lasted for nine hours.

In August 2010, a massive part of the flank of B.C.'s Mount Meager gave way and about 48 million cubic meters (m^3) of rock rushed down the valley, one of the largest slope failures in Canada in historical times (Figure 4.4.3). More than 25 slope failures have taken place at Mount Meager in the past 8,000 years, some of them more than 10 times larger than the 2010 failure.



Figure 4.4.3 The August 2010 Mount Meager rock avalanche, showing where the slide originated (arrow, 4 km upstream), its path down a steep narrow valley, and the debris field (and the stream that eventually cut through it) in the foreground.

Lava Flows

As we saw in Exercise 4.4, lava flows at volcanoes like Kilauea do not advance very quickly, and in most cases, people can get out of the way. Of course, it is more difficult to move infrastructure, and so buildings and roads are typically the main casualties of lava flows.

Exercise 4.5 Volcanic hazards in Squamish

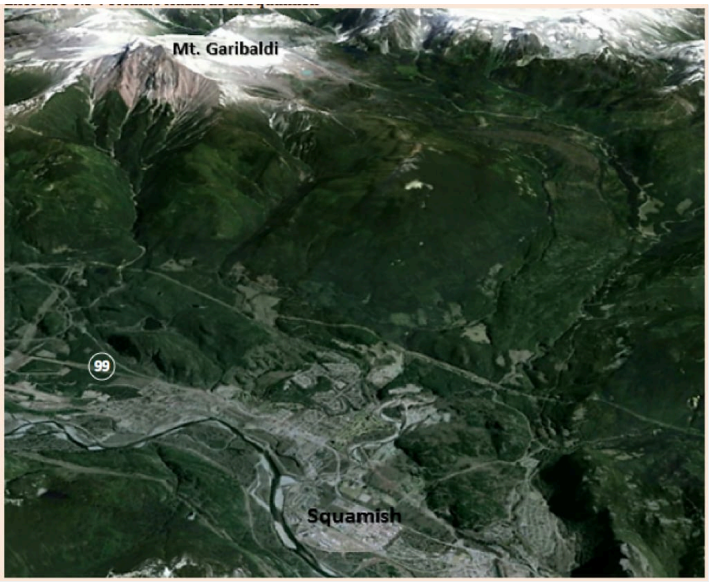


Figure 4.4.4

The town of Squamish is situated approximately 10 km from Mount Garibaldi, as shown in the photo. If Mount Garibaldi were to erupt, which of the following hazards could be an issue for people in and around Squamish? Explain why or why not.

1. Tephra emission.
2. Gas emission.
3. Pyroclastic denisty current.

4. Pyroclastic fall.
5. Lahar.
6. Sector collapse.
7. Lava flow.

See Appendix 3 for Exercise 4.5 answers.

Media Attributions

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4.5: Monitoring Volcanoes and Predicting Eruptions

In 2005 USGS geologist Chris Newhall made a list of the six most important signs of an imminent volcanic eruption. They are as follows:

1. *Gas leaks* — the release of gases (mostly H_2O, CO_2, and SO_2) from the magma into the atmosphere through cracks in the overlying rock
2. *Bit of a bulge* — the deformation of part of the volcano, indicating that a magma chamber at depth is swelling or becoming more pressurized
3. *Getting shaky* — many (hundreds to thousands) of small earthquakes, indicating that magma is on the move. The quakes may be the result of the magma forcing the surrounding rocks to crack, or a harmonic vibration that is evidence of magmatic fluids moving underground.
4. *Dropping fast* — a sudden decrease in the rate of seismicity, which may indicate that magma has stalled, which could mean that something is about to give way
5. *Big bump* — a pronounced bulge on the side of the volcano (like the one at Mount St. Helens in 1980), which may indicate that magma has moved close to surface
6. *Blowing off steam* — steam eruptions (a.k.a. **phreatic eruptions**) that happen when magma near the surface heats groundwater to the boiling point. The water eventually explodes, sending fragments of the overlying rock far into the air.

With these signs in mind, we can make a list of the equipment we should have and the actions we can take to monitor a volcano and predict when it might erupt.

Assessing Seismicity

The simplest and cheapest way to monitor a volcano is with seismometers. In an area with several volcanoes that have the potential to erupt (e.g., the Squamish-Pemberton area), a few well-placed seismometers can provide us with an early warning that something is changing beneath one of the volcanoes, and that we need to take a closer look. There are currently enough seismometers in the Lower Mainland and on Vancouver Island to provide this information.[1]

If there is seismic evidence that a volcano is coming to life, more seismometers should be placed in locations within a few tens of kilometers of the source of the activity (Figure 4.5.1). This will allow geologists to determine the exact location and depth of the seismic activity so that they can see where the magma is moving.



Figure 4.5.1 A seismometer and related power and communication equipment installed in 2007 in the vicinity of the Nazco Cone, British Columbia

Detecting Gases

Water vapor quickly turns into clouds of liquid water droplets and is relatively easy to detect just by looking, but CO_2 and SO_2 are not as obvious. It's important to be able to monitor changes in the composition of volcanic gases, and we need instruments to do that. Some can be monitored from a distance (from the ground or even from the air) using infrared devices, but to obtain more accurate data, we need to sample the air and do chemical analysis. This can be achieved with instruments placed on the ground close to the source of the gases (see Figure 4.5.12), or by collecting samples of the air and analyzing them in a lab.

Measuring Deformation

Measurement of deformation allows us to determine if a volcano is expanding or contracting. That can typically be related to the movement of magma into or out of a magma chamber near to surface and so is an indicator of the potential for an eruption. There are two main ways to measure ground deformation. One is known as a **tiltmeter**, which is a sensitive three-directional level that can sense small changes in the tilt of the ground at a specific location. Another is through the use of GPS (global positioning system) technology (Figure 4.5.2). Both are effective, but GPS provides more information than a tiltmeter because it can tell us how far the ground has actually moved in the east-west, north-south, and up-down directions.



Figure 4.5.2 A GPS unit installed at Hualalai volcano, Hawaii. The dish-shaped antenna on the right is the GPS receiver. The antenna on the left is for communication with a base station.

By combining information from these types of sources, along with careful observations made on the ground and from the air, and a thorough knowledge of how volcanoes work, geologists can get a good idea of the potential for a volcano to erupt in the near future (months to weeks, but not days). They can then make recommendations to authorities about the need for evacuations and restricting transportation corridors. Our ability to predict volcanic eruptions has increased dramatically in recent decades because of advances in our understanding of how volcanoes behave and in monitoring technology. Providing that careful work is done, there is no longer a large risk of surprise eruptions, and providing that public warnings are issued and heeded, it is less and less likely that thousands will die from sector collapse, pyroclastic flows, ash falls, or lahars. Indirect hazards are still very real, however, and we can expect the next eruption like the one at Laki in 1783 to take an even greater toll than it did then, especially since there are now roughly eight times as many people on Earth.

Exercise 4.6 Volcano alert!

You're the chief volcanologist for the Geological Survey of Canada (GSC), based in Vancouver. At 10:30 a.m. on a Tuesday, you receive a report from a seismologist at the GSC in Sidney saying that there has been a sudden increase in the number of small earthquakes in the vicinity of Mount Garibaldi. You have two technicians available, access to some monitoring

equipment, and a four-wheel-drive vehicle. At noon, you meet with your technicians and a couple of other geologists. By the end of the day, you need to have a plan to implement, starting tomorrow morning, and a statement to release to the press. What should your first day's fieldwork include? What should you say later today in your press release?

See Appendix 3 for Exercise 4.6 Answers.

Media Attributions

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- Figure 4.5.2: A Typical GPS Monitoring Station by USGS. Public domain.

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1. See: http://www.earthquakescanada.nrcan.gc.ca/stndon/CNSN-RNSC/stnbook-cahierstn/index-eng.php?tpl_sorting=map&CHIS_SZ=west↵

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4.6: Volcanoes in British Columbia

As shown on the Figure 4.6.1, three types of volcanic environments are represented in British Columbia:

- The Cascade Arc (a.k.a. the Garibaldi Volcanic Belt in Canada) is related to subduction of the Juan de Fuca Plate beneath the North America plate. This extends along the south coast from the U.S. border to the northern end of Vancouver Island.
- The Anahim Volcanic Belt is assumed to be related to a mantle plume. It is in central BC and stretches inland from the coast.
- The Stikine Volcanic Belt (northwestern BC) and the Wells Gray-Clearwater Volcanic Field (central Alberta border area) are assumed to be related to crustal rifting.

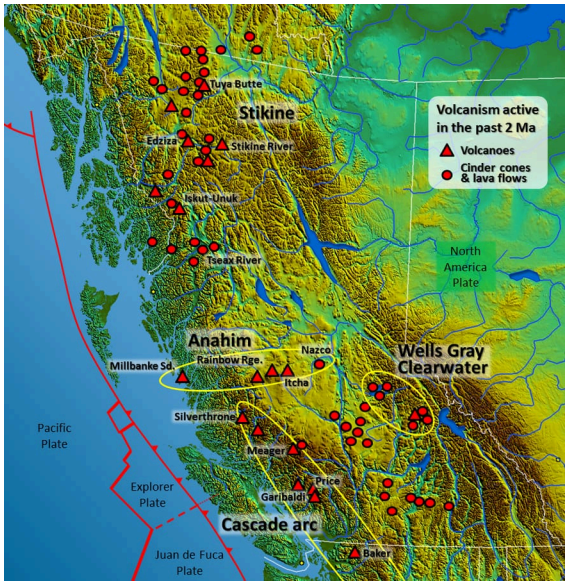


Figure 4.6.1 Major volcanic centres in British Columbia active within the last 2 million years

Subduction Volcanism

Southwestern British Columbia is at the northern end of the Juan de Fuca (Cascadia) subduction zone, and the volcanism there is related to magma generation by flux melting in the upper mantle above the subducting plate. In general, there has been a much lower rate and volume of volcanism in the B.C. part of this belt than in the U.S. part. One possible reason for this is that the northern part of the Juan de Fuca Plate (i.e., the Explorer Plate) is either not subducting, or is subducting at a slower rate than the rest of the plate. There are several volcanic centres in the Garibaldi Volcanic Belt: the Garibaldi centre (including Mount Garibaldi and the Black Tusk-Mount Price area adjacent to Garibaldi Lake (Figures 4.0.1 and 4.0.2), Mount Cayley, and Mount. Meager (Figure 4.6.3). The most recent volcanic activity in this area was at Mount Meager. Approximately 2,400 years ago, an explosive eruption of about the same magnitude as the 1980 Mount St. Helens eruption took place at Mount Meager. Ash spread as far east as Alberta. There was also significant eruptive activity at Mounts Price and Garibaldi approximately 12,000 and 10,000 years ago during the last glaciation; in both cases, lava and tephra built up against glacial ice in the adjacent valley (Figure 4.6.2). The Table in Figure 4.6.2 at the beginning of this chapter is a **tuya**, a volcano that formed beneath glacial ice and had its top eroded by the lake that formed around it in the ice.

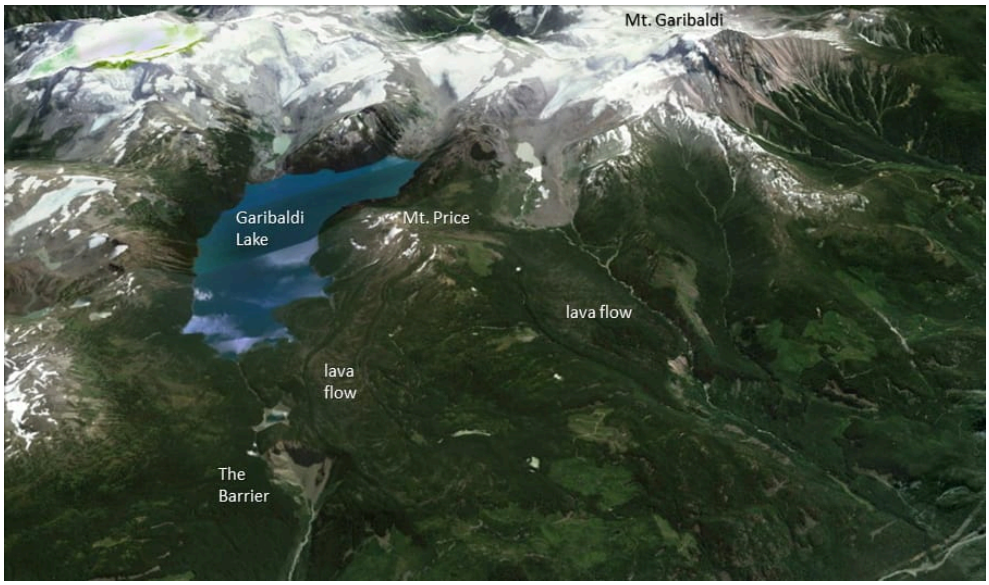


Figure 4.6.2 Perspective view of the Garibaldi region (looking east) showing the outlines of two lava flows from Mount Price. Volcanism in this area last took place when the valley in the foreground was filled with glacial ice. The cliff known as the Barrier formed when part of the Mount Price lava flow failed after deglaciation. The steep western face of Mount Garibaldi formed by sector collapse, also because the rocks were no longer supported by glacial ice.

Mantle Plume Volcanism

The chain of volcanic complexes and cones extending from Milbanke Sound to Nazko Cone is interpreted as being related to a mantle plume currently situated close to the Nazko Cone, just west of Quesnel. The North America Plate is moving in a westerly direction at about 2 cm per year with respect to this plume, and the series of now partly eroded shield volcanoes between Nazco and the coast is interpreted to have been formed by the plume as the continent moved over it.

The Rainbow Range, which formed at approximately 8 Ma, is the largest of these older volcanoes. It has a diameter of about 30 km and an elevation of 2,495 m (Figure 4.6.3). The name “Rainbow” refers to the bright colors displayed by some of the volcanic rocks as they weather.



Figure 4.6.3 Rainbow Range, Chilcotin Plateau, B.C.

Rift-Related Volcanism

While B.C. is not about to split into pieces, two areas of volcanism are related to rifting—or at least to stretching-related fractures that might extend through the crust. These are the Wells Gray-Clearwater volcanic field southeast of Quesnel, and the Northern Cordillera Volcanic Field, which ranges across the northwestern corner of the province (as already discussed in section 4.1). This area includes Canada's most recent volcanic eruption, a cinder cone and mafic lava flow that formed around 250 years ago at the Tseax River Cone in the Nass River area north of Terrace. According to Nisga'a oral history, as many as 2,000 people died during that eruption, in which lava overran their village on the Nass River. Most of the deaths are attributed to asphyxiation from volcanic gases, probably carbon dioxide.

The Mount Edziza Volcanic Field near the Stikine River is a large area of lava flows, sulfurous ridges, and cinder cones. The most recent eruption in this area was about 1,000 years ago. While most of the other volcanism in the Edziza region is mafic and involves lava flows and cinder cones, Mount Edziza itself (Figure 4.6.4) is a composite volcano with rock compositions ranging from rhyolite to basalt. A possible explanation for the presence of composite volcanism in an area dominated by mafic flows and cinder cones is that there is a magma chamber beneath this area, within which magma differentiation is taking place.



Figure 4.6.4 Mount Edziza, in the Stikine area, B.C., with Eve Cone in the foreground.

Exercise 4.7 Volcanoes down under

This map shows the plate tectonic situation in the area around New Zealand.

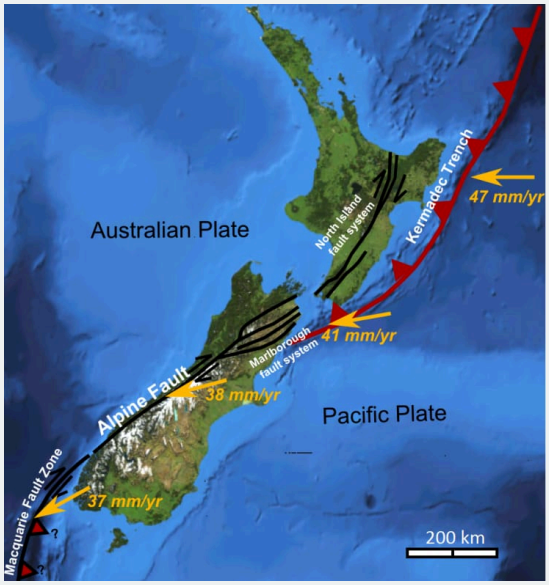


Figure 4.6.5

1. Based on what you know about volcanoes in B.C., predict where you might expect to see volcanoes in and around New Zealand.
2. What type of volcanoes would you expect to find in and around New Zealand?

See Appendix 3 for Exercise 4.7 Answers.

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4.7: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
4.1 Plate Tectonics and Volcanism	Volcanism is closely related to plate tectonics. Most volcanoes are associated with convergent plate boundaries (at subduction zones), and there is also a great deal of volcanic activity at divergent boundaries and areas of continental rifting. At convergent boundaries magma is formed where water from a subducting plate acts as a flux to lower the melting temperature of the adjacent mantle rock. At divergent boundaries magma forms because of decompression melting. Decompression melting also takes place within a mantle plume.
4.2 Magma Composition and Eruption Style	The initial magmas in most volcanic regions are mafic in composition, but they can evolve into more felsic types through interaction with crustal rock, and as a result of crystal settling within a magma chamber. Felsic magmas tend to have higher gas contents than mafic magmas, and they are also more viscous. The higher viscosity prevents gases from escaping from the magma, and so felsic magmas are more pressurized and more likely to erupt explosively.
4.3 Types of Volcanoes	Cinder cones, which can form in various volcanic settings, are relatively small volcanoes that were formed during a single eruptive event and are composed mostly of mafic rock fragments. Composite volcanoes are normally associated with subduction, and while their magma tends to be intermediate on average, it can range all the way from felsic to mafic. The corresponding differences in magma viscosity lead to significant differences in eruptions style. Most shield volcanoes are associated with mantle plumes, and have consistently mafic magma which typically erupts as lava flows.
4.4 Volcanic Hazards	Most direct volcanic hazards are related to volcanoes that erupt explosively, especially composite volcanoes. Pyroclastic density currents, some as hot as 1000°C can move at hundreds of kilometers per hour and will kill anything in the way. Lahars, volcano-related mudflows, can be large enough to destroy entire towns. Lava flows will destroy anything in their paths, but tend to move slowly enough so that people can get to safety. But the indirect effects of volcanism have been more deadly in the past, mostly because volcanic ash and gases can lead to short-term significant climate cooling.
4.5 Monitoring Volcanoes and Predicting Eruptions	We have the understanding and technology to predict volcanic eruptions with some success, and to ensure that people are not harmed. The prediction techniques include monitoring seismicity in volcanic regions, detecting volcanic gases, and measuring deformation of the flanks of a volcano.
4.6 Volcanoes in British Columbia	There are examples of all of the important types of volcanoes in British Columbia, including subduction volcanism north of Vancouver, mantle-plume volcanism along the Nazco trend, and rift-related volcanism in the Wells Gray and Stikine regions.

Questions for Review

Answers to Review Questions at the end of each chapter are provided in Appendix 2.

1. What are the three main tectonic settings for volcanism on Earth?
2. What is the primary mechanism for partial melting at a convergent plate boundary?
3. Why are the viscosity and gas content of a magma important in determining the type of volcanic rocks that will be formed when that magma is extruded?
4. Why do the gases in magma not form gas bubbles when the magma is deep within the crust?
5. Where do pillow lavas form? Why do they form and from what type of magma?
6. What two kinds of rock textures are typically found in a composite volcano?
7. What is a lahar, and why are lahars commonly associated with eruptions of composite volcanoes?
8. Under what other circumstances might a lahar form?

9. Explain why shield volcanoes have such gentle slopes.
10. In very general terms, what is the lifespan difference between a composite volcano and a shield volcano?
11. Why is weak seismic activity (small earthquakes) typically associated with the early stages of a volcanic eruption?
12. How can GPS technology be used to help monitor a volcano in the lead-up to an eruption?
13. What type of eruption at Mount St. Helens might have produced columnar basalts?
14. What is the likely geological origin of the Nazko Cone?
15. What might be the explanation for southwestern B.C. having much less subduction-related volcanism than adjacent Washington and Oregon?
16. What was the likely cause of most of the deaths from the most recent eruption at the Tseax River Cone?

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5: WEATHERING AND SOIL

Weathering is what takes place when a body of rock is exposed to the forces and conditions that exist at Earth's surface. Weathering is a key part of the process of soil formation, which lies within the top few tens of centimeters of the Earth's surface and is important in sustaining plant growth.

5.1: Mechanical Weathering

5.2: Chemical Weathering

5.3: The Products of Weathering and Erosion

5.4: Weathering and the Formation of Soil

5.5: The Soils of Canada

5.6: Weathering and Climate Change

5.7: Summary

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5.1: Mechanical Weathering

Intrusive igneous rocks form at depths of several hundreds of meters to several tens of kilometers. Sediments are turned into sedimentary rocks only when they are buried by other sediments to depths in excess of several hundreds of meters. Most metamorphic rocks are formed at depths of kilometers to tens of kilometers. Weathering cannot even begin until these rocks are uplifted through various processes of mountain building—most of which are related to plate tectonics—and the overlying material has been eroded away and the rock is exposed as an **outcrop**.[1]

The most important agents of mechanical weathering are:

- The decrease in pressure that results from removal of overlying rock
- Freezing and thawing of water in cracks in the rock
- Formation of salt crystals within the rock
- Cracking from plant roots and removal of material by burrowing animals

When a mass of rock is exposed by weathering and removal of the overlying rock, there is a decrease in the confining pressure on the rock, and the rock expands. This unloading promotes cracking of the rock, known as **exfoliation**, as shown in the granitic rock in Figure 5.1.1., which, in places, is peeling off like the layers of an onion.



Figure 5.1.1 Exfoliation fractures in granitic rock exposed on the side of the Coquihalla Highway north of Hope, B.C.



Figure 5.1.2 Exfoliation of slate at a road cut in the Columbia Mountains west of Golden, B.C.

Granitic rock tends to exfoliate parallel to the exposed surface because the rock is typically homogenous, and it doesn't have predetermined planes along which it must fracture. Sedimentary and metamorphic rocks, on the other hand, tend to exfoliate along predetermined planes (Figure 5.1.2).

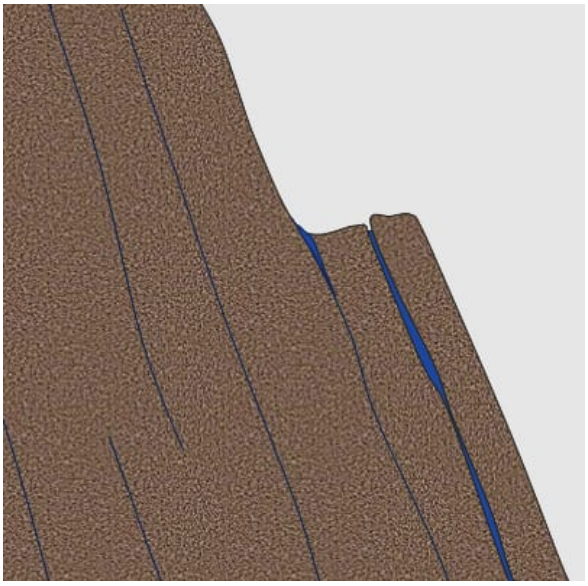


Figure 5.1.3 The process of frost wedging on a steep slope. Water gets into fractures and then freezes, expanding the fracture a little. When the water thaws it seeps a little farther into the expanded crack. The process is repeated many times, and eventually a piece of rock will be wedged away.

Frost wedging is the process by which water seeps into cracks in a rock, expands on freezing, and thus enlarges the cracks (Figure 5.1.3). The effectiveness of frost wedging is related to the frequency of freezing and thawing. Frost wedging is most effective in a climate like Canada's. In warm areas where freezing is infrequent, in very cold areas where thawing is infrequent, or in very dry areas, where there is little water to seep into cracks, the role of frost wedging is limited. If you are ever hiking in the mountains you might hear the effects of frost wedging when the Sun warms a steep rocky slope and the fragments of rock that were pried away from the surface by freezing the night before are released as that ice melts.

In many parts of Canada, the transition between freezing nighttime temperatures and thawing daytime temperatures is frequent — tens to hundreds of times a year. Even in warm coastal areas of southern B.C., freezing and thawing transitions are common at higher elevations. A common feature in areas of effective frost wedging is a **talus slope**—a fan-shaped deposit of fragments removed by frost wedging from the steep rocky slopes above (Figure 5.1.4).



Figure 5.1.4 An area with very effective frost-wedging near Keremeos, B.C. The fragments that have been wedged away from the cliffs above have accumulated in a talus deposit at the base of the slope. The rocks in this area have quite varied colors, and those are reflected in the colors of the talus.

A related process, frost heaving, takes place within unconsolidated materials on gentle slopes. In this case, water in the soil freezes and expands, pushing the overlying material up. Frost heaving is responsible for winter damage to roads all over North America.

When salt water seeps into rocks and then evaporates on a hot sunny day, salt crystals grow within cracks and pores in the rock. The growth of these crystals exerts pressure on the rock and can push grains apart, causing the rock to weaken and break. There are many examples of this on the rocky shorelines of Vancouver Island and the Gulf Islands, where sandstone outcrops are common and salty seawater is readily available (Figure 5.1.5). Salt weathering can also occur away from the coast, because most environments have some salt in them.



Figure 5.1.5 Honeycomb weathering of sandstone on Gabriola Island, B.C. The holes are caused by crystallization of salt within rock pores, and the seemingly regular pattern is related to the original roughness of the surface. It's a positive-feedback process because the holes collect salt water at high tide, and so the effect is accentuated around existing holes. This type of weathering is most pronounced on south-facing sunny exposures

The effects of plants and animals are significant in mechanical weathering. Roots can force their way into even the tiniest cracks, and then they exert tremendous pressure on the rocks as they grow, widening the cracks and breaking the rock (Figure 5.1.6).

Although animals do not normally burrow through solid rock, they can excavate and remove huge volumes of soil, and thus expose the rock to weathering by other mechanisms.



Figure 5.1.6 Conifers growing on granitic rocks at The Lions, near Vancouver, B.C.

Mechanical weathering is greatly facilitated by erosion, which is the removal of weathering products, allowing for the exposure of more rock for weathering. A good example of this is shown in Figure 5.1.4. On the steep rock faces at the top of the cliff, rock fragments have been broken off by ice wedging, and then removed by gravity. This is a form of mass wasting, which is discussed in more detail in Chapter 15. Other important agents of erosion that also have the effect of removing the products of weathering include water in streams (Chapter 13), glacial ice (Chapter 16), and waves on the coasts (Chapter 17).

Exercise 5.1 Mechanical weathering

This photo shows granitic rock at the top of Stawamus Chief near Squamish, B.C. Identify the mechanical weathering processes that you can see taking place, or you think probably take place at this location.

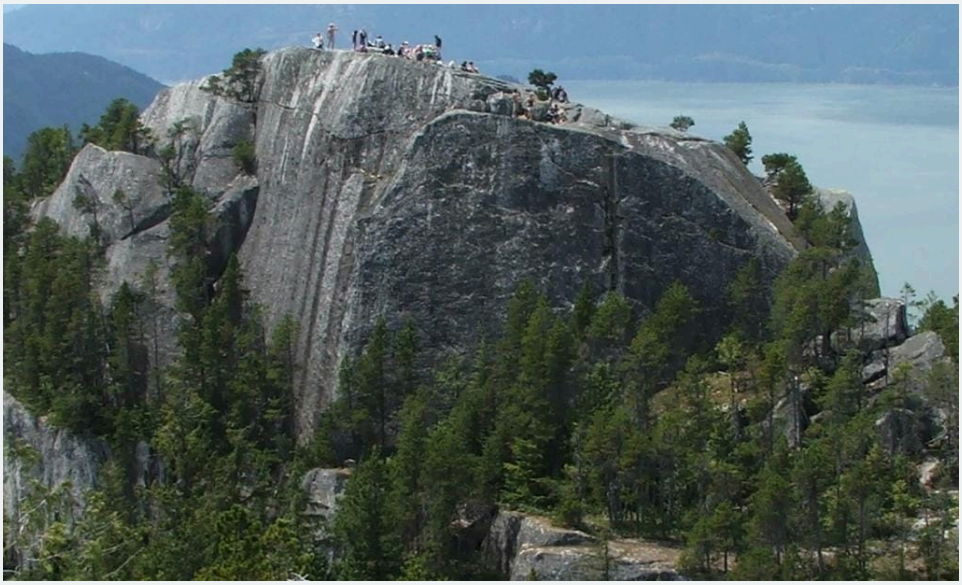


Figure 5.1.7

See Appendix 3 for Exercise 5.1 answers.

Media Attributions

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1. To a geologist, an outcrop is an exposure of bedrock, the solid rock of the crust. ↵

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5.2: Chemical Weathering

Chemical weathering results from chemical changes to minerals that become unstable when they are exposed to surface conditions. The kinds of changes that take place are highly specific to the mineral and the environmental conditions. Some minerals, like quartz, are virtually unaffected by chemical weathering, while others, like feldspar, are easily altered. In general, the degree of chemical weathering is greatest in warm and wet climates, and least in cold and dry climates. The important characteristics of surface conditions that lead to chemical weathering are the presence of water (in the air and on the ground surface), the abundance of oxygen, and the presence of carbon dioxide, which produces weak carbonic acid when combined with water. That process, which is fundamental to most chemical weathering, can be shown as follows:

$$H_2O + CO_2 \leftrightarrow H_2CO_3 \text{ then } H_2CO_3 \leftrightarrow H^+ + HCO_3^-$$

water + carbon dioxide ↔ carbonic acid then carbonic acid ↔ dissolved hydrogen ions + dissolved bicarbonate ions

Yikes! Chemical formulas

> Lots of people seize up when they are asked to read chemical or mathematical formulas. It's OK, you don't necessarily have to! If you don't like the formulas just read the text underneath them. In time you may get used to reading the formulas.
>
> The double-ended arrow "↔ " indicates that the reaction can go either way, but for our purposes these reactions are going towards the right.

Here we have water (e.g., as rain) plus carbon dioxide in the atmosphere, combining to create carbonic acid. Then carbonic acid dissociates (comes apart) to form hydrogen and bicarbonate ions. The amount of CO_2 in the air is enough to make weak carbonic acid. There is typically much more CO_2 in the soil, so water that percolates through the soil can become more acidic. In either case, this acidic water is a critical to chemical weathering.

In some types of chemical weathering the original mineral becomes altered to a different mineral. For example, feldspar is altered—by **hydrolysis**—to form **clay minerals** plus some ions in solution. In other cases the minerals dissolve completely, and their components go into solution. For example, calcite ($CaCO_3$) is soluble in acidic solutions.

The hydrolysis of feldspar can be written like this:

$$CaAl_2Si_2O_8 + H_2CO_3 + ½O_2 \leftrightarrow Al_2Si_2O_5(OH)_4 + Ca^{2+} + CO_3^{2-}$$

plagioclase feldspar + carbonic acid ↔ kaolinite + dissolved calcium ions + dissolved carbonate ions

This reaction shows calcium-bearing plagioclase feldspar, but similar reactions could also be written for sodium or potassium feldspars. In this case, we end up with the mineral kaolinite, along with calcium and carbonate ions in solution. Those ions can eventually combine (probably in the ocean) to form the mineral calcite. The hydrolysis of feldspar to clay is illustrated in Figure 5.2.1, which shows two images of the same granitic rock, a recently broken fresh surface on the left and a clay-altered weathered surface on the right. Other silicate minerals can also go through hydrolysis, although the end results will be a little different. For example, pyroxene can be converted to the clay minerals chlorite or smectite, and olivine can be converted to the clay mineral serpentine.

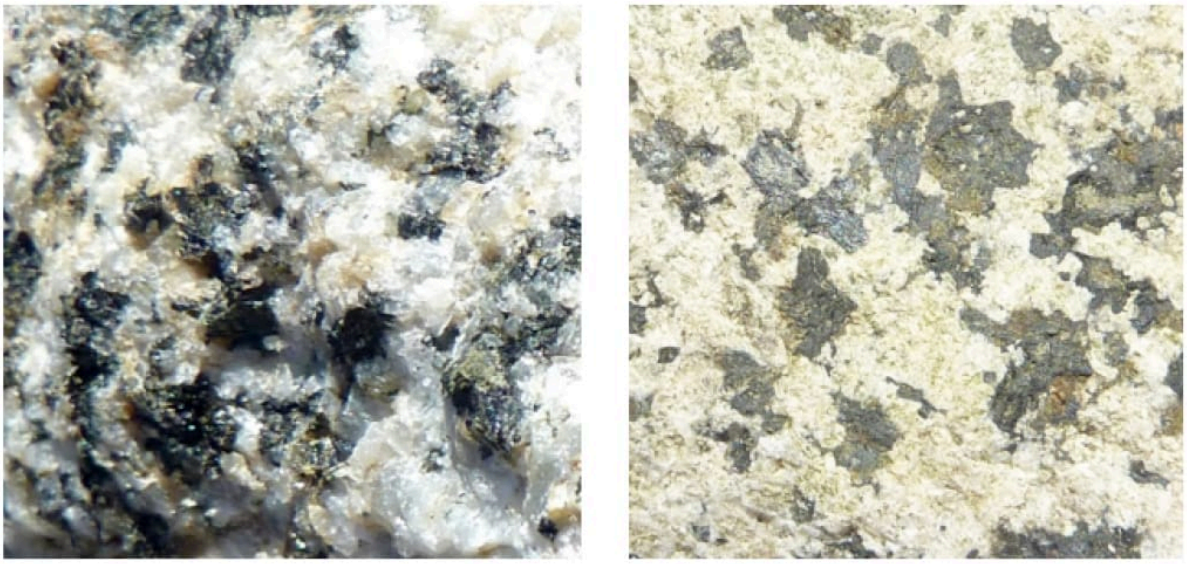


Figure 5.2.1 Unweathered (left) and weathered (right) surfaces of the same piece of granitic rock. On the unweathered surfaces the feldspars are still fresh and glassy-looking. On the weathered surface much of the feldspar has been altered to the chalky-looking clay mineral kaolinite.

Oxidation is another very important chemical weathering process. The oxidation of the iron in a ferromagnesian silicate starts with the dissolution of the iron. For olivine, the process looks like this, where olivine in the presence of carbonic acid is converted to dissolved iron, carbonate, and silicic acid:

$$Fe_2SiO_4 + 4H_2CO_3 \leftrightarrow 2Fe^{2+} + 4HCO_3^- + H_4SiO_4$$

olivine + (carbonic acid) ↔ dissolved iron ions + dissolved carbonate ions + dissolved silicic acid

But in the presence of oxygen and carbonic acid, the dissolved iron is then quickly converted to the mineral hematite:

$$2Fe^{2+} + 4HCO_3^- + ½\ O_2 + 2H_2O \leftrightarrow Fe_2O_3 + 4H_2CO_3$$

dissolved iron ions + dissolved bicarbonate ions + oxygen + water ↔ hematite + carbonic acid

The equation shown here is for olivine, but it could apply to almost any other ferromagnesian silicate, including pyroxene, amphibole, or biotite. Iron in the sulfide minerals (e.g., pyrite) can also be oxidized in this way. And the mineral hematite is not the only possible end result, as there is a wide range of iron oxide minerals that can form in this way. The results of this process are illustrated in Figure 5.2.2, which shows a granitic rock in which some of the biotite and amphibole have been altered to form the iron oxide mineral limonite.



Figure 5.2.2 A granitic rock containing biotite and amphibole which have been altered near to the rock's surface to limonite, which is a mixture of iron oxide minerals.

A special type of oxidation takes place in areas where the rocks have elevated levels of sulfide minerals, especially pyrite (FeS_2). Pyrite reacts with water and oxygen to form sulfuric acid, as follows:

$$2FeS_2 + 7O_2 + 2H_2O \leftrightarrow 2Fe^{2+} H_2SO_4 + 2H^+$$

pyrite + oxygen + water ↔ dissloved iron ions + sulfuric acid + dissolved hydrogen ions

The runoff from areas where this process is taking place is known as **acid rock drainage** (ARD), and even a rock with 1% or 2% pyrite can produce significant ARD. Some of the worst examples of ARD are at metal mine sites, especially where pyrite-bearing rock and waste material have been mined from deep underground and then piled up and left exposed to water and oxygen. One example of that is the Mt. Washington Mine near Courtenay on Vancouver Island (Figure 5.2.3), but there are many similar sites across Canada and around the world.



Figure 5.2.3 Exposed oxidizing and acid generating rocks and mine waste at the abandoned Mt. Washington Mine, B.C. (left), and an example of acid drainage downstream from the mine site (right).

At many ARD sites, the pH of the runoff water is less than 4 (very acidic). Under these conditions, metals such as copper, zinc, and lead are quite soluble, and this can lead to toxicity for aquatic and other organisms. For many years, the river downstream from the Mt. Washington Mine had so much dissolved copper in it that it was toxic to salmon. Remediation work has since been carried out at the mine and the situation has improved.

The hydrolysis of feldspar and other silicate minerals and the oxidation of iron in ferromagnesian silicates all serve to create rocks that are softer and weaker than they were to begin with, and thus more susceptible to mechanical weathering.

The weathering reactions that we've discussed so far involved the transformation of one mineral to another mineral (e.g., feldspar to clay), and the release of some ions in solution (e.g., Ca^{2+} or Fe^{2+}). Some weathering processes involve the complete dissolution of a mineral. Calcite, for example, will dissolve in weak acid, to produce calcium and bicarbonate ions. The equation is as follows:

$$\mathbf{CaCO_3 + H^+ + HCO_3^- \leftrightarrow Ca^{2+} + 2HCO_3^-}$$

calcite + dissolved hydrogen ions + dissolved bicarbonate ions ↔ dissolved calcium ions + dissolved bicarbonate ions

Calcite is the major component of limestone (typically more than 95%), and under surface conditions, limestone can dissolve completely, as shown in Figure 5.2.4. Limestone also dissolves at relatively shallow depths underground, forming limestone caves. This is discussed in more detail in Chapter 14, where we look at **groundwater**.



Figure 5.2.4 A limestone outcrop on Quadra Island, B.C. The limestone, which is primarily made up of the mineral calcite, has been dissolved to different degrees in different areas because of compositional differences. The buff-colored bands are chert, which stands out because it is not soluble.

Exercise 5.2 Chemical Weathering

The main processes of chemical weathering are hydrolysis, oxidation, and dissolution. Indicate which process is primarily involved during each of the following chemical weathering changes:

1. Pyrite to hematite
2. Calcite to calcium and bicarbonate ions
3. Feldspar to clay
4. Olivine to serpentine

See Appendix 3 for Exercise 5.2 answers.

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5.3: The Products of Weathering and Erosion

The products of weathering and erosion are the unconsolidated materials that we find around us on slopes, beneath, beside and on top of glaciers, in stream valleys, on beaches, and in deserts. The nature of these materials—their composition, size, degree of sorting, and degree of rounding—is determined by the type of rock that is being weathered, the nature of the weathering, the erosion and transportation processes, and the climate.

In addition to these solid sediments, the other important products of weathering are several different types of ions in solution.

A summary of the weathering products of some of the common minerals present in rocks is provided in Table 5.1. In addition to the weathering products listed in the table, most of the larger fragments—larger than sand grains—that make up sediments will be pieces of rock as opposed to individual minerals.

Table 5.1 A list of the typical weathering products of some of the minerals in common rocks

Common Mineral	Typical Weathering Products
Quartz	Quartz as sand grains
Feldspar	Clay minerals plus potassium, sodium, and calcium in solution
Biotite and amphibole	Chlorite plus iron and magnesium in solution
Pyroxene and olivine	Serpentine plus iron and magnesium in solution
Calcite	Calcium and carbonate in solution
Pyrite	Iron oxide minerals plus iron in solution and sulfuric acid

Some examples of the products of weathering are shown in Figure 5.3.1. They range widely in size and shape depending on the processes involved in their transportation. If and when deposits like these are turned into sedimentary rocks, the textures of those rocks will vary significantly. Importantly, when we describe sedimentary rocks that formed millions of years in the past, we can use those properties to make inferences about the conditions that existed during their formation.



Boulders in a talus deposit at Keremeos. All are angular fragments from the same rock source.

Pebbles on a beach in Victoria. All are rounded fragments of rock from different sources.

Sand from a beach at Gabriola. Most are angular quartz grains, some are fragments of rock.

Sand from a dune in Utah. All are rounded quartz grains.

Figure 5.3.1 Products of weathering and erosion formed under different conditions.

We'll talk more about the nature and interpretation of sediments and sedimentary rocks in Chapter 6, but it's worth considering here why the sand-sized sediments shown in Figure 5.3.1 are so strongly dominated by the mineral quartz, even though quartz makes up less than 20% of Earth's crust. The explanation is that quartz is highly resistant to the types of weathering that occur at Earth's surface. It is not affected by weak acids or the presence of oxygen. This makes it unique among the minerals that are

common in igneous rocks. Quartz is also very hard, and doesn't have cleavage, so it is resistant to mechanical erosion.

So when a rock like granite is subject to chemical weathering the feldspar and the ferromagnesian silicates get converted to clays and dissolved ions such as: Ca^{2+}, Na^{+}, K^{+}, Fe^{2+}, Mg^{2+}, and H_4SiO_4, but the quartz is resistant to those processes and remains intact. The clay gradually gets eroded away, then the rock breaks apart leaving lots of grains of quartz. In other words, quartz, clay minerals, and dissolved ions are the most common products of weathering. Quartz and some of the clay minerals tend to form sedimentary deposits on and at the edges of continents, while the rest of the clay minerals and the dissolved ions tend to be washed out into the oceans to form sediments on the sea floor.

Exercise 5.3 Describing the weathering origins of sand

In the left side of the following table, a number of different sands are pictured and described. Describe some of the important weathering processes that might have led to the development of these sands.

See Appendix 3 for Exercise 5.3 answers.

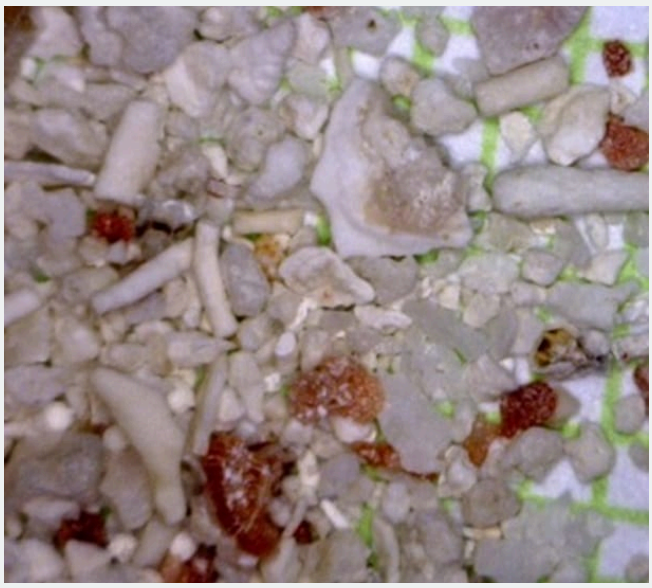
Image	Description and Location
	Fragments of coral, algae, and urchin from a shallow water area (roughly 2 meters deep) near a reef in Belize. The grain diameters are between 0.1 and 1 millimeters.
	Angular quartz and rock fragments from a glacial stream deposit near Osoyoos, B.C. The grain diameters are between 0.25 and 0.5 milimeters.

Image	Description and Location
	Rounded grains of olivine (green) and volcanic glass (black) from a beach on the big island of Hawaii. The grains are approximately 1 millimetre across.

Image Descriptions

Figure 5.3.1 image description: Examples of weathering and erosion.

1. Boulders in a talus deposit at Keremeos. All are angular fragments from the same rock source.
2. Pebbles on a beach in Victoria. All are rounded fragments of rock from different sources.
3. Sand from a beach at Gabriola Island. most are angular quartz grains, some are sand-sized fragments of rock.
4. Sand from a dune in Utah. All are rounded quartz grains.

[Return to Figure 5.3.1]

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5.4: Weathering and the Formation of Soil

Weathering is a key part of the process of soil formation, and soil is critical to our existence on Earth. In other words, we owe our existence to weathering, and we need to take care of our soil!

Many people refer to any loose material on Earth's surface as soil, but to geologists (and geology students) soil is the material that includes organic matter, lies within the top few tens of centimeters of the surface, and is important in sustaining plant growth.

Soil is a complex mixture of minerals (approximately 45%), organic matter (approximately 5%), and empty space (approximately 50%, filled to varying degrees with air and water). The mineral content of soils is variable, but is dominated by clay minerals and quartz, along with minor amounts of feldspar and small fragments of rock. The types of weathering that take place within a region have a major influence on soil composition and texture. For example, in a warm climate, where chemical weathering dominates, soils tend to be richer in clay. Soil scientists describe soil texture in terms of the relative proportions of sand, silt, and clay, as shown in Figure 5.4.1. The sand and silt components in this diagram are dominated by quartz, with lesser amounts of feldspar and rock fragments, while the clay component is dominated by the clay minerals.

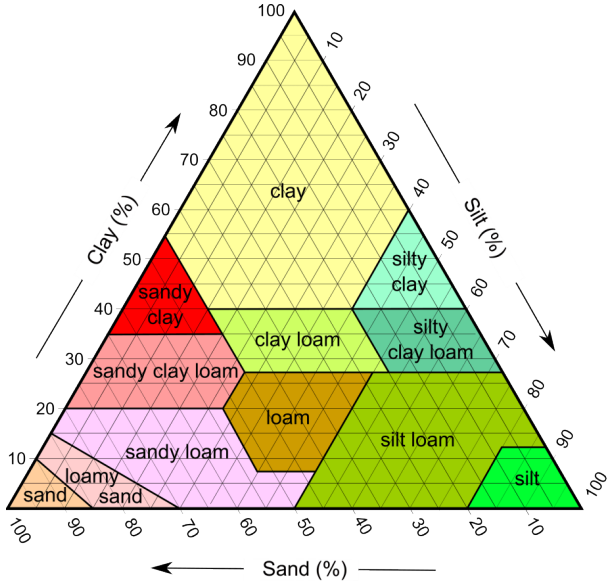


Figure 5.4.1 This diagram applies only to the mineral component of soils, and the names are textural descriptions, not soil classes.

Soil forms through accumulation and decay of organic matter and through the mechanical and chemical weathering processes described above. The factors that affect the nature of soil and the rate of its formation include climate (especially average temperature and precipitation amounts, and the consequent types and intensity of vegetation), the type of parent material, the slope of the surface, and the amount of time available.

Climate

Soils develop because of the weathering of materials on Earth's surface, including the mechanical breakup of rocks, and the chemical weathering of minerals. Soil development is facilitated by the downward percolation of water. Soil forms most readily under temperate to tropical conditions (not cold) and where precipitation amounts are moderate (not dry, but not too wet). Chemical weathering reactions (especially the formation of clay minerals) and biochemical reactions proceed fastest under warm conditions, and plant growth is enhanced in warm climates. Too much water (e.g., in rainforests) can lead to the leaching of important chemical nutrients and hence to acidic soils. In humid and poorly drained regions, swampy conditions may prevail, producing soil that is dominated by organic matter. Too little water (e.g., in deserts and semi-deserts), results in very limited downward chemical transportation and the accumulation of salts and carbonate minerals (e.g., calcite) from upward-moving water. Soils in dry regions also suffer from a lack of organic material (Figure 5.4.2).



Figure 5.4.2 Poorly developed soil on wind-blown silt (loess) in an arid part of northeastern Washington State. The thickness shown is about 1 m, and the "soil" is just the upper 2 or 3 cm.

Parent Material

Soil parent materials can include all different types of bedrock and any type of unconsolidated sediments, such as glacial deposits and stream deposits. Soils are described as **residual soils** if they develop on bedrock, and transported soils if they develop on transported material such as glacial sediments. Other sources may use the term "transported soil" to imply that the soil itself has been transported, but in this text "transported soil" is soil that is developed on transported materials, like the very thin soil shown in Figure 5.4.2. When referring to such soil, it is better to be specific and say "soil developed on unconsolidated material," because that distinguishes it from soil developed on bedrock.

Quartz-rich parent material, such as granite, sandstone, or loose sand, leads to the development of sandy soils. Quartz-poor material, such as shale or basalt, generates soils with little sand.

Parent materials provide important nutrients to residual soils. For example, a minor constituent of granitic rocks is the calcium-phosphate mineral apatite ($Ca_5(PO_4)_3(F,Cl,OH)$), which is a source of the important soil nutrient phosphorus. Basaltic parent material tends to generate very fertile soils because it also provides phosphorus, along with significant amounts of iron, magnesium, and calcium.

Some unconsolidated materials, such as river-flood deposits, make for especially good soils because they tend to be rich in clay minerals. Clay minerals have large surface areas with negative charges that are attractive to positively charged elements like calcium, magnesium, iron, and potassium—important nutrients for plant growth.

Slope

Soil can only develop where surface materials remain in place and are not frequently moved away by mass wasting. Soils cannot develop where the rate of soil formation is less than the rate of erosion, so steep slopes tend to have little or no soil.

Time

Even under ideal conditions, soil takes thousands of years to develop. Virtually all of southern Canada was still glaciated up until 14 ka, and most of the central and northern parts of B.C., the prairies, Ontario, and Quebec were still glaciated at 12 ka. Glaciers still dominated the central and northern parts of Canada until around 10 ka, and so, at that time, conditions were still not ideal for soil development even in the southern regions. Therefore, soils in Canada, and especially in central and northern Canada, are relatively young and not well developed.

The same applies to soils that are forming on newly created surfaces, such as recent deltas or sand bars, or in areas of mass wasting.

Soil Horizons

The process of soil formation generally involves the downward movement of clay, water, and dissolved ions, and a common result of that is the development of chemically and texturally different layers known as **soil horizons**. The typically developed soil horizons, as illustrated in Figure 5.4.3, are:

- O — the layer of organic matter
- A — the layer of partially decayed organic matter mixed with mineral material
- E— the eluviated (leached) layer from which some of the clay and iron have been removed to create a pale layer that may be sandier than the other layers
- B — the layer of accumulation of clay, iron, and other elements from the overlying soil
- C — the layer of incomplete weathering

Although rare in Canada, another type of layer that develops in hot arid regions is known as **caliche** (pronounced *ca-lee-chee*). It forms from the downward (or in some cases upward) movement of calcium ions, and the precipitation of calcite within the soil. When well developed, caliche cements the surrounding material together to form a layer that has the consistency of concrete.

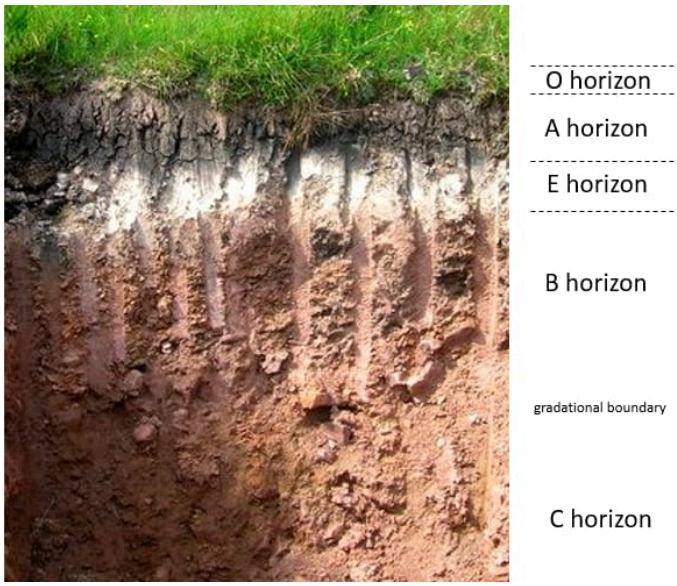


Figure 5.4.3 Soil horizons in a podsol from a site in northeastern Scotland.

Like all geological materials, soil is subject to erosion, although under natural conditions on gentle slopes, the rate of soil formation either balances or exceeds the rate of erosion. Human practices, especially those related to forestry and agriculture, have significantly upset this balance.

Soils are held in place by vegetation. When vegetation is removed, either through cutting trees or routinely harvesting crops and tilling the soil, that protection is either temporarily or permanently lost. The primary agents of the erosion of unprotected soil are water and wind.

Water erosion is accentuated on sloped surfaces because fast-flowing water obviously has greater eroding power than slow-flowing or still water (Figure 5.4.4). Raindrops can disaggregate exposed soil particles, putting the finer material (e.g., clays) into suspension in the water. **Sheetwash—**unchannelled flow across a surface—carries suspended material away, and channels erode right through the soil layer, removing both fine and coarse material.



Figure 5.4.4 Soil erosion by rain and channelled runoff on a field in Alberta.

Wind erosion is exacerbated by the removal of trees that act as wind breaks and by agricultural practices that leave bare soil exposed (Figure 5.4.5).

Tillage is also a factor in soil erosion, especially on slopes, because each time the soil is lifted by a cultivator, it is moved a few centimeters down the slope.



Figure 5.4.5 Soil erosion by wind in Alberta.

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5.5: The Soils of Canada

Up until the 1950s, the classification of soils in Canada was based on the system used in the United States. However, it was long recognized that the U.S .system did not apply well to many parts of Canada because of climate and environmental differences. The Canadian System of Soil Classification was first outlined in 1955 and has been refined and modified numerous times since then.

There are 10 orders of soil recognized in Canada. Each one is divided into groups, and then families, and then series, but we will only look at the orders, some of which are summarized in Table 5.2. The distribution of these types of soils (and a few others) in Canada is shown in Figure 5.5.1.

Table 5.2 The nature, origins and distributions of the more important soil orders in Canada

Order	Type	Brief Description	Environment
Forest soils	Podsol	Well-developed A and B horizons	Coniferous forests throughout Canada
	Luvisol	Clay rich B horizon	Northern prairies and central B.C., mostly on sedimentary rocks
	Brunisol	Poorly developed or immature soil, that does not have the well-defined horizons of podsol or luvisol	Boreal-forest soils in the discontinuous permafrost areas of central and western Canada, and also in southern B.C.
Grassland soils	Chernozem	High levels of organic matter and an A horizon at least 10 centimeters thick	Southern prairies (and parts of B.C.'s southern interior), in areas that experience water deficits during the summer
	Solonetzic	A clay-rich B horizon, commonly with a salt-bearing C horizon	Southern prairies, in areas that experience water deficits during the summer
Other important soils	Organic	Dominated by organic matter; mineral horizons are typically absent	Wetland areas, especially along the western edge of Hudson Bay, and in the area between the prairies and the boreal forest
	Cryosol	Poorly developed soil, mostly C horizon	Permafrost areas of northern Canada

There is an excellent website on Canadian soils, with videos describing the origins and characteristics of the soils, at: Soil Classification: Soil Orders of Canada.

As we've discussed, the processes of soil formation are dominated by the downward transportation of clays and certain elements dissolved in water, and the nature of those processes depends in large part on the climate. In Canada's predominantly cool and humid climate (which applies to most places other than the far north), **podsolization** is the norm. This involves downward transportation of hydrogen, iron, and aluminum (and other elements) from the upper part of the soil profile, and accumulation of clay, iron, and aluminum in the B horizon. Most of the **podsols, luvisols**, and **brunisols** of Canada form through various types of podsolization.

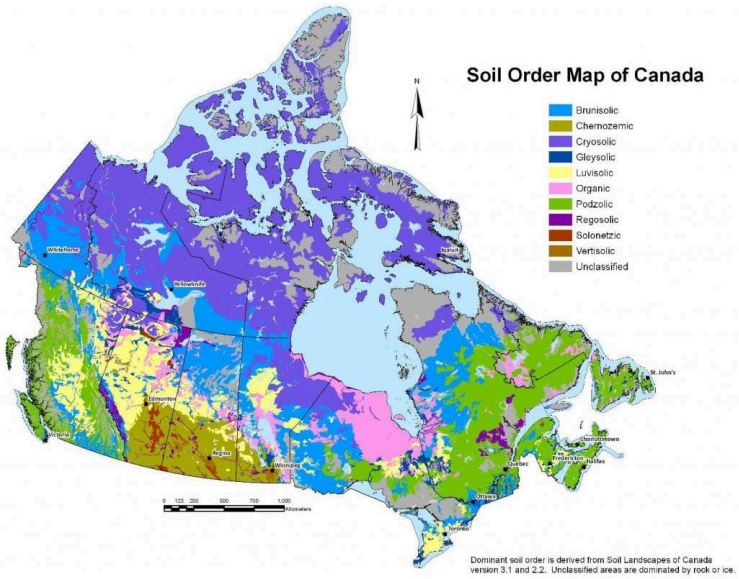


Figure 5.5.1 The soil order map of Canada.

In the grasslands of the dry southern parts of the prairie provinces and in some of the drier parts of southern B.C., dark brown organic-rich chernozem soils are dominant. In some parts of these areas, weak calcification takes place with leaching of calcium from the upper layers and accumulation of calcium in the B layer. Development of caliche layers is rare in Canada.

Organic soils form in areas with poor drainage (i.e., swamps) and a rich supply of organic matter. These soils have very little mineral matter.

In the permafrost regions of the north, where glacial retreat was most recent, the time available for soil formation has been short and the rate of soil formation is very slow. The soils are called cryosols (*cryo* means “ice cold”). Permafrost areas are also characterized by the churning of the soil by freeze-thaw processes, and as a result, development of soil horizons is very limited.

Exercise 5.4 The soils of Canada

Examine Figure 5.5.1, which shows the distribution of soils in Canada. Briefly describe the distributions of the five soils types listed. For each one, explain its distribution based on what you know about the conditions under which the soil forms and the variations in climate and vegetation related to it.

1. Chernozem
2. Luvisol
3. Podsol
4. Brunisol
5. Organic

See Appendix 3 for Exercise 5.4 answers.

Image Descriptions

Figure 5.5.1 image description: Soil order map of Canada. Chernozem is found in the southern parts of the prairies provinces (Alberta, Saskachewan, and Manitoba). Luvisol is found in the BC interior, most of central and northern Alberta, and a strip through central Saskatchewan and Manitoba. Podsol is common on the eastern and western coastal areas, as well as northern BC, and much of Quebec and the maritime provinces. Brunisol is common in the southern parts of the territories, northern Sakachewan and Manitoba, as well as parts of Ontario and Quebec. Organic soil is scattered all over the country, but most concentrated in northern Ontario and the lowlands around Hudson Bay. Cryosol is predominant in the far north. [Return to Figure 5.5.1]

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5.6: Weathering and Climate Change

Earth has two important carbon cycles. One is the biological one, wherein living organisms—mostly plants—consume carbon dioxide from the atmosphere to make their tissues. After they die almost all of that carbon is released back into the atmosphere when they decay over a period of years or decades. A small proportion of this biological-cycle carbon becomes buried in sedimentary rocks: during the slow formation of coal, as tiny fragments and molecules in organic-rich shale, and as the shells and other parts of marine organisms in limestone. This then becomes part of the geological carbon cycle, a cycle that actually involves a majority of Earth's carbon, but one that operates very slowly.

The geological carbon cycle is shown diagrammatically in Figure 5.6.1. The various steps in the process (not necessarily in this order) are as follows (the letters are shown on Figure 5.6.1):

a) Organic matter from plants is stored in lake sediments, peat, and permafrost for up to tens of thousands of years, and some may be buried deeper to form coal that can be stored for tens of millions of years.

b) Weathering of silicate minerals converts atmospheric carbon dioxide to dissolved bicarbonate, which is stored in the oceans for thousands to tens of thousands of years.

c) Dissolved carbon is converted by marine organisms to calcite, which is stored in carbonate rocks for hundreds of millions of years.

d) Organic and inorganic carbon compounds are stored in sediments for tens to hundreds of millions of years; some end up in petroleum deposits.

e) Carbon-bearing sediments are transferred to the mantle, where the carbon may be stored for tens of millions to billions of years.

f) During volcanic eruptions, carbon dioxide is released back to the atmosphere, where it is stored for years to decades.

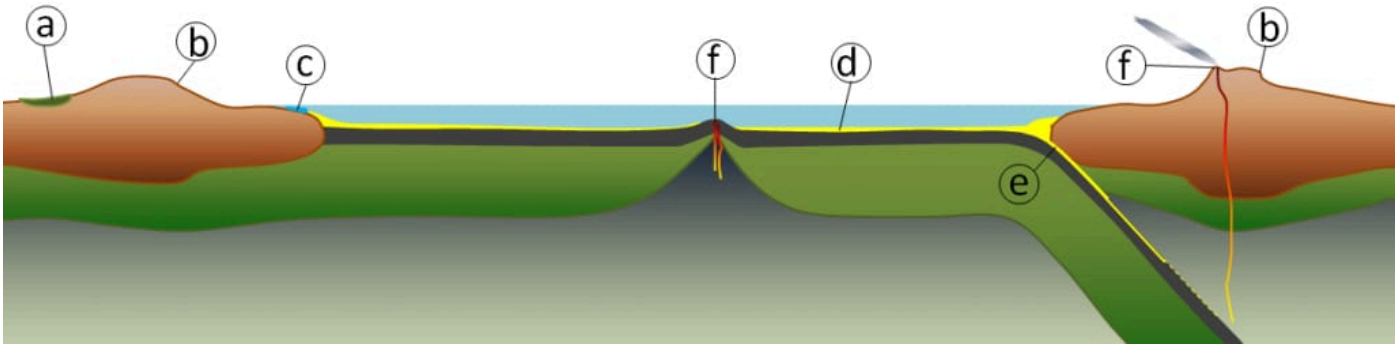


Figure 5.6.1 A representation of the geological carbon cycle. The processes represented by the letters are described in the text above.

During much of Earth's history, the geological carbon cycle has been balanced, with carbon being released by volcanism at approximately the same rate that it is stored by the other processes. Under these conditions, the climate remains relatively stable.

During some periods of Earth's history, that balance has been upset. This can happen during prolonged periods of greater than average volcanism. One example is the eruption of the Siberian Traps at around 250 Ma, which appears to have led to strong climate warming over a few million years because of the slow but steady input of extra volcanic CO_2 into the atmosphere.

A carbon imbalance is also associated with significant mountain-building events. For example, the Himalayan Range was formed between about 40 and 10 Ma and over that time period—and still today—the rate of weathering on Earth has been enhanced because those mountains are so high and steep and the range is so extensive. The weathering of these rocks—most importantly the hydrolysis of feldspar—has resulted in consumption of atmospheric carbon dioxide and transfer of the carbon to the oceans

and to ocean-floor carbonate minerals. The steady drop in carbon dioxide levels over the past 40 million years, which led to the Pleistocene glaciations, is partly attributable to the formation of the Himalayan Range.

A non-geological form of carbon-cycle imbalance is happening today on a very rapid time scale. We are in the process of extracting vast volumes of fossil fuels (coal, oil, and gas) that was stored in rocks over the past several hundred million years, and then converting these fuels to energy and carbon dioxide. By doing so, we are changing the climate faster than has ever happened in the past, and putting both ecosystems and our descendants at significant risk.

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5.7: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
5.1 Mechanical Weathering	Rocks weather when they are exposed to surface conditions, which in most case are quite different from those at which they formed. The main processes of mechanical weathering include exfoliation, freeze-thaw, salt crystallization, and the effects of plant growth.
5.2 Chemical Weathering	Chemical weathering takes place when minerals within rocks are not stable in their existing environment. Some of the important chemical weathering processes are hydrolysis of silicate minerals to form clay minerals, oxidation of iron in silicate and other minerals to form iron oxide minerals, and dissolution of calcite.
5.3 The Products of Weathering and Erosion	The main products of weathering and erosion are grains of quartz (because quartz is resistant to chemical weathering), clay minerals, iron oxide minerals, rock fragments, and a wide range of ions in solution.
5.4 Weathering and the Formation of Soil	Soil is a mixture of fine mineral fragments (including quartz and clay minerals), organic matter, and empty spaces that may be partially filled with water. Soil formation is controlled by climate (especially temperature and humidity), the nature of the parent material, the slope (because soil can't accumulate on steep slopes), and the amount of time available. Typical soils have layers called horizons which form because of differences in the conditions with depth.
5.5 The Soils of Canada	Canada has a range of soil types related to our unique conditions. The main types of soil form in forested and grassland regions, but there are extensive wetlands in Canada that produce organic soils, and large areas where soil development is poor because of cold conditions.
5.6 Weathering and Climate Change	The geological carbon cycle plays a critical role in balancing Earth's climate. Carbon is released to the atmosphere during volcanic eruptions. Carbon is extracted from the atmosphere during weathering of silicate minerals and this is eventually stored in the ocean and in sediments. Atmospheric carbon is also transferred to organic matter and some of that is later stored in soil, permafrost, and rocks. Our use of geologically stored carbon (fossil fuels) has upset this balance and that has created a climate crisis.

Questions for Review

Answers to Review Questions at the end of each chapter can be found in Appendix 2.

1. What has to happen to a body of rock before exfoliation can take place?
2. The climate of central B.C. is consistently cold in the winter and consistently warm in the summer. At what times of year would you expect frost wedging to be most effective?
3. What are the likely products of the hydrolysis of the feldspar albite ($NaAlSi_3O_8$)?
4. Oxidation weathering of the sulfide mineral pyrite (FeS_2) can lead to development of acid rock drainage (ARD). What are the environmental implications of ARD?
5. Most sand deposits are dominated by quartz, with very little feldspar. Under what weathering and erosion conditions would you expect to find feldspar-rich sand?
6. What ultimately happens to most of the clay that forms during the hydrolysis of silicate minerals?
7. Why are the slope and the parent materials important factors in soil formation?
8. Which soil constituents move downward to produce the B horizon of a soil?
9. What are the main processes that lead to the erosion of soils in Canada?
10. Where in Canada would you expect to find a chernozemic soil? What characteristics of this region produce this type of soil?
11. Where are luvisolic soils found in B.C.?

12. Why does weathering of silicate minerals, especially feldspar, lead to consumption of atmospheric carbon dioxide? What eventually happens to the carbon that is involved in that process?

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6: SEDIMENTS AND SEDIMENTARY ROCKS

Sedimentary rocks are formed by the deposition and subsequent cementation of that material at the Earth's surface and within bodies of water. Sedimentation is the collective name for processes that cause mineral and/or organic particles (detritus) to settle in place.

6.1: Clastic Sedimentary Rocks

6.2: Chemical Sedimentary Rocks

6.3: Depositional Environments and Sedimentary Basins

6.4: Sedimentary Structures and Fossils

6.5: Groups, Formations, and Members

6.6: Summary

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6.1: Clastic Sedimentary Rocks

A **clast** is a fragment of rock or mineral, ranging in size from less than a micron[1] (too small to see) to as big as an apartment block. Various types of clasts are shown in Figure 6.1.1 and in Exercise 5.3. The smaller ones tend to be composed of a single mineral crystal, and the larger ones are typically composed of pieces of rock. As we've seen in Chapter 5, most sand-sized clasts are made of quartz because quartz is more resistant to weathering than any other common mineral. Many of the clasts that are smaller than sand size (less than $^1/_{16}$th millimetre) are made of clay minerals. Most clasts larger than sand size (greater than 2 millimeters) are actual fragments of rock, and commonly these might be fine-grained rock like basalt or andesite, or if they are bigger, coarse-grained rock like granite or gneiss. Sedimentary rocks that are made up of "clasts" are called clastic sedimentary rocks. A comparable term is "detrital sedimentary rocks".

Grain-Size Classification

Geologists that study sediments and sedimentary rocks use the Udden-Wentworth grain-size scale for describing the sizes of the grains in these materials (Table 6.1).

Table 6.1 The Udden-Wentworth grain-size scale for classifying sediments and the grains that make up sedimentary rocks

[Skip Table]

Type	Description	Size range (millimeters)	Size range (microns)
Boulder	large	1024 and up	
medium	512 to 1024		
small	256 to 512		
Cobble	large	128 to 256	
small	64 to 128		
Pebble (Granule)	very coarse	32 to 64	
coarse	16 to 32		
medium	8 to 16		
fine	4 to 8		
very fine	2 to 4		
Sand	very coarse	1 to 2	1000 to 2000
coarse	0.5 to 1	500 to 1000	
medium	0.25 to 0.5 ($^1/_4$ to $^1/_2$ mm)	250 to 500	
fine	0.125 to 0.25 ($^1/_8$th to $^1/_4$ mm)	125 to 250	
very fine	0.063 to 0.125 (or $^1/_{16}$th to $^1/_8$th mm)	63 to 125	
Silt	very course		32 to 63
course		16 to 32	
medium		8 to 16	
fine		4 to 8	
very fine		2 to 4	
Clay	clay		0 to 2

There are six main grain-size categories; five are broken down into subcategories, with **clay** being the exception. The diameter limits for each successive subcategory are twice as large as the one beneath it. In general, a **boulder** is bigger than a toaster and difficult to lift. There is no upper limit to the size of boulder.[2] A small **cobble** will fit in one hand, a large one in two hands. A **pebble** is something that you could throw quite easily. The smaller ones—known as **granules**—are gravel size, but still you could throw one. You can't really throw a single grain of **sand**. Sand ranges from 2 millimeters down to 0.063 millimeters, and its key characteristic is that it feels "sandy" or gritty between your fingers—even the finest sand grains feel that way. **Silt** is essentially too small for individual grains to be visible, and while sand feels sandy to your fingers, silt feels smooth to your fingers but gritty in your mouth. Clay is so fine that it feels smooth even in your mouth.

Exercise 6.1 Describe the sediment on a beach

Providing that your landscape isn't covered in deep snow at present, visit a beach somewhere nearby—an ocean shore, a lake shore, or a river bank. Look carefully at the size and shape of the beach sediments. Are they sand, pebbles, or cobbles? If they are not too fine, you should be able to tell if they are well rounded or more angular.

The beach in Figure 6.1.1 is at Sechelt, B.C. Although there is a range of clast sizes, it's mostly made up of well-rounded cobbles interspersed with pebbles. This beach is subject to strong wave activity, especially when winds blow across the Strait of Georgia from the south. That explains why the clasts are relatively large and are well rounded.



Figure 6.1.1 Pebbles on an ocean beach at Sechelt, B.C.

See Appendix 3 for Exercise 6.1 answers.

If you drop a granule into a glass of water, it will sink quickly to the bottom (less than half a second). If you drop a grain of sand into the same glass, it will sink more slowly (a second or two depending on the size). A grain of silt will take several seconds to get to the bottom, and a particle of fine clay may never get there. The rate of settling is determined by the balance between gravity and friction, as shown in Figure 6.1.2. Large particles settle quickly because the gravitational force (which is proportional to the mass, and therefore to the volume of the particle) is much greater than the frictional resistance (which is proportional to the surface area of the particle). For smaller particles the difference between gravitational push and frictional resistance is less, so they settle slowly.

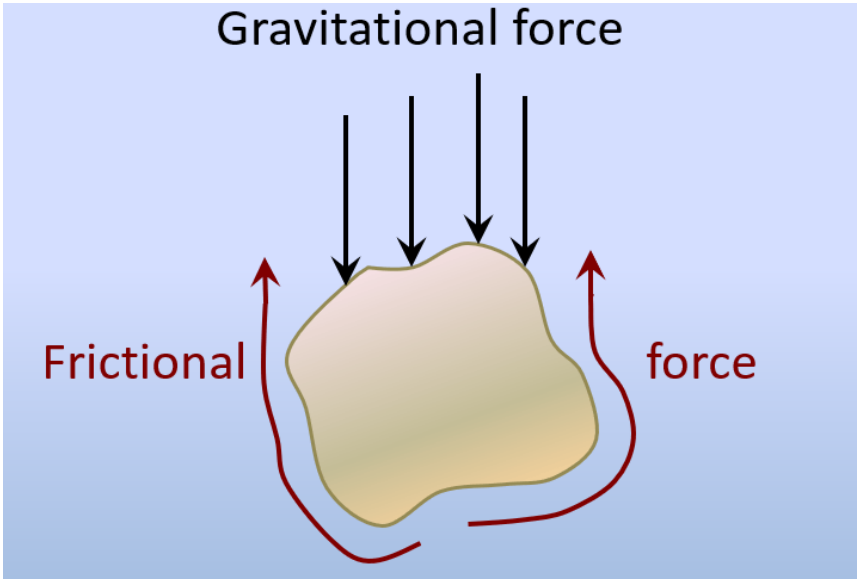


Figure 6.1.2 The two forces operating on a grain of sand in water. Gravity is pushing it down, and the friction between the grain and the water is resisting that downward force.

Small particles that settle slowly spend longer suspended in the water, and therefore tend to get moved farther than large particles if the water is moving.

Transportation

One of the key principles of sedimentary geology is that the ability of a moving medium (air or water) to move sedimentary particles—and keep them moving—is dependent on the velocity of flow. The faster the medium flows, the larger the particles it can move. This is illustrated in Figure 6.1.3. Parts of the river are moving faster than other parts, especially where the slope is greatest and the channel is narrow. Not only does the velocity of a river change from place to place, but it changes from season to season. During peak **discharge**[3] at the location of Figure 6.1.3, the water is high enough to flow over the embankment on the right, and it flows fast enough to move the boulders that cannot be moved during low flows.



Figure 6.1.3 Variations in flow velocity on the Englishman River near Parksville, B.C. When the photo was taken the river was not flowing fast enough anywhere to move the boulders and cobbles visible here. During flood events the water flows right over the snow-covered bank on the right, and is fast enough to move boulders.

Clasts within streams are moved in several different ways, as illustrated in Figure 6.1.4. Large **bed load** clasts are pushed (by

traction) or bounced along the bottom (by saltation), while smaller clasts are suspended in the water and kept there by the turbulence of the flow. As the flow velocity changes, different-sized clasts may be either incorporated into the flow or deposited on the bottom. At various places along a river, there are always some clasts being deposited, some staying where they are, and some being eroded and transported. This changes over time as the discharge of the river changes in response to changing weather conditions.

Other sediment transportation media, such as waves, ocean currents, and wind, operate under similar principles, with flow velocity as the key underlying factor that controls transportation and deposition.

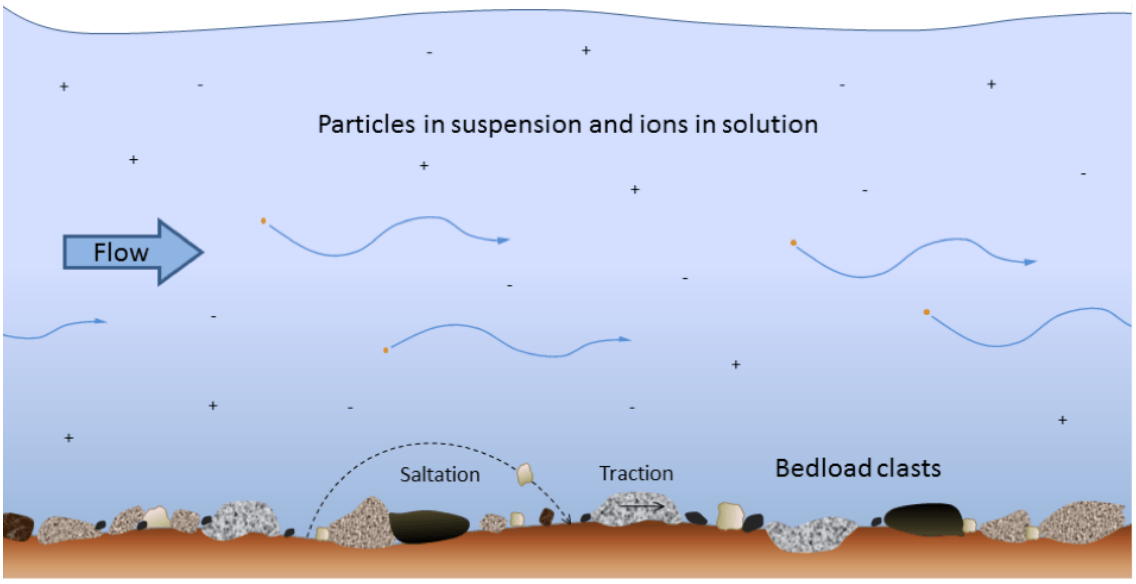


Figure 6.1.4 Transportation of sediment clasts by stream flow. The larger clasts, resting on the bottom (bedload), are moved by traction (sliding) or by saltation (bouncing). Smaller clasts are kept in suspension by turbulence in the flow. Ions (depicted as + and – in the image, but invisible in real life) are dissolved in the water.

Clastic sediments are deposited in a wide range of environments, including glaciers, slope failures, rivers—both fast and slow—lakes, deltas, and ocean environments—both shallow and deep. If the sedimentary deposits last long enough to get covered with other sediments they may eventually form into rocks ranging from fine mudstone to coarse breccia and conglomerate.

Lithification is the term used to describe a number of different processes that take place within a deposit of sediment to turn it into solid rock (Figure 6.1.5). One of these processes is burial by other sediments, which leads to compaction of the material and removal of some of the intervening water and air. After this stage, the individual clasts are touching one another. **Cementation** is the process of crystallization of minerals within the pores between the small clasts, and especially at the points of contact between clasts. Depending on the pressure, temperature, and chemical conditions, these crystals might include a range of minerals, the common ones being calcite, hematite, quartz and clay minerals.

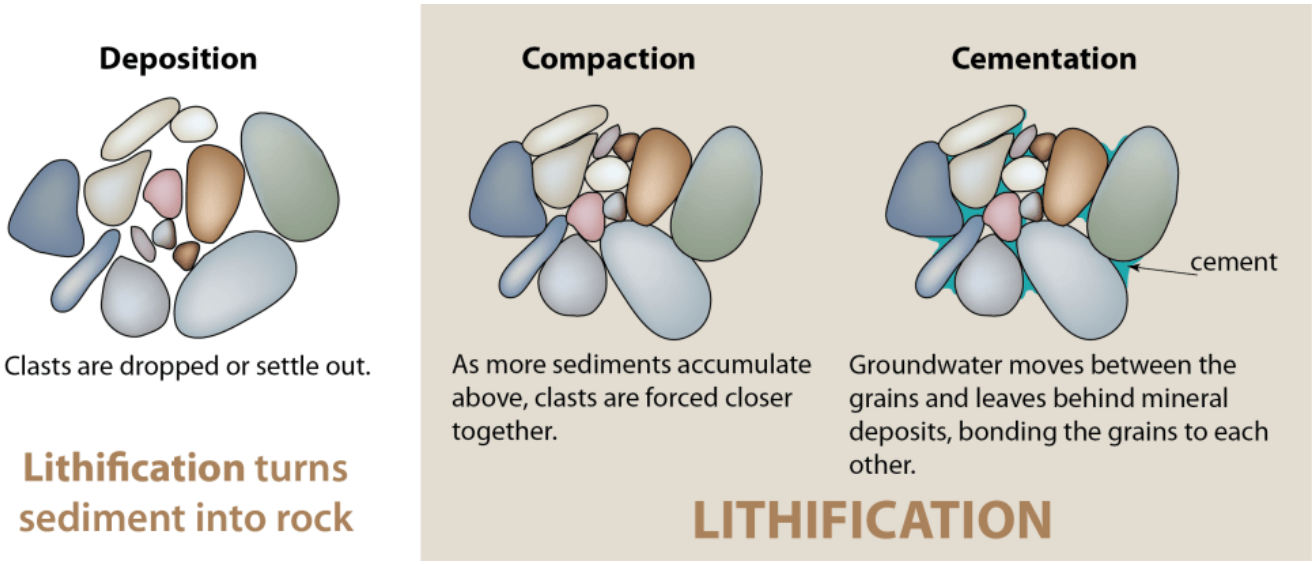


Figure 6.1.5 Lithification turns sediments into solid rock. Lithification involves the compaction of sediments and then the cementation of grains by minerals that precipitate from groundwater in the spaces between these grains. *Source: Karla Panchuk (2016) CC BY 4.0*

The characteristics and distinguishing features of clastic sedimentary rocks are summarized in Table 6.2. **Mudrock** is composed of at least 75% silt- and clay-sized fragments. If it is dominated by clay, it is called **claystone**. If it shows evidence of bedding or fine laminations, it is **shale**; otherwise, it is mudstone. Mudrocks form in very low energy environments, such as lakes, river backwaters, and the deep ocean.

Table 6.2 The main types of clastic sedimentary rocks and their characteristics.

[Skip Table]

Group	Examples	Characteristics
Mudrock	mudstone	Greater than 75% silt and clay, not bedded
	shale	Greater than 75% silt and clay, thinly bedded
Coal		Dominated by fragments of partially decayed plant matter often enclosed between beds of sandstone or mudrock.
Sandstone	quartz sandstone	Dominated by sand, greater than 90% quartz
	arkose	Dominated by sand, greater than 10% feldspar
	lithic wacke	dominated by sand, greater than 10% rock fragments, greater than 15% silt and clay
Conglomerate		Dominated by rounded clasts, granule size and larger
Breccia		Dominated by angular clasts, granule size and larger

Most coal forms in fluvial or delta environments where vegetation growth is vigorous and where decaying plant matter accumulates in long-lasting swamps with low oxygen levels. To avoid oxidation and breakdown, the organic matter must remain submerged for centuries or millennia, until it is covered with another layer of either muddy or sandy sediments. It is important to note that in some textbooks coal is described as an “organic sedimentary rock.” In this book, coal is included with the clastic rocks for two reasons: first, because it is made up of fragments of organic matter; and second, because coal seams (sedimentary layers) are almost always interbedded with layers of clastic rocks, such as mudrock or sandstone. In other words, coal accumulates in environments where other clastic rocks accumulate.

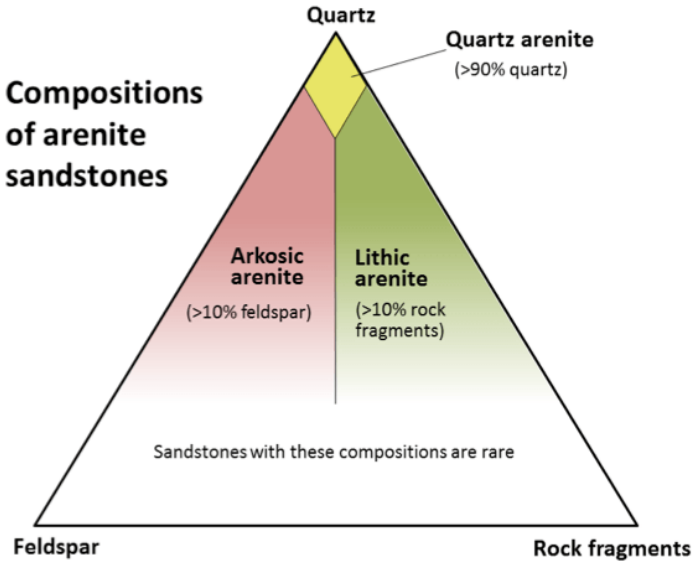


Figure 6.1.6 A compositional triangle for arenite sandstones, with the three most common components of sand-sized grains: quartz, feldspar, and rock fragments. Arenites have less than 15% silt or clay. Sandstones with more than 15% silt and clay are called wackes (e.g., quartz wacke, lithic wacke).

It's worth taking a closer look at the different types of sandstone because sandstone is a common and important sedimentary rock. Typical sandstone compositions are shown in Figure 6.1.6. Sandstones are mostly made up of sand grains of course, but they also include finer material—both silt and clay. The term **arenite** applies to a so-called clean sandstone, meaning one with less than 15% silt and clay. Considering the sand-sized grains only (the grains larger than $^{1}/_{16}$th mm), arenites with 90% or more quartz are called quartz arenites. If they have more than 10% feldspar and more feldspar than rock fragments, they are called feldspathic arenites or **arkosic arenites** (or just **arkose**). If they have more than 10% rock fragments, and more rock fragments than feldspar, they are **lithic arenites.**[4] A sandstone with more than 15% silt or clay is called a **wacke** (pronounced *wackie*). The terms *quartz wacke*, *lithic wacke*, and *feldspathic wacke* are used with limits similar to those on the arenite diagram. Another name for a lithic wacke is **greywacke**.

Some examples of sandstones, magnified in thin section are shown in Figure 6.1.7. (A thin section is rock sliced thin enough so that light can shine through.)

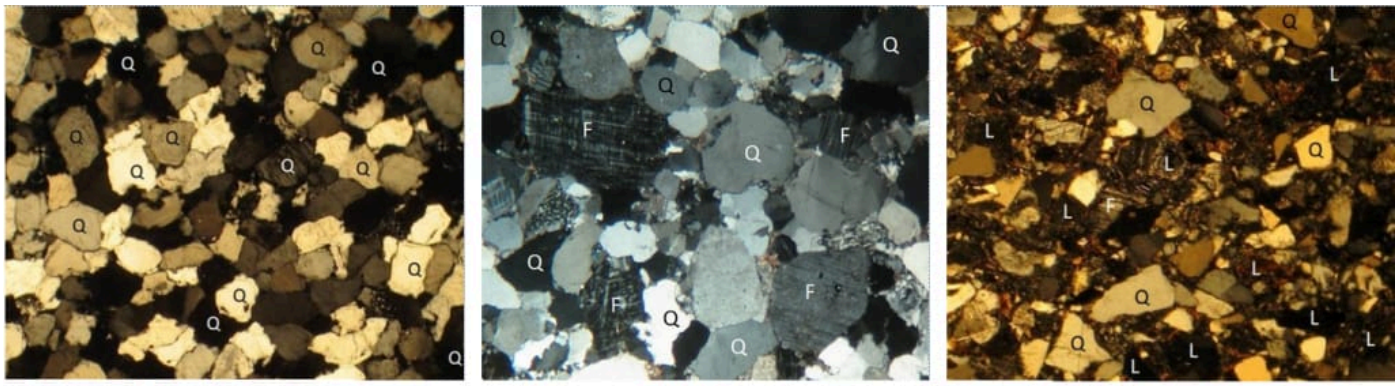


Figure 6.1.7 Microscope photos of three types of sandstone in thin-section. Some of the minerals are labelled: Q=quartz, F=feldspar and L= lithic (rock fragments). The quartz arenite and arkose have relatively little silt-clay matrix, while the lithic wacke has abundant matrix.

Clastic sedimentary rocks in which a significant proportion of the clasts are larger than 2 millimeters are known as **conglomerate** if the clasts are well rounded, and **breccia** if they are angular. Conglomerates form in high-energy environments such as fast-flowing rivers, where the particles can become rounded. Breccias typically form where the particles are not transported a significant distance in water, such as alluvial fans and talus slopes. Some examples of clastic sedimentary rocks are shown on Figure 6.1.8.

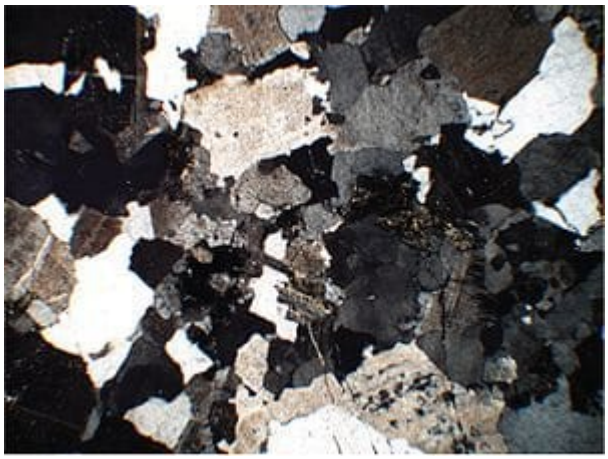


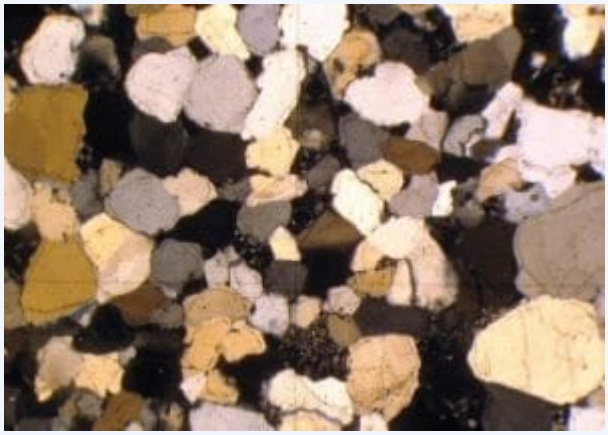
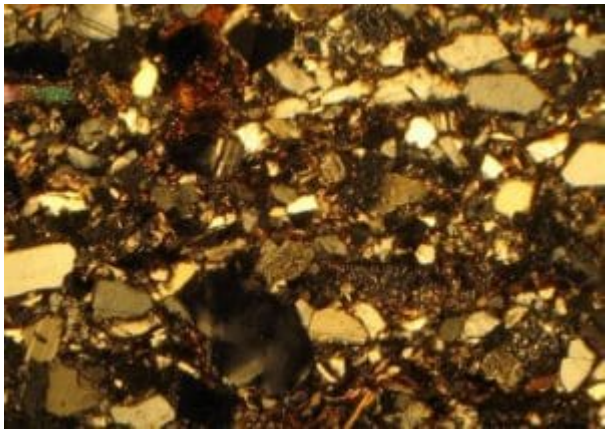
Figure 6.1.8 Examples of various clastic sedimentary rocks.

Exercise 6.2 Classifying sandstones

Table 6.3 below shows magnified thin sections of three sandstones, along with descriptions of their compositions. Using Table 6.1 and Figure 6.1.6, find an appropriate name for each of these rocks.

Table 6.3 Classifying sandstones

Magnified Thin Section	Description
	Angular sand-sized grains are approximately 85% quartz and 15% feldspar. Silt and clay make up less than 5% of the rock.

Magnified Thin Section	Description
	Rounded sand-sized grains are approximately 99% quartz and 1% feldspar. Silt and clay make up less than 2% of the rock.
	Angular sand-sized grains are approximately 70% quartz, 20% lithic, and 10% feldspar. Silt and clay make up about 20% of the rock.

See Appendix 3 for Exercise 6.2 answers.

Image Descriptions

Figure 6.1.8 image description: (A) Mudrock with bivalve impressions, Cretaceous Nanaimo group, Browns River, Vancouver Island. A very fine-grained rock with shell impressions. (B) Coarse sandstone with cross-bedding, Cambrian Tapeats Formation Chino Valley, Arizona. (C) Conglomerate with imbricate (aligned, tilted down to the left) cobbles, Cretaceous Geoffrey Formation, Hornby Island, BC. (D) Sedimentary breccia, the Pre-Cambrian Toby Formation, east of Castlegar, BC. [Return to Figure 6.1.8]

Media Attributions

- Figures 6.1.1, 6.1.2, 6.1.3, 6.1.4, 6.1.5, 6.1.6, 6.1.7, 6.1.8: © Steven Earle. CC BY.
- Exercise 6.2, first image: Aplite Red © Rudolf Pohl. CC BY-SA.

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1. A micron is a millionth of a metre. There are 1,000 microns in a millimetre. ↵
2. The largest known free-standing rock (i.e., not part of bedrock) is Giant Rock in the Mojave Desert, California. It's about as big as an apartment building—seven stories high! ↵
3. Discharge of a stream is the volume of flow passing a point per unit time. It's normally measured in cubic meters per second (m^3/s). ↵
4. "Lithic" means "rock." Lithic clasts are rock fragments, as opposed to mineral fragments. ↵

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6.2: Chemical Sedimentary Rocks

Whereas clastic sedimentary rocks are dominated by components that have been transported as solid clasts (clay, silt, sand, etc.), chemical sedimentary rocks are dominated by components that have been transported as ions in solution (Na^+, Ca^{2+}, HCO_3^-, etc.). There is some overlap between the two because almost all clastic sedimentary rocks contain cement formed from dissolved ions, and many chemical sedimentary rocks include some clasts. Since ions can stay in solution for tens of thousands of years (some much longer), and can travel for tens of thousands of kilometers, it is virtually impossible to relate chemical sediments back to their source rocks.

Chemical weathering and chemical sedimentary rocks

> Many students confuse chemical weathering with chemical sedimentary rocks, or mistakenly assume that when and where chemical weathering is taking place, chemical sedimentary rocks will accumulate. Most ions in solution in rivers, lakes and the ocean are produced during chemical weathering, but those ions can remain in solution for millions of years, and during that time they can travel hundreds of thousands of km (yes, literally around the world, several times). They might eventually come out of solution as a result of a biological process or a change in the chemical conditions and will then become a mineral crystal that can settle to form a chemical sediment.
>
> So the calcium ions that are part of a calcite mud on the sea floor near Australia's Great Barrier Reef could literally have come from anywhere on Earth (and almost certainly came from many different places), and might have been in solution for as little as a few days or for as long as tens of millions of years.

The most common chemical sedimentary rock, by far, is **limestone**. Others include **chert**, **banded iron formation,** and **evaporites**. Biological processes are important in the formation of some chemical sedimentary rocks, especially limestone and chert. For example, limestone is made up almost entirely of fragments of marine[1] organisms that manufacture calcite for their shells and other hard parts, and most chert includes at least some of the silica tests (shells) of tiny marine organisms (such as diatoms and radiolarians).

Limestone

Almost all limestone forms in the oceans, and most of that forms on the shallow continental shelves, especially in tropical regions with coral reefs. Reefs are highly productive ecosystems populated by a wide range of organisms, many of which use calcium and bicarbonate ions in seawater to make carbonate minerals (especially calcite) for their shells and other structures. These include corals, of course, but also green and red algae, urchins, sponges, molluscs, and crustaceans. The hard parts of these organisms are eroded by waves and currents to produce carbonate fragments that accumulate in the surrounding region, as illustrated in Figure 6.2.1.

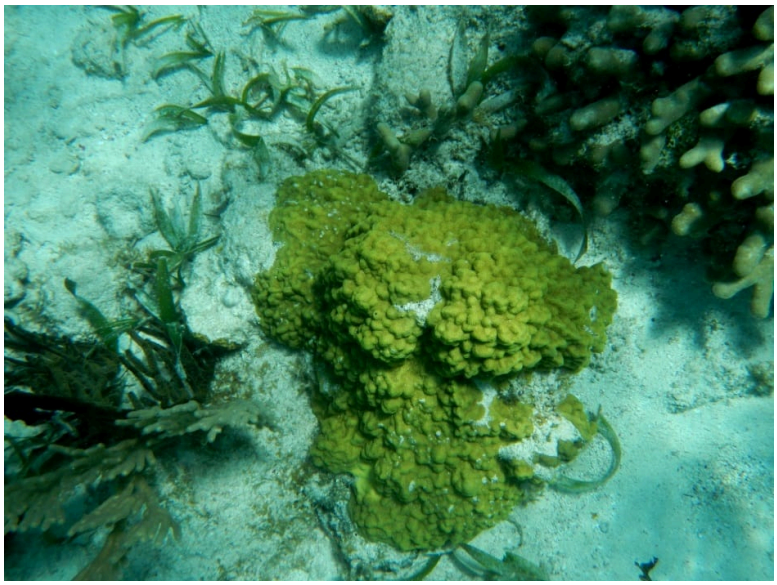


Figure 6.2.1 Various corals and green algae on a reef at Ambergris, Belize. The light-colored sand consists of carbonate fragments eroded from the reef organisms, and this is the type of material that will eventually become limestone.

Figure 6.2.2 shows a cross-section through a typical reef in a tropical environment (normally between 40° N and 40° S). Reefs tend to form in areas with clear water (e.g., not close to the mouths of large rivers), and near the edges of steep drop-offs because the reef organisms thrive on nutrient-rich upwelling currents. As the reef builds up, it is eroded by waves and currents to produce carbonate sediments that are transported into the steep offshore **fore-reef** area and the shallower inshore **back-reef** area. These sediments are dominated by reef-type carbonate fragments of all sizes, including mud. In many such areas, carbonate-rich sediments also accumulate in quiet lagoons, where mud and mollusc-shell fragments predominate (Figure 6.2.3a) or in offshore areas with strong currents, where either foraminifera tests accumulate (Figure 6.2.3b) or calcite crystallizes inorganically to form **ooids**—spheres of calcite that form in shallow tropical ocean water with strong currents (Figure 6.2.3c).

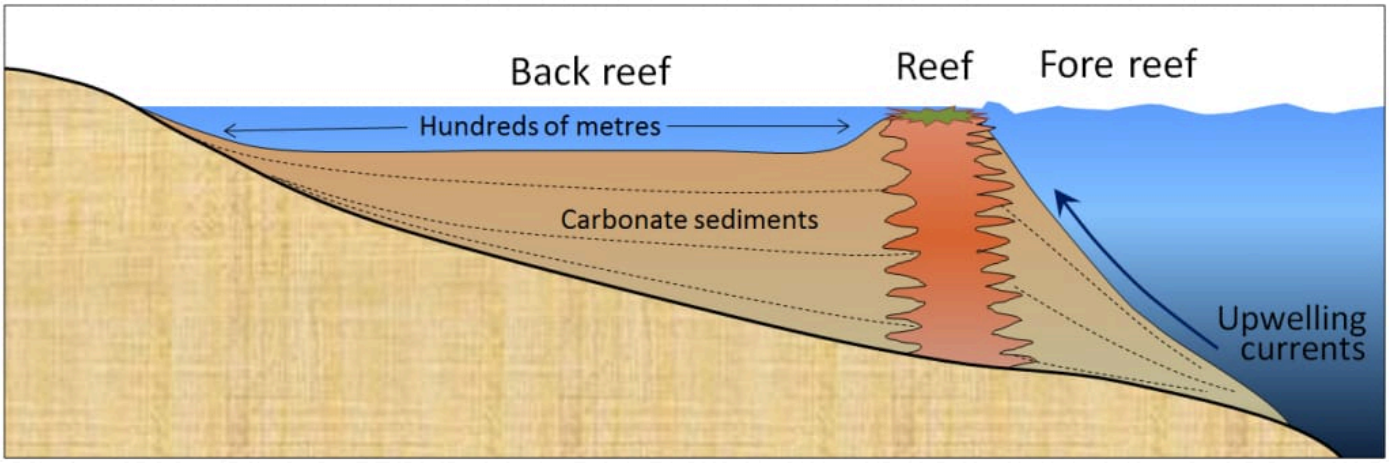


Figure 6.2.2 Schematic cross-section through a typical tropical reef.

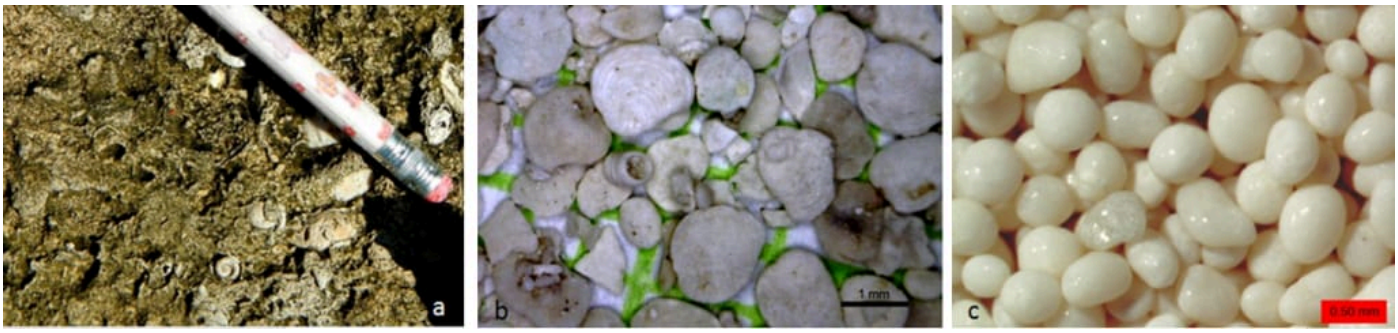


Figure 6.2.3 Carbonate rocks and sediments: (a) mollusc-rich limestone formed in a lagoon area at Ambergris, Belize, (b) foraminifera-rich sediment from a submerged carbonate sandbar in Belize (c) ooids from a beach at Joulters Cay, Bahamas.

Limestone also accumulates in deeper water, from the steady rain of the carbonate shells of tiny organisms that lived near the ocean surface. The lower limit for limestone accumulation is around 4,000 meters. Beneath that depth, calcite is soluble so limestone does not accumulate.

Calcite can also form on land in a number of environments. **Tufa** forms at springs (Figure 6.2.4) and **travertine** forms at hot springs. Similar material precipitates within limestone caves to form **stalactites**, **stalagmites,** and a wide range of other **speleothems**. Tufa, travertine and speleothems make up only a tiny proportion of all limestone.



Figure 6.2.4 Tufa formed at a spring at Johnston Creek, Alberta. The bedded grey rock to the left is limestone.

Dolomite ($CaMg(CO_3)_2$) is another carbonate mineral, but *dolomite* is also the name for a rock composed of the mineral dolomite (although some geologists use the term **dolostone** to avoid confusion). Dolomite rock is quite common (there's a whole Italian mountain range named after it), which is surprising since marine organisms don't make dolomite. All of the dolomite found in ancient rocks has been formed through magnesium replacing some of the calcium in the calcite in carbonate muds and sands. This process is known as **dolomitization**, and it is thought to take place where magnesium-rich water percolates through the sediments in carbonate tidal flat environments.

Chert

As we've seen, not all marine organisms make their hard parts out of calcite; some, like radiolarians and diatoms, use silica, and when they die their tiny shells (or tests) settle slowly to the bottom where they accumulate as chert. In some cases, chert is deposited along with limestone in the moderately deep ocean, but the two tend to remain separate, so chert beds within limestone are quite common (Figure 6.2.5), as are nodules, like the flint nodules of the Cretaceous chalk of southeastern England. In other situations, and especially in very deep water, chert accumulates on its own, commonly in thin beds.



Figure 6.2.5 Chert (brown layers) interbedded with Triassic Quatsino Fm. limestone on Quadra Island, B.C. All of the layers have been folded, and the chert, being insoluble and harder than limestone, stands out.

Banded iron formation

Banded iron formation (BIF) is a deep sea-floor deposit of iron oxide that is a common ore of iron (Figure 6.2.6). BIF forms when iron dissolved in seawater is oxidized, becomes insoluble, and sinks to the bottom in the same way that silica tests do to form chert. BIF is prevalent in rocks dating from 2400 to 1800 Ma, a result off changes in the atmosphere and oceans that took place over that time period. Photosynthetic bacteria (i.e., **cyanobacteria**, a.k.a. blue-green algae) consume carbon dioxide from the atmosphere and use solar energy to convert it to oxygen. These bacteria first evolved around 3500 Ma, and for the next billion years, almost all of that free oxygen was used up by chemical and biological processes, but by 2400 Ma free oxygen levels started to increase in the atmosphere and the oceans. Over a period of 600 million years, that oxygen gradually converted soluble ferrous iron (Fe^{2+}) to insoluble ferric iron (Fe^{3+}), which combined with oxygen to form the mineral hematite (Fe_2O_3), leading to the accumulation of BIFs on the sea floor. After 1800 Ma, little dissolved iron was left in the oceans and the formation of BIF essentially stopped.



Figure 6.2.6 An example of a banded iron formation with dark iron oxide layers interspersed with chert stained red by hematite. This rock is 2.1 billion years old.

Evaporites

In arid regions many lakes and inland seas have no stream outlet and the water that flows into them is removed only by evaporation. Under these conditions, the water becomes increasingly concentrated with dissolved salts, and eventually some of these salts reach saturation levels and start to crystallize (Figure 6.2.7). Although all evaporite deposits are unique because of differences in the chemistry of the water, in most cases minor amounts of carbonates start to precipitate when the solution is reduced to about 50% of its original volume. Gypsum ($CaSO_4 \cdot H_2O$) precipitates at about 20% of the original volume and halite (NaCl) precipitates at 10%. Other important evaporite minerals include sylvite (KCl) and borax ($Na_2B_4O_7 \cdot 10H_2O$). Sylvite is mined at numerous locations across Saskatchewan (Figure 6.2.8) from evaporites that were deposited during the Devonian (~385 Ma) when an inland sea occupied much of the region.



Figure 6.2.7 Spotted Lake, near Osoyoos, B.C. The patterns on the surface are salt. This photo was taken in May when the water was relatively fresh because of winter rains. By the end of the summer the surface of this lake is typically fully encrusted with salt deposits.



Figure 6.2.8 A mining machine at the face of potash ore (sylvite) in the Lanigan Mine near Saskatoon, Saskatchewan. The mineable potash layer (on the right) is about 3 meters thick.

Exercise 6.3 Making evaporite



Figure 6.2.9

This is an easy experiment that you can do at home. Pour about 50 mL (just less than 1/4 cup) of very hot water into a cup and add 2 teaspoons (10 mL) of salt. Stir until all or almost all of the salt has dissolved, then pour the salty water (leaving any undissolved salt behind) into a shallow wide dish or a small plate. Leave it to evaporate for a few days and observe the result. What is the size range and shape of the crystals you grew?

It may look a little like Figure 6.2.9. These crystals are up to about 3 millimeters across.

See Appendix 3 for Exercise 6.3 answers.

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1. We use the word *marine* when referring to salt water (i.e., oceanic) environments, and the word *aquatic* when referring to freshwater environments. ↵

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6.3: Depositional Environments and Sedimentary Basins

Sediments accumulate in a wide variety of environments, both on the continents and in the oceans. Some of the more important of these environments are illustrated in Figure 6.3.1.

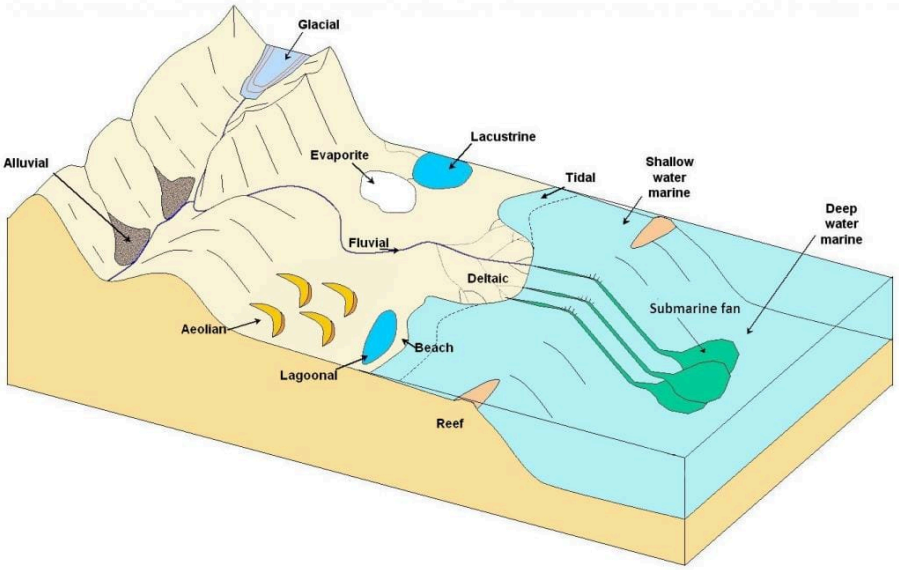


Figure 6.3.1 Some of the important depositional environments for sediments and sedimentary rocks.

Table 6.4 provides a summary of the processes and sediment types that pertain to the various depositional environments illustrated in Figure 6.3.1. We'll look more closely at the types of sediments that accumulate in these environments in the last section of this chapter. The characteristics of these various environments, and the processes that take place within them, are also discussed in later chapters on glaciation, mass wasting, streams, coasts, and the sea floor.

Table 6.4 The important terrestrial depositional environments and their characteristics

Environment	Important transport processes	Depositional environments	Typical sediment types
Glacial	gravity, moving ice, moving water	valleys, plains, streams, lakes	glacial till, gravel, sand, silt, and clay
Alluvial	gravity	steep-sided valleys	coarse angular fragments
Fluvial	moving water	streams	gravel, sand, silt, and organic matter (in swampy parts only)
Aeolian	wind	deserts and coastal regions	sand, silt
Lacustrine	moving water (flowing into a lake)	lakes	sand (near the edges only), silt, clay, and organic matter
Evaporite	moving water (flowing into a lake)	lakes in arid regions	salts, clay

Table 6.5 The important marine depositional environments and their characteristics

Environment	Important Transport Processes	Depositional Environments	Typical Sediment Types
Deltaic	moving water	deltas	sand, silt, clay, and organic matter (in swampy parts only)
Beach	waves, longshore currents	beaches, spits, sand bars	gravel, sand
Tidal	tidal currents	tidal flats	silt, clay
Reefs	waves and tidal currents	reefs and adjacent basins	carbonates
Shallow water marine	waves and tidal currents	shelves and slopes, lagoons	carbonates in tropical climates, sand/silt/clay elsewhere
Lagoonal	little transportation	lagoon bottom	carbonates in tropical climates
Submarine fan	underwater gravity flows	continental slopes and abyssal plains	gravel, sand, mud
Deep water marine	ocean currents	deep-ocean abyssal plains	clay, carbonate mud, silica mud

Most of the sediments that you might see around you, including talus on steep slopes, sand bars in streams, or gravel in road cuts, will never become sedimentary rocks because they have only been deposited relatively recently—perhaps a few centuries or millennia ago—and are likely to be re-eroded before they are buried deep enough beneath other sediments to be lithified. In order for sediments to be preserved long enough to be turned into rock—a process that takes millions or tens of millions of years—they need to have been deposited in a basin that will last that long. Most such basins are formed by plate tectonic processes, and some of the more important examples are shown in Figure 6.3.2.

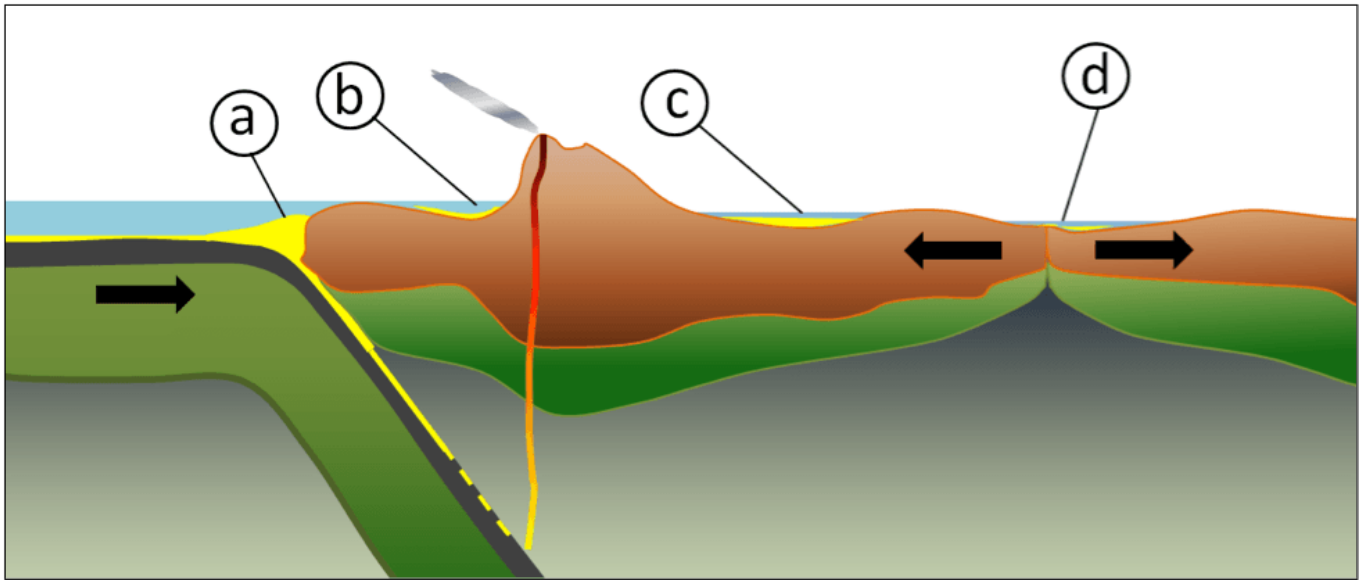


Figure 6.3.2 Some of the more important types of tectonically produced basins: (a) trench basin, (b) forearc basin, (c) foreland basin, and (d) rift basin.

Trench basins form where a subducting oceanic plate dips beneath the overriding continental or oceanic crust. They can be several kilometers deep, and in many cases, host thick sequences of sediments from eroding coastal mountains. There is a well-developed trench basin off the west coast of Vancouver Island. A forearc basin lies between the subduction zone and the volcanic arc, and may be formed in part by friction between the subducting plate and the overriding plate, which pulls part of the overriding plate down. The Strait of Georgia is a forearc basin. A foreland basin is caused by the mass of the volcanic range depressing the crust on either side. Foreland basins are not only related to volcanic ranges, but can form adjacent to fold belt mountains like the Canadian Rockies. A rift basin forms where continental crust is being pulled apart, and the crust on both sides of the rift subsides. As rifting continues this eventually becomes a narrow sea, and then an ocean basin. The East African rift basin represents an early stage in this process.

Media Attributions

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6.4: Sedimentary Structures and Fossils

Through careful observation over the past few centuries, geologists have discovered that the accumulation of sediments and sedimentary rocks takes place according to some important geological principles, as follows:

- The **principle of original horizontality** is that sediments accumulate in essentially horizontal layers. The implication is that tilted sedimentary layers observed to day must have been subjected to tectonic forces.
- The **principle of superposition** is that sedimentary layers are deposited in sequence, and that unless the entire sequence has been turned over by tectonic processes, the layers at the bottom are older than those at the top.
- The **principle of inclusions** is that any rock fragments in a sedimentary layer must be older than the layer. For example, the cobbles in a conglomerate must have been formed before the conglomerate was formed.
- The **principle of faunal succession** is that there is a well-defined order in which organisms have evolved through geological time, and therefore the identification of specific fossils in a rock can be used to determine its age.

In addition to these principles, that apply to all sedimentary rocks (as well as volcanic rocks), a number of other important characteristics of sedimentary processes result in the development of distinctive sedimentary features in specific sedimentary environments. By understanding the origins of these features, we can make some very useful inferences about the processes that led to deposition the rocks that we are studying.

Bedding, for example, is the separation of sediments into layers that either differ from one another in textures, composition, color, or weathering characteristics, or are separated by **partings**—narrow gaps between adjacent beds (Figure 6.4.1). Bedding is an indication of changes in depositional processes that may be related to seasonal differences, changes in climate, changes in locations of rivers or deltas, or tectonic changes. Partings may represent periods of non-deposition that could range from a few decades to a few millennia. Bedding can form in almost any sedimentary depositional environment.



Figure 6.4.1 The Triassic Sulphur Mt. Formation near Exshaw, Alberta. Bedding is defined by differences in color and texture, and also by partings (gaps) between beds that may otherwise appear to be similar.

Cross-bedding is bedding that contains angled layers within otherwise horizontal beds, and it forms when sediments are deposited by flowing water or wind. Some examples are shown in Figures 6.0.11, 6.1.7b, and 6.4.2. Cross-beds formed in streams tend to be on the scale of centimeters to tens of centimeters, while those in **aeolian** (wind deposited) sediments can be on the scale of meters to several meters.



Figure 6.4.2 Cross-bedded Jurassic Navajo Formation aeolian sandstone at Zion National Park, Utah. In most of the layers the cross-beds dip down toward the right, implying a consistent wind direction from right to left during deposition.

Cross-beds form as sediments are deposited on the leading edge of an advancing ripple or dune under steady state conditions (similar flow rate and same flow direction). Each layer is related to a different ripple that advances in the direction of flow, and is partially eroded by the following ripple (Figure 6.4.3). Cross-bedding is a very important sedimentary structure to be able to recognize because it can provide information on the process of deposition, the direction of current flows and, when analyzed in detail, on other features like the rate of flow and the amount of sediment available.

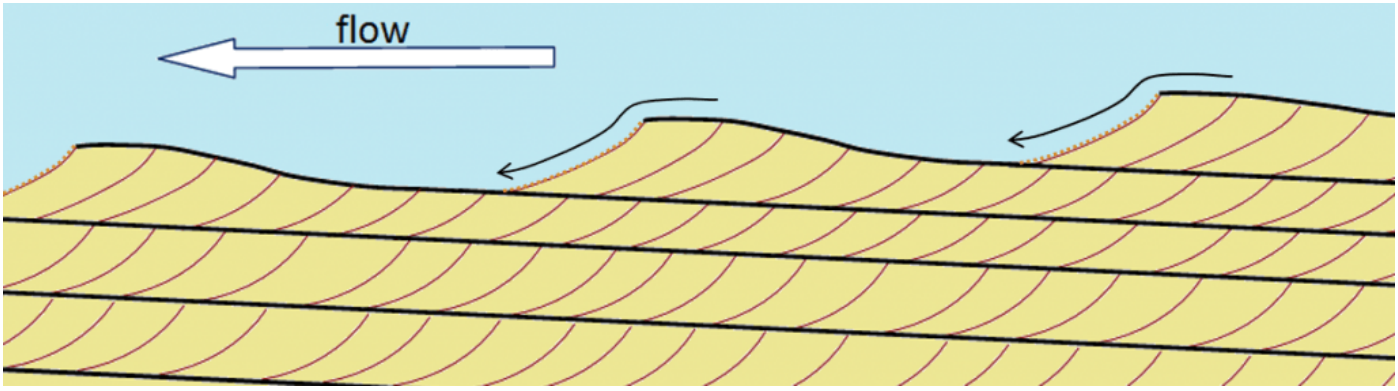


Figure 6.4.3 Formation of cross-beds as a series of ripples or dunes migrates with the flow. Each ripple advances forward (right to left in this view) as more sediment is deposited on its leading face (small arrows). (On each ripple the last deposited layer is represented by small dots.)

Graded bedding is characterized by a gradation in grain size from bottom to top within a single bed. “Normal” graded beds are coarse at the bottom and become finer toward the top. They are a product of deposition from a slowing current (Figure 6.4.4). Most graded beds form in a submarine-fan environment (see Figure 6.4.1), where sediment-rich flows descend periodically from a shallow marine shelf down a slope and onto the deeper sea floor. Some graded beds are reversed (coarser at the top), and this normally results from deposition by a fast-moving debris flow (see Chapter 15).

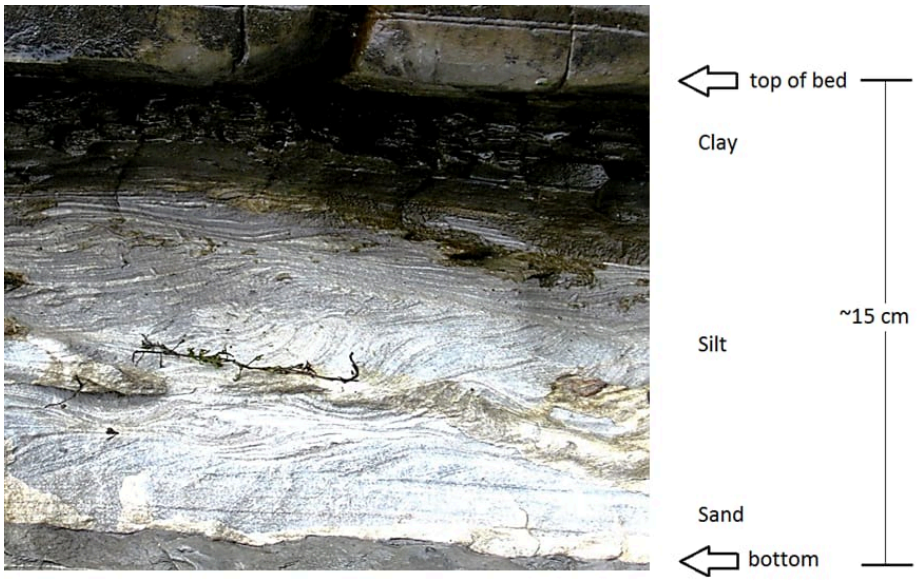


Figure 6.4.4 A graded turbidite bed in Cretaceous Spray Formation rocks on Gabriola Island, B.C. The lower several centimeters of sand and silt probably formed over the duration of less than an hour. The upper few centimeters of fine clay may have accumulated over several hundred years.

Ripples, which are associated with the formation of cross-bedding, may be preserved on the surfaces of sedimentary beds. Ripples can also help to determine flow direction as they tend to have their steepest surface facing in the direction of the flow (see Figure 6.4.3).

In a stream environment, boulders, cobbles, and pebbles can become **imbricated**, meaning that they are generally tilted in the same direction. Clasts in streams tend to tilt with their upper ends pointing downstream because this is the most stable position with respect to the stream flow (Figure 6.4.5 and Figure 6.4.7c).

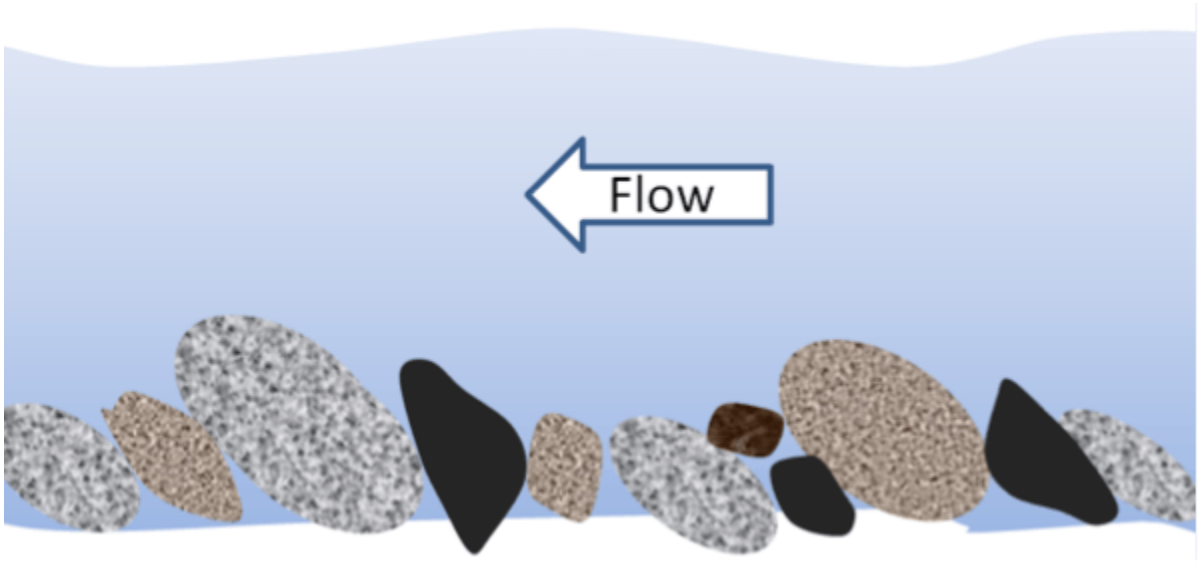


Figure 6.4.5 An illustration of imbrication of clasts in a fluvial environment.

Mud cracks form when a shallow body of water (e.g., a tidal flat or pond or even a puddle), into which muddy sediments have been deposited, dries up and cracks (Figure 6.4.6). This happens because the clay in the upper mud layer tends to shrink on drying, and so it cracks because it occupies less space when it is dry.



Figure 6.4.6 Mudcracks in volcanic mud at a hot-spring area near Myvatn, Iceland.

The various structures described above are critical to understanding and interpreting the conditions that existed during the formation of sedimentary rocks. In addition to these, geologists also look very closely at sedimentary grains to determine their mineralogy or lithology (in order to make inferences about the type of source rock and the weathering processes), their degree of rounding, their sizes, and the extent to which they have been sorted by transportation and depositional processes. Some of the types of differences that we might want to look for are illustrated in Figure 6.4.7.

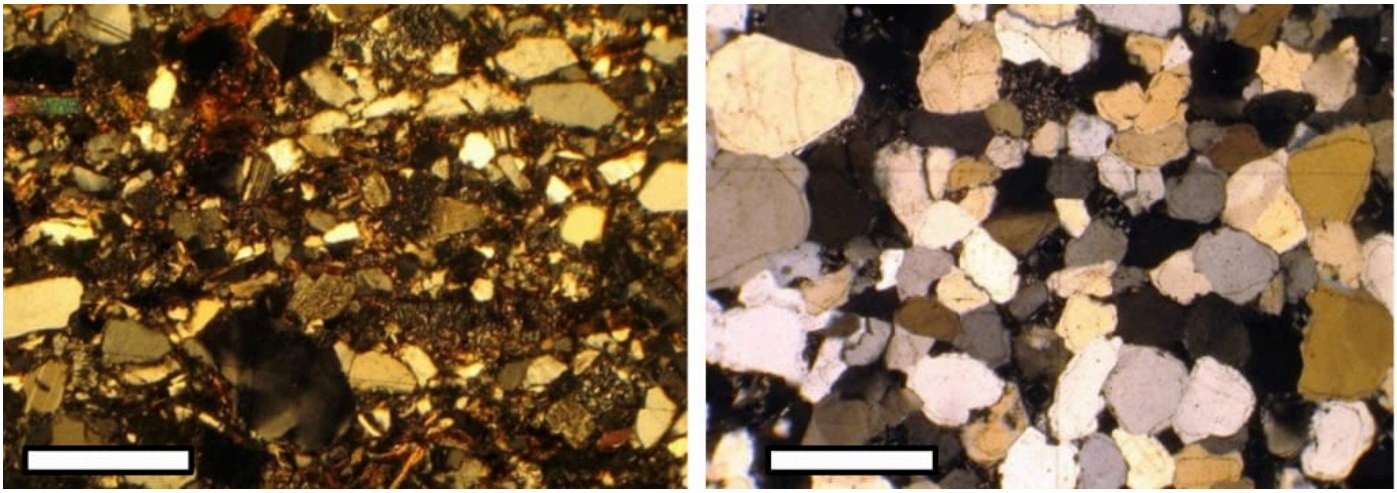


Figure 6.4.7 Thin section photos of two sandstones with very different grain characteristics. The one on the left has angular grains with a wide range of different types (quartz, feldspar, biotite, rock fragments), and is poorly sorted (grains range from less than 0.005 mm to ~1 mm). The one on the right has relatively well-rounded grains of quartz only, and the size range is much less (approx. 0.25 to 1 mm). (Scale bars are 1 mm.)

We won't be covering fossils in any detail in this book, but they are extremely important for understanding sedimentary rocks. Of course, fossils can be used to date sedimentary rocks, but equally importantly, they tell us a great deal about the depositional environment of the sediments and the climate at the time. For example, they can help to differentiate marine versus terrestrial environments; estimate the depth of the water; detect the existence of currents; and estimate average temperature and precipitation. For example, the tests of tiny marine organisms (mostly foraminifera) have been recovered from deep-ocean sediment cores from all over the world, and their isotopic signatures have been measured. As we'll see in Chapter 19, this has provided us with information about the changes in average global temperatures over the past 65 million years.

Exercise 6.4 Interpretation of past environments

Sedimentary rocks can tell us a great deal about the environmental conditions that existed during the time of their formation. Make some inferences about the source rock, weathering environment, type and distance of sediment transportation, and deposition conditions that existed during the formation of the following rocks:

1. **Quartz sandstone:** no feldspar, well-sorted and well-rounded quartz grains, cross-bedding
2. **Feldspathic sandstone and mudstone:** feldspar, volcanic fragments, angular grains, repetitive graded bedding from sandstone upwards to mudstone
3. **Conglomerate:** well-rounded pebbles and cobbles of granite and basalt; imbrication
4. **Breccia:** poorly sorted, angular limestone fragments; orange-red matrix

See Appendix 3 for Exercise 6.4 answers.

Media Attributions

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6.5: Groups, Formations, and Members

Geologists who study sedimentary rocks need ways to divide them into manageable units, and they also need to give those units names so that they can easily be referred to and compared with other rocks deposited in other places. The International Commission on Stratigraphy (ICS) has established a set of conventions for grouping, describing, and naming sedimentary rock units.

The main stratigraphic unit is a **formation**, which according to the ICS, should be established with the following principles in mind:

> *The contrast in lithology between formations required to justify their establishment varies with the complexity of the geology of a region and the detail needed for geologic mapping and to work out its geologic history. No formation is considered justifiable and useful that cannot be delineated at the scale of geologic mapping practiced in the region. The thickness of formations may range from less than a meter to several thousand meters.*

In other words, a formation is a series of beds that is distinct from other beds above and below, and is thick enough to be shown on the geological maps that are widely used within the area in question. In most parts of the world, geological mapping is done at a relatively coarse scale, and so most formations are in the order of a few hundred meters thick. At that thickness, a typical formation would appear on a typical geological map as an area that is at least a few millimeters thick.

A series of formations can be classified together to define a **group**, which could be as much as a few thousand meters thick, and represents a series of rocks that were deposited within a single basin (or a series of related and adjacent basins) over a few million to a few tens of millions of years.

In areas where detailed geological information is needed (for example, within a mining or petroleum district) a formation might be divided into **members**, where each member has a specific and distinctive lithology. For example, a formation that includes both shale and sandstone might be divided into members, each of which is either shale or sandstone. In some areas, where particular detail is needed, members may be divided into beds, but this is only applicable to beds that have a special geological significance. Groups, formations, and members are typically named for the area where they are found.

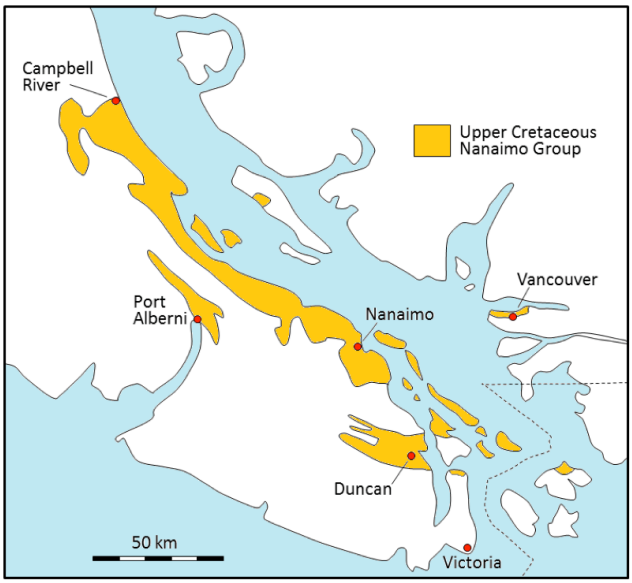


Figure 6.5.1 The distribution of the Upper Cretaceous Nanaimo Group rocks on Vancouver Island, the Gulf Islands, and in the Vancouver area.

The sedimentary rocks of the Nanaimo Group provide a useful example for understanding groups, formations, and members. During the latter part of the Cretaceous Period, from about 90 Ma to 65 Ma, a thick sequence of clastic rocks was deposited in a foreland basin between what is now Vancouver Island and the B.C. mainland (Figure 6.5.1). The Nanaimo Group strata comprise a 5000-metre-thick sequence of conglomerate, sandstone, and mudstone layers. Coal was mined from Nanaimo Group rocks from around 1850 to 1950 in the Nanaimo region, and even more recently in the Campbell River area.

The Nanaimo Group is divided into 11 formations as described in Table 6.6. In general, the boundaries between formations are based on major lithological differences. As can be seen in the far-right column of Table 6.6, a wide range of depositional environments existed during the accumulation of the Nanaimo Group rocks, from nearshore marine for the Comox and Haslam Formation, to fluvial and deltaic with backwater swampy environments for the coal-bearing Extension, Pender, and Protection Formations, to a deep-water submarine fan environment for the upper six formations.

Table 6.6 The formations of the Nanaimo Group[1]

[Skip Table 6.6]

Approximate Age (Ma)	Formation name	Lithologies	Depositional Environment
65 to 66	Gabriola	Sandstone with minor mudstone	Submarine fan, high energy
66 to 67	Spray (Fine grained)	Mudstone/sandstone turbidites	Submarine fan, low energy
67 to 68	Geoffrey	Sandstone and conglomerate	Submarine fan, high energy
68 to 70	Northumberland (Fine grained)	Mudstone turbidites	Submarine fan, low energy
70	De Courcy	Sandstone	Submarine fan, high energy
70 to 72	Cedar District (Fine grained)	Mudstone turbidites	Submarine fan, low energy
72 to 75	Protection	Sandstone and minor coal	Nearshore marine and onshore deltaic and fluvial
75 to 80	Pender	Sandstone and minor coal	Nearshore marine and onshore deltaic and fluvial
80	Extension	Conglomerate, with minor sandstone and some coal	Nearshore marine and onshore deltaic and fluvial
80 to 85	Haslam (Fine grained)	Mudstone and siltstone	Shallow marine
85 to 90	Comox	Conglomerate, sandstone, mudstone (coal in the Campbell River area)	Nearshore fluvial and marine

In tables like this one, the layers are always listed in order, with the oldest at the bottom and the youngest at the top.

The five lower formations of the Nanaimo Group are all exposed in the Nanaimo area, and were well studied during the coal mining era between 1850 and 1950. All of these formations (except Haslam) have been divided into members, as that was useful for understanding the rocks in the areas where coal mining was taking place. Within some of those members even some individual beds have been named if they were of specific importance to the mining industry.

Although there is a great deal of variety in the Nanaimo Group rocks, and it would take hundreds of photographs to illustrate all of the different types of rocks, a few representative examples are provided in Figure 6.5.2.



Figure 6.5.2a Representative photos of Nanaimo Group rocks. Turbidite layers in the Spray Formation on Gabriola Island. Each turbidite set consists of a lower sandstone layer (light color) that grades upward into siltstone, and then into mudstone. (See Figure 6.5.2 for detail.)



Figure 6.5.2b Two separate layers of fluvial sandstone with a thin (approx. 75 centimeters) coal seam in between. Pender Formation in Nanaimo.

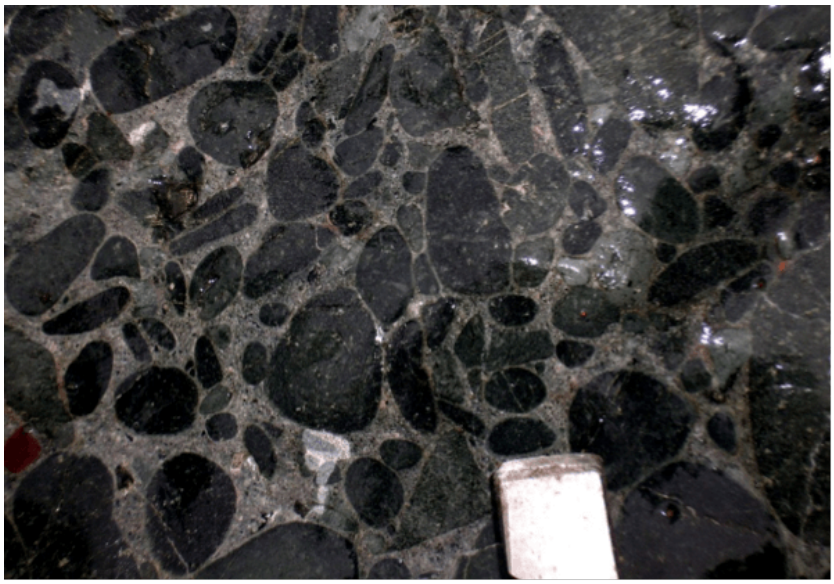


Figure 6.5.2c Comox Formation conglomerate at the very base of the Nanaimo Group in Nanaimo. The metal object is the end of a rock hammer that is 3 centimeters wide. Almost all of the clasts in this view are well-rounded basalt pebbles cobbles eroded from the Triassic Karmutsen Formation which makes up a major part of Vancouver Island.

Image Descriptions

Figure 6.5.1 image description: A map showing that Nanaimo Group rocks are present along the east coast of Vancouver Island from Nanaimo to Campbell River, farther inland in areas around Port Alberni and Duncan, on much of the Gulf Islands and a bit in the Vancouver area. [Return to Figure 6.5.1]

Media Attributions

- Figure 6.5.1: Redrawn based on Mustard, P., 1994, The Upper Cretaceous Nanaimo Group, Georgia Basin, in J. Monger (ed) Geology and Geological Hazards of the Vancouver Region, Geol. Survey of Canada, Bull. 481, pp. 27-95. © Steven Earle. CC BY.
- Figure 6.5.2abc: © Steven Earle. CC BY.

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1. [Based on data in Mustard, P., 1994, The Upper Cretaceous Nanaimo Group, Georgia Basin, in J. Monger (ed) Geology and Geological Hazards of the Vancouver Region, Geol. Survey of Canada, Bull. 481, p. 27-95.] ↩

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6.6: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
6.1 Clastic Sedimentary Rocks	Sedimentary clasts are classified based on their size, and variations in clast size have important implications for transportation and deposition. Clastic sedimentary rocks range from conglomerate to mudstone. Clast size, sorting, composition, and shape are important features that allow us to differentiate clastic rocks and understand the processes that took place during their deposition.
6.2 Chemical Sedimentary Rocks	Chemical sedimentary rocks form from ions that were transported in solution, and then converted into minerals by biological and/or chemical processes. The most common chemical rock, limestone, typically forms in shallow tropical environments, where biological activity is a very important factor. Chert and banded iron formation are deep-ocean sedimentary rocks. Evaporites form where the water of lakes and inland seas becomes supersaturated due to evaporation.
6.3 Depositional Environments and Sedimentary Basins	There is a wide range of depositional environments, both on land (glaciers, lakes, rivers, etc.) and in the ocean (deltas, reefs, shelves, and the deep-ocean floor). In order to be preserved, sediments must accumulate in long-lasting sedimentary basins, most of which form through plate-tectonic processes.
6.4 Sedimentary Structures and Fossils	The deposition of sedimentary rocks takes place according to a series of important principles, including original horizontality, superposition, and faunal succession. Sedimentary rocks can also have distinctive structures, such as cross bedding, graded bedding and mud cracks, that are important in determining their depositional environments. Fossils are useful for determining the age of a rock, the depositional environment, and the climate at the time of deposition.
6.5 Groups, Formations, and Members	Sedimentary sequences are classified into groups, formations, and members so that they can be referred to easily and without confusion.

Questions for Review

Answers to Review Questions at the end of each chapter can be found in Appendix 2.

1. What are the minimum and maximum sizes of sand grains?
2. How can you easily distinguish between a silty deposit and one that has only clay-sized material?
3. What factors control the rate at which a clast settles in water?
4. The material that makes up a rock such as conglomerate cannot be deposited by a slow-flowing river. Why not?
5. Describe the two main processes of lithification.
6. What is the difference between a lithic arenite and a lithic wacke?
7. How does a feldspathic arenite differ from a quartz arenite?
8. What can we say about the source area lithology and the weathering and transportation history of a sandstone that is primarily composed of rounded quartz grains?
9. What is the original source of the carbon that is present within carbonate deposits such as limestone?
10. What long-term environmental change on Earth led to the deposition of banded iron formations?
11. Name two important terrestrial depositional environments and two important marine ones.
12. What is the origin of a foreland basin, and how does it differ from a forearc basin?
13. Explain the origin of (a) bedding, (b) cross-bedding, (c) graded bedding, and (d) mud cracks.
14. Under what conditions is reverse graded bedding likely to form?
15. What are the criteria for the application of a formation name to a series of sedimentary rocks?
16. Explain why some of the Nanaimo Group formations have been divided into members, while others have not.

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7: METAMORPHISM AND METAMORPHIC ROCKS

Metamorphism is the change that takes place within a body of rock as a result of it being subjected to conditions that are different from those in which it formed. In most cases, but not all, this involves the rock being deeply buried beneath other rocks, where it is subjected to higher temperatures and pressures than those under which it formed. Metamorphic rocks typically have different mineral assemblages and different textures from their parent rocks, but they may have the same overall composition.

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7.1: Controls Over Metamorphic Processes

The main factors that control metamorphic processes are:

- the mineral composition of the parent rock,
- the temperature at which metamorphism takes place,
- the amount and type of pressure during metamorphism,
- the types of fluids (mostly water) that are present during metamorphism, and
- the amount of time available for metamorphism.

Parent Rock

The **parent rock** is the rock that exists before metamorphism starts. Sedimentary or igneous rocks can be considered the parent rocks for metamorphic rocks. Although an existing metamorphic rock can be further metamorphosed or re-metamorphosed, metamorphic rock doesn't normally qualify as a "parent rock". For example, if a mudstone is metamorphosed to slate and then buried deeper where it is metamorphosed to schist, the parent rock of the schist is mudstone, not slate. The critical feature of the parent rock is its mineral composition because it is the stability of minerals that counts when metamorphism takes place. In other words, when a rock is subjected to increased temperatures, certain minerals may become unstable and start to recrystallize into new minerals.

Temperature

The temperature that the rock is subjected to is a key variable in controlling the type of metamorphism that takes place. As we learned in the context of igneous rocks, mineral stability is a function of temperature, pressure, and the presence of fluids (especially water). All minerals are stable over a specific range of temperatures. For example, quartz is stable from environmental temperatures (whatever the weather can throw at it) all the way up to about 1800°C. If the pressure is higher, that upper limit will be even higher. If there is water present, it will be lower. On the other hand, most clay minerals are only stable up to about 150° or 200°C; above that, they transform into micas. Most feldspars are stable up to between 1000°C and 1200°C. Most other common minerals have upper limits between 150°C and 1000°C.

Some minerals will crystallize into different **polymorphs** (same composition, but different crystalline structure) depending on the temperature and pressure. The minerals kyanite, andalusite, and sillimanite are polymorphs with the composition Al_2SiO_5. They are stable at different pressures and temperatures, and, as we will see later, they are important indicators of the pressures and temperatures that existed during the formation of metamorphic rocks (Figure 7.1.1).

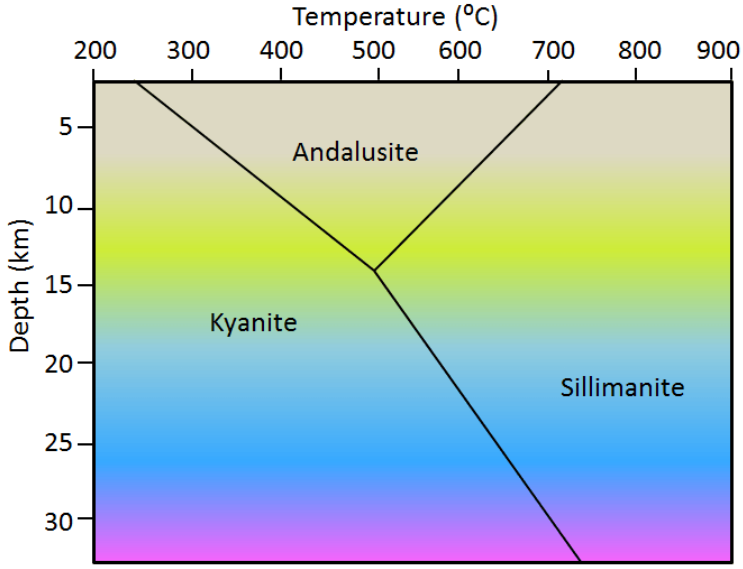


Figure 7.1.1 The temperature and pressure stability fields of the three polymorphs of Al_2SiO_5 (Pressure is equivalent to depth. Kyanite is stable at low to moderate temperatures and low to high pressures, andalusite at moderate temperatures and low pressures, and sillimanite at higher temperatures.)

Pressure

Pressure is important in metamorphic processes for two main reasons. First, it has implications for mineral stability (Figure 7.1.1). Second, it has implications for the texture of metamorphic rocks. Rocks that are subjected to very high confining pressures are typically denser than others because the mineral grains are squeezed together (Figure 7.1.2a), and also because they may contain minerals that have greater density because the atoms are more closely packed.

Because of plate tectonics, pressures within the crust are typically not applied equally in all directions. In areas of plate convergence, for example, the pressure in one direction (perpendicular to the direction of convergence) is typically greater than in the other directions (Figure 7.1.2b). In situations where different blocks of the crust are being pushed in different directions, the rocks will likely be subjected to sheer stress (Figure 7.1.2c).

One of the results of directed pressure and shear stress is that rocks become **foliated**—meaning that they'll have a directional fabric. Foliation a very important aspect of metamorphic rocks, and is described in more detail later in this chapter.

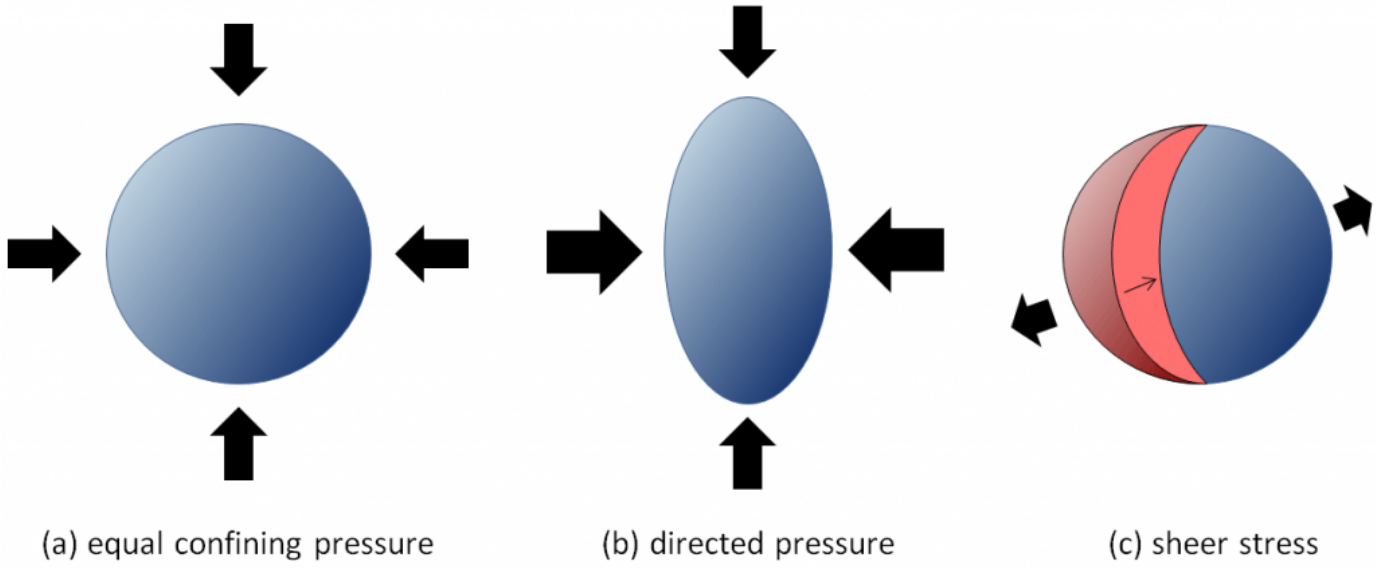


Figure 7.1.2 An illustration of different types of pressure on rocks. (a) confining pressure, where the pressure is essentially equal in all directions, (b) directed pressure, where the pressure form the sides is greater than that from the top and bottom, and (c) shear stress caused by different blocks of rock being pushed in different directions. (In a and b there is also pressure in and out of the page.)

Fluids

Water is the main fluid present within rocks of the crust, and the only one that we'll consider here. The presence of water is important for two main reasons. First, water facilitates the transfer of ions between minerals and within minerals, and therefore increases the rates at which metamorphic reactions take place. So, while the water doesn't necessarily change the outcome of a metamorphic process, it speeds the process up so metamorphism might take place over a shorter time period, or metamorphic processes that might not otherwise have had time to be completed are completed.

Secondly, water, especially hot water, can have elevated concentrations of dissolved elements (ions), and therefore it is an important medium for moving certain elements around within the crust. So not only does water facilitate metamorphic reactions on a grain-to-grain basis, it also allows for the transportation of elements from one place to another. This is very important in hydrothermal processes, which are discussed toward the end of this chapter, and in the formation of mineral deposits.

Time

Most metamorphic reactions take place at very slow rates. For example, the growth of new minerals within a rock during metamorphism has been estimated to be about 1 millimetre per million years. For this reason, it is very difficult to study metamorphic processes in a lab.

While the rate of metamorphism is slow, the tectonic processes that lead to metamorphism are also very slow, so in most cases, the chance for metamorphic reactions to be completed is high. For example, one important metamorphic setting is many kilometers deep within the roots of mountain ranges. A mountain range takes tens of millions of years to form, and tens of millions of years more to be eroded to the extent that we can see the rocks that were metamorphosed deep beneath it.

Exercise 7.1



Figure 7.1.3 Garnets in a rock. Euro coin (23 mm) is for scale.

This photo shows a sample of garnet-mica schist from the Greek island of Syros. The large reddish crystals are garnet, and the surrounding light colored rock is dominated by muscovite mica. The Euro coin is 23 millimeters in diameter. Assume that the diameters of the garnets increased at a rate of 1 millimetre per million years.

Based on the approximate average diameter of the garnets visible, estimate how long this metamorphic process might have taken.

See Appendix 3 for Exercise 7.1 answers.

Image Descriptions

Figure 7.1.1 image description: The temperature ranges that polymorphs of Al_2SiO_5 are stable at at various depths.

Depth (kilometers)	Kyanite	Andalusite	Sillimanite
5	Less than 300°C	300 to 650°C	Greater than 670°C
10	Less than 400°C	410 to 580°C	Greater than 590°C
15	Less than 500°C	Not stable	Greater than 500°C
20	Less than 570°C	Not stable	Greater than 590°C
25	Less than 640°C	Not stable	Greater than 620°C
30	Less than 700°C	Not stable	Greater than 700°C

[Return to Figure 7.1.1]

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7.2: Classification of Metamorphic Rocks

There are two main types of metamorphic rocks: those that are foliated because they have formed in an environment with either directed pressure or shear stress, and those that are not foliated because they have formed in an environment without directed pressure or relatively near the surface with very little pressure at all. Some types of metamorphic rocks, such as quartzite and marble, which can form whether there is directed-pressure or not, do not typically exhibit foliation because their minerals (quartz and calcite respectively) do not tend to show alignment (see Figure 7.2.8).

When a rock is squeezed under directed pressure during metamorphism it is likely to be deformed, and this can result in a textural change such that the minerals appear elongated in the direction perpendicular to the main stress (Figure 7.2.1). This contributes to the formation of foliation.

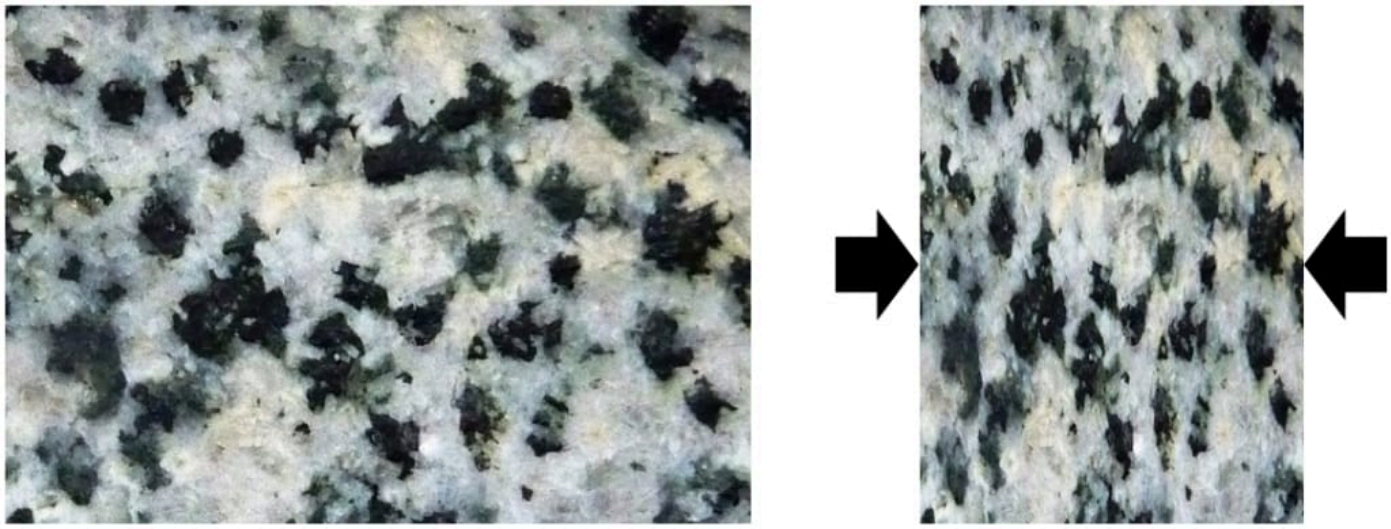


Figure 7.2.1 The textural effects of squeezing during metamorphism. In the original rock (left) there is no alignment of minerals. In the squeezed rock (right) the minerals have been elongated in the direction perpendicular to the squeezing.

When a rock is both heated and squeezed during metamorphism, and the temperature change is enough for new minerals to form from existing ones, there is a strong tendency for new minerals to grow with their long axes perpendicular to the direction of squeezing. This is illustrated in Figure 7.2.2, where the parent rock is shale, with bedding as shown. After both heating and squeezing, new minerals have formed within the rock, generally parallel to each other, and the original bedding has been largely obliterated.

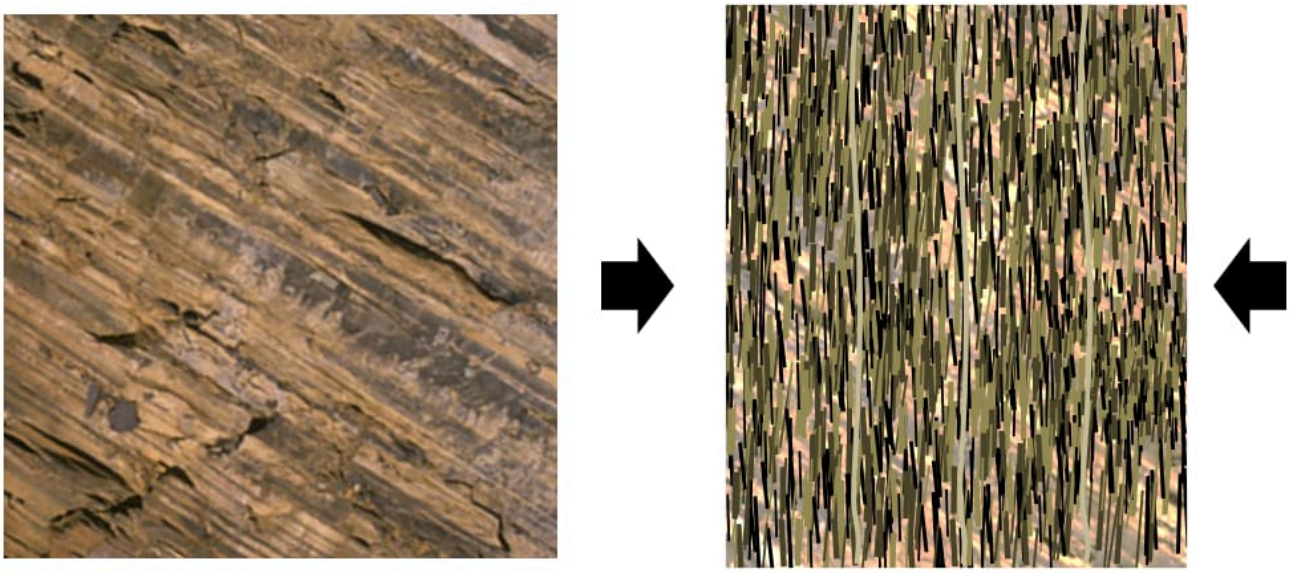


Figure 7.2.2 The textural effects of squeezing and mineral growth during regional metamorphism. The left diagram is shale with bedding slanting down to the right. The right diagram represents schist (derived from that shale), with mica crystals orientated perpendicular to the main stress direction and the original bedding no longer easily visible.

Figure 7.2.3 shows an example of this effect. This large boulder has bedding visible as dark and light bands sloping steeply down to the right. The rock also has a strong slaty foliation, which is horizontal in this view (parallel to the surface that the person is sitting on), and has developed because the rock was being squeezed during metamorphism. The rock has split from bedrock along this foliation plane, and you can see that other weaknesses are present in the same orientation.

Squeezing and heating alone (as shown in Figure 7.2.1) can contribute to foliation, but most foliation develops when new minerals are formed and are forced to grow perpendicular to the direction of greatest stress (Figure 7.2.2). This effect is especially strong if the new minerals are platy like mica or elongated like amphibole. The mineral crystals don't have to be large to produce foliation. Slate, for example, is characterized by aligned flakes of mica that are too small to see.



Figure 7.2.3 A slate boulder on the side of Mt. Wapta in the Rockies near Field, BC. Bedding is visible as light and dark bands sloping steeply to the right (yellow arrows). Slaty cleavage is evident from the way the rock has broken (along the flat surface that the person is sitting on) and also from lines of weakness that are parallel to that same trend (red arrows).

The various types of foliated metamorphic rocks, listed in order of the **grade** or intensity of metamorphism and the type of foliation are: **slate**, **phyllite**, **schist**, and **gneiss** (Figure 7.2.4). As already noted, slate is formed from the low-grade metamorphism of shale, and has microscopic clay and mica crystals that have grown perpendicular to the stress. Slate tends to break into flat

sheets. Phyllite is similar to slate, but has typically been heated to a higher temperature; the micas have grown larger and are visible as a sheen on the surface. Where slate is typically planar, phyllite can form in wavy layers. In the formation of schist, the temperature has been hot enough so that individual mica crystals are big enough to be visible, and other mineral crystals, such as quartz, feldspar, or garnet may also be visible. In gneiss, the minerals may have separated into bands of different colors. In the example shown in Figure 7.2.4d, the dark bands are largely amphibole while the light-colored bands are feldspar and quartz. Most gneiss has little or no mica because it forms at temperatures higher than those under which micas are stable. Unlike slate and phyllite, which typically only form from mudrock, schist, and especially gneiss, can form from a variety of parent rocks, including mudrock, sandstone, conglomerate, and a range of both volcanic and intrusive igneous rocks.

Schist and gneiss can be named on the basis of important minerals that are present. For example a schist derived from basalt is typically rich in the mineral chlorite, so we call it chlorite schist. One derived from shale may be a muscovite-biotite schist, or just a mica schist, or if there are garnets present it might be mica-garnet schist. Similarly, a gneiss that originated as basalt and is dominated by amphibole, is an amphibole gneiss or, more accurately, an **amphibolite**.



a) Slate, near to Golden, BC
b) Phyllite, location unknown
c) Schist, location unknown
d) Gneiss from the Victoria area, BC

Figure 7.2.4 Examples of foliated metamorphic rocks: (A) Slate, (B) Phyllite, (C) Schist, (D) Gneiss.

If a rock is buried to a great depth and encounters temperatures that are close to its melting point, it may partially melt. The resulting rock, which includes both metamorphosed and igneous material, is known as **migmatite** (Figure 7.2.5).



Figure 7.2.5 Migmatite from Prague, Czech Republic

As already noted, the nature of the parent rock controls the types of metamorphic rocks that can form from it under differing metamorphic conditions. The kinds of rocks that can be expected to form at different metamorphic grades from various parent rocks are listed in Table 7.1. Some rocks, such as granite, do not change much at the lower metamorphic grades because their minerals are still stable up to several hundred degrees.

Table 7.1 A rough guide to the types of metamorphic rocks that form from different parent rocks at different grades of regional metamorphism.

Parent Rock	Very Low Grade (150-300°C)	Low Grade (300-450°C)	Medium Grade (450-550°C)	High Grade (Above 550°C)
Mudrock	slate	phyllite	schist	gneiss
Granite	no change	no change	almost no change	granite gneiss
Basalt	chlorite schist	chlorite schist	amphibolite	amphibolite
Sandstone	no change	little change	quartzite	quartzite
Limestone	little change	marble	marble	marble

Metamorphic rocks that form under either low-pressure conditions or just confining pressure do not become foliated. In most cases, this is because they are not buried deeply, and the heat for the metamorphism comes from a body of magma that has moved into the upper part of the crust. This is **contact metamorphism**. Some examples of non-foliated metamorphic rocks are **marble**, **quartzite**, and **hornfels**.

Marble is metamorphosed limestone. When it forms, the calcite crystals tend to grow larger, and any sedimentary textures and fossils that might have been present are destroyed. If the original limestone was pure calcite, then the marble will likely be white (as in Figure 7.2.6), but if it had various impurities, such as clay, silica, or magnesium, the marble could be “marbled” in appearance. Marble that forms during regional metamorphism—and in fact that includes most marble—may or may not develop a foliated texture, but foliation is typically not easy to see in marble.

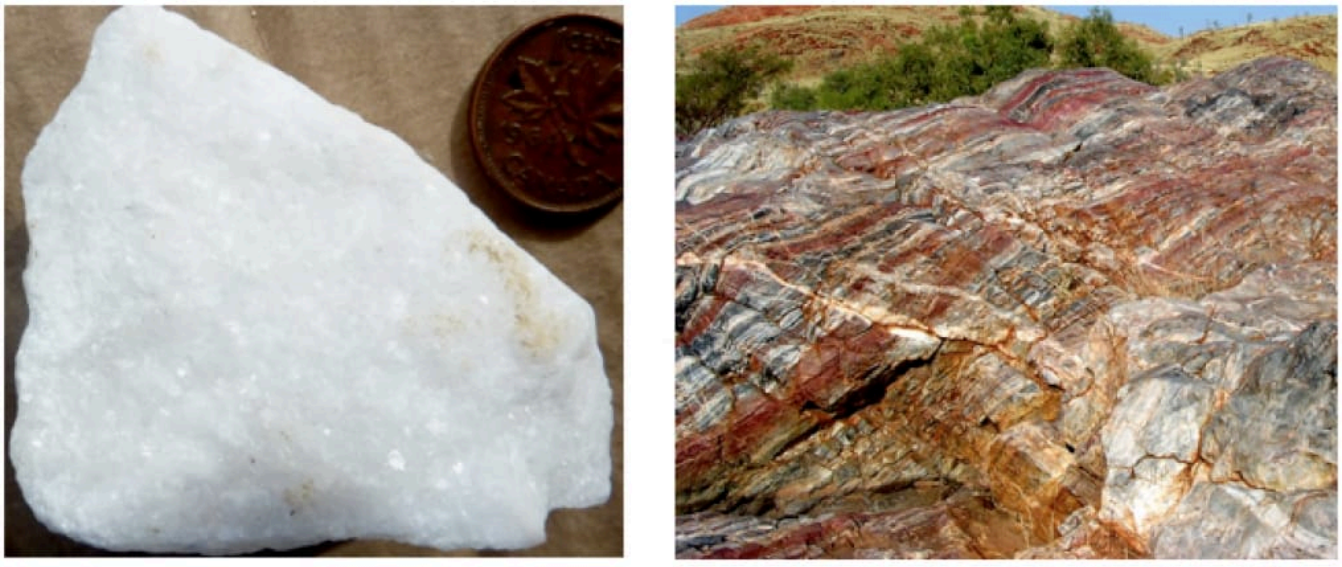


Figure 7.2.6 Marble with visible calcite crystals (left) and an outcrop of banded marble (right).

Quartzite is metamorphosed sandstone (Figure 7.2.7). It is dominated by quartz, and in many cases, the original quartz grains of the sandstone are welded together with additional silica. Most sandstone contains some clay minerals and may also include other minerals such as feldspar or fragments of rock, so most quartzite has some impurities with the quartz.



Figure 7.2.7 Quartzite from the Rocky Mountains, found in the Bow River at Cochrane, Alberta.

Even if formed during **regional metamorphism**, quartzite (like marble) does not tend to look foliated because quartz crystals don't align with the directional pressure. On the other hand, any clay present in the original sandstone is likely to be converted to mica during metamorphism, and any such mica is likely to align with the directional pressure. An example of this is shown in Figure 7.2.8. The quartz crystals show no alignment, but the micas are all aligned, indicating that there was directional pressure during regional metamorphism of this rock. Since these micas are very small, this rock would not appear to be foliated to the naked eye.

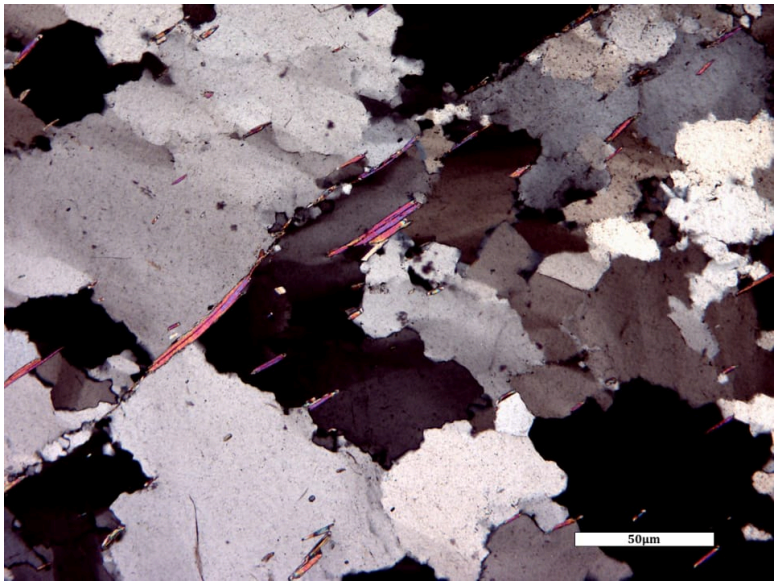


Figure 7.2.8 Magnified thin section of quartzite in polarized light. The irregular-shaped white, grey, and black crystals are all quartz. The small, thin, brightly colored crystals are mica. This rock is foliated, even though it might not appear to be if examined without a microscope, and so it must have formed under directed-pressure conditions.

Hornfels is another non-foliated metamorphic rock that normally forms during contact metamorphism of fine-grained rocks like mudstone or volcanic rock (Figure 7.2.9). In some cases, hornfels has visible crystals of minerals like biotite or andalusite. If the hornfels formed in a situation without directed pressure, then these minerals would be randomly orientated, not aligned with one-another, as they would be if formed with directed pressure.



Figure 7.2.9 Hornfels from the Novosibirsk region of Russia. The dark and light bands are bedding. The rock has been recrystallized during contact metamorphism and does not display foliation. (scale in centimeters).

Exercise 7.2 Naming metamorphic rocks

Provide reasonable names for the following metamorphic rocks based on the description:

1. A rock with visible crystals of mica and with small crystals of andalusite. The mica crystals are consistently parallel to one another.
2. A very hard rock with a granular appearance and a glassy lustre. There is no evidence of foliation.
3. A fine-grained rock that splits into wavy sheets. The surfaces of the sheets have a sheen to them.
4. A rock that is dominated by aligned crystals of amphibole.

See Appendix 3 for Exercise 7.2 answers.

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- Figure 7.2.5: Migmatite in Geopark on Albertov © Chmee2. CC BY.
- Figure 7.2.6 (right): An outcrop of banded marble by the USGS. Public domain.
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7.3: Plate Tectonics and Metamorphism

All of the important processes of metamorphism that we are familiar with can be understood in the context of geological processes related to plate tectonics. The relationships between plate tectonics and metamorphism are summarized in Figure 7.3.1, and in more detail in Figures 7.3.2, 7.3.3, 7.3.4, and 7.3.6.

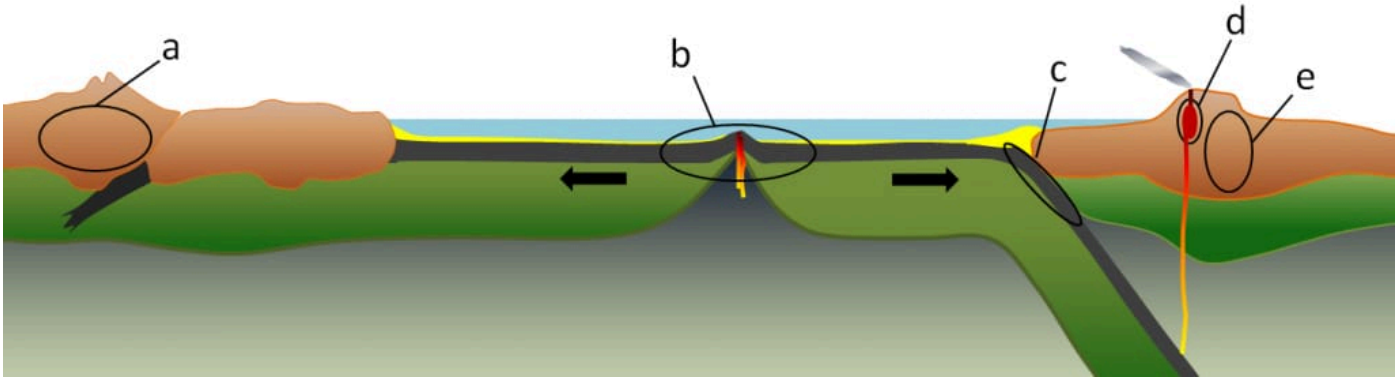


Figure 7.3.1 Environments of metamorphism in the context of plate tectonics: (a) regional metamorphism related to mountain building at a continent-continent convergent boundary, (b) regional metamorphism of oceanic crust in the area on either side of a spreading ridge, (c) regional metamorphism of oceanic crustal rocks within a subduction zone, (d) contact metamorphism adjacent to a magma body at a high level in the crust, and (e) regional metamorphism related to mountain building at a convergent boundary.

Most regional metamorphism takes place within the continental crust. While rocks can be metamorphosed at depth in most areas, the potential for metamorphism is greatest in the roots of mountain ranges where there is a strong likelihood for burial of relatively young sedimentary rock to great depths, as depicted in Figure 7.3.2. An example would be the Himalayan Range. At this continent-continent convergent boundary, sedimentary rocks have been both thrust up to great heights (nearly 9,000 meters above sea level) and also buried to great depths. Considering that the normal geothermal gradient (the rate of increase in temperature with depth) is around 30°C per kilometre, rock buried to 9 kilometers below sea level in this situation could be close to 18 kilometers below the surface of the ground, and it is reasonable to expect temperatures up to 500°C. Metamorphic rocks formed there are likely to be foliated because of the strong directional pressure (compression) of converging plates.

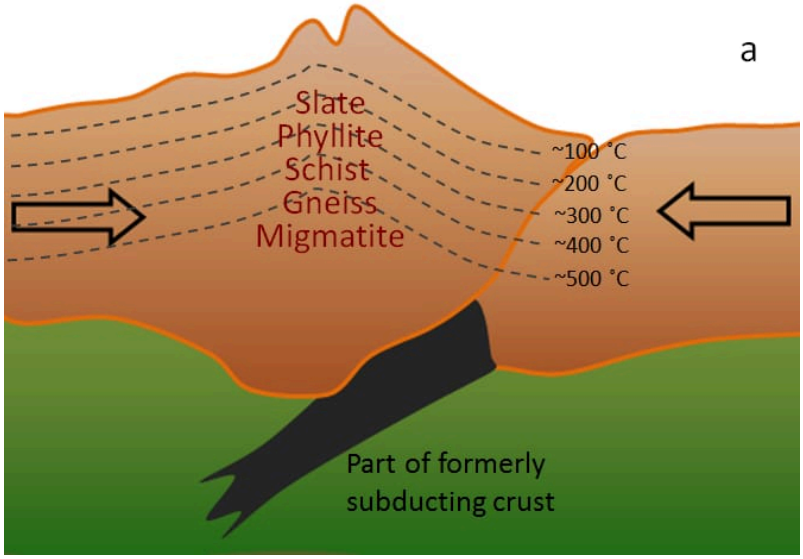


Figure 7.3.2 Regional metamorphism beneath a mountain range related to continent-continent collision (typical geothermal gradient). (Example: Himalayan Range)

At an oceanic spreading ridge, recently formed oceanic crust of gabbro and basalt is slowly moving away from the plate boundary

(Figure 7.3.3). Water within the crust is forced to rise in the area close to the source of volcanic heat, and this draws more water in from farther out, which eventually creates a convective system where cold seawater is drawn into the crust and then out again onto the sea floor near the ridge. The passage of this water through the oceanic crust at 200° to 300°C promotes metamorphic reactions that change the original pyroxene in the rock to chlorite and serpentine. Because this metamorphism takes place at temperatures well below the temperature at which the rock originally formed (~1200°C), it is known as **retrograde metamorphism**. The rock that forms in this way is known as **greenstone** if it isn't foliated, or **greenschist** if it is. Chlorite ($(Mg_5Al)(AlSi_3)O_{10}(OH)_8$) and serpentine ($(Mg, Fe)_3Si_2O_5(OH)_4$) are both "**hydrated minerals**" meaning that they have water (as OH) in their chemical formulas. When metamorphosed ocean crust is later subducted, the chlorite and serpentine are converted into new non-hydrous minerals (e.g., garnet and pyroxene) and the water that is released migrates into the overlying mantle, where it contributes to flux melting (Chapter 3, section 3.2).

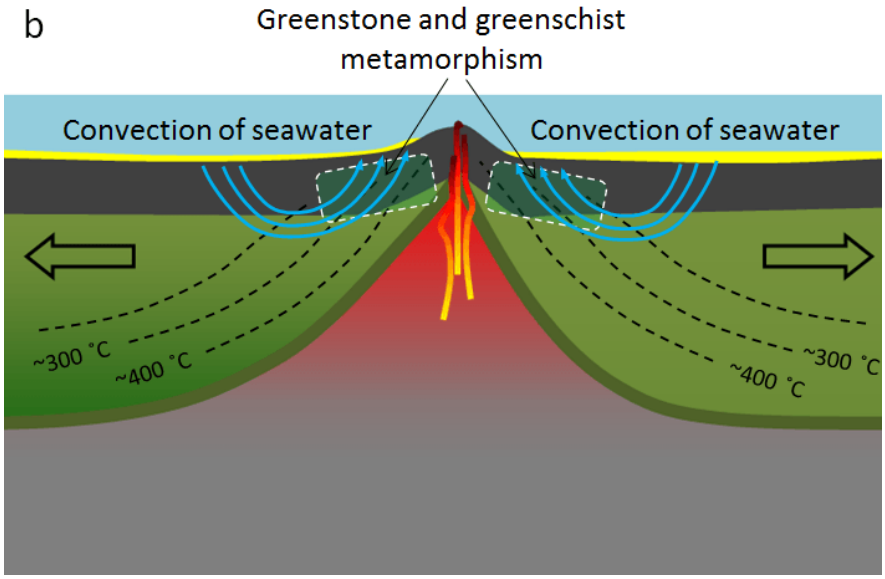


Figure 7.3.3 Regional metamorphism of oceanic crustal rock on either side of a spreading ridge. The dotted rectangles are the areas where metamorphism is taking place. (Example: Juan de Fuca spreading ridge)

At a subduction zone, oceanic crust is forced down into the hot mantle. But because the oceanic crust is now relatively cool, especially along its sea-floor upper surface, it does not heat up quickly, and the subducting rock remains several hundreds of degrees cooler than the surrounding mantle (Figure 7.3.4). A special type of metamorphism takes place under these very high-pressure but relatively low-temperature conditions, producing an amphibole mineral known as **glaucophane** ($Na_2(Mg_3Al_2)Si_8O_{22}(OH)_2$), which is blue in color, and is an important component of a rock known as **blueschist**.

You've probably never seen or even heard of blueschist; that's not surprising. What is a little surprising is that anyone has seen it! Most blueschist forms in subduction zones, continues to be subducted, turns into **eclogite** at about 35 kilometers depth, and then eventually sinks deep into the mantle—never to be seen again because that rock will eventually melt. In only a few places in the world, where the subduction process has been interrupted by some other tectonic process, has partially subducted blueschist rock returned to the surface. One such place is the area around San Francisco; the rock is known as the Franciscan Complex (Figure 7.3.5).

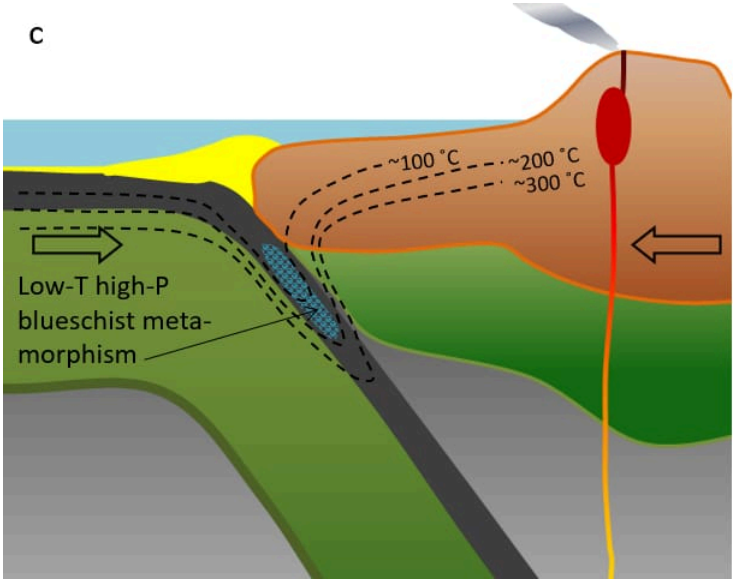


Figure 7.3.4 Regional metamorphism of oceanic crust at a subduction zone. (Example: Cascadia subduction zone. Rock of this type is exposed in the San Francisco area.)



Figure 7.3.5 Franciscan Complex blueschist rock exposed north of San Francisco. The blue color of rock is due to the presence of the amphibole mineral glaucophane.

Magma is produced at convergent boundaries and rises toward the surface, where it can form magma bodies in the upper part of the crust. Such magma bodies, at temperatures of around 1000°C, heat up the surrounding rock, leading to contact metamorphism (Figure 7.3.6). Because this happens at relatively shallow depths, in the absence of directed pressure, the resulting rock does not normally develop foliation. The zone of contact metamorphism around an intrusion is very small (typically meters to tens of meters) compared with the extent of regional metamorphism in other settings (tens of thousands of square kilometers).

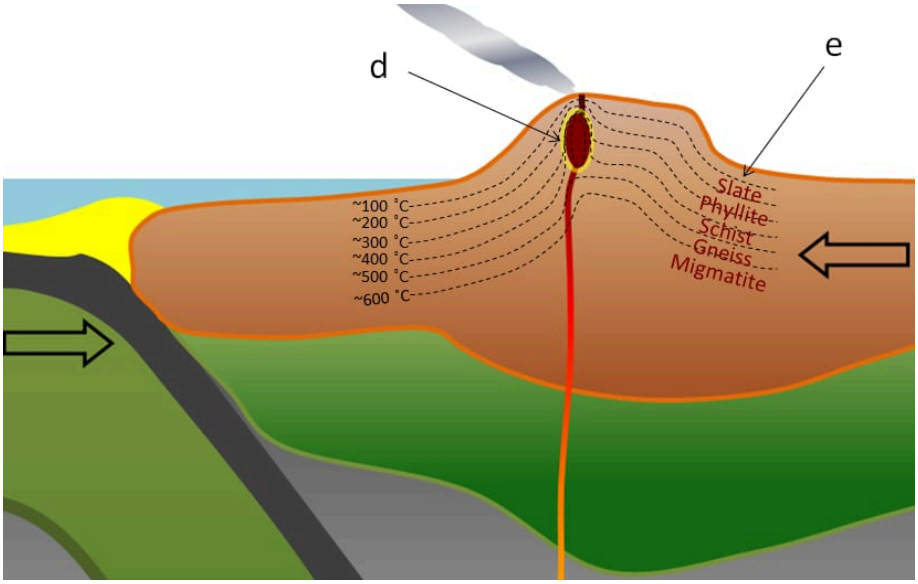


Figure 7.3.6 d: Contact metamorphism around a high-level crustal magma chamber (Example: the magma chamber beneath Mt. St. Helens.) e: Regional metamorphism in a volcanic-arc related mountain range (volcanic-region temperature gradient) (Example: The southern part of the Coast Range, B.C.)

Regional metamorphism also takes place within volcanic-arc mountain ranges, and because of the extra heat associated with the volcanism, the geothermal gradient is typically a little steeper in these settings (somewhere between 40° and 50°C per kilometre). As a result higher grades of metamorphism can take place closer to surface than is the case in other areas (Figure 7.3.6).

Another way to understand metamorphism is by using a diagram that shows temperature on one axis and depth—which is equivalent to pressure—on the other (Figure 7.3.7). The three heavy dotted lines on this diagram represent Earth's geothermal gradients under different conditions. In most areas, the rate of increase in temperature with depth is 30°C per kilometre. In other words, if you go 1,000 meters down into a mine, the temperature will be roughly 30°C warmer than the average temperature at the surface. In most parts of southern Canada, the average surface temperature is about 10°C, so at a 1,000 metre depth, it will be about 40°C. That's uncomfortably hot, so deep mines must have effective ventilation systems. This typical geothermal gradient is shown by the green dotted line in Figure 7.3.7. At a 10 kilometre depth, the temperature is about 300°C and at 20 kilometers it's about 600°C.

In volcanic areas, the geothermal gradient is more like 40° to 50°C per kilometre, so the temperature at a 10 kilometre depth is in the 400° to 500°C range. Along subduction zones, as described above, the cold oceanic crust keeps temperatures low, so the gradient is typically less than 10°C per kilometre. The various types of metamorphism described above are represented in Figure 7.3.7 with the same letters (a through e) used in Figures 7.3.1 to 7.3.4 and 7.3.6.

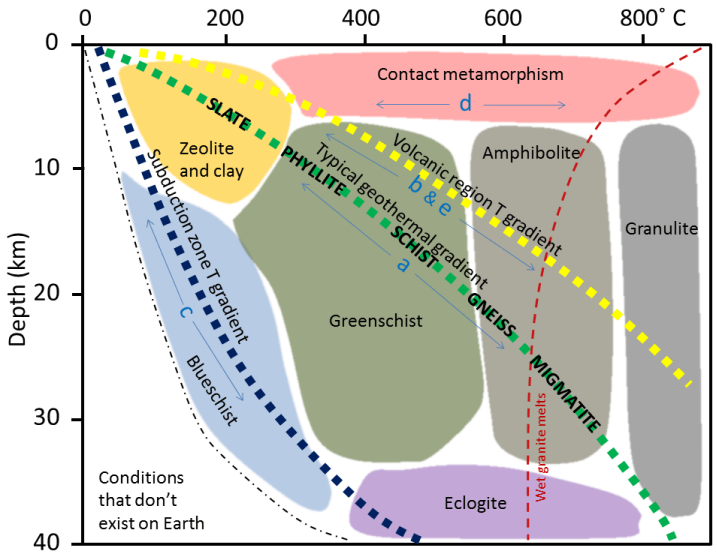


Figure 7.3.7 Types of metamorphism shown in the context of depth and temperature under different conditions. The metamorphic rocks formed from mudrock under regional metamorphosis with a typical geothermal gradient are listed. The letters a through e correspond with those shown in Figures 7.3.1 to 7.3.4 and 7.3.6.

By way of example, if we look at regional metamorphism in areas with typical geothermal gradients, we can see that burial in the 5 kilometre to 10 kilometre range puts us in the zeolite[1] and clay mineral zone (see Figure 7.3.7), which is equivalent to the formation of slate. At 10 to 15 kilometers, we are in the greenschist zone (where chlorite would form in mafic volcanic rock) and very fine micas form in mudrock, to produce phyllite. At 15 to 20 kilometers, larger micas form to produce schist, and at 20 to 25 kilometers amphibole, feldspar, and quartz form to produce gneiss. Beyond a depth of 25 kilometers in this setting, we cross the partial melting line for granite (or gneiss) with water present, and so we can expect migmatite to form.

Exercise 7.3 Metamorphic rocks in areas with higher geothermal gradients

Figure 7.3.7 shows the types of rock that might form from mudrock at various points along the curve of the "typical" geothermal gradient (dotted green line). Looking at the geothermal gradient for volcanic regions (dotted yellow line in Figure 7.3.7), estimate the depths at which you would expect to find the same types of rock forming from a mudrock parent.

1. Slate
2. Phyllite
3. Schist
4. Gneiss
5. Migmatite

See Appendix 3 for Exercise 7.3 answers.

Image Descriptions

Figure 7.3.2 image description: Regional metamorphism occurring beneath a mountain range due to continent-continent collision. The typical geothermal gradient for slate is 100°C, for phyllite 200°C, for schist 300°C, for gneiss °C, for migmatite 500°C. [Return to Figure 7.3.2]

Media Attributions

- Figures 7.3.1, 7.3.2, 7.3.3, 7.3.4, 7.3.5, 7.3.6, 7.3.7: © Steven Earle. CC BY.

1. Zeolites are silicate minerals that typically form during low-grade metamorphism of volcanic rocks. ↵

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7.4: Regional Metamorphism

As described above, regional metamorphism occurs when rocks are buried deep in the crust. This is commonly associated with convergent plate boundaries and the formation of mountain ranges. Because burial to 10 to 20 kilometers is required, the areas affected tend to be large—thousands of square kilometers.

Rather than focusing on metamorphic rock textures (slate, schist, gneiss, etc.), geologists tend to look at specific minerals within the rocks that are indicative of different grades of metamorphism. Some common minerals in metamorphic rocks are shown in Figure 7.4.1, arranged in order of the temperature ranges over which they tend to be stable. The upper and lower limits of the ranges are intentionally vague because these limits depend on a number of different factors, such as the pressure, the amount of water present, and the overall composition of the rock.

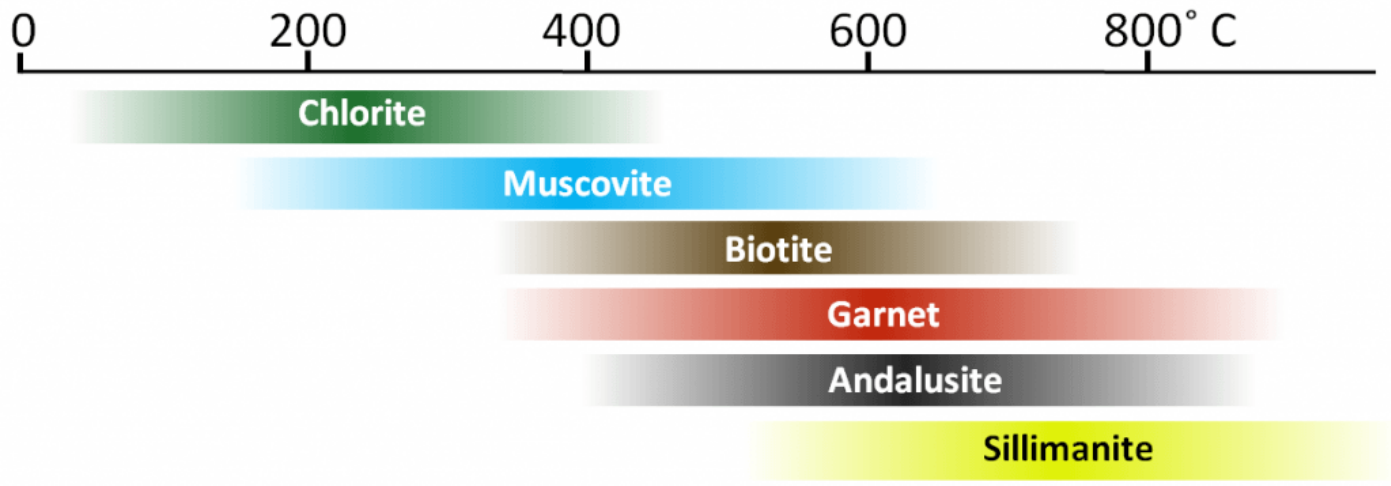


Figure 7.4.1 Metamorphic index minerals and their approximate temperature ranges

The southern and southwestern parts of Nova Scotia were regionally metamorphosed during the Devonian Acadian Orogeny (around 400 Ma), when a relatively small continental block (the Meguma **Terrane**[1]) was pushed up against the existing eastern margin of North America. As shown in Figure 7.4.2, clastic sedimentary rocks within this terrane were variably metamorphosed, with the strongest metamorphism in the southwest (the sillimanite zone), and progressively weaker metamorphism toward the east and north. The rocks of the sillimanite zone were likely heated to over 700°C, and therefore must have buried to depths between 20 and 25 kilometers. The surrounding lower-grade rocks were not buried as deep, and the rocks within the peripheral chlorite zone were likely not buried to more than about 5 kilometers.

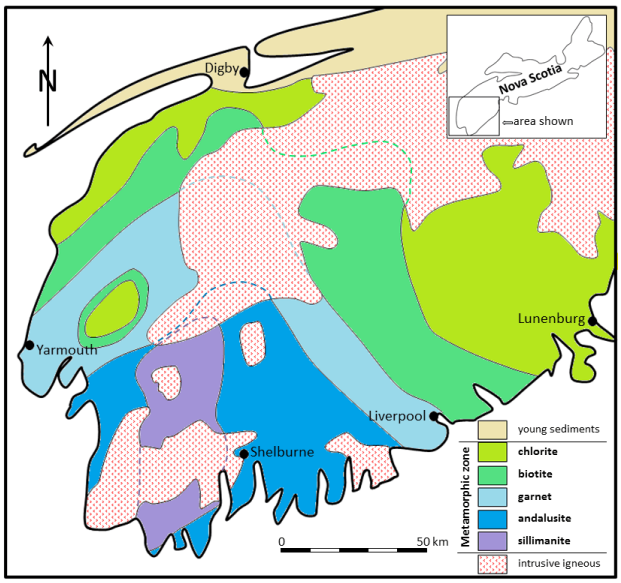


Figure 7.4.2 Regional metamorphic zones in the Meguma Terrane of southwestern Nova Scotia.

A probable explanation for this pattern is that the area with the highest-grade rocks was buried beneath the central part of a mountain range formed by the collision of the Meguma Terrane with North America. As is the case with all mountain ranges, the crust became thickened as the mountains grew, and it was pushed farther down into the mantle than the surrounding crust. This happens because Earth's crust is floating on the underlying mantle—and that is known as an isostatic relationship. As the formation of mountains adds weight, the crust in that area sinks farther down into the mantle to compensate for the added weight. The likely pattern of metamorphism in this situation is shown in cross-section in Figure 7.4.3a. The mountains were eventually eroded (over tens of millions of years), allowing the crust to rebound upward, thus exposing the metamorphic rock (Figure 7.4.3b).

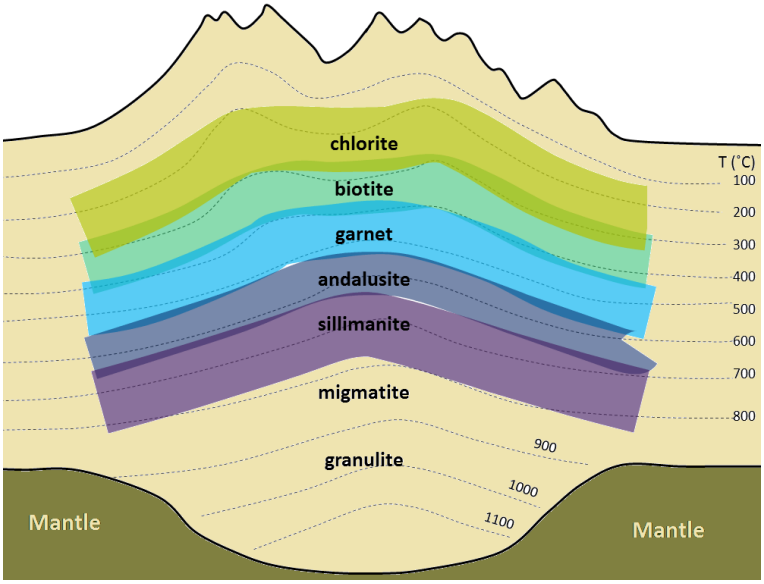


Figure 7.4.3a Schematic cross-section through the Meguma Terrane during the Devonian period. The crust is thickened underneath the mountain range to compensate for the added weight of the mountains above and has sunk into the mantle.

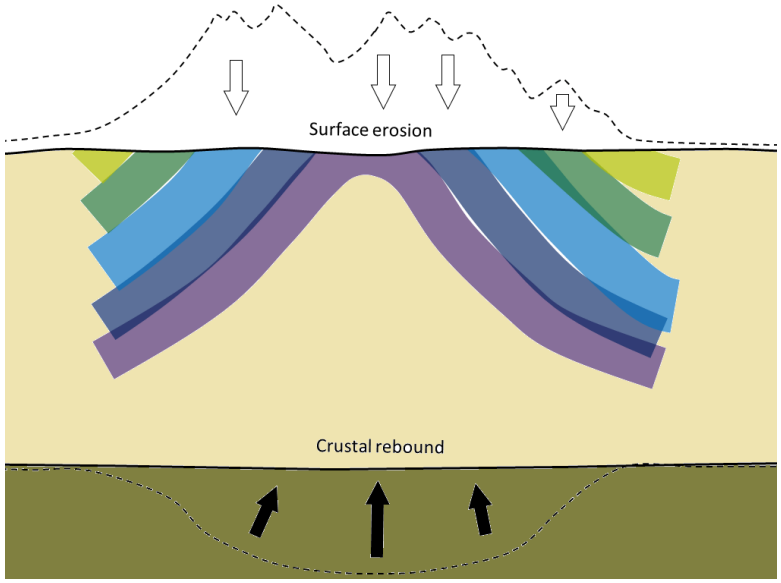


Figure 7.4.3b Schematic present-day cross-section through the Meguma Terrane. The mountains have been eroded. As they lost mass the base of the crust gradually rebounded, pushing up the core of the metamorphosed region so that the once deeply buried metamorphic zones are now exposed at surface.

The metamorphism in Nova Scotia's Meguma Terrane is just one example of the nature of regional metamorphism. Obviously many different patterns of regional metamorphism exist, depending on the parent rocks, the geothermal gradient, the depth of burial, the pressure regime, and the amount of time available. The important point is that regional metamorphism happens only at significant depths. The greatest likelihood of attaining those depths, and then having the once-buried rocks eventually exposed at the surface, is where mountain ranges existed and have since been largely eroded away. As this happens typically at convergent plate boundaries, directed pressures can be strong, and regionally metamorphosed rocks are almost always foliated.

Exercise 7.4 Scottish metamorphic zones

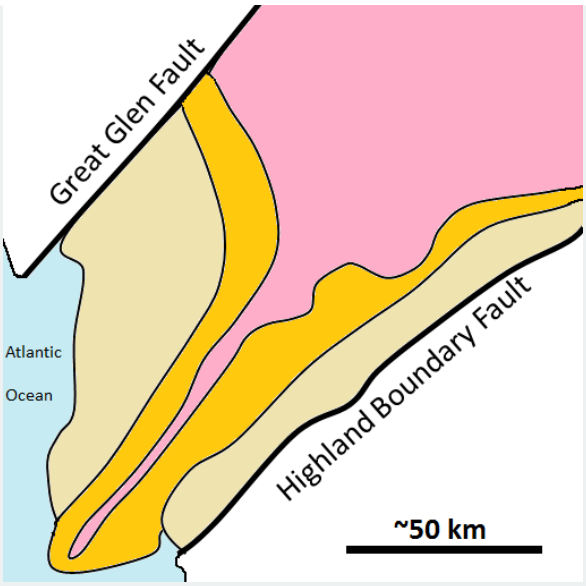


Figure 7.4.4

The map shown here represents the part of western Scotland between the Great Glen Fault and the Highland Boundary Fault. The shaded areas are metamorphic rock, and the three metamorphic zones represented are garnet, chlorite, and biotite.

Label the three colored areas of the map with the appropriate zone names (garnet, chlorite, and biotite). Hint: refer to Figure 7.4.1 above to work out which of these zones might represent the peripheral area of low-grade metamorphism, and which might

represent the core area of higher-grade metamorphism.

Indicate which part of the region was likely to have been buried the deepest during metamorphism.

British geologist George Barrow studied this area in the 1890s and was the first person anywhere to map metamorphic zones based on their mineral assemblages. This pattern of metamorphism is sometimes referred to as “Barrovian.”

See Appendix 3 for Exercise 7.4 answers.

Image Descriptions

Figure 7.4.1 **image description:** Approximate temperature range of metamorphic minerals: Chlorite, 50 to 450°C. Muscovite, 175 to 625°C. Biotite, 350 to 725°C. Garnet, 375 to 900°C. Andalusite, 400 to 850°C. Sillimanite, 575 to 1000°C. [Return to Figure 7.4.1]

Media Attributions

- Figure 7.4.1, 7.4.3ab, 7.4.4: © Steven Earle. CC BY.
- Figure 7.4.2: Image edited by Steven Earle, after Keppie, D, and Muecke, G, 1979, Metamorphic map of Nova Scotia, N.S. Dept. of Mines and Energy, Map 1979-006., and from White, C and Barr, S., 2012, Meguma Terrane revisted, Stratigraphy, metamorphism, paleontology and provenance, Geoscience Canada, V. 39, No.1.

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1. No, it’s not a spelling mistake! A terrane is a distinctive block of crust that is now part of a continent, but is thought to have come from elsewhere, and was added on by plate-tectonic processes. ↵

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7.5: Contact Metamorphism and Hydrothermal Processes

Contact metamorphism takes place where a body of magma intrudes into the upper part of the crust. Any type of magma body can lead to contact metamorphism, from a thin dyke to a large stock. The type and intensity of the metamorphism, and width of the metamorphic **aureole** will depend on a number of factors, including the type of country rock, the temperature of the intruding body and the size of the body (Figure 7.5.1). A large intrusion will contain more thermal energy and will cool much more slowly than a small one, and therefore will provide a longer time and more heat for metamorphism. That will allow the heat to extend farther into the country rock, creating a larger aureole.

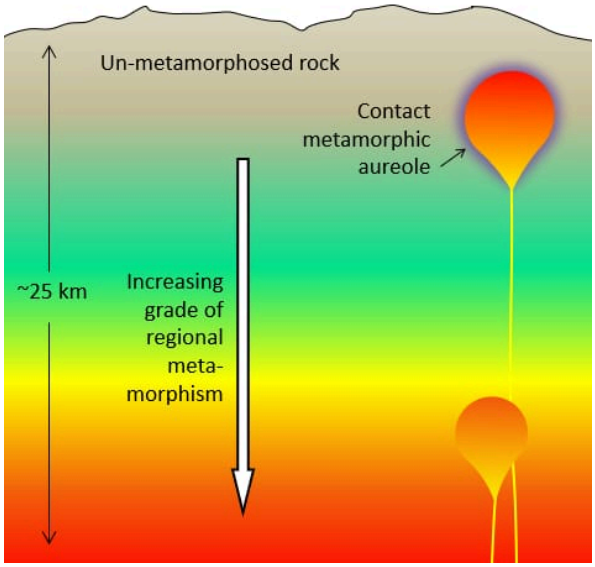


Figure 7.5.1 Schematic cross-section of the middle and upper crust showing two magma bodies. The upper body has intruded into cool unmetamorphosed rock near to the surface and has created a zone of contact metamorphism. The lower body is surrounded by rock that is already hot (and probably already metamorphosed), and so it does not have a significant metamorphic aureole.

Contact metamorphic aureoles are typically quite small, from just a few centimeters around small dykes and sills, to several 10s of meters around a large stock. As was shown in Figure 7.5.7, contact metamorphism can take place over a wide range of temperatures—from around 300° to over 800°C—and of course the type of metamorphism, and new minerals formed, will vary accordingly. The nature of the country rock (or parent rock) is also important.

A hot body of magma in the upper crust can create a very dynamic situation that may have geologically interesting and economically important implications. In the simplest cases, water does not play a big role, and the main process is transfer of heat from the pluton to the surrounding rock, creating a zone of contact metamorphism (Figure 7.5.2a). In that situation mudrock or volcanic rock will likely be metamorphosed to hornfels (Figure 7.5.9), limestone will be metamorphosed to marble (Figure 7.5.6), and sandstone to quartzite (Figure 7.5.7). (But don't forget that marble and quartzite can also form during regional metamorphism!)

In many cases, however, water is released from the magma body as crystallization takes place, and this water is dispersed along fractures in the country rock (Figure 7.5.2b). The water released from a magma chamber is typically rich in dissolved minerals. As this water cools, is chemically changed by the surrounding rocks, or boils because of a drop in pressure, minerals are deposited, forming veins within the fractures in the country rock. Quartz veins are common in this situation, and they might also include pyrite, hematite, calcite, and even silver and gold.

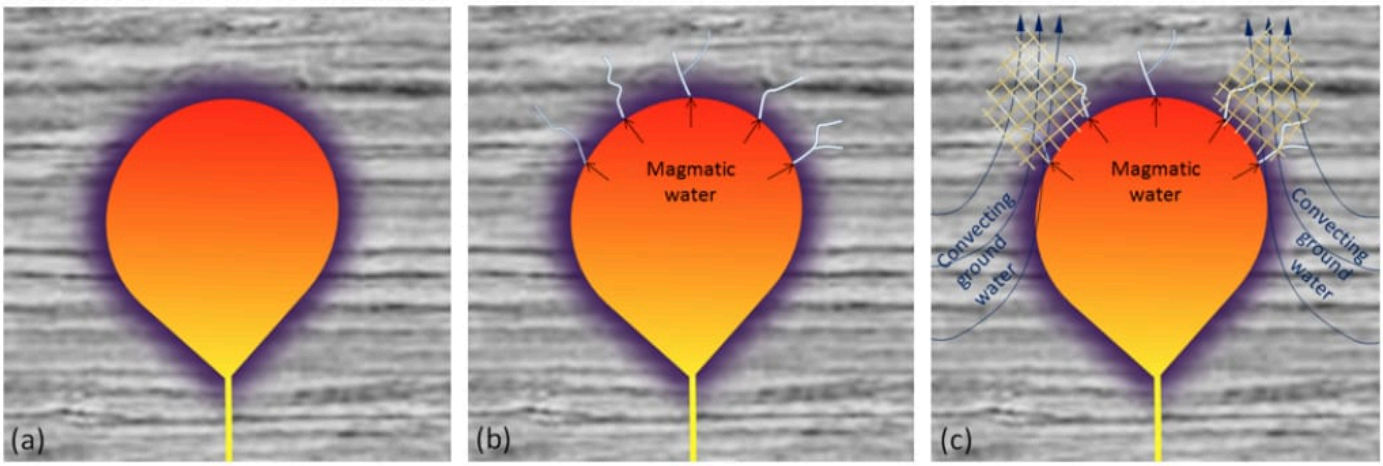


Figure 7.5.2 Depiction of metamorphism and alteration around a pluton in the upper crust. (a) Thermal metamorphism only (within the purple zone). (b) Thermal metamorphism plus veining (white) related to dispersal of magmatic fluids into the overlying rock. (c) Thermal metamorphism plus veining from magmatic fluids plus alteration and possible formation of metallic minerals (hatched yellow areas) from convection of groundwater.

Heat from the magma body will heat the surrounding groundwater, causing it to expand and then rise toward the surface. In some cases, this may initiate a convection system where groundwater circulates past the pluton. Such a system could operate for many thousands of years, resulting in the circulation of billions of litres of groundwater from the surrounding region past the pluton. Hot water circulating through the rocks can lead to significant changes in the mineralogy of the rock, including alteration of feldspars to clays, and deposition of quartz, calcite, and other minerals in fractures and other open spaces (Figure 7.5.3). As with the magmatic fluids, the nature of this circulating groundwater can also change adjacent to, or above, the pluton, resulting in deposition of other minerals, including ore minerals. Metamorphism in which much of the change is derived from fluids passing through the rock is known as **metasomatism**. When hot water contributes to changes in rocks, including mineral alteration and formation of veins, it is known as **hydrothermal alteration**.



Figure 7.5.3 Calcite veins in limestone of the Comox Formation, Nanaimo, B.C

A special type of metasomatism can take place where a hot pluton intrudes into carbonate rock such as limestone. When magmatic fluids rich in silica, calcium, magnesium, iron, and other elements flow through the carbonate rock, their chemistry can change dramatically, resulting in the deposition of minerals that would not normally exist in either the igneous rock or limestone. These include garnet, epidote (another silicate), magnetite, pyroxene, and a variety of copper and other minerals (Figure 7.5.4). This type of metamorphism is known as **skarn**, and again, some important types of mineral deposits can form this way.

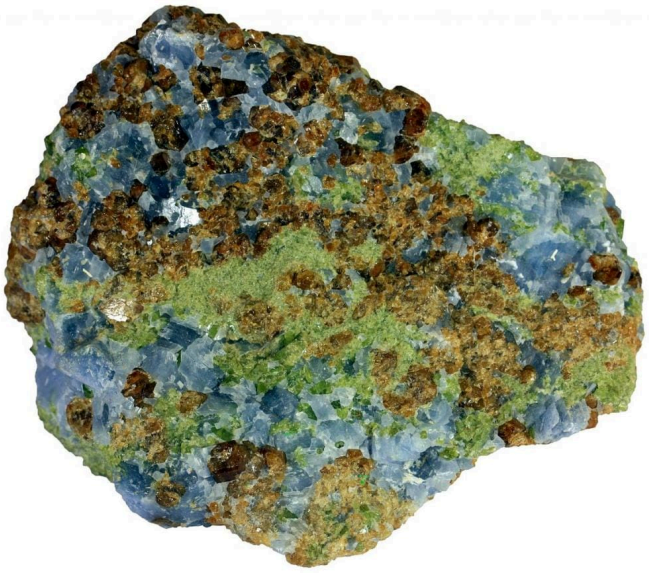


Figure 7.5.4 A skarn rock from Mount Monzoni, Northern Italy, with recrystallized calcite (blue), garnet (brown), and pyroxene (green). The rock is 6 centimeters across.

Exercise 7.5 Contact metamorphism and metasomatism

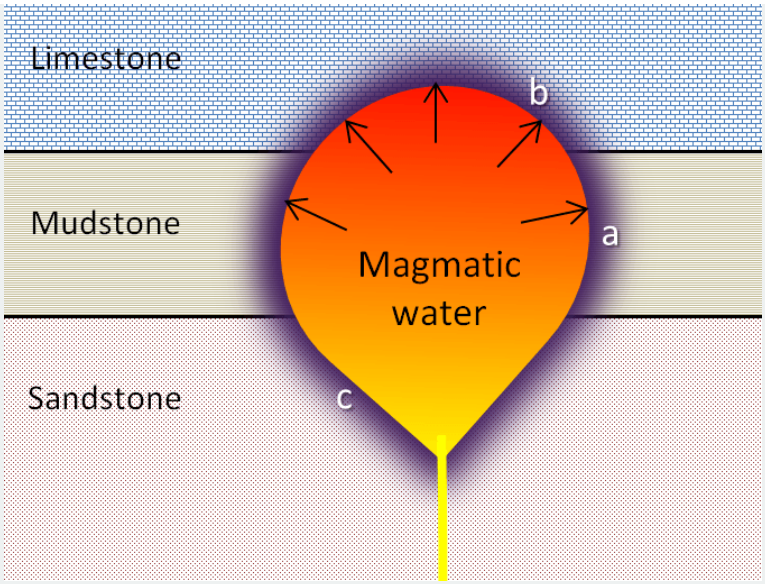


Figure 7.5.5

This diagram shows a pluton that has intruded into a series of sedimentary rocks. What type of metamorphic rock would you expect to see at locations:

a) Mudstone? _____

b) Limestone? _____

c) Sandstone? _____

See Appendix 3 for Exercise 7.5 answers.

Media Attributions

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- Figure 7.5.4: 00031 6 cm grossular calcite augite skarn © Siim. CC BY SA.

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7.6: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
7.1 Controls Over Metamorphic Processes	Metamorphism is controlled by five main factors: the composition of the parent rock, the temperature to which the rock is heated, the amount and type of pressure, the volumes and compositions of aqueous fluids that are present, and the amount of time available for metamorphic reactions to take place.
7.2 Classification of Metamorphic Rocks	Metamorphic rocks are classified on the basis of texture and mineral composition. Foliation is a key feature of metamorphic rocks formed under directed pressure. Foliated metamorphic rocks include slate, phyllite, schist, and gneiss. Metamorphic rocks formed in environments without strong directed pressure include hornfels, marble, and quartzite, although the latter two may form in high-pressure situations but not develop obvious foliated textures.
7.3 Plate Tectonics and Metamorphism	Almost all metamorphism can be explained by plate-tectonic processes. Oceanic crustal rock can be metamorphosed near the spreading ridge where it was formed, but most other regional metamorphism takes place in areas where mountain ranges have been created, which are most common at convergent boundaries. Contact metamorphism takes place around magma bodies in the upper part of the crust, which are also most common above convergent boundaries.
7.4 Regional Metamorphism	Geologists classify metamorphic rocks based on some key minerals—such as chlorite, garnet, andalusite, and sillimanite—that form at specific temperatures and pressures. Most regional metamorphism takes place beneath mountain ranges because the crust becomes thickened and rocks are pushed down to great depths because of the isostatic relationship between the crust and mantle. When mountains erode, those metamorphic rocks are uplifted by crustal rebound.
7.5 Contact Metamorphism and Hydrothermal Processes	Contact metamorphism takes place around magma bodies that have intruded into cool rocks at high levels in the crust. Heat from the magma is transferred to the surrounding country rock, resulting in mineralogical and textural changes. Water from a cooling body of magma, or from convection of groundwater produced by the heat of the pluton, can also lead to metasomatism, hydrothermal alteration, and accumulation of valuable minerals in the surrounding rocks.

Questions for Review

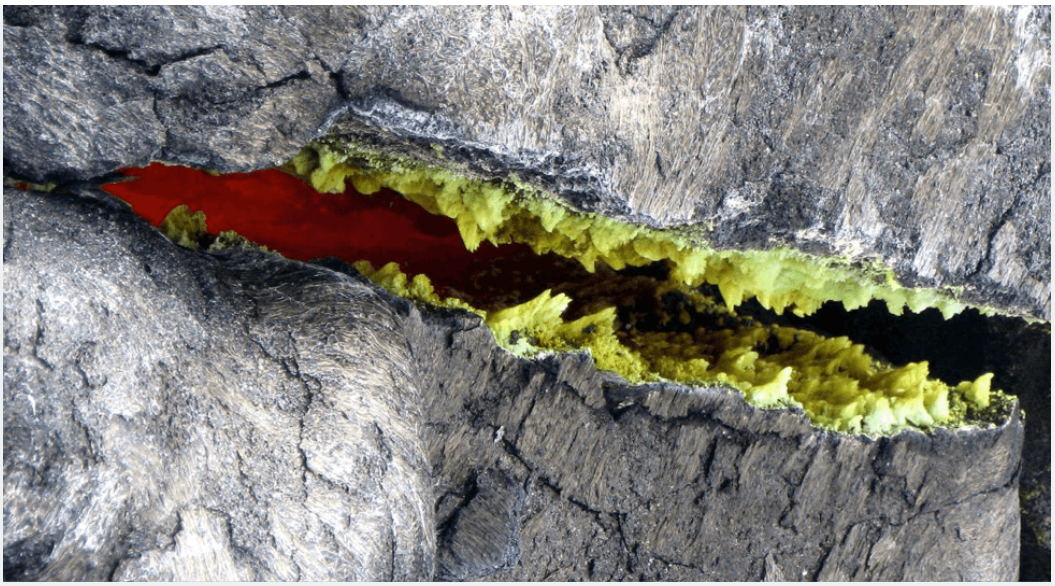
Answers to Review Questions can be found in Appendix 2.

1. What are the two main agents of metamorphism, and what are their respective roles in producing metamorphic rocks?
2. Into what metamorphic rocks will a mudrock be transformed at very low, low, medium, and high metamorphic grades?
3. Why doesn't granite change very much at lower metamorphic grades?
4. Describe the main process of foliation development in a metamorphic rock such as schist.
5. What process contributes to metamorphism of oceanic crust at a spreading ridge?
6. How do variations in the geothermal gradient affect the depth at which different metamorphic rocks form?
7. Blueschist metamorphism takes place within subduction zones. What are the unique temperature and pressure characteristics of this geological setting?
8. Rearrange the following minerals in order of increasing metamorphic grade: biotite, garnet, sillimanite, chlorite.
9. Why does contact metamorphism not normally take place at significant depth in the crust?
10. What is the role of magmatic fluids in metamorphism that takes place adjacent to a pluton?
11. How does metasomatism differ from regional metamorphism?
12. How does the presence of a hot pluton contribute to the circulation of groundwater that facilitates metasomatism and hydrothermal processes?
13. What must be present in the country rock to produce a skarn?

14. Two things that a geologist first considers when looking at a metamorphic rock are what the parent rock might have been, and what type of metamorphism has taken place. This can be difficult to do, even if you have the actual rock in your hand, but give it a try for the following metamorphic rocks:
    a. Chlorite schist
    b. Slate
    c. Mica-garnet schist
    d. Amphibolite
    e. Marble

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7.7: Review of Minerals and Rocks



Crystals of the mineral native sulfur growing on the rock basalt at an outlet of volcanic gases, Kilauea volcano, Hawaii

Now that we've covered minerals and all three types of rocks it's important for you to convince yourself that you've got them straight in your mind. As already noted, one of the most common mistakes that geology students make on assignments, tests and exams is to confuse minerals with rocks and then give a wrong answer when asked to name one or the other based on information provided.

In this exercise you are given a list of names of minerals and rocks and asked to determine which ones are minerals and which are rocks. For those that you think are minerals you should then indicate which mineral group it belongs to (e.g., *oxide, sulphate, silicate, carbonate, halide* etc.). For those that you think are rocks, you should describe what type of rock it is (e.g., *intrusive igneous, extrusive igneous, clastic sedimentary, chemical sedimentary, foliated metamorphic* and *non-foliated metamorphic)*. The answers can be found in Rock and mineral review exercise answers in Appendix 3.

Mineral or rock name	Rock or mineral?	If it's a mineral, which group does it belong to? If it's a rock, what type is it?
Feldspar		
Calcite		
Slate		
Hematite		
Rhyolite		
Sandstone		
Diorite		
Olivine		
Pyrite		
Quartzite		
Granite		

Amphibole		
Conglomerate		
Chert		
Halite		
Gneiss		
Mica		
Pyroxene		
Chlorite		
Limestone		
Andesite		

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8: MEASURING GEOLOGICAL TIME

8.1: The Geological Time Scale
8.2: Relative Dating Methods
8.3: Dating Rocks Using Fossils
8.4: Isotopic Dating Methods
8.5: Other Dating Methods
8.6: Understanding Geological Time
8.7: Summary

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8.1: The Geological Time Scale

William Smith worked as a surveyor in the coal-mining and canal-building industries in southwestern England in the late 1700s and early 1800s. While doing his work, he had many opportunities to look at the Paleozoic and Mesozoic sedimentary rocks of the region, and he did so in a way that few had done before. Smith noticed the textural similarities and differences between rocks in different locations, and more importantly, he discovered that fossils could be used to correlate rocks of the same age. Smith is credited with formulating the **principle of faunal succession** (the concept that specific types of organisms lived during different time intervals), and he used it to great effect in his monumental project to create a geological map of England and Wales, published in 1815. For more on William Smith, including a large-scale digital copy of the famous map, see the William Smith Wikipedia page.

Inset into Smith's great geological map is a small diagram showing a schematic geological cross-section extending from the Thames estuary of eastern England all the way to the west coast of Wales. Smith shows the sequence of rocks, from the Paleozoic rocks of Wales and western England, through the Mesozoic rocks of central England, to the Cenozoic rocks of the area around London (Figure 8.1.1). Although Smith did not put any dates on these—because he didn't know them—he was aware of the **principle of superposition** (the idea, developed much earlier by the Danish theologian and scientist Nicholas Steno, that young sedimentary rocks form on top of older ones), and so he knew that this diagram represented a stratigraphic column. And because almost every period of the Phanerozoic is represented along that section through Wales and England, it is a primitive geological time scale.

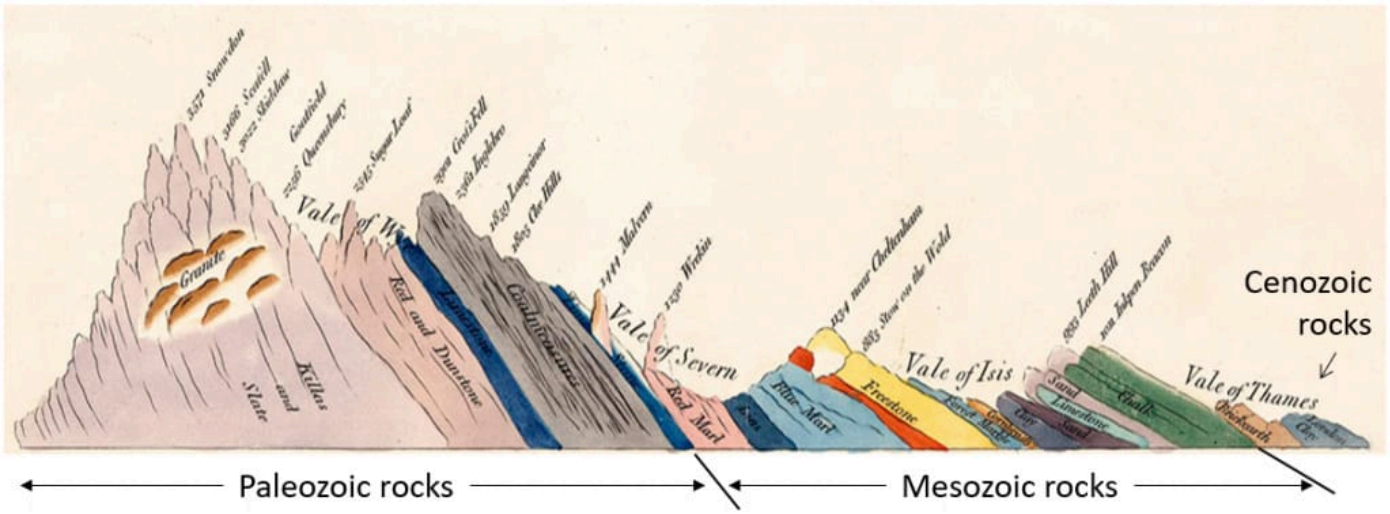


Figure 8.1.1 William Smith's "Sketch of the succession of strata and their relative altitudes," an inset on his geological map of England and Wales (with era names added).

Smith's work set the stage for the naming and ordering of the geological periods, which was initiated around 1820, first by British geologists, and later by other European geologists. Many of the periods are named for places where rocks of that age are found in Europe, such as Cambrian for Cambria (Wales), Devonian for Devon in England, Jurassic for the Jura Mountains in France and Switzerland, and Permian for the Perm region of Russia. Some are named for the type of rock that is common during that age, such as Carboniferous for the coal- and carbonate-bearing rocks of England, and Cretaceous for the chalks of England and France.

The early time scales were only relative because 19th century geologists did not know the ages of the rocks. That information was not available until the development of isotopic dating techniques early in the 20th century.

The geological time scale is currently maintained by the International Commission on Stratigraphy (ICS), which is part of the International Union of Geological Sciences. The time scale is continuously being updated as we learn more about the timing and nature of past geological events. You can view the ICS time scale online. It would be a good idea to print a copy (in color) to put on your wall while you are studying geology.

Geological time has been divided into four eons: Hadean (4570 to 4850 Ma), Archean (3850 to 2500 Ma), Proterozoic (2500 to 540 Ma), and Phanerozoic (540 Ma to present). As shown in Figure 8.1.2, the first three of these represent almost 90% of Earth's

history. The last one, the Phanerozoic (meaning “visible life”), is the time that we are most familiar with because Phanerozoic rocks are the most common on Earth, and they contain evidence of the life forms that we are familiar with to varying degrees.

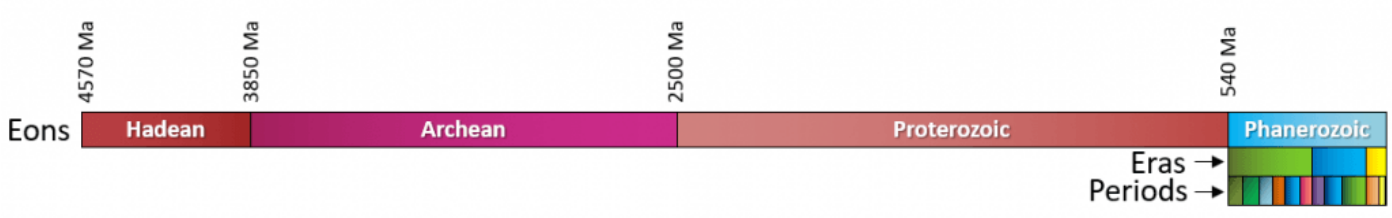


Figure 8.1.2 The four eons of Earth’s history.

The Phanerozoic eon—the past 540 Ma of Earth’s history—is divided into three eras: the Paleozoic (“early life”), the Mesozoic (“middle life”), and the Cenozoic (“new life”), and each of these is divided into a number of periods (Figure 8.1.3). Most of the organisms that we share Earth with evolved at various times during the Phanerozoic.

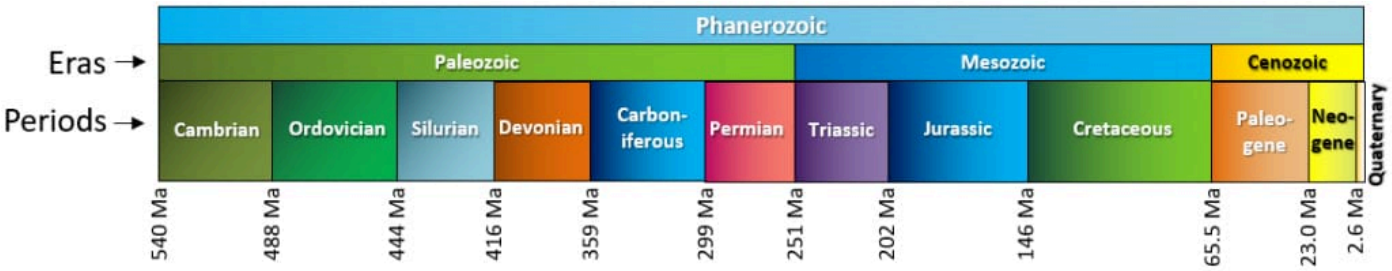


Figure 8.1.3 The eras (middle row) and periods (bottom row) of the Phanerozoic eon.

The Cenozoic era, which represents the past 65.5 Ma, is divided into three periods: Paleogene, Neogene, and Quaternary, and seven epochs (Figure 8.1.4). Dinosaurs became extinct at the start of the Cenozoic, after which birds and mammals radiated to fill the available habitats. Earth was very warm during the early Eocene and has steadily cooled ever since. Glaciers first appeared on Antarctica in the Oligocene and then on Greenland in the Miocene, and covered much of North America and Europe by the Pleistocene. The most recent of the Pleistocene glaciations ended around 11,700 years ago. The current epoch is known as the Holocene. Epochs are further divided into ages (a.k.a. stages), but we won’t be going into that level of detail here.

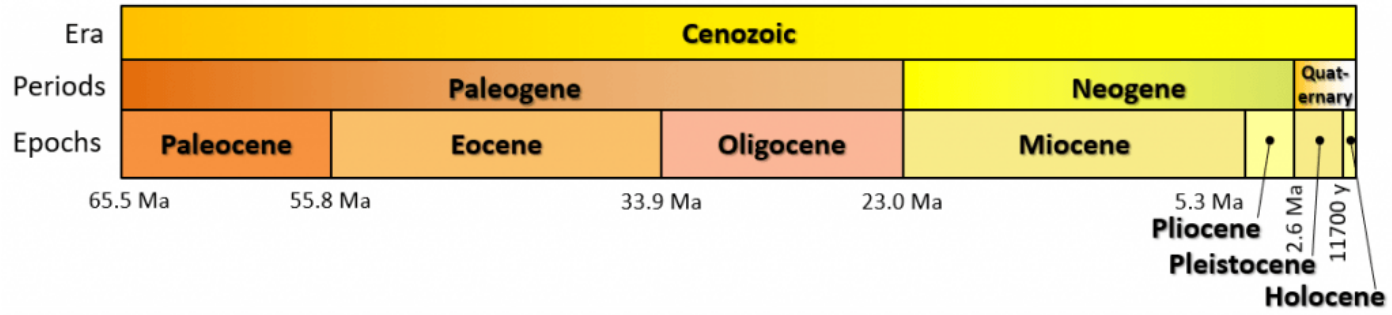


Figure 8.1.4 The periods (middle row) and epochs (bottom row) of the Cenozoic era.

Most of the boundaries between the periods and epochs of the geological time scale have been fixed on the basis of significant changes in the fossil record. For example, as already noted, the boundary between the Cretaceous and the Paleogene coincides exactly with a devastating mass extinction. That’s not a coincidence. The dinosaurs and many other types of organisms went extinct at this time, and the boundary between the two periods marks the division between sedimentary rocks with Cretaceous organisms (including dinosaurs) below, and Paleogene organisms above.

Image Descriptions

Figure 8.1.3 image description: The eras and periods that make up the Phanerozoic Eon.

Era	Period	Time span
Paleozoic	Cambrian	488 to 540 Ma
Paleozoic	Ordovician	488 to 444 Ma
Paleozoic	Silurian	444 to 416 Ma
Paleozoic	Devonian	416 to 359 Ma
Paleozoic	Carboniferous	359 to 299 Ma
Paleozoic	Permian	299 to 251 Ma
Mesozoic	Triassic	251 to 202 Ma
Mesozoic	Jurassic	202 to 146 Ma
Mesozoic	Cretaceous	146 to 65.5 Ma
Cenozoic	Paleogene	65.5 to 23 Ma
Cenozoic	Neogene	23 to 2.6 Ma
Cenozoic	Quaternary	2.6 Ma to present

[Return to Figure 8.1.3]

Figure 8.1.4 image description: The periods and epochs that make up the Cenozoic era.

Period	Epoch	Time span
Paleogene	Paleocene	65.5 to 55.8 Ma
Paleogene	Eocene	55.8 to 33.9 Ma
Paleogene	Oligocene	33.9 to 23.0 Ma
Neogene	Miocene	23.0 to 5.3 Ma
Neogene	Pliocene	5.3 to 2.6 Ma
Quaternary	Pleistocene	2.6 Ma to 11,700 years ago
Quaternary	Holocene	11,700 years ago to the present

[Return to Figure 8.1.4]

Media Attributions

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8.2: Relative Dating Methods

The simplest and most intuitive way of dating geological features is to look at the relationships between them. There are a few simple rules for doing this, some of which we've already looked at in Chapter 6. For example, the principle of superposition states that sedimentary layers are deposited in sequence, and, unless the entire sequence has been turned over by tectonic processes or disrupted by faulting, the layers at the bottom are older than those at the top. The **principle of inclusions** states that any rock fragments that are included in rock must be older than the rock in which they are included. For example, a **xenolith** in an igneous rock or a clast in sedimentary rock must be older than the rock that includes it (Figure 8.2.1).



Figure 8.2.1a A xenolith of diorite incorporated into a basalt lava flow, Mauna Kea volcano, Hawaii. The lava flow took place some time after the diorite cooled, was uplifted, and then eroded. (geological rock hammer head for scale).



Figure 8.2.1b Rip-up clasts of shale embedded in Gabriola Formation sandstone, Gabriola Island, B.C. The pieces of shale were eroded as the sandstone was deposited, so the shale is older than the sandstone.

The **principle of cross-cutting relationships** states that any geological feature that cuts across, or disrupts another feature must be younger than the feature that is disrupted. An example of this is given in Figure 8.2.2, which shows three different sedimentary layers. The lower sandstone layer is disrupted by two **faults**, so we can conclude that the faults are younger than that layer. But the faults do not appear to continue into the coal seam, and they certainly do not continue into the upper sandstone. So we can

infer that coal seam is younger than the faults (because it cuts them off), and of course the upper sandstone is youngest of all, because it lies on top of the coal seam.

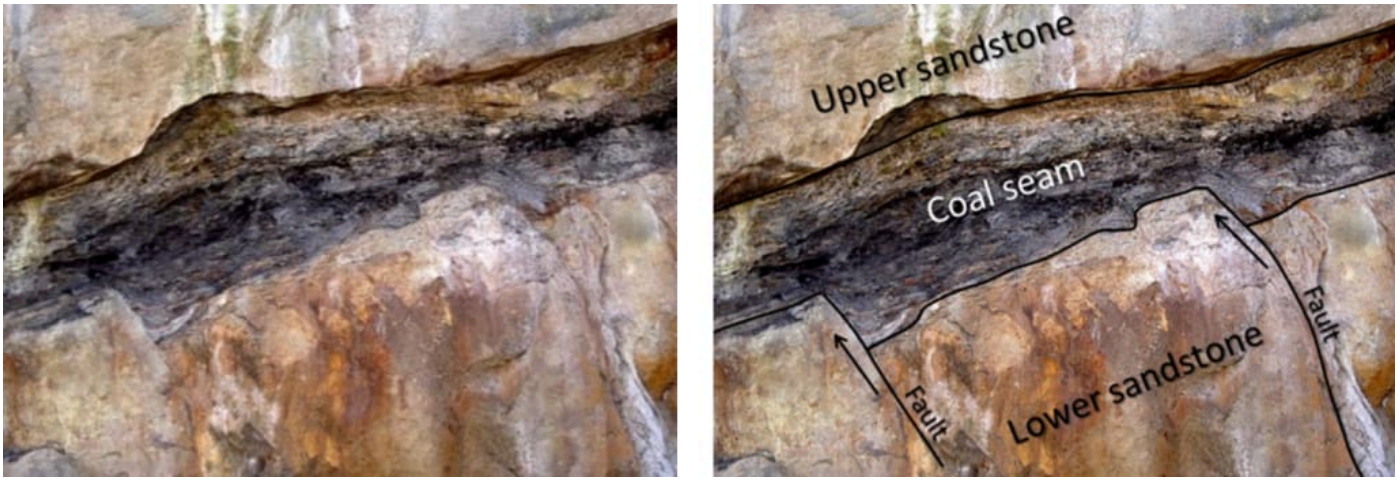


Figure 8.2.2 Superposition and cross-cutting relationships in Cretaceous Nanaimo Group rocks in Nanaimo, B.C. The coal seam is about 50 centimeters thick. The sequence of events is as follows: a) deposition of lower sandstone, b) faulting of lower sandstone, c) deposition of coal seam and d) deposition of upper sandstone.

Exercise 8.1 Cross-Cutting Relationships



Figure 8.2.3

The outcrop shown here (at Horseshoe Bay, B.C.) has three main rock types:

1. Buff/pink felsic intrusive igneous rock present as somewhat irregular masses trending from lower right to upper left
2. Dark grey metamorphosed basalt
3. A 50 centimeters wide light-grey felsic intrusive igneous dyke extending from the lower left to the middle right – offset in several places

Using the principle of cross-cutting relationships outlined above, determine the relative ages of these three rock types.

(The near-vertical stripes are blasting drill holes. The image is about 7 meters across.)

See Appendix 3 for Exercise 8.1 answers.

An **unconformity** represents an interruption in the process of deposition of sedimentary rocks. Recognizing unconformities is important for understanding time relationships in sedimentary sequences. An example of an unconformity is shown in Figure 8.2.4. The Proterozoic rocks of the Grand Canyon Group have been tilted and then eroded to a flat surface prior to deposition of the younger Paleozoic rocks. The difference in time between the youngest of the Proterozoic rocks and the oldest of the Paleozoic rocks is close to 300 million years. Tilting and erosion of the older rocks took place during this time, and if there was any deposition going on in this area, the evidence of it is now gone.



Figure 8.2.4 The great angular unconformity in the Grand Canyon, Arizona. The tilted rocks at the bottom are part of the Proterozoic Grand Canyon Group (aged 825 to 1,250 Ma). The flat-lying rocks at the top are Paleozoic (540 to 250 Ma). The boundary between the two represents a time gap of nearly 300 million years.

There are four types of unconformities, as summarized in Table 8.1, and illustrated in Figure 8.2.5.

Table 8.1 The characteristics of the four types of unconformities

Unconformity Type	Description
Nonconformity	A boundary between non-sedimentary rocks (below) and sedimentary rocks (above)
Angular unconformity	A boundary between two sequences of sedimentary rocks where the underlying ones have been tilted (or folded) and eroded prior to the deposition of the younger ones (as in Figure 8.2.4)
Disconformity	A boundary between two sequences of sedimentary rocks where the underlying ones have been eroded (but not tilted) prior to the deposition of the younger ones (as in Figure 8.2.2)
Paraconformity	A time gap in a sequence of sedimentary rocks that does not show up as an angular unconformity or a disconformity

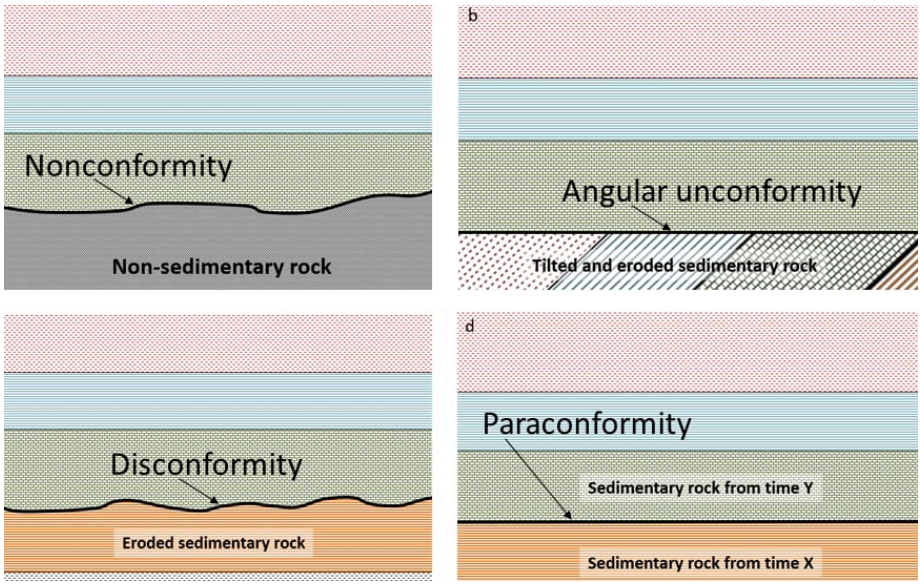


Figure 8.2.5 The four types of unconformities: (a) a nonconformity between older non-sedimentary rock and sedimentary rock, (b) an angular unconformity, (c) a disconformity between layers of sedimentary rock, where the older rock has been eroded but not tilted, and (d) a paraconformity where there is a long period (typically millions of years) of non-deposition between two parallel layers.

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8.3: Dating Rocks Using Fossils

Geologists get a wide range of information from fossils. They help us to understand evolution and life in general; they provide critical information for understanding depositional environments and changes in Earth's climate; and, of course, they can be used to date rocks.

Although the recognition of fossils goes back hundreds of years, the systematic cataloguing and assignment of relative ages to different organisms from the distant past—paleontology—only dates back to the earliest part of the 19th century. The oldest undisputed fossils are from rocks dated around 3.5 Ga, and although fossils this old are tiny, typically poorly preserved and are not useful for dating rocks, they can still provide important information about conditions at the time. The oldest well-understood fossils are from rocks dating back to around 600 Ma, and the sedimentary record from that time forward is rich in fossil remains that provide a detailed record of the history and evolution of life on Earth. However, as anyone who has gone hunting for fossils knows, that does not mean that all sedimentary rocks have visible fossils, or that they are easy to find. Fossils alone cannot provide us with numerical ages of rocks, but over the past century geologists have acquired enough isotopic dates from rocks associated with fossil-bearing rocks (such as igneous dykes cutting through sedimentary layers, or volcanic layers between sedimentary layers) to be able to put specific time limits on most fossils.

A very selective history of life on Earth over the past 600 million years is provided in Figure 8.3.1. The major groups of organisms that we are familiar with evolved between the late Proterozoic and the Cambrian (approximately 600 to 520 Ma). Plants, which evolved in the oceans as green algae, came onto land during the Ordovician (approximately 450 Ma). Insects, which evolved from marine arthropods, came onto land during the Devonian (400 Ma), and amphibians (i.e., vertebrates) came onto land about 50 million years later. By the late Carboniferous, trees had evolved from earlier plants, and reptiles had evolved from amphibians. By the mid-Triassic, dinosaurs and mammals had evolved from very different branches of the reptiles; birds evolved from dinosaurs during the Jurassic. Flowering plants evolved in the late Jurassic or early Cretaceous. The earliest primates evolved from other mammals in the early Paleogene, and the genus *Homo* evolved during the late Neogene (roughly 2.8 Ma).

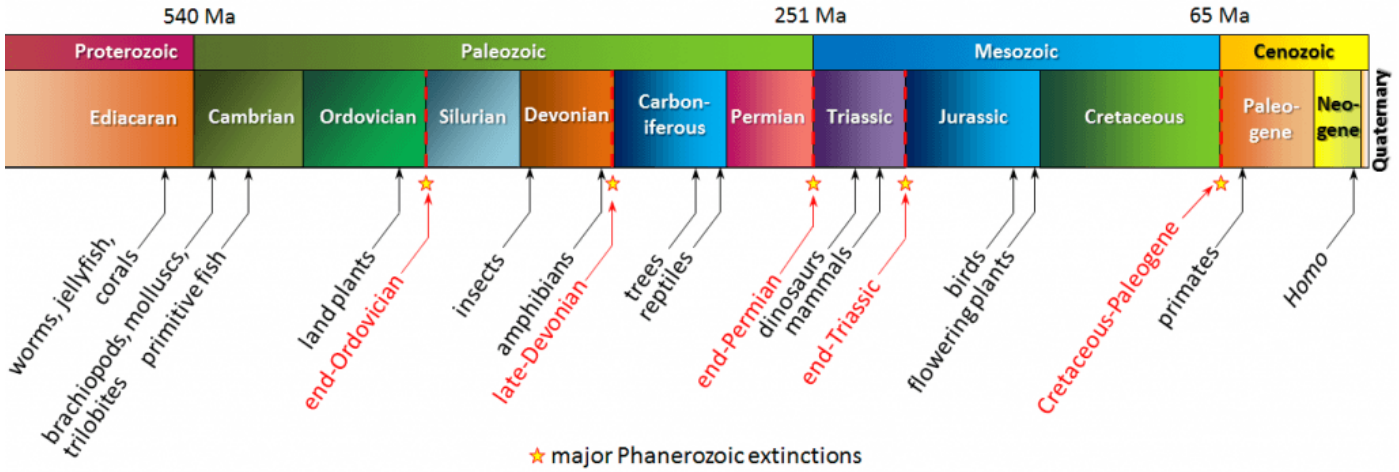


Figure 8.3.1 A summary of life on Earth during the late Proterozoic and the Phanerozoic. The top row shows geological eras, and the lower row shows the periods.

If we understand the sequence of evolution on Earth, we can apply knowledge to determining the relative ages of rocks. This is William Smith's principle of faunal succession, although of course it doesn't just apply to "fauna" (animals); it can also apply to fossils of plants and those of simple organisms.

The Phanerozoic has seen five major extinctions, as indicated in Figure 8.3.1. The most significant of these was at the end of the Permian, which saw the extinction of over 80% of all species and over 90% of all marine species. Most well-known types of organisms were decimated by this event, but only a few became completely extinct, including trilobites. The second most significant extinction was at the Cretaceous-Paleogene boundary (K-Pg, a.k.a. the K-T extinction). At that time, about 75% of marine species disappeared. Again, a few well-known types of organisms disappeared altogether, including the dinosaurs (but not birds) and the pterosaurs. Many other types were badly decimated by that event but survived, and then flourished in the Paleogene. The K-Pg extinction is thought to have been caused by the impact of a large extraterrestrial body (10 to 15 kilometers

across), but it is generally agreed that the other four Phanerozoic extinctions had other causes, although their exact nature is not clearly understood.

As already stated, it is no coincidence that the major extinctions all coincide with boundaries of geological periods and even eras. Geologists have placed most of the divisions of the geological time scale at points in the fossil record where there are major changes in the type of organisms observed, and most of these correspond with minor or major extinctions.

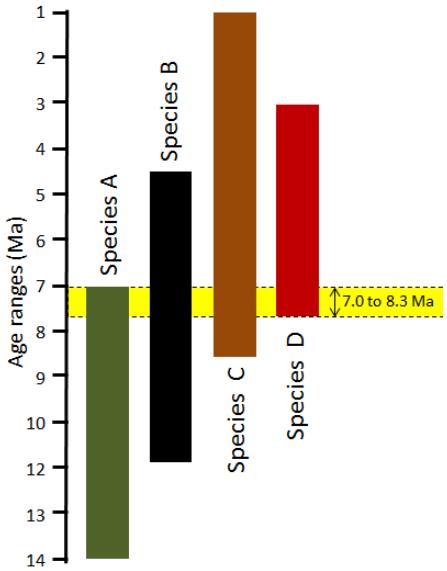


Figure 8.3.2 The application of bracketing to constrain the age of a rock based on several fossils. In this diagram, the colored bars represent the time range during which each of the four species (A, B, C, D) existed on Earth. Although each species lived for several million years, we can narrow down the likely age of the rock to just 700,000 years during which all four species coexisted.

If we can identify a fossil to the species level, or at least to the genus level, and we know the time period when the organism lived, we can assign a range of time to the rock. That range might be several million years because some organisms survived for a very long time. If the rock we are studying has several types of fossils in it, and we can assign time ranges to several of them, we might be able to narrow the time range for the age of the rock considerably. An example of this is given in Figure 8.3.2.

Some organisms survived for a very long time, and are not particularly useful for dating rocks. Sharks, for example, have been around for over 400 million years, and the great white shark has survived for 16 million years, so far. Organisms that lived for relatively short time periods are particularly useful for dating rocks, especially if they were distributed over a wide geographic area and so can be used to compare rocks from different regions. These are known as **index fossils**. There is no specific limit on how short the time span has to be to qualify as an index fossil. Some lived for millions of years, and others for much less than a million years.

Some well-studied groups of organisms qualify as **biozone** fossils because, although the genera and families lived over a long time, each species lived for a relatively short time and can be easily distinguished from others on the basis of specific features. For example, ammonites have a distinctive feature known as the **suture** line—where the internal shell layers that separate the individual chambers (**septae**) meet the outer shell wall, as shown in Figure 8.3.3. These suture lines are sufficiently variable to identify species that can be used to estimate the relative or absolute ages of the rocks in which they are found.



Figure 8.3.3 The septum of an ammonite (white part, left), and the suture lines where the septae meet the outer shell (right).

Foraminifera (small, carbonate-shelled marine organisms that originated during the Triassic and are still around today) are also useful biozone fossils. As shown in Figure 8.3.4, numerous different foraminifera lived during the Cretaceous. Some lasted for over 10 million years, but others for less than 1 million years. If the foraminifera in a rock can be identified to the species level, we can get a good idea of its age.

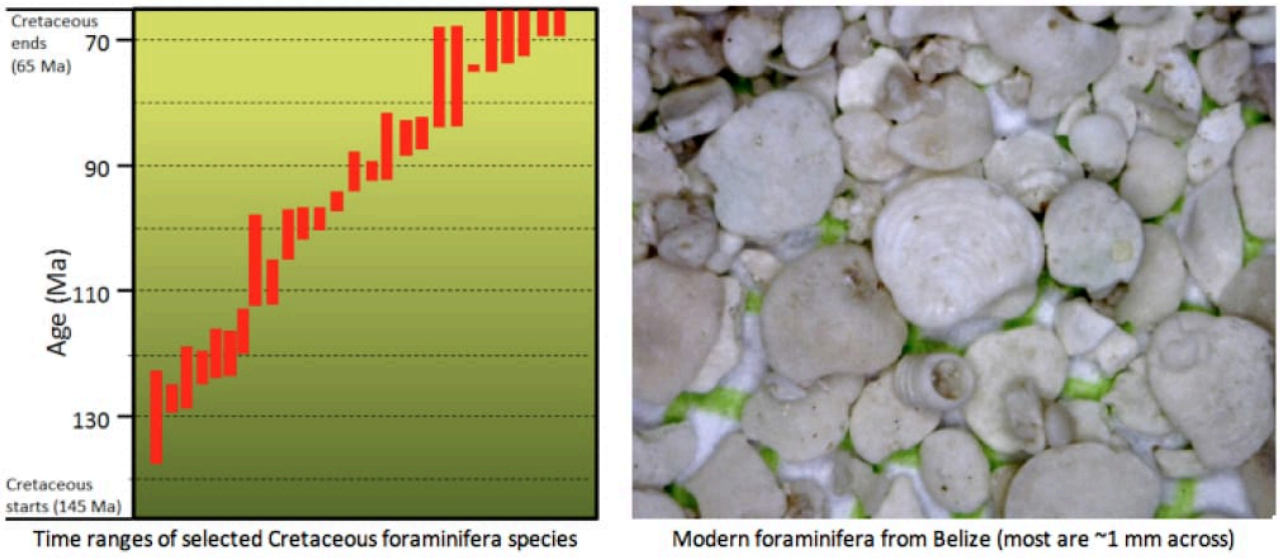


Figure 8.3.4 Time ranges for Cretaceous foraminifera (left). Modern foraminifera from the Ambergris area of Belize (right).

Exercise 8.2 Dating rocks using index fossils

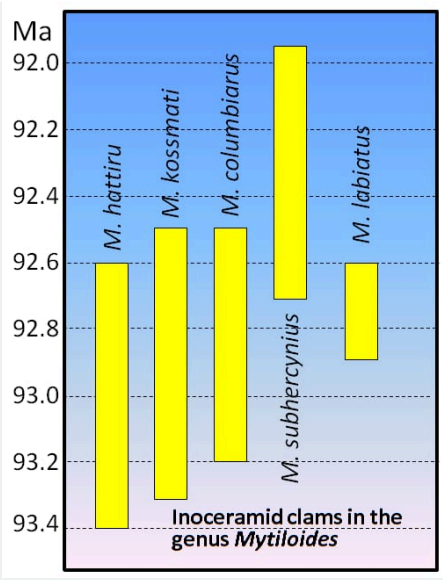


Figure 8.3.5 shows the age ranges for some late Cretaceous inoceramid clams in the genus Mytiloides:

Figure 8.3.5 shows the age ranges for some late Cretaceous inoceramid clams in the genus *Mytiloides:*

- *M. hattiru*, 93.4 to 92.6 Ma
- *M. kossmati*, 93.3 to 92.5 Ma
- *M. columbiarus*, 93.2 to 92.5 Ma
- *M. subhercynius*, 92.7 to 91.9 Ma
- *M. labiatus*, 92.9 to 92.6 Ma

Using the bracketing method described above, determine the possible age range of the rock that these five organisms were found in.

How would that change if *M. subhercynius* was not present in these rocks?

See Appendix 3 for Exercise 8.2 answers.

Image Descriptions

Figure 8.3.1 image description: Life on earth during the late Proterozoic and the Phanerozoic.

Eon	Ero	Period	Life on Earth
Proterozoic		Ediacaran	Worms, jellyfish, corals
Phanerozoic	Paleozoic (540 to 251 Ma)	Cambrian	Brachiopods, molluscs, trilobites, primitive fish
Phanerozoic	Paleozoic (540 to 251 Ma)	Ordovician	Land plants, the period ends with a major extinction
Phanerozoic	Paleozoic (540 to 251 Ma)	Silurian	
Phanerozoic	Paleozoic (540 to 251 Ma)	Devonian	Insects, amphibians, the period ends with a major extinction
Phanerozoic	Paleozoic (540 to 251 Ma)	Carboniferous	Trees, reptiles
Phanerozoic	Paleozoic (540 to 251 Ma)	Permian	The period ends with a major extinction
Phanerozoic	Mesozoic (251 to 65 Ma)	Triassic	Dinosaurs, mammals, the period ends with a major extinction
Phanerozoic	Mesozoic (251 to 65 Ma)	Jurrassic	Birds, flowering plants

Eon	Ero	Period	Life on Earth
Phanerozoic	Mesozoic (251 to 65 Ma)	Cretaceous	The period ends with a major extinction
Phanerozoic	Cenozoic, (65 Ma to present)	Paleogene	Primates
Phanerozoic	Cenozoic, (65 Ma to present)	Neogene	*Homo*
Phanerozoic	Cenozoic, (65 Ma to present)	Quaternary	

[Return to Figure 8.3.1]

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- Figure 8.3.5: © Steven Earle. CC BY. Based on data at obtained from Lower Turonian Euramerican Inoceramidae: A morphologic, taxonomic, and biostratigraphic overview.

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8.4: Isotopic Dating Methods

Originally fossils only provided us with relative ages because, although early paleontologists understood biological succession, they did not know the absolute ages of the different organisms. It was only in the early part of the 20th century, when isotopic dating methods were first applied, that it became possible to discover the absolute ages of the rocks containing fossils. In most cases, we cannot use isotopic techniques to directly date fossils or the sedimentary rocks they are found in, but we can constrain their ages by dating igneous rocks that cut across sedimentary rocks, or volcanic layers that lie within sedimentary layers.

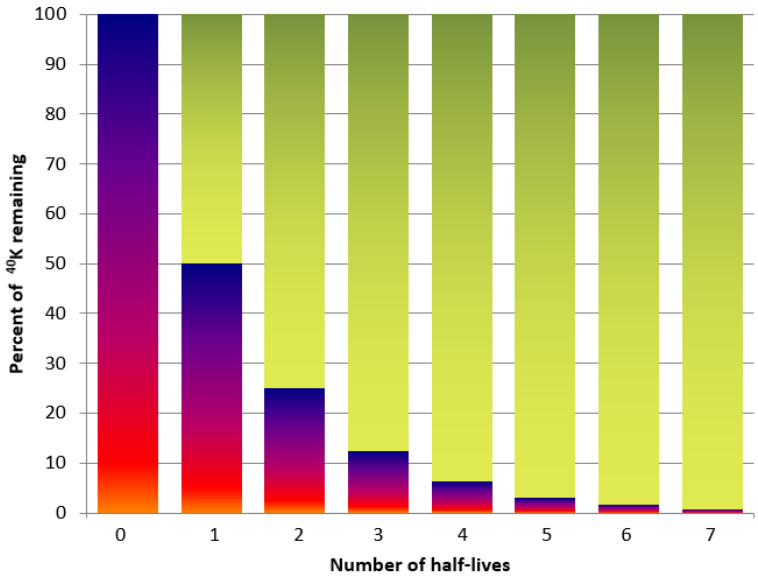


Figure 8.4.1 The decay of ^{40}K over time. Each half-life is 1.3 billion years, so after 3.9 billion years (three half-lives) 12.5% of the original ^{40}K will remain. The red-blue bars represent ^{40}K and the green-yellow bars represent ^{40}Ar.

Isotopic dating of rocks, or the minerals in them, is based on the fact that we know the decay rates of certain unstable **isotopes** of elements and that these rates have been constant over geological time. It is also based on the premise that when the atoms of an element decay within a mineral or a rock, they stay there and don't escape to the surrounding rock, water, or air. One of the isotope pairs widely used in geology is the decay of ^{40}K to ^{40}Ar (potassium-40 to argon-40). ^{40}K is a radioactive isotope of potassium that is present in very small amounts in all minerals that have potassium in them. It has a half-life of 1.3 billion years, meaning that over a period of 1.3 Ga one-half of the ^{40}K atoms in a mineral or rock will decay to ^{40}Ar, and over the next 1.3 Ga one-half of the remaining atoms will decay, and so on (Figure 8.4.1).

In order to use the K-Ar dating technique, we need to have an igneous or metamorphic rock that includes a potassium-bearing mineral. One good example is granite, which normally has some potassium feldspar (Figure 8.4.2). Feldspar does not have any argon in it when it forms. Over time, the ^{40}K in the feldspar decays to ^{40}Ar. Argon is a gas and the atoms of ^{40}Ar remain embedded within the crystal, unless the rock is subjected to high temperatures after it forms. The sample must be analyzed using a very sensitive mass-spectrometer, which can detect the differences between the masses of atoms, and can therefore distinguish between ^{40}K and the much more abundant ^{39}K. Biotite and hornblende are also commonly used for K-Ar dating.

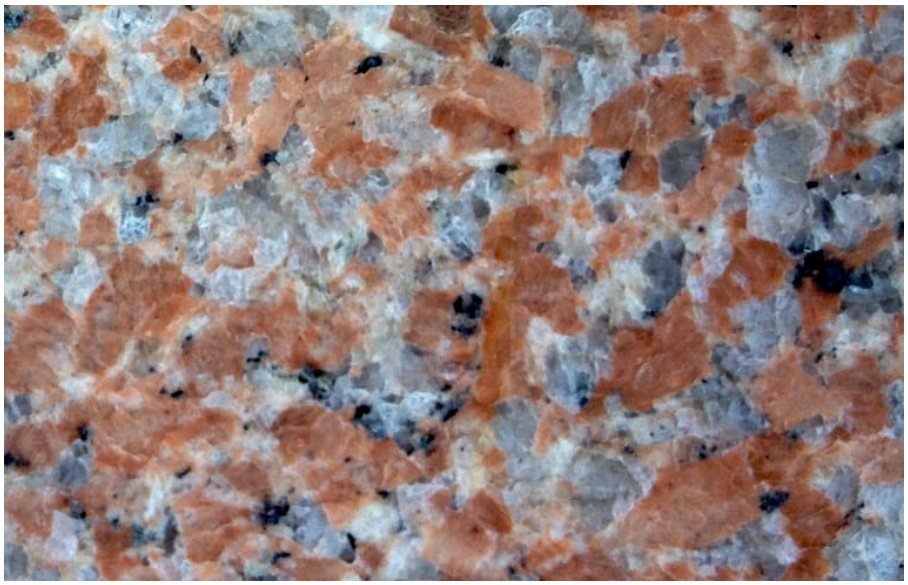


Figure 8.4.2 Crystals of potassium feldspar (pink) in a granitic rock are candidates for isotopic dating using the K-Ar method because they contained potassium and no argon when they formed.

Why can't we use isotopic dating techniques to accurately date sedimentary rocks?



Figure 8.4.3

An important assumption that we have to be able to make when using isotopic dating is that when the rock formed none of the daughter isotope was present (e.g., ^{40}Ar in the case of the K-Ar method). A clastic sedimentary rock is made up of older rock and mineral fragments, and when the rock forms it is almost certain that all of the fragments already have daughter isotopes in them. Furthermore, in almost all cases, the fragments have come from a range of source rocks that all formed at different times. If we dated a number of individual grains in the sedimentary rock, we would likely get a range of different dates, all older than the age of the rock. That could be useful information, but it would not provide an accurate date for the rock in question.

It might be possible to directly date some chemical sedimentary rocks isotopically, but there are no useful isotopes that can be used on old chemical sedimentary rocks. Radiocarbon dating can be used on sediments or sedimentary rocks that contain carbon, but it cannot be used on materials older than about 60 ka.

K-Ar is just one of many isotope-pairs that are useful for dating geological materials. Some of the other important pairs are listed in Table 8.2, along with the age ranges that they apply to and some comments on their applications. When radiometric techniques are applied to metamorphic rocks, the results normally tell us the date of metamorphism, not the date when the parent rock

formed.

Table 8.2 A few of the isotope systems that are widely used for dating geological materials

[Skip Table]

Isotope System	Half-Life	Useful Range	Comments
Potassium-argon	1.3 Ga	10 Ka to 4.57 Ga	Widely applicable because most rocks have some potassium
Uranium-lead	4.5 Ga	1 Ma to 4.57 Ga	The rock must have uranium-bearing minerals, but most have enough.
Rubidium-strontium	47 Ga	10 Ma to 4.57 Ga	Less precision than other methods at old dates
Carbon-nitrogen (a.k.a. radiocarbon dating)	5,730 years	100 to 60,000 years	Sample must contain wood, bone, or carbonate minerals; can be applied to young sediments

Exercise 8.3 Isotopic dating

Assume that a feldspar crystal from the granite shown in Figure 8.4.2 was analyzed for ^{40}K and ^{40}Ar. The proportion of ^{40}K remaining is 0.91. Using the decay curve shown on the graph below, estimate the age of the rock.

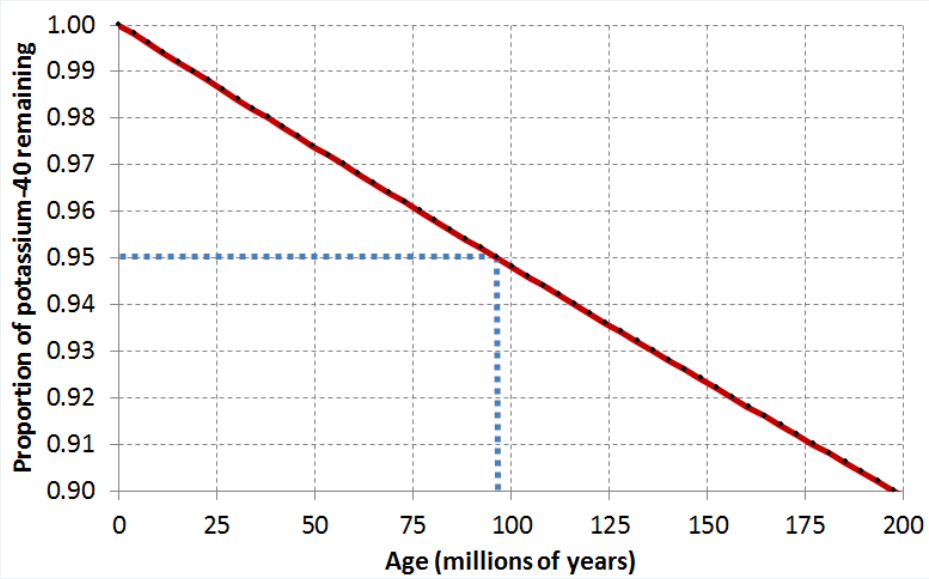


Figure 8.4.4

An example is provided (in blue) for a ^{40}K proportion of 0.95, which is equivalent to an age of approximately 96 Ma. This is determined by drawing a horizontal line from 0.95 to the decay curve line, and then a vertical line from there to the time axis.

See Appendix 3 for Exercise 8.3 answers.

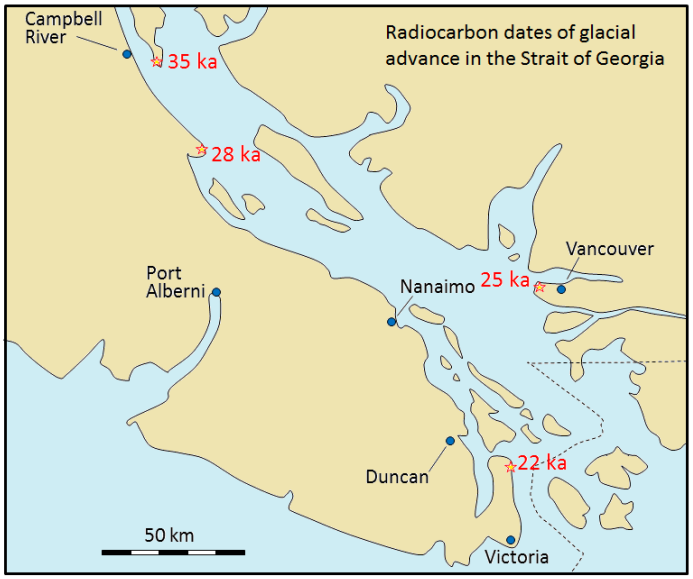


Figure 8.4.5 Radiocarbon dates on wood fragments in glacial sediments in the Strait of Georgia.

Radiocarbon dating (using ^{14}C) can be applied to many geological materials, including sediments and sedimentary rocks, but the materials in question must be younger than 60 ka. Fragments of wood incorporated into young sediments are good candidates for carbon dating, and this technique has been used widely in studies involving late Pleistocene glaciers and glacial sediments. An example is shown in Figure 8.4.5; radiocarbon dates from wood fragments in glacial sediments have been used to estimate the timing of the last glacial advance along the Strait of Georgia. It is evident that the ice-front of the major glacier that occupied the Strait of Georgia was near to Campbell River at around 35 ka, near to Nanaimo and Vancouver at about 25 ka, and had reached the Victoria area by around 22 ka.

Over the past decade there has been increasing use of U-Pb dating to study sedimentary rocks, not necessarily to find out the age of the rock, but to discover something about its history and origins. All clastic sedimentary rocks contain some tiny clasts of the silicate mineral zircon ($ZrSiO_4$), derived from the weathering of the sediment parent rocks. Zircon always has some uranium in it (but no lead) so it is a good candidate for U-Pb dating, and it isn't too difficult to separate the grains of zircon from the other grains in a sandstone. The procedure is to isolate a few hundred tiny zircons from a rock sample, and then carry out U-Pb dating on each one of them. An example of the types of results obtained are shown on Figure 8.4.6. All of the samples are from Nanaimo Gp. rocks on Vancouver Island and nearby Salt Spring Island.

The three samples from Vancouver Island have zircons aged around 90 Ma, 118 Ma and 150 Ma. The Salt Spring Island sample has some zircons aged around 150 Ma, but most are much older, at 200 Ma and 340 to 360 Ma. It is interpreted that the younger zircons (90 to 150 Ma) are mostly derived from granitic rocks in the Coast Range, while the older ones (>200 Ma) are from older rocks on Vancouver Island (Huang, 2018).

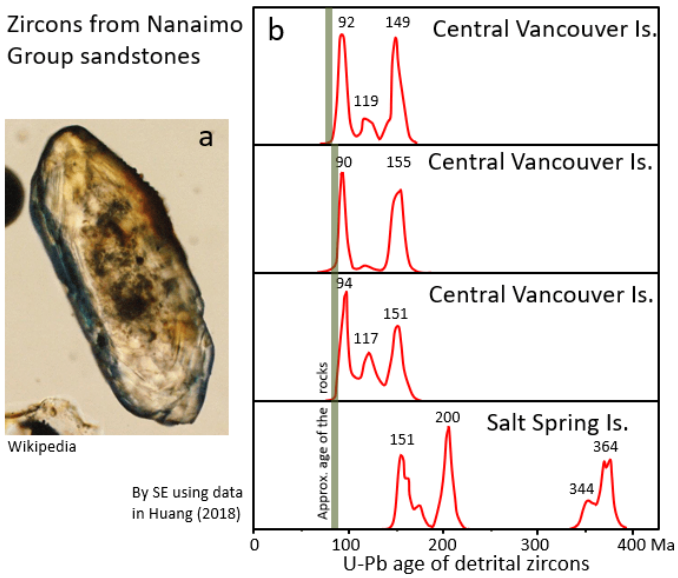


Figure 8.4.6 U-Pb dates for zircon samples from the Nanaimo Gp. (after Krause, 2018) a: a typical zircon clast (this one is about 1/4 mm long). b: plots of zircon ages for 4 sandstone samples.

Image Descriptions

Figure 8.4.1 image description: Decay of 40K over time.

Number of half-lives	Percent of 40K remaining	Percent of 40Ar
0	100	0
1	50	50
2	25	75
3	12.5	87.5
4	6.25	93.75
5	3.125	96.875
6	1.5625	98.4375
7	0.78125	99.21875

[Return to Figure 8.4.1]

Figure 8.4.4 image description: isotopic dating graph

Proportion of Potassium-40 remaining	Age (in millions of Years)
1.00	0
0.99	19
0.98	37
0.97	55
0.96	75
0.95	96
0.94	114
0.93	134

Proportion of Potassium-40 remaining	Age (in millions of Years)
0.92	156
0.91	175
0.90	194

[Return to Figure 8.4.4]

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- Figure 8.4.5: © Steven Earle. CC BY. From J. Clague, 1976, Quadra Sand and its relation to late Wisconsin glaciation of southeast British Columbia, Can. J. Earth Sciences, V. 13, p. 803-815.
- Figure 8.4.6 (left): "Zircon microscope" © Chd. CC BY-SA.
- Figure 8.4.6b: © Steven Earle. CC BY. From data in Huang, C, 2018, Refining the chronostratigraphy of the lower Nanaimo Group, Vancouver Island, Canada, using Detrital Zircon Geochronology, MSc thesis, Department of Earth Science, Simon Fraser University, 74 p.

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8.5: Other Dating Methods

There are numerous other techniques for dating geological materials, but we will examine just two of them here: tree-ring dating (i.e., dendrochronology) and dating based on the record of reversals of Earth's magnetic field.

Dendrochronology can be applied to dating very young geological materials based on reference records of tree-ring growth going back many millennia. The longest such records can take us back to 25 ka, to the height of the last glaciation. One of the advantages of dendrochronology is that, providing reliable reference records are available, the technique can be used to date events to the nearest year.

Dendrochronology has been used to date the last major subduction-zone earthquake on the coast of B.C., Washington, and Oregon. When large earthquakes strike in this setting, there is a tendency for some coastal areas to subside by one or two meters. Seawater then rushes in, flooding coastal flats and killing trees and other vegetation within a few months. There are at least four locations along the coast of Washington that have such dead trees (and probably many more in other areas). Wood samples from these trees have been studied and the ring patterns have been compared with patterns from old living trees in the region (Figure 8.5.1).

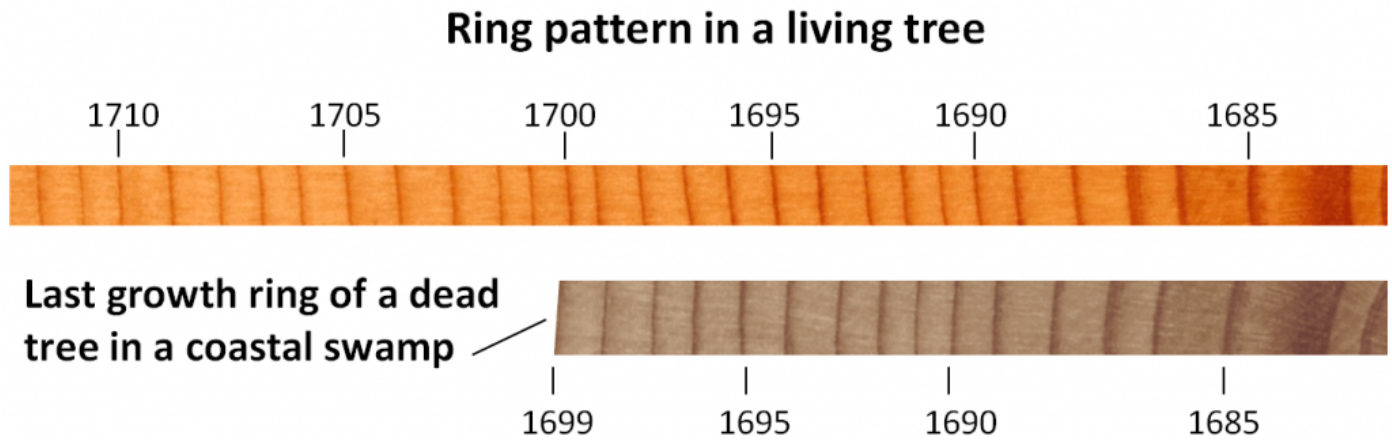


Figure 8.5.1 Example of tree-ring dating of dead trees.

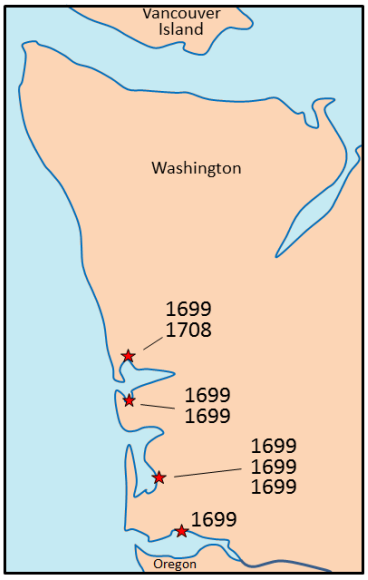


Figure 8.5.2 Sites in Washington where dead trees are present in coastal flats. The outermost wood of eight trees was dated using dendrochronology, and of these, seven died during the year 1699, suggesting that the land near to the coast was inundated by water at that time.

At all of the locations studied, the trees were found to have died either in the year 1699, or very shortly thereafter (Figure 8.5.2). On the basis of these results, it was concluded that a major earthquake took place in this region sometime between the end of growing season in 1699 and the beginning of the growing season in 1700. Evidence from a major tsunami that struck Japan on January 27, 1700, narrowed the timing of the earthquake to sometime in the evening of January 26, 1700. For more information, see The 1700 Juan de Fuca Earthquake.

Changes in Earth's magnetic field can also be used to date events in geologic history. The magnetic field makes compasses point toward the North Pole, but, as we'll see in Chapter 10, this hasn't always been the case. At various times in the past, Earth's magnetic field has reversed itself completely, and during those times a compass would have pointed to the South Pole. By studying magnetism in volcanic rocks that have been dated isotopically, geologists have been able to delineate the chronology of magnetic field reversals going back to 250 Ma. About 5 million years of this record is shown in Figure 8.5.3, where the black bands represent periods of normal magnetism ("normal" meaning similar to the current magnetic field) and the white bands represent periods of reversed magnetism. These periods of consistent magnetic polarity are given names to make them easier to reference. The current normal magnetic field, known as Brunhes, has lasted for the past 780,000 years. Prior to that there was a short reversed period and then a short normal period known as Jaramillo.

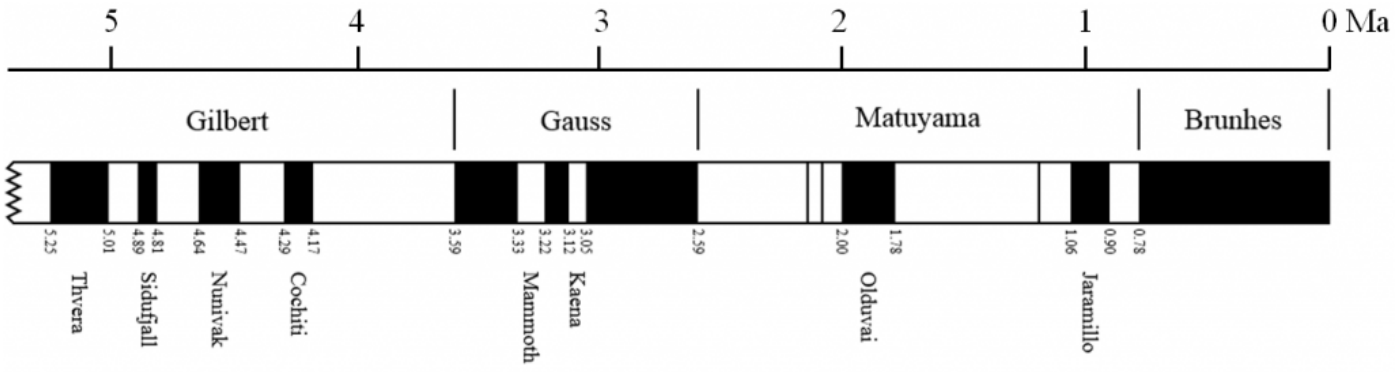


Figure 8.5.3 The last 5 Ma of magnetic field reversals.

Oceanic crust becomes magnetized by the magnetic field that exists as the crust forms from magma. As it cools, tiny crystals of magnetite that form within the magma become aligned with the existing magnetic field and then remain that way after all of the rock has hardened, as shown in Figure 8.5.4. Crust that is forming today is being magnetized in a "normal" sense, but crust that formed 780,000 to 900,000 years ago, in the interval between the Brunhes and Jaramillo normal periods, was magnetized in the "reversed" sense.

Chapter 9 has a discussion of Earth's magnetic field, including where and how it is generated and why its polarity changes periodically.

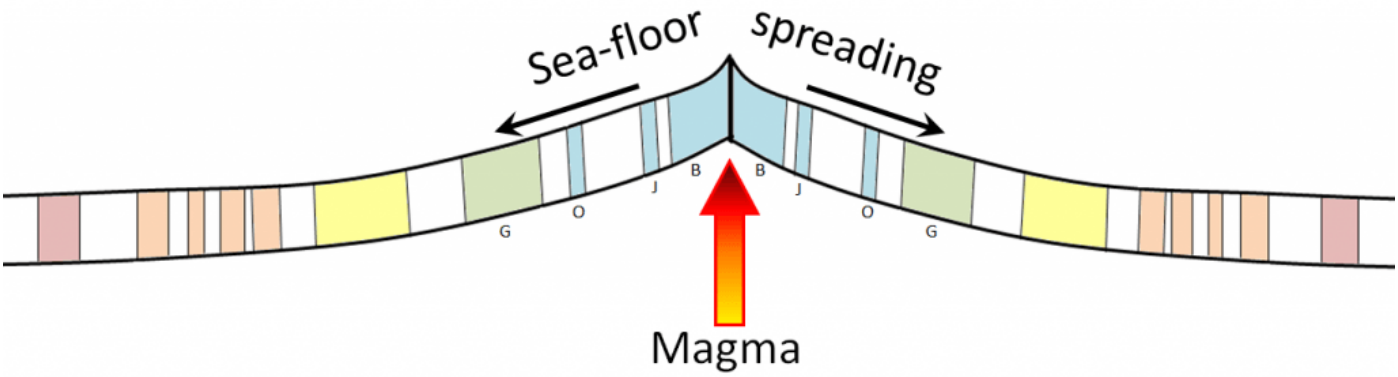


Figure 8.5.4 Depiction of the formation of magnetized oceanic crust at a spreading ridge. Coloured bars represent periods of normal magnetism, and the small capital letters denote the Brunhes, Jaramillio, Olduvai, and Gauss normal magnetic periods (see Figure 8.5.2).

Magnetic chronology can be used as a dating technique because we can measure the magnetic field of rocks using a magnetometer in a lab, or of entire regions by towing a magnetometer behind a ship or an airplane. For example, the Juan de Fuca Plate, which lies off of the west coast of B.C., Washington, and Oregon, is being and has been formed along the Juan de Fuca spreading ridge (Figure 8.5.5). The parts of the plate that are still close to the ridge have normal magnetism, while parts that

are farther away (and formed much earlier) have either normal or reversed magnetism, depending on when the rock formed. By carefully matching the sea-floor magnetic stripes with the known magnetic chronology, we can determine the age at any point on the plate. We can see, for example, that the oldest part of the Juan de Fuca Plate that has not subducted (off the coast of Oregon) is just over 8 million years old, while the part that is subducting underneath Vancouver Island is between 0 and about 6 million years old.

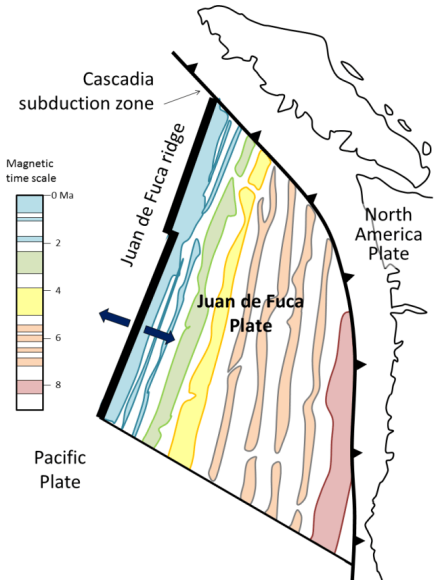


Figure 8.5.5 The pattern of magnetism within the area of the Juan de Fuca Plate, off the west coast of North America. The colored shapes represent parts of the sea floor that have normal magnetism, and the magnetic time scale is shown using the same colors. The blue bands represent Brunhes, Jaramillo, and Olduvai; the green represents Gauss; and so on. (Note that in this diagram, sea-floor magnetism is only shown for the Juan de Fuca Plate, although similar patterns exist on the Pacific Plate.)

Exercise 8.4 Magnetic dating

The fact that magnetic intervals can only be either *normal* or *reversed* places significant limits on the applicability of magnetic dating. If we find a rock with normal magnetism, we can't know which normal magnetic interval it represents, unless we have some other information.

Using Figure 8.5.3 for reference, determine the age of a rock with normal magnetism that has been found to be between 1.5 and 2.0 Ma based on fossil evidence.

How about a rock that is limited to 2.6 to 3.2 Ma by fossils and has reversed magnetism?

See Appendix 3 for Exercise 8.4 answers.

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- Figure 8.5.3: "Geomagnetic polarity late Cenozoic" by the USGS. Adapted by Steven Earle. Public domain.

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8.6: Understanding Geological Time

It's one thing to know the facts about geological time—how long it is, how we measure it, how we divide it up, and what we call the various periods and epochs—but it is quite another to really understand geological time. The problem is that our lives are short and our memories are even shorter. Our experiences span only a few decades, so we really don't have a way of knowing what 11,700 years means. What's more, it's hard for us to understand how 11,700 years differs from 65.5 million years, or even from 1.8 billion years. It's not that we can't comprehend what the numbers mean—we can all get that figured out with a bit of practice—but even if we do know the numerical meaning of 65.5 Ma, we can't really appreciate how long ago it was.

You may be wondering why it's so important to really "understand" geological time. There are some very good reasons. One is so that we can fully understand how geological processes that seem impossibly slow can produce anything of consequence. For example, we are familiar with the concept of driving from one major city to another: a journey of several hours at around 100 kilometers per hour. Continents move toward each other at rates of a fraction of a millimetre per day, or something in the order of 0.000000001 kilometers per hour, and yet, at this impossibly slow rate (try walking at that speed!), they can move thousands of kilometers. Sediments typically accumulate at even slower rates—less than a millimetre per year—but still they are thick enough to be thrust up into monumental mountains and carved into breathtaking canyons.

Another reason is that for our survival on this planet, we need to understand issues like extinction of endangered species and **anthropogenic** (human-caused) climate change. Some people, who don't understand geological time, are quick to say that the climate has changed in the past, and that what is happening now is no different. And it certainly has changed in the past—many times. For example, from the Eocene (50 Ma) to the present day, Earth's climate cooled by about 12°C. That's a huge change that ranks up there with many of the important climate changes of the distant past, and yet the rate of change over that time was only 0.000024°C/century. Anthropogenic climate change has been 1.1°C over the past century,[1]; that is 45,800 times faster than the rate of natural climate change since the Eocene!

One way to wrap your mind around geological time is to put it into the perspective of single year, because we all know how long it is from one birthday to the next. At that rate, each hour of the year is equivalent to approximately 500,000 years, and each day is equivalent to 12.5 million years.

If all of geological time is compressed down to a single year, Earth formed on January 1, and the first life forms evolved in late March (roughly 3,500 Ma). The first large life forms appeared on November 13 (roughly 600 Ma), plants appeared on land around November 24, and amphibians on December 3. Reptiles evolved from amphibians during the first week of December and dinosaurs and early mammals evolved from reptiles by December 13, but the dinosaurs, which survived for 160 million years, were gone by Boxing Day (December 26). The Pleistocene Glaciation got started at around 6:30 p.m. on New Year's Eve, and the last glacial ice left southern Canada by 11:59 p.m.

It's worth repeating: on this time scale, the earliest ancestors of the animals and plants with which we are familiar did not appear on Earth until mid-November, the dinosaurs disappeared after Christmas, and most of Canada was periodically locked in ice from 6:30 to 11:59 p.m. on New Year's Eve. As for people, the first to inhabit B.C. got here about one minute before midnight, and the first Europeans arrived about two seconds before midnight.

It is common for the popular press to refer to distant past events as being "prehistoric." For example, dinosaurs are reported as being "prehistoric creatures," even by the esteemed National Geographic Society.[2] The written records of our history date back to about 6,000 years ago, so anything prior to that can be considered "prehistoric." But to call the dinosaurs prehistoric is equivalent to—and about as useful as—saying that Singapore is beyond the city limits of Kamloops! If we are going to become literate about geological time, we have to do better than calling dinosaurs, or early horses (54 Ma), or even early humans (2.8 Ma), "prehistoric."

Exercise 8.5 What happened on your birthday?

Using the "all of geological time compressed to one year" concept, determine the geological date that is equivalent to your birthday. First go to Day Number of the Year Calculator to find out which day of the year your birth date is. Then divide that number by 365, and multiply that number by 4,570 to determine the time (in millions since the *beginning* of geological time). Finally subtract that number from 4,570 to determine the date back from the present.

For example, April Fool's Day (April 1) is day 91 of the year: 91/365 = 0.2493. 0.2493 x 4,570 = 1,139 million years from the start of time, and 4,570 – 1,193 = 3,377 Ma is the geological date.

Finally, go to the Foundation for Global Community's "Walk through Time" website to find out what was happening on your day. The nearest date to 3,377 Ma is 3,400 Ma. Bacteria ruled the world at 3,400 Ma, and there's a discussion about their lifestyles.

See Appendix 3 for Exercise 8.5 answers.

1. Climate change data from NASA Goddard Institute for Space Studies: http://data.giss.nasa.gov/gistemp/tabledata_v3/GLB.Ts.txt↵
2. http://science.nationalgeographic.com/science/prehistoric-world/ ↵

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8.7: Summary

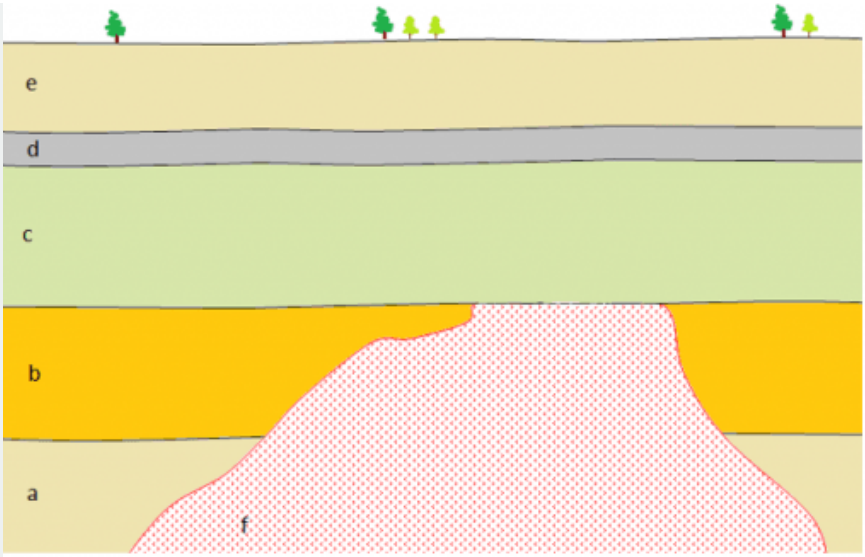
The topics covered in this chapter can be summarized as follows:

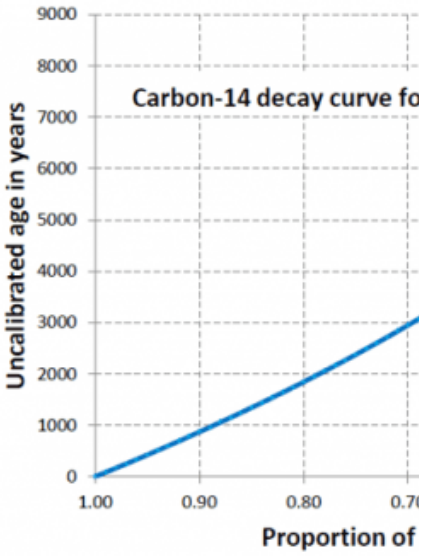
Section	Summary
8.1 The Geological Time Scale	The work of William Smith was critical to the establishment of the first geological time scale early in the 19th century, but it wasn't until the 20th century that geologists were able to assign reliable dates to the various time periods. Geological time is divided into eons, eras, periods, and epochs and the geological time scale is maintained and updated by the International Commission on Stratigraphy.
8.2 Relative Dating Methods	We can determine the relative ages of different rocks by observing and interpreting relationships among them, such as superposition, cross-cutting, and inclusions. Gaps in the geological record are represented by various types of unconformities.
8.3 Dating Rocks Using Fossils	Fossils are useful for dating rocks date back to about 600 Ma. If we know the age range of a fossil, we can date the rock, but some organisms lived for many millions of years. Index fossils represent shorter geological times, and if a rock has several different fossils with known age ranges, we can normally narrow the time during which the rock formed.
8.4 Isotopic Dating Methods	Radioactive isotopes decay at predictable and known rates, and can be used to date igneous and metamorphic rocks. Some of the more useful isotope systems are potassium-argon, rubidium-strontium, uranium-lead, and carbon-nitrogen. Radiocarbon dating can be applied to sediments and sedimentary rocks, but only if they are younger than 60 ka.
8.5 Other Dating Methods	There are many other methods for dating geological materials. Two that are widely used are dendrochronology and magnetic chronology. Dendrochronology, based on studies of tree rings, is widely applied to dating glacial events. Magnetic chronology is based on the known record of Earth's magnetic field reversals.
8.6 Understanding Geological Time	While knowing about geological time is relatively easy, actually comprehending the significance of the vast amounts of geological time is a great challenge. To be able to solve important geological problems and critical societal challenges, like climate change, we need to really understand geological time.

Questions for Review

Answers to Review Questions can be found in Appendix 2.

1. A granitic rock contains inclusions (xenoliths) of basalt. What can you say about the relative ages of the granite and the basalt?
2. Explain the differences between:
    1. a disconformity and a paraconformity
    2. a nonconformity and an angular unconformity
3. What are the features of a useful index fossil?
4. The bellow shows a geological cross-section. The granitic rock "f" at the bottom is the one that you estimated the age of in Exercise 8.3. A piece of wood from layer "d" has been sent for radiocarbon dating and the result was 0.55 ^{14}C remaining. How old is layer "d"? (You can use the carbon-14 decay curve of Figure 8.27 to answer this question.)





Left: Cross section through the crust for questions 4 to 7. Right: ^{14}C decay curve for question 4.

5. Based on your answer to question 4, what can you say about the age of layer “c” in Figure 8.27.?
6. What type of unconformity exists between layer “c” and rock “f”?
7. What about between layer “c” and layer “b”?
8. We can’t use magnetic chronology to date anything younger than 780,000 years. Why not?
9. How did William Smith apply the principle of faunal succession to determine the relative ages of the sedimentary rocks of England and Wales?
10. Access a copy of the geological time scale (International Commission on Stratigraphy). What are the names of the last age (or stage) of the Cretaceous and the first age of the Paleogene? Print out the time scale and stick it on the wall above your desk!

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9: EARTH'S INTERIOR

9.1: Understanding Earth Through Seismology
9.2: The Temperature of Earth's Interior
9.3: Earth's Magnetic Field
9.4: Isostasy
9.5: Summary

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9.1: Understanding Earth Through Seismology

Seismology is the study of vibrations within Earth. These vibrations are caused by various events: earthquakes, extraterrestrial impacts, explosions, storm waves hitting the shore, and tidal effects. Of course, seismic techniques have been most widely applied to the detection and study of earthquakes, but there are many other applications, and arguably seismic waves provide the most important information that we have concerning Earth's interior. Before going any deeper into Earth, however, we need to take a look at the properties of seismic waves. The types of waves that are useful for understanding Earth's interior are called **body waves**, meaning that, unlike the surface waves on the ocean, they are transmitted through Earth materials.

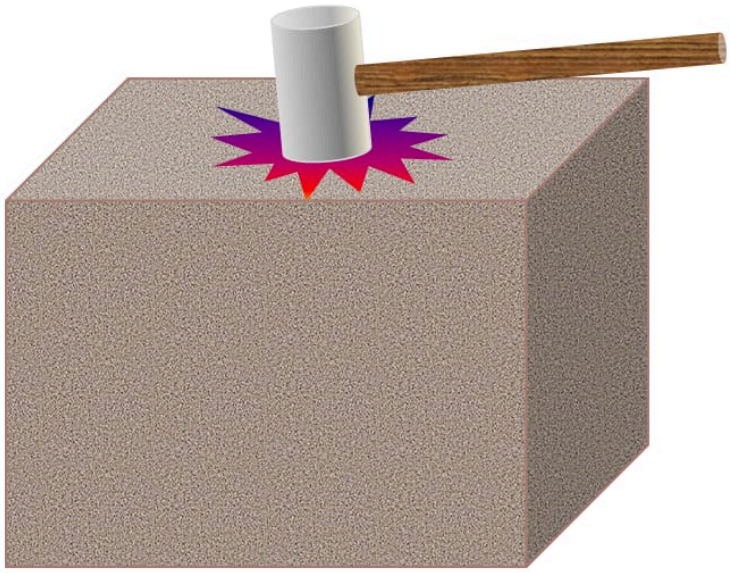


Figure 9.1.1 Hitting a large block of rock with a heavy hammer will create seismic waves within the rock. Please don't try this at home!

Imagine hitting a large block of strong rock (e.g., granite) with a heavy sledgehammer (Figure 9.1.1). At the point where the hammer strikes it, a small part of the rock will be compressed by a fraction of a millimetre. That compression will transfer to the neighbouring part of the rock, and so on through to the far side of the rock—all in a fraction of a second. This is known as a compression wave, and it can be illustrated by holding a loose spring (like a Slinky) that is attached to something (or someone) at the other end. If you give it a sharp push so the coils are compressed, the compression propagates (travels) along the length of the spring and back (Figure 9.1.2). You can think of a compression wave as a "push" wave—it's called a **P wave** (although the "P" stands for "primary" because P waves arrive first at seismic stations).

When we hit a rock with a hammer, we also create a different type of body wave, one that is characterized by back-and-forth vibrations (as opposed to compressions). This is known as a shear wave (**S wave**, where the "S" stands for "secondary"), and an analogy would be what happens when you flick a length of rope with an up-and-down motion. As shown in Figure 9.1.2, a wave will form in the rope, which will travel to the end of the rope and back.

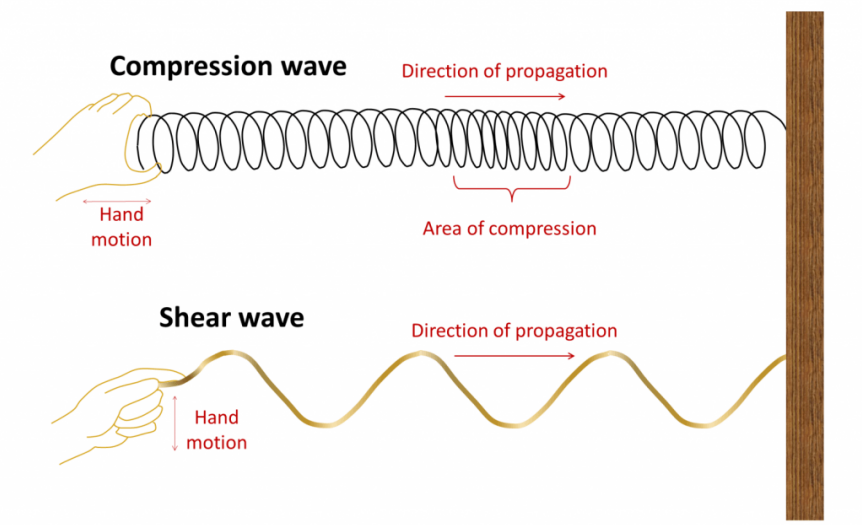


Figure 9.1.2 A compression wave can be illustrated by a spring (like a Slinky) that is given a sharp push at one end. A shear wave can be illustrated by a rope that is given a quick flick.

Compression waves and shear waves travel very quickly through geological materials. As shown in Figure 9.1.3, typical P wave velocities are between 0.5 kilometers per second (km/s) and 2.5 km/s in unconsolidated sediments, and between 3.0 km/s and 6.5 km/s in solid crustal rocks. Of the common rocks of the crust, velocities are greatest in basalt and granite. S waves are slower than P waves, with velocities between 0.1 km/s and 0.8 km/s in soft sediments, and between 1.5 km/s and 3.8 km/s in solid rocks.

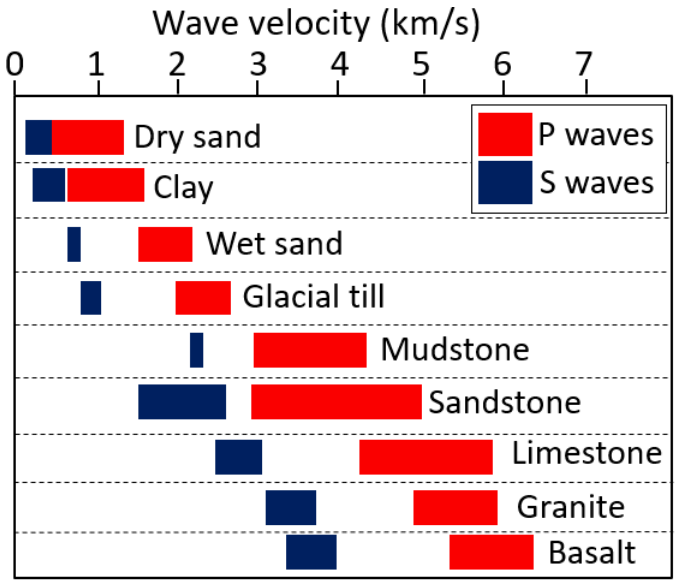


Figure 9.1.3 Typical velocities of P-waves (red) and S-waves (blue) in sediments and in solid crustal rocks.

Exercise 9.1 How soon will seismic waves get here?

Imagine that a strong earthquake takes place on Vancouver Island within Strathcona Park (west of Courtenay). Assuming that the crustal average P wave velocity is 5 km per second, how long will it take (in seconds) for the first seismic waves (P waves) to reach you in the following places (distances from the epicentre are shown)?

1. Nanaimo (120 km away)
2. Surrey (200 km away)
3. Kamloops (390 km away)

See Appendix 3 for Exercise 9.1 answers.

Mantle rock is generally denser and stronger than crustal rock and both P- and S-waves travel faster through the mantle than they do through the crust. Moreover, seismic-wave velocities are related to how tightly compressed a rock is, and the level of compression increases dramatically with depth. Finally, seismic waves are affected by the phase state of rock. They are slowed if there is any degree of melting in the rock. If the material is completely liquid, P waves are slowed dramatically and S waves are stopped altogether.

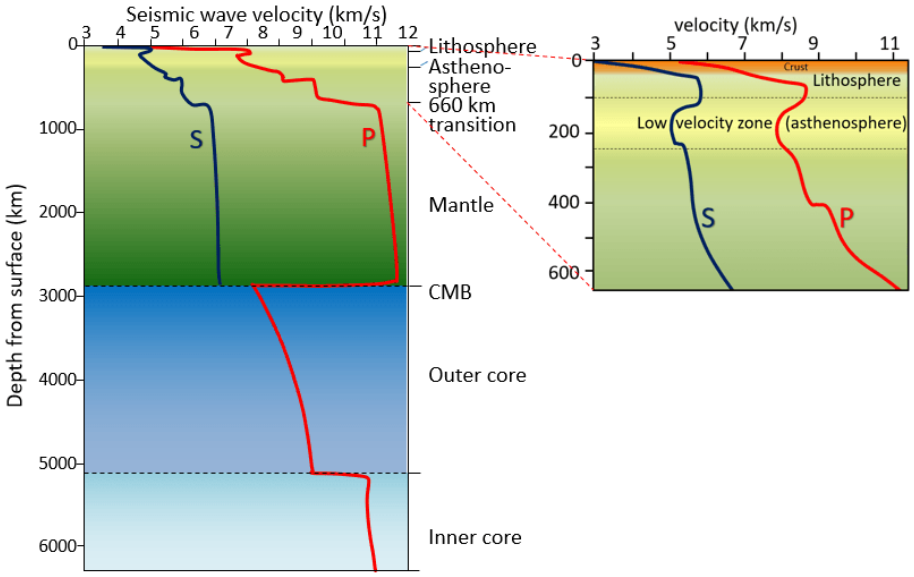


Figure 9.1.4 P wave (red) and S wave (blue) velocity variations with depth in Earth. The diagram on the right shows an expanded view of the upper 660 kilometers of the curves in the diagram on the left.

As shown on the right-hand part of Figure 9.1.4, the upper approximately 100 km of the Earth is known as the lithosphere. This includes the rigid upper part of the mantle (or lithospheric mantle) and the crust. The next 150 km is the asthenosphere or low velocity zone (because seismic waves are slowed as they pass through that material). As we'll see below, that part of the mantle is close to it's melting point and in some regions may be partially molten.

Accurate seismometers have been used for earthquake studies since the late 1800s, and systematic use of seismic data to understand Earth's interior started in the early 1900s. The rate of change of seismic waves with depth in Earth (as shown in Figure 9.1.4) has been determined over the past several decades by analyzing seismic signals from large earthquakes at seismic stations around the world. Small differences in arrival time of signals at different locations have been interpreted to show that:

- Velocities are greater in mantle rock than in the crust.
- Velocities generally increase with pressure, and therefore with depth.
- Velocities slow in the area between a 100 and 250 kilometre depth (called the "low-velocity zone"; equivalent to the asthenosphere).
- Velocities increase dramatically at 660 kilometre depth (because of a mineralogical transition).
- Velocities slow in the region just above the core-mantle boundary (the D" (d-double-prime) layer or "ultra-low-velocity zone").
- S waves do not pass through the outer part of the core.
- P wave velocities increase dramatically at the boundary between the liquid outer core and the solid inner core.

One of the first discoveries about Earth's interior made through seismology was in 1909 when Croatian seismologist Andrija Mohorovičić (pronounced *Moho-ro-vi-chich*) realized that at certain distances from an earthquake, two separate sets of seismic waves arrived at a seismic station within a few seconds of each other. He reasoned that the waves that went down into the mantle, traveled through the mantle, and then were bent upward back into the crust, reached the seismic station first because although they had farther to go, they traveled faster through mantle rock (as shown in Figure 9.1.5). The boundary between the crust and the mantle is known as the **Mohorovičić discontinuity** (or Moho). Its depth is between 30 and 40 kilometers beneath most of the continental crust, and between 5 and 10 kilometers beneath the oceanic crust.

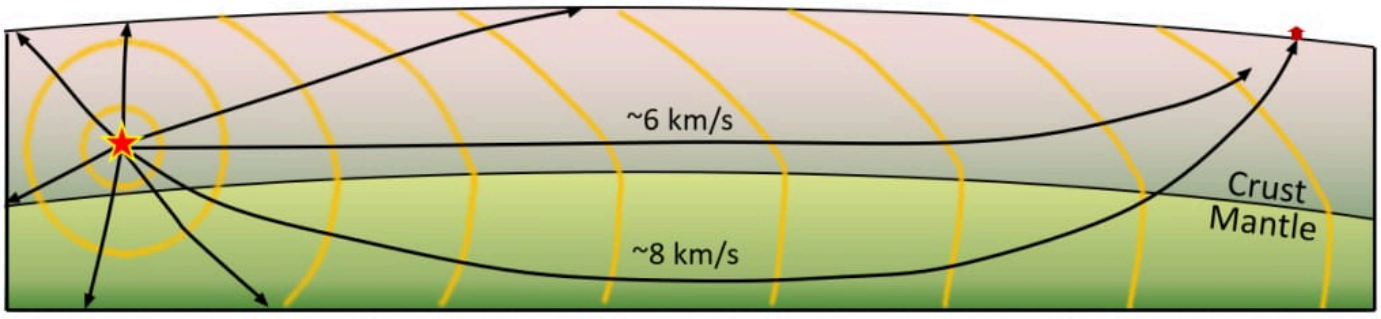


Figure 9.1.5 Depiction of seismic waves emanating from an earthquake (red star). Some waves travel through the crust to the seismic station (at about 6 km/s), while others go down into the mantle (where they travel at around 8 km/s) and are bent upward toward the surface, reaching the station before the ones that traveled only through the crust.

Our current understanding of the patterns of seismic wave transmission through Earth is summarized in Figure 9.1.6. Because of the gradual increase in density (and therefore rock strength) with depth, all waves are refracted (toward the lower density material) as they travel through homogenous parts of Earth and thus tend to curve outward toward the surface. Waves are also refracted at boundaries within Earth, such as at the Moho, at the core-mantle boundary (CMB), and at the outer-core/inner-core boundary.

S waves do not travel through liquids—they are stopped at the CMB—and there is an S wave shadow on the side of Earth opposite a seismic source. The angular distance from the seismic source to the shadow zone is 103° on either side, so the total angular distance of the shadow zone is 154°. We can use this information to infer the depth to the CMB.

P waves do travel through liquids, so they can make it through the liquid part of the core. Because of the refraction that takes place at the CMB, waves that travel through the core are bent away from the surface, and this creates a P wave shadow zone on either side, from 103° to 150°. This information can be used to discover the differences between the inner and outer parts of the core.

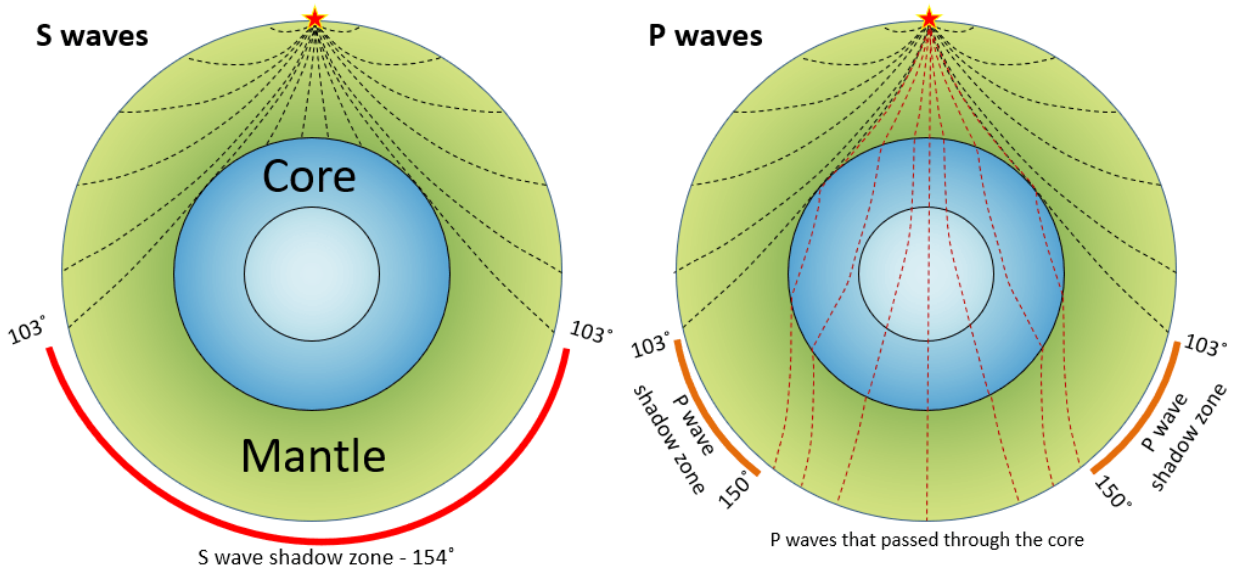


Figure 9.1.6 Patterns of seismic wave propagation through Earth's mantle and core. S waves do not travel through the liquid outer core, so they leave a shadow on Earth's far side where they cannot get to. P waves do travel through the core, but because the waves that enter the core are refracted, there are also P wave shadow zones.

Exercise 9.2 Liquid Cores in Other Planets

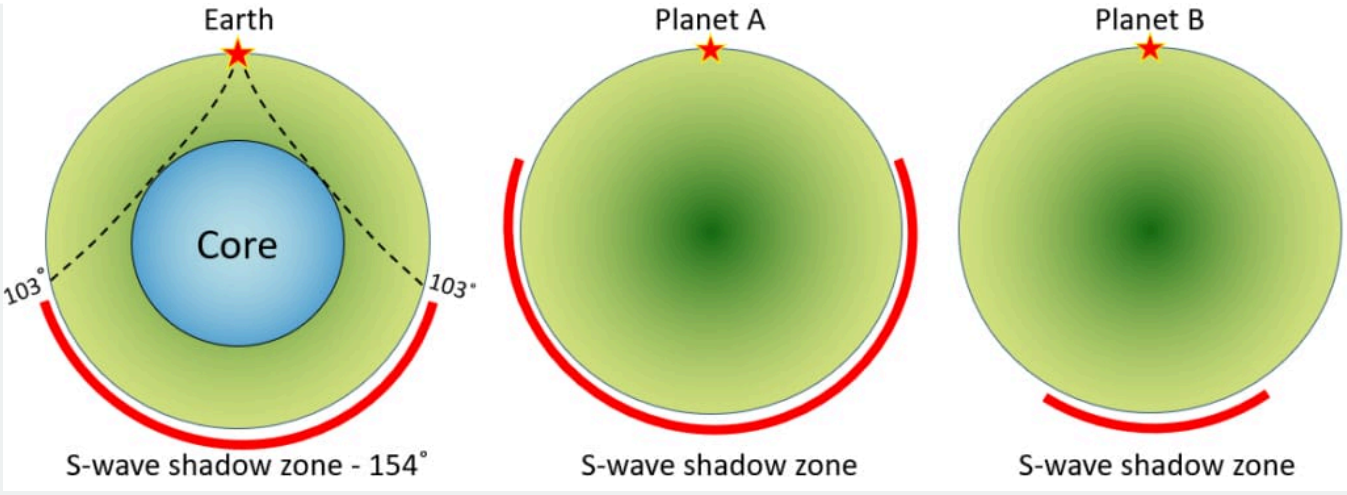


Figure 9.1.7

We know that other planets must have (or at least did have) liquid cores like ours, and we could use seismic data to find out how big they are. The S wave shadow zones on planets A and B are shown. Using the same method used for Earth (on the left), sketch in the outlines of the cores for these two other planets.

See Appendix 3 for Exercise 9.2 answers.

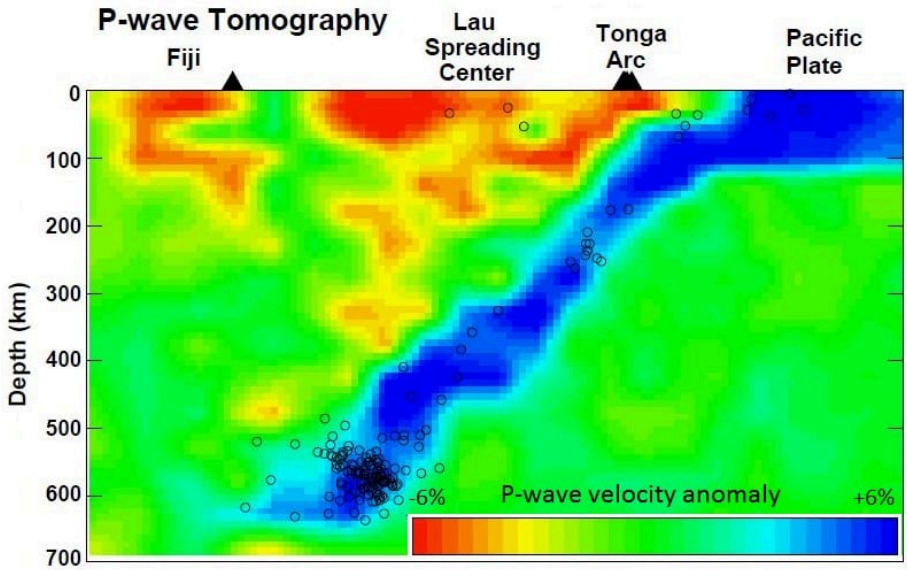


Figure 9.1.8 P-wave tomographic profile of area in the southern Pacific Ocean from southeast of Tonga to Fiji. Blue represents rock that has relatively high seismic velocities, while yellow and red represent rock with low velocities. Open circles are earthquakes used in the study.

Using data from many seismometers and hundreds of earthquakes, it is possible to create a two- or three-dimensional image of the seismic properties of part of the mantle. This technique is known as seismic tomography, and an example of the result is shown in Figure 9.1.8.

The Pacific Plate subducts beneath Tonga and appears in Figure 9.1.8 as a 100 kilometre thick slab of cold (blue-colored) oceanic crust that has pushed down into the surrounding hot mantle. The cold rock is more rigid than the surrounding hot mantle rock, so it is characterized by slightly faster seismic velocities. There is volcanism in the Lau spreading centre and also in the Fiji area, and the warm rock in these areas has slower seismic velocities (yellow and red colors).

Image descriptions

Figure 9.1.3 image description: Wave velocity in different materials in kilometers per second.

Material	S Wave (kilometers per second)	P Wave (kilometers per second)
Dry sand	0.1 to 0.4	0.4 to 1.3
Clay	0.2 to 0.6	0.6 to 1.6
Wet sand	0.7 to 0.8	1.5 to 2.2
Till	0.8 to 1.0	1.9 to 2.6
Mudstone	2.1 to 2.3	3.0 to 4.3
Sandstone	1.4 to 2.5	3.0 to 5.0
Limestone	2.4 to 3.1	4.2 to 5.8
Granite	3.0 to 3.7	4.9 to 5.9
Basalt	3.3 to 4.0	5.2 to 6.2

[Return to Figure 9.1.3]

Figure 9.1.4 image description: P-wave and S-wave velocity variations with depth in Earth.

Layer	Depth from surface (km)	S-Wave velocity (kilometers per second)	P-Wave velocity (kilometers per second)
Crust	0 to 30	3.0 to 4.6	5.3 to 7.0
Lithosphere	30 to 100	4.6 to 5.8	7.0 to 8.7
Asthenosphere	100 to 250	5.0 to 5.9	7.8 to 8.5
Mantle	250 to 2890	5. to 7.0	8.2 to 12.6
Outer core	2890 to 5100	0	8.0 to 10.1
Inner core	5100 to 6370	0	11.8 to 12.0

[Return to Figure 9.1.4]

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9.2: The Temperature of Earth's Interior

As we've discussed in the context of metamorphism, Earth's internal temperature increases with depth. However, as shown in Figure 9.2.1 (right), that rate of increase is not linear. The temperature gradient is around 15° to 30°C per kilometre within the upper 100 kilometers; it then drops off dramatically through the mantle, increases more quickly at the base of the mantle, and then increases slowly through the core. The temperature is around 1000°C at the base of the crust, around 3500°C at the base of the mantle, and around 5,000°C at Earth's centre. The temperature gradient within the lithosphere (upper 100 kilometers) is quite variable depending on the tectonic setting. Gradients are lowest in the central parts of continents, higher in the vicinity of subduction zones, and higher still at divergent boundaries.

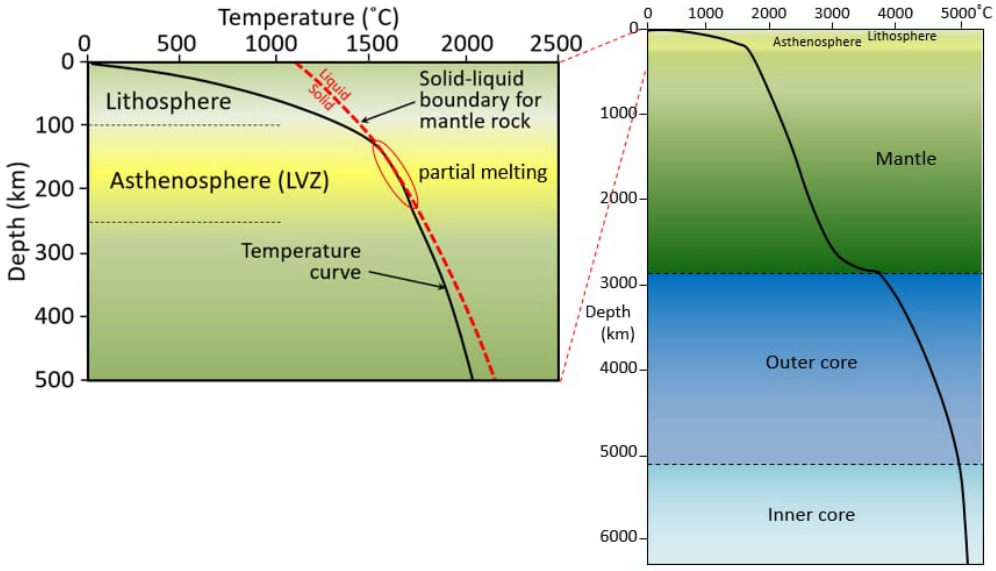


Figure 9.2.1 Right: generalized rate of temperature increase with depth within Earth. Temperature increases to the right, so the flatter the line, the steeper the temperature gradient. Our understanding of the temperature gradient comes from seismic wave information and knowledge of the melting points of Earth's materials. Left: Rate of temperature increase with depth in Earth's upper 500 kilometers, compared with the dry mantle rock melting curve (red dashed line). LVZ= low-velocity zone.

Figure 9.2.1 (left) shows a typical temperature curve for the upper 500 kilometers of the mantle in more detail, along with the melting curve for dry mantle rock. (Mantle rock will melt under conditions to the right of the dashed red line.) In general the mantle is not molten because the temperature lies to the left of the melting curve, but within the depth interval between 100 and 250 kilometers the temperature curve comes very close to the melting boundary for dry mantle rock. At these depths, therefore, mantle rock is either very nearly melted or partially melted. In some situations, where extra heat is present and the temperature line crosses over the melting line, or where water is present, it may be completely molten. This region of the mantle—the asthenosphere—is also known as the low-velocity zone because seismic waves are slowed within rock that is near its melting point. Below 250 kilometers the temperature stays on the left side of the melting line; in other words, the mantle is solid from here all the way down to the D" layer near the core-mantle-boundary.

The fact that the temperature gradient is much less in the main part of the mantle than in the lithosphere has been interpreted to indicate that the mantle is convecting, and therefore that heat from depth is being brought toward the surface faster than it would be with only heat conduction. As we'll see in Chapter 10, a convecting mantle is an key feature of plate tectonics.

The convection of the mantle is a product of the upward transfer of heat from the core to the lower mantle. As in a pot of soup on a hot stove (Figure 9.2.2), the material near the heat source becomes hot and expands, making it lighter than the material above. The force of buoyancy causes it to rise, and cooler material flows in from the sides. The mantle convects in this way because the heat transfer from below is not perfectly even, and also because, even though mantle material is solid rock, it is sufficiently plastic to slowly flow (at rates of centimeters per year) as long as a steady force is applied to it.

As in the soup pot example, Earth's mantle will no longer convect once the core has cooled to the point where there is not enough heat transfer to overcome the strength of the rock. This has already happened on smaller planets like Mercury and Mars, as well

as on Earth's Moon.

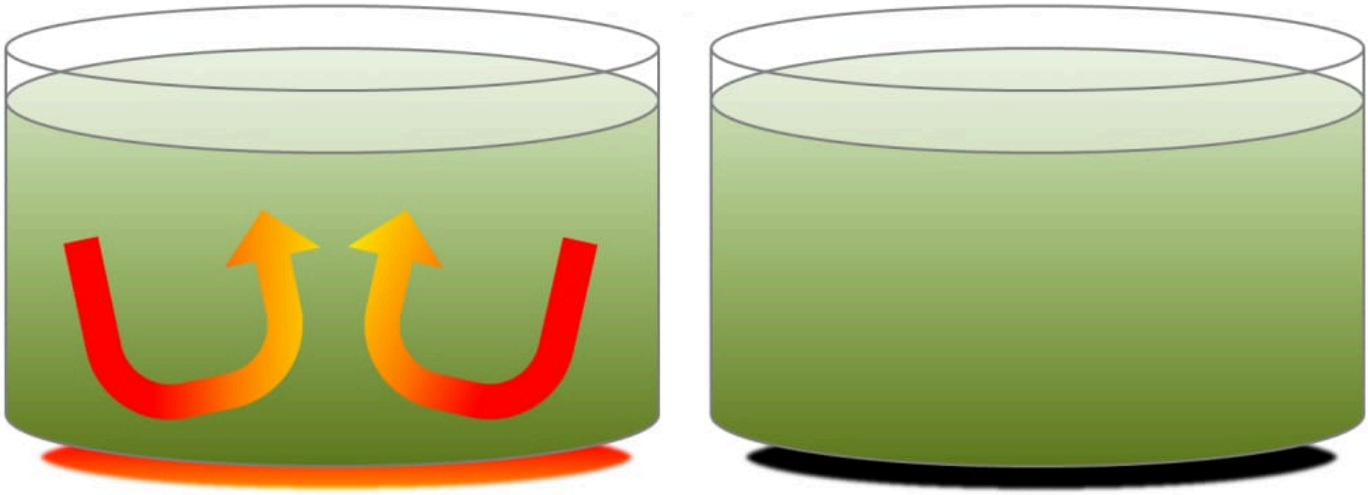


Figure 9.2.2 Convection in a pot of soup on a hot stove (left). As long as heat is being transferred from below, the liquid will convect. If the heat is turned off (right), the liquid remains hot for a while, but convection will cease.

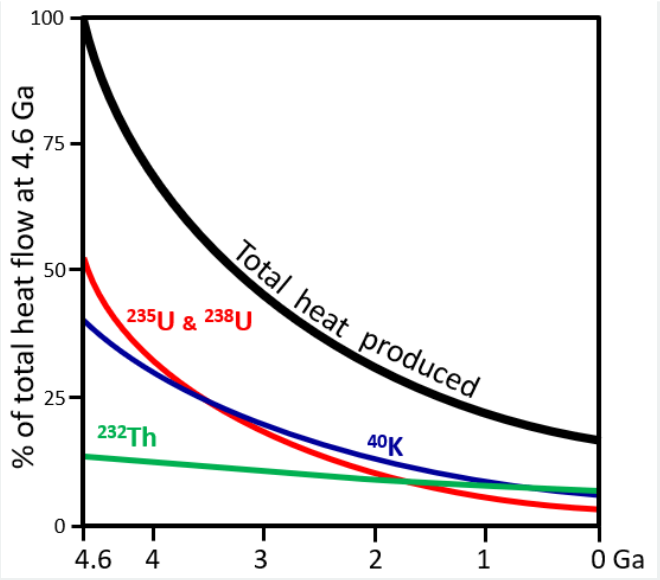


Figure 9.2.3 Heat flow on Earth from radioactive decay

The heat of Earth's interior comes from two main sources, each contributing about 50% of the heat. One of those is the frictional heat left over from the collisions of large and small particles that created Earth in the first place, plus the subsequent frictional heat of redistribution of material within Earth by gravitational forces (e.g., sinking of iron to form the core).

The other source is **radioactivity**, specifically the spontaneous radioactive decay of the isotopes ^{235}U, ^{238}U, ^{40}K, and ^{232}Th, which are primarily present in the mantle. As shown on Figure 9.2.3, the total heat produced that way has been decreasing over time (because these isotopes are getting used up), and is now roughly 25% of what it was when Earth formed. This means that Earth's interior is slowly becoming cooler.

Image Descriptions

Figure 9.2.1 image description: Temperature increase within the Earth.

Layer	Depth (kilometers)	Temperature increase (Celsius)	Temperature increase rate (Degrees per kilometre)
Lithosphere	0 to 100	0 to 1400	14
Asthenosphere	100 to 250	1400 to 1700	2
Mantle	250 to 1000	1700 to 2100	0.53
	1000 to 2000	2100 to 2600	0.5
	2000 to 2890	2600 to 3800	1.35
Outer Core	2890 to 4000	3800 to 4600	0.72
	4000 to 5100	4600 to 5000	0.36
Inner core	5100 to 6370	5000 to 5100	0.079

[Return to Figure 9.2.1]

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9.3: Earth's Magnetic Field

Heat is also being transferred from the solid inner core to the liquid outer core, and this leads to convection of the liquid iron of the outer core. Because iron is a metal and conducts electricity (even when molten), its motion generates a magnetic field.

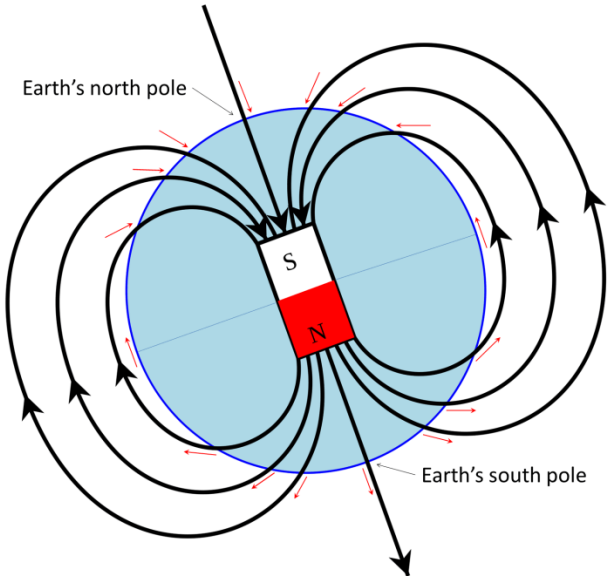


Figure 9.3.1 Depiction of Earth's magnetic field as a bar magnet coinciding with the core. The south pole of such a magnet points to Earth's North Pole. The red arrows represent the orientation of the magnetic field at various locations on Earth's surface.

Earth's magnetic field is defined by the North and South Poles that align generally with the axis of rotation (Figure 9.3.1). The lines of magnetic force flow into Earth in the northern hemisphere and out of Earth in the southern hemisphere. Because of the shape of the field lines, the magnetic force trends at different angles to the surface in different locations (red arrows of Figure 9.3.1). At the North and South magnetic poles, the force is vertical. Anywhere near to the equator the force is horizontal, and everywhere in between, the magnetic force is at some intermediate angle to the surface. As we'll see in Chapter 10, the variations in these orientations provide a critical piece of evidence to the understanding of continental drift as an aspect of plate tectonics.

Earth's magnetic field is generated within the outer core by the convective movement of liquid iron, but as we discovered in Chapter 8, the magnetic field is not stable over geological time. For reasons that are not completely understood, the magnetic field decays periodically and then becomes re-established. When it does re-establish, it may be oriented the way it was before the decay, or it may be oriented with the reversed polarity. Over the past 250 Ma, there have been a few hundred magnetic field reversals, and their timing has been anything but regular. The shortest ones that geologists have been able to define lasted only a few thousand years, and the longest one was more than 30 million years, during the Cretaceous (Figure 9.3.2).

Exercise 9.3 What would a magnetic dip meter tell you?

Regular compasses point only to the north magnetic pole, but if you have a magnetic dip meter you could also measure the angle of the magnetic field at your location in the up-and-down sense.

Using Figure 9.3.1 as a guide, describe where you'd be on Earth if the vertical angles are as follows:

1. Up at a shallow angle
2. Parallel to the ground
3. Down at a steep angle
4. Straight down

See Appendix 3 for Exercise 9.3 answers.

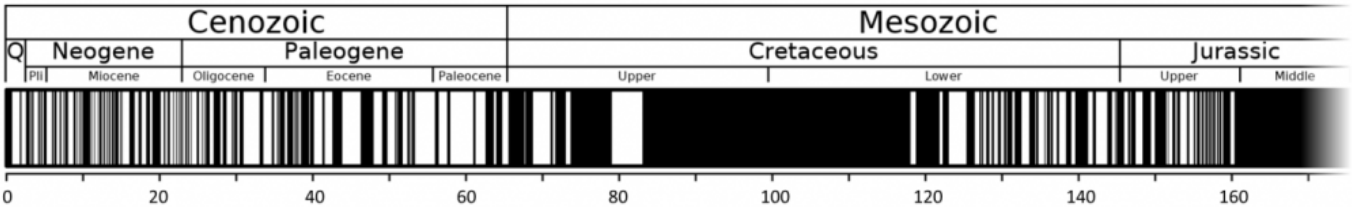


Figure 9.3.2 Magnetic field reversal chronology for the past 170 Ma. The first 5 Ma of the magnetic chronology are shown in more detail in Figure 9.3.3, although the time scale is in the opposite direction in that figure.

Changes in Earth's magnetic field have been studied using a mathematical model, and reversals have been shown to take place when the model was run to simulate a period of several hundred thousand years. The fact that field reversals took place shows that the model is a reasonably accurate representation of the Earth. According to the lead author of the study, Gary Glatzmaier, of University of California at Santa Cruz: "Our solution shows how convection in the fluid outer core is continually trying to reverse the field but that the solid inner core inhibits magnetic reversals because the field in the inner core can only change on the much longer time scale of diffusion. Only once in many attempts is a reversal successful, which is probably the reason why the times between reversals of the Earth's field are long and randomly distributed." A depiction of Earth's magnetic field lines during a stable period and during a reversal is shown in Figure 9.3.3. To read more about these phenomena see Glatzmaier's Geodynamo website.

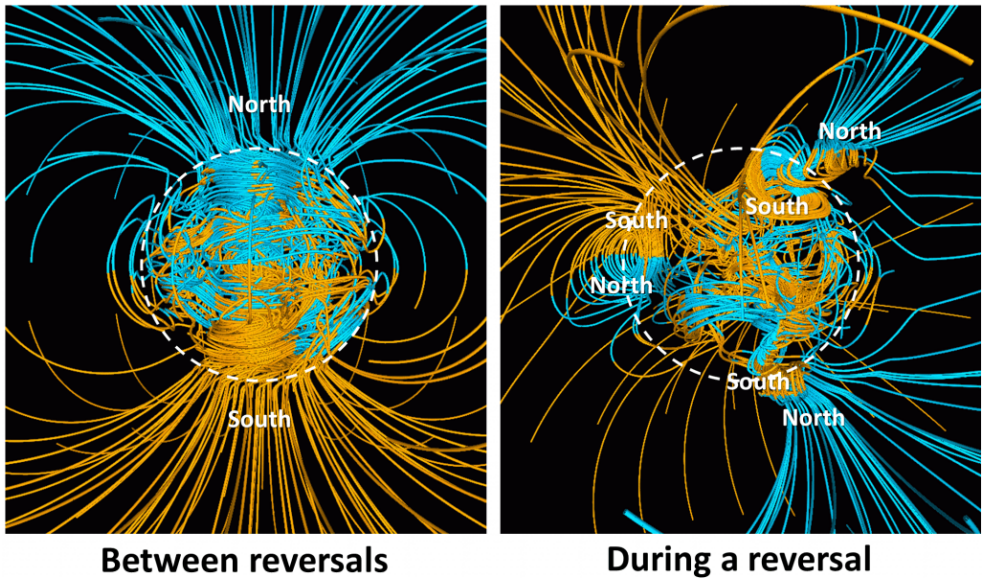


Figure 9.3.3 Depiction of Earth's magnetic field between reversals (left) and during a reversal (right). The lines represent magnetic field lines: blue where the field points toward Earth's centre and yellow where it points away. The rotation axis of Earth is vertical, and the outline of the core is shown as a dashed white circle.

Image Descriptions

Figure 9.3.3 image description: The earth's magnetic fields is normally very uniform with the magnetic field pointing towards the earth in the north and away from the earth in the south. During a reversal, the Earth's magnetic field becomes very convoluted. [Return to Figure 9.3.3]

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9.4: Isostasy

Theory holds that the mantle is able to convect because of its plasticity, and this plasticity also allows for another very important Earth process known as **isostasy**. The literal meaning of the word isostasy is "equal standstill," but the importance behind it is the principle that Earth's crust is *floating* on the mantle, like a raft floating in the water, rather than *resting* on the mantle like a raft sitting on the ground.

The relationship between the crust and the mantle is illustrated in Figure 9.4.1 On the right is an example of a non-isostatic relationship between a raft and solid concrete. It's possible to load the raft up with lots of people, and it still won't sink into the concrete. On the left, the relationship is an isostatic one between two different rafts and a swimming pool full of peanut butter. With only one person on board, the raft floats high in the peanut butter, but with three people, it sinks dangerously low. We're using peanut butter here, rather than water, because its viscosity more closely represents the relationship between the crust and the mantle. Although it has about the same density as water, peanut butter is much more viscous (stiff), and so although the three-person raft will sink into the peanut butter, it will do so quite slowly.

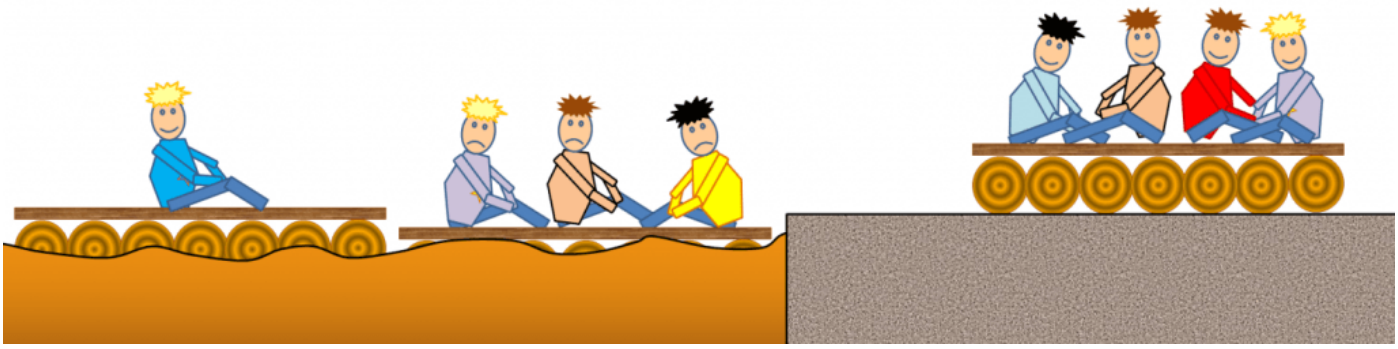


Figure 9.4.1 Illustration of a non-isostatic relationship between a raft and solid ground (right) and of isostatic relationships between rafts and peanut butter (left).

The relationship of Earth's crust to the mantle is similar to the relationship of the rafts to the peanut butter. The raft with one person on it floats comfortably high. Even with three people on it the raft is less dense than the peanut butter, so it floats, but it floats uncomfortably low for those three people. The crust, with an average density of around 2.6 grams per cubic centimetre (g/cm^3), is less dense than the mantle (average density of approximately 3.4 g/cm^3 near the surface, but more than that at depth), and so it is floating on the "plastic" mantle. When more weight is added to the crust, through the process of mountain building, it slowly sinks deeper into the mantle and the mantle material that was there is pushed aside (Figure 9.4.2, left). When that weight is removed by erosion over tens of millions of years, the crust rebounds and the mantle rock flows back (Figure 9.4.2, right).

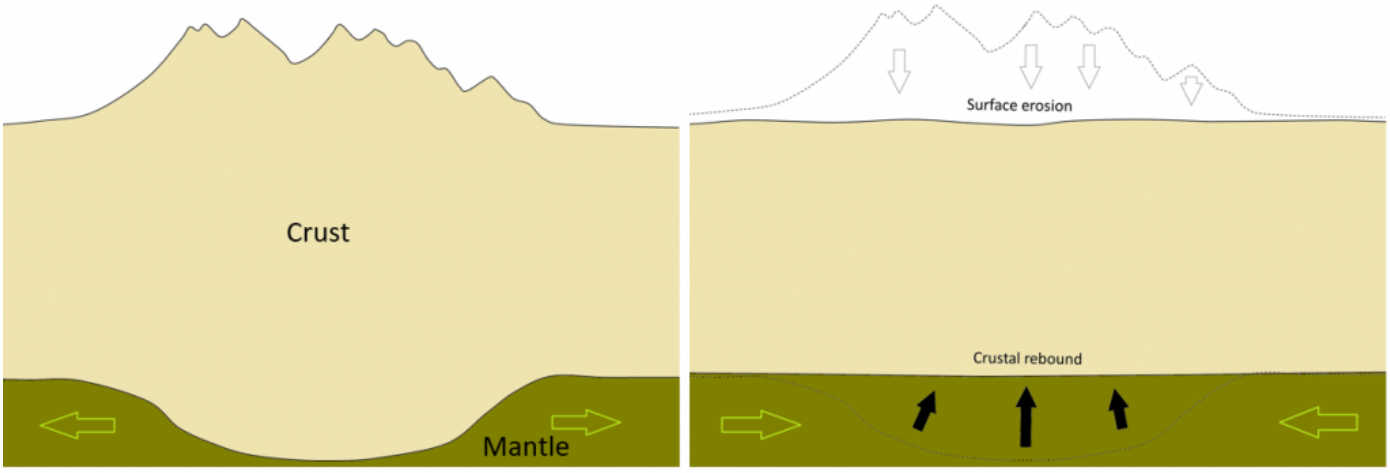


Figure 9.4.2 Illustration of the isostatic relationship between the crust and the mantle. Following a period of mountain building, mass has been added to a part of the crust, and the thickened crust has pushed down into the mantle (left). Over the following tens of millions of years, the mountain chain is eroded and the crust rebounds (right). The green arrows represent slow mantle flow.

The crust and mantle respond in a similar way to glaciation and deglaciation as they do to the growth and erosion of mountain ranges. Thick accumulations of glacial ice add weight to the crust, and as the mantle beneath is squeezed to the sides, the crust subsides. This process is illustrated for the current ice sheet on Greenland in Figure 9.4.3 (a and b). The Greenland Ice Sheet at this location is over 2,500 meters thick, and the crust beneath the thickest part has been depressed to the point where it is below sea level over a wide area. When the ice eventually melts, the crust and mantle will slowly rebound, but full rebound will likely take more than 10,000 years (Figure 9.4.3c).

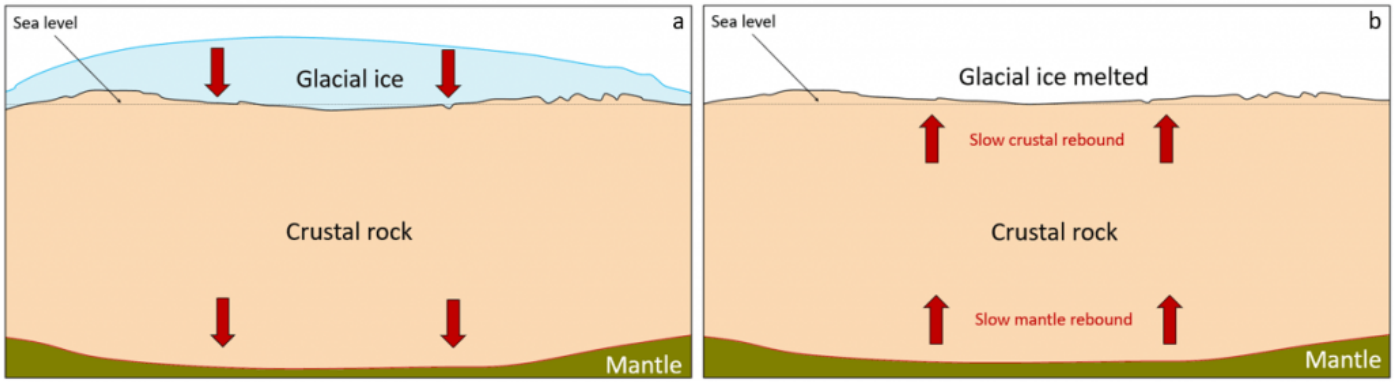


Figure 9.4.3 (a) A cross-section through the crust in the northern part of Greenland (The ice thickness is based on data from NASA and the Center for Remote Sensing of Ice Sheets, but the crust thickness is less than it should be for the sake of illustration.) The maximum ice thickness is over 2,500 m. The red arrows represent downward pressure on the mantle because of the mass of the ice. (b) Depiction of the situation after complete melting of the ice sheet, a process that could happen within 2,000 years if people and their governments continue to ignore climate change. The isostatic rebound of the mantle would not be able to keep up with this rate of melting, so for several thousand years the central part of Greenland would remain close to sea level, in some areas even below sea level.

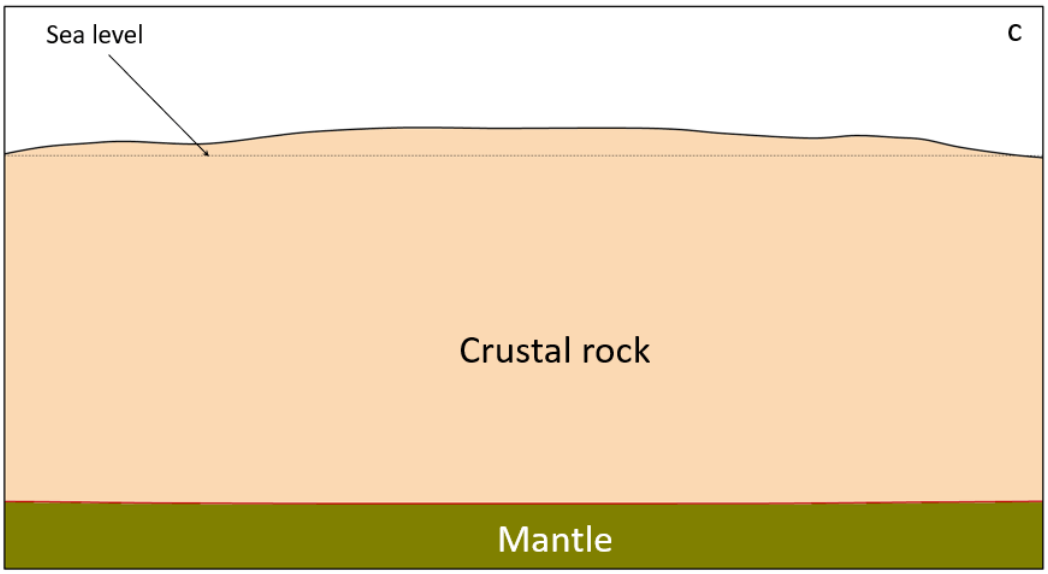


Figure 9.4.3c Depiction of the crust beneath a post-glacial Greenland once isostatic equilibrium is achieved. It is likely that complete rebound of the mantle would take more than 10,000 years.



Figure 9.4.4

You might be wondering how it is possible that Earth's mantle is rigid enough to break during an earthquake, and yet it convects and flows like a very viscous liquid. The explanation is that the mantle behaves as a non-Newtonian fluid, meaning that it responds differently to stresses depending on how quickly the stress is applied. A good example of this is the behaviour of the material known as Silly Putty, which can bounce and will break if you pull on it sharply, but will deform like a liquid if stress is applied slowly. In this photo, Silly Putty was placed over a hole in a glass tabletop, and in response to gravity, it slowly flowed into the hole. The mantle will flow when placed under the slow but steady stress of a growing (or melting) ice sheet.

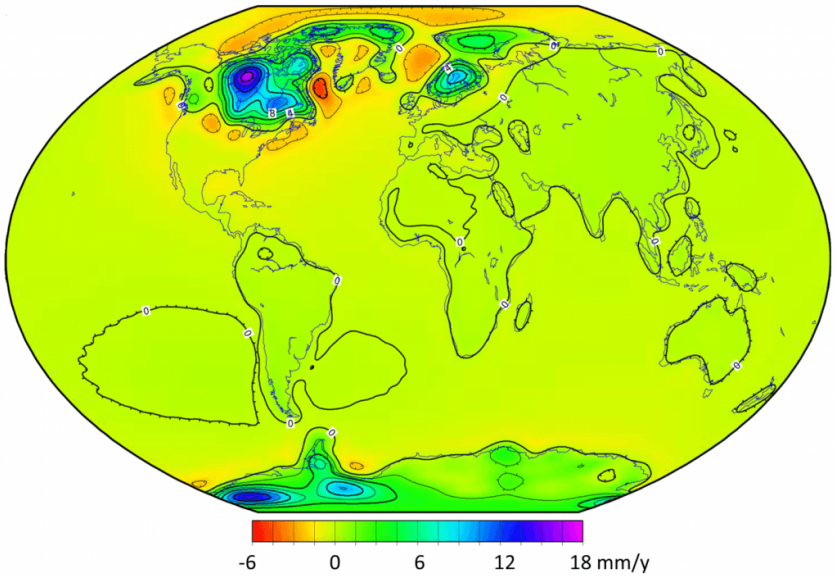


Figure 9.4.5 The current rates of post-glacial isostatic uplift (green, blue, and purple shades) and subsidence (yellow and orange). Subsidence is taking place where the mantle is slowly flowing back toward areas that are experiencing post-glacial uplift.

Large parts of Canada are still rebounding as a result of the loss of glacial ice over the past 12 ka, and as shown in Figure 9.4.5, other parts of the world are also experiencing isostatic rebound. The highest rate of uplift is in within a large area to the west of Hudson Bay, which is where the Laurentide Ice Sheet was the thickest (over 3,000 m). Ice finally left this region around 8,000 years ago, and the crust is currently rebounding at a rate of nearly 2 centimeters per year. Strong isostatic rebound is also occurring in northern Europe where the Fenno-Scandian Ice Sheet was thickest, and in the eastern part of Antarctica, which also experienced significant ice loss during the Holocene.

There are also extensive areas of subsidence surrounding the former Laurentide and Fenno-Scandian Ice Sheets. During glaciation, mantle rock flowed away from the areas beneath the main ice sheets, and this material is now slowly flowing back, as illustrated in Figure 9.4.3b.

Exercise 9.4 Rock density and isostasy

The densities (also known as "specific gravity") of a number of common minerals are given in Table 9.1.

Table 9.1 Densities of common minerals.

Mineral	Density (grams per cubic centimetre, g/cm^3)
Quartz	2.65
Feldspar	2.63
Amphibole	3.25
Pyroxene	3.4
Olivine	3.3

The following table provides the approximate proportions of these minerals in the continental crust (typified by granite), oceanic crust (mostly basalt), and mantle (mainly the rock known as peridotite). Assuming that you have 1,000 cm^3 of each rock type, estimate the respective rock-type densities. For each rock type, you will need to multiply the volume of the different minerals in the rock by their density, and then add those numbers to get the total weight for 1,000 cm^3 of that rock. The density is that number divided by 1,000. The continental crust is done for you.

Table 9.2 Determine the density of different kinds of crusts

Rock Type	Volumes of individual minerals in 1000 cm^3.	Grams of individual minerals in 1000 cm^3	Total Weight (grams)	Density (grams per cubic centimetre, g/cm^3)
Continental Crust (Granite)	Quartz – 180 cm^3 Feldspar – 760 cm^3 Amphibole – 70 cm^3	Quartz – 477 g Feldspar – 1999 g Amphibole – 277 g	2703 g	2.70
Oceanic Crust (Basalt)	Feldspar – 450 cm^3 Amphibole – 50 cm^3 Pyroxene – 500 cm^3	Feldspar – Amphibole – Pyroxene –		
Mantle (Peridotite)	Pyroxene – 450 cm^3 Olivine – 550 cm^3	Pyroxene – Olivine –		

If continental crust (represented by granite) and oceanic crust (represented by basalt) are like rafts floating on the mantle, what does this tell you about how high or low they should float?

This concept is illustrated in Figure 9.4.6. The dashed line is for reference, showing points at equal distance from Earth's centre.

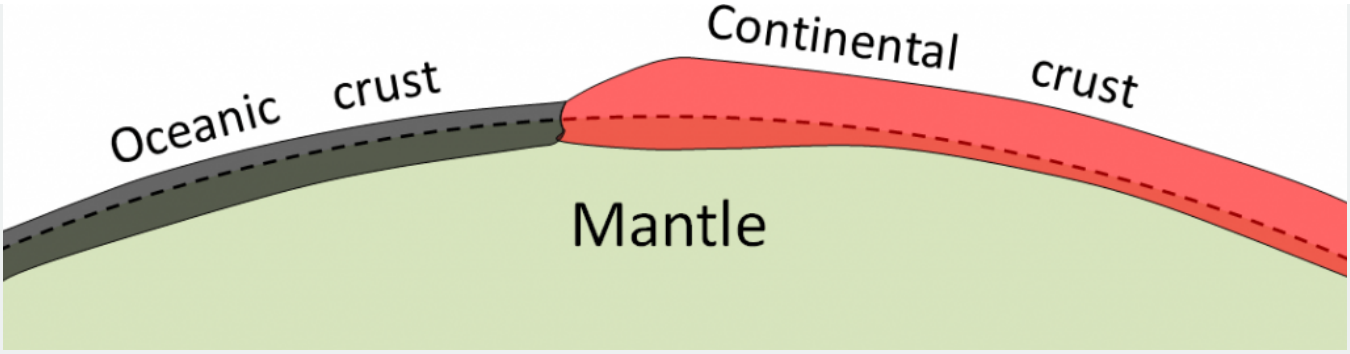


Figure 9.4.6

See Appendix 3 for Exercise 9.4 answers.

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9.5: Summary

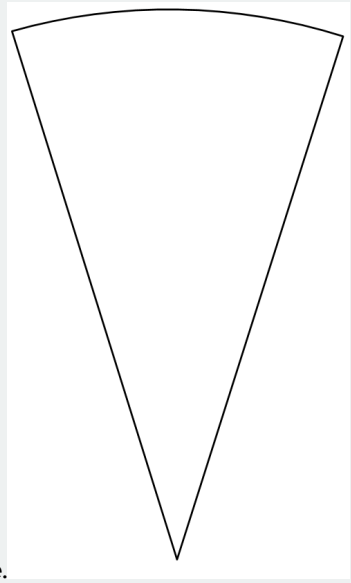
The topics covered in this chapter can be summarized as follows:

Section	Summary
9.1 Understanding Earth Through Seismology	Seismic waves that travel through Earth are either P-waves (compression, or “push” waves) or S-waves (shear waves). P-waves are faster than S-waves, and can pass through fluids. By studying seismic waves, we can discover the nature and temperature characteristics of the various parts of Earth’s interior.
9.2 The Temperature of Earth’s Interior	Earth’s temperature increases with depth (to around 5000°C at the centre), but there are significant variations in the rate of temperature increase. These variations are related to differences in composition and the existence of convection in the mantle and liquid part of the core.
9.3 Earth’s Magnetic Field	Because of outer-core convection, Earth has a magnetic field. The magnetic force directions are different at different latitudes. The polarity of the field is not constant, and has flipped from “normal” (as it is now) to reversed and back to normal hundreds of times in the past.
9.4 Isostasy	The “plastic” nature of the mantle, which allows for mantle convection, also determines the nature of the relationship between the crust and the mantle. The crust floats on the mantle in an isostatic relationship. Where the crust becomes thicker because of mountain building, it pushes farther down into the mantle. Oceanic crust, being heavier than continental crust, floats lower on the mantle.

Questions for Review

Answers to Review Questions can be found in Appendix 2.

1. What parts of Earth are most closely represented by typical stony meteorites and typical iron meteorites?
2. On the below diagram draw (from memory) and label the approximate locations of the following boundaries: crust/mantle, mantle/core, outer core/inner core.



3. Describe the important differences between P-waves and S-waves.
4. Why does P-wave velocity decrease dramatically at the core-mantle boundary?
5. Why do both P-waves and S-waves gradually bend as they move through the mantle?
6. What is the evidence for mantle convection, and what is the mechanism that causes it?
7. Where and how is Earth’s magnetic field generated?

8. When were the last two reversals of Earth's magnetic field?
9. What property of the mantle is essential for the isostatic relationship between the crust and the mantle?
10. How would you expect the depth to the crust-mantle boundary in the area of the Rocky Mountains to differ from that in central Saskatchewan?
11. As you can see in Figure 9.22, British Columbia is still experiencing weak post-glacial isostatic uplift, especially in the interior, but also along the coast. Meanwhile offshore areas are experiencing weak isostatic subsidence. Why?

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10: PLATE TECTONICS

10.1: Alfred Wegener- The Father of Plate Tectonics
10.2: Global Geological Models of the Early 20th Century
10.3: Geological Renaissance of the Mid-20th Century
10.4: Plate, Plate Motions, and Plate Boundary Processes
10.5: Mechanisms for Plate Motion
10.6: Summary

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10.1: Alfred Wegener- The Father of Plate Tectonics



Figure 10.1.1 Alfred Wegener a few years before his death in 1930.

Alfred Wegener (1880-1930) (Figure 10.1.1) earned a PhD in astronomy at the University of Berlin in 1904, but he had always been interested in geophysics and meteorology and spent most of his academic career working in meteorology. In 1911 he happened on a scientific publication that included a description of the existence of matching Permian-aged terrestrial fossils in various parts of South America, Africa, India, Antarctica, and Australia (Figure 10.1.2).

Wegener concluded that this distribution of terrestrial organisms could only exist if these continents were joined together during the Permian, and he coined the term **Pangea** ("all land") for the supercontinent that he thought included all of the present-day continents.

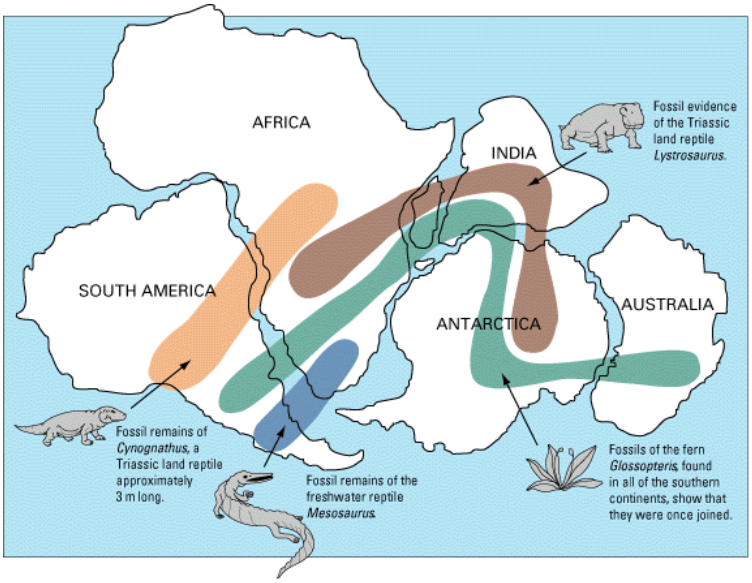


Figure 10.1.2 The distribution of several Permian terrestrial fossils that are present in various parts of continents that are now separated by oceans.

Wegener pursued his theory with determination—combing the libraries, consulting with colleagues, and making observations—looking for evidence to support it. He relied heavily on matching geological patterns across oceans, such as

sedimentary strata in South America matching those in Africa (Figure 10.1.3), North American coalfields matching those in Europe, and the mountains of Atlantic Canada matching those of northern Britain—both in morphology and rock type.

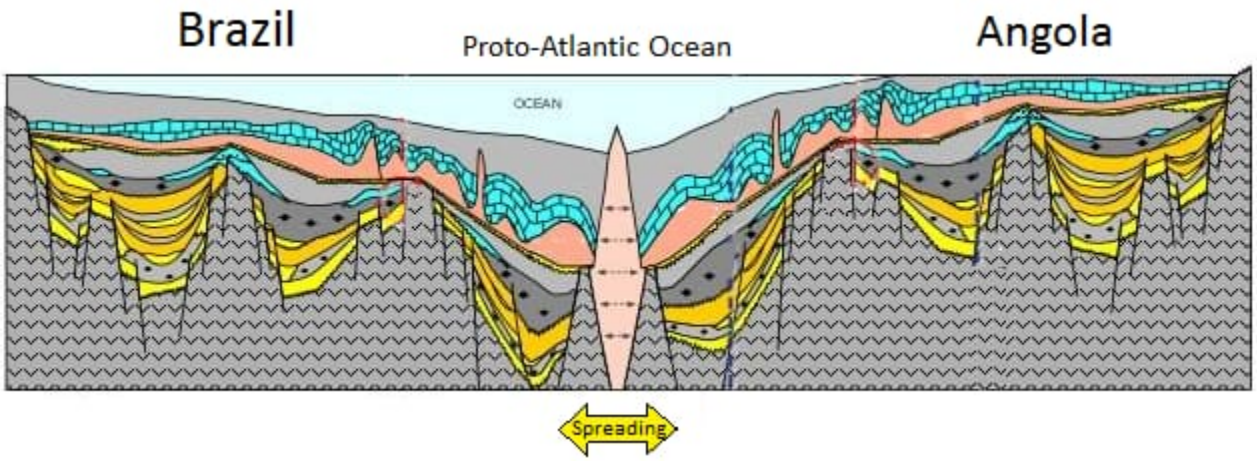


Figure 10.1.3 A cross-section showing the geological similarities between parts of Brazil (South America) on the left and Angola (Africa) on the right. The pink layer is a salt deposit, which is now known to be common in areas of continental rifting.

Wegener referred to the evidence for the Carboniferous and Permian (~300 Ma) Karoo Glaciation in South America, Africa, India, Antarctica, and Australia (Figure 10.1.4). He argued that this could only have happened if these continents were once all connected as a single supercontinent. He also cited evidence (based on his own astronomical observations) that showed that the continents were moving with respect to each other, and determined a separation rate between Greenland and Scandinavia of 11 meters per year, although he admitted that the measurements were not accurate. In fact they weren't even close—the separation rate is actually about 2.5 centimeters per year!



Figure 10.1.4 The distribution of the Carboniferous and Permian Karoo Glaciation (outlined in blue).

Wegener first published his ideas in 1912 in a short book called *Die Entstehung der Kontinente* (*The Origin of Continents*), and then in 1915 in *Die Entstehung der Kontinente und Ozeane* (*The Origin of Continentsand Oceans*). He revised this book several times up to 1929. It was translated into French, English, Spanish, and Russian in 1924.

In fact the continental fits were not perfect and the geological matchups were not always consistent, but the most serious problem of all was that Wegener could not conceive of a credible mechanism for moving the continents around. It was understood by this time that the continents were primarily composed of **sialic** material (**SIAL**: silicon and aluminum dominated, similar to "felsic"), and that the ocean floors were primarily **simatic** (**SIMA**: silicon and magnesium dominated, similar to "mafic"). Wegener proposed that the continents were like icebergs floating on the heavier SIMA crust, but the only forces that he could invoke to

propel continents around were *poleflucht,* the effect of Earth's rotation pushing objects toward the equator, and the lunar and solar tidal forces, which tend to push objects toward the west. It was quickly shown that these forces were far too weak to move continents, and without any reasonable mechanism to make it work, Wegener's theory was quickly dismissed by most geologists of the day.

Alfred Wegener died in Greenland in 1930 while carrying out studies related to glaciation and climate. At the time of his death, his ideas were tentatively accepted by only a small minority of geologists, and soundly rejected by most. However, within a few decades that was all to change. For more about his extremely important contributions to Earth science, visit the NASA website to see a collection of articles on Alfred Wegener.

Image Descriptions

Figure 10.1.2 **image description:** Fossils found across different continents suggest that these continents were once joined as a super-continent. Fossil remains of *Cynognathus* (a terrestrial reptile) and *Mesosaurus* (a freshwater reptile) have been found in South America and Africa. Fossil evidence of the *Lystrosaurus*, a land reptile from the Triassic period, has been found in India, Africa, and Antarctica. Fossils of the fern *Glossopteris* have been found in Australia, Antarctica, India, Africa, and South America. When you position these continents so they fit together, the areas where these fossils were found line up. [Return to Figure 10.1.2]

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- Figure 10.1.2: "Snider-Pellegrini Wegener fossil map" by Osvaldocangaspadilla. Public domain.
- Figure 10.1.3: © Steven Earle. CC BY. Based on "Angola -Brazil sub-sea geology" by Cobalt International Energy can be found at U.S. Energy Information Administration: Country Analysis Brief: Angola (May 2016) [PDF].
- Figure 10.1.4: "Karoo Glaciation" © GeoPotinga. Adapted by Steven Earle. CC BY-SA.

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10.2: Global Geological Models of the Early 20th Century

The untimely death of Alfred Wegener didn't solve any problems for those who opposed his ideas because they still had some inconvenient geological truths to deal with. One of those was explaining the distribution of terrestrial species across five continents that are currently separated by hundreds or thousands of kilometers of ocean water (Figure 10.2.2), and another was explaining the origin of extensive fold-belt mountains, such as the Appalachians, the Alps, the Himalayas, and the Canadian Rockies.

Before we go any further, it is important to know what was generally believed about global geology before plate tectonics. At the beginning of the 20th century, geologists had a good understanding of how most rocks were formed and understood their relative ages through interpretation of fossils, but there was considerable controversy regarding the origin of mountain chains, especially fold-belt mountains. At the end of the 19th century, one of the prevailing views on the origin of mountains was the theory of **contractionism**—the idea that since Earth is slowly cooling, it must also be shrinking. In this scenario, mountain ranges had formed like the wrinkles on a dried-up apple, and the oceans had submerged parts of former continents. While this theory helped to address the dilemma of the terrestrial fossils, it came with its own set of problems, one being that the amount of cooling couldn't produce the necessary amount of shrinking, and the other being the principle of isostasy (which had already been around for several decades), which wouldn't allow continents to sink. (See Section 9.4 for a review of the important principle of isostasy.)

Another widely held view was **permanentism**, in which it was believed that the continents and oceans have always been generally as they are today. This view incorporated a mechanism for creation of mountain chains known as the **geosyncline** theory. A geosyncline is a thick deposit of sediments and sedimentary rocks, typically situated along the edge of a continent (Figure 10.2.1).

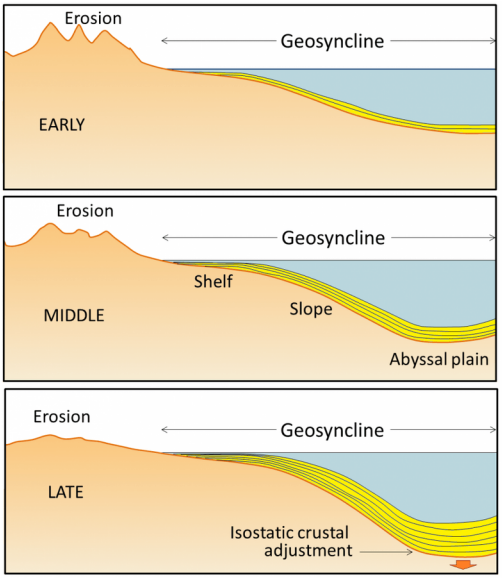


Figure 10.2.1 The development of a geosyncline along a continental margin. (Note that a geosyncline is not related to a syncline, which is a downward fold in sedimentary rocks.)

The idea of geosynclines developing into fold-belt mountains originated in the middle of the 19th century, proposed first by James Hall and later elaborated by Dwight Dana, both of whom worked extensively in the Appalachian Mountains of the eastern United States. The process of converting a geosyncline into a mountain belt was never really adequately explained, although it was widely believed that mountain belts formed when geosynclines were compressed by forces pushing from either side. The problem is that, without the lateral forces related to plate tectonics, no one was able to adequately describe what would do the pushing. The sediments that accumulate within a geosyncline are derived from erosion of the adjacent continent. Geosynclinal sediments—which eventually turn into sedimentary rocks—may be many thousands of meters thick. As they accumulate, they push down the pre-existing crustal rocks. Extensive geosynclinal deposits exist around much of the coastline of most of the continents; there is a large geosyncline along the eastern edge of North America.

Proponents of the geosyncline theory of mountain formation—and there were many well into the 1960s—also had the problem of

explaining the intercontinental terrestrial fossil matchups. The simple explanation was that there were “land bridges” across the Atlantic along which animals and plants could migrate back and forth. One proponent of this idea was the American naturalist Ernest Ingersoll. Referring to evidence of past climate changes, Ingersoll contributed the following to the *Encyclopedia Americana* in 1920: “The most interesting feature of these changes, however, is that by which, now and again, the Old World was connected with the New by necks or spaces of land, known as “land-bridges”; especially as these permitted an interchange of plants and animals, giving to us many new ones from the other side of the ocean, including, finally, man himself.”[1]

There are serious problems with the land-bridge theory. One is that it is completely inconsistent with isostasy, and another is that there is no evidence of the remnants of the land bridges. The Atlantic Ocean is several thousand meters deep over wide areas, and so the underwater slopes leading up to a land bridge would have to have been at least tens of kilometers wide in most places, and many times that in others. A land bridge of that size would certainly have left some trace.

Exercise 10.1 Fitting the continents together

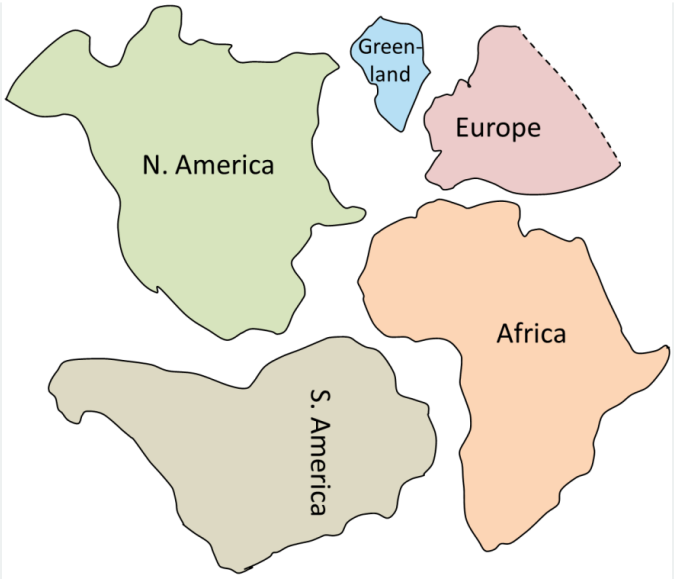


Figure 10.2.2

The main continents around the Atlantic Ocean are depicted here in the shapes that they might have had during the Mesozoic, including the extents of their continental shelves. Cut these shapes out and see how well you can fit them together in the positions that these areas occupied within Pangea. You can refer to a map of Pangea to help you make the fit.

See Appendix 3 for Exercise 10.1 answers.

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1. en.wikisource.org/wiki/The_Encyclopedia_Americana_(1920)/Land-Bridges_Across_the_Oceans ↵

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10.3: Geological Renaissance of the Mid-20th Century

As the mineral magnetite (Fe_3O_4) crystallizes from magma, it becomes magnetized with an orientation parallel to that of Earth's magnetic field at that time. This is called **remnant magnetism**. Rocks like basalt, which cool from a high temperature and commonly have relatively high levels of magnetite (up to 1 or 2%), are particularly susceptible to being magnetized in this way, but even sediments and sedimentary rocks, as long as they have small amounts of magnetite, will take on remnant magnetism because the magnetite grains gradually become reoriented following deposition. By studying both the horizontal and vertical components of the remnant magnetism, one can tell not only the direction to magnetic north at the time of the rock's formation, but also the latitude where the rock formed relative to magnetic north.

In the early 1950s, a group of geologists from Cambridge University, including Keith Runcorn, Ted Irving,[1] and several others, started looking at the remnant magnetism of Phanerozoic British and European volcanic rocks, and collecting **paleomagnetic** data. They found that rocks of different ages sampled from generally the same area showed quite different apparent magnetic pole positions (Figure 10.3.1). They initially assumed that this meant that Earth's magnetic field had, over time, departed significantly from its present position—which is close to the rotational pole.

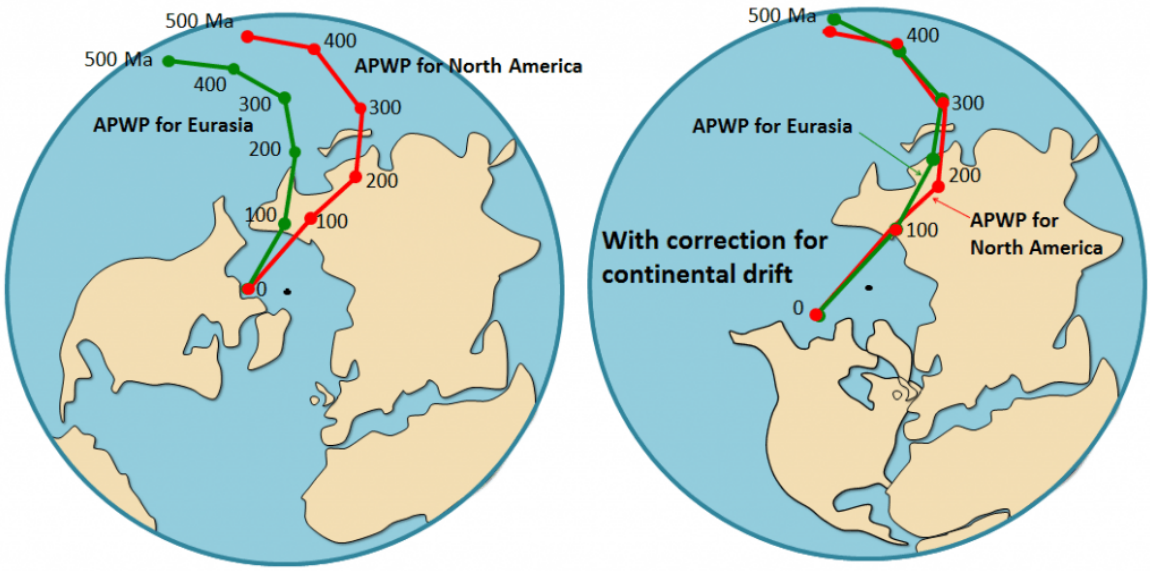


Figure 10.3.1 Apparent polar-wandering paths (APWP) for Eurasia and North America. The view is from the North Pole (black dot) looking down. The outer circle is the equator. In the diagram to the right the curve locations have been corrected taking continental drift into account.

The curve defined by the paleomagnetic data was called a **polar wandering path** because Runcorn and his students initially thought that their data represented actual movement of the magnetic poles (since geophysical models of the time suggested that the magnetic poles did not need to be aligned with the rotational poles). We now know that the magnetic data define movement of continents, and *not* of the magnetic poles, so we call it an *apparent* polar wandering path (APWP).

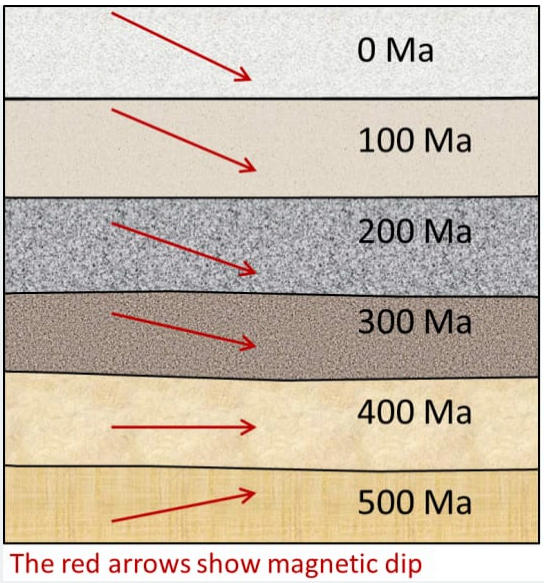


Figure 10.3.2

At around 500 Ma, what we now call Europe was south of the equator, and so European rocks formed then would have acquired an upward-pointing magnetic field orientation (see Figure 10.3.2 and Figure 10.3.2). Between then and now, Europe gradually moved north, and the rocks forming at various times acquired steeper and steeper *downward-pointing* magnetic orientations.When researchers evaluated magnetic data in this way in the 1950s, they plotted where the North Pole would have appeared to be based on the magnetic data and assumed that the continent was always where it is now. That means that the 500 Ma "apparent" north pole would have been somewhere in the South Pacific, and that over the following 500 million years it would have gradually moved north.Of course we now know that the magnetic poles don't move around much (although polarity reversals do take place) and that the reason Europe had a magnetic orientation characteristic of the southern hemisphere is that it was in the southern hemisphere at 500 Ma.Runcorn and colleagues soon extended their work to North America, and this also showed *apparent* polar wandering, but the results were not consistent with those from Europe. For example, the 200 Ma pole for North America plotted somewhere in China, while the 200 Ma pole for Europe plotted in the Pacific Ocean. Since there could only have been one pole position at 200 Ma, this evidence strongly supported the idea that North America and Europe had moved relative to each other since 200 Ma. Subsequent paleomagnetic work showed that South America, Africa, India, and Australia also have unique polar wandering curves. In 1956, Runcorn changed his mind and became a proponent of continental drift.This paleomagnetic work of the 1950s was the first new evidence in favour of continental drift, and it led a number of geologists to start thinking that the idea might have some merit. Nevertheless, for a majority of geologists working on global geology at the time, this type of evidence was not sufficiently convincing to get them to change their views.

During the 20th century, our knowledge and understanding of the ocean basins and their geology increased dramatically. Before 1900, we knew virtually nothing about the bathymetry and geology of the oceans. By the end of the 1960s, we had detailed maps of the topography of the ocean floors, a clear picture of the geology of ocean floor sediments and the solid rocks underneath them, and almost as much information about the geophysical nature of ocean rocks as of continental rocks.

Up until about the 1920s, ocean depths were measured using weighted lines dropped overboard. In deep water this is a painfully slow process and the number of soundings in the deep oceans was probably fewer than 1,000. That is roughly one depth sounding for every 350,000 square kilometers of the ocean. To put that in perspective, it would be like trying to describe the topography of British Columbia with elevation data for only a half a dozen points! The voyage of the *Challenger* in 1872 and the laying of trans-Atlantic cables had shown that there were mountains beneath the seas, but most geologists and oceanographers still believed that the oceans were essentially vast basins with flat bottoms, filled with thousands of meters of sediments.

Following development of acoustic depth sounders in the 1920s (Figure 10.3.3), the number of depth readings increased by many orders of magnitude, and by the 1930s, it had become apparent that there were major mountain chains in all of the world's oceans. During and after World War II, there was a well-organized campaign to study the oceans, and by 1959, sufficient bathymetric data had been collected to produce detailed maps of all the oceans (Figure 10.3.4).

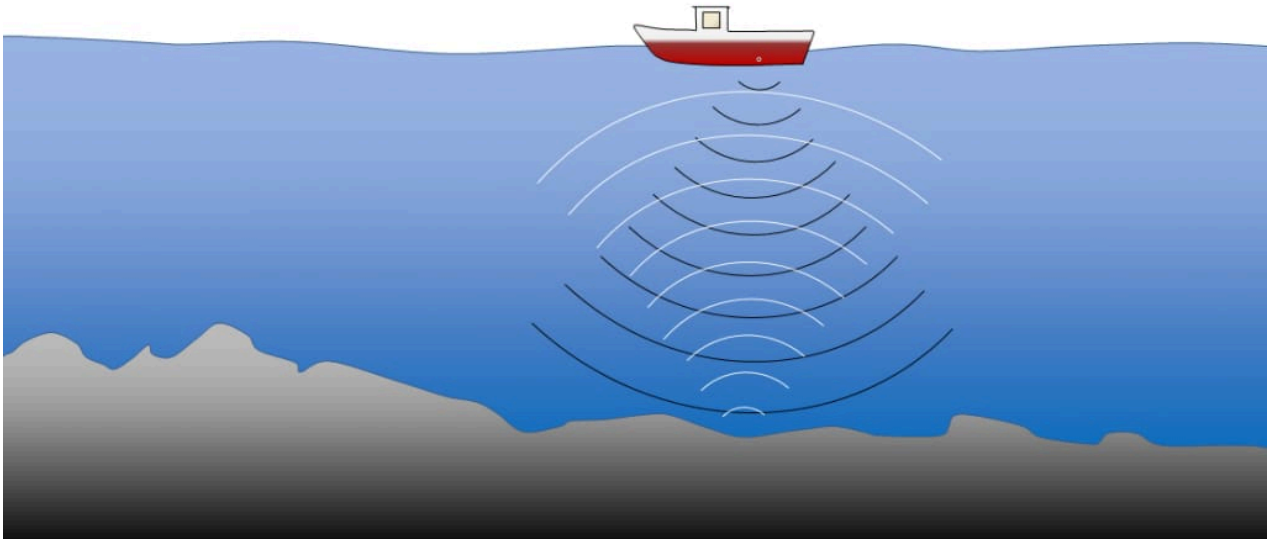


Figure 10.3.3 Depiction of a ship-borne acoustic depth sounder. The instrument emits a sound (black arcs) that bounces off the sea floor and returns to the surface (white arcs). The travel time is proportional to the water depth.

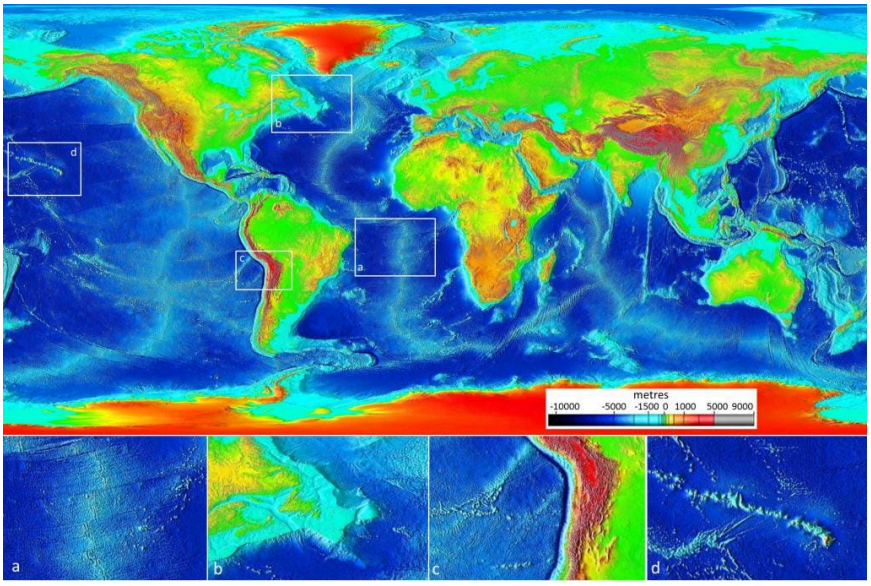


Figure 10.3.4 Ocean floor bathymetry (and continental topography). Inset (a): the mid-Atlantic ridge, (b): the Newfoundland continental shelf, (c): the Nazca trench adjacent to South America, and (d): the Hawaiian Island chain.

The important physical features of the ocean floor are:

- Extensive linear ridges (commonly in the central parts of the oceans) with water depths in the order of 2,000 to 3,000 m (Figure 10.3.4, inset a)
- Fracture zones perpendicular to the ridges (inset a)
- Deep-ocean plains at depths of 5,000 to 6,000 m (insets a and d)
- Relatively flat and shallow continental shelves with depths under 500 m (inset b)
- Deep trenches (up to 11,000 m deep), most near the continents (inset c)
- Seamounts and chains of seamounts (inset d)

Seismic reflection sounding involves transmitting high-energy sound bursts and then measuring the echos with a series of geophones towed behind a ship. The technique is related to *acoustic sounding* as described above; however, much more energy is transmitted and the sophistication of the data processing is much greater. As the technique evolved, and the amount of energy was increased, it became possible to *see through* the sea-floor sediments and map the bedrock topography and crustal thickness. Hence sediment thicknesses could be mapped, and it was soon discovered that although the sediments were up to several thousands of meters thick near the continents, they were relatively thin — or even non-existent — in the ocean ridge areas

(Figure 10.3.5). The seismic studies also showed that the crust is relatively thin under the oceans (5 km to 6 km) compared to the continents (30 km to 60 km) and geologically very consistent, composed almost entirely of basalt.

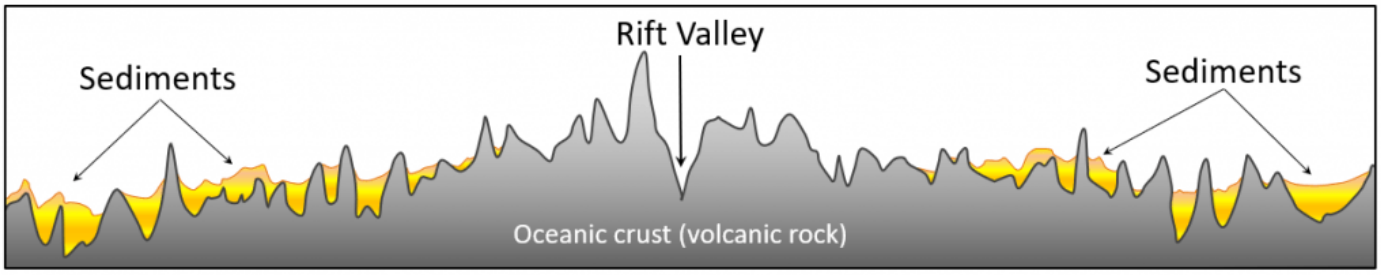


Figure 10.3.5 Topographic section at an ocean ridge based on reflection seismic data. Sediments are not thick enough to be detectable near the ridge, but get thicker on either side. The diagram represents approximately 50 km width, and has a 10x vertical exaggeration.

In the early 1950s, Edward Bullard, who spent time at the University of Toronto but is mostly associated with Cambridge University, developed a probe for measuring the flow of heat from the ocean floor. Bullard and colleagues found the rate to be higher than average along the ridges, and lower than average in the trench areas. Although Bullard was a plate-tectonics sceptic, these features were interpreted to indicate that there is convection within the mantle — the areas of high heat flow being correlated with upward convection of hot mantle material, and the areas of low heat flow being correlated with downward convection.

With developments of networks of seismographic stations in the 1950s, it became possible to plot the locations *and* depths of both major and minor earthquakes with great accuracy. It was found that there is a remarkable correspondence between earthquakes and both the mid-ocean ridges and the deep ocean trenches. In 1954 Gutenberg and Richter showed that the ocean-ridge earthquakes were all relatively shallow, and confirmed what had first been shown by Benioff in the 1930s — that earthquakes in the vicinity of ocean trenches were both shallow and deep, but that the deeper ones were situated progressively farther inland from the trenches (Figure 10.3.6).

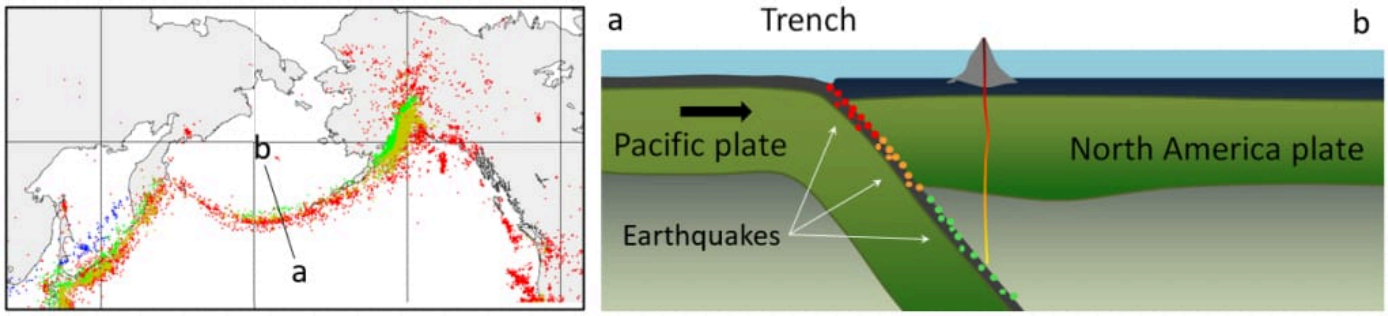


Figure 10.3.6 Cross-section through the Aleutian subduction zone with a depiction of the increasing depth of earthquakes “inshore” from the trench.

In the 1950s, scientists from the Scripps Oceanographic Institute in California persuaded the U.S. Coast Guard to include magnetometer readings on one of their expeditions to study ocean floor topography. The first comprehensive magnetic data set was compiled in 1958 for an area off the coast of B.C. and Washington State. This survey revealed a bewildering pattern of low and high magnetic intensity in sea-floor rocks (Figure 10.3.7). When the data were first plotted on a map in 1961, nobody understood them — not even the scientists who collected them. Although the patterns made even less sense than the stripes on a zebra, many thousands of kilometers of magnetic surveys were conducted over the next several years.



Figure 10.3.7 Pattern of sea-floor magnetism off of the west coast of British Columbia and Washington.

The wealth of new data from the oceans began to significantly influence geological thinking in the 1960s. In 1960, Harold Hess, a widely respected geologist from Princeton University, advanced a theory with many of the elements that we now accept as **plate tectonics**. He maintained some uncertainty about his proposal however, and in order to deflect criticism from mainstream geologists, he labelled it *geopoetry*. In fact, until 1962, Hess didn't even put his ideas in writing—except internally to the U.S. Navy (which funded his research)—but presented them mostly in lectures and seminars. Hess proposed that new sea floor was generated from mantle material at the ocean ridges, and that old sea floor was dragged down at the ocean trenches and re-incorporated into the mantle. He suggested that the process was driven by mantle convection currents, rising at the ridges and descending at the trenches (Figure 10.3.8). He also suggested that the less-dense continental crust did not descend with oceanic crust into trenches, but that colliding land masses were thrust up to form mountains. Hess's theory formed the basis for our ideas on **sea-floor spreading** and **continental drift,** but it did not deal with the concept that the crust is made up of specific **plates**. Although the Hess model was not roundly criticized, it was not widely accepted (especially in the U.S.), partly because it was not well supported by hard evidence.

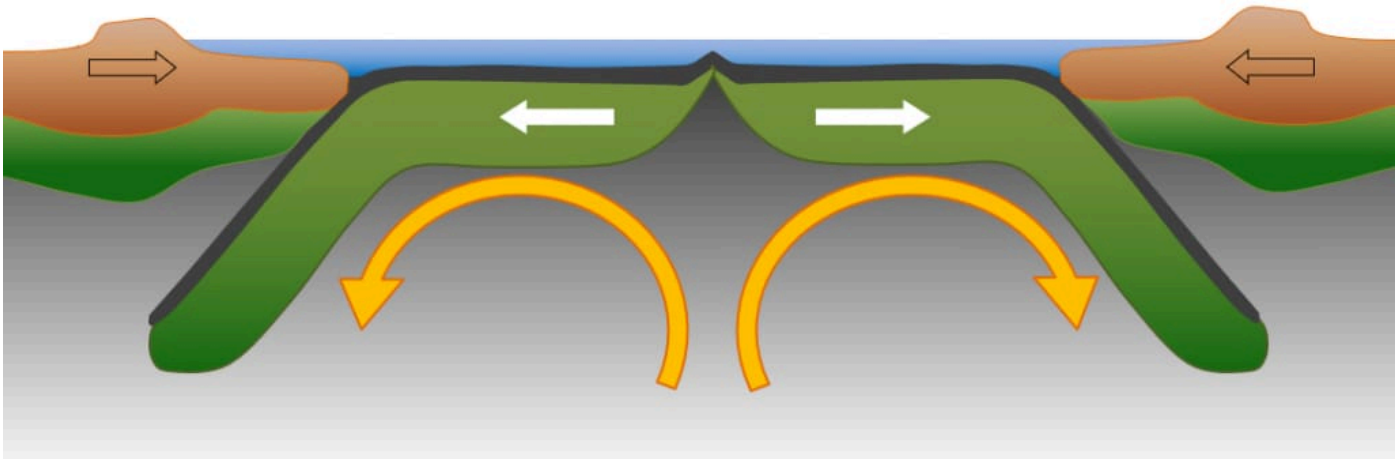


Figure 10.3.8 A representation of Harold Hess's model for sea-floor spreading and subduction.

Collection of magnetic data from the oceans continued in the early 1960s, but still nobody could explain the origin of the zebra-like patterns. Most assumed that they were related to variations in the composition of the rocks—such as variations in the amount of magnetite—as this is a common explanation for magnetic variations in rocks of the continental crust. The first real understanding of the significance of the striped anomalies was the interpretation by Fred Vine, a Cambridge graduate student. Vine was examining magnetic data from the Indian Ocean and, like others before, he noted the symmetry of the magnetic patterns with respect to the oceanic ridge.

At the same time, other researchers, led by groups in California and New Zealand, were studying the phenomenon of reversals in Earth's magnetic field. They were trying to determine when such reversals had taken place over the past several million years by analyzing the magnetic characteristics of hundreds of samples from basaltic flows. As discussed in Chapter 9, it is evident that Earth's magnetic field becomes weakened periodically and then virtually non-existent, before becoming re-established with the reverse polarity. During periods of reversed polarity, a compass would point south instead of north.

The time scale of magnetic reversals is irregular. For example, the present "normal" event, known as the Bruhnes magnetic chron, has persisted for about 780,000 years. This was preceded by a 190,000-year reversed event; a 50,000-year normal event known as Jaramillo; and then a 700,000-year reversed event (see Figure 10.3.3).

In a paper published in September 1963, Vine and his PhD supervisor Drummond Matthews proposed that the patterns associated with ridges were related to the magnetic reversals, and that oceanic crust created from cooling basalt during a *normal* event would have polarity aligned with the present magnetic field, and thus would produce a positive anomaly (a black stripe on the sea-floor magnetic map), whereas oceanic crust created during a *reversed* event would have polarity opposite to the present field and thus would produce a negative magnetic anomaly (a white stripe). The same idea had been put forward a few months earlier by Lawrence Morley, of the Geological Survey of Canada; however, his papers submitted earlier in 1963 to *Nature* and *The Journal of Geophysical Research* were rejected. Many people refer to the idea as the Vine-Matthews-Morley (VMM) hypothesis.

Vine, Matthews, and Morley were the first to show this type of correspondence between the relative widths of the stripes and the periods of the magnetic reversals. The VMM hypothesis was confirmed within a few years when magnetic data were compiled from spreading ridges around the world. It was shown that the same general magnetic patterns were present straddling each ridge, although the widths of the anomalies varied according to the spreading rates characteristic of the different ridges. It was also shown that the patterns corresponded with the chronology of Earth's magnetic field reversals. This global consistency provided strong support for the VMM hypothesis and led to rejection of the other explanations for the magnetic anomalies.

In 1963, J. Tuzo Wilson of the University of Toronto proposed the idea of a **mantle plume** or **hot spot**—a place where hot mantle material rises in a stationary and semi-permanent plume, and affects the overlying crust. He based this hypothesis partly on the distribution of the Hawaiian and Emperor Seamount island chains in the Pacific Ocean (Figure 10.3.9). The volcanic rock making up these islands gets progressively younger toward the southeast, culminating with the island of Hawaii itself, which consists of rock that is almost all younger than 1 Ma. Wilson suggested that a stationary plume of hot upwelling mantle material is the source of the Hawaiian volcanism, and that the ocean crust of the Pacific Plate is moving toward the northwest over this hot spot. Near the Midway Islands, the chain takes a pronounced change in direction, from northwest-southeast for the Hawaiian Islands and to nearly north-south for the Emperor Seamounts. This change is widely ascribed to a change in direction of the Pacific Plate moving over the stationary mantle plume, but a more plausible explanation is that the Hawaiian mantle plume has not actually been stationary throughout its history, and in fact moved at least 2,000 km south over the period between 81 and 45 Ma.[2]

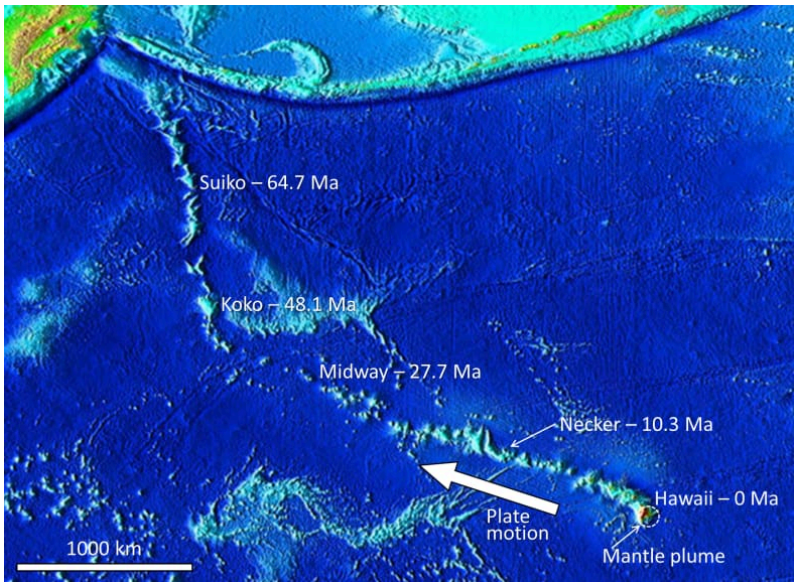


Figure 10.3.9 The ages of the Hawaiian Islands and the Emperor Seamounts in relation to the location of the Hawaiian mantle plume.

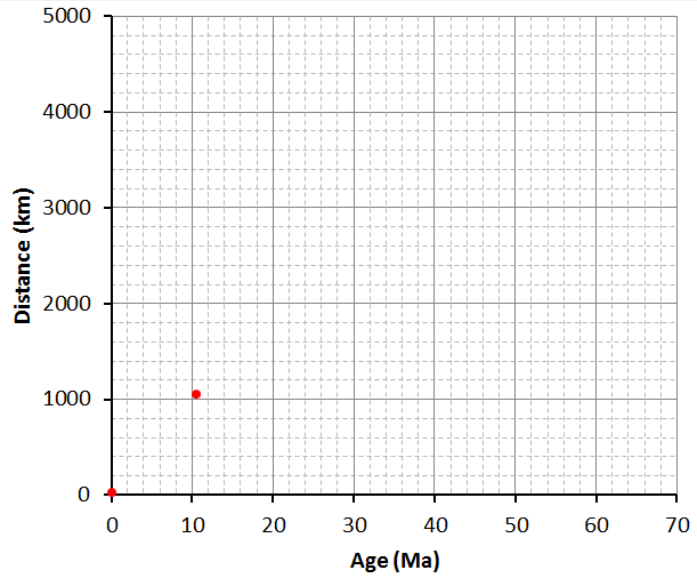
Exercise 10.2 Volcanoes and the Rate of Plate Motion

The Hawaiian and Emperor volcanoes shown in Figure 10.3.9 are listed in the table below along with their ages and their distances from the centre of the mantle plume under Hawaii (the Big Island).

Ages of Hawaiian and Emperor volcanoes and their distances from the centre of the mantle plume. Calculate their rate of movement in centimeters per year.

Island	Age	Distance	Rate
Hawaii	0 Ma	0 km	–
Necker	10.3 Ma	1,058 km	10.2 cm/y
Midway	27.7 Ma	2,432 km	
Koko	48.1 Ma	3,758 km	
Suiko	64.7 Ma	4,860 km	

Plot the data on the graph provided here, and use the numbers in the table to estimate the rates of plate motion for the Pacific Plate in cm/year. (The first two are plotted for you.)



See Appendix 3 for Exercise 10.2 answers.

There is evidence of many such mantle plumes around the world (Figure 10.3.10). Most are within the ocean basins—including places like Hawaii, Iceland, and the Galapagos Islands—but some are under continents. One example is the Yellowstone hot spot in the west-central United States, and another is the one responsible for the Anahim Volcanic Belt in central British Columbia. It is evident that mantle plumes are very long-lived phenomena, lasting for at least tens of millions of years, possibly for hundreds of millions of years in some cases.

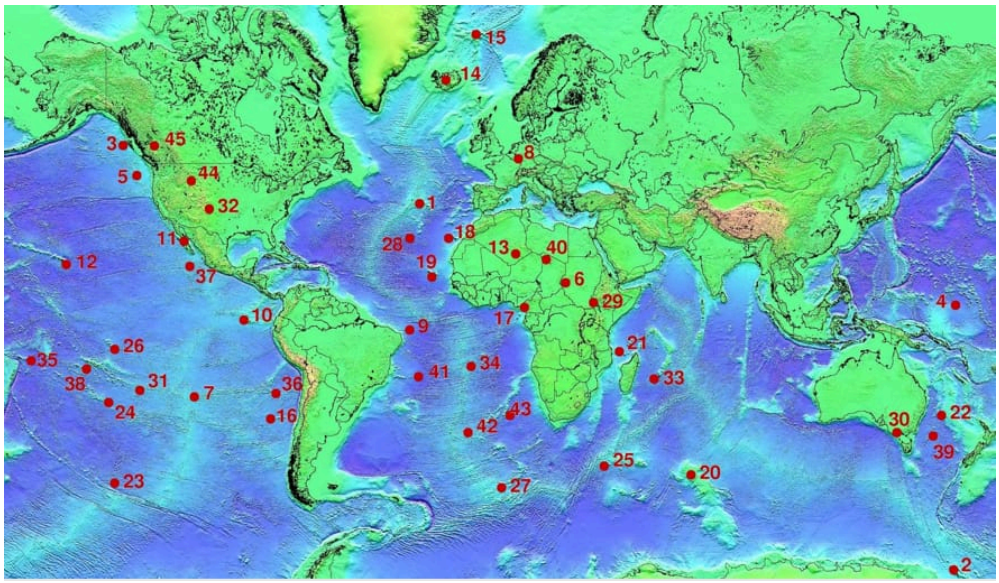


Figure 10.3.10 Mantle plume locations. Selected Mantle plumes: 1: Azores, 3: Bowie, 5: Cobb, 8: Eifel, 10: Galapagos, 12: Hawaii, 14: Iceland, 17: Cameroon, 18: Canary, 19: Cape Verde, 35: Samoa, 38: Tahiti, 42: Tristan, 44: Yellowstone, 45: Anahim

Although oceanic spreading ridges appear to be curved features on Earth's surface, in fact the ridges are composed of a series of straight-line segments, offset at intervals by faults perpendicular to the ridge (Figure 10.3.11). In a paper published in 1965, Tuzo Wilson termed these features **transform faults**. He described the nature of the motion along them, and showed why there are earthquakes only on the section of a transform fault between two adjacent ridge segments. The San Andreas Fault in California is a very long transform fault that links the southern end of the Juan de Fuca spreading ridge to the East Pacific Rise spreading ridges situated in the Gulf of California (see Figure 10.3.9). The Queen Charlotte Fault, which extends north from the northern end of the Juan de Fuca spreading ridge (near the northern end of Vancouver Island) toward Alaska, is also a transform fault.

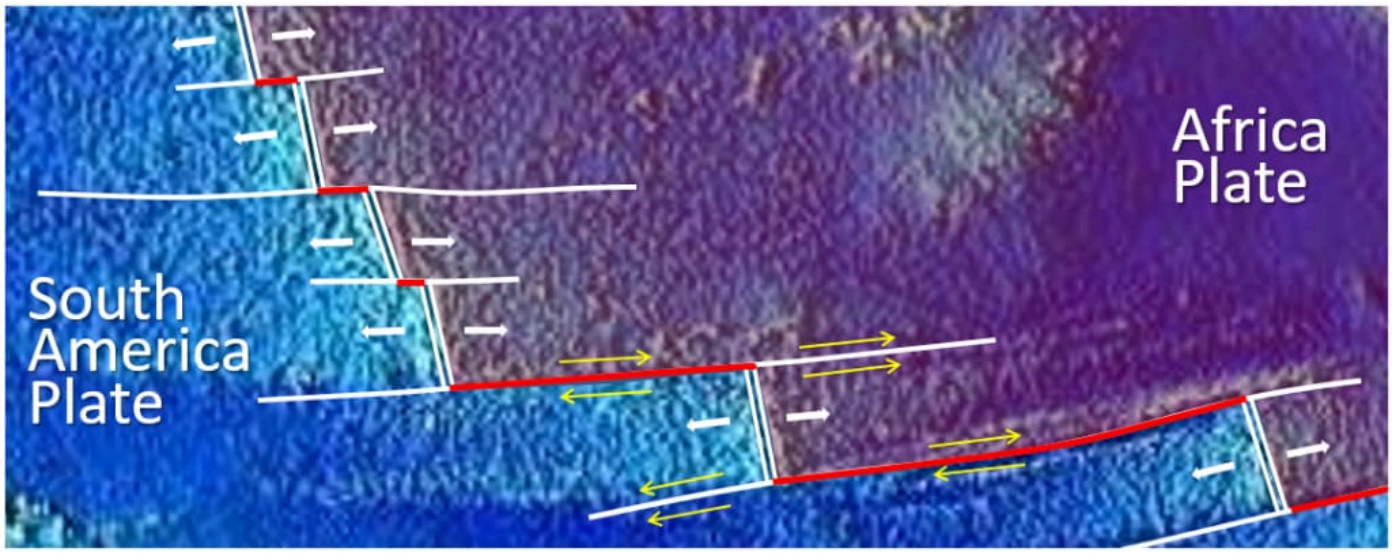


Figure 10.3.11 A part of the mid-Atlantic ridge near the equator. The double white lines are spreading ridges. The solid white lines are fracture zones. As shown by the yellow arrows, the relative motion of the plates on either side of the fracture zones can be similar (arrows pointing the same direction) or opposite (arrows pointing opposite directions). Transform faults (red lines) are in between the ridge segments, where the yellow arrows point in opposite directions.

In the same 1965 paper, Wilson introduced the idea that the crust can be divided into a series of rigid plates, and thus he is responsible for the term **plate tectonics**.

Exercise 10.3 Paper transform fault model

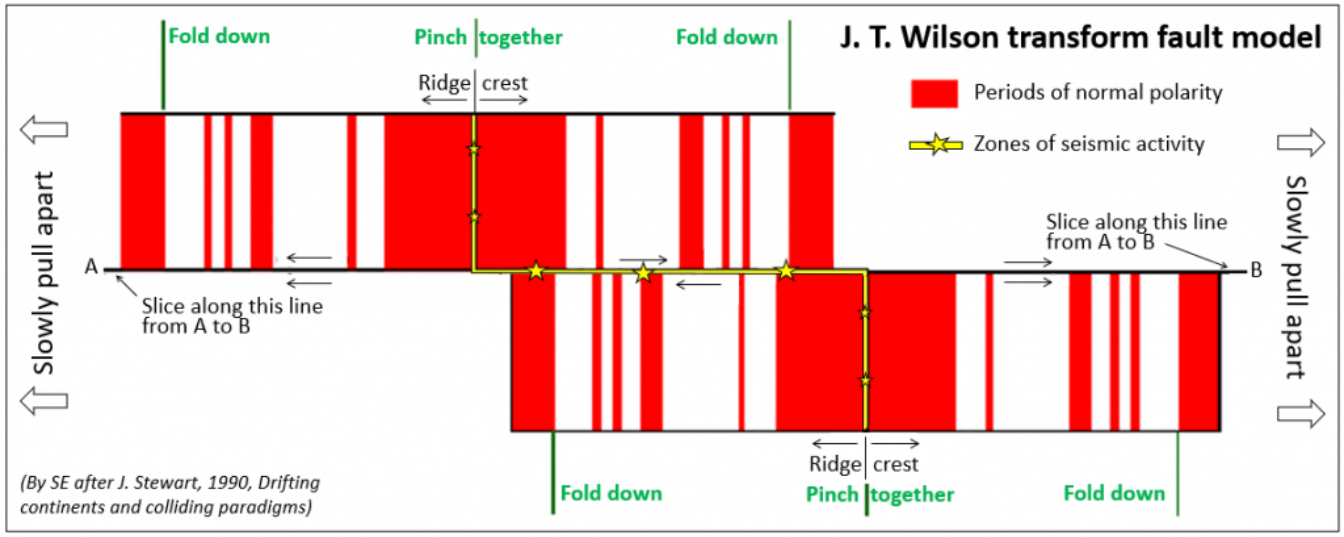


Figure 10.3.12

Tuzo Wilson used a paper model, a little bit like the one shown here, to explain transform faults to his colleagues. To use this model either print this page or download the image above and print that, then cut around the outside, and then slice along the line A-B (the fracture zone) with a sharp knife. Fold down the top half where shown, and then pinch together in the middle. Do the same with the bottom half.

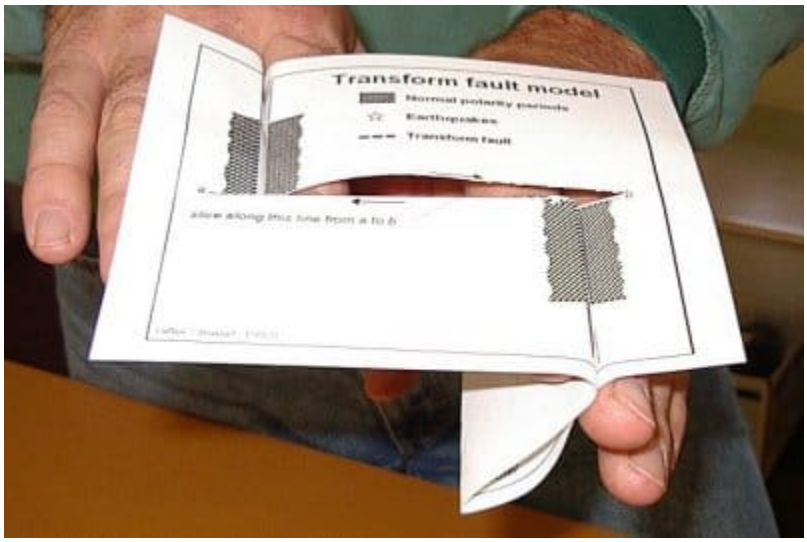


Figure 10.3.13

When you're done, you should have something like the example shown on Figure 10.3.13, with two folds of paper extending underneath. Find someone else to pinch those folds with two fingers just below each ridge crest, and then gently pull apart where shown. As you do, the oceanic crust will emerge from the middle, and you'll see that the parts of the fracture zone between the ridge crests will be moving in opposite directions (this is the transform fault) while the parts of the fracture zone outside of the ridge crests will be moving in the same direction. You'll also see that the oceanic crust is being magnetized as it forms at the ridge. The magnetic patterns shown are accurate, and represent the last 2.5 Ma of geological time.

There are other versions of this model available here: Paper Models of Transform Faults.[3]

See Appendix 3 for Exercise 10.3 answers.

Image Descriptions

Figure 10.3.2 **image description:** At 500 Ma, rocks in Europe had upward-pointing magnetic orientations. At 400 Ma, the magnetic orientation leveled. From 300 Ma to the present, rocks in Europe shown an increasingly downward-pointing magnetic orientation. [Return to Figure 10.3.2]

Figure 10.3.6 **image description:** A cross section of the trench formed at the Aleutian subduction zone as the Pacific plate subducts under the North American plate in the middle of the Pacific Ocean. The farther away an earthquake is from this trench (on the North America plate side), the deeper it is. [Return to Figure 10.3.6]

Media Attributions

- Figures 10.3.1, 10.3.2, 10.3.3, 10.3.5, 10.3.6, 10.3.8, 10.3.11, 10.3.12, 10.3.13: © Steven Earle. CC BY.
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- Figure 10.3.7: "Juan de Fuca Ridge" by USGS. Adapted by Steven Earle. Public domain. Based on Raff, A. and Mason, R., 1961, Magnetic survey off the west coast of North America, 40° N to 52° N latitude, Geol. Soc. America Bulletin, V. 72, p. 267-270.
- Figure 10.3.9: "Hawaii Hotspot" by National Geophysical Data Center. Adapted by Steven Earle. Public domain.
- Figure 10.3.10: "Hotspots" by Ingo Wölbern. Public domain.

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1. Ted Irving later set up a paleomagnetic lab at the Geological Survey of Canada in Sidney, B.C., and did a great deal of important work on understanding the geology of western North America. ↩
2. J. A. Tarduno et al., 2003, The Emperor Seamounts: Southward Motion of the Hawaiian Hotspot Plume in Earth's Mantle, Science 301 (5636): 1064–1069. ↩
3. For more information see: Earle, S., 2004, *A simple paper model of a transform fault at a spreading ridge*, J. Geosc. Educ. V. 52, p. 391-2. ↩

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10.4: Plate, Plate Motions, and Plate Boundary Processes

Continental drift and sea-floor spreading became widely accepted around 1965 as more and more geologists started thinking in these terms. By the end of 1967 the Earth's surface had been mapped into a series of plates (Figure 10.4.1). The major plates are Eurasia, Pacific, India, Australia, North America, South America, Africa, and Antarctic. There are also numerous small plates (e.g., Juan de Fuca, Cocos, Nazca, Scotia, Philippine, Caribbean), and many very small plates or sub-plates. For example the Juan de Fuca Plate is actually three separate plates (Gorda, Juan de Fuca, and Explorer) that all move in the same general direction but at slightly different rates.

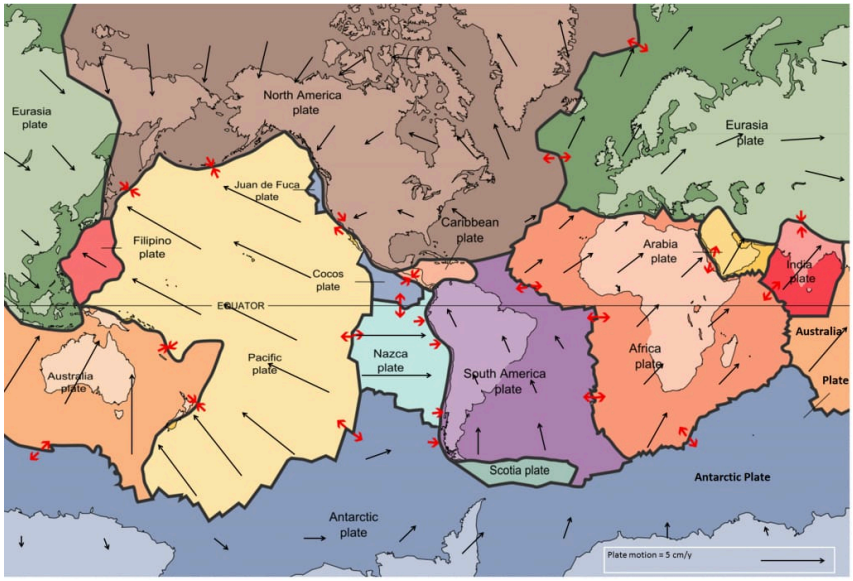


Figure 10.4.1 A map showing 15 of the Earth's tectonic plates and the approximate rates and directions of plate motions.

Rates of motions of the major plates range from less than 1 cm/y to over 10 cm/y. The Pacific Plate is the fastest, followed by the Australian and Nazca Plates. The North American Plate is one of the slowest, averaging around 1 cm/y in the south up to almost 4 cm/y in the north.

Plates move as rigid bodies, so it may seem surprising that the North American Plate can be moving at different rates in different places. The explanation is that plates move in a rotational manner. The North American Plate, for example, rotates counter-clockwise; the Eurasian Plate rotates clockwise.

Boundaries between the plates are of three types: **divergent** (i.e., moving apart), **convergent** (i.e., moving together), and **transform** (moving side by side). Before we talk about processes at plate boundaries, it's important to point out that there are never gaps between plates. The plates are made up of crust and the lithospheric part of the mantle (Figure 10.4.2), and even though they are moving all the time, and in different directions, there is never a significant amount of space between them. Plates are thought to move along the lithosphere-asthenosphere boundary, as the asthenosphere is the zone of partial melting. It is assumed that the relative lack of strength of the partial melting zone facilitates the sliding of the lithospheric plates.

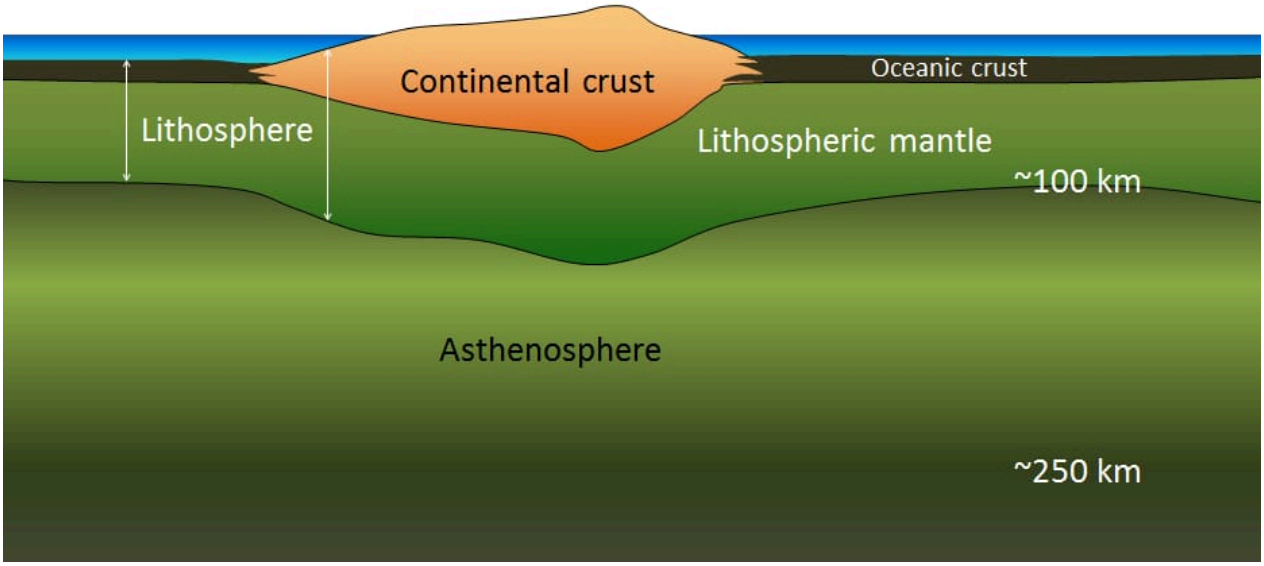


Figure 10.4.2 The crust and upper mantle. Tectonic plates consist of lithosphere, which includes the crust and the lithospheric (rigid) part of the mantle.

At spreading centres, the lithospheric mantle may be very thin because the upward convective motion of hot mantle material generates temperatures that are too high for the existence of a significant thickness of rigid lithosphere (Figure 10.4.8). The fact that the plates include both crustal material and lithospheric mantle material makes it possible for a single plate to be made up of both oceanic and continental crust. For example, the North American Plate includes most of North America, plus half of the northern Atlantic Ocean. Similarly the South American Plate extends across the western part of the southern Atlantic Ocean, while the European and African plates each include part of the eastern Atlantic Ocean. The Pacific Plate is almost entirely oceanic, but it does include the part of California west of the San Andreas Fault.

Divergent Boundaries

Divergent boundaries are spreading boundaries, where new oceanic crust is created from magma derived from partial melting of the mantle caused by decompression as hot mantle rock from depth is moved toward the surface (Figure 10.4.3). The triangular zone of partial melting near the ridge crest is approximately 60 km thick and the proportion of magma is about 10% of the rock volume, thus producing crust that is about 6 km thick. Most divergent boundaries are located at the oceanic ridges (although some are on land), and the crustal material created at a spreading boundary is always oceanic in character; in other words, it is mafic igneous rock (e.g., basalt or gabbro, rich in ferromagnesian minerals). Spreading rates vary considerably, from 2 cm/y to 6 cm/y in the Atlantic, to between 12 cm/y and 20 cm/y in the Pacific. (Note that spreading rates are typically double the velocities of the two plates moving away from a ridge.)

Some of the processes taking place in this setting include:

- Magma from the mantle pushing up to fill the voids left by divergence of the two plates
- **Pillow lavas** forming where magma is pushed out into seawater (Figure 10.4.4)
- Vertical **sheeted dykes** intruding into cracks resulting from the spreading
- Magma cooling more slowly in the lower part of the new crust and forming gabbro bodies

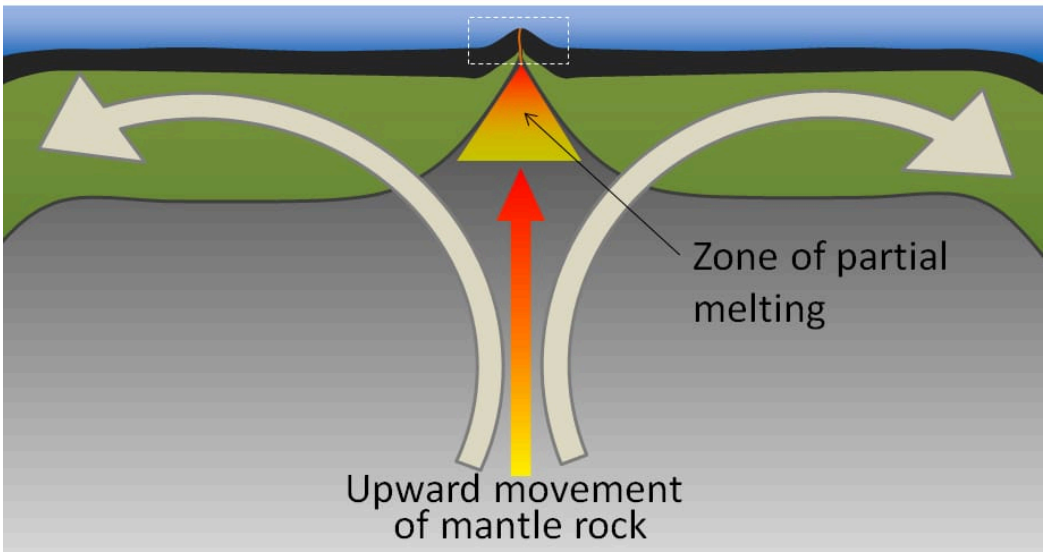


Figure 10.4.3 The general processes that take place at a divergent boundary. The area within the dashed white rectangle is shown in Figure 10.4.4.

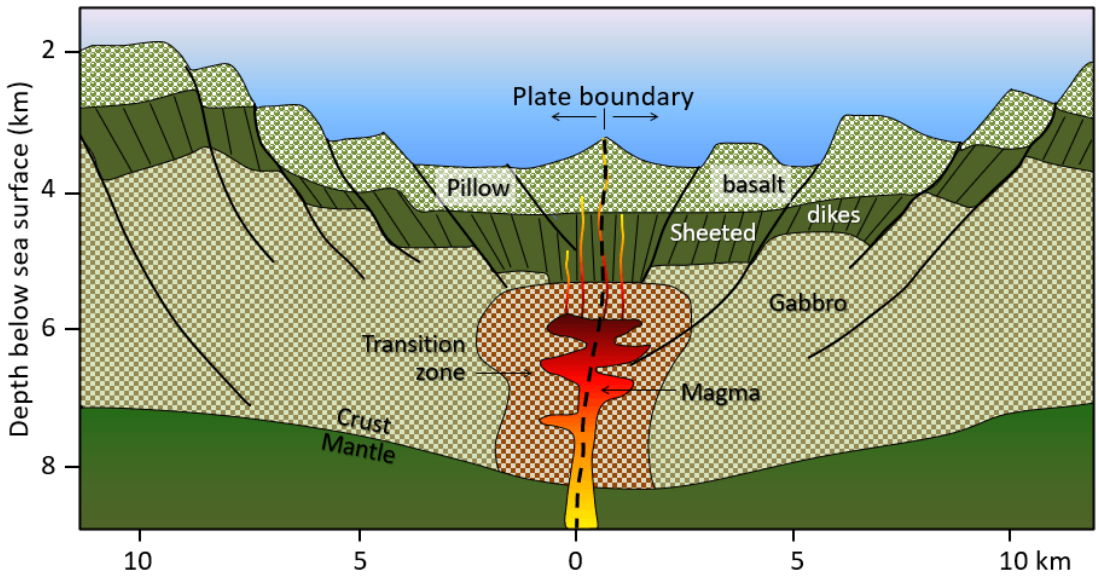


Figure 10.4.4 Depiction of the processes and materials formed at a divergent boundary.

Spreading is hypothesized to start within a continental area with up-warping or doming related to an underlying mantle plume or series of mantle plumes. The buoyancy of the mantle plume material creates a dome within the crust, causing it to fracture in a radial pattern, with three arms spaced at approximately 120° (Figure 10.4.5). When a series of mantle plumes exists beneath a large continent, the resulting rifts may align and lead to the formation of a rift valley (such as the present-day Great Rift Valley in eastern Africa). It is suggested that this type of valley eventually develops into a linear sea (such as the present-day Red Sea), and finally into an ocean (such as the Atlantic). It is likely that as many as 20 mantle plumes, many of which still exist, were responsible for the initiation of the rifting of Pangea along what is now the mid-Atlantic ridge (see Figure 10.4.10).

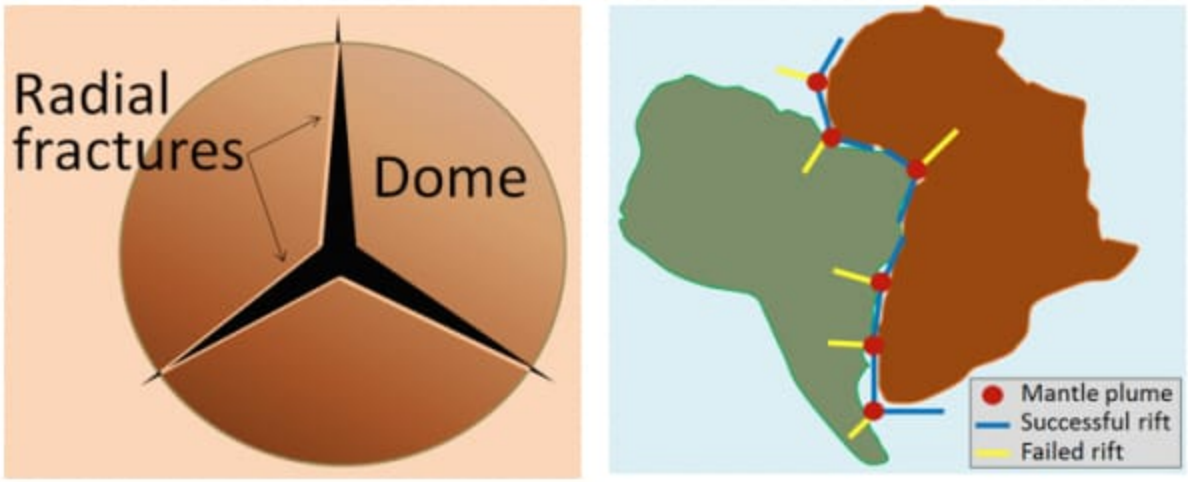


Figure 10.4.5 Depiction of the process of dome and three-part rift formation (left) and of continental rifting between the African and South American parts of Pangea at around 200 Ma (right).

Convergent Boundaries

Convergent boundaries, where two plates are moving toward each other, are of three types, depending on whether oceanic or continental crust is present on either side of the boundary. The types are ocean-ocean, ocean-continent, and continent-continent.

At an ocean-ocean convergent boundary, one of the plates (oceanic crust and lithospheric mantle) is pushed, or subducted**,** under the other. Often it is the older and colder plate that is denser and subducts beneath the younger and hotter plate. There is commonly an ocean trench along the boundary. The subducted lithosphere descends into the hot mantle at a relatively shallow angle close to the subduction zone, but at steeper angles farther down (up to about 45°). As discussed in the context of subduction-related volcanism in Chapter 4, the significant volume of water within the subducting material is released as the subducting crust is heated. Most of this water is present within the sheet silicate mineral serpentine which is derived from alteration of pyroxene and olivine near the spreading ridge shortly after the rock's formation. It is released when the oceanic crust is heats and then rises and mixes with the overlying mantle. The addition of water to the hot mantle lowers the rocks's melting point and leads to the formation of magma (flux melting) (Figure 10.4.6). The magma, which is lighter than the surrounding mantle material, rises through the mantle and the overlying oceanic crust to the ocean floor where it creates a chain of volcanic islands known as an island arc. A mature island arc develops into a chain of relatively large islands (such as Japan or Indonesia) as more and more volcanic material is extruded and sedimentary rocks accumulate around the islands.

As described above in the context of Benioff zones (Figure 10.4.6), earthquakes take place close to the boundary between the subducting crust and the overriding crust. The largest earthquakes occur near the surface where the subducting plate is still cold and strong.

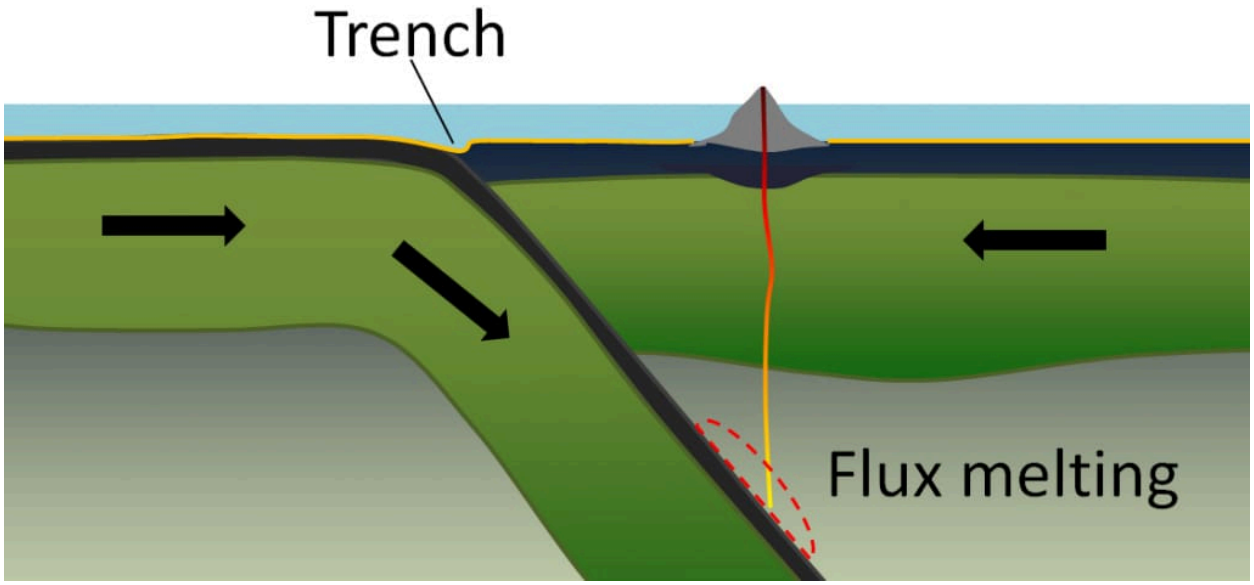


Figure 10.4.6 Configuration and processes of an ocean-ocean convergent boundary.

Examples of ocean-ocean convergent zones are subduction of the Pacific Plate beneath the North America Plate south of Alaska (Aleutian Islands) and beneath the Philippine Plate west of the Philippines, subduction of the India Plate beneath the Eurasian Plate south of Indonesia, and subduction of the Atlantic Plate beneath the Caribbean Plate (see Figure 10.4.1).

At an ocean-continent convergent boundary, the oceanic plate is pushed under the continental plate in the same manner as at an ocean-ocean boundary. Sediment that has accumulated on the **continental slope** is thrust up into an accretionary wedge, and compression leads to thrusting within the continental plate (Figure 10.4.7). The mafic magma produced adjacent to the subduction zone rises to the base of the continental crust and leads to partial melting of the crustal rock. The resulting magma ascends through the crust, producing a mountain chain with many volcanoes.

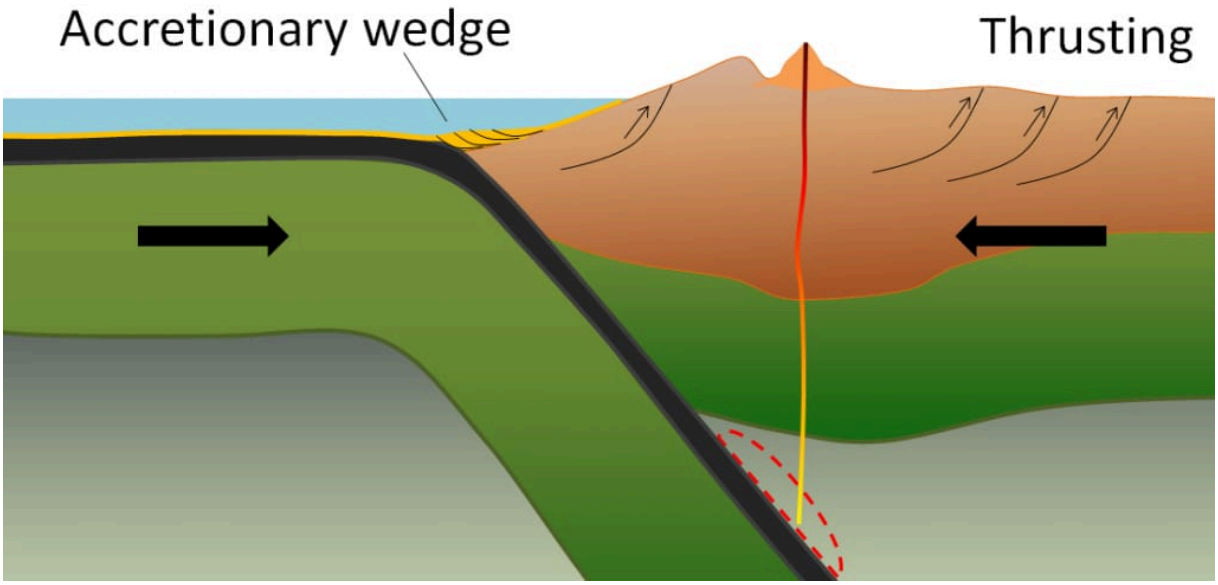


Figure 10.4.7 Configuration and processes of an ocean-continent convergent boundary.

Examples of ocean-continent convergent boundaries are subduction of the Nazca Plate under South America (which has created the Andes Range) and subduction of the Juan de Fuca Plate under North America (creating the mountains Garibaldi, Baker, St. Helens, Rainier, Hood, and Shasta, collectively known as the Cascade Range).

A continent-continent collision occurs when a continent or large island that has been moved along with subducting oceanic crust collides with another continent (Figure 10.4.8). The colliding continental material will not be subducted because it is too light (i.e., because it is composed largely of light continental rocks [SIAL]), but the root of the oceanic plate will eventually break off

and sink into the mantle. There is tremendous deformation of the pre-existing continental rocks, and creation of mountains from that rock, from any sediments that had accumulated along the shores (i.e., within geosynclines) of both continental masses, and commonly also from some ocean crust and upper mantle material.

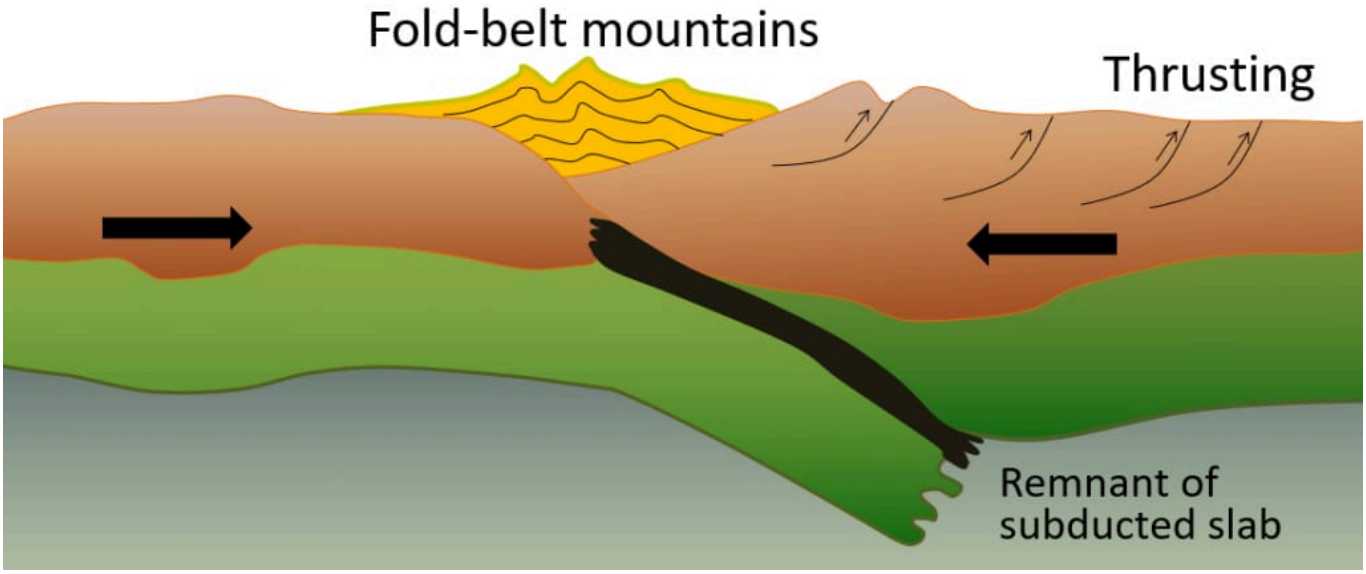


Figure 10.4.8 Configuration and processes of a continent-continent convergent boundary.

Examples of continent-continent convergent boundaries are the collision of the India Plate with the Eurasian Plate, creating the Himalaya Mountains, and the collision of the African Plate with the Eurasian Plate, creating the series of ranges extending from the Alps in Europe to the Zagros Mountains in Iran. The Rocky Mountains in B.C. and Alberta are also a result of continent-continent collisions.

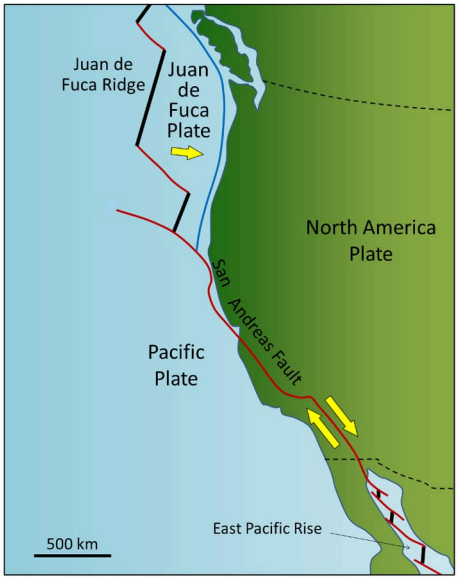


Figure 10.4.9 The San Andreas Fault extends from the north end of the East Pacific Rise in the Gulf of California to the southern end of the Juan de Fuca Ridge. All of the red lines on this map are transform faults.

Transform boundaries exist where one plate slides past another without production or destruction of crustal material. As explained above, most transform faults connect segments of mid-ocean ridges and are thus ocean-ocean plate boundaries (Figure 10.4.11). Some transform faults connect continental parts of plates. An example is the San Andreas Fault, which extends from the southern end of the Juan de Fuca Ridge to the northern end of the East Pacific Rise (ridge) in the Gulf of California (Figures 10.28 and 10.29). The part of California west of the San Andreas Fault and all of Baja California are on the Pacific Plate. Transform faults do not just connect divergent boundaries. For example, the Queen Charlotte Fault connects the north end of the Juan de Fuca Ridge, starting at the north end of Vancouver Island, to the Aleutian subduction zone.



Figure 10.4.10 The San Andreas Fault at Parkfield in central California. The person with the orange shirt is standing on the Pacific Plate and the person at the far side of the bridge is on the North American Plate. The bridge is designed to accommodate motion on the fault by sliding on its foundation.

Exercise 10.4 A different type of transform fault

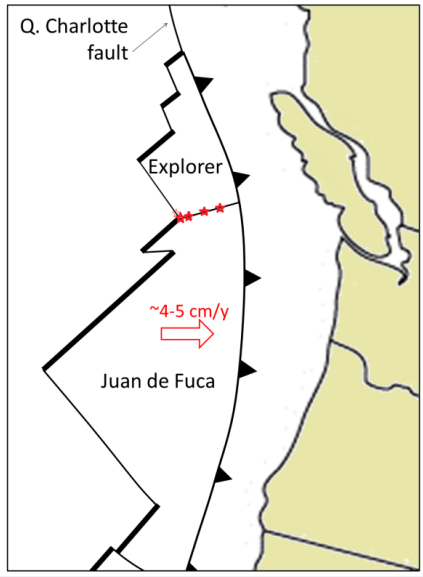


Figure 10.4.11

This map shows the Juan de Fuca (JDF) and Explorer Plates off the coast of Vancouver Island. We know that the JDF Plate is moving toward the North American Plate at around 4 centimeters per year to 5 centimeters per year. We think that the Explorer Plate is also moving east, but we don't know the rate, and there is evidence that it is slower than the JDF Plate.

The boundary between the two plates is the Nootka Fault, which is the location of frequent small-to-medium earthquakes (roughly up to magnitude 5), as depicted by the red stars. Explain why the Nootka Fault is a transform fault, and show the *relative* sense of motion along the fault with two small arrows.

See Appendix 3 for Exercise 10.4 answers.

As originally described by Wegener in 1915, the present continents were once all part of a supercontinent, which he termed **Pangea** (meaning *all land*). More recent studies of continental matchups and the magnetic ages of ocean-floor rocks have

enabled us to reconstruct the history of the break-up of Pangea.

Pangea began to rift apart along a line between Africa and Asia and between North America and South America at around 200 Ma. During the same period, the Atlantic Ocean began to open up between northern Africa and North America, and India broke away from Antarctica. Between 200 and 150 Ma, rifting started between South America and Africa and between North America and Europe, and India moved north toward Asia. By 80 Ma, Africa had separated from South America, most of Europe had separated from North America, and India had separated from Antarctica. By 50 Ma, Australia had separated from Antarctic, and shortly after that, India collided with Asia. To see the timing of these processes for yourself, go to time lapse of Continental Movements.

Within the past few million years, rifting has taken place in the Gulf of Aden and the Red Sea, and also within the Gulf of California. Incipient rifting has begun along the Great Rift Valley of eastern Africa, extending from Ethiopia and Djibouti on the Gulf of Aden (Red Sea) all the way south to Malawi.

Over the next 50 million years, it is likely that there will be full development of the east African rift and creation of new ocean floor. Eventually Africa will split apart. There will also be continued northerly movement of Australia and Indonesia. The western part of California (including Los Angeles and part of San Francisco) will split away from the rest of North America, and eventually sail right by the west coast of Vancouver Island, en route to Alaska. Because the oceanic crust formed by spreading on the mid-Atlantic ridge is not currently being subducted (except in the Caribbean), the Atlantic Ocean is slowly getting bigger, and the Pacific Ocean is getting smaller. If this continues without changing for another couple hundred million years, we will be back to where we started, with one supercontinent.

Pangea, which existed from about 350 to 200 Ma, was not the first supercontinent. It was preceded by Pannotia (600 to 540 Ma), by Rodinia (1,100 to 750 Ma), and by others before that.

In 1966, Tuzo Wilson proposed that there has been a continuous series of cycles of continental rifting and collision; that is, break-up of supercontinents, drifting, collision, and formation of other supercontinents. At present, North and South America, Europe, and Africa are moving with their respective portions of the Atlantic Ocean. The eastern margins of North and South America and the western margins of Europe and Africa are called **passive margins** because there is no subduction taking place along them.

This situation may not continue for too much longer, however. As the Atlantic Ocean floor gets weighed down around its margins by great thickness of continental sediments (i.e., geosynclines), it will be pushed farther and farther into the mantle, and eventually the oceanic lithosphere may break away from the continental lithosphere (Figure 10.4.12). A subduction zone will develop, and the oceanic plate will begin to descend under the continent. Once this happens, the continents will no longer continue to move apart because the spreading at the mid-Atlantic ridge will be taken up by subduction. If spreading along the mid-Atlantic ridge continues to be slower than spreading within the Pacific Ocean, the Atlantic Ocean will start to close up, and eventually (in a 100 million years or more) North and South America will collide with Europe and Africa.

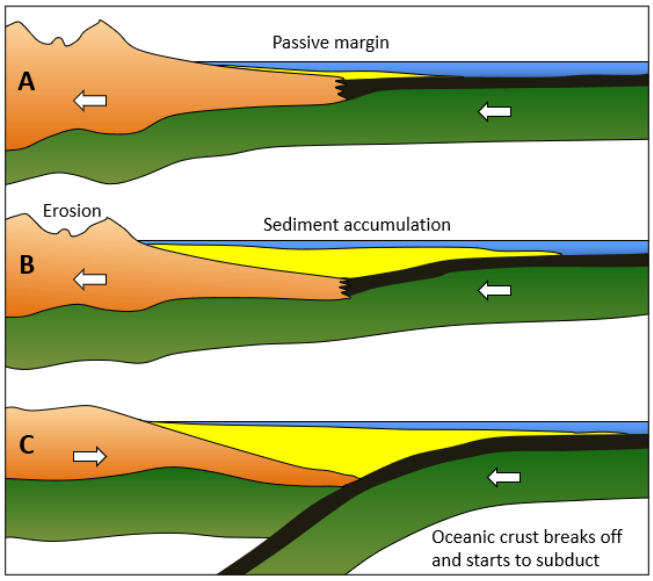


Figure 10.4.12 Development of a subduction zone at a passive margin. Times A, B, and C are separated by tens of millions of years. Once the oceanic crust breaks off and starts to subduct the continental crust (North America in this case) will no longer be pushed to the west and will likely start to move east because the rate of spreading in the Pacific basin is faster than that in the Atlantic basin.

There is strong evidence around the margins of the Atlantic Ocean that this process has taken place before. The roots of ancient mountain belts, which are present along the eastern margin of North America, the western margin of Europe, and the northwestern margin of Africa, show that these land masses once collided with each other to form a mountain chain, possibly as big as the Himalayas. The apparent line of collision runs between Norway and Sweden, between Scotland and England, through Ireland, through Newfoundland, and the Maritimes, through the northeastern and eastern states, and across the northern end of Florida. When rifting of Pangea started at approximately 200 Ma, the fissuring was along a different line from the line of the earlier collision. This is why some of the mountain chains formed during the earlier collision can be traced from Europe to North America and from Europe to Africa.

That the Atlantic Ocean rift may have occurred in approximately the same place during two separate events several hundred million years apart is probably no coincidence. The series of hot spots that has been identified in the Atlantic Ocean may also have existed for several hundred million years, and thus may have contributed to rifting in roughly the same place on at least two separate occasions (Figure 10.4.13).

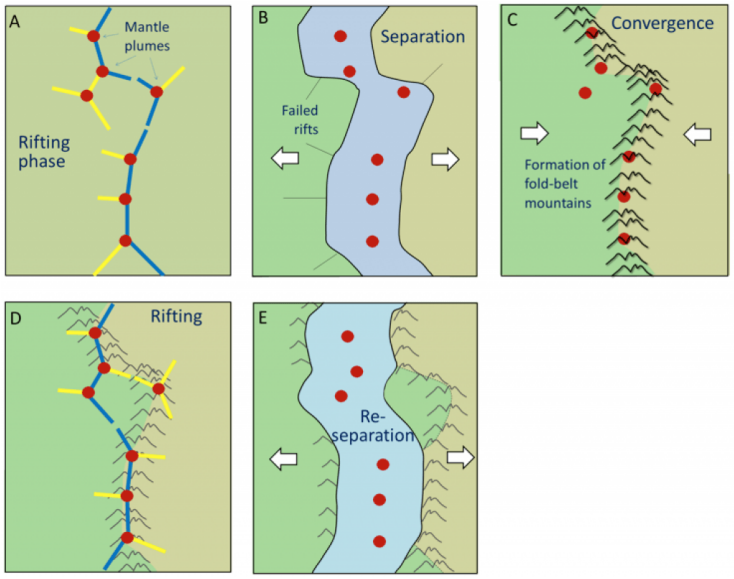


Figure 10.4.13 A scenario for the Wilson cycle. (A) The cycle starts with continental rifting above a series of mantle plumes. (B) The continents separate, and then (C) re-converge some time later, forming a fold-belt mountain chain. (D) Eventually rifting is repeated, possibly because of the same set of mantle plumes, but this time the rift is in a different place.

Exercise 10.5 Getting to know the plates and their boundaries

This map shows the boundaries between the major plates. Without referring to the plate map in Figure 10.4.1, or any other resources, write in the names of as many of the plates as you can. Start with the major plates, and then work on the smaller ones. Don't worry if you can't name them all.

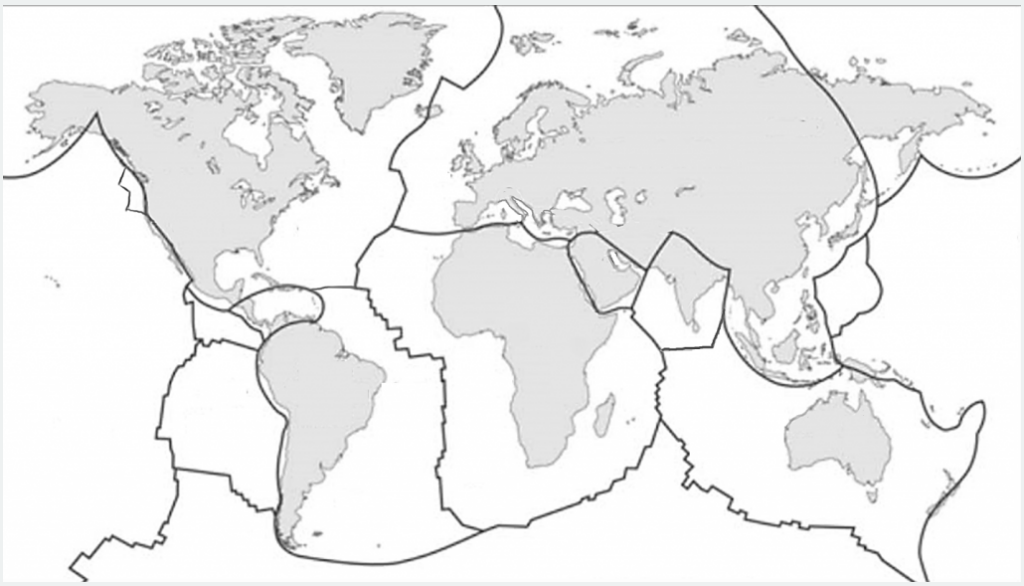


Figure 10.4.14

Once you've named most of the plates, draw arrows to show the general plate motions. Finally, using a highlighter or colored pencil, label as many of the boundaries as you can as divergent, convergent, or transform.

See Appendix 3 for Exercise 10.5 answers.

Image descriptions

Figure 10.4.1 image description: Descriptions of 15 different plates and their movements.

Plate name	Description of plate	Bordering plates (ordered from longest border to shortest)	Description of movement
Africa plate	This plate includes all of Africa and the surrounding ocean, including the eastern Atlantic Ocean, the surrounding Antarctic Ocean, and the western Indian ocean.	South America plate, Antarctic plate, Eurasia plate, North America plate, Arabia plate, India plate, Australia plate	This plate is moving north east towards the Arabia and Eurasia plates.
Antarctic plate.	This plate makes up all of Antarctica and much of the surrounding ocean.	Pacific plate, Australia plate, Africa plate, Scotia plate, Nazca plate, South America plate.	The part of the plate around the South America plate is moving northwards and a little east. The part of the plate around the Australia plate is moving southwards.
Arabia plate	This plate includes all of Saudi Arabia, and much of the Levant (up to Iraq and Syria).	Eurasia plate, Africa plate, India plate	This plate is moving north east towards the Eurasia plate.

Plate name	Description of plate	Bordering plates (ordered from longest border to shortest)	Description of movement
Australia plate	This plate includes Australia and much of the surrounding ocean. New Guinea and the northern parts of New Zealand are part of the Australia plate. The ocean area along southern Asia up to the India plate is also a part of the Australia plate.	Antarctic plate, Pacfic plate, Eurasia plate, India Plate, Africa plate.	This plate is moving north east towards the Eurasia plate and the Pacific plate.
Caribbean plate	This plate is small. It includes the central Caribbean countries and runs along the northern edge of South America.	North America plate, South America plate, Cocos plate.	N/A
Cocos plate	This plate is small. It runs along the west coast of Mexico and western Caribbean countries.	Nazca plate, Pacific plate, North America plate, Caribbean plate.	This plate is moving north east towards the Caribbean and North America plates.
Eurasia plate	This plate includes the northeastern part of the Atlantic ocean, all of Europe, all of Russia (except its most eastern part), and down through southeast Asia, including China and Indonesia.	North America plate, Africa plate, Australia plate, Arabia plate, India plate, Filipino plate.	This plate is rotating in a clockwise direction towards the Pacific plate.
Filipino plate	This plate includes the islands that make up the Philipines and north to include parts of southern Japan.	Eurasia plate, Pacific plate.	This plate is moving north west towards the Eurasia plate.
India plate	This plate includes India and the surrounding India Ocean.	Australia plate, Eurasia plate, Africa plate, Arabia plate.	This plate is moving north north east towards the Eurasia plate.
Juan de Fuca plate	This plate is small. It runs along the north western coast of the United States and the southern British Columbia coast.	Pacific plate, North America plate.	N/A
Nazca plate	This plate is in the Pacific Ocean between the Pacific plate and the South America plate.	South America plate, Pacific plate, Antarctic plate, Cocos plate	This plate is moving directly east towards the South America plate.
North America Plate	This plate includes all of North America, Greenland, the eastern most part of Russia, northern Japan, and the northwestern part of the Atlantic ocean.	Eurasia plate, Pacific plate, Africa plate, Caribbean plate, South America plate, Cocos plate, Juan de Fuca plate	This plate is rotating counter clockwise in towards the Pacific plate.
Pacific plate	This plate makes up most of the Pacific Ocean.	North America plate, Australia plate, Antarctic plate, Nazca plate, Filipino plate, Cocos plate, Juan de Fuca plate	This plate is moving northwest towards the Australia, Filipino, and Eurasia plates.
Scotia plate	This plate is small. It runs from the tip of South America eastwards to form a barrier between the Antarctic plate and the South America plate.	Antarctic plate, South America plate.	N/A
South America plate	This plate starts at the western edge of South America and stretches east into the southwestern parst of the Atlantic Ocean.	Africa plate, Nazca plate, Scotia plate, Caribbean plate, Antarctic plate, North America plate	This plate moves north and slightly west towards the Caribbean plate and the North America plate.

[Return to Figure 10.4.1]

Media Attributions

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10.5: Mechanisms for Plate Motion

It has been often repeated in this text and elsewhere that convection of the mantle is critical to plate tectonics, and while this is almost certainly so, other forces likely play a significant role. One side in the argument holds that the plates are only moved by the traction caused by mantle convection. The other side holds that traction plays only a minor role and that two other forces, ridge-push and slab-pull, are more important (Figure 10.5.1). Some argue that the real answer lies somewhere in between.

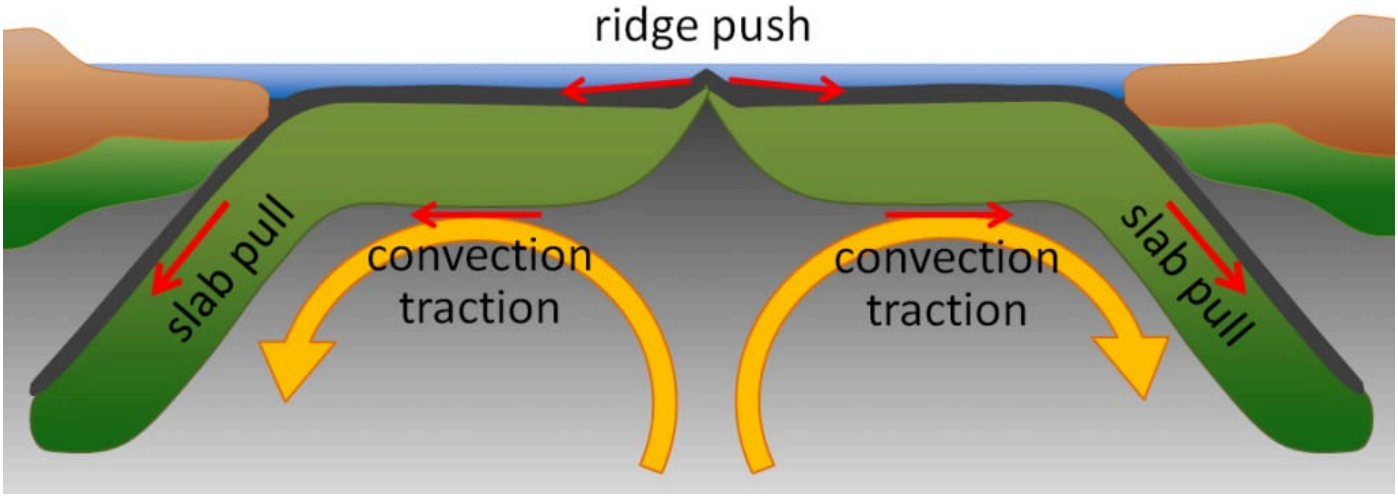


Figure 10.5.1 Models for plate motion mechanisms.

Kearey and Vine (1996)[1] have listed some compelling arguments in favour of the **ridge-push/slab-pull** model, as follows: (a) plates that are attached to subducting slabs (e.g., Pacific, Australian, and Nazca Plates) move the fastest, and plates that are not (e.g., North American, South American, Eurasian, and African Plates) move significantly slower; (b) in order for the traction model to apply, the mantle would have to be moving about five times faster than the plates are moving (because the coupling between the partially liquid asthenosphere and the plates is not strong), and such high rates of convection are not supported by geophysical models; and (c) although large plates have potential for much higher convection traction, plate velocity is not related to plate area.

In the ridge-push/slab-pull model, which is the one that has been adopted by most geologists working on plate-tectonic problems, the lithosphere is the upper surface of the convection cells, as is illustrated in Figure 10.5.2.

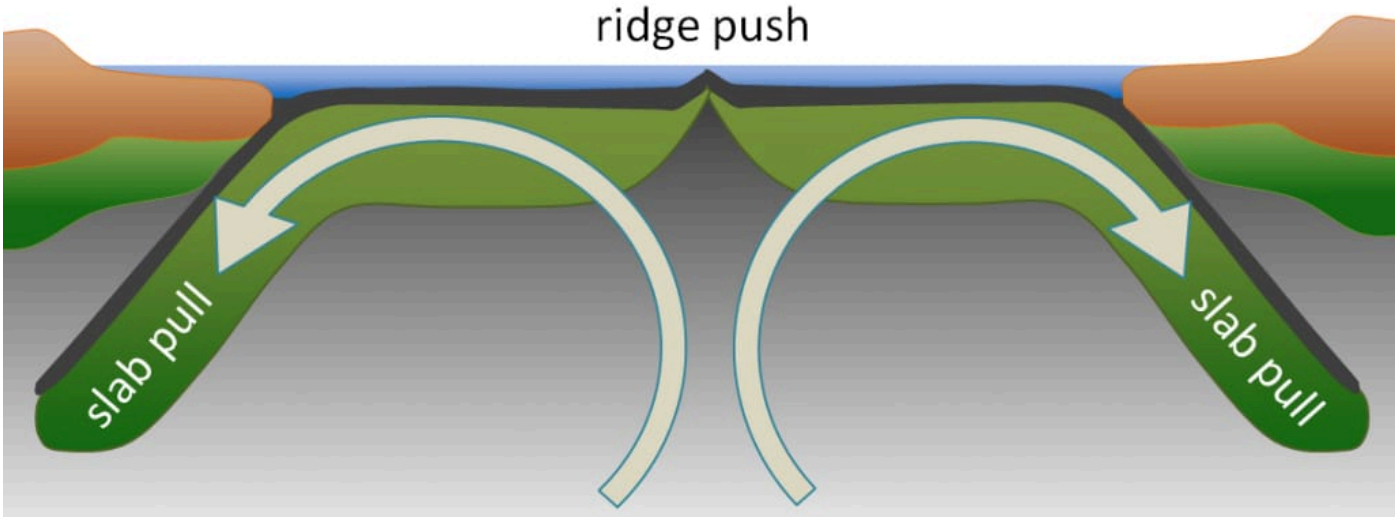


Figure 10.5.2 The ridge-push/slab-pull model for plate motion, in which the lithosphere is the upper surface of the convective systems.

Although ridge-push/slab-pull is the widely favoured mechanism for plate motion, it's important not to underestimate the role of mantle convection. Without convection, there would be no ridges to push from because upward convection brings hot buoyant rock to surface. Furthermore, many plates, including our own North American Plate, move along nicely—albeit slowly—without any slab-pull happening.

Image Descriptions

Figure 10.5.1 **image description:** In this model, there are three forces working to move the plates. Ridge-push forces cause two plates to pull apart on the surface. Slab-pull forces pull the plates down. This movement of out and down is also encouraged by convection traction, or clockwise and counterclockwise currents that are present beneath the plates. [Return to Figure 10.5.1]

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1. Kearey and Vine , 1996, Global Tectonics (2ed), Blackwell Science Ltd., Oxford ↩

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10.6: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
10.1 Alfred Wegener: The Father of Plate Tectonics	The evidence for continental drift in the early 20th century included the matching of continental shapes on either side of the Atlantic and the geological and fossil matchups between continents that are now thousands of kilometers apart.
10.2 Global Geological Models of the Early 20th Century	The established theories of global geology were permanentism and contractionism, but neither of these theories was able to explain some of the evidence that supported the idea of continental drift.
10.3 Geological Renaissance of the Mid-20th Century	Giant strides were made in understanding Earth during the middle decades of the 20th century, including discovering magnetic evidence of continental drift, mapping the topography of the ocean floor, describing the depth relationships of earthquakes along ocean trenches, measuring heat flow differences in various parts of the ocean floor, and mapping magnetic reversals on the sea floor. By the mid-1960s, the fundamentals of the theory of plate tectonics were in place.
10.4 Plate, Plate Motions, and Plate Boundary Processes	Earth's lithosphere is made up of over 20 plates that are moving in different directions at rates of between 1 cm/y and 10 cm/y. The three types of plate boundaries are divergent (plates moving apart and new crust forming), convergent (plates moving together and one being subducted), and transform (plates moving side by side). Divergent boundaries form where existing plates are rifted apart, and it is hypothesized that this is caused by a series of mantle plumes. Subduction zones are assumed to form where accumulation of sediment at a passive margin leads to separation of oceanic and continental lithosphere. Supercontinents form and break up through these processes.
10.5 Mechanisms for Plate Motion	It is widely believed that ridge-push and slab-pull are the main mechanisms for plate motion, as opposed to traction by mantle convection. Mantle convection is a key factor for producing the conditions necessary for ridge-push and slab-pull.

Questions for Review

1. List some of the evidence used by Wegener to support his idea of moving continents.
2. What was the primary technical weakness with Wegener's continental drift theory?
3. How were mountains thought to be formed (a) by contractionists and (b) by permanentists?
4. How were the trans-Atlantic paleontological matchups explained in the late 19th century?
5. In the context of isostasy, what would prevent an area of continental crust from becoming part of an ocean?
6. How did we learn about the topography of the sea floor in the early part of the 20th century?
7. How does the temperature profile of the crust and the mantle indicate that part of the mantle must be convecting?
8. What evidence from paleomagnetic studies provided support for continental drift?
9. Which parts of the oceans are the deepest?
10. Why is there less sediment in the ocean ridge areas than in other parts of the sea floor?
11. How were the oceanic heat-flow data related to mantle convection?
12. Describe the spatial and depth distribution of earthquakes at ocean ridges and ocean trenches.
13. In the model for ocean basins developed by Harold Hess, what took place at oceanic ridges and what took place at oceanic trenches?



Figure A

- What aspect of plate tectonics was not included in the Hess theory?
- Figure 10.36 shows the pattern of sea-floor magnetic anomalies in the area of a spreading ridge. Draw in the likely location of the ridge.
- What is a mantle plume and what is its expected lifespan?
- Describe the nature of movement at an ocean ridge transform fault (a) between the ridge segments, and (b) outside the ridge segments.
- How is it possible for a plate to include both oceanic and continental crust?
- What is the likely relationship between mantle plumes and the development of a continental rift?
- Why does subduction not take place at a continent-continent convergent zone?
- Divergent, convergent, and transform boundaries are shown in different colors on Fiugre 10.37. Which colors are the divergent boundaries, which are the convergent boundaries, and which are the transform boundaries?

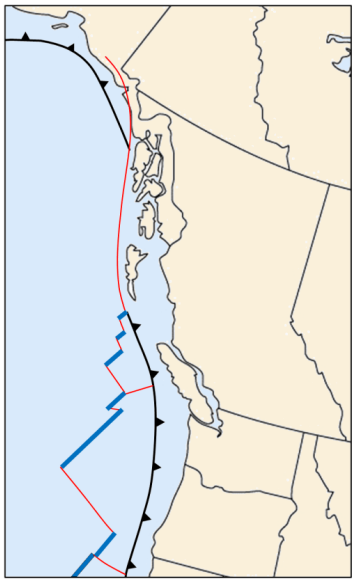


Figure B

- Name the plates on this map and show their approximate motion directions.
- Show the sense of motion on either side of the plate boundary to the west of Haida Gwaii (Queen Charlotte Islands).
- Where are Earth's most recent sites of continental rifting and creation of new ocean floor?
- What is likely to happen to western California over the next 50 million years?
- Answers to Review Questions can be found in Appendix 2.

Image Descriptions

Figure B image description: A black line with triangles pointing towards the coast stretches from the Oregon and Washington state up just past Vancouver Island to the southern tip of Haida Gwaii. This line also appears along the Alaskan coast and stretches part way down the Alaskan Pan-Handle. A thin red line stretches from the Alaskan Pan-Handle down just past the southern tip of Haida Gwaii. From that point, it alternates from being a thin red to a thick blue line to form uneven angles zig zagging south past Oregon state. [Return to Figure B]

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11: EARTHQUAKES

11.1: What is an Earthquake?
11.2: Earthquakes and Plate Tectonics
11.3: Measuring Earthquakes
11.4: The Impacts of Earthquakes
11.5: Forecasting Earthquakes and Minimizing Damage and Casualties
11.6: Summary

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11.1: What is an Earthquake?

An earthquake is the shaking caused by the **rupture** (breaking) and subsequent displacement of rocks (one body of rock moving with respect to another) beneath Earth's surface.

A body of rock that is under stress becomes deformed. When the rock can no longer withstand the deformation, it breaks and the two sides slide past each other. Most earthquakes take place near plate boundaries—but not necessarily right on a boundary—and not necessarily even on a pre-existing fault.

The engineering principle of **elastic deformation**, which can be used to understand earthquakes, is illustrated in Figure 11.1.1. The stress applied to a rock—typically because of ongoing plate movement—results in strain or deformation of the rock (Figure 11.1.1b). Because most rock is strong (unlike loose sand, for example), it can withstand a significant amount of deformation without breaking. But every rock has a deformation limit and will rupture (break) once that limit is reached. At that point, in the case of rocks within the crust, the rock breaks and there is displacement along the **rupture surface** (Figure 11.1.1c). The magnitude of the earthquake depends on the extent of the area that breaks (the area of the rupture surface) and the average amount of displacement (sliding).

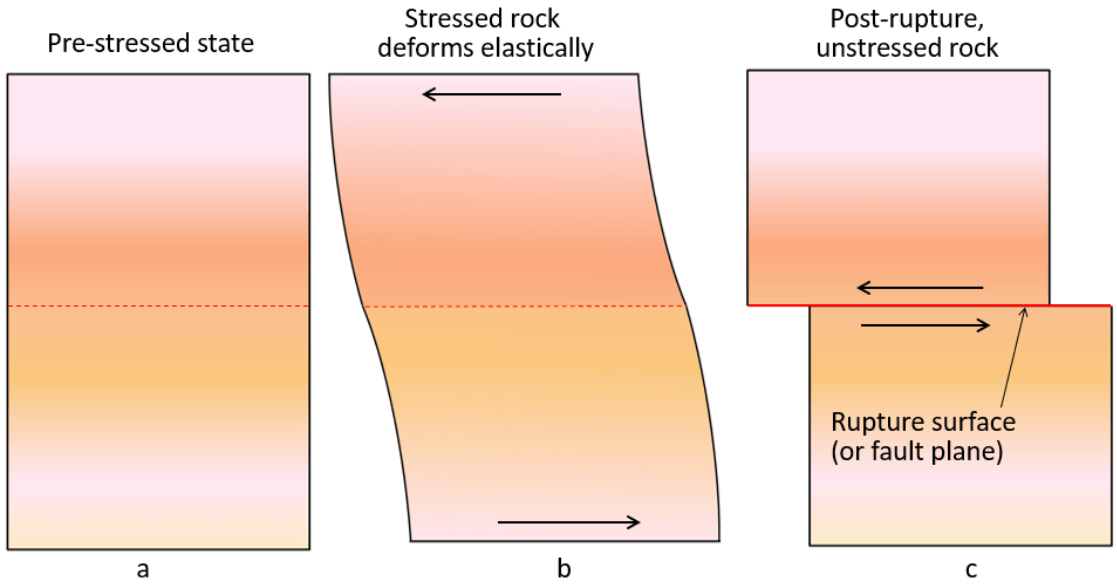


Figure 11.1.1 Depiction of the concept of elastic deformation and rupture. The plate boundary is shown as a dashed red line. In b the two plates are moving as shown by the arrows, but they are locked against each along the plate boundary so they are both deforming and the rocks are stressed. In c there has been a rupture along the boundary and the stress is released.

The concept of a rupture surface, which is critical to understanding earthquakes, is illustrated in Figure 11.1.2. An earthquake does not happen at a point, it happens over an area within a plane, although not necessarily a *flat* plane. Within the area of the rupture surface, the amount of displacement is variable (Figure 11.1.2), and, by definition, it decreases to zero at the edges of the rupture surface because the rock beyond that point isn't displaced at all. The extent of a rupture surface and the amount of displacement will depend on a number of factors, including the type and strength of the rock, and the degree to which it was stressed beforehand.

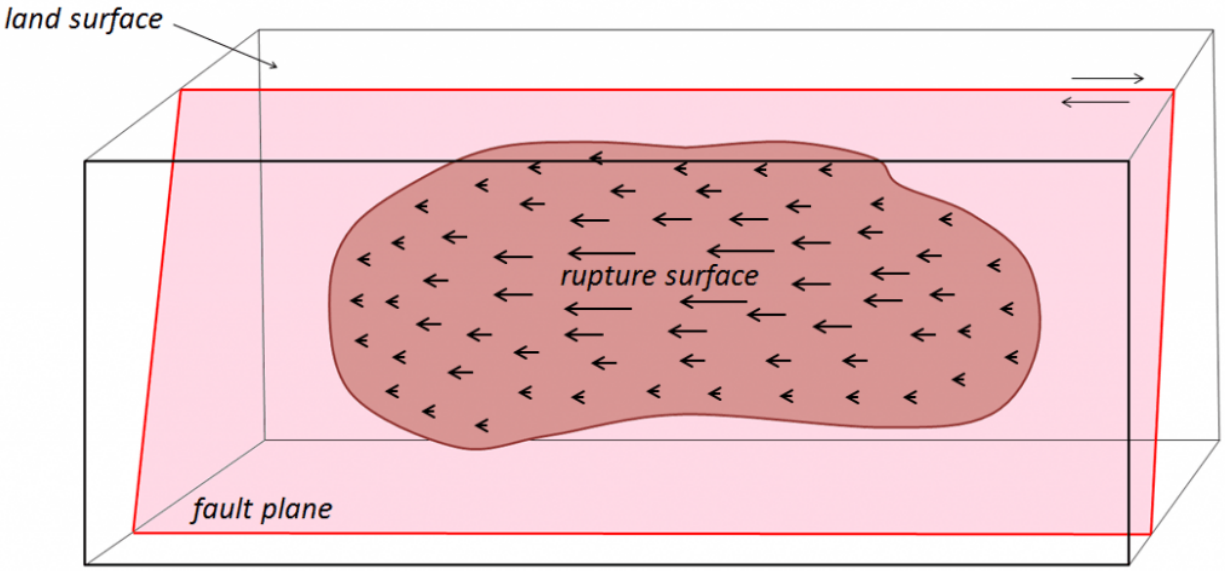


Figure 11.1.2 A rupture surface (dark pink), on a steeply dipping fault plane (light pink). The diagram represents a part of the crust that may be up to tens or hundreds of kilometers long. The rupture surface is the part of the fault plane along which displacement occurred. In this example, the near side of the fault is moving to the left, and the lengths of the arrows within the rupture surface represent relative amounts of displacement.

Earthquake rupture doesn't happen all at once; it starts at a single point and spreads rapidly from there. Depending on the extent of the rupture surface, the propagation of failures out from the point of initiation is typically completed within seconds to several tens of seconds (Figure 11.1.3). The initiation point isn't necessarily in the centre of the rupture surface; it may be close to one end, near the top, or near the bottom.

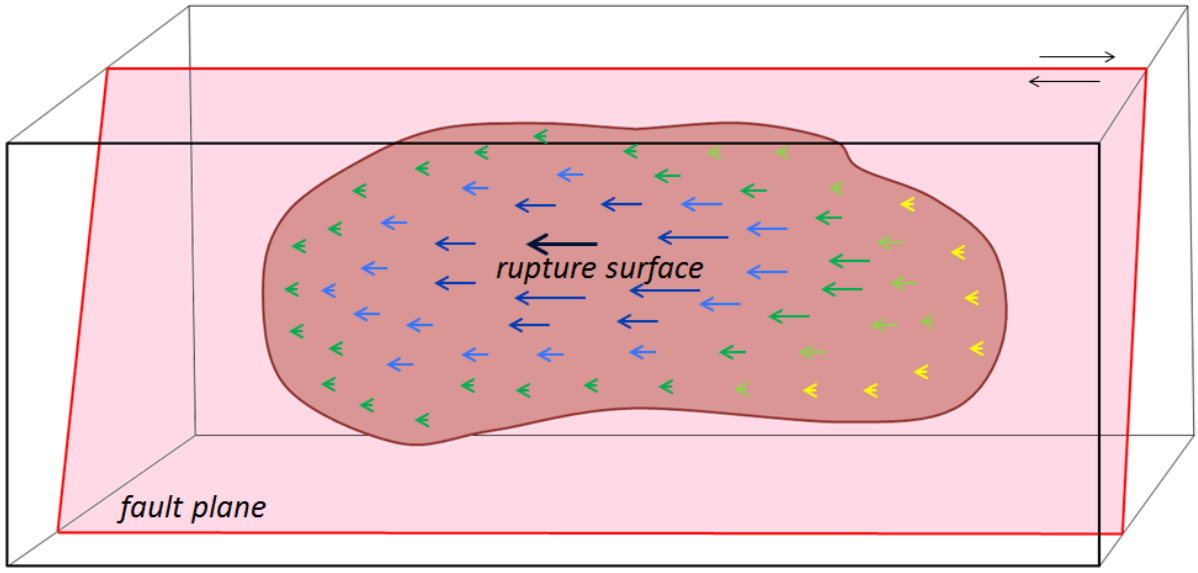


Figure 11.1.3 Propagation of failure on a rupture surface. In this case, the failure starts at the dark blue heavy arrow in the centre and propagates outward, reaching the left side first (green arrows) and the right side last (yellow arrows).

Figure 11.1.4 shows the distribution of immediate **aftershocks** associated with the 1989 Loma Prieta earthquake. Panel (b) is a section along the San Andreas Fault; this view is equivalent to what is shown in Figures 11.1.2 and 11.1.3. The area of red dots is the rupture surface; each red dot is a specific aftershock that was recorded on a seismometer. The hexagon labelled "main earthquake" represents the first or main shock. When that happened, the rock at that location broke and was displaced. That released the stress on that particular part of the fault, but it resulted in an increase of the stress on other nearby parts of the fault, and contributed to a cascade of smaller ruptures (aftershocks), in this case, over an area about 60 kilometers long and 15 kilometers wide.

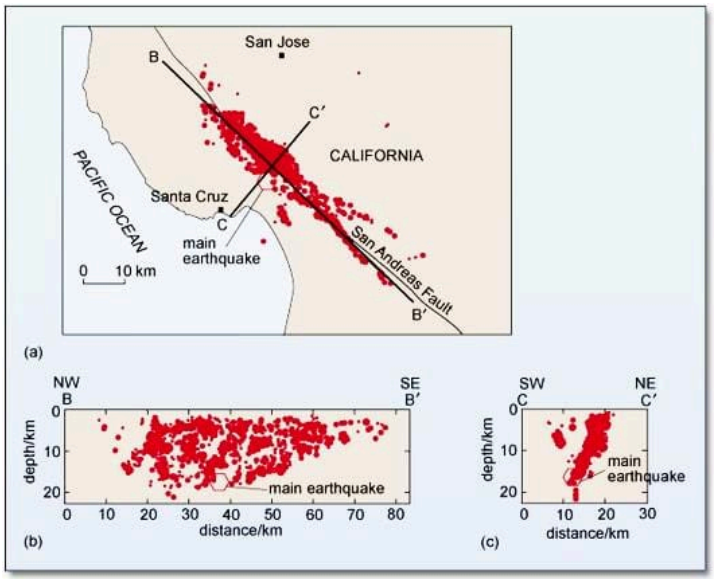


Figure 11.1.4 Distribution of the aftershocks of the 1989 M 6.9 Loma Prieta earthquake (a: plan view, b: section along the fault, c: section across the fault.)

So, what exactly is an aftershock then? An aftershock is an earthquake just like any other, but it is one that can be shown to have been triggered by **stress transfer** from a preceding earthquake. Within a few tens of seconds of the main Loma Prieta earthquake, there were hundreds of smaller aftershocks; their distribution defines the area of the rupture surface.

Aftershocks can be of any magnitude. Most are smaller than the earthquake that triggered them, but they can be bigger. The aftershocks shown in Figure 11.1.4 all happened within seconds or minutes of the main shock, but aftershocks can be delayed for hours, days, weeks, or even years. As already noted, aftershocks are related to stress transfer. For example, the main shock of the Loma Prieta earthquake triggered aftershocks in the immediate area, which triggered more in the surrounding area, eventually extending for 30 kilometers along the fault in each direction and for 15 kilometers toward the surface. But the earthquake as a whole also changed the stress on adjacent parts of the San Andreas Fault. This effect, which has been modelled for numerous earthquakes and active faults around the world, is depicted in Figure 11.1.5. Stress was reduced in the area of the rupture (blue), but was increased at either end of the rupture surface (red and yellow).

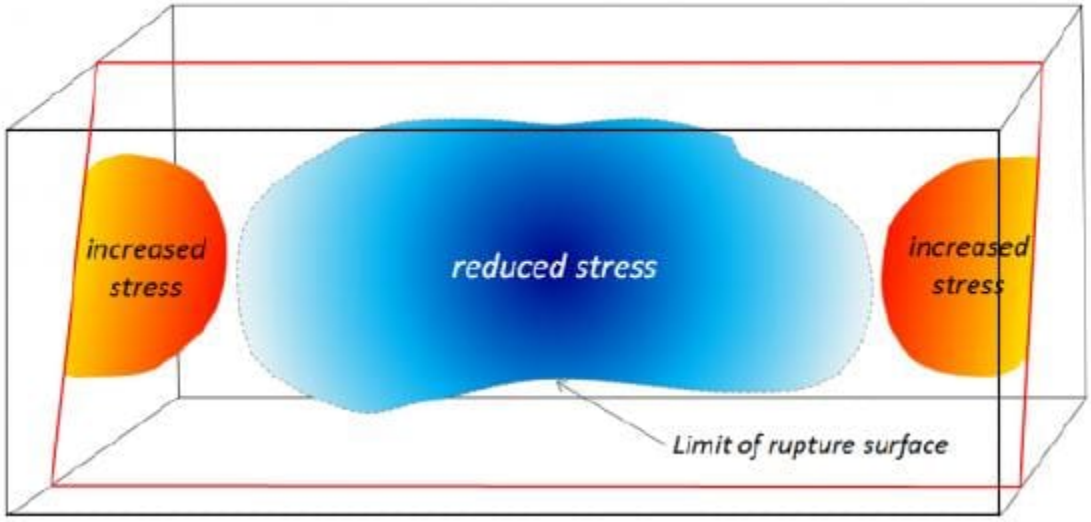


Figure 11.1.5 Depiction of stress changes related to an earthquake. Stress decreases in the area of the rupture surface, but increases on adjacent parts of the fault.

Stress transfer isn't necessarily restricted to the fault along which an earthquake happened. It will affect the rocks in general around the site of the earthquake and may lead to increased stress on other faults in the region. The effects of stress transfer don't necessarily show up right away. Segments of faults are typically in some state of stress, and the transfer of stress from another

area is only rarely enough to push a fault segment beyond its limits to the point of rupture. The stress that is added by stress transfer accumulates along with the ongoing buildup of stress from plate motion and eventually leads to another earthquake.

Episodic Tremor and Slip

Episodic tremor and slip (ETS) is periodic slow sliding along part of a subduction boundary. It does not produce recognizable earthquakes, but does produce seismic tremor (rapid seismic vibrations on a seismometer). It was first discovered on the Vancouver Island part of the Cascadia subduction zone by Geological Survey of Canada geologists Herb Dragert and Garry Rogers.[1]

The boundary between the subducting Juan de Fuca Plate and the North America Plate can be divided into three segments (Figure 11.1.6). The cold upper part of the Juan de Fuca Plate boundary is locked. The plates are stuck and don't move, except with very large earthquakes that happen *approximately* every 500 years (the last one was approximately M9 on January 26, 1700). The warm lower part of the boundary is sliding continuously because the warm rock is weaker. The central part of the boundary isn't cold enough to be stuck, but isn't warm enough to slide continuously. Instead it slips episodically, approximately every 14 months for about 2 weeks, moving a few centimeters each time.

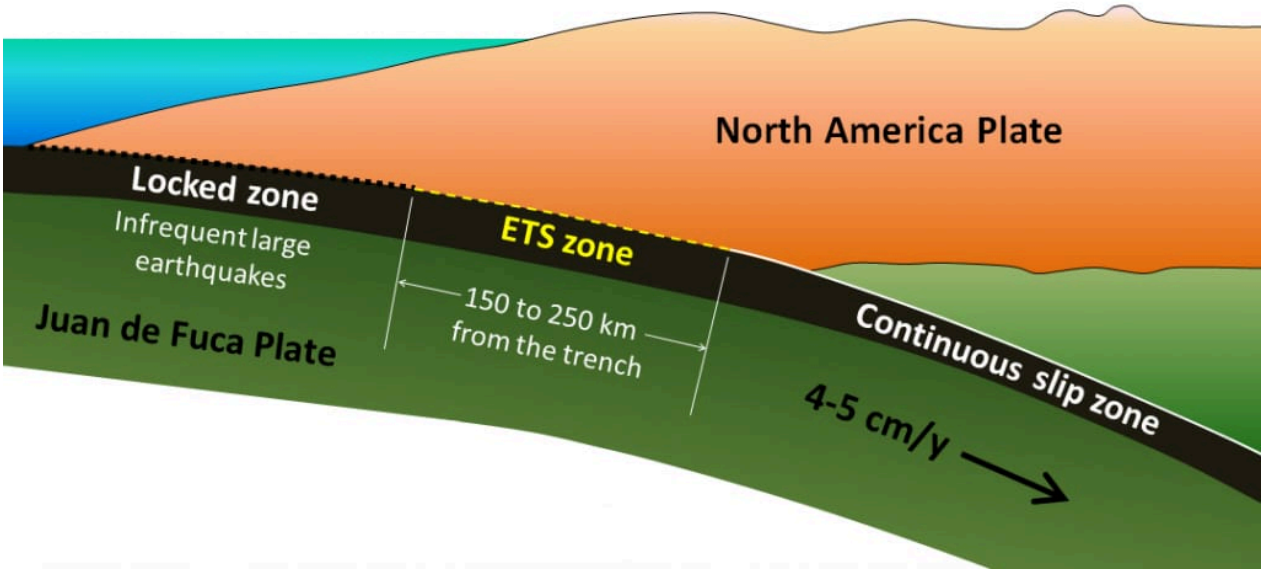


Figure 11.1.6 The boundary between the subducting Juan de Fuca Plate and the North America Plate is locked in the upper part, slides continuously in the lower part and slides episodically in the middle part.

You might be inclined to think that it's a good thing that there is periodic slip on this part of the plate because it releases some of the tension and reduces the risk of a large earthquake. In fact, the opposite is likely the case. The movement along the ETS part of the plate boundary acts like a medium-sized earthquake and leads to stress transfer to the adjacent locked part of the plate. Approximately every 14 months, during the two-week ETS period, there is a transfer of stress to the shallow locked part of the Cascadia subduction zone, and therefore an increased chance of a large earthquake.

Since 2003, ETS processes have also been observed on subduction zones in Mexico, New Zealand and Japan.

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1. Rogers, G. and Dragert, H., 2003, Episodic tremor and slip on the Cascadia subduction zone: the chatter of silent slip, Science, V. 300, p. 1942-1943. ↩

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11.2: Earthquakes and Plate Tectonics

The distribution of earthquakes across the globe is shown in Figure 11.2.1. It is relatively easy to see the relationships between earthquakes and the plate boundaries. Along divergent boundaries like the mid-Atlantic ridge and the East Pacific Rise, earthquakes are common, but restricted to a narrow zone close to the ridge, and consistently at less than a 30 kilometre depth. Shallow earthquakes are also common along transform faults, such as the San Andreas Fault. Along subduction zones, as we saw in Chapter 10, earthquakes are very abundant, and they are increasingly deep on the landward side of the subduction zone.

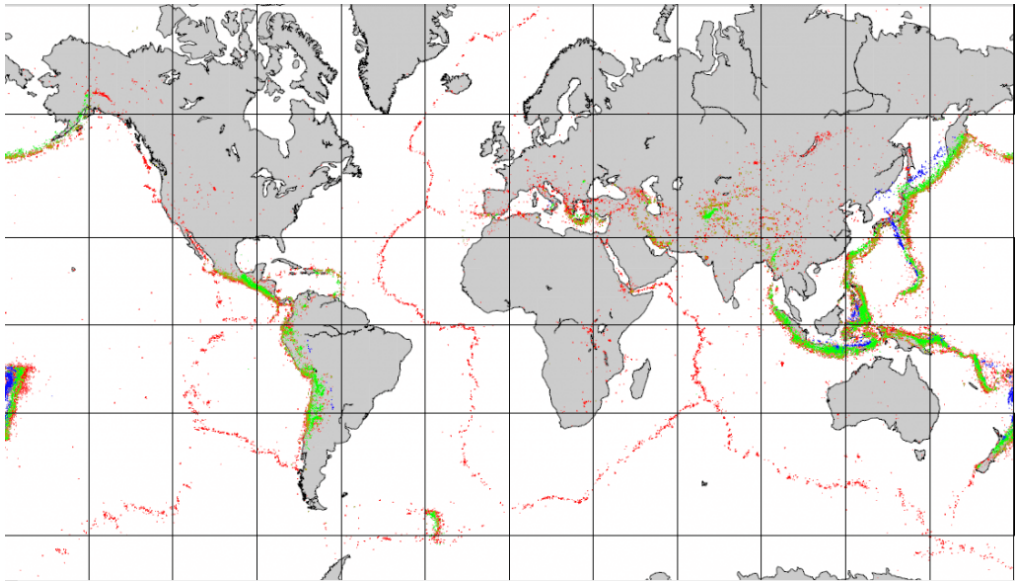


Figure 11.2.1 General distribution of global earthquakes of magnitude 4 and greater from 2004 to 2011, color coded by depth (red: 0 to 33 kilometers, orange 33 to 70 kilometers, green: 70 to 300 kilometers, blue: 300 to 700 kilometers).

Earthquakes are also relatively common at a few intraplate locations. Some are related to the buildup of stress due to continental rifting or the transfer of stress from other regions, and some are not well understood. Examples of intraplate earthquake regions include the Great Rift Valley area of Africa, the Tibet region of China, and the Lake Baikal area of Russia.

Earthquakes at Divergent and Transform Boundaries

Figure 11.2.2 provides a closer look at magnitude (M) 4 and larger earthquakes in an area of divergent boundaries in the mid-Atlantic region near the equator. Here, as we saw in Chapter 10, the segments of the mid-Atlantic ridge are offset by some long transform faults. Most of the earthquakes are located along the transform faults, rather than along the spreading segments, although there are clusters of earthquakes at some of the ridge-transform boundaries. Some earthquakes do occur on spreading ridges, but they tend to be small and infrequent because of the relatively high rock temperatures in the areas where spreading is taking place.

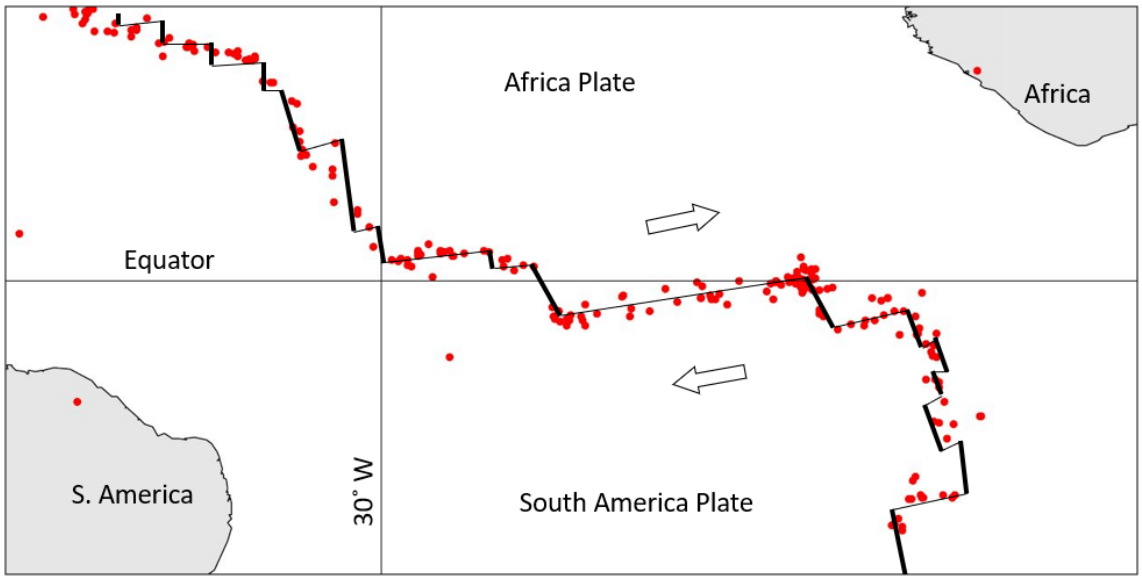


Figure 11.2.2 Distribution of earthquakes of M4 and greater in the area of the mid-Atlantic ridge near the equator from 1990 to 1996. All are at a depth of 0 to 33 kilometers.

Earthquakes at Convergent Boundaries

The distribution and depths of earthquakes in the Caribbean and Central America area are shown in Figure 11.2.3. In this region, the Cocos Plate is subducting beneath the North America and Caribbean Plates (ocean-continent convergence), and the South and North America Plates are subducting beneath the Caribbean Plate (ocean-ocean convergence). In both cases, the earthquakes get deeper with distance from the trench. In Figure 11.2.3, the South America Plate is shown as being subducted beneath the Caribbean Plate in the area north of Colombia, but since there is almost no earthquake activity along this zone, it is questionable whether subduction is actually taking place.

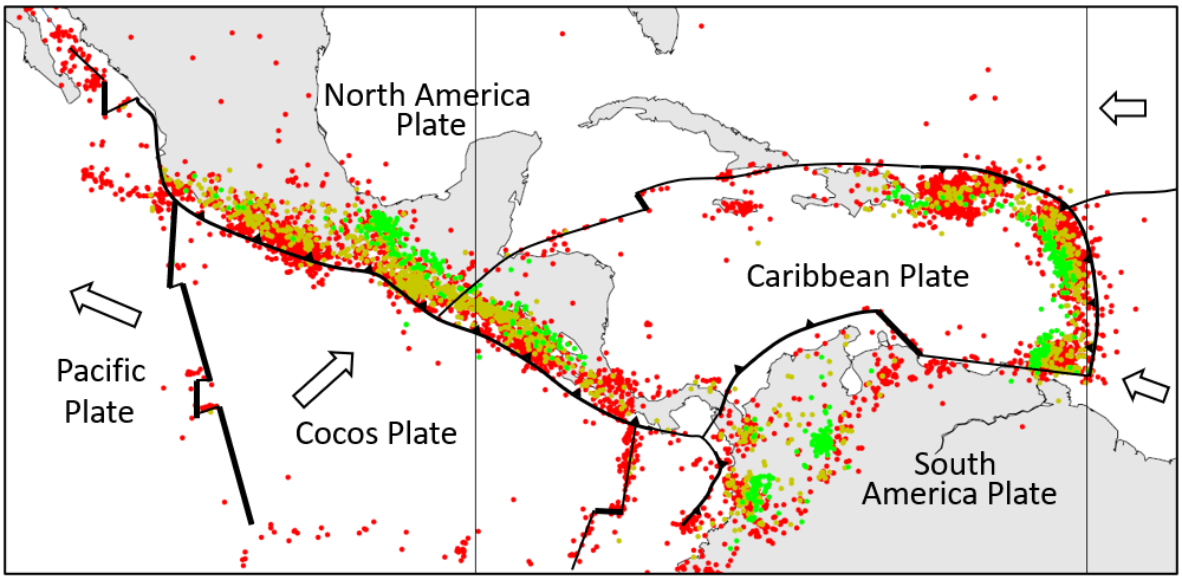


Figure 11.2.3 Distribution of earthquakes of M4 and greater in the Central America region from 1990 to 1996 (red: 0 to 33 kilometers, orange: 33 to 70 kilometers, green: 70 to 300 kilometers, blue: 300 to 700 kilometers) (Spreading ridges are heavy lines, subduction zones are toothed lines, and transform faults are light lines.)

There are also various divergent and transform boundaries in the area shown in Figure 11.2.3, and as we've seen in the mid-Atlantic area, most of these earthquakes occur along the transform faults.

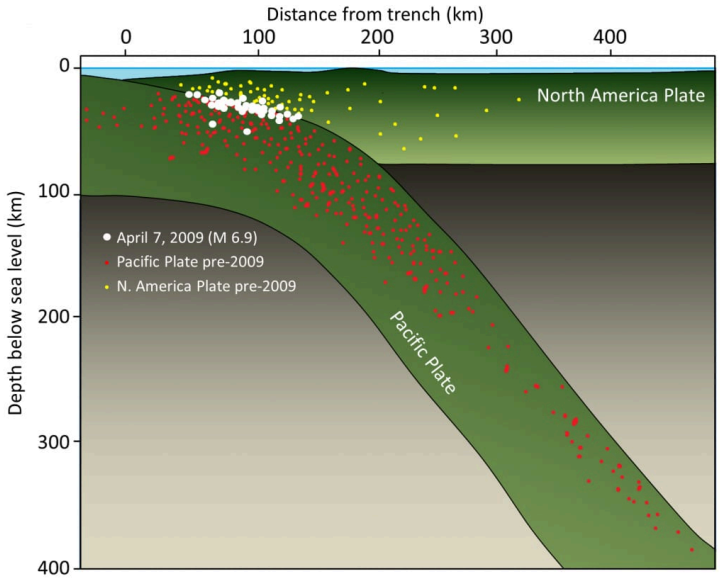


Figure 11.2.4 Distribution of earthquakes in the area of the Kuril Islands, Russia (just north of Japan) (White dots represent the April 2009 M6.9 earthquake. Red and yellow dots are from background seismicity over several years prior to 2009.)

The distribution of earthquakes with depth in the Kuril Islands of Russia in the northwest Pacific is shown in Figure 11.2.4. This is an ocean-ocean convergent boundary. The small red and yellow dots show background seismicity over a number of years, while the larger white dots are individual shocks associated with a M6.9 earthquake in April 2009. The relatively large earthquake took place on the upper part of the plate boundary between 60 kilometers and 140 kilometers inland from the trench. As we saw for the Cascadia subduction zone, this is where large subduction earthquakes are expected to occur.

In fact, all of the very large earthquakes — M9 or higher — take place at subduction boundaries because there is the potential for a greater width of rupture zone on a gently dipping boundary than on a steep transform boundary. The largest earthquakes on transform boundaries are in the order of M8.

The background seismicity at this convergent boundary, and on other similar ones, is predominantly near the upper side of the subducting plate. The frequency of earthquakes is greatest near the surface and especially around the area where large subduction quakes happen, but it extends to at least a 400 kilometre depth. There is also significant seismic activity in the overriding North America Plate, again most commonly near the region of large quakes, but also extending for a few hundred kilometers away from the plate boundary.

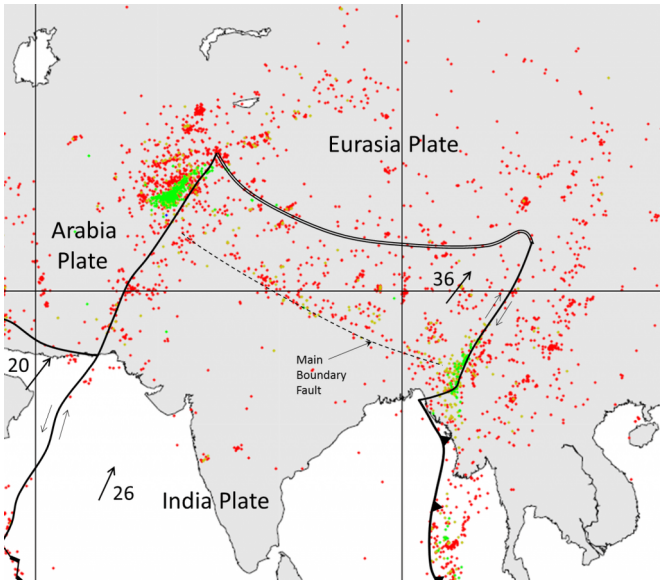


Figure 11.2.5 Distribution of earthquakes in the area where the India Plate is converging with the Asia Plate (data from 1990 to 1996, red: 0 to 33 kilometers, orange: 33 to 70 kilometers, green: 70 to 300 kilometers). (Spreading ridges are heavy lines, subduction zones are toothed lines, and transform faults are light lines. The double line along the northern edge of the India Plate indicates convergence, but not subduction. Plate motions are shown in millimeters per year).

The distribution of earthquakes in the area of the India-Eurasia plate boundary is shown in Figure 11.2.5. This is a continent-continent convergent boundary, and it is generally assumed that although the India Plate continues to move north toward the Asia Plate, there is no actual subduction taking place. There are transform faults on either side of the India Plate in this area.

The entire northern India and southern Asia region is very seismically active. Earthquakes are common in northern India, Nepal, Bhutan, Bangladesh and adjacent parts of China, and throughout Pakistan and Afghanistan. Many of the earthquakes are related to the transform faults on either side of the India Plate, and most of the others are related to the significant tectonic squeezing caused by the continued convergence of the India and Asia Plates. That squeezing has caused the Asia Plate to be thrust over top of the India Plate, building the Himalayas and the Tibet Plateau to enormous heights. Most of the earthquakes of Figure 11.2.5 are related to the thrust faults shown in Figure 11.2.6 (and to hundreds of other similar ones that cannot be shown at this scale). The southernmost thrust fault in Figure 11.2.6 is equivalent to the Main Boundary Fault in Figure 11.2.5.

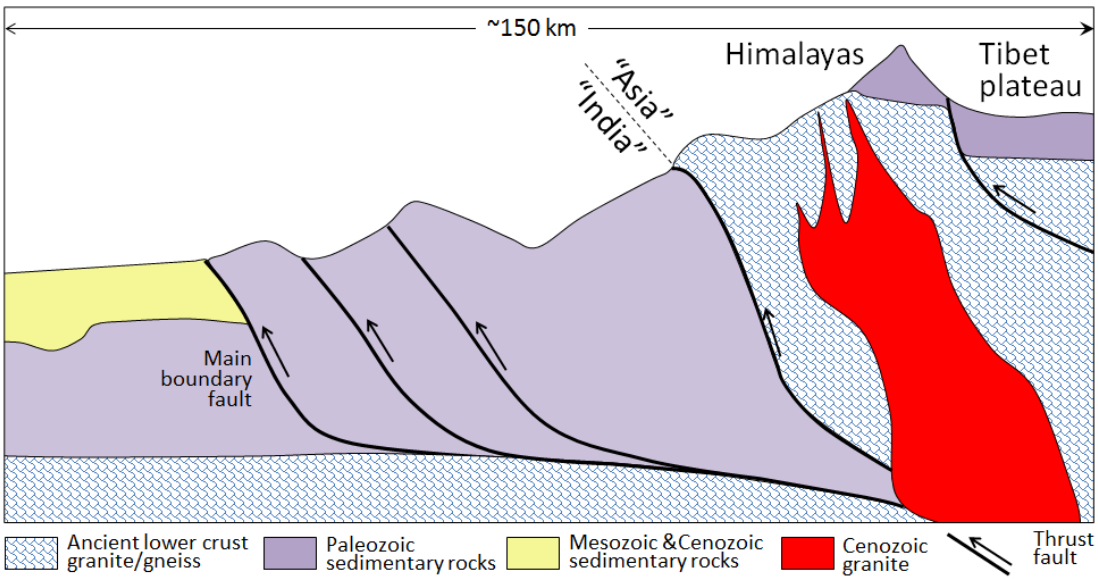


Figure 11.2.6 Schematic diagram of the India-Asia convergent boundary, showing examples of the types of faults along which earthquakes are focused. The devastating Nepal earthquake of May 2015 took place along one of these thrust faults.

There is a very significant concentration of both shallow and deep (greater than 70 kilometers) earthquakes in the northwestern

part of Figure 11.2.5. This is northern Afghanistan, and at depths of more than 70 kilometers, many of these earthquakes are within the mantle as opposed to the crust. It is interpreted that these deep earthquakes are caused by northwestward subduction of part of the India Plate beneath the Asia Plate in this area.

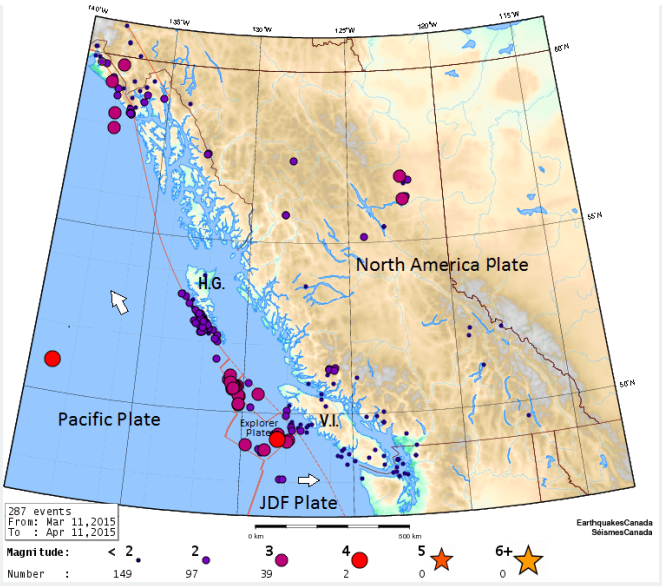


Figure 11.2.7

This map shows the incidence and magnitude of earthquakes in British Columbia over a one-month period in March and April 2015.

1. What is the likely origin of the earthquakes between the Juan de Fuca (JDF) and Explorer Plates?
2. The string of small earthquakes adjacent to Haida Gwaii (H.G.) coincides closely with the rupture surface of the 2012 M7.8 earthquake in that area. How might these earthquakes be related to that one?
3. Most of the earthquakes around Vancouver Island (V.I.) are relatively shallow. What is their likely origin?
4. Some of the earthquakes in B.C. are interpreted as being caused by natural gas extraction (including fracking). Which of the earthquakes here could fall into this category?

See Appendix 3 for Exercise 11.1 answers.

Image descriptions

Figure 11.2.7 image description: The incidence and magnitude of earthquakes in British Columbia over a one-month period in March and April 2015: There were a few dozen smaller earthquakes spread out around Vancouver Island and the sunshine coast with a magnitude of 2. Farther west along the Explorer Plate, which is between the North American plate, the Juan de Fuca Plate, and the Pacific Plate, there were quite a few earthquakes with a magnitude of 3 and at least one earthquake with a magnitude of 4. Between the North American Plate and the Pacific Plate off the south-west coast of Haida Gwaii, there was a large cluster of earthquakes with magnitudes of 2. Along the Alaskan panhandle, there was a collection of 2- and 3-magnitude earthquakes. In addition, there were two 3-magnitude earthquakes west of Fort St. John in northern British Columbia and one or two 2-magnitude earthquakes. In total, this map shows one hundred and forty-nine earthquakes with a magnitude less than 2, ninety-seven earthquakes with a magnitude of 2, thirty-nine earthquakes with a magnitude of 3, and two earthquakes with a magnitude of 4 [Return to Figure 11.2.7]

Media Attributions

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- Figure 11.2.3: Earthquakes Around the Central-American Region © Steven Earle after Dale Sawyer, Rice University.
- Figure 11.2.4: Earthquakes Around the Kuril Islands. © Steven Earle after Gavin Hayes, from data from the USGS [PDF].
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11.3: Measuring Earthquakes

There are two main ways to measure earthquakes. The first of these is an estimate of the energy released, and the value is referred to as **magnitude**. This is the number that is typically used by the press when a big earthquake happens. It is often referred to as "Richter magnitude," but that is a misnomer, and it should be just "magnitude." There are many ways to measure magnitude—including Charles Richter's method developed in 1935—but they are all ways to estimate the same number, which is proportional to the amount of energy released.

The other way of assessing the impact of an earthquake is to assess what people felt and how much damage was done. This is known as **intensity**. Intensity values are assigned to locations, rather than to the earthquake itself, and therefore intensity can vary widely, depending on the proximity to the earthquake and the types of materials underneath and the local conditions.

Earthquake Magnitude

Before we look more closely at magnitude we need to review what we know about body waves, and look at surface waves. Body waves are of two types, P waves, or primary or compression waves (like the compression of the coils of a spring), and S waves, or secondary or shear waves (like the flick of a rope). An example of P and S seismic wave records is shown in Figure 11.3.1 . The critical parameters for the measurement of magnitude are labelled, including the time interval between the arrival of the P- and S-waves—which is used to determine the distance from the earthquake to the seismic station, and the amplitude of the S-waves—which is used to estimate the magnitude of the earthquake.

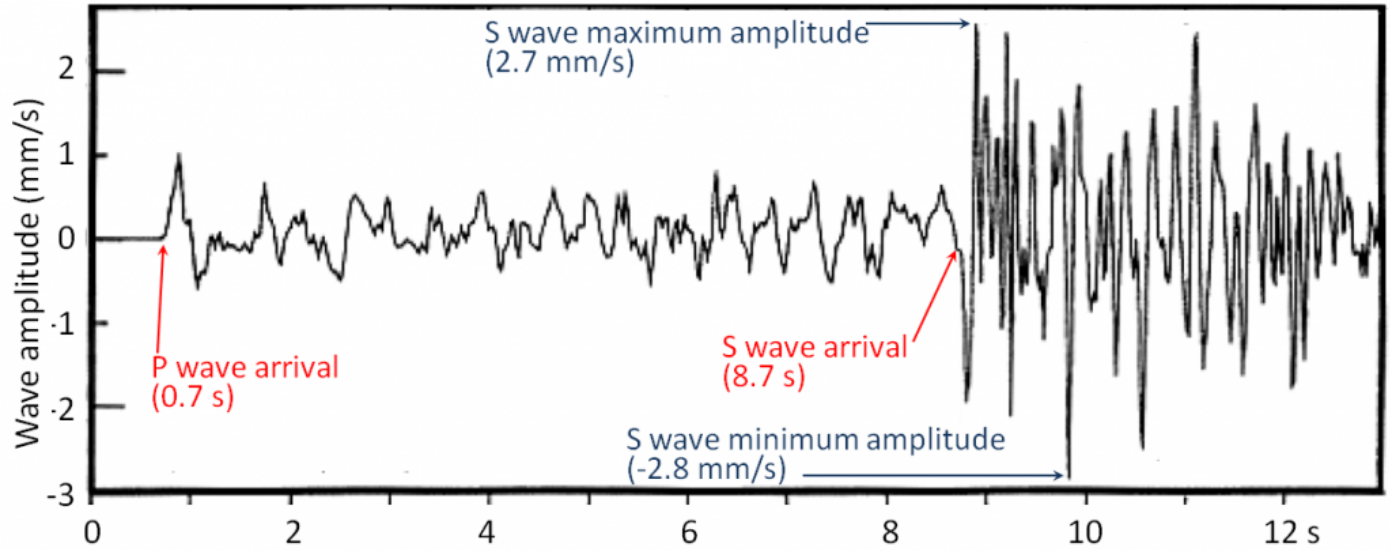


Figure 11.3.1 P-waves and S-waves from a small (M4) earthquake that took place near Vancouver Island in 1997.

When body waves (P or S) reach Earth's surface, some of their energy is transformed into surface waves, of which there are two main types, as illustrated in Figure 11.3.2. **Rayleigh waves** are characterized by vertical motion of the ground surface, like waves on water, while **Love waves** are characterized by horizontal motion. Both Rayleigh and Love waves are about 10% slower than S-waves (so they arrive later at a seismic station). Surface waves typically have greater amplitudes than body waves, and they do more damage.

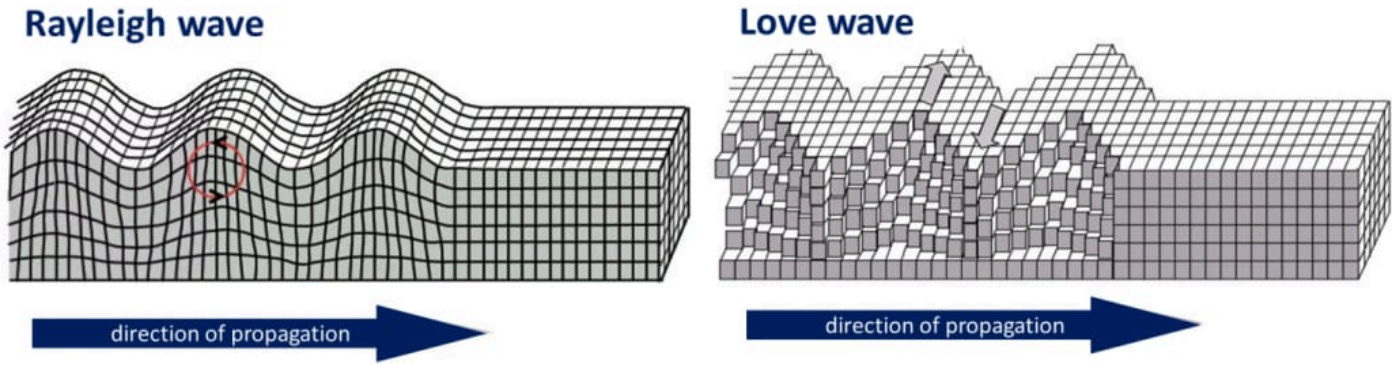


Figure 11.3.2 Depiction of seismic surface waves.

Other important terms for describing earthquakes are **hypocentre** (or **focus**) and **epicentre**. The hypocentre is the actual location of an individual earthquake shock at depth in the ground, and the epicentre is the point on the land surface vertically above the hypocentre (Figure 11.3.3).

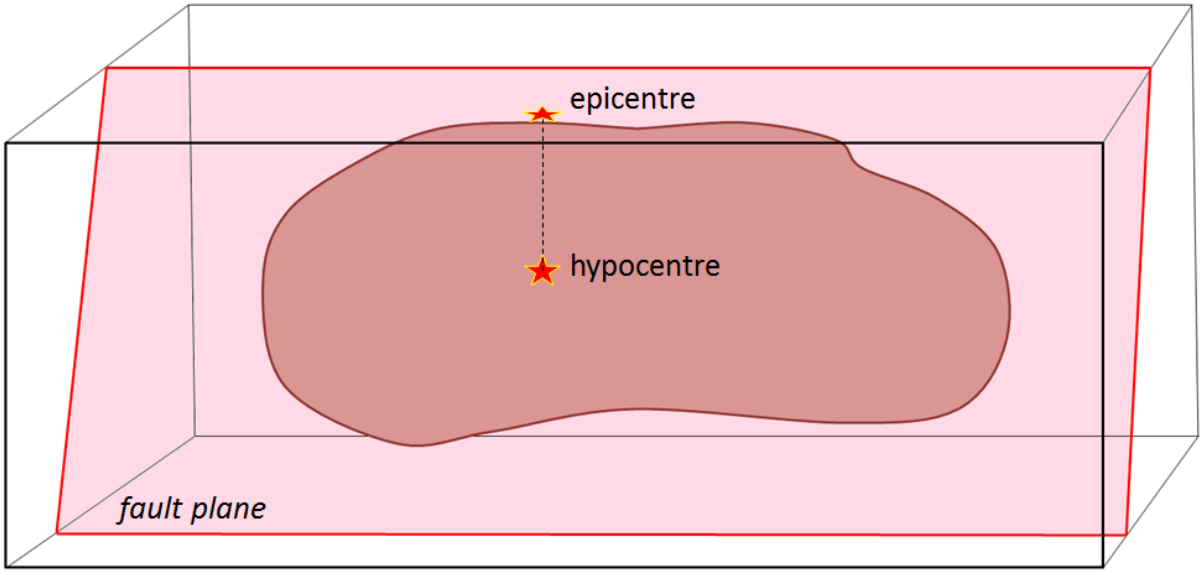


Figure 11.3.3 Epicentre and hypocentre.

A number of methods for estimating magnitude are listed in Table 11.1. Local magnitude (ML) was widely used until late in the 20th century, but **moment magnitude** (MW) is now more commonly used because it gives more accurate estimates (especially with larger earthquakes) and can be applied to earthquakes at any distance from a seismometer. Surface-wave magnitudes can also be applied to measure distant large earthquakes.

Because of the increasing size of cities in earthquake-prone areas (e.g., China, Japan, California) and the increasing sophistication of infrastructure, it is becoming important to have very rapid warnings and magnitude estimates of earthquakes that have already happened. This can be achieved by using P-wave data to determine magnitude because P-waves arrive first at seismic stations, in many cases several seconds ahead of the more damaging S-waves and surface waves. Operators of electrical grids, pipelines, trains, and other infrastructure can use the information to automatically shut down systems so that damage and casualties can be limited.

Table 11.1 A summary of some of the different methods for estimating earthquake magnitude.[1]

[Skip Table]

Type	M Range	Dist. Range	Comments
Local or Richter (M_L)	2 to 6	0 to 400 kilometers	The original magnitude relationship defined in 1935 by Richter and Gutenberg. It is based on the maximum amplitude of S-waves recorded on a Wood-Anderson torsion seismograph. ML values can be calculated using data from modern instruments. L stands for local because it only applies to earthquakes relatively close to the seismic station.
Moment (M_W)	Greater than 3.5	All	Based on the **seismic moment** of the earthquake, which is equal to the average amount of displacement on the fault times the fault area that slipped. It can also be estimated from seismic data if the seismometer is tuned to detect long-period body waves.
Surface wave (M_S)	5 to 8	20 to 180°	A magnitude for distant earthquakes based on the amplitude of surface waves measured at a period near 20 seconds.
P-wave	2 to 8	Local	Based on the amplitude of P-waves. This technique is being increasingly used to provide very rapid magnitude estimates so that early warnings can be sent to utility and transportation operators to shut down equipment before the larger (but slower) S-waves and surface waves arrive.

Exercise 11.2 Moment magnitude estimates from earthquake parameters

Use this moment magnitude calculation tool to estimate the moment magnitude based on the approximate length, width, and displacement values provided in the following table:

Table 11.2 Calculate Moment Magnitude Based on Length, Width, and Displacement Values

[Skip Table]

Length (kilometers)	Width (kilometers)	Displacement (meters)	Earthquake	MW?
60	15	4	The 1946 Vancouver Island earthquake	
0.4	0.2	.5	The small Vancouver Island earthquake shown in Figure 11.3.1	
20	8	4	The 2001 Nisqually earthquake described in Exercise 11.3	
1,100	120	10	The 2004 Indian Ocean earthquake	
30	11	4	The 2010 Haiti earthquake	

The largest recorded earthquake had a magnitude of 9.5. Could there be a 10? You can answer that question using this tool. See what numbers are needed to make MW = 10. Are they reasonable?

See Appendix 3 for Exercise 11.2 answers.

The magnitude scale is logarithmic; in fact, the amount of energy released by an earthquake of M4 is 32 times higher than that released by one of M3, and this ratio applies to all intervals in the scale. If we assign an arbitrary energy level of 1 unit to a M1 earthquake the energy for quakes up to M8 will be as shown on the following table:

Table 11.3 The energy of an earthquake increases by 32 times at each magnitude level.

Magnitude	Energy
1	1
2	32
3	1,024
4	32,768
5	1,048,576
6	33,554,432
7	1,073,741,824
8	34,359,738,368

In any given year, when there is a large earthquake on Earth (M8 or M9), the amount of energy released by that one event will likely exceed the energy released by all smaller earthquake events combined.

Earthquake Intensity

The intensity of earthquake shaking at any location is determined not only by the magnitude of the earthquake and its distance, but also by the type of underlying rock or unconsolidated materials. If buildings are present, the size and type of buildings (and their inherent natural vibrations) are also important.

Intensity scales were first used in the late 19th century, and then adapted in the early 20th century by Giuseppe Mercalli and modified later by others to form what we know call the modified Mercalli intensity scale (Table 11.4). Intensity estimates are important because they allow us to characterize parts of any region into areas that are especially prone to strong shaking versus those that are not. The key factor in this regard is the nature of the underlying geological materials, and the weaker those are, the more likely it is that there will be strong shaking. Areas underlain by strong solid bedrock tend to experience much less shaking than those underlain by unconsolidated river or lake sediments.

Table 11.4 The modified Mercalli intensity scale.

[Skip Table]

Level of intensity	Description
Not felt (1)	Not felt except by a very few under especially favourable conditions
Weak (2)	Felt only by a few persons at rest, especially on upper floors of buildings
Weak (3)	Felt quite noticeably by persons indoors, especially on upper floors of buildings; many people do not recognize it as an earthquake; standing motor cars may rock slightly; vibrations similar to the passing of a truck; duration estimated
Light (4)	Felt indoors by many, outdoors by few during the day; at night, some awakened; dishes, windows, doors disturbed; walls make cracking sound; sensation like heavy truck striking building; standing motor cars rocked noticeably
Moderate (5)	Felt by nearly everyone; many awakened; some dishes, windows broken; unstable objects overturned; pendulum clocks may stop
Strong (6)	Felt by all, many frightened; some heavy furniture moved; a few instances of fallen plaster; damage slight
Very Strong (7)	Damage negligible in buildings of good design and construction; slight to moderate in well-built ordinary structures; considerable damage in poorly built or badly designed structures; some chimneys broken

[Skip Table]

Level of intensity	Description
Severe (8)	Damage slight in specially designed structures; considerable damage in ordinary substantial buildings with partial collapse; damage great in poorly built structures; fall of chimneys, factory stacks, columns, monuments, walls; heavy furniture overturned
Violent (9)	Damage considerable in specially designed structures; well-designed frame structures thrown out of plumb; damage great in substantial buildings, with partial collapse; buildings shifted off foundations
Extreme (10)	Some well-built wooden structures destroyed; most masonry and frame structures destroyed with foundations; rails bent
Extreme (11)	Few, if any (masonry), structures remain standing; bridges destroyed; broad fissures in ground; underground pipelines completely out of service; earth slumps and land slips in soft ground; rails bent greatly
Extreme (12)	Damage total; waves seen on ground surfaces; lines of sight and level distorted; objects thrown upward into the air

An example of this effect is the 1985 M8 earthquake that struck the Michoacán region of western Mexico, southwest of Mexico City. There was relatively little damage in the area around the epicentre, but there was tremendous damage and about 5,000 deaths in heavily populated Mexico City some 350 kilometers from the epicentre. The key reason for this is that Mexico City was built largely on the unconsolidated and water-saturated sediment of former Lake Texcoco. These sediments resonate at a frequency of about two seconds, which was similar to the frequency of the body waves that reached the city. For the same reason that a powerful opera singer can break a wine glass by singing the right note, the amplitude of the seismic waves was amplified by the lake sediments. Survivors of the disaster recounted that the ground in some areas moved up and down by about 20 centimeters every two seconds for over two minutes. Damage was greatest to buildings between 5 and 15 storeys tall, because they also resonated at around two seconds, which amplified the shaking.

Exercise 11.3 Estimating intensity from personal observations

The following observations were made by residents of the Nanaimo area during the M6.8 Nisqually earthquake near Olympia, Washington in 2001. Estimate the Mercalli intensities using Table 11.4.

Table 11.5

[Skip Table]

Building Type	Floor	Shaking Felt	How long it lasted (in seconds)	Description of Motion	Intensity?
House	1	no	10	Heard a large rumble lasting not even 10 seconds, mirror swayed	
House	2	moderate	60	Candles, pictures and CDs on bookshelf moved, towels fell off racks	
House	1	no		Pots hanging over stove moved and crashed together	
House	1	weak		Rolling feeling with a sudden stop, picture fell off mantle, chair moved	
Apartment	1	weak	10	Sounded like a big truck then everything shook for a short period	
House	1	moderate	20-30	Teacups rattled but didn't fall off	
Institution	2	moderate	15	Creaking sounds, swaying movement of shelving	
House	1	moderate	15-30	Bed banging against the wall with me in it, dog barking aggressively	

See Appendix 3 for Exercise 11.3 answers.

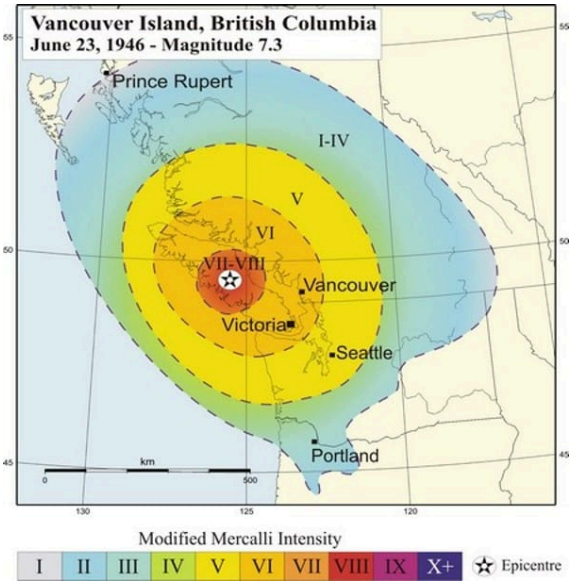


Figure 11.3.4 Intensity map for the 1946 M7.3 Vancouver Island earthquake. [Long Decsription]

An intensity map for the 1946 M7.3 Vancouver Island earthquake is shown in Figure 11.3.4. The intensity was greatest in the central island region where, in some communities, chimneys were damaged on more than 75% of buildings, some roads were made impassable, and a major rock slide occurred. The earthquake was felt as far north as Prince Rupert, as far south as Portland Oregon, and as far east as the Rockies.

Image Descriptions

Figure 11.3.1 **image description:** P-waves and S-waves from a small (M4) earthquake near Vancouver Island in 1997. The P-wave arrived in 0.7 seconds with an amplitude ranging from negative 0.7 millimeters per second to 1.1 millimeters per second and lasting until the arrival of the S-wave. The S-wave arrived at 8.7 seconds, with a minimum amplitude of negative 2.8 millimeters per second and a maximum amplitude of 2.7 millimeters per second. The S-wave's net amplitude gradually decreased over the next 5 seconds. [Return to Figure 11.3.1]

Figure 11.3.4 **image description:** The graduated intensity of the 1945 M7.3 Vancouver Island earthquake based on the modified Mercalli intensity scale. The area surrounding the epicentre of the earthquake which included central Vancouver Island ranged between a very strong (7) and severe (8) intensity. The next ring included the northern and southern parts of Vancouver Island, as well as a part of the main land coast including Vancouver and much of the Sunshine coast a strong (6) intensity. The next ring, which reached experienced a moderate (5) intensity, included Seattle and much of the BC interior. The outermost ring ranged between not felt (1) and light (4) intensity. It was felt as far north as Prince Rupert and the southern tip of Haida Gwaii, south eastern BC, and as far south as north western Oregon. [Return to Figure 11.3.4]

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- **Figure** 11.3.2 **(right):** "Love Wave" © Nicoguaro. Adapted by Steven Earle. CC BY.
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1. Table 11.1 by Steven Earle. ↵

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11.4: The Impacts of Earthquakes

Some of the common impacts of earthquakes include structural damage to buildings, fires, damage to bridges, highways, pipelines and electrical transmission lines, initiation of slope failures, liquefaction, and tsunami. The types of impacts depend to a large degree on where the earthquake is located: whether it is predominantly urban or rural, densely or sparsely populated, highly developed or underdeveloped, and of course on the ability of the infrastructure to withstand shaking.

As we've seen from the example of the 1985 Mexico earthquake, the geological foundations on which structures are built can have a significant impact on earthquake shaking. When an earthquake happens, the seismic waves produced have a wide range of frequencies. The energy of the higher frequency waves tends to be absorbed by solid rock, while the lower frequency waves (with periods slower than one second) pass through the solid rock without being absorbed, but are eventually absorbed and amplified by soft sediments. It is therefore very common to see much worse earthquake damage in areas underlain by soft sediments than in areas of solid rock. A good example of this is in the Oakland area near San Francisco, where parts of a two-layer highway built on soft sediments collapsed during the 1989 Loma Prieta earthquake (Figure 11.4.1).



Figure 11.4.1 A part of the Cypress Freeway in Oakland California that collapsed during the 1989 Loma Prieta earthquake.

Building damage is also greatest in areas of soft sediments, and multi-storey buildings tend to be more seriously damaged than smaller ones. Buildings can be designed to withstand most earthquakes, and this practice is increasingly applied in earthquake-prone regions. Turkey is one such region, and even though Turkey had a relatively strong building code in the 1990s, adherence to the code was poor, as builders did whatever they could to save costs, including using inappropriate materials in concrete and reducing the amount of steel reinforcing. The result was that there were over 17,000 deaths in the 1999 M7.6 Izmit earthquake (Figure 11.4.2). After two devastating earthquakes that year, Turkish authorities strengthened the building code further, but the new code has been applied only in a few regions, and enforcement of the code is still weak, as revealed by the amount of damage from a M7.1 earthquake in eastern Turkey in 2011.



Figure 11.4.2 Buildings damaged by the 1999 earthquake in the Izmit area, Turkey.

Fires are commonly associated with earthquakes because fuel pipelines rupture and electrical lines are damaged when the ground shakes (Figure 11.4.3). Most of the damage in the great 1906 San Francisco earthquake was caused by massive fires in the downtown area of the city (Figure 11.4.4). Some 25,000 buildings were destroyed by those fires, which were fuelled by broken gas pipes. Fighting the fires was difficult because water mains had also ruptured. The risk of fires can be reduced through P-wave early warning systems if utility operators can reduce pipeline pressure and close electrical circuits.



Figure 11.4.3 Some of the effects of the 2011 Tohoku earthquake in the Sendai area of Japan. An oil refinery is on fire, and a vast area has been flooded by a tsunami.



Figure 11.4.4 Fires in San Francisco following the 1906 earthquake.



Figure 11.4.5 The Las Colinas debris flow at Santa Tecla (a suburb of the capital San Salvador) triggered by the January 2001 El Salvador earthquake. This is just one of many hundreds of slope failures that resulted from that earthquake. Over 500 people died in the area affected by this slide.

Earthquakes are important triggers for failures on slopes that are already weak. An example is the Las Colinas slide in the city of Santa Tecla, El Salvador, which was triggered by a M7.6 offshore earthquake in January 2001 (Figure 11.4.5).

Ground shaking during an earthquake can be enough to weaken rock and unconsolidated materials to the point of failure, but in many cases the shaking also contributes to a process known as **liquefaction**, in which an otherwise solid body of sediment is transformed into a liquid mass that can flow. When water-saturated sediments are shaken, the grains become rearranged to the point where they are no longer supporting one another. Instead, the water between the grains is holding them apart and the material can flow. Liquefaction can lead to the collapse of buildings and other structures that might be otherwise undamaged. A good example is the collapse of apartment buildings during the 1964 Niigata earthquake (M7.6) in Japan (Figure 11.4.6). Liquefaction can also contribute to slope failures and to fountains of sandy mud (sand volcanoes) in areas where there is loose saturated sand beneath a layer of more cohesive clay.



Figure 11.4.6 Collapsed apartment buildings in the Niigata area of Japan. The material beneath the buildings was liquefied to varying degrees by the 1964 earthquake.

Parts of the Fraser River delta are prone to liquefaction-related damage because the region is characterized by a 2 metre to 3 metre thick layer of fluvial silt and clay over top of at least 10 meters of water-saturated fluvial sand (Figure 11.4.7). Under these

conditions, it can be expected that seismic shaking will be amplified and that the sandy sediments will liquefy. This could lead to subsidence and tilting of buildings, and to failure and sliding of the silt and clay layer. Current building-code regulations in the Fraser delta area require that measures be taken to strengthen the ground underneath multi-storey buildings prior to construction.

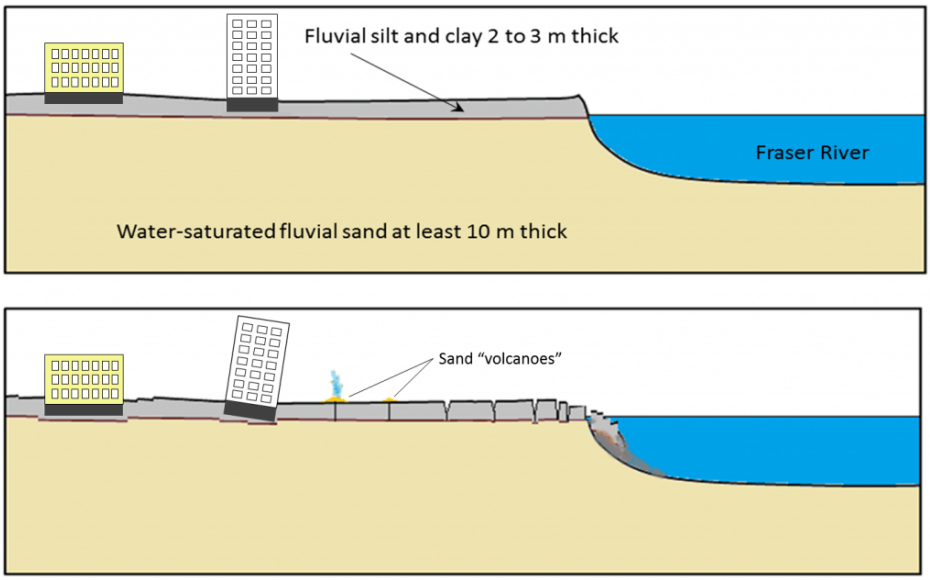


Figure 11.4.7 Recent unconsolidated sedimentary layers in the Fraser River delta area (top) and the potential consequences in the event of a damaging earthquake.

Exercise 11.4 Creating liquefaction and discovering the harmonic frequency

There are a few ways that you can demonstrate the process of liquefaction for yourself. The simplest is to go to a sandy beach (lake, ocean, or river) and find a place near the water's edge where the sand is wet. This is best done with your shoes off, so let's hope it's not too cold! While standing in one place on a wet part of the beach, start moving your feet up and down at a frequency of about once per second. Within a few seconds the previously firm sand will start to lose strength, and you'll gradually sink in up to your ankles.



Figure 11.4.8

If you can't get to a beach, or if the weather isn't cooperating, put some sand (sandbox sand will do) into a small container, saturate it with water, and then pour the excess water off. You can shake it gently to get the water to separate and then pour the excess water away, and you may have to do that more than once. Place a small rock on the surface of the sand; it should sit there

for hours without sinking in. Now, holding the container in one hand gently thump the side or the bottom with your other hand, about twice a second. The rock should gradually sink in as the sand around it becomes liquefied.

As you were moving your feet up and down or thumping the pot, it's likely that you soon discovered the most effective rate for getting the sand to liquefy; this would have been close to the natural harmonic frequency for that body of material. Stepping up and down as fast as you can (several times per second) on the wet beach would not have been effective, nor would you have achieved much by stepping once every several seconds. The body of sand vibrates most readily in response to shaking that is close to its natural harmonic frequency, and liquefaction is also most likely to occur at that frequency.

See Appendix 3 for Exercise 11.4 answers.

Earthquakes that take place beneath the ocean have the potential to generate **tsunami**. (Tsunami is the Japanese word for harbour wave. It is the same in both singular and plural.) The most likely situation for a significant tsunami is a large (M7 or greater) subduction-related earthquake. As shown in Figure 11.4.9, during the time between earthquakes the overriding plate becomes distorted by elastic deformation; it is squeezed laterally (Figure 11.4.9B) and pushed up.

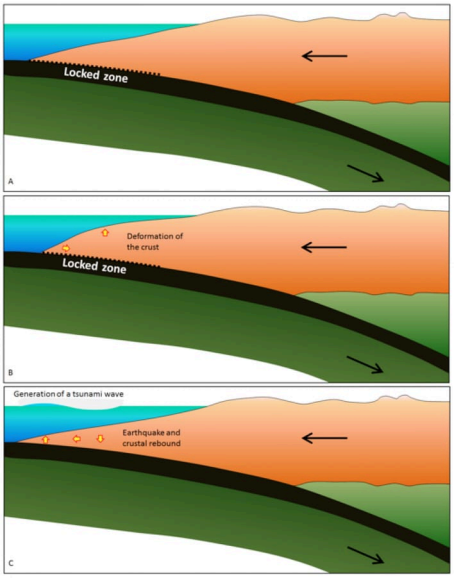


Figure 11.4.9 Elastic deformation and rebound of overriding plate at a subduction setting (B). The release of the locked zone during an earthquake (C) results in both uplift and subsidence on the sea floor, and this is transmitted to the water overhead, resulting in a tsunami.

When an earthquake happens (Figure 11.4.9C), the plate rebounds and there is both uplift and subsidence on the sea floor, in some cases by as much as several meters vertically over an area of thousands of square kilometers. This vertical motion is transmitted through the water column where it generates a series of waves that then spread across the ocean.

Subduction earthquakes with magnitude less than 7 do not typically generate significant tsunami because the amount of vertical displacement of the sea floor is minimal. Sea-floor transform earthquakes, even large ones (M7 to M8), don't typically generate tsunami either, because the motion is mostly side to side, not vertical.

Tsunami waves travel at velocities of several hundred kilometers per hour and easily make it to the far side of an ocean in about the same time as a passenger jet. The simulated one shown in Figure 11.4.10 is similar to that created by the 1700 Cascadia earthquake off the coast of British Columbia, Washington, and Oregon, which was recorded in Japan nine hours later.

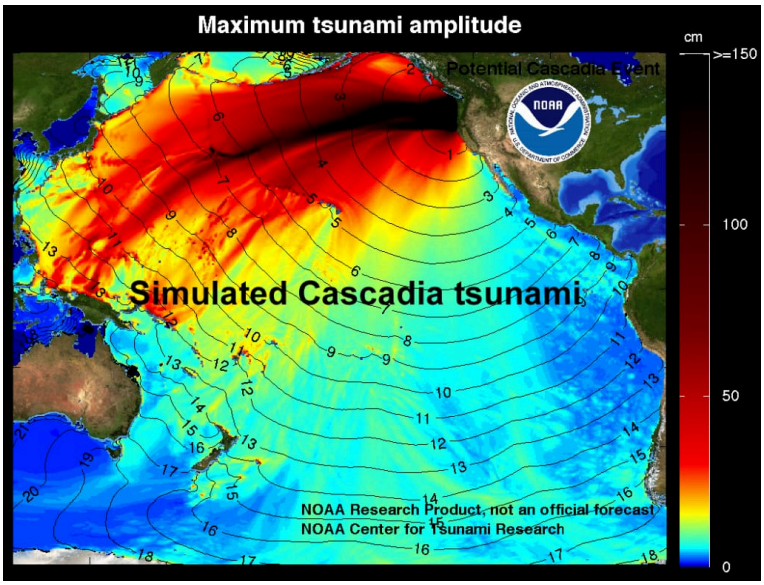


Figure 11.4.10 Model of the tsunami from the 1700 Cascadia earthquake (around M9) showing open-ocean wave heights (colors) and travel time contours. Tsunami wave amplitudes typically increase in shallow water.

Tsunami are discussed further in Chapter 17 under the topic of waves and coasts.

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11.5: Forecasting Earthquakes and Minimizing Damage and Casualties

It has long been a dream of seismologists, geologists, and public safety officials, to be able to accurately predict the location, magnitude, and timing of earthquakes on time scales that would be useful for minimizing danger to the public and damage to infrastructure (e.g., weeks, days, hours). Many different avenues of prediction have been explored, such as using observations of warning foreshocks, changes in magnetic fields, seismic tremor, changing groundwater levels, strange animal behaviour, observed periodicity, stress transfer considerations, and others. So far, none of the research into earthquake prediction has provided a reliable method. Although there are some reports of successful earthquake predictions, they are rare, and many are surrounded by doubtful circumstances.

The problem with earthquake predictions, as with any other type of prediction, is that they have to be accurate *most* of the time, not just *some* of the time. We have come to rely on weather predictions because they are generally (and increasingly) accurate. But if we try to predict earthquakes and are only accurate 10% of the time (and even that isn't possible with the current state of knowledge), the public will lose faith in the process very quickly, and then will ignore all of the predictions. Efforts are currently focused on forecasting earthquake probabilities, rather than predicting their occurrence.

There was great hope for earthquake predictions late in the 1980s when attention was focused on part of the San Andreas Fault at Parkfield, about 200 kilometers south of San Francisco. Between 1881 and 1965 there were five earthquakes at Parkfield, most spaced at approximately 20-year intervals, all confined to the same 20 kilometre-long segment of the fault, and all very close to M6 (Figure 11.5.1). Both the 1934 and 1966 earthquakes were preceded by small foreshocks exactly 17 minutes before the main quake.

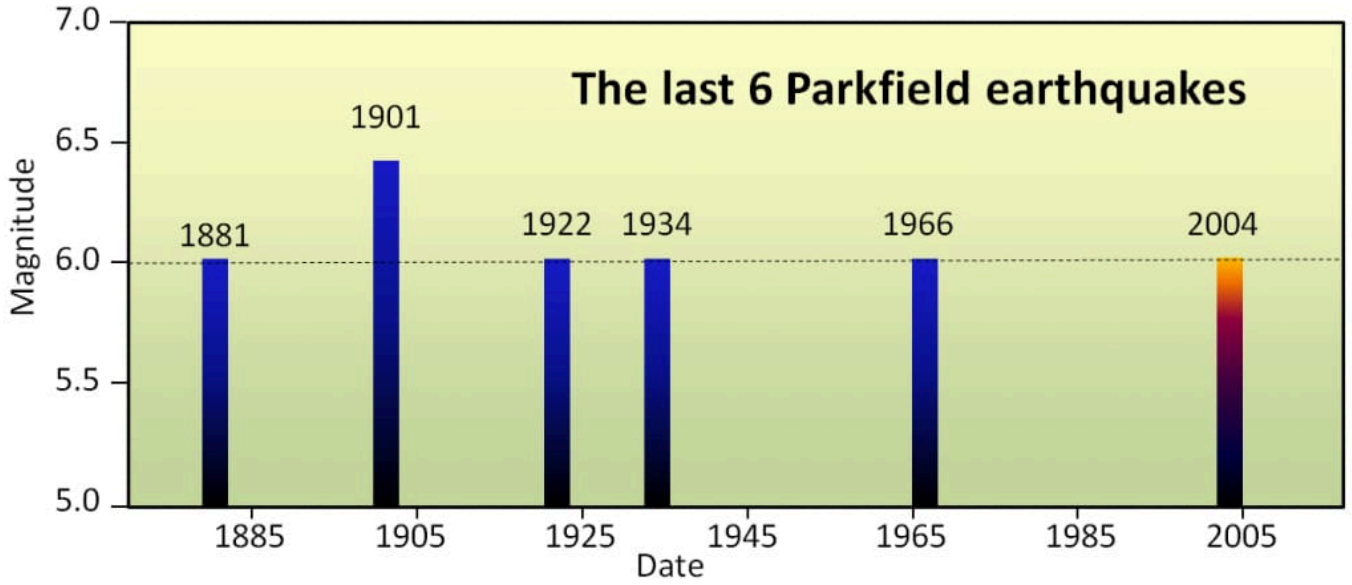


Figure 11.5.1 Earthquakes on the Parkfield segment of the San Andreas Fault between 1881 and 2004.

The U.S. Geological Survey recognized this as an excellent opportunity to understand earthquakes and earthquake prediction, so they armed the Parkfield area with a huge array of geophysical instruments and waited for the next quake, which was expected to happen around 1987. Nothing happened! The "1987 Parkfield earthquake" finally struck in September 2004. Fortunately all of the equipment was still there, but it was no help from the perspective of earthquake prediction. There were no significant precursors to the 2004 Parkfield earthquake in any of the parameters measured, including seismicity, harmonic tremor, strain (rock deformation), magnetic field, the conductivity of the rock, or creep, and there was no foreshock. In other words, even though every available technique was used to monitor it, the 2004 earthquake came as a complete surprise, with no warning whatsoever.

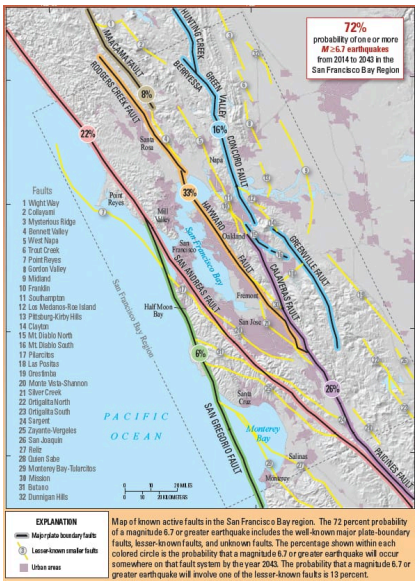


Figure 11.5.2 Probabilities of a M6.7 or larger earthquake over the period 2014 to 2043 on various faults in the San Francisco Bay region of California.

The hope for earthquake prediction is not dead, but it was hit hard by the Parkfield experiment. The current focus in earthquake-prone regions is to provide forecasts of the probability of an earthquake of a certain magnitude within a certain time period—typically a number of decades—while officials focus on ensuring that the population is educated about earthquake risks and that buildings and other infrastructure are as safe as can be. An example of this approach for the San Francisco Bay region of California is shown in Figure 11.5.2. Based on a wide range of information, including past earthquake history, accumulated stress from plate movement, and known stress transfer, seismologists and geologists have predicted the likelihood of a M6.7 or greater earthquake on each of eight major faults that cut through the region. The greatest probabilities are on the Hayward, Rogers Creek, Calaveras, and San Andreas Faults. As shown in Figure 11.5.2, there is a 72% chance that a major and damaging earthquake will take place somewhere in the region prior to 2043.

As we've discussed already, it's not sufficient to have strong building codes, they have to be enforced. Building code compliance is quite robust in most developed countries, but is sadly inadequate in many developing countries.

It's also not enough just to focus on new buildings; we have to make sure that existing buildings—especially schools and hospitals—and other structures such as bridges and dams, are as safe as they can be. An example of how this is applied to schools in B.C. is described in Exercise 11.5.

Making the Seismic Upgrade in B.C.'s Schools

British Columbia is in the middle of a multi-billion-dollar program to make schools safer for students. The program is focused on older schools, because, according to the government, those built since 1992 already comply with modern seismic codes. Some schools would require too much work to make upgrading economically feasible and they are slowly being replaced. Where upgrading is feasible, the school is assessed carefully before any upgrade work is initiated.

An example is Sangster Elementary in Colwood on southern Vancouver Island. The school was originally built in 1957, with a major addition in 1973. Ironically, the newer part of the school, built of concrete blocks, required strengthening with the addition of a steel framework, while the 1957 part, which is a wood-frame building, did not require seismic upgrading. The work was completed in 2014.



Figure 11.5.3 Sangster Elementary image from Google Maps – street view.

As of July 2019, upgrades had been completed at 168 B.C. schools, 17 were underway, and an additional 26 were ready to proceed, with funding identified. Another 274 schools were listed as needing upgrades.[1]

Exercise 11.5 Is your local school on the seismic upgrade list?

Here is The B.C. Ministry of Education's list of schools in the seismic mitigation program as of July 2019: Progress Report: Seismic Mitigation Program (BC Schools).

If you live in B.C., you can check to see if any of the schools in your area are on the list. If so, you might be able to find out, either from the school or on the Internet, what type of work has been done or is planned.

The seismic mitigation program has a strong focus on the Lower Mainland and Vancouver Island. Why do you think that is the case, and is it reasonable?

See Appendix 3 for Exercise 11.5 answers.

The final part of earthquake preparedness involves the formulation of public emergency plans, including escape routes, medical facilities, shelters, and food and water supplies. It also includes personal planning, such as emergency supplies (food, water, shelter, and warmth), escape routes from houses and offices, and communication strategies (with a focus on ones that don't involve the cellular network).

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1. "Seismic Mitigation Program Progress Report, July 2019." Accessed July 21, 2019 from https://www2.gov.bc.ca/gov/content/e...mic-mitigation ↵

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11.6: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
11.1 What is an Earthquake?	An earthquake is the shaking that results when a body of rock that has been deformed breaks and the two sides quickly slide past each other. The rupture is initiated at a point but quickly spreads across an area of a fault, via a series of aftershocks initiated by stress transfer. Episodic tremor and slip is a periodic slow movement, accompanied by harmonic tremors, along the middle part of a subduction zone boundary.
11.2 Earthquakes and Plate Tectonics	Most earthquakes take place at or near plate boundaries, especially at transform boundaries (where most quakes are at less than a 30 kilometre depth) and at convergent boundaries (where they can be at well over a 100 kilometre depth). The largest earthquakes happen at subduction zones, typically in the upper section where the rock is relatively cool.
11.3 Measuring Earthquakes	Magnitude is a measure of the amount of energy released by an earthquake, and it is proportional to the area of the rupture surface and to the amount of displacement. Although any earthquake has only one magnitude value, it can be estimated in various ways, mostly involving seismic data. Intensity is a measure of the amount of shaking experienced and damage done at a particular location around the earthquake. Intensity will vary depending on the distance to the epicentre, the depth of the earthquake, and the geological nature of the material below surface.
11.4 The Impacts of Earthquakes	Damage to buildings is the most serious consequence of most large earthquakes. The amount of damage is related to the type and size of buildings, how they are constructed, and the nature of the material on which they are built. Other important consequences are fires, damage to bridges and highways, slope failures, liquefaction, and tsunami. Tsunami, which are almost all related to large subduction earthquakes, can be devastating.
11.5 Forecasting Earthquakes and Minimizing Damage and Casualties	There is no reliable technology for predicting earthquakes, but the probability of one happening within a certain time period can be forecast. We can minimize earthquake impacts by ensuring that citizens are aware of the risk, that building codes are enforced, that existing buildings like schools and hospitals are seismically sound, and that both public and personal emergency plans are in place.

Questions for Review

1. Define the term *earthquake*.
2. How does elastic rebound theory help to explain how earthquakes happen?
3. What is a rupture surface, and how does the area of a rupture surface relate to earthquake magnitude?
4. What is an aftershock and what is the relationship between aftershocks and stress transfer?
5. Episodic slip on the middle part of the Cascadia subduction zone is thought to result in an increase in the stress on the upper part where large earthquakes take place. Why?
6. Explain the difference between magnitude and intensity as expressions of the size of an earthquake.
7. How much more energy is released by an M7.3 earthquake compared with an M5.3 earthquake?
8. Figure A shows a map of earthquake locations with the depths coded according the color scheme used in Figure 11.6.5. What type of plate boundary is this?

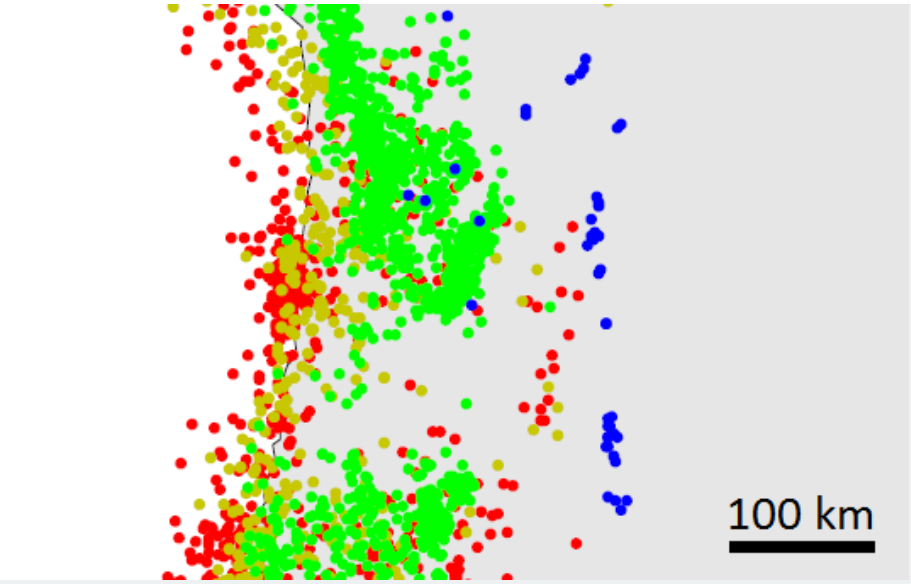


Figure A

- Draw a line on the map to show approximately where the plate boundary is situated.
- In which directions are the plates moving, and where in the world might this be?
- Earthquakes are relatively common along the mid-ocean ridges. At what type of plate boundary do most such quakes occur?
- The northward motion of the Pacific Plate relative to the North America Plate takes place along two major transform faults. What are they called?
- Why is earthquake damage likely to be more severe for buildings built on unconsolidated sediments as opposed to solid rock?
- Why are fires common during earthquakes?
- What type of earthquake is likely to lead to a tsunami?
- What did we learn about earthquake prediction from the 2004 Parkfield earthquake?
- What are some of the things we should know about an area in order to help minimize the impacts of an earthquake?
- Answers to Review Questions can be found in Appendix 2.

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12: GEOLOGICAL STRUCTURES

12.1: Stress and Strain
12.2: Folding
12.3: Fracturing and Faulting
12.4: Measuring Geological Structures
12.5: Summary

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12.1: Stress and Strain

Rocks are subject to **stress**—mostly related to plate tectonics but also to the weight of overlying rocks—and their response to that stress is **strain** (deformation). In regions close to where plates are converging stress is typically compressive—the rocks are being squeezed. Where plates are diverging the stress is extensive—rocks are being pulled apart. At transform plate boundaries, where plates are moving side by side there is sideways or **shear stress**—meaning that there are forces in opposite directions parallel to a plane. Rocks have highly varying strain responses to stress because of their different compositions and physical properties, and because temperature is a big factor and rock temperatures within the crust can vary greatly.

We can describe the stress applied to a rock by breaking it down into three dimensions—all at right angles to one-another (Figure 12.1.1). If the rock is subject only to the pressure of burial, the stresses in all three directions will likely be the same. If it is subject to both burial and tectonic forces, the pressures will be different in different directions.

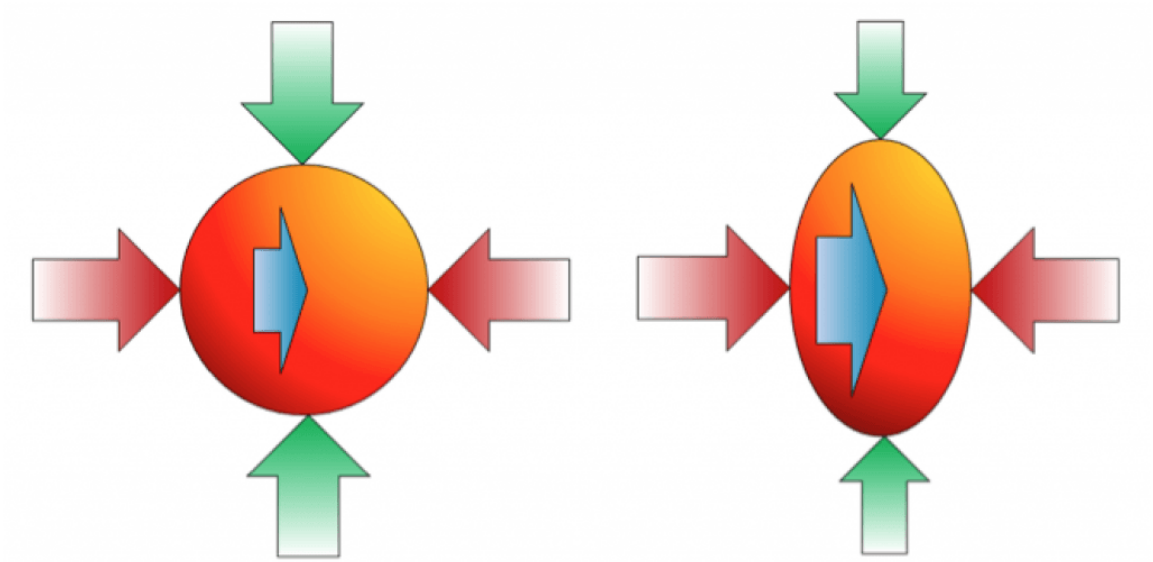


Figure 12.1.1 Depiction of the stress applied to rocks within the crust. The stress can be broken down into three components. Assuming that we're looking down in this case, the green arrows represent north-south stress, the red arrows represent east-west stress, and the blue arrows (the one underneath is not visible) represent up-down stress. On the left, all of the stress components are the same. On the right, the north-south stress is least and the up-down stress is greatest.

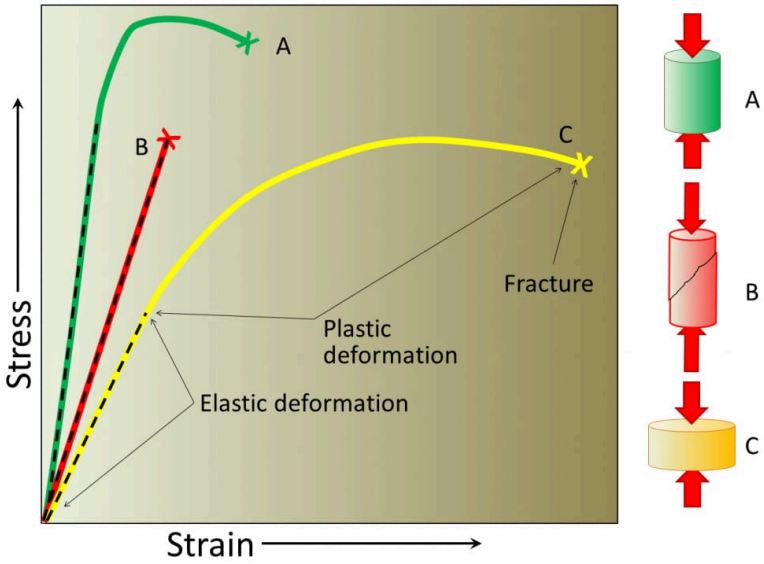


Figure 12.1.2 The varying types of response of geological materials to stress. The straight dashed parts are elastic strain and the curved parts are plastic strain. In each case the X marks where the material fractures. A, the strongest material, deforms relatively little and breaks at a high stress level. B, strong but brittle, shows no plastic deformation and breaks after relatively little elastic deformation. C, the most deformable, breaks only after significant elastic and plastic strain. The three deformation diagrams on the right show A and C before breaking and B after breaking.

Rock can respond to stress in three ways: it can deform elastically, it can deform plastically, and it can break or fracture. Elastic strain is reversible; if the stress is removed, the rock will return to its original shape just like a rubber band that is stretched and released. Plastic strain is not reversible. As already noted, different rocks at different temperatures will behave in different ways to stress. Higher temperatures lead to more plastic behaviour. Some rocks or sediments are also more plastic when they are wet. Another factor is the rate at which the stress is applied. If the stress is applied quickly (for example, because of an extraterrestrial impact or an earthquake), there will be an increased tendency for the rock to fracture. Some different types of strain response are illustrated in Figure 12.1.2.

The outcomes of placing rock under stress are highly variable, but they include fracturing, tilting and folding, stretching and squeezing, and faulting. A fracture is a simple break that does not involve significant movement of the rock on either side. Fracturing is particularly common in volcanic rock, which shrinks as it cools. The basalt columns in Figure 12.1.3a are a good example of fracture. Beds are sometimes tilted by tectonic forces, as shown in Figure 12.1.3b, or folded as shown in Figure 12.1.1.



Figure 12.1.3 Rock structures caused by various types of strain within rocks that have been stressed. (A) Fracturing in basalt near to Whistler, BC; (B) Tilting of sedimentary rock near to Exshaw, Alberta; (C) Stretching of limestone at Quadra Island, BC. The light grey rock is limestone and the dark rock is chert. The body of rock has been stretched parallel to bedding. The chert, which is not elastic, has broken into fragments which are called boudins; (D) Faulting within shale beds at McAbee, near to Cache Creek, BC. The fault runs from the lower right to the upper left, and the upper rock body has been pushed up and to the left.

When a body of rock is compressed in one direction it is typically extended (or stretched) in another. This is an important concept because some geological structures only form under compression, while others only form under tension. Most of the rock in Figure 12.1.3c is limestone, which is relatively easily deformed when heated. The dark rock is chert, which remains brittle. As the limestone stretched (parallel to the hammer handle) the brittle chert was forced to break into fragments to accommodate the change in shape of the body of rock. A fault is a rock boundary along which the rocks on either side have been displaced relative to each other (Figure 12.1.3d).

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12.2: Folding

When a body of rock, especially sedimentary rock, is squeezed from the sides by tectonic forces, it is likely to fracture and/or become faulted if it is cold and brittle, or become folded if it is warm enough to behave in a plastic manner.

The nomenclature and geometry of folds are summarized on Figure 12.2.1. An upward fold is called an **anticline** (or, more accurately, an antiform if we don't know if the beds have been overturned or not), while a downward fold is called a **syncline**, (or a synform if we don't if the beds have been overturned). In many areas it's common to find a series of antiforms and synforms (as in Figure 12.5), although some sequences of rocks are folded into a single antiform or synform. A plane drawn through the crest of a fold in a series of beds is called the **axial plane** of the fold. The sloping beds on either side of an axial plane are **limbs**. An antiform or synform is described as **symmetrical** if the angles between each of limb and the axial plane are generally similar, and **asymmetrical** if they are not. If the axial plane is sufficiently tilted that the beds on one side have been tilted past vertical, the fold is known as an **overturned** antiform or synform.

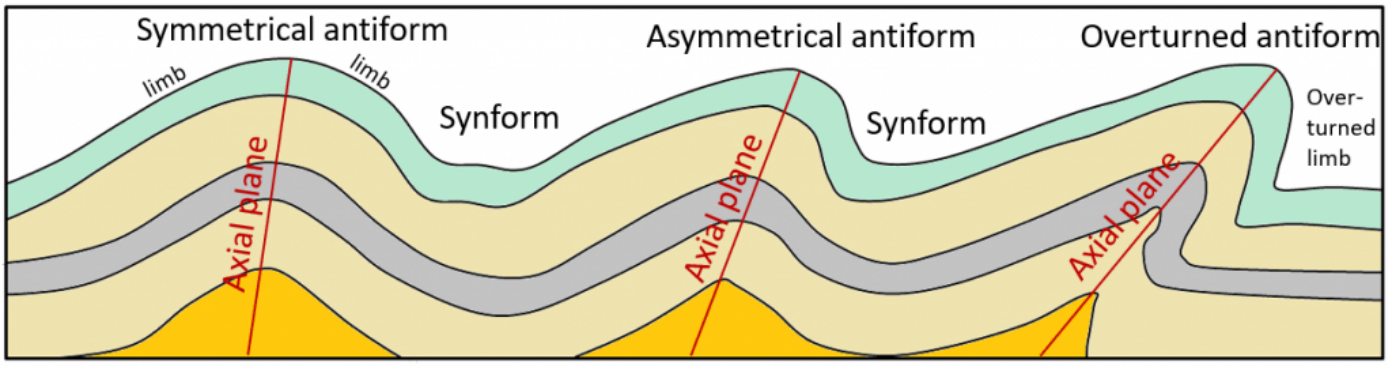


Figure 12.2.1 Examples of different types of folds and fold nomenclature. Axial planes are only shown for the antiforms, but synforms also have axial planes.

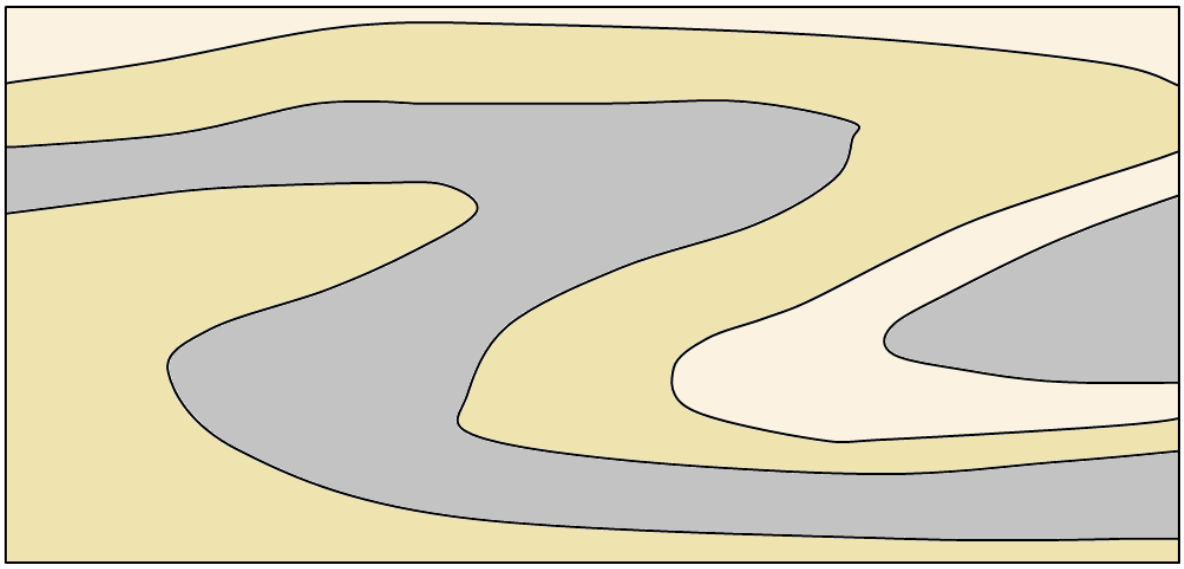


Figure 12.2.2 An isoclinal recumbent fold.

A very tight fold, in which the limbs are parallel or nearly parallel to one another is called an **isoclinal fold** (Figure 12.2.2). Isoclinal folds that have been overturned to the extent that their limbs are nearly horizontal are called **recumbent folds**.

Folds can be of any size, and it's very common to have smaller folds within larger folds (Figure 12.2.3). Large folds can have wavelengths of tens of kilometers, and very small ones might be visible only under a microscope.



Figure 12.2.3 Folded limestone (grey) and chert (rust-colored) in Triassic Quatsino Formation rocks on Quadra Island, B.C. The image is about 1 metre across.

Antiforms are not necessarily, or even typically, expressed as ridges in the terrain, nor synforms as valleys. Folded rocks get eroded just like all other rocks and the topography that results is typically controlled mostly by the resistance of different layers to erosion (Figure 12.2.4).

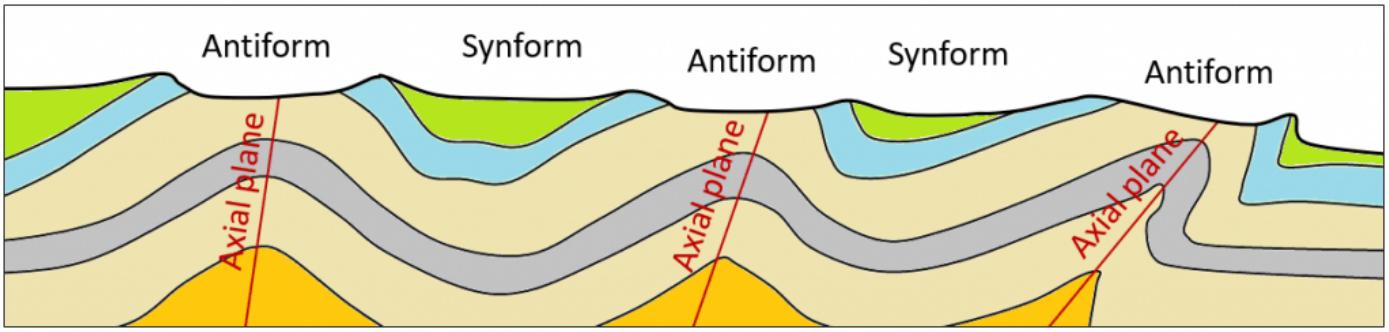


Figure 12.2.4 Example of the topography in an area of folded rocks that has been eroded. In this case the blue and green rocks are most resistant to erosion, and are represented by hills.

Exercise 12.1 Folding style

Figure 12.2.5 shows folding in the same area of the Rocky Mountains as Figure 12.2.1. Describe the types of folds using the appropriate terms from above (symmetrical, asymmetrical, isoclinal, overturned, recumbent etc.). You might find it useful to first sketch in the axial planes.



Figure 12.2.5

See Appendix 3 for Exercise 12.1 answers.

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12.3: Fracturing and Faulting

A body of rock that is brittle—either because it is cold or because of its composition, or both— is likely to break rather than fold when subjected to stress, and the result is fracturing or faulting.

Fracturing

Fracturing is common in rocks near the surface, either in volcanic rocks that have shrunk on cooling (Figure 12.3.3a), or in other rocks that have been exposed by erosion and have expanded (Figure 12.3.1).



Figure 12.3.1 Granite in the Coquihalla Creek area, B.C. (left) and sandstone at Nanoose, B.C. (right), both showing fracturing that has resulted from expansion due to removal of overlying rock.

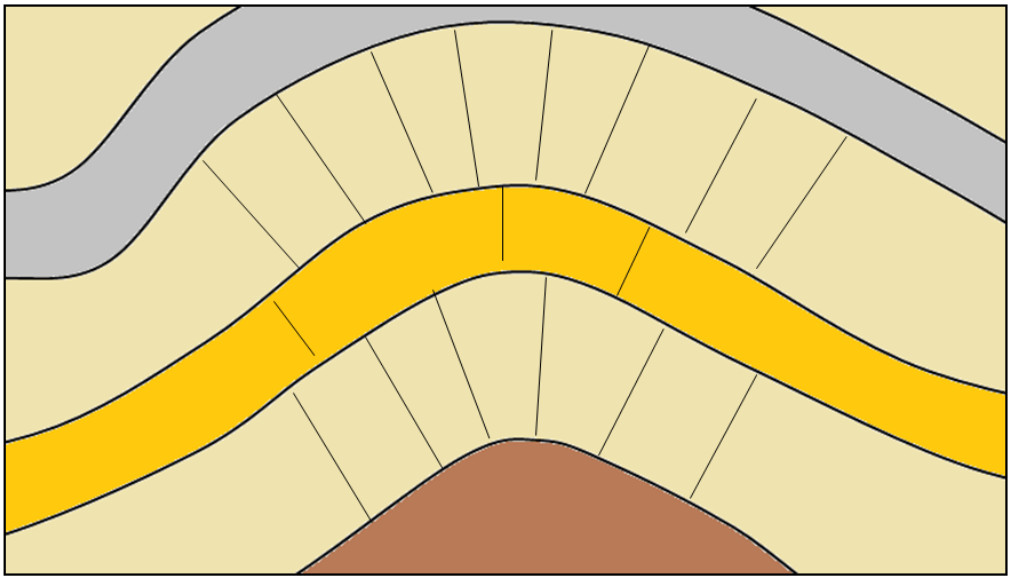


Figure 12.3.2 A depiction of joints developed in the hinge area of folded rocks. Note that in this situation some rock types are more likely to fracture than others.

A fracture in a rock is also called a **joint**. There is no side-to-side movement of the rock on either side of a joint. Most joints form where a body of rock is expanding because of reduced pressure, as shown by the two examples in Figure 12.3.1, or where the rock itself is contracting but the body of rock remains the same size (the cooling volcanic rock in Figure 12.3.3a). In all of these cases, the pressure regime is one of *tension* as opposed to *compression*. Joints can also develop where rock is being folded because,

while folding typically happens during compression, there may be some parts of the fold that are in tension (Figure 12.3.2).

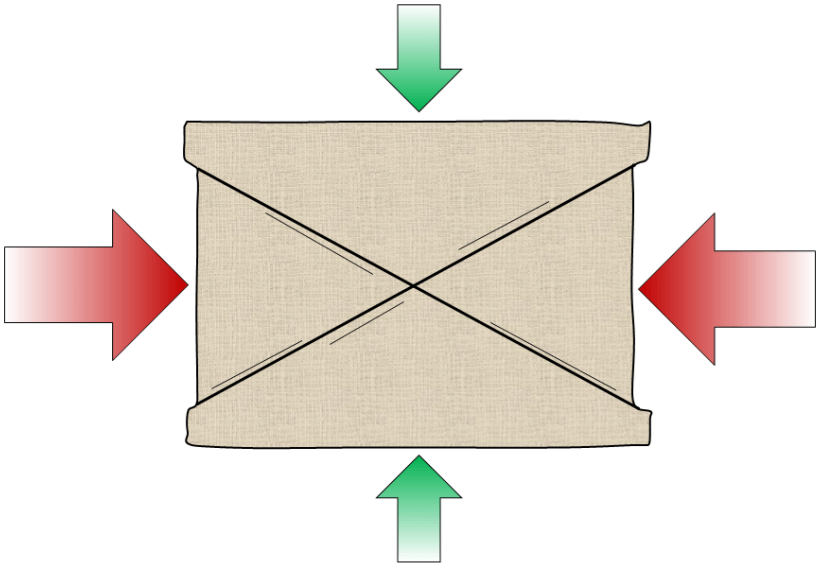


Figure 12.3.3 A depiction of joints developed in a rock that is under stress.

Finally joints can also develop when rock is under compression as shown on Figure 12.3.3, where there is differential stress on the rock, and joint sets develop at angles to the compression directions.

Faulting

A fault is a boundary between two bodies of rock along which there has been relative motion (Figure 12.3.3d). As we discussed in Chapter 11, an earthquake involves the sliding of one body of rock past another. Earthquakes don't necessarily happen on existing faults, but once an earthquake takes place a fault will exist in the rock at that location. Some large faults, like the San Andreas Fault in California or the Tintina Fault, which extends from northern B.C. through central Yukon and into Alaska, show evidence of hundreds of kilometers of motion, while others show less than a millimetre. In order to estimate the amount of motion on a fault, we need to find some geological feature that shows up on both sides and has been offset (Figure 12.3.4).

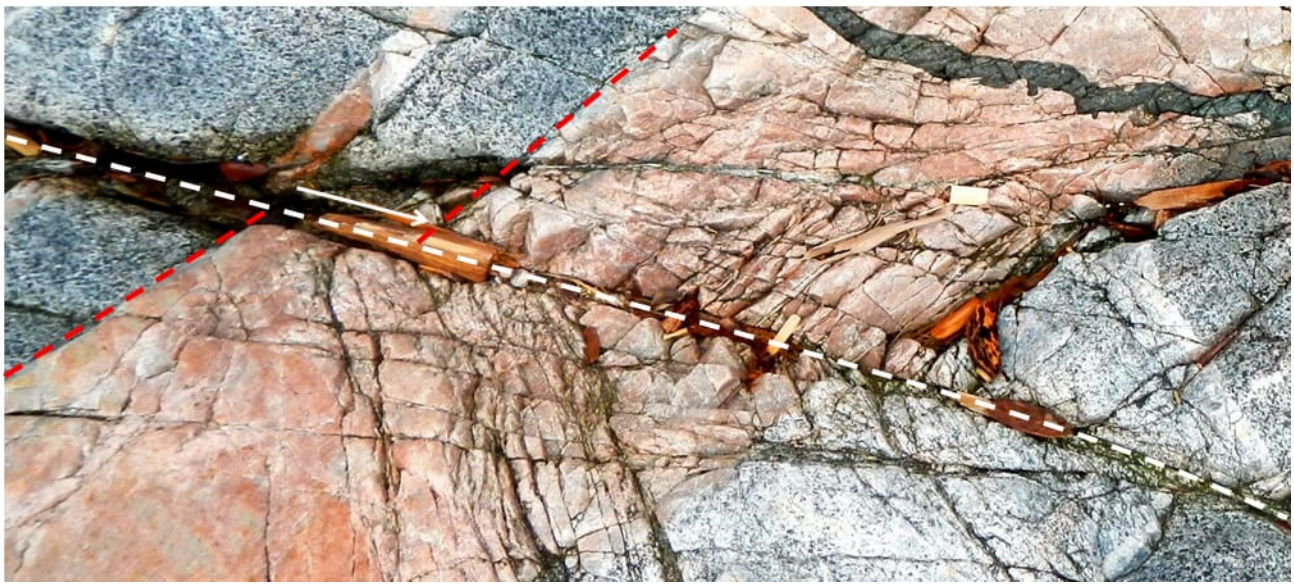


Figure 12.3.4 A fault (white dashed line) in intrusive rocks on Quadra Island, B.C. The pink dyke has been offset by the fault and the extent of the offset is shown by the white arrow (approximately 10 centimeters). Because the far side of the fault has moved to the right, this is a right-lateral fault. If the photo had been taken from the other side, the fault would still appear to have a right-lateral offset.

There are several kinds of faults, as illustrated on Figure 12.3.5, and they develop under different stress conditions. The terms *hanging wall* and *footwall* in the diagrams apply to situations where the fault is not vertical. The body of rock above the fault is called the **hanging wall**, and the body of rock below it is called the **footwall**. If the fault develops in a situation of compression, then it will be a **reverse fault** because the compression causes the hanging wall to be pushed up relative to the footwall. If the fault develops in a situation of extension, then it will be a **normal fault**, because the extension allows the hanging wall to slide down relative to the footwall in response to gravity.

The third situation is where the bodies of rock are sliding sideways with respect to each other, as is the case along a transform fault (see Chapter 10). This is known as a **strike-slip fault** because the displacement is along the "strike" or the length of the fault. On strike-slip faults the motion is typically only horizontal, or with a very small vertical component, and as discussed above the sense of motion can be right lateral (the far side moves to the right), as in Figures 12.12 and 12.13, or it can be left lateral (the far side moves to the left). Transform faults are strike-slip faults.

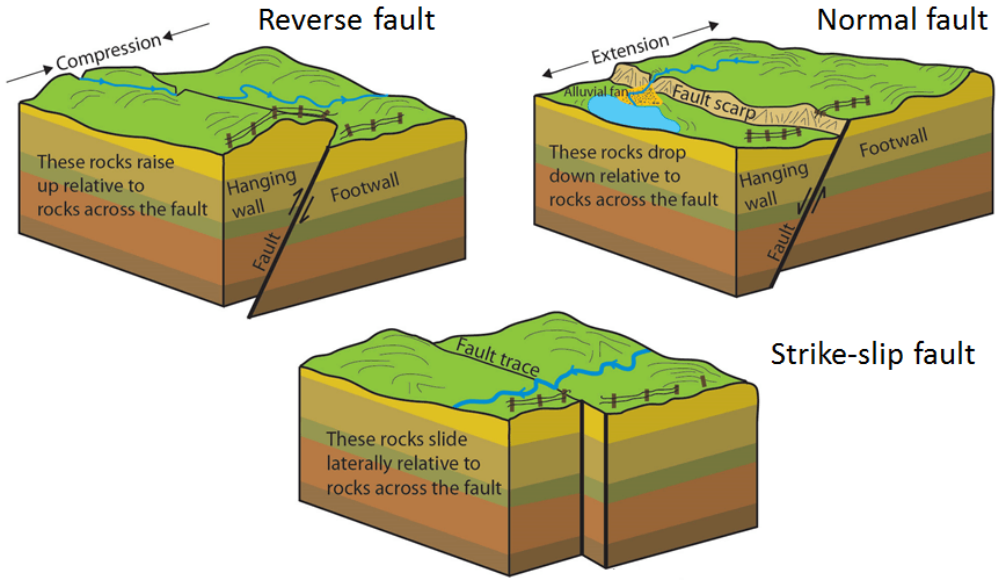


Figure 12.3.5 Depiction of reverse, normal, and strike-slip faults. Reverse faults happen during compression while normal faults happen during extension. Most strike-slip faults are related to transform boundaries.

In areas that are characterized by extensional tectonics, it is not uncommon for a part of the upper crust to subside with respect to neighbouring parts. This is typical along areas of continental rifting, such as the Great Rift Valley of East Africa or in parts of Iceland, but it is also seen elsewhere. In such situations a down-dropped block is known as a **graben** (German for ditch), while an adjacent block that doesn't subside is called a **horst** (German for heap) (Figure 12.3.6). There are many horsts and grabens in the Basin and Range area of the western United States, especially in Nevada. Part of the Fraser Valley region of B.C., in the area around Sumas Prairie is a graben.

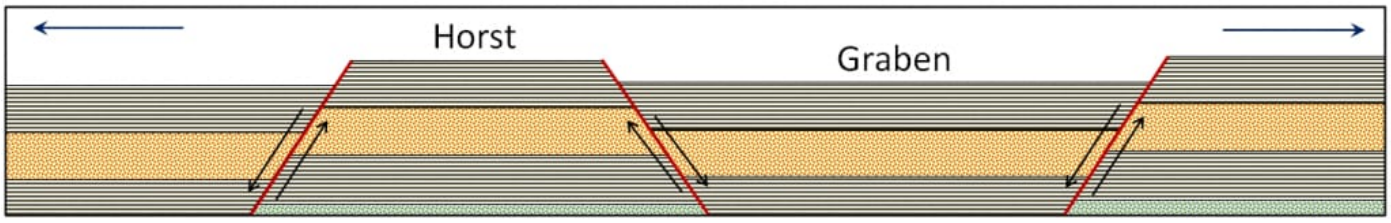


Figure 12.3.6 Depiction of graben and horst structures that form in extensional situations. All of the faults are normal faults.

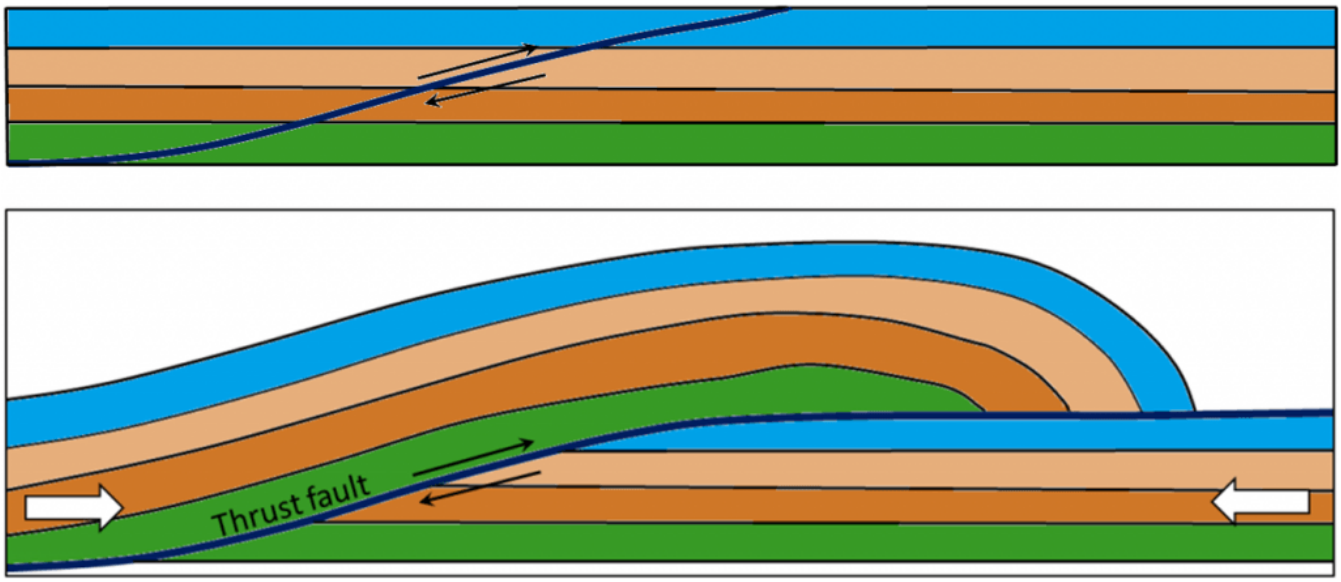


Figure 12.3.7 Depiction a thrust fault. Top: prior to faulting. Bottom: after significant fault offset.

A special type of reverse fault, with a very low-angle fault plane, is known as a **thrust fault**. Thrust faults are relatively common in areas where fold-belt mountains have been created during continent-continent collision. Some represent tens of kilometers of thrusting, where thick sheets of sedimentary rock have been pushed up and over top of other rock (Figure 12.3.7).

There are numerous thrust faults in the Rocky Mountains, and a well-known example is the McConnell Thrust, along which a sequence of sedimentary rocks about 800 meters thick has been pushed for about 40 kilometers from west to east (Figure 12.3.8). The thrusted rocks range in age from Cambrian to Cretaceous, so in the area around Mt. Yamnuska Cambrian-aged rock (around 500 Ma) has been thrust over, and now lies on top of Cretaceous-aged rock (around 75 Ma) (Figure 12.3.9).

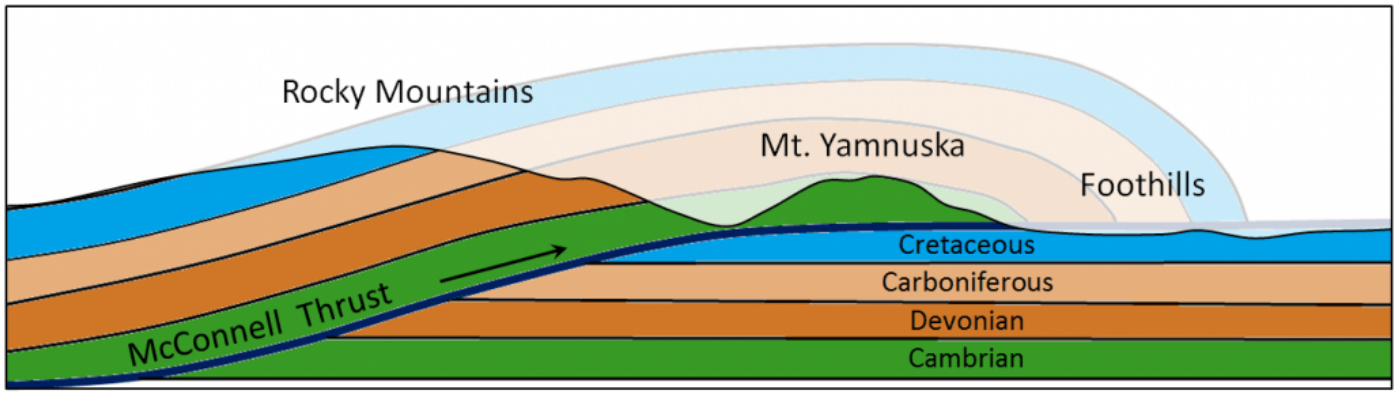


Figure 12.3.8 Depiction of the McConnell Thrust in the eastern part of the Rocky Mountinas. The rock within the faded area has been eroded

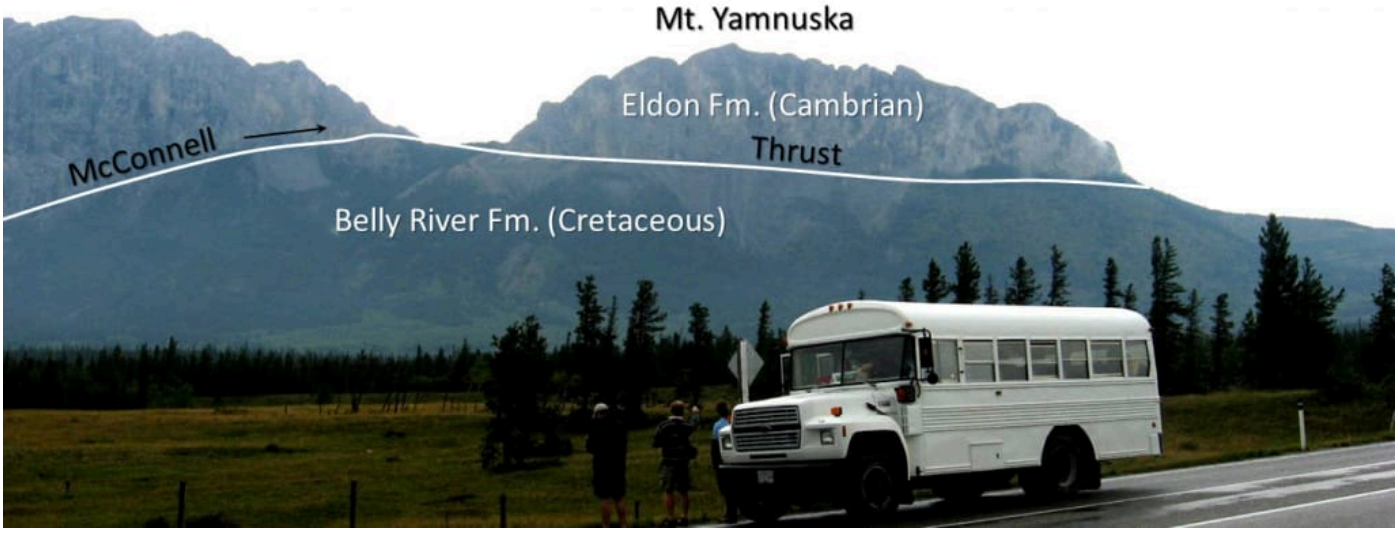


Figure 12.3.9 The McConnell Thrust at Mt. Yamnuska near Exshaw, Alberta. Carbonate rocks (limestone) of Cambrian age have been thrust over top of Cretaceous mudstone.

Exercise 12.2 Types of faults



Figure 12.3.10

The four images are faults that formed in different tectonic settings. Identifying the type of fault allows us to determine if the body of rock was under compression or extension at the time of faulting. Complete the table below the images, identifying the types of faults (normal or reversed) and whether each one formed under compression or extension.

Type of Fault and Tectonic Situation
Top left:
Bottom left:
Top right:
Bottom right:

See Appendix 3 for Exercise 12.2 answers.

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12.4: Measuring Geological Structures

Geologists take great pains to measure and record geological structures because they are critically important to understanding the geological history of a region. One of the key features to measure is the orientation, or **attitude**, of bedding. We know that sedimentary beds are deposited in horizontal layers, so if the layers are no longer horizontal, then we can infer that they have been affected by tectonic forces and have become either tilted, or folded. We can express the orientation of a bed (or any other planar feature) with two values: first, the compass orientation of a horizontal line on the surface—the **strike**—and second, the angle at which the surface dips below a horizontal plane, (perpendicular to the strike)—the **dip** (Figure 12.4.1).

It may help to imagine a vertical surface, such as a wall in your house. The strike is the compass orientation of the wall and the dip is 90° from horizontal. If you could push the wall so it's leaning over, but still attached to the floor, the strike direction would be the same, but the dip angle would be less than 90°. If you pushed the wall over completely so it was lying on the floor, it would no longer have a strike direction and its dip would be 0°. When describing the dip it is important to include the direction. In other words. if the strike is 0° (i.e., north) and the dip is 30°, it would be necessary to say "to the west" or "to the east." Similarly if the strike is 45° (i.e., northeast) and the dip is 60°, it would be necessary to say "to the northwest" or "to the southeast."

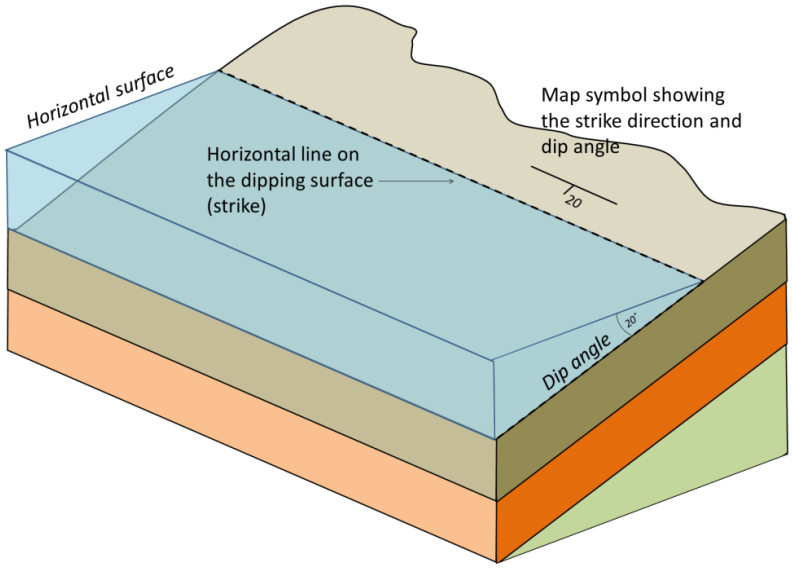


Figure 12.4.1 A depiction of the strike and dip of some tilted sedimentary beds. The dipping beds are shown partially covered with water so that you can visualize a horizontal line on the rock surface. The notation for expressing strike and dip on a map is also shown.

Measurement of geological features is done with a special compass that has a built-in clinometer—a device for measuring vertical angles. An example of how this is done is shown on Figure 12.4.2.



Figure 12.4.2 Measuring the compass direction of the strike (left) and the vertical angle of the dip (right) using a compass with a clinometer.

Strike and dip are also used to describe any other planar features, including joints, faults, dykes, sills, and even the foliation planes in metamorphic rocks. Figure 12.4.3 shows an example of how we would depict the beds that make up an anticline on a map.

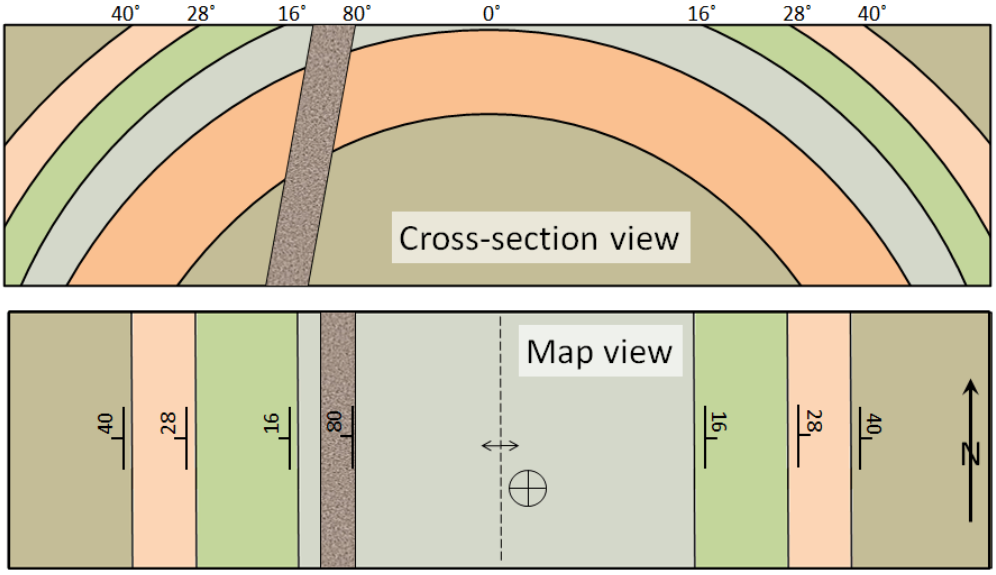


Figure 12.4.3 A depiction of an antiform and a dyke in cross-section (looking from the side) and in map view (a.k.a. plan view) with the appropriate strike-dip and antiform symbols.

The beds on the west (left) side of the map are dipping at various angles to the west. The beds on the east side are dipping to the east. The middle bed (light grey) is horizontal; this is denoted by a cross within a circle. The dyke is dipping at 80° to the west. The hinge of the fold is denoted with a dashed line with two arrows that point away from it. If it was a synform, the arrows would point towards the line.

Exercise 12.3 Putting strike and dip on a map

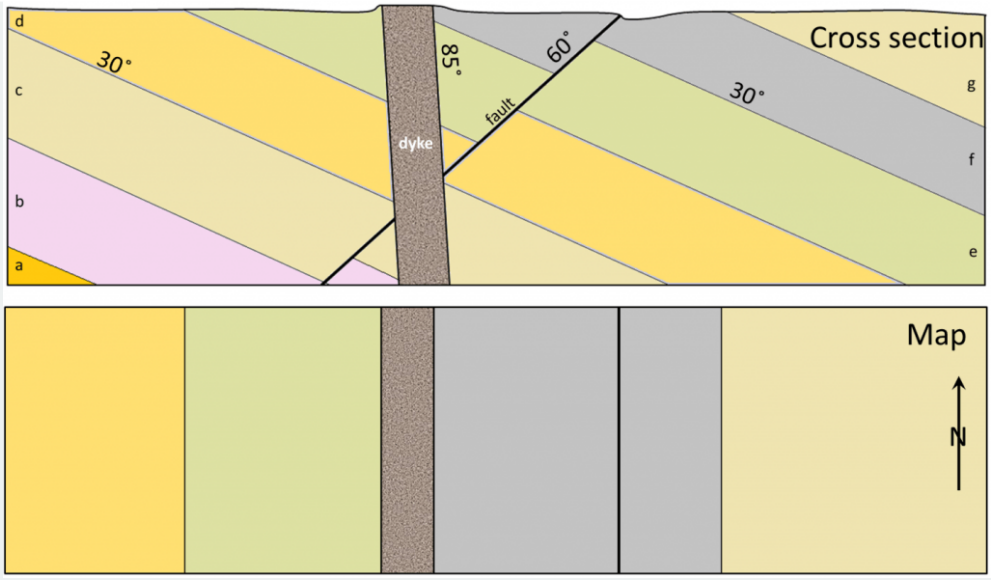


Figure 12.4.4

This cross-section shows seven tilted sedimentary layers (a to g), a fault, and a steeply dipping dyke. Place strike and dip symbols on the map to indicate the orientations of the beds shown, the fault, and the dyke. Then answer the questions.

1. What type of fault is this, and is this an extensional or compressional situation?
2. What are the relative ages of the nine geological features shown here (seven beds, dyke, and fault)? Which are the youngest and oldest?

See Appendix 3 for Exercise 12.3 answers.

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12.5: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
12.1 Stress and Strain	Stress within rocks—which includes compression, extension and shearing—typically originates from plate-boundary processes. Rock that is stressed responds with either elastic or plastic strain, and may eventually break. The way a rock responds to stress depends on its composition and structure, the rate at which strain is applied, and also to the temperature of the rock body and the presence of water.
12.2 Folding	Folding is generally a plastic response to compressive stress, although some brittle behaviour can happen during folding. An upward fold is an antiform. A downward fold is a synform. The axis of a fold can be vertical, inclined, or even horizontal. If we know that the folded beds have not been overturned, then we can use the more specific terms: anticline and syncline.
12.3 Fracturing and Faulting	Fractures (joints) typically form during extension, but can also form during compression. Faulting, which involves the displacement of rock, can take place during compression or extension, as well as during shearing at transform boundaries. Thrust faulting is a special form of reverse faulting.
12.4 Measuring Geological Structures	It is important to be able to measure the strike and dip of planar surfaces, such as a bedding planes, fractures or faults. Special symbols are used to show the orientation of structural features on geological maps.

Questions for Reviews

1. What types of plate boundaries are most likely to contribute to the following?: a) compression, b) extension, and c) shearing.
2. Explain the difference between elastic strain and plastic strain.
3. List some of the factors that influence whether a rock will deform (in either an elastic or plastic manner) or break when placed under stress.
4. Label the types of folds in this diagram, and label any of the important features of the folds.

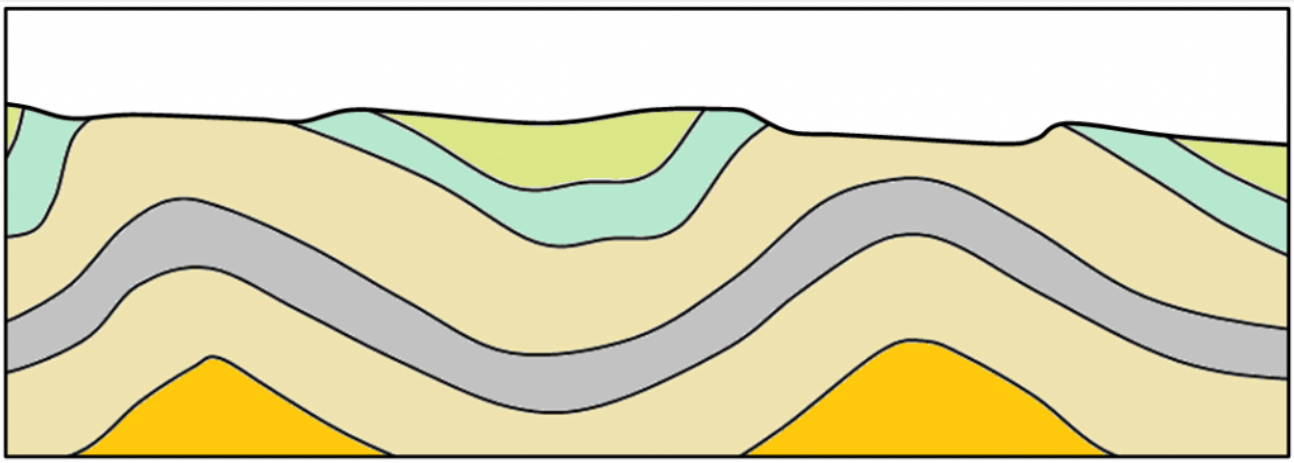


Figure A

- Explain why fractures are common in volcanic rocks.
- What is the difference between a normal fault and a reverse fault, and under what circumstances would you expect these to form?
- What type of fault would you expect to see near to a transform plate boundary?

- i) Describe in words the *general* attitude (strike and dip) of these beds.

  ii) Which of these beds is the oldest?

  iii) What is “a” and what is its attitude?

  iv) What is “b” and what is its attitude?

  v) Which of these terms applies to “b”: “left lateral” or “right lateral”?

  Answers to Review Questions can be found in Appendix 2.

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13: STREAMS AND FLOODS

13.1: The Hydrological Cycle
13.2: Drainage Basins
13.3: Stream Erosion and Deposition
13.4: Stream Types
13.5: Flooding
13.6: Summary

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13.1: The Hydrological Cycle

Water is constantly on the move. It is evaporated from the oceans, lakes, streams, the surface of the land, and plants (transpiration) by solar energy (Figure 13.1.1). It is moved through the atmosphere by winds and condenses to form clouds of water droplets or ice crystals. In response to the pull of gravity it comes back down as rain or snow and then flows through streams, into lakes, and eventually back to the oceans. Water on the surface and in streams and lakes infiltrates the ground to become groundwater. Groundwater slowly moves through the rock and surficial materials. Some groundwater returns to other streams and lakes, and some goes directly back to the oceans.

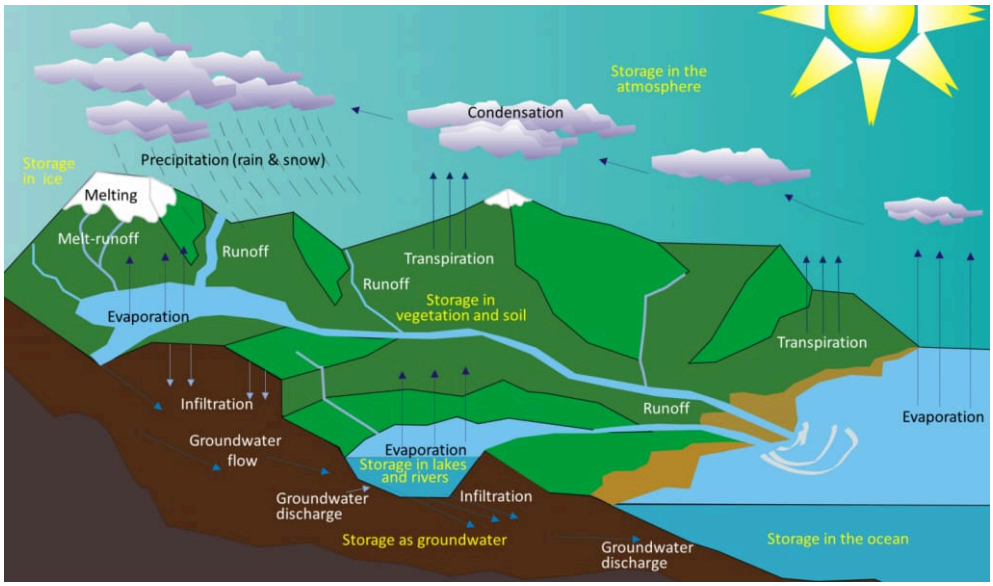


Figure 13.1.1 The various components of the water cycle. Black or white text indicates the movement or transfer of water from one reservoir to another. Yellow text indicates the storage of water.



Figure 13.1.2 Representation of the Earth's water. The 1 litre jug is filled with salty sea water (97%). The ice-cube is glacial ice (2%). The 2 teaspoons represent groundwater (1%), and the three drops represent all of the fresh water in lakes, streams, and wetlands, plus all of the water in the atmosphere.

Even while it's moving around, water is stored in various reservoirs. The largest, by far, is the oceans, accounting for 97% of the volume. Of course, that water is salty. The remaining 3% is fresh water. Two-thirds of our fresh water is stored in ice and one-third is stored in the ground. The remaining fresh water—about 0.03% of the total—is stored in lakes, streams, vegetation, and the

atmosphere. To put that in perspective, let's imagine putting all of Earth's water into a 1 litre jug (Figure 13.1.2). We start by almost filling the jug with 970 millilitres of water and 34 grams of salt. Then we add one regular-sized (roughly 20 millilitres) ice cube (representing glacial ice) and two teaspoons (roughly 10 millilitres) of groundwater. All of the water that we see around us in lakes and streams and up in the sky can be represented by adding three more drops from an eyedropper.

Although the proportion of Earth's water that is in the atmosphere is tiny, the actual volume is huge. At any given time, there is the equivalent of approximately 13,000 cubic kilometers (km^3) of water in the air in the form of water vapor and water droplets in clouds. Water is evaporated from the oceans, vegetation, and lakes at a rate of 1,580 km^3 per day, and just about exactly the same volume falls as rain and snow every day—over both the oceans and land. The precipitation that falls on land goes back to the ocean in the form of stream flow (117 km^3/day) and groundwater flow (6 km^3/day). Most of the rest of this chapter is about that 117 km^3/day of streamflow. The average discharge of the Fraser River into the ocean is approximately 0.31 km^3/day, or 0.26% of the total flow of all rivers.

Exercise 13.1 How long does water stay in the atmosphere and oceans?

The residence time of a water molecule in the atmosphere (or any of the other reservoirs) can be estimated by dividing the amount that is there by the rate at which it is transferred in and out. For the atmosphere, we know that the reservoir size is 13,000 km^3, and the rate of flux is 1,580 km^3 per day. That's the amount that goes into or comes out of the atmosphere in a day. If we divide 13,000 by 1,580, we get 8.22 days. This means that, *on average*, a molecule of water stays in the atmosphere for just over eight days. "Average" needs to be emphasized here because obviously some molecules stay in the air for only a few hours, while others may stay up there for weeks.

The volume of the oceans is 1,338,000,000 km^3 and the flux rate is approximately the same (1,580 km^3 per day). What is the average residence time of a water molecule in the ocean?

See Appendix 3 for Exercise 13.1 answers.

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13.2: Drainage Basins

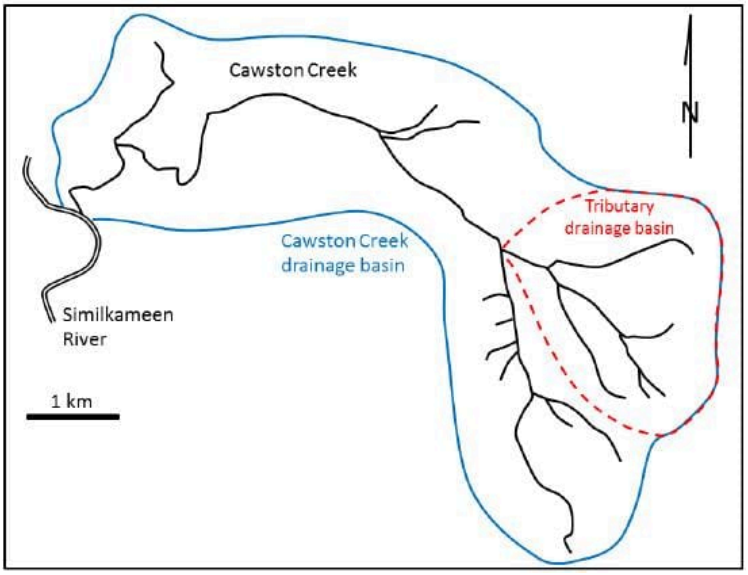


Figure 13.2.1 Cawston Creek near Keremeos, B.C. The blue line shows the extent of the drainage basin. The dashed red line is the drainage basin of one of its tributaries.

A **stream** is a body of flowing surface water of any size, ranging from a tiny trickle to a mighty river. The area from which the water flows to form a stream is known as its **drainage basin**. All of the precipitation (rain or snow) that falls within a drainage basin eventually flows into its stream, unless some of that water is able to cross into an adjacent drainage basin via groundwater flow. An example of a drainage basin is shown in Figure 13.2.1.

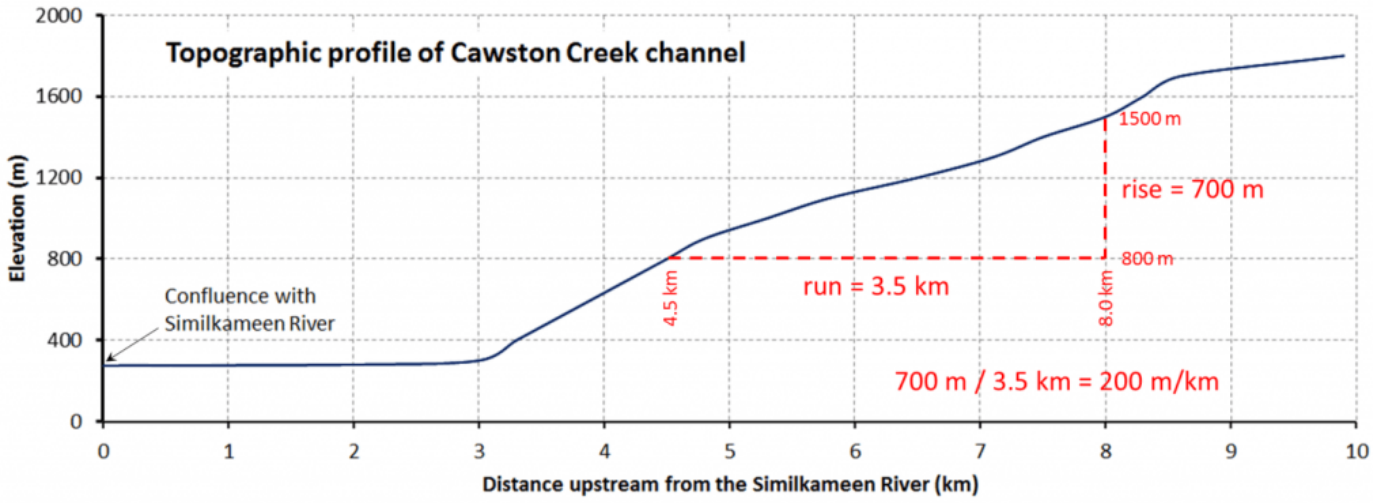


Figure 13.2.2 Profile of the main stem of Cawston Creek near Keremeos, B.C. The maximum elevation of the drainage basin is about 1,840 meters, near Mount Kobau. The base level is 275 meters, at the Similkameen River. As shown, the gradient of the stream can be determined by dividing the change in elevation between any two points (rise) by the distance between those two points (run).

Cawston Creek is a typical small drainage basin (approximately 25 square kilometers) within a very steep glaciated valley. As shown in Figure 13.2.2, the upper and middle parts of the creek have steep **gradients** (averaging about 200 meters per kilometre but ranging from 100 to 350 meters per kilometre), and the lower part, within the valley of the Similkameen River, is relatively flat (less than 5 meters per kilometre). The shape of the valley has been controlled first by tectonic uplift (related to plate convergence), then by pre-glacial stream erosion and mass wasting, then by several episodes of glacial erosion, and finally by post-glacial stream erosion. The lowest elevation of Cawston Creek (275 meters at the Similkameen River) is its **base level**.

Cawston Creek cannot erode below that level unless the Similkameen River erodes deeper into its flood plain (the area that is inundated during a flood).

Metro Vancouver's water supply comes from three large drainage basins on the north shore of Burrard Inlet, as shown in Figure 13.2.3. This map illustrates the concept of a drainage basin divide. The boundary between two drainage basins is the height of land between them. A drop of water falling on the boundary between the Capilano and Seymour drainage basins (a.k.a., watersheds), for example, could flow into either one of them.



Figure 13.2.3 The three drainage basins that are used for the Metro Vancouver water supply.

The pattern of tributaries within a drainage basin depends largely on the type of rock beneath, and on structures within that rock (folds, fractures, faults, etc.). The three main types of drainage patterns are illustrated in Figure 13.2.4. **Dendritic** patterns, which are by far the most common, develop in areas where the rock (or unconsolidated material) beneath the stream has no particular fabric or structure and can be eroded equally easily in all directions. Examples would be granite, gneiss, volcanic rock, and sedimentary rock that has not been folded. Most areas of British Columbia have dendritic patterns, as do most areas of the prairies and the Canadian Shield. **Trellis** drainage patterns typically develop where sedimentary rocks have been folded or tilted and then eroded to varying degrees depending on their strength. The Rocky Mountains of B.C. and Alberta are a good example of this, and many of the drainage systems within the Rockies have trellis patterns. **Rectangular** patterns develop in areas that have very little topography and a system of bedding planes, fractures, or faults that form a rectangular network. Rectangular drainage patterns are rare in Canada.

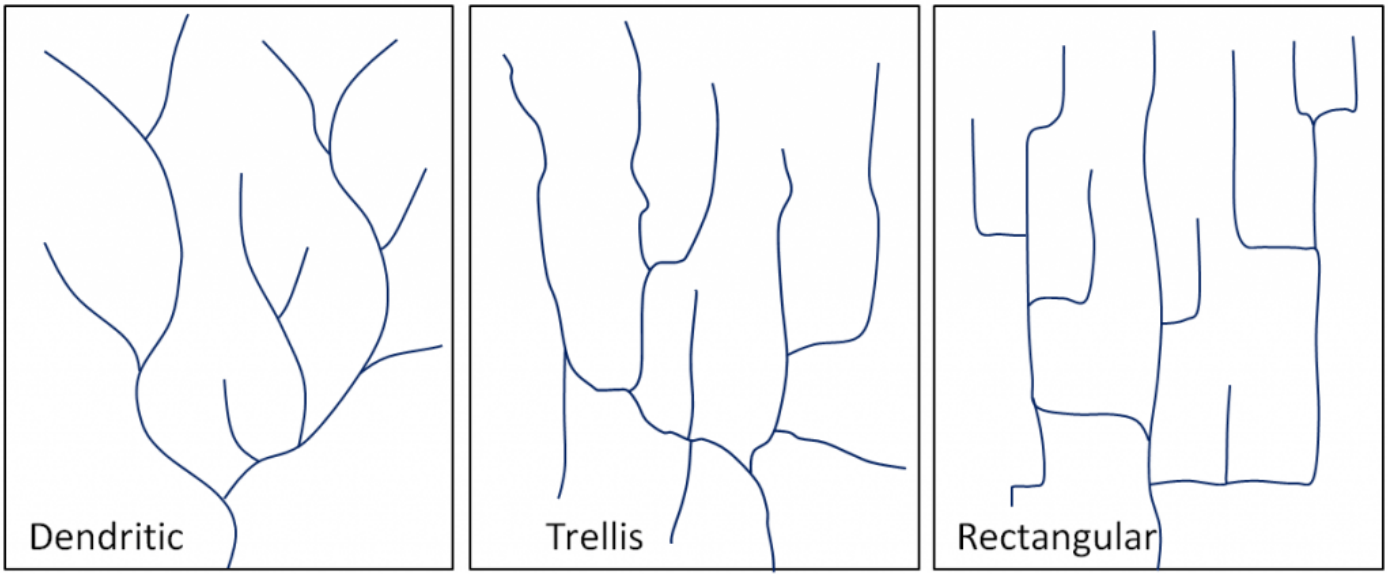


Figure 13.2.4 Typical dendritic, trellis, and rectangular stream drainage patterns.

In many parts of Canada, especially relatively flat areas with thick glacial sediments, and throughout much of Canadian Shield in eastern and central Canada, drainage patterns are chaotic, or what is known as **deranged** (Figure 13.2.5, left). Lakes and wetlands are common in this type of environment.

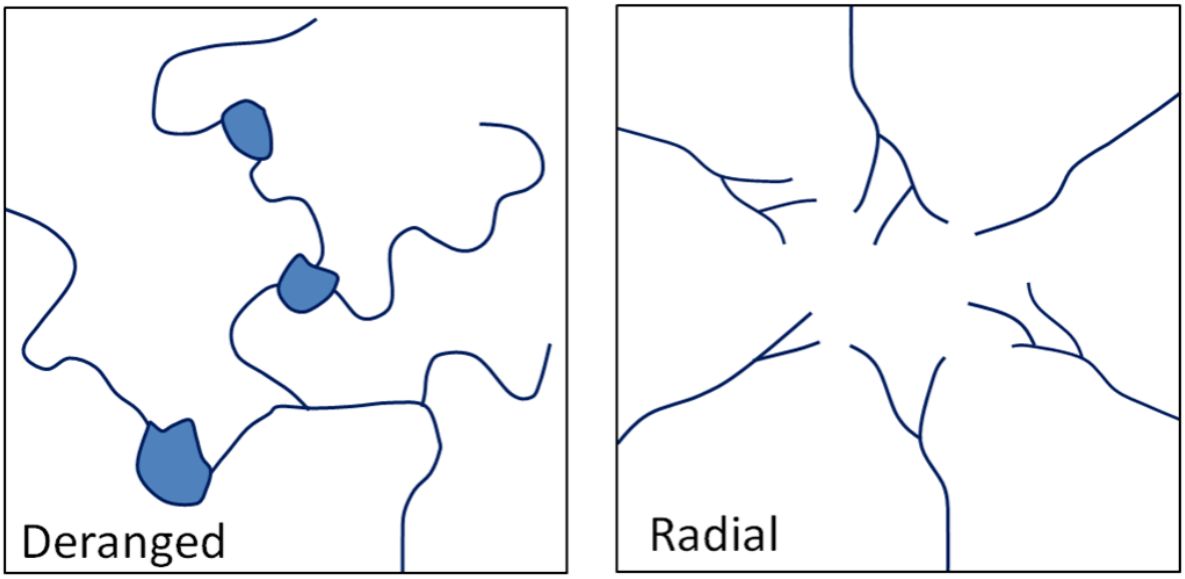


Figure 13.2.5 Left: a typical deranged pattern; right: a typical radial drainage pattern developed around a mountain or hill.

A fourth type of drainage pattern, which is not specific to a drainage basin, is known as **radial** (Figure 13.2.5, right). Radial patterns form around isolated mountains (such as volcanoes) or hills, and the individual streams typically have dendritic drainage patterns.

Over geological time, a stream will erode its drainage basin into a smooth profile similar to that shown in Figure 13.2.6. If we compare this with an ungraded stream like Cawston Creek (Figure 13.2.2), we can see that graded streams are steepest in their headwaters and their gradient gradually decreases toward their mouths. Ungraded streams have steep sections at various points, and typically have rapids and waterfalls at numerous locations along their lengths.

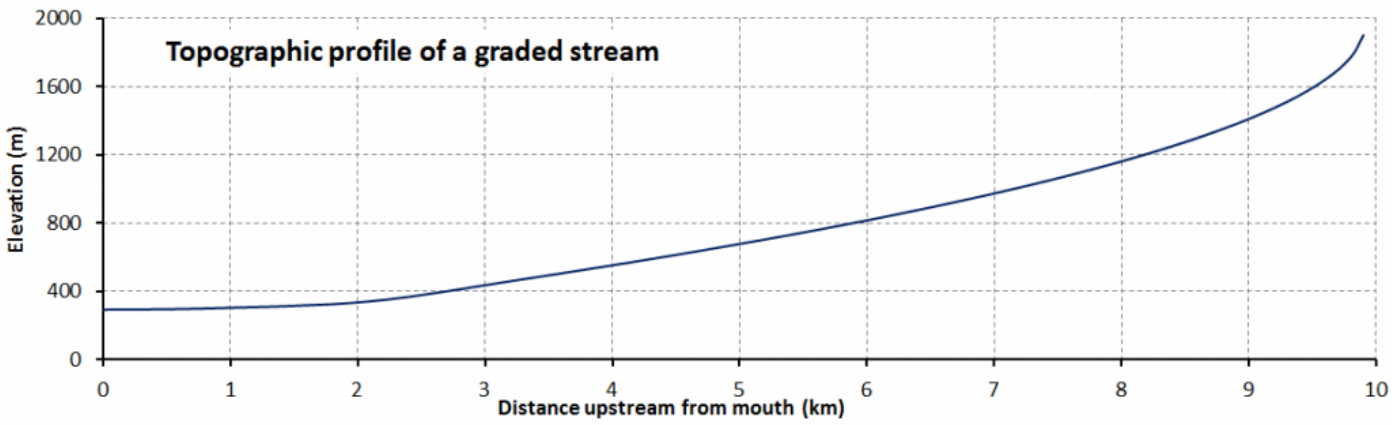


Figure 13.2.6 The topographic profile of a typical graded stream.

A graded stream can become ungraded if there is renewed tectonic uplift, or if there is a change in the base level, either because of tectonic uplift or some other reason. As stated earlier, the base level of Cawston Creek is defined by the level of the Similkameen River, but this can change, and has done so in the past. Figure 13.2.7 shows the valley of the Similkameen River in the Keremeos area. The river channel is just beyond the row of trees. The green field in the distance is underlain by material eroded from the hills behind and deposited by a small creek (not Cawston Creek) adjacent to the Similkameen River when its level was higher than it is now. Sometime in the past several centuries, the Similkameen River eroded down through these deposits (forming the steep bank on the other side of the river), and the base level of the small creek was lowered by about 10 meters. Over the next few centuries, this creek will seek to become graded again by eroding down through its own alluvial fan.

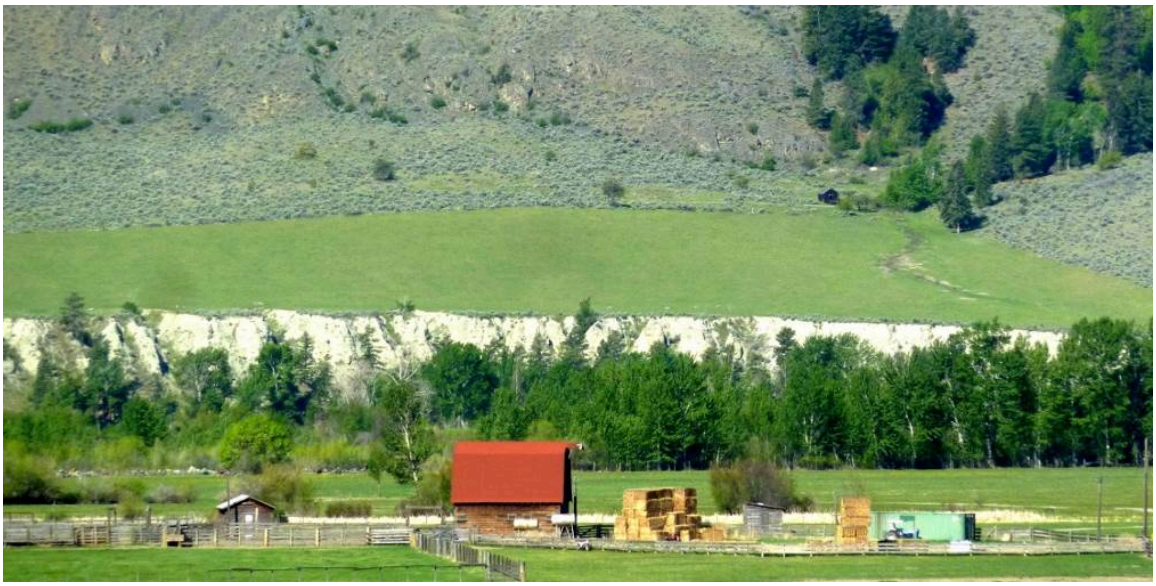


Figure 13.2.7 An example of a change in the base level of a small stream that flows into the Similkameen river near Keremeos. The previous base level was near the top of the sandy bank. The current base level is the river (hidden behind the trees) which has eroded down into the sandy deposit.

Another example of a change in base level can be seen along the Juan de Fuca Trail on southwestern Vancouver Island. As shown in Figure 13.2.8, many of the small streams along this part of the coast flow into the ocean as waterfalls. It is evident that the land in this area has risen by about 5 meters in the past few thousand years, probably in response to deglaciation. The streams that used to flow directly into the ocean now have a lot of down-cutting to do to become regraded.



Figure 13.2.8 Two streams with a lowered base level on the Juan de Fuca Trail, southwestern Vancouver Island.

The ocean is the ultimate base level, but lakes and other rivers act as base levels for many smaller streams. We can create an artificial base level on a stream by constructing a dam, as illustrated in Exercise 13.2.

Exercise 13.2 The effect of a dam on base level



Figure 13.2.9 Revelstoke Dam and Revelstoke Lake on the Columbia River at Revelstoke, BC.

When a dam is built on a stream, a reservoir (artificial lake) forms behind the dam. This temporarily (for many decades at least) creates a new base level for the part of the stream above the reservoir.

How does the formation of a reservoir affect the stream where it enters the reservoir, and what happens to the sediment it was carrying?

The water leaving the dam has no sediment in it. How does this affect the stream below the dam?

See Appendix 3 for Exercise 13.2 answers.

Sediments accumulate within the flood plain of a stream, and then, if the base level falls, or if there is less sediment to deposit, the stream may cut down through those existing sediments to form terraces. A terrace on the Similkameen River is shown in Figure 13.2.7 and some on the Fraser River are shown in Figure 13.2.10. The Fraser River photo shows at least two levels of terraces.



Figure 13.2.10 Terraces on the Fraser River at High Bar.

In the late 19th century, American geologist William Davis proposed that streams and the surrounding terrain develop in a cycle of erosion (Figure 13.2.11). Following tectonic uplift, streams erode quickly, developing deep V-shaped valleys that tend to follow relatively straight paths. Gradients are high, and profiles are ungraded. Rapids and waterfalls are common. During the mature stage, streams erode wider valleys and start to deposit thick sediment layers. Gradients are slowly reduced and grading increases. In old age, streams are surrounded by rolling hills, and they occupy wide sediment-filled valleys. Meandering patterns are common.

Davis's work was done long before the idea of plate tectonics, and he was not familiar with the impacts of glacial erosion on streams and their environments. While some parts of his theory are out of date, it is still a useful way to understand streams and their evolution.

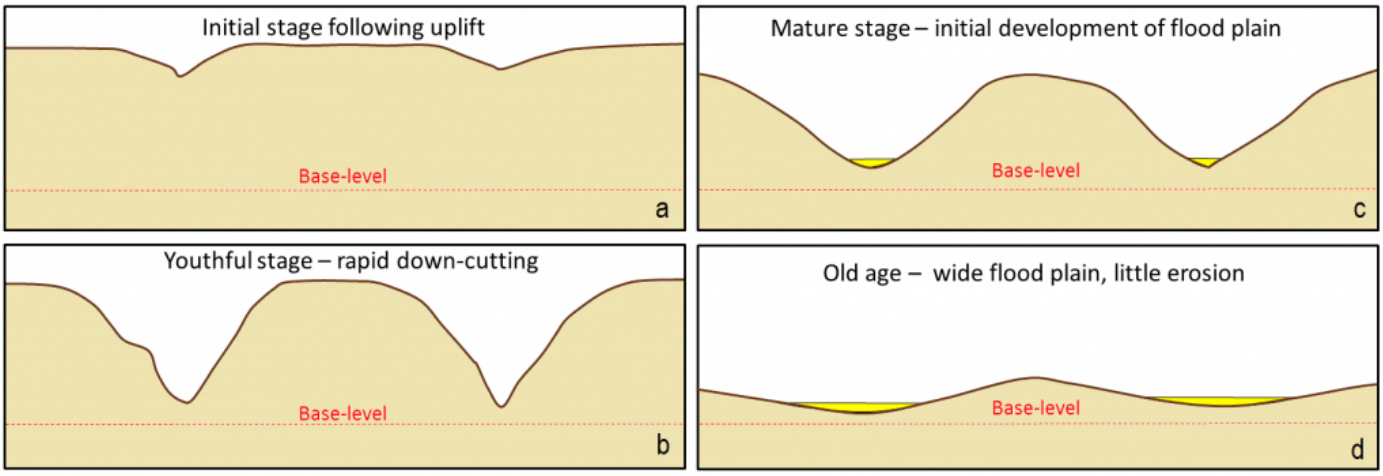


Figure 13.2.11 A depiction of the Davis cycle of erosion: a: initial stage, b: youthful stage, c: mature stage, and d: old age.

Image Descriptions

Figure 13.2.1 image description: Cawston Creek is a drainage basin made up of a number of small streams that flow into the Similkameen River. The Cawston Creek drainage basin also has a smaller tributary drainage basin that flows into it. [Return to

Figure 13.2.1]

Figure 13.2.2 image description: The graph shows the topographical profile of the Cawston Creek channel. Starting at 10 kilometers from the Similkameen River, the channel is at an elevation of 1800 meters. It flows downhill towards the river at a steady decline for seven kilometers. At three kilometers from the Similkameen River at an elevation of 275 meters, the channel levels out and stays at that elevation until it joins with the Similkameen River. This is conveyed in the following table.

Distance from the Similkameen River (kilometers)	Elevation (meters)
10	1,800
9	1,750
8	1,500
7	1,300
6	1,125
5	950
4	600
3	280
2	275
1	275
0	275

To determine the gradient of the stream, you must divide the change in elevation (rise) by the change in distance (run). For example, the elevation at 4.5 kilometers from the river is 800 meters and at 8.0 kilometers from the river, the elevation is 1500 meters. That means that in 3.5 kilometers (run), elevation increased by 700 meters (rise). To determine the gradient, you must divide rise (700 meters) by run (3.5 kilometers), which equals a vertical increase of 200 meters per kilometre.

[Return to Figure 13.2.2]

Figure 13.2.4 image description: Three types of drainage patterns:

- Dendritic: a pattern of drainage channels that resembles the branches in a tree.
- Trellis: a drainage pattern in which tributaries typically flow parallel to one other but meet at right angles.
- Rectangular: a drainage pattern in which tributaries typically flow at right angles to each other and meet at right angles.

[Return to Figure 13.2.4]

Figure 13.2.5 image description: Two types of drainage patterns:

- Deranged: a pattern of drainage channels that is chaotic.
- Radial: a pattern of streams radiating out from a central point, typically an isolated mountain.

[Return to Figure 13.2.5]

Figure 13.2.6 image description: The topographic profile of a typical graded stream. While the topographical profile of Cawston Creek fell at a steady rate and then leveled out for the last three kilometers, the topographical profile of a typical graded stream starts out steep, but the slope (or gradient) gradually decreases towards the bottom to form a soft curve. [Return to Figure 13.2.6]

Figure 13.2.10 image description: The Fraser River snaking through a channel that it has worn down over the years. The sides of the bank rise steeply, level out, and then rise again to produce terraces. [Return to Figure 13.2.10]

Figure 13.2.11 image description: William Davis's cycle of erosion.

1. Initial stage: A tectonic uplift forms two sharp valleys on each side.
2. Youthful stage: Steams flowing through those valleys erode the surrounding earth quickly and cut deep to form sharp V's.

3. Mature stage: The streams become wider and the banks more gradual. This marks the initial development of a flood plain.
4. Old age: Steams for a wide flood plain where little erosion occurs.

[Return to Figure 13.2.11]

Media Attributions

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13.3: Stream Erosion and Deposition

As we discussed in Chapter 6, flowing water is a very important mechanism for erosion, transportation and deposition of sediments. Water flow in a stream is primarily related to the stream's gradient, but it is also controlled by the geometry of the stream channel. As shown in Figure 13.3.1, water flow velocity is decreased by friction along the stream bed, so it is slowest at the bottom and edges and fastest near the surface and in the middle. In fact, the velocity just below the surface is typically a little higher than right at the surface because of friction between the water and the air. On a curved section of a stream, flow is fastest on the outside and slowest on the inside.

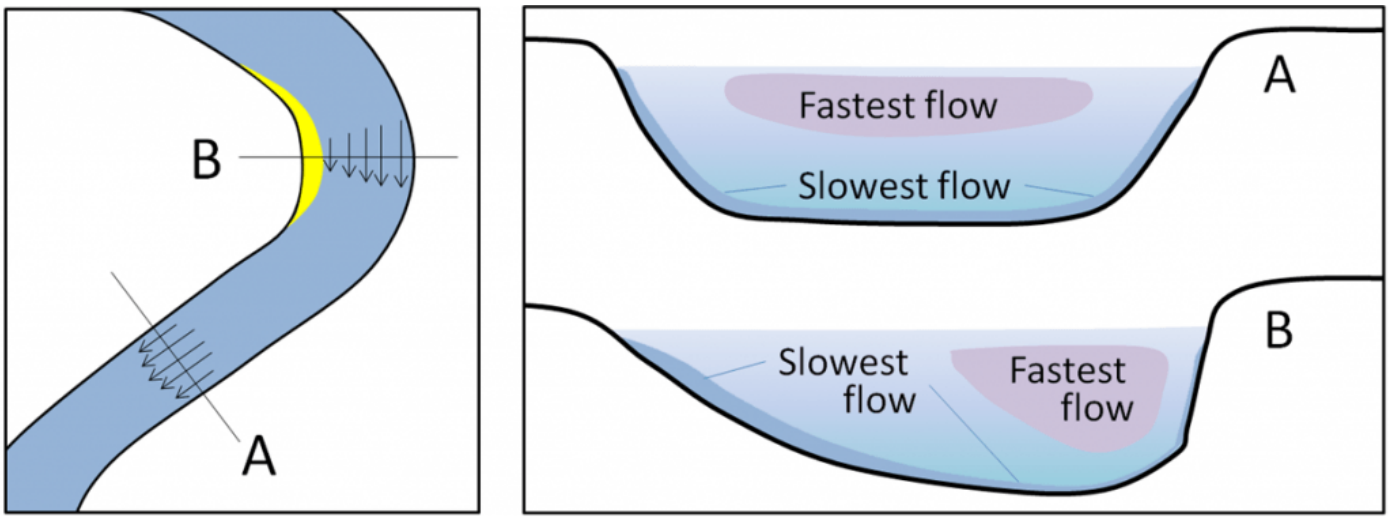


Figure 13.3.1 The relative velocity of stream flow depending on whether the stream channel is straight or curved (left), and with respect to the water depth (right).

Other factors that affect stream-water velocity are the size of sediments on the stream bed—because large particles tend to slow the flow more than small ones—and the **discharge**, or volume of water passing a point in a unit of time (e.g., cubic meters (m^3) per second). During a flood, the water level always rises, so there is more cross-sectional area for the water to flow in, however, as long as a river remains confined to its channel, the velocity of the water flow also increases.

Figure 13.3.2 shows the nature of sediment transportation in a stream. Large particles rest on the bottom—**bed load**—and may only be moved during rapid flows under flood conditions. They can be moved by **saltation** (bouncing) and by **traction** (being pushed along by the force of the flow).

Smaller particles may rest on the bottom some of the time, where they can be moved by saltation and traction, but they can also be held in suspension in the flowing water, especially at higher velocities. As you know from intuition and from experience, streams that flow fast tend to be turbulent (flow paths are chaotic and the water surface appears rough) and the water may be muddy, while those that flow more slowly tend to have laminar flow (straight-line flow and a smooth water surface) and clear water. Turbulent flow is more effective than laminar flow at keeping sediments in suspension.

Stream water also has a dissolved load, which represents (on average) about 15% of the mass of material transported, and includes ions such as calcium (Ca^{+2}) and chloride (Cl^{-}) in solution. The solubility of these ions is not affected by flow velocity.

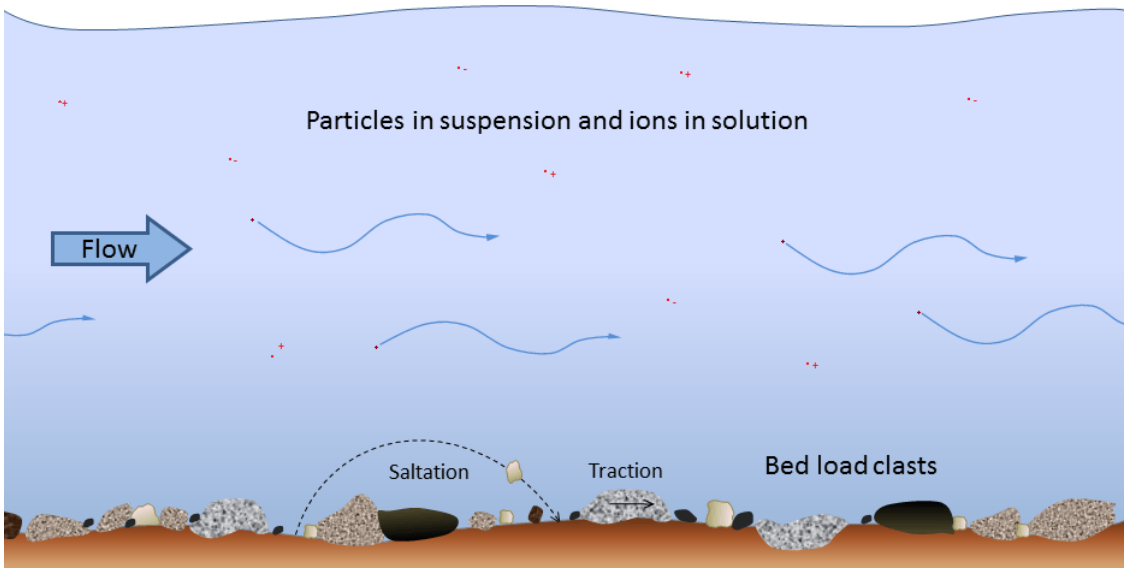


Figure 13.3.2 Modes of transportation of sediments and dissolved ions (represented by red dots with + and – signs) in a stream.

The faster the water is flowing, the larger the particles that can be kept in suspension and transported within the flowing water. However, as Swedish geographer Filip Hjulström discovered in the 1940s, the relationship between grain size and the likelihood of a grain being eroded, transported, or deposited is not as simple as one might imagine (Figure 13.3.3). Consider, for example, a 1 millimetre grain of sand. If it is resting on the bottom, it will remain there until the velocity is high enough to erode it, around 20 centimeters per second (cm/s). But once it is in suspension, that same 1 mm particle will remain in suspension as long as the velocity doesn't drop below 10 cm/s. For a 10 mm gravel grain, the velocity is 105 cm/s to be eroded from the bed but only 80 cm/s to remain in suspension.

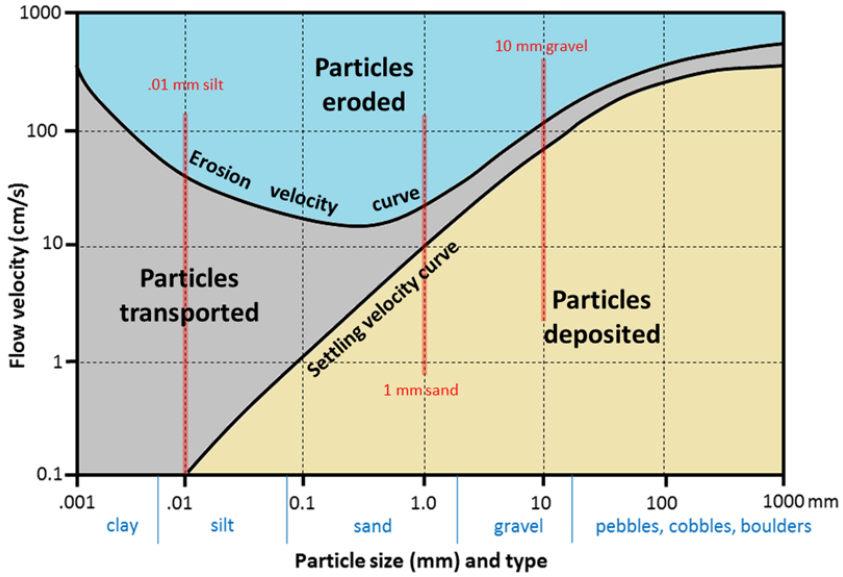


Figure 13.3.3 The Hjulström-Sundborg diagram showing the relationships between particle size and the tendency to be eroded, transported, or deposited at different current velocities.

On the other hand, a 0.01 mm silt particle only needs a velocity of 0.1 centimeters per second (cm/s) to remain in suspension, but requires 60 cm/s to be eroded. In other words, a tiny silt grain requires a greater velocity to be eroded than a grain of sand that is 100 times larger! For clay-sized particles, the discrepancy is even greater. In a stream, the most easily eroded particles are small sand grains between 0.2 mm and 0.5 mm. Anything smaller or larger requires a higher water velocity to be eroded and entrained in the flow. The main reason for this is that small particles, and especially the tiny grains of clay, have a strong tendency to stick together, and so are difficult to erode from the stream bed.

It is important to be aware that a stream can both erode and deposit sediments at the same time. At 100 cm/s, for example,

silt, sand, and medium gravel will be eroded from the stream bed and transported in suspension, coarse gravel will be held in suspension, pebbles will be both transported and deposited, and cobbles and boulders will remain stationary on the stream bed.

Exercise 13.3 Understanding the Hjulström-Sundborg diagram

Refer to the Hjulström-Sundborg diagram (Figure 13.3.3) to answer these questions.

1. A fine sand grain (0.1 millimeters) is resting on the bottom of a stream bed.
    1. What stream velocity will it take to get that sand grain into suspension?
    2. Once the particle is in suspension, the velocity starts to drop. At what velocity will it finally come back to rest on the stream bed?
2. A stream is flowing at 10 centimeters per second (which means it takes 10 seconds to go 1 meters, and that's pretty slow).
    1. What size of particles can be eroded at 10 centimeters per second?
    2. What is the largest particle that, once already in suspension, will remain in suspension at 10 centimeters per second?

See Appendix 3 for Exercise 13.3 answers.

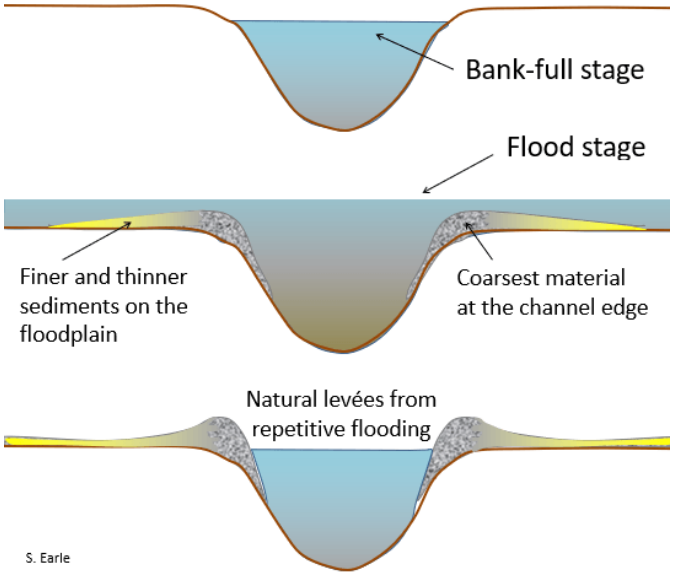


Figure 13.3.4 The development of natural levées during flooding of a stream. The sediments of the levée become increasingly fine away from the stream channel, and even finer sediments—clay, silt, and fine sand—are deposited across most of the flood plain.

A stream typically reaches its greatest velocity when it is close to flooding over its banks. This is known as the bank-full stage, as shown in Figure 13.3.4. As soon as the flooding stream overtops its banks and occupies the wide area of its flood plain, the water has a much larger area to flow through and the velocity drops significantly. At this point, sediment that was being carried by the high-velocity water is deposited near the edge of the channel, forming a natural bank or **levée**.

Image Descriptions

Figure 13.3.1 image description: When a stream curves, the flow of water is fastest on the outside of the curve and slowest on the inside of the curve. When the stream is straight and a uniform depth, the stream flows fastest in the middle near the top and slowest along the edges. When the depth is not uniform, the stream flows fastest in the deeper section. [Return to Figure 13.3.1]

Figure 13.3.3 image description:

- Erosion velocity curve: A 0.001 millimetre particle would erode at a flow velocity of 500 centimeters per second or greater. As the particle size gets larger, the minimum flow velocity needed to erode the particle decreases, with the lowest flow velocity being 30 centimeters per second to erode a 0.5 millimetre particle. To erode particles larger than 0.5 millimeters,

the minimum flow velocity rises again.
- Settling velocity curve: A 0.01 millimetre particle would be deposited with a flow velocity of 0.1 centimetre per second or less. As the flow velocity increases, only larger and larger particles will be deposited.
- Particles between these two curves (either moving too slow or being too small to be eroded or deposited) will be transported in the stream.

[Return to Figure 13.3.3]

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13.4: Stream Types

Stream channels can be straight or curved, deep and slow, or rapid and choked with coarse sediments. The cycle of erosion has some influence on the nature of a stream, but there are several other factors that are important.



Figure 13.4.1 The Cascade Falls area of the Kettle River, near Christina Lake, B.C. This stream has a step-pool morphology and a deep bedrock channel.

Youthful streams that are actively down-cutting their channels tend to be relatively straight and are typically ungraded (meaning that rapids and falls are common). As shown in Figures 13.0.1 and 13.4.1, youthful streams commonly have a **step-pool** morphology, meaning that the stream consists of a series of pools connected by rapids and waterfalls. They also have steep gradients and steep and narrow V-shaped valleys—in some cases steep enough to be called canyons.

In mountainous terrain, such as that in western Alberta and B.C., steep youthful streams typically flow into wide and relatively low-gradient U-shaped glaciated valleys. The youthful streams have high sediment loads, and when they flow into the lower-gradient glacial valleys where the velocity isn't high enough to carry all of the sediment **braided** patterns develop, characterized by a series of narrow channels separated by gravel bars (Figure 13.4.2).

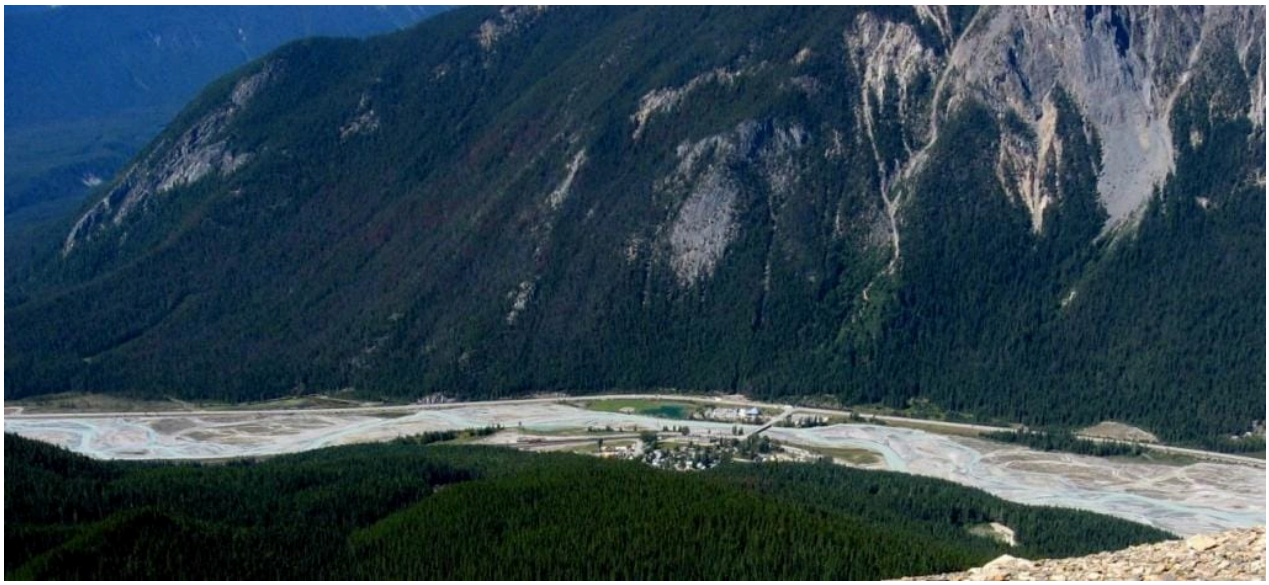


Figure 13.4.2 The braided channel of the Kicking Horse River at Field, B.C.



Figure 13.4.3 The braided channel of Meager Creek in the Mt. Meager area

Braided streams can develop anywhere there is more sediment than a stream is able to transport. One such environment is in volcanic regions, where explosive eruptions produce large amounts of unconsolidated material that gets washed into streams. Streams in the volcanic Mt. Meager area of southwestern British Columbia are good examples of this (Figure 13.4.3).

A stream that occupies a wide, flat flood plain with a low gradient typically carries only sand-sized and finer sediments and develops a sinuous flow pattern. As you saw in Figure 13.4.1, when a stream flows around a corner, the water on the outside has farther to go and tends to flow faster. This leads to erosion of the banks on the outside of the curve, deposition on the inside, and formation of a point bar (Figure 13.4.4). Over time, the sinuosity of the stream becomes increasingly exaggerated, and the channel migrates around within its flood plain, forming a **meandering** pattern.



Figure 13.4.4 The meandering channel of the Bonnell Creek, Nanoose, B.C. The stream is flowing toward the viewer. The sand and gravel point bar must have formed when the creek was higher and the flow faster than it was when the photo was taken.

A well-developed meandering river is shown in Figure 13.4.5. The meander in the middle of the photo has reached the point where the thin neck of land between two parts of the channel is about to be eroded through. When this happens, another **oxbow** lake will form like the others in the photo.



Figure 13.4.5 The meandering channel of the Nowitna River, Alaska. Numerous oxbow lakes are present and another meander cutoff will soon take place.

Exercise 13.4 Determining stream gradients

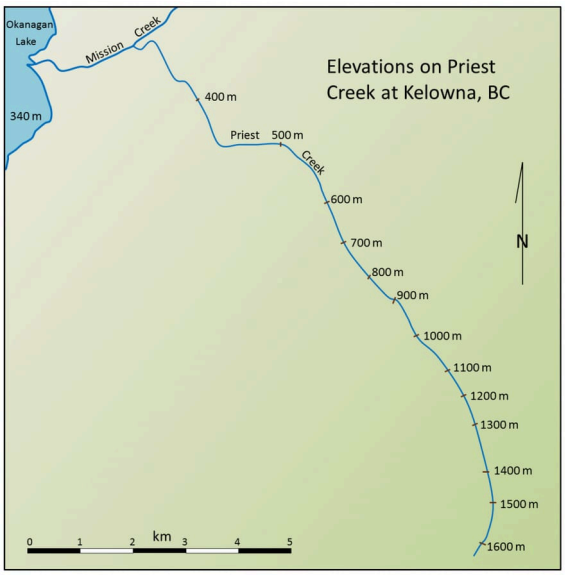


Figure 13.4.6 Elevations on Priest Creek at Kelowna, BC.

Gradient is the key factor controlling stream velocity, and of course, velocity controls sediment erosion and deposition. This map shows the elevations of Priest Creek in the Kelowna area. The length of the creek between 1,600 meters and 1,300 meters elevation is 2.4 kilometers, so the gradient is 300/2.4 = 125 meters per kilometre.

1. Use the scale bar to estimate the distance between 1,300 meters and 600 meters and then calculate that gradient.
2. Estimate the gradient between 600 and 400 meters.
3. Estimate the gradient between 400 meters on Priest Creek and the point where Mission Creek enters Okanagan Lake.

See Appendix 3 for Exercise 13.4 answers.

At the point where a stream enters a still body of water—a lake or the ocean—sediment is deposited and a delta forms. The Fraser River has created a large delta, which extends out into the Strait of Georgia (Figure 13.4.7). Much of the Fraser delta is very young in geological terms. Shortly after the end of the last glaciation (10,000 years ago), the delta did not extend past New Westminster. Since that time, all of the land that makes up Richmond, Delta, and parts of New Westminster and south Surrey has formed from sediment from the Fraser River. (You can see this in more detail at Geoscape Vancouver.)



Figure 13.4.7 The delta of the Fraser River and the plume of sediment that extends across the Strait of Georgia. The land outlined in red has formed over the past 10,000 years.

Image Descriptions

Figure 13.4.5 image description: A part of the Nowitna River has curved around so sharply that it almost forms a circle before curving the other way again. Eventually, as the barrier between these two parts of the channel erodes, they will be joined and form an oxbow lake. [Return to Figure 13.4.5]

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- Figure 13.4.7: Delta of the Fraser River by NASA. Taken September 2011. Adapted by Steven Earle. Public domain.

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13.5: Flooding

The discharge levels of streams are highly variable depending on the time of year and on specific variations in the weather from one year to the next. In Canada, most streams show discharge variability similar to that of the Stikine River in northwestern B.C., as illustrated in Figure 13.5.1. The Stikine River has its lowest discharge levels in the depths of winter when freezing conditions persist throughout most of its drainage basin. Discharge starts to rise slowly in May, and then rises dramatically through the late spring and early summer as a winter's worth of snow melts. For the year shown, the minimum discharge on the Stikine River was 56 cubic meters per second (m^3/s) in March, and the maximum was 37 times higher, 2,470 m^3/s, in May.

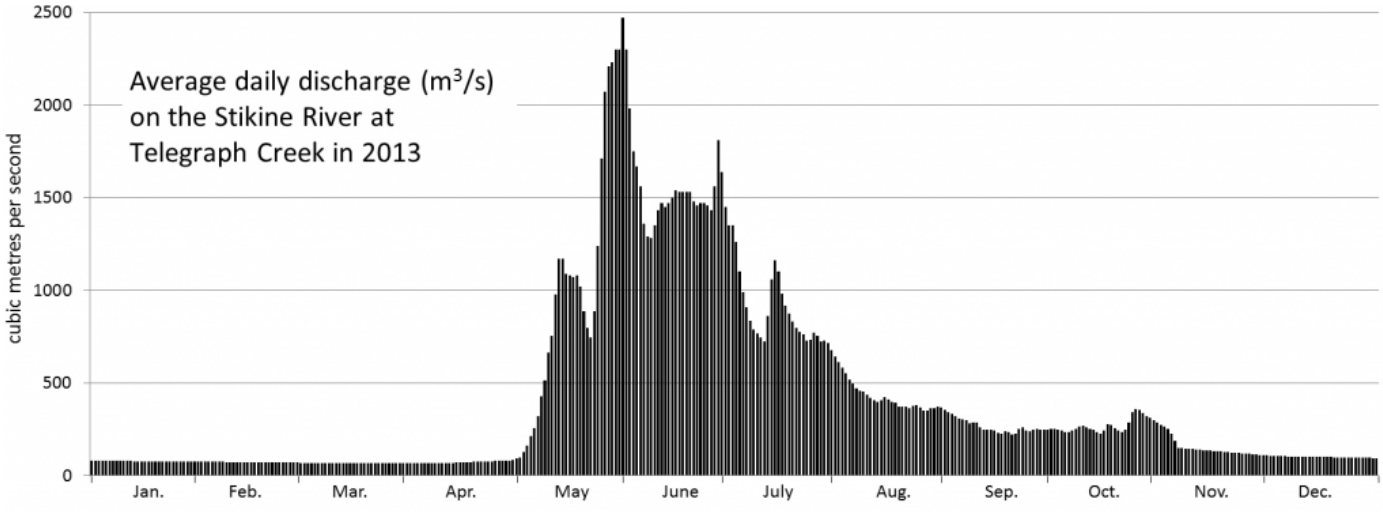


Figure 13.5.1 Variations in discharge of the Stikine River during 2013.

Streams in coastal areas of southern British Columbia show a very different pattern from those in most of the rest of the country because their drainage basins do not remain entirely frozen and because they receive a lot of rain (rather than snow) during the winter. The Qualicum River on Vancouver Island typically has its highest discharge levels in January or February and its lowest levels in late summer (Figure 13.5.2). In 2013, the minimum discharge was 1.6 m^3/s, in August, and the maximum was 34 times higher, 53 m^3/s, in March.

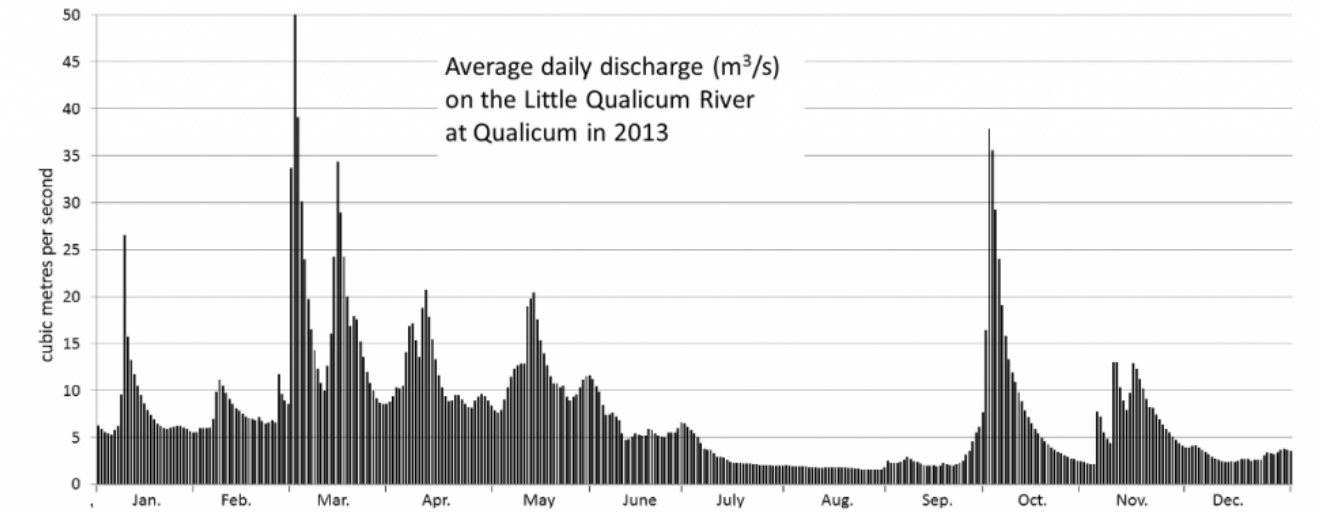


Figure 13.5.2 Variations in discharge of the Qualicum River during 2013.

When a stream's discharge increases, both the water level (**stage**) and the velocity increase as well. Rapidly flowing streams become muddy and large volumes of sediment are transported both in suspension and along the stream bed. In extreme

situations, the water level reaches the top of the stream's banks (the **bank-full stage**, see Figure 13.5.4), and if it rises any more, it floods the surrounding terrain. In the case of mature or old-age streams, this could include a vast area of relatively flat ground known as a flood plain, which is the area that is typically covered with water during a major flood. Because fine river sediments are deposited on flood plains, they are ideally suited for agriculture, and thus are typically occupied by farms and residences, and in many cases, by towns or cities. Such infrastructure is highly vulnerable to damage from flooding, and the people that live and work there are at risk.

Most streams in Canada have the greatest risk of flooding in the late spring and early summer when stream discharges rise in response to melting snow. In some cases, this is exacerbated by spring storms. In years when melting is especially fast and/or spring storms are particularly intense, flooding can be very severe.

One of the worst floods in Canadian history took place in the Fraser Valley in late May and early June of 1948. The early spring of that year had been cold, and a large snow pack in the interior was slow to melt. In mid-May, temperatures rose quickly and melting was accelerated by rainfall. Fraser River discharge levels rose rapidly over several days during late May, and the dykes built to protect the valley were breached in a dozen places. Approximately one-third of the flood plain was inundated and many homes and other buildings were destroyed, but there were no deaths. The Fraser River flood of 1948, which was the highest in the past century, was followed by very high river levels in 1950 and 1972 and by relatively high levels several times since then, the most recent being 2007 (Table 13.1). In the years following 1948, millions of dollars were spent repairing and raising the existing dykes and building new ones; since then damage from flooding in the Fraser Valley has been relatively limited.

Table 13.1 Ranking of the maximum stage and discharge values for the Fraser River at Hope between 1948 and 2008. Typical discharge levels are around 1,000 m^3/s.[1]

Rank	Year	Month	Date	Stage (meters)	Discharge (cubic meters per second)
1	1948	May	31	11.0	15,200
2	1972	June	16	10.1	12,900
3	1950	June	20	9.9	12,500
4	1964	June	21	9.6	11,600
5	1997	June	5	9.5	11,300
6	1955	June	29	9.4	11,300
7	1999	June	22	9.4	11,000
8	2007	June	10	9.3	10,850
9	1974	June	22	9.3	10,800
10	2002	June	21	9.2	10,600

Serious flooding happened in July in 1996 in the Saguenay-Lac St. Jean region of Quebec. In this case, the floods were caused by two weeks of heavy rainfall followed by one day of exceptional rainfall. July 19 saw 270 millimeters of rain, equivalent to the region's normal rainfall for the entire month of July. Ten deaths were attributed to the Saguenay floods, and the economic toll was estimated at $1.5 billion.

Just a year after the Saguenay floods, the Red River in Minnesota, North Dakota, and Manitoba reached its highest level since 1826. As is typical for the Red River, the 1997 flooding was due to rapid snowmelt. Because of the south to north flow of the river, the flooding starts in Minnesota and North Dakota, where melting starts earlier, and builds toward the north. The residents of Manitoba had plenty of warning that the 1997 flood was coming because there was severe flooding at several locations on the U.S. side of the border.

After the 1950 Red River flood, the Manitoba government built a channel around the city of Winnipeg to reduce the potential of flooding in the city (Figure 13.5.3). Known as the Red River Floodway, the channel was completed in 1964 at a cost of $63 million. Since then it has been used many times to alleviate flooding in Winnipeg, and is estimated to have saved many billions of dollars in flood damage. The massive 1997 flood was almost too much for the floodway; in fact the amount of water diverted was greater than the designed capacity. The floodway has recently been expanded so that it can be used to divert more of the Red River's flow away from Winnipeg.

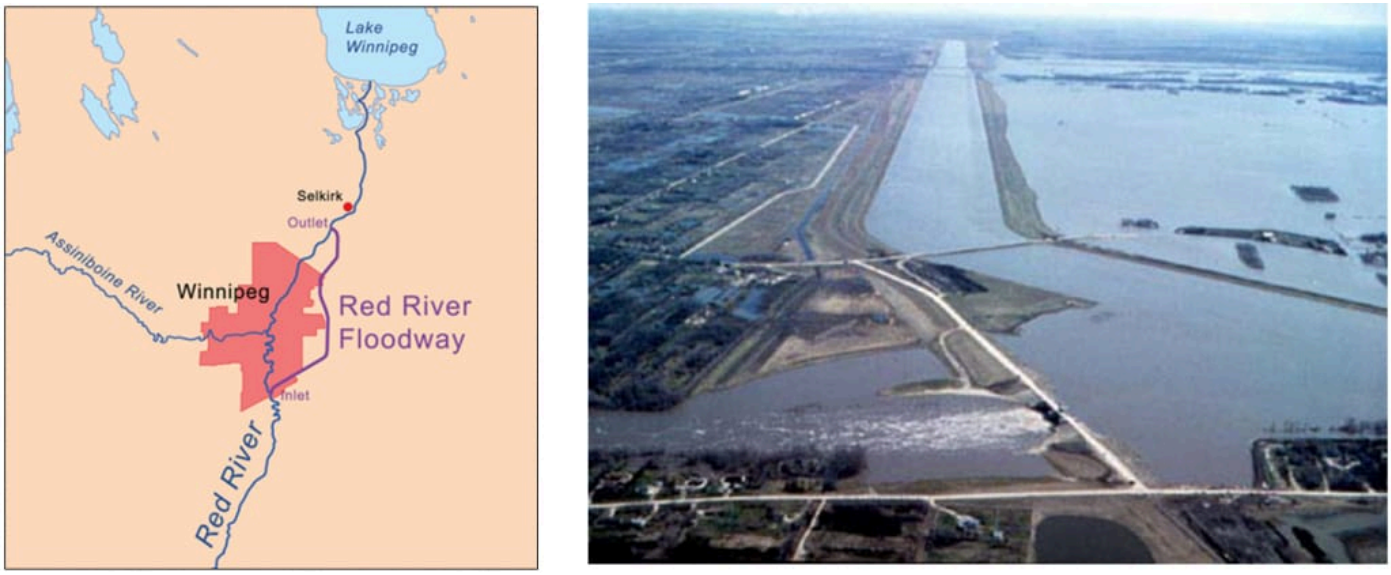


Figure 13.5.3 Map of the Red River floodway around Winnipeg, MB (left), and aerial view of the southern (inlet) end of the floodway (right).

Canada's most costly flood ever was the June 2013 flood in southern Alberta. The flooding was initiated by snowmelt and worsened by heavy rains in the Rockies due to an anomalous flow of moist air from the Pacific and the Caribbean. At Canmore, rainfall amounts exceeded 200 millimeters in 36 hours, and at High River, 325 millimeters of rain fell in 48 hours.

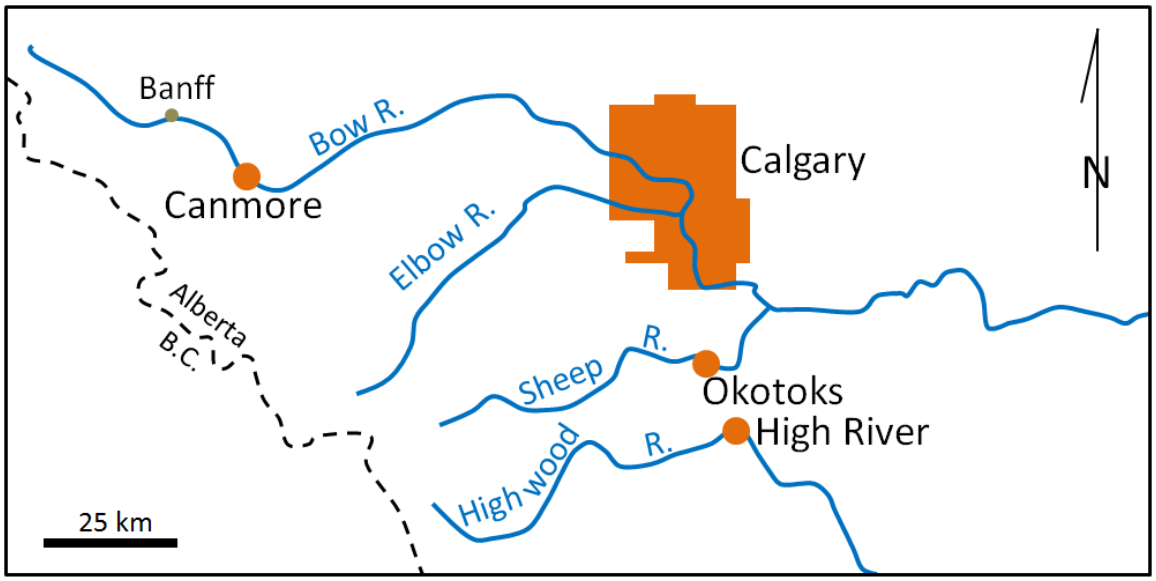


Figure 13.5.4 Map of the communities most affected by the 2013 Alberta floods (in orange).

In late June and early July, the discharges of several rivers in the area, including the Bow River in Banff, Canmore, and Exshaw, the Bow and Elbow Rivers in Calgary, the Sheep River in Okotoks, and the Highwood River in High River, reached levels that were 5 to 10 times higher than normal for the time of year (see Exercise 13.5). Large areas of Calgary, Okotoks, and High River were flooded and five people died (see Figures 13.5.2 and 13.5.3). The cost of the 2013 flood is estimated to be approximately $5 billion. Learn more about Alberta's flood of the century.

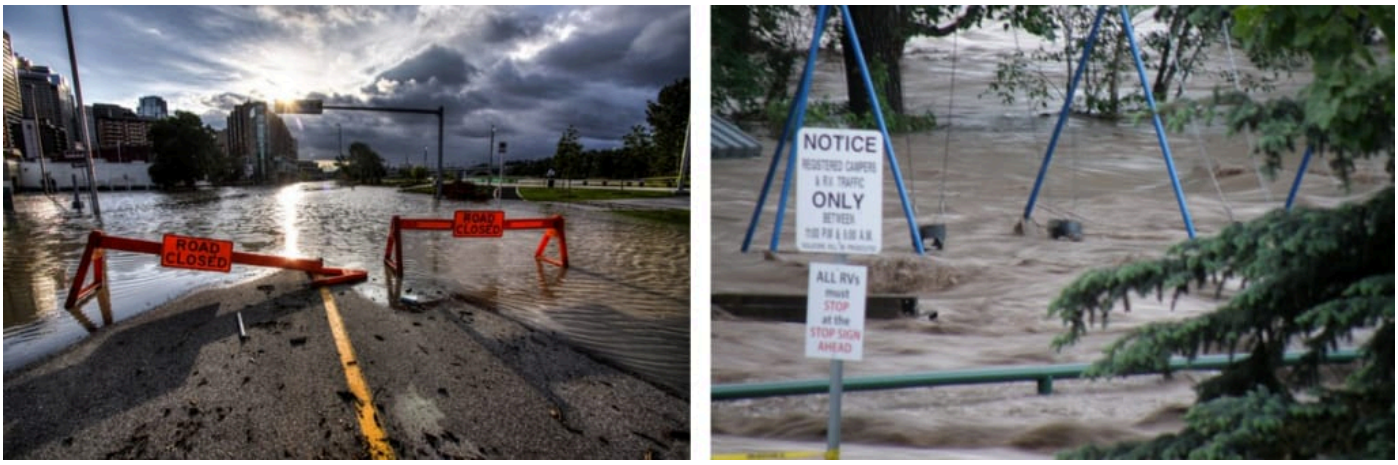


Figure 13.5.5 Flooding in Calgary (June 21, left) and Okotoks (June 20, right) during the 2013 southern Alberta flood.

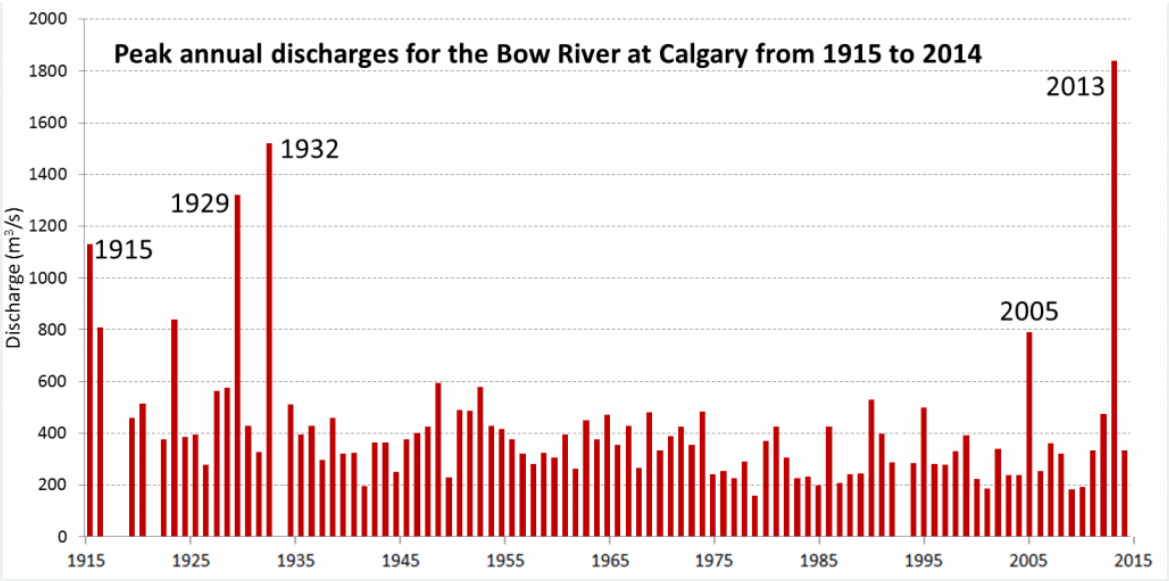


Figure 13.5.6 Graph of the highest discharge from the Bow River each year from 1915 to 2014.

Figure 13.5.6 shows the highest discharges per year between 1915 and 2014 on the Bow River at Calgary. Using this data set, we can calculate the recurrence interval (Ri) for any particular flood magnitude using the equation: Ri = (n+1)/r (where n is the number of floods in the record being considered, and r is the rank of the particular flood). There are a few years missing in this record, and the actual number of data points is 95.

The largest flood recorded on the Bow River over that period was the one in 2013, 1,840 cubic meters per second (m^3/s) on June 21. Ri for that flood is (95+1)/1 = 96 years. The probability of such a flood in any future year is 1/Ri, which is 1%. The fifth largest flood was just a few years earlier in 2005, at 791 m^3/s. Ri for that flood is (95+1)/5 = 19.2 years. The recurrence probability is 5%.

1. Calculate the recurrence interval for the second largest flood (1932, 1,520 m^3/s).
2. What is the probability that a flood of 1,520 m^3/s will happen next year?
3. Examine the 100-year trend for floods on the Bow River. If you ignore the major floods (the labelled ones), what is the general trend of peak discharges over that time?

See Appendix 3 for Exercise 13.5 answers.

One of the things that the 2013 flood on the Bow River teaches us is that we can't predict when a flood will occur or how big it will be, so in order to minimize damage and casualties we need to be prepared. Some of the ways of doing that are as follows:

- Mapping flood plains and not building within them
- Building dykes or dams where necessary
- Monitoring the winter snowpack, the weather, and stream discharges
- Creating emergency plans
- Educating the public

Image Descriptions

Figure 13.5.1 image description: The highest and lowest daily average discharge from the Stikine River by month in 2013 in cubic meters per second (m^3/s).

Month	Lowest daily average discharge (m^3/s)	Highest daily average discharge (m^3/s)
January	100	100
February	100	100
March	100	100
April	100	100
May	150	2,450
June	1,250	1,800
July	700	1,450
August	300	700
September	250	300
October	250	750
November	150	600
December	100	100

[Return to Figure 13.5.1]

Figure 13.5.2 image description: The highest and lowest daily average discharge from the Qualicum River by month in 2013 in cubic meters per second (m^3/s).

Month	Lowest daily average discharge (m^3/s)	Highest daily average discharge (m^3/s)
January	5	24
February	6	14
March	8	50
April	7	21
May	7	20
June	5	12
July	3	7
August	2	3
September	3	8
October	3	37
November	3	13
December	3	4

[Return to Figure 13.5.2]

Figure 13.5.6 image description: Peak annual discharge from the Bow River from 1915 to 2014 in cubic meters per second (m3/s)

Year	Peak Annual Discharge (m^3/s)
1915	1,125 (major)
1916	800
1919	475
1920	525
1922	375
1923	850
1924	400
1925	400
1926	275
1927	575
1928	580
1929	1,325 (major)
1930	425
1931	325
1932	1,520 (major)
1934	525
1935	400
1936	425
1937	300
1938	475
1939	325
1940	325
1941	200
1942	375
1943	375
1944	250
1945	375
1946	400
1947	425
1948	600
1949	225
1950	500
1951	500
1952	575
1953	425

Year	Peak Annual Discharge (m^3/s)
1954	400
1955	375
1956	325
1957	275
1958	325
1959	300
1960	400
1961	250
1962	450
1963	375
1964	475
1965	350
1966	425
1967	275
1968	475
1969	325
1970	400
1971	425
1972	350
1973	350
1974	250
1975	250
1976	225
1977	300
1978	150
1979	375
1980	425
1981	300
1982	225
1983	225
1984	200
1985	425
1986	200
1987	250
1988	250
1989	550
1990	400

Year	Peak Annual Discharge (m^3/s)
1991	300
1994	300
1995	500
1996	300
1997	300
1998	350
1999	400
2000	225
2001	175
2002	350
2003	250
2004	250
2005	791 (major)
2006	250
2007	375
2008	325
2009	175
2010	200
2011	350
2012	475
2013	1,840 (major)
2014	350

[Return to Figure 13.5.6]

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- Figure 13.5.5: “Riverfront Ave Calgary Flood” © Ryan L. C. Quan. CC BY-SA. “Okotoks – June 20, 2013 – Flood waters in local campground playground-03” © Stephanie N. Jones. CC BY-SA.
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1. Mannerström, M, 2008, Comprehensive Review of Fraser River at Hope Flood Hydrology and Flows Scoping Study, Report prepared for the B.C. Ministry of the Environment [PDF]. Available at: http://www.env.gov.bc.ca/wsd/public_safety/flood/pdfs_word/review_fraser_flood_flows_hope.pdf↵

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13.6: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
13.1 The Hydrological Cycle	Water is stored in the oceans, glacial ice, the ground, lakes, rivers, and the atmosphere. Its movement is powered by the sun and gravity.
13.2 Drainage Basins	All of the precipitation that falls within a drainage basin flows into the stream that drains that area. Stream drainage patterns are determined by the type of rock within the basin. Over geological time, streams change the landscape that they flow within, and eventually they become graded, meaning their profile is a smooth curve. A stream can lose that gradation if there is renewed uplift or if their base level changes for some reason.
13.3 Stream Erosion and Deposition	Erosion and deposition of particles within streams is primarily determined by the velocity of the water. Erosion and deposition of different-sized particles can happen at the same time. Some particles are moved along the bottom of a river while some are suspended in the water. It takes a greater velocity of water to erode a particle from a stream bed than it does to keep it in suspension. Ions are also transported in solution. When a stream rises and then occupies its flood plain, the velocity slows and natural levées form along the edges of the channel.
13.4 Stream Types	Youthful streams in steep areas erode rapidly, and they tend to have steep, rocky, and relatively straight channels. Where sediment-rich streams empty into areas with lower gradients, braided streams can form. In areas with even lower gradients, and where silt and sand are the dominant sediments, meanders are common. Deltas form where streams flow into standing water.
13.5 Flooding	Most streams in Canada have their highest discharge rates in spring and early summer, although many of B.C.'s coastal streams are highest in the winter. Floods happen when a stream rises high enough to spill over its banks and spread across its flood plain. Some of the more significant floods in Canada include the Fraser River flood of 1948, the Saguenay River flood of 1996, the Red River flood of 1997, and the Alberta floods of 2013. We can estimate the probability of a specific flood level based on the record of past floods, and we can take steps to minimize the impacts of flooding.

Questions for Review

1. What is the proportion of liquid (not frozen) ***fresh***water on Earth expressed as a percentage of all water on Earth?
2. What percentage of that fresh water is groundwater?
3. What type of rock, and what processes, can lead to the formation of a trellis drainage pattern?
4. Why do many of the streams in the southwestern part of Vancouver Island flow to the ocean as waterfalls?
5. Where would you expect to find the fastest water flow on a straight stretch of a stream?
6. Sand grains can be moved by traction and saltation. What minimum stream velocities might be required to move 1 millimetre sand grains?
7. If the flow velocity of a stream is 1 cm per second, what sizes of particles can be eroded, what sizes can be transported if they are already in suspension, and what sizes of particles cannot be moved at all?
8. Under what circumstances might a braided stream develop?
9. How would the gradient of a stream be affected if a meander is cut off?
10. The elevation of the Fraser River at Hope is 41 meters. From there it flows approximately 147 kilometers to the sea. What is the average gradient of the river (meters per kilometre) over that distance?
11. How do B.C.'s coastal streams differ from most of the rest of the streams in Canada in terms of their annual flow patterns? Why?
12. Why do most serious floods in Canada happen in late May, June, or early July?
13. There is a 65-year record of peak annual discharges on the Ashnola River near Princeton, B.C. During this time, the second highest discharge was 175 m^3/s. Based on this information, what is the recurrence interval (Ri) for that discharge level, and what is the probability that there will be a similar peak discharge next year?

Answers to Review Questions can be found in Appendix 2.

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14: GROUNDWATER

14.1: Groundwater and Aquifers
14.2: Groundwater Flow
14.3: Groundwater Extraction
14.4: Groundwater Quality
14.5: Summary

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14.1: Groundwater and Aquifers

Groundwater is stored in the open spaces within rocks and within unconsolidated sediments. Rocks and sediments near the surface are under less pressure than those at significant depth and therefore tend to have more open space. For this reason, and because it's expensive to drill deep wells, most of the groundwater that is accessed by individual users is within the first 100 meters of the surface. Some municipal, agricultural, and industrial groundwater users get their water from greater depth, but deeper groundwater tends to be of lower quality than shallow groundwater, so there is a limit as to how deep we can go.

Porosity is the percentage of open space within an unconsolidated sediment or a rock. Primary porosity is represented by the spaces between grains in a sediment or sedimentary rock. Secondary porosity is porosity that has developed after the rock has formed. It can include fracture porosity—space within fractures in any kind of rock. Some volcanic rock has a special type of porosity related to vesicles, and some limestone has extra porosity related to cavities within fossils.

Porosity is expressed as a percentage calculated from the volume of open space in a rock compared with the total volume of rock. The typical ranges in porosity of a number of different geological materials are shown in Figure 14.1.1. Unconsolidated sediments tend to have higher porosity than consolidated ones because they have no cement, and most have not been strongly compressed. Finer-grained materials (e.g., silt and clay) tend to have greater porosity—some as high as 70%—than coarser materials (e.g., gravel). Primary porosity tends to be higher in well-sorted sediments compared to poorly sorted sediments, where there is a range of smaller particles to fill the spaces made by the larger particles. Glacial till, which has a wide range of grain sizes and is typically formed under compression beneath glacial ice, has relatively low porosity.

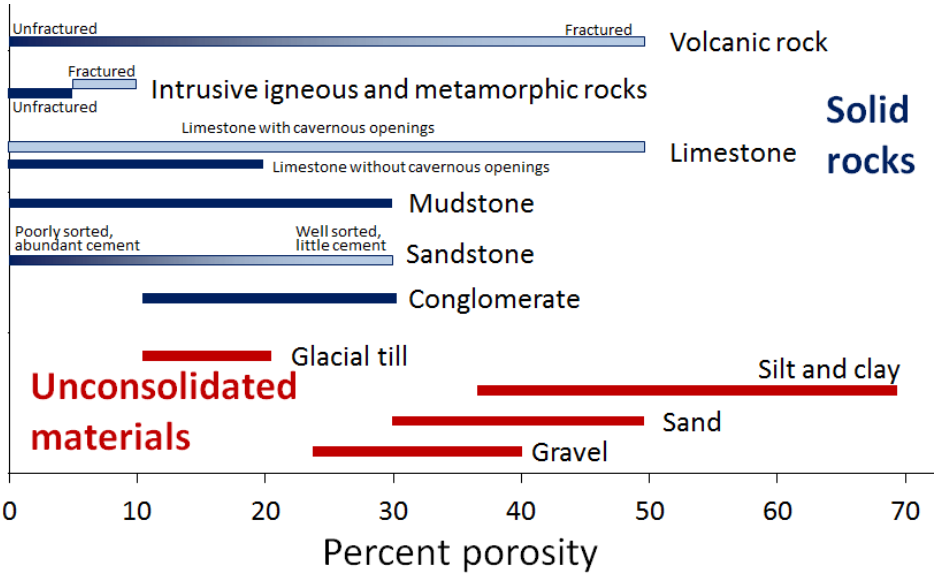


Figure 14.1.1 Variations in porosity of unconsolidated materials (in red) and rocks (in blue).

Consolidation and cementation during the process of lithification of unconsolidated sediments into sedimentary rocks reduces primary porosity. Sedimentary rocks generally have porosities in the range of 10% to 30%, some of which may be secondary (fracture) porosity. The grain size, sorting, compaction, and degree of cementation of the rocks all influence primary porosity. For example, poorly sorted and well-cemented sandstone and well-compressed mudstone can have very low porosity. Igneous or metamorphic rocks have the lowest primary porosity because they commonly form at depth and have interlocking crystals. Most of their porosity comes in the form of secondary porosity in fractures. Of the consolidated rocks, well-fractured volcanic rocks and limestone that has cavernous openings produced by dissolution have the highest potential porosity, while intrusive igneous and metamorphic rocks, which formed under great pressure, have the lowest.

Porosity is a measure of how much water can be stored in geological materials. Almost all rocks contain some porosity and therefore contain groundwater. Groundwater is found under your feet and everywhere on the planet. Considering that sedimentary rocks and unconsolidated sediments cover about 75% of the continental crust with an average thickness of a few hundred meters, and that they are likely to have around 20% porosity on average, it is easy to see that a huge volume of water can be stored in the ground.

Porosity is a description of how much space there could be to hold water under the ground, while **permeability** describes how those pores are shaped and interconnected. This determines how easy it is for water to flow from one pore to the next. Larger pores mean there is less friction between flowing water and the sides of the pores. Smaller pores mean more friction along pore walls, and also more twists and turns for the water to have to flow-through. A permeable material has a greater number of larger, well-connected pores spaces, whereas an impermeable material has fewer, smaller pores that are poorly connected. Permeability is the most important variable in groundwater. Permeability describes how easily water can flow through the rock or unconsolidated sediment and how easy it will be to extract the water for our purposes. The characteristic of permeability of a geological material is quantified by geoscientists and engineers using a number of different units, but the most common is the **hydraulic conductivity**. The symbol used for hydraulic conductivity is *K*. Although hydraulic conductivity can be expressed in a range of different units, in this book we will always use meters per second.

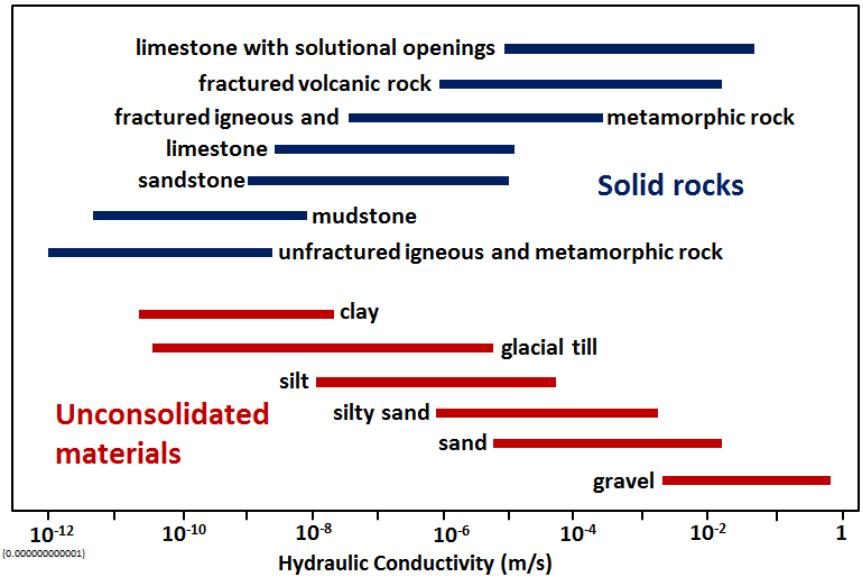
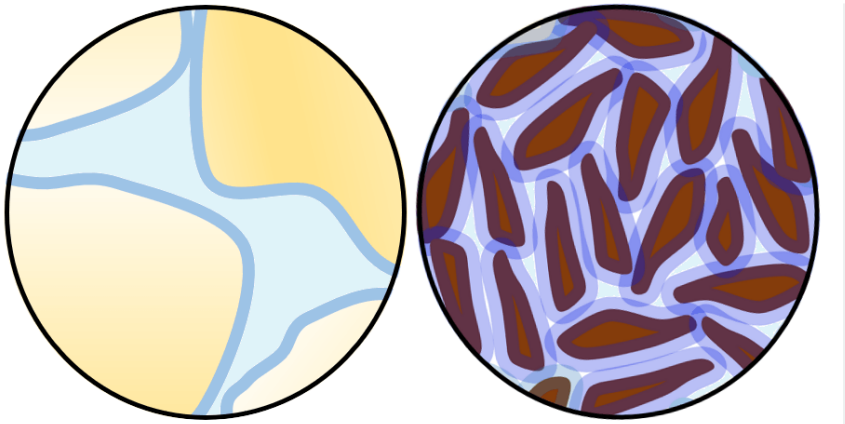


Figure 14.1.2 Variations in hydraulic conductivity (in meters per second) of unconsolidated materials (in red) and of rocks (in blue).

As shown on Figure 14.1.2 there is a wide range of permeability in geological materials from 10^{-12} meters per second (0.000000000001 meters per second) to around 1 metre per second. Unconsolidated materials are generally more permeable than the corresponding rocks (compare sand with sandstone, for example), and the coarser materials are much more permeable than the finer ones. The least permeable rocks are unfractured intrusive igneous and metamorphic rocks, followed by unfractured mudstone, sandstone, and limestone. The permeability of sandstone can vary widely depending on the degree of sorting and the amount of cement that is present. Fractured igneous and metamorphic rocks, and especially fractured volcanic rocks, can be highly permeable, as can limestone that has been dissolved along fractures and bedding planes to create solutional openings.



Sand deposit. The light blue represents water that is free to move, while the medium blue is water held to the grains by surface tension.
Most of the water is free to move.

Clay deposit. The light blue represents water that is free to move, while the blue is water held to the grains by surface tension.
Most of the water is unable to move.

Figure 14.1.3 Porosity versus permeability in sand and clay

Both sand and clay deposits (and sandstone and mudstone) are quite porous (30% to 50% for sand and 40% to 70% for silt and clay), but while sand can be quite permeable, clay and mudstone are not.

The surface of most silicate mineral grains has a slight negative charge due to imperfections in the mineral structure. Water (H_2O) is a polar molecule. This means that while it has no overall electrical charge, one side of the molecule has a slight positive charge (the side with the two hydrogens), compared to a slight negative charge on the other side. Water is strongly attracted to all mineral grains and water within that bound water layer (a few microns around each grain) is not able to move and flow along with the rest of the groundwater.

In the two diagrams of Figure 14.1.3 the bound water is represented by medium blue lines around each grain, and the water that is not bound, and can flow, is shown in light blue. In the sand, there is still a lot of water that is able to move through the sediment, but in the clay deposit almost all of the water is held tightly to the grains and this reduces the permeability.

We have now seen that there is a wide range of porosity in geological materials and an even wider range of permeability. Groundwater exists everywhere there is porosity. However, whether that groundwater is able to flow in significant quantities depends on the permeability. An **aquifer** is defined as a body of rock or unconsolidated sediment that has sufficient permeability to allow water to flow through it. Unconsolidated materials like gravel, sand, and even silt make relatively good aquifers, as do rocks like sandstone. Other rocks can be good aquifers if they are well fractured. An **aquitard** is a body that does not allow transmission of a significant amount of water, such as a clay, a till, or a poorly fractured igneous or metamorphic rock. These are relative terms, not absolute, and are usually defined based on someone's desire to pump groundwater; what is an aquifer to someone who does not need a lot of water, may be an aquitard to someone else who does. An aquifer that is exposed at the ground surface is called an **unconfined aquifer**. An aquifer where there is a lower permeability material between the aquifer and the ground surface is known as a **confined aquifer**, and the aquitard separating ground surface and the aquifer is known as the **confining layer**.

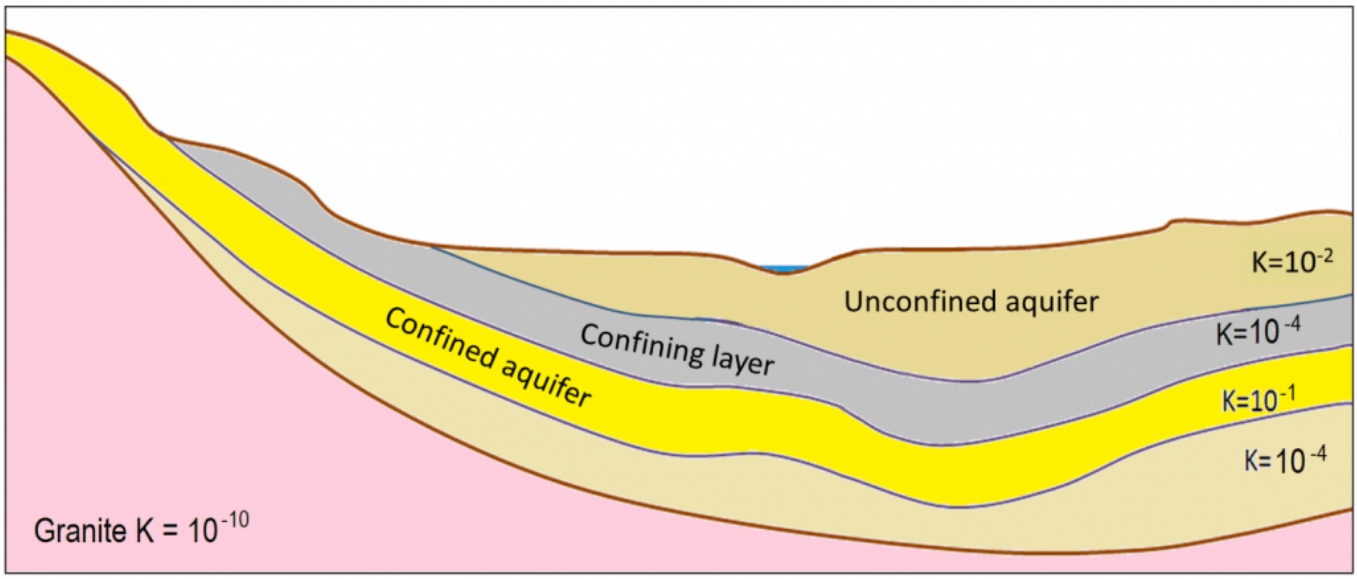


Figure 14.1.4 A cross-section showing materials that might serve as aquifers and confining layers. The relative permeabilities are denoted by hydraulic conductivity (K in meters per second). The pink rock is granite; the other layers are various sedimentary layers.

Figure 14.1.4 shows a cross-section of a series of rocks and unconsolidated materials, some of which might serve as aquifers and others as aquitards or confining layers. The granite is much less permeable than the other materials, and so is an aquitard in this context. The yellow layer is very permeable and would make an ideal aquifer. The overlying grey layer is a confining layer.

The upper buff-colored layer (K = 10^{-2} meters per second) does not have a confining layer and is an unconfined aquifer. The yellow layer (K = 10^{-1} meters per second) is "confined" by the confining layer (K = 10^{-4} meters per second), and is a confined aquifer. The confined aquifer gets most of its water from the upper part of the hill where it is exposed at the surface, and relatively little by seepage through the fine silt layer.

Image Descriptions

Figure 14.1.1 image description part 1: Percent porosity for solid rocks

Solid rocks	Percent porosity (Unfractured)	Percent porosity (Fractured)
Volcanic Rock	0% to roughly 20%	Roughly 20% to 50%
Intrusive igneous and metamorphic rocks	0% to 5%	5% to 10%
Limestone with cavernous openings	N/A	0% to 50%
Limestone without cavernous openings	0% to 20%	N/A
Mudstone	0% to 30%	N/A
Sandstone	0% to roughly 10%	Roughly 10% to 30%
Conglomerate	10% to 30%	N/A

Figure 14.1.1 image description part 2: Percent porosity of unconsolidated materials

Unconsolidated Materials	Percent porosity
Glacial Till	11% to 20%
Silt and clay	36% to 70%
Sand	30% to 50%
Gravel	24% to 40%

[Return to Figure 14.1.1]

Figure 14.1.2 image description part 1: Variations in hydraulic conductivity in meters per second in solid rocks.

Material type	Hydraulic conductivity (meters per second)
Limestone with solutional openings	10 to the power of negative 5 to 10 to the power of negative 1.2 (10^{-5} to $10^{-1.2}$)
Fractured volcanic rock	10 to the power of negative 6 to 10 to the power of negative 1.7 (10^{-6} to $10^{-1.7}$)
Fractured igneous and metamorphic rock	10 to the power of negative 7.5 to 10 to the power of negative 3.5 ($10^{-7.5}$ to $10^{-3.5}$)
Limestone	10 to the power of negative 8.4 to 10 to the power of negative 4.9 ($10^{-8.4}$ to $10^{-4.9}$)
Sandstone	10 to the power of negative 8.9 to 10 to the power of negative 5 ($10^{-8.9}$ to 10^{-5})
Mudstone	10 to the power of negative 11.3 to 10 to the power of negative 8 ($10^{-11.3}$ to 10^{-8})
Unfractured igneous and metamorphic rock	10 to the power of negative 12 to 10 to the power of negative 8.6 (10^{-12} to $10^{-8.6}$)

Figure 14.1.2 image description part 2: Variations in hydraulic conductivity in meters per second in unconsolidated materials.

Material Type	Hydraulic conductivity (meters per second)
Clay	10 to the power of negative 10.6 to 10 to the power of negative 7.7 ($10^{-10.6}$ to $10^{-7.7}$)
Glacial till	10 to the power of negative 10.4 to 10 to the power of negative 5.2 ($10^{-10.4}$ to $10^{-5.2}$)
Silt	10 to the power of negative 8 to 10 to the power of negative 4.3 (10^{-8} to $10^{-4.3}$)
Silty sand	10 to the power of negative 6 to 10 to the power of negative 2.7 (10^{-6} to $10^{-2.7}$)
Sand	10 to the power of negative 5.2 to 10 to the power of negative 1.7 ($10^{-5.2}$ to $10^{-1.7}$)
Gravel	10 to the power of negative 2.7 to 10 to the power of negative 0.1 ($10^{-2.7}$ to $10^{-0.1}$)

[Return to Figure 14.1.2]

Figure 14.1.4 image description: The relative permeability of various sedimentary layers. Permeability is denoted by hydraulic conductivity (K = meters per second). The buff-colored, top layer is an unconfined aquifer with K = 10^{-2}. The grey-colored second layer is a confining layer with K = 10^{-4}. The yellow-colored third layer is a confined aquifer with K = 10^{-1}. The fourth layer is unlabelled, with K = 10^{-4}. The bottom layer is granite, K = 10^{-10}. [Return to Figure 14.1.4]

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14.2: Groundwater Flow

If you go out into your garden or into a forest or a park and start digging, you will find that the soil is moist (unless you're in a desert), but it's not saturated with water. This means that some of the pore space in the soil is occupied by water, and some of the pore space is occupied by air (unless you're in a swamp). This is known as the **unsaturated zone**. If you could dig down far enough, you would get to the point where all of the pore spaces are 100% filled with water (saturated) and the bottom of your hole would fill up with water. The level of water in the hole represents the **water table**, which is the surface of the **saturated zone**. In most parts of British Columbia, the water table is several meters below the surface.

Water falling on the ground surface as precipitation (rain, snow, hail, fog, etc.) may flow off a hill slope directly to a stream in the form of **runoff**, or it may **infiltrate** the ground, where it is stored in the unsaturated zone. The water in the unsaturated zone may be used by plants (transpiration), evaporate from the soil (evaporation), or continue past the root zone and flow downward to the water table, where it **recharges** the groundwater.

A cross-section of a typical hillside with an unconfined aquifer is illustrated in Figure 14.2.1. In areas with topographic relief, the water table generally follows the land surface, but tends to come closer to surface in valleys, and intersects the surface where there are streams or lakes. The water table can be determined from the depth of water in a well that isn't being pumped, although, as described below, that only applies if the well is within an unconfined aquifer. In this case, most of the hillside forms the **recharge area**, where water from precipitation flows downward through the unsaturated zone to reach the water table. The area at the stream or lake to which the groundwater is flowing is a **discharge area**.

What makes water flow from the recharge areas to the discharge areas? Recall that water is flowing in pores where there is friction, which means it takes work to move the water. There is also some friction between water molecules themselves, which is determined by the viscosity. Water has a low viscosity, but friction is still a factor. All flowing fluids are always losing energy to friction with their surroundings. Water will flow from areas with high energy to those with low energy. Recharge areas are at higher elevations, where the water has high gravitational energy. It was energy from the sun that evaporated the water into the atmosphere and lifted it up to the recharge area. The water loses this gravitational energy as it flows from the recharge area to the discharge area.

In Figure 14.2.1, the water table is sloping; that slope represents the change in gravitational potential energy of the water at the water table. The water table is higher under the recharge area (90 meters) and lower at the discharge area (82 meters). Imagine how much work it would be to lift water 8 meters high in the air. That is the energy that was lost to friction as the groundwater flowed from the top of the hill to the stream.

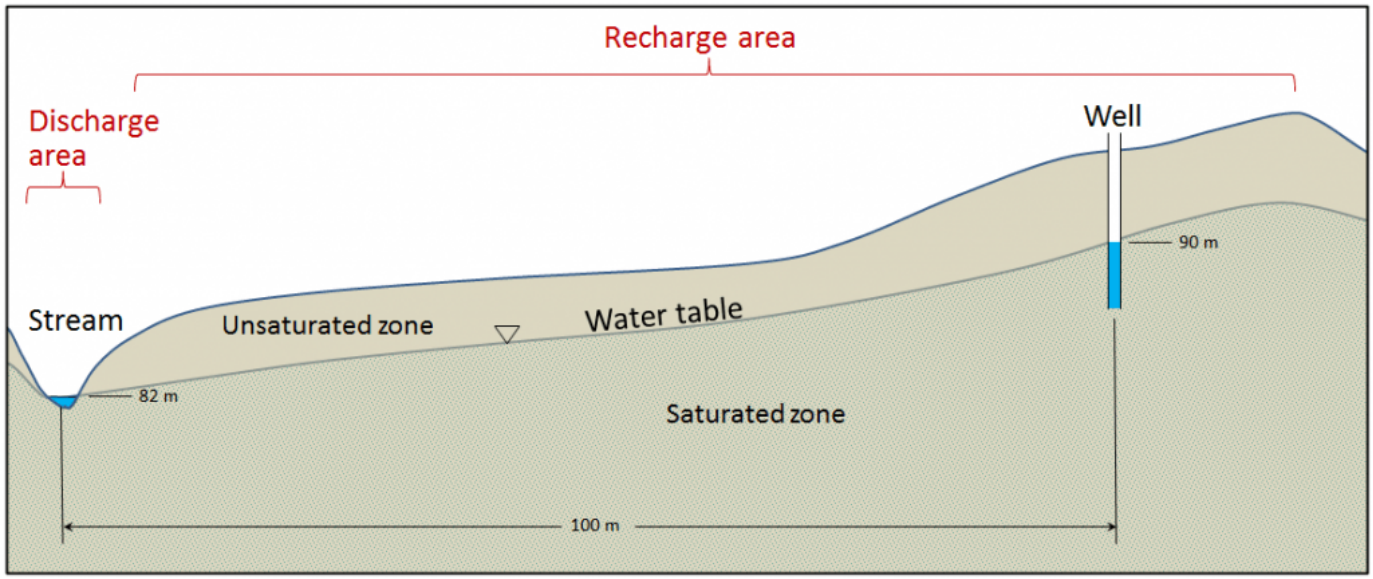


Figure 14.2.1 A depiction of the water table in cross-section, with the saturated zone below and the unsaturated zone above. The water table is denoted with a small upside-down triangle.

The situation gets a lot more complicated in the case of confined aquifers, but they are important sources of water so we need to understand how they work. As shown in Figure 14.2.2, there is always a water table, and that applies even if the geological materials at the surface have very low permeability. Where there is a confined aquifer — meaning one that is separated from the surface by a confining layer — this aquifer will have its own "water table," which is actually called a **potentiometric surface**, as it is a measure of the total potential energy of the water. The red dashed line in Figure 14.2.2 is the potentiometric surface for the confined aquifer, and it describes the total energy that water is under within the confined aquifer. If we drill a well into the unconfined aquifer, the water will rise to the level of the water table (well A in Figure 14.2.2). But if we drill a well through both the unconfined aquifer and the confining layer and into the confined aquifer, the water will rise above the top of the confined aquifer to the level of its potentiometric surface (well B in Figure 14.2.2). This is known as an **artesian well**, because the water rises above the top of the aquifer. In some situations, the potentiometric surface may be above the ground level. The water in a well drilled into the confined aquifer in this situation would rise above ground level, and flow out, if it's not capped (well C in Figure 14.2.2). This is known as a **flowing artesian well**.

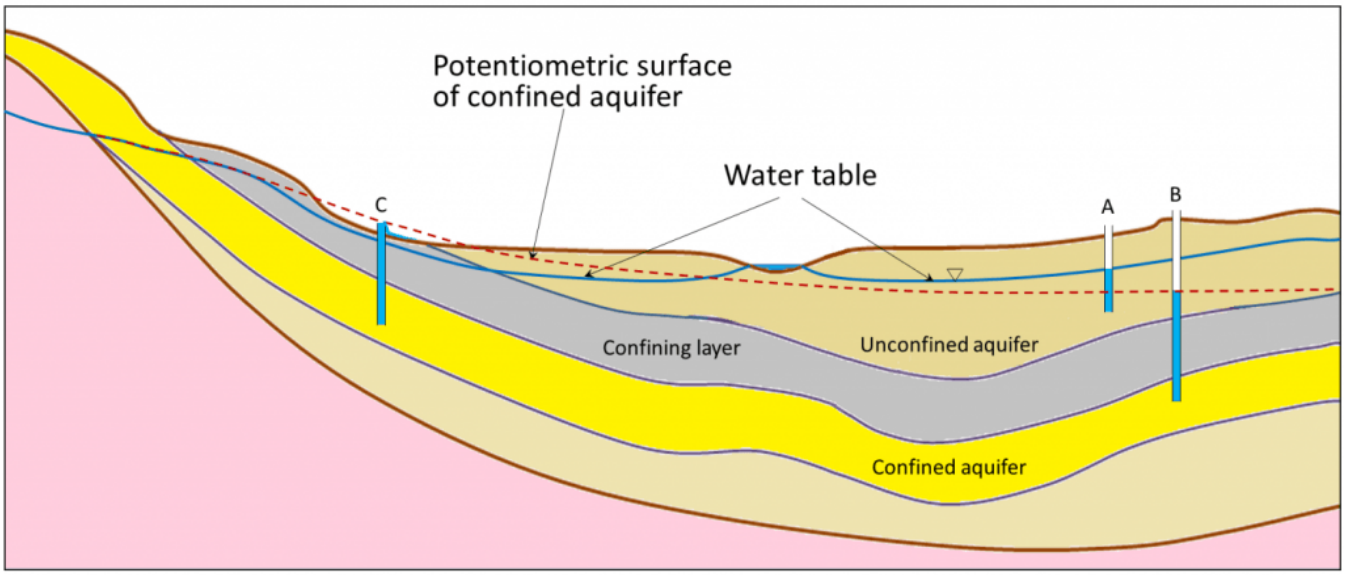


Figure 14.2.2 A depiction of the water table and the potentiometric surface of a confined aquifer.

In situations where there is an aquitard of limited extent, it is possible for a perched aquifer to exist as shown in Figure 14.2.3. Although perched aquifers may be good water sources at some times of the year, they tend to be relatively thin and small, and so can easily be depleted with over-pumping.

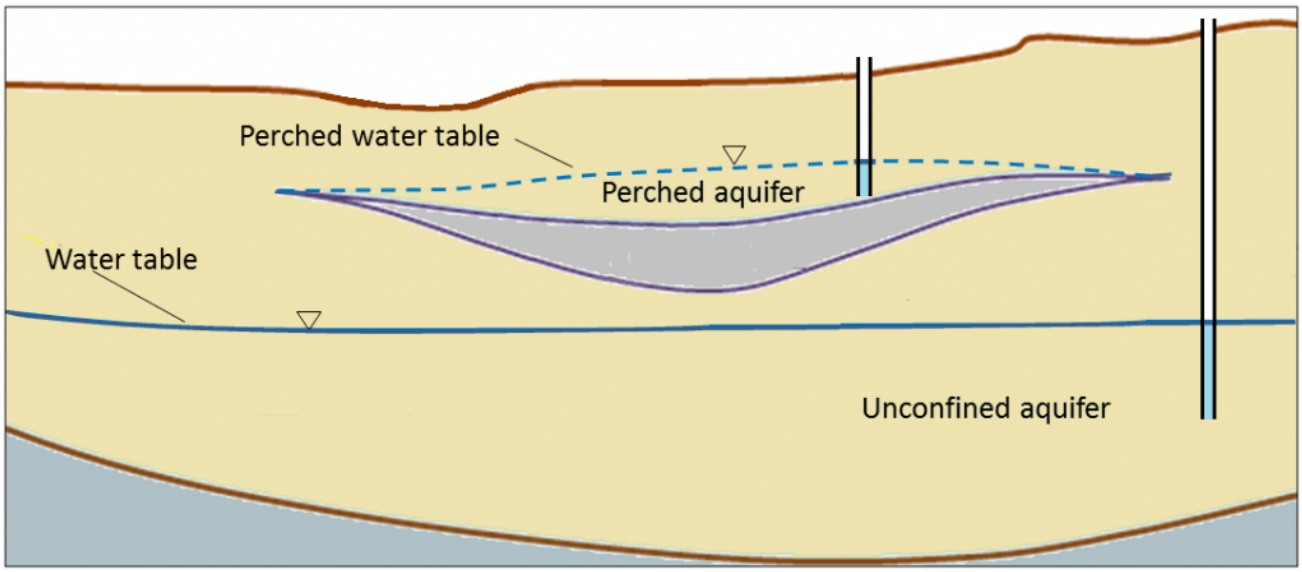


Figure 14.2.3 A perched aquifer above a regular unconfined aquifer.

In 1856, French engineer Henri Darcy carried out some experiments from which he derived a method for estimating the rate of groundwater flow based on the hydraulic gradient and the permeability of an aquifer, expressed using *K*, the hydraulic conductivity. Darcy's equation, which has been used widely by hydrogeologists ever since, looks like this:

$$V = K * i$$

(where *V* is the velocity of the groundwater flow, *K* is the hydraulic conductivity, and *i* is the hydraulic gradient).

We can apply this equation to the scenario in Figure 14.2.1. If we assume that the permeability is 0.00001 meters per second we get: *V* = 0.00001 * 0.08 = 0.0000008 meters per second. That is equivalent to 0.000048 meters per minute, 0.0029 meters/hour or 0.069 m/day. That means it would take 1,450 days (nearly four years) for water to travel the 100 meters from the vicinity of the well to the stream. Groundwater moves slowly, and that is a reasonable amount of time for water to move that distance. In fact it would likely take longer than that, because it doesn't travel in a straight line.

Exercise 14.1 How long will it take?

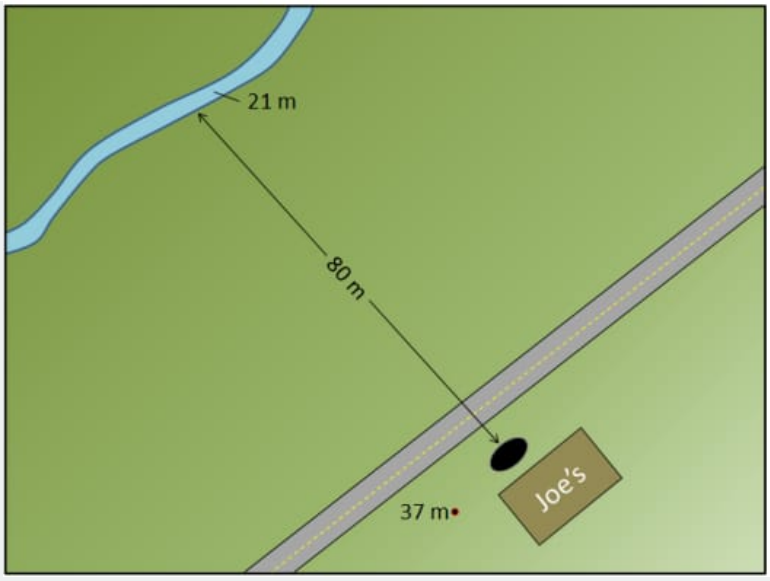


Figure 14.2.4

Sue, the owner of Joe's 24-Hour Gas, has discovered that her underground storage tank (UST) is leaking fuel. She calls in a hydrogeologist to find out how long it might take for the fuel contamination to reach the nearest stream. They discover that the well at Joe's has a water level that is 37 meters above sea level and the elevation of the stream is 21 meters above sea level. The sandy sediment in this area has a permeability of 0.0002 meters per second.

Using *V* = *K* * *i*, estimate the velocity of groundwater flow from Joe's to the stream, and determine how long it might take for contaminated groundwater to flow the 80 meters to the stream.

See Appendix 3 for Exercise 14.1 answers.

It's critical to understand that groundwater does not flow in underground streams, nor does it form underground lakes. With the exception of **karst** areas, with caves in limestone, groundwater flows very slowly through granular sediments, or through solid rock that has fractures in it. Flow velocities of several centimeters per day are possible in significantly permeable sediments with significant hydraulic gradients. But in many cases, permeabilities are lower than the ones we've used as examples here, and in many areas, gradients are much lower. It is not uncommon for groundwater to flow at velocities of a few millimeters to a few centimeters per year.

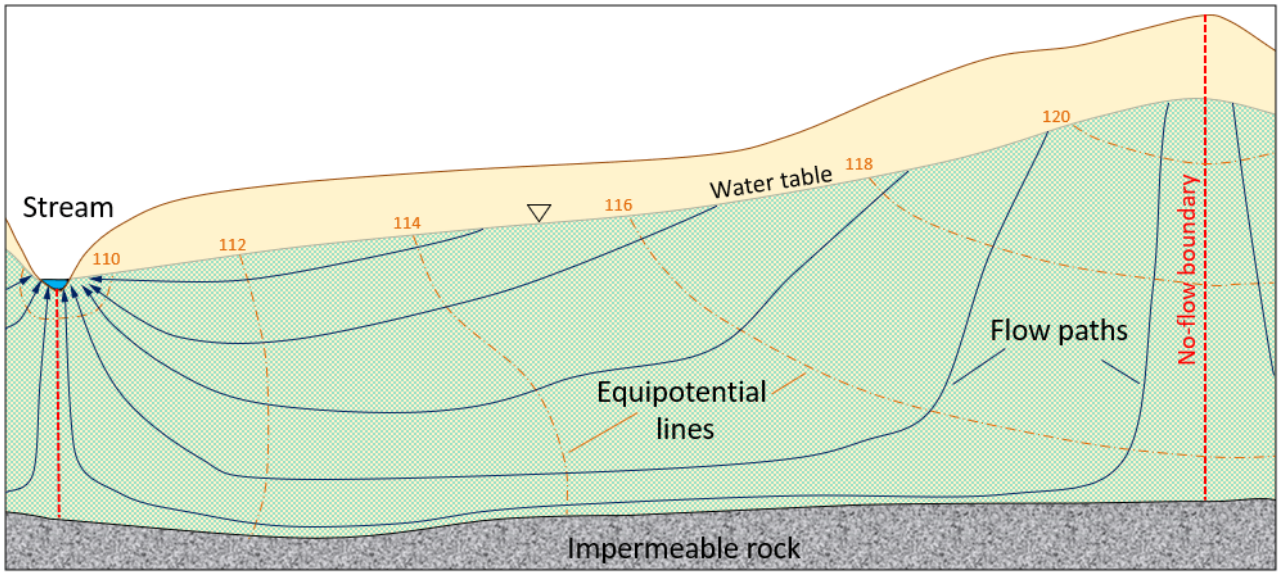


Figure 14.2.5 Predicted equipotential lines (orange) and groundwater flow paths (blue) in an unconfined aquifer. The orange numbers are the elevations of the water table at the locations shown, and therefore they represent the pressure along the equipotential lines.

As already noted, groundwater does not flow in straight lines. It flows from areas of higher hydraulic head to areas of lower hydraulic head, and this means that it can flow "uphill" in many situations. This is illustrated in Figure 14.2.5. The dashed orange lines are **equipotential lines,** meaning lines of equal pressure. The blue lines are the predicted groundwater **flow paths**. Note that in some areas the groundwater will flow steeply down, in other areas it will flow down only at a shallow angle, and in some areas—especially near to discharge zones—it will flow upward. The dashed red lines are no-flow boundaries, meaning that water cannot flow across these lines. That's not because there is something there to stop it, but because there's no pressure gradient that will cause water to flow in that direction.

Groundwater flows at right angles to the equipotential lines in the same way that water flowing down a slope would flow at right angles to the contour lines. The stream in this scenario is the location with the lowest hydraulic potential, so the groundwater that flows to the lower parts of the aquifer has to flow upward to reach this location. It is forced upward by the pressure differences, for example, the difference between the 112 and 110 equipotential lines.

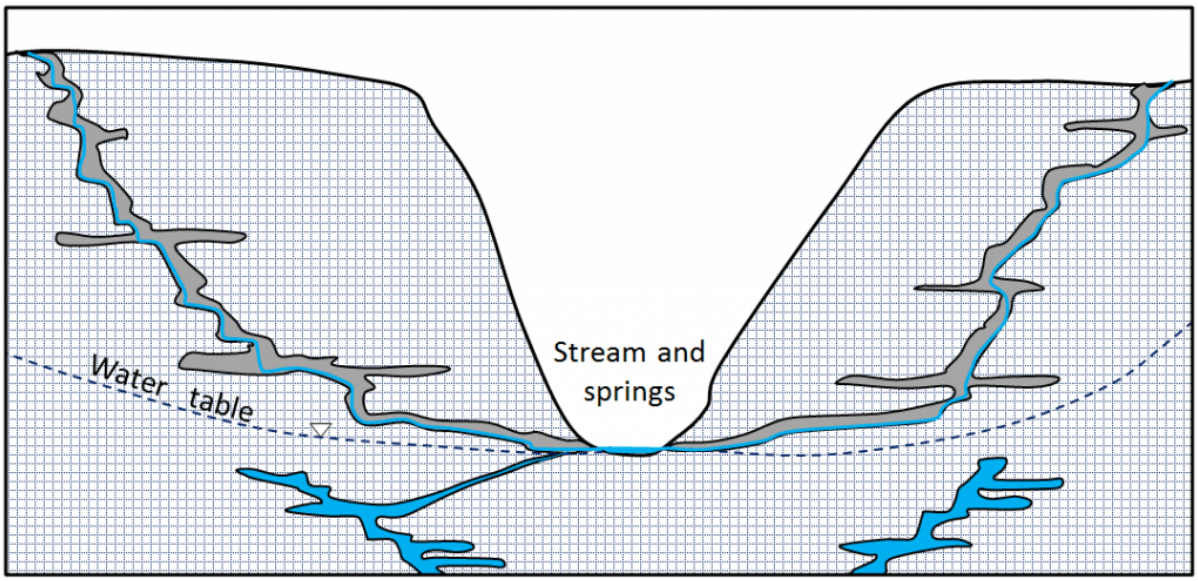


Figure 14.2.6 Groundwater in a limestone karst region. The water in the caves above the water table does not behave like true groundwater because its flow is not controlled by water pressure, only by gravity. The water below the water table does behave like true groundwater.

Groundwater that flows through caves, including those in **karst** areas—where caves have been formed in limestone because of dissolution—behaves differently from groundwater in other situations. Caves above the water table are air-filled conduits, and the

water that flows within these conduits is not under pressure; it responds only to gravity. In other words, it flows downhill along the gradient of the cave floor (Figure 14.2.6). Many limestone caves also extend below the water table and into the saturated zone. Here water behaves in a similar way to any other groundwater, and it flows according to the hydraulic gradient and Darcy's law.

Image Descriptions

Figure 14.2.2 image description: A cross-section showing three layers. The top layer is an unconfined aquifer. A well has been drilled into the unconfined aquifer and filled up to the height of the water table. Below the unconfined aquifer is a confining layer, and below that is a confined aquifer. A well has been drilled through the confining layer and into the confined aquifer. Water fills that well up to the potentiometric surface of the confined aquifer, which, in this case, is above the confining layer. [Return to Figure 14.2.2]

Media Attributions

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14.3: Groundwater Extraction

Except in areas where groundwater comes naturally to the surface at a **spring** (a place where the water table intersects the ground surface), we have to construct wells in order to extract it. If the water table is relatively close to the surface, a well can be dug by hand or with an excavator, but in most cases we need to use a drill to go down deep enough. There are many types of drills that can be used; an example is shown in Figure 14.3.1. A well has to be drilled at least as deep as the water table, but in fact must go much deeper; first, because the water table may change from season to season and from year to year, and second, because when water is being pumped, the water level will drop, at least temporarily.



Figure 14.3.1 A water-well drilling rig in operation near Nanaimo, B.C. In the photo on the right the well is being test-pumped with air pressure. The casing (yellow arrow) is about 40 centimeters in diameter.

Where a well is drilled in unconsolidated sediments or relatively weak rock, it has to be lined with casing (steel pipe in most cases) in order to ensure that it doesn't cave in. A specially designed well screen is installed at the bottom of the casing. The size of the holes in the screen is carefully chosen to make sure that it allows the water to move into the well freely, but prevents aquifer particles from entering the well. A submersible pump is typically used to lift water from within the well up to the surface where it is needed. The well shown in Figure 14.3.1 has casing that is about 40 centimeters in diameter, which might be typical for a municipal water supply well, or a very large well for irrigation. Most domestic wells have 15 centimetre casing.

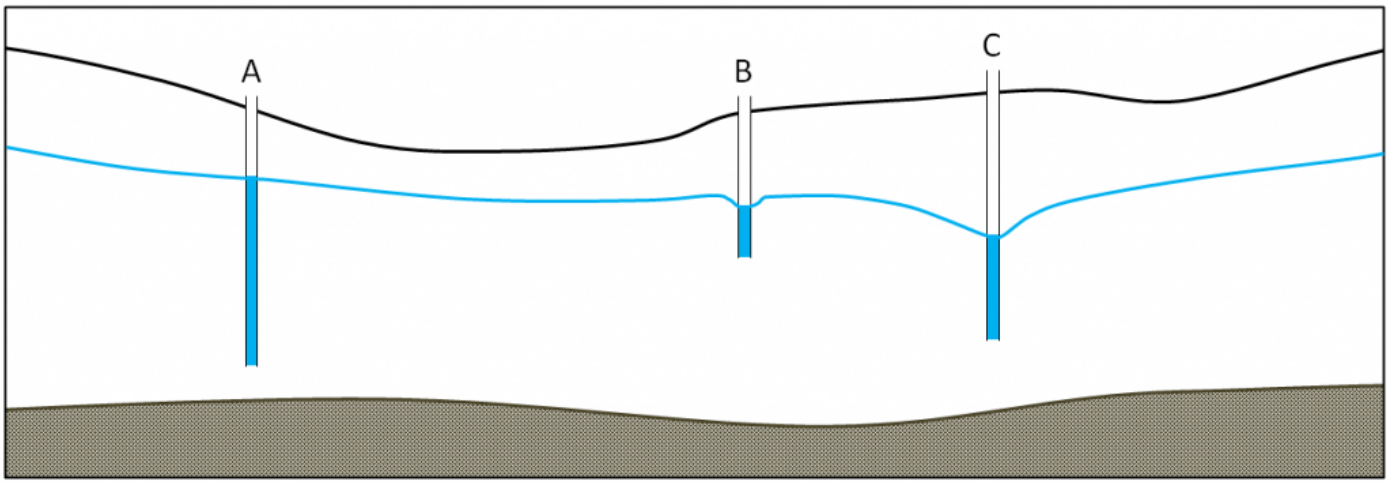


Figure 14.3.2 Three wells in an unconfined aquifer. Well A is not being pumped. Well B is being pumped at a slow rate and well C, which has a larger cone of depression, is being pumped at a faster rate.

Pumping water from the well removes water from inside the well at first. That lowers the water level inside the well. This means that water will flow from the surrounding aquifer toward the pumping well where the water level is now lower.

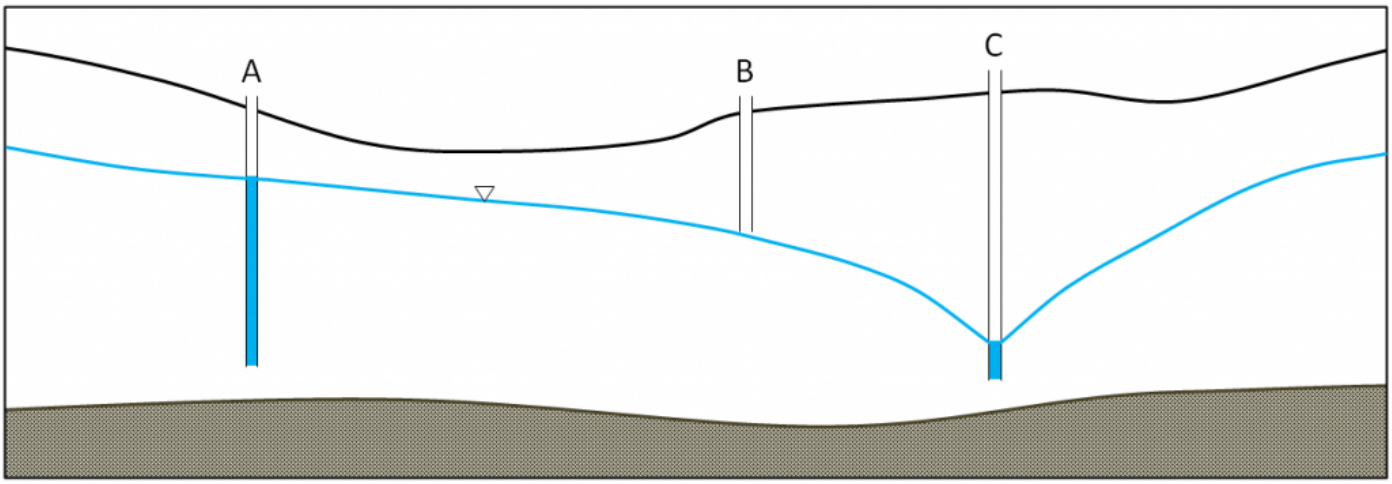


Figure 14.3.3 A similar scenario to that in Figure 14.3.2, but in this case, well C has been pumped unsustainably for a long time. The cone of depression from well C has reached well B and has contributed to it going dry because the water level is now below the bottom of well B.

That is how a well gets water from the ground. The water table, or potentiometric surface, will slope in toward the well where the water is being withdrawn. That indicates the energy gradient that is allowing water to flow toward the well. This creates a shape known as a **cone of depression** surrounding the well, as illustrated in Figure 14.3.2. If pumping from a well continues for hours to days, the cone of depression may result in a loss of water in nearby wells. As shown in Figure 14.3.3, pumping of well C has contributed to well B going dry because the water level is now below the bottom of well B. If pumping continues in well C, it too may go dry.

Exercise 14.2 Cone of depression

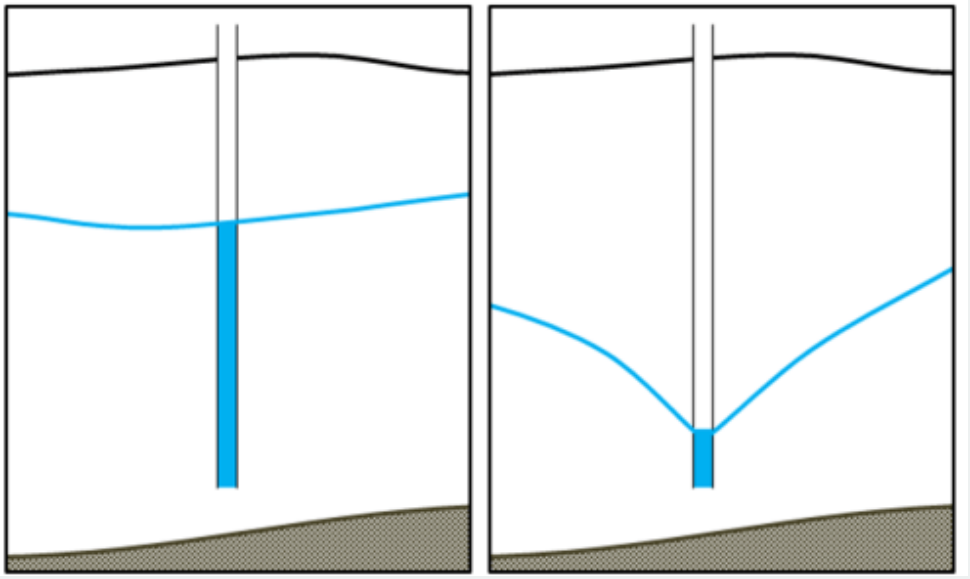


Figure 14.3.4

The two diagrams here show the same well before (left) and after (right) long-term pumping. A cone of depression has developed. This provides the energy gradient for water to flow toward the well so that it can be pumped out.

How will this likely affect the rate of flow into the well?

See Appendix 3 for Exercise 14.2 answers.

Like other provinces in Canada and states in the US, British Columbia has a network of observation wells administered by the Ministry of the Environment. These are wells that are installed to measure water levels; they are not pumped. There are 145 active observation wells in B.C. (in 2015), most equipped with automatic recorders that monitor water levels continuously. The main purpose of the observation wells is to monitor water table levels so that we can see if there are long-term natural fluctuations in groundwater quantity, and shorter-term fluctuations related to overuse of the resource. Most are also sampled regularly to monitor groundwater chemistry and quality.

An example of an observation well is illustrated in Figure 14.3.5. This one is situated at Cassidy on Vancouver Island and is used to monitor an unconsolidated aquifer that is widely used by residents with private wells.



Figure 14.3.5 B.C. observation well 228 near Cassidy, Vancouver Island. The installation also has a solar panel, which is not visible in this view.

The water-level data from B.C.'s observation wells are available to the public, and an example data set is illustrated in Figure 14.3.6 . The water level in Ministry of Environment observation well 232 (OW-232), situated in Lantzville on Vancouver Island, dropped significantly from 1979 (average depth around 1.5 meters), to 2003 (average depth around 5.5 meters), but has recovered a little since then.

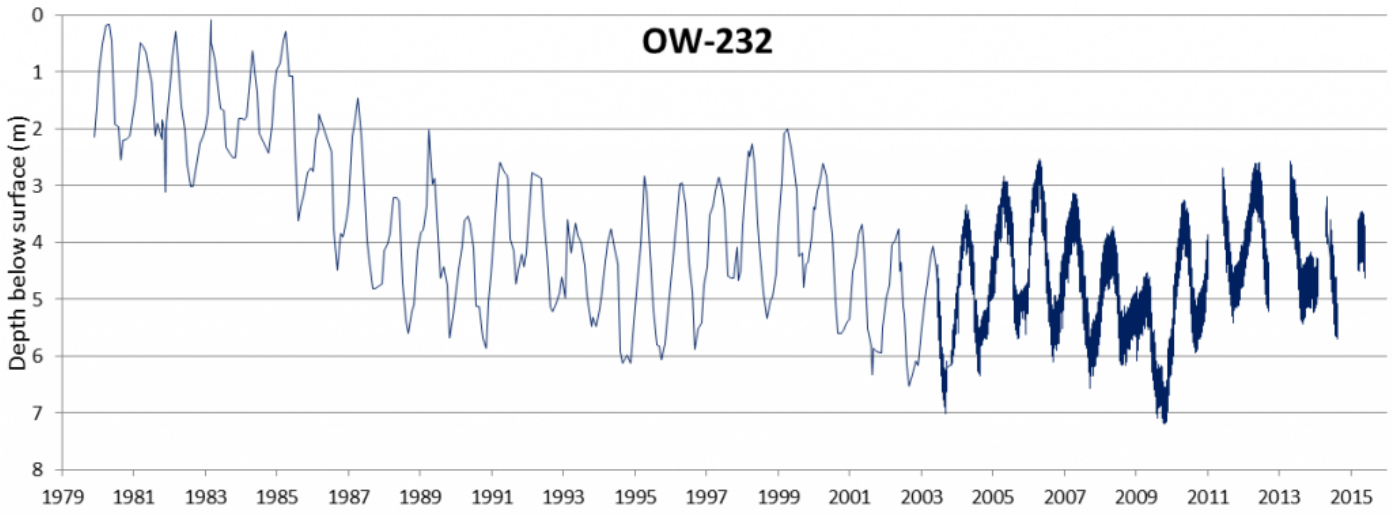


Figure 14.3.6 Water level data for B.C. observation well 232 on Harby Rd., Lantzville, Vancouver Island. From 1979 to 2003, depths were recorded monthly. Automated equipment was installed in 2003, and the depths were recorded hourly since that time.

The short-term variations in the level of well 232 are at a period of one year and are related to annual cycles of recharge and discharge governed by the wet winter climate and drier summers. The data for part of the period are shown in more detail in Figure 14.3.7. In most of British Columbia most wells drop to their lowest levels in September or October after the relatively dry

summer period. Levels increase rapidly from October through February as high winter precipitation adds recharge to the aquifer, and water is stored. The water table reaches a peak in March or April. Most wells then drop over the summer as groundwater continues to flow, but no new recharge is added. The water is drained from storage into streams or lakes and eventually into the ocean, and as a result, the water table decreases, reaching its lowest level again in September or October. Similar fluctuations are observed at most observation wells around the province, although the timing is slightly different from region to region.

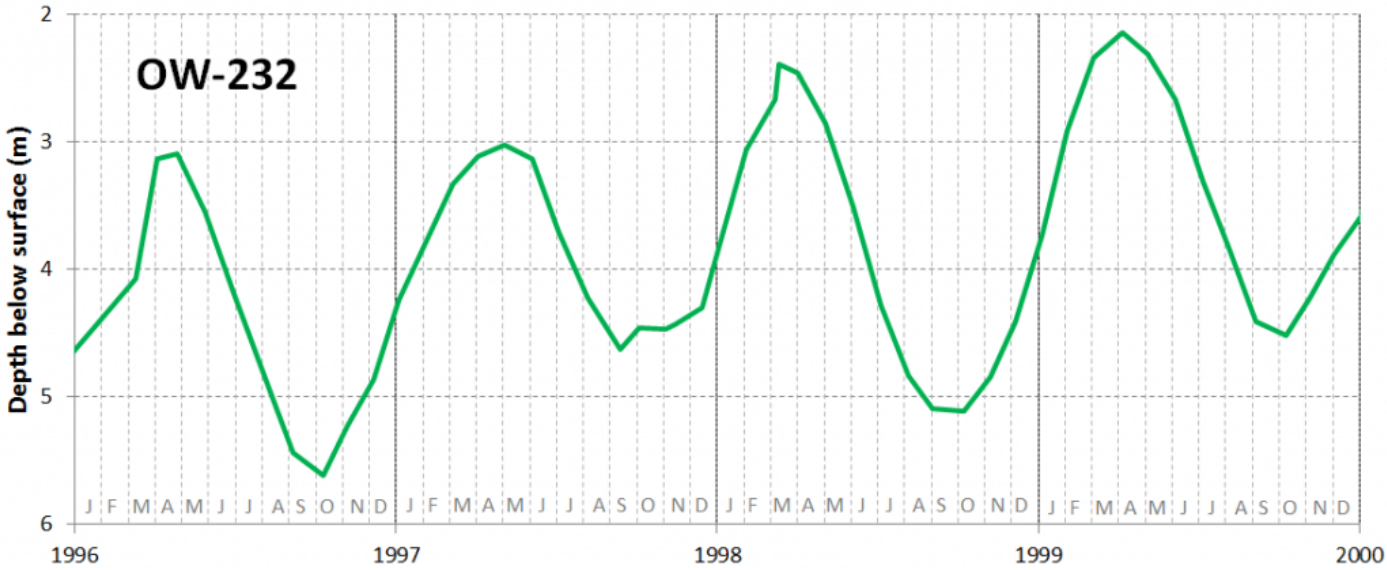


Figure 14.3.7 Water level data for B.C. observation well 232 for the period 1996 to 2000, showing seasonal variations.

The long-term fluctuations in levels in observation wells in different places are also quite variable. Long-term changes in climate can lead to gradual natural changes in water levels. These long-term cycles lasting years or decades are mixed with the effects of well pumping. Some observation wells show consistent decreases in water level that may indicate long-term over-extraction. Many others show generally consistent levels over several decades, and some show increases in water level. One of the important jobs performed by hydrogeologists working for government agencies is to examine these long-term records of water levels for indications of how sustainable the groundwater use might be.

Exercise 14.3 What's your water table doing?

Visit the B.C. Ministry of the Environment observation well website and use the map to find an observation well near you. When you click on a point, a window pops up with a link that says, "Click for details about this well." Click on that link and then choose one of the options available. The "Graphs" tab will show you a graph of the water levels, and you should be able to tell if the level is generally increasing or decreasing. If there isn't much data to see, choose a different well.

(Similar data are available for Alberta and Saskatchewan and other jurisdictions as well. Search using terms like "Alberta observation wells.")

See Appendix 3 for Exercise 14.3 answers.

In 2016 the new Water Sustainability Act came into effect in British Columbia, requiring licensing groundwater extraction for the first time. The new Act also includes provisions for determining "environmental flow needs"—the amount of water that must be in surface water streams at different times of the year to meet the needs of the ecosystem that depends on the streamflow. For example, many streams in B.C. support populations of salmon that live in the stream for part of their life cycle or return to their home stream for spawning. Groundwater forms a part of the baseflow in a watershed, and is therefore an important part of the environmental flow needs. Careful work is needed in the coming years to ensure that the amount of water licensed to be extracted from surface water and groundwater for human use does not interfere with the amount of water needed for the natural water-dependent ecosystems to function.

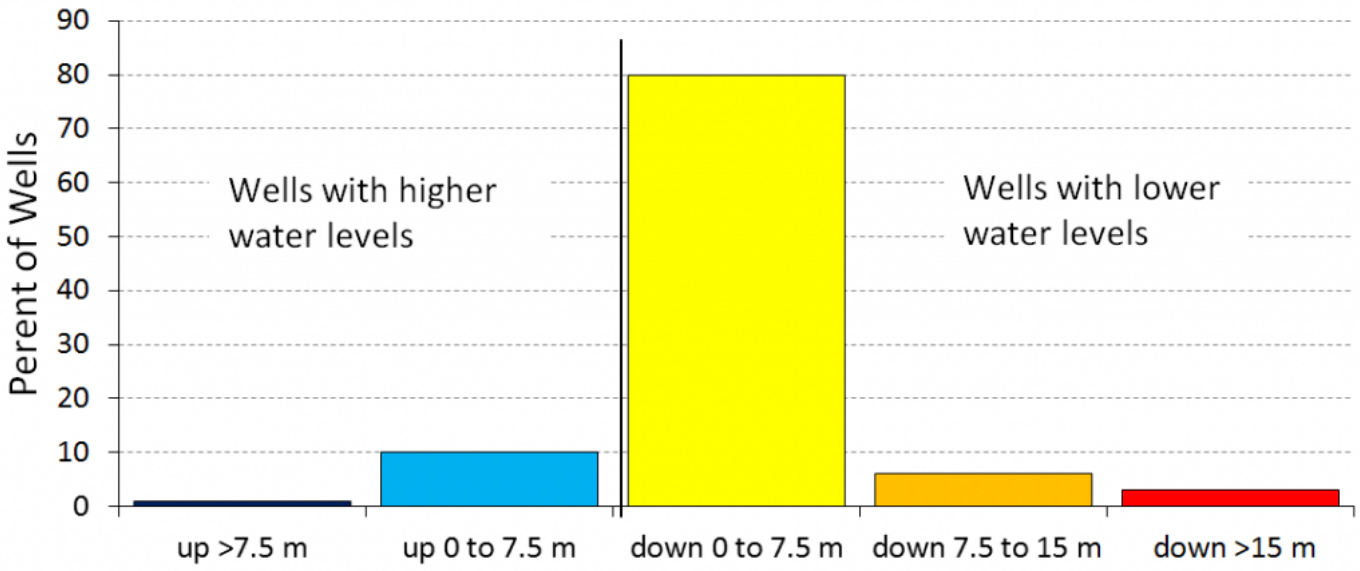


Figure 14.3.8 Changes in water levels in wells in California from 2011 to 2013.

The situation in California, where groundwater extraction over large areas is leading to declining water levels, is quite different from that in B.C. According to the state Department of Water Resources, 80% of groundwater wells showed drops in water level of 0 meters to 7.5 meters between 2011 and 2013, another 6% dropped by 7.5 meters to 15 meters, and 3% dropped by more than 15 meters (Figure 14.3.8). Over the same time period, only 10% of well levels increased by 0 meters to 7.5 meters, and 1% increased by more than 7.5 meters. The drought that gripped California in 2013 had worsened significantly by 2015, and California farmers — and the people across North America that eat the food they produce — continue to have a prodigious appetite for irrigation water. California, like B.C., is introducing new groundwater regulations to try to control water usage and halt water table declines.

Impermeable Surfaces

Even if groundwater supplies are not being depleted by overuse, or by a changing climate, we are continuing to put stress on aquifers by covering vast areas with impermeable surfaces that don't allow rain and snow-melt to infiltrate and become groundwater. Instead, water that falls on these surfaces is channeled into drainage systems, then into storm sewers, and then directly into rivers and the ocean. In cities and their suburbs, the biggest culprits are parking lots, roads, and highways. While it would great if we didn't dedicate such huge areas to cars, that's not about to change quickly, so we need to think about ways that we can improve surface water infiltration in cities. One way is to use road and parking surfaces that will allow water to seep through, although this is not practical in many cases. Another way is to ensure that runoff from pavement is channeled into existing or constructed wetlands that serve to decontaminate the water, and then allow it to infiltrate into the ground.

Image Descriptions

Figure 14.3.5 image description: Inside a box attached to the well casing, there is a battery, a depth sensor wheel, a spare float, and an instrument that provides a depth readout. This one says 4.047 meters. There is also a communication antenna. [Return to Figure 14.3.5]

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- Figure 14.3.7: © Steven Earle. CC BY. Based on data from "Observation Well #232."
- Figure 14.3.8:© Steven Earle. CC BY. Based on State of California Department of Water Resources data.

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14.4: Groundwater Quality

As was noted at the very beginning of this chapter, one of the good things about groundwater as a source of water is that it is not as easily contaminated as surface water is. But there are two caveats to that: one is that groundwater can become naturally contaminated because of its very close connection to the materials of its aquifer, and the second is that once contaminated by human activities on surface, groundwater is very difficult to clean up.

Natural Contamination of Groundwater

Groundwater moves slowly through an aquifer, and unlike the surface water of a stream, it has a lot of contact with the surrounding rock or sediment. In most aquifers, the geological materials that make up the aquifer are relatively inert, or are made up of minerals that dissolve very slowly into the groundwater. Over time, however, all groundwater gradually has more and more material dissolved within it as it remains in contact with the aquifer. In some areas, that rock or sediment includes some minerals that could potentially contaminate the water with elements that might make the water less than ideal for human consumption or agricultural use. Examples include copper, arsenic, mercury, fluorine, sodium, and boron. In some cases, contamination may occur because the aquifer material has particularly high levels of the element in question. In other cases, the aquifer material is just normal rock or sediment, but some particular feature of the water or the aquifer allows the contaminant to build up to significant levels.

An example of natural contamination takes place in the bedrock aquifers of the east coast of Vancouver Island and the adjacent Gulf Islands. The aquifer is the Cretaceous (90 Ma to 65 Ma) Nanaimo Group, which is made up of sandstone, mudstone, and conglomerate (Figure 14.4.1).



Figure 14.4.1 Cretaceous Nanaimo Group sandstone exposed in a Nanaimo parking lot.

The rocks of the Nanaimo Group are not particularly enriched in any trace elements, but the submarine-fan sandstone that makes up much of the group has relatively high levels of clay (for a sandstone). This clay is good at adsorbing[1] some elements from the water and desorbing others, and in the process, its pH goes up (it becomes alkaline). At high pH levels (some as high as 9 in groundwater in the Nanaimo Group), the element fluorine that is naturally present in the rock (as it is in almost any rock) has an increased tendency to dissolve in the water. In some areas, groundwater in the Nanaimo Group has fluorine levels that are well above recommended levels for drinking water. The World Health Organization (WHO) maximum acceptable concentration (MAC) for fluorine is 1.5 milligrams per litre (milligrams per litre). Between 5% and 10% of the domestic wells around Nanaimo and adjacent Gabriola Island have more than that, some as much as 10 milligrams per litre. A small amount of fluorine in the human diet is considered important for maintaining dental health, but high levels can lead to malformation and discoloration of teeth, and long-term exposure can lead to other more serious health effects such as skeletal problems.

Nanaimo Group groundwater can also have elevated levels of boron, again related to pH and adsorption from clay minerals. While

boron at the levels found there is not toxic to humans, there is enough boron in some wells to be toxic to plants, and the water cannot be used for irrigation.

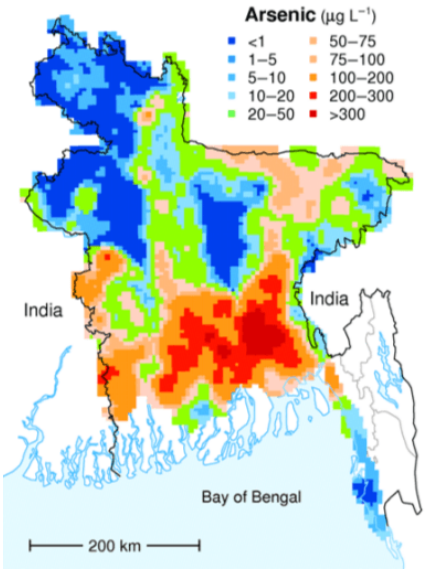


Figure 14.4.2 The distribution of arsenic in groundwater in Bangladesh. The WHO recommended safe level for arsenic is 10 µg/L. All of the green, orange, and red areas on the map exceed that limit.

Rural residents in the densely populated country of Bangladesh (over 1,000 residents per square kilometre, compared with 3.4 per square kilometre in Canada) used to rely mostly on surface supplies for their drinking water, and many of these were subject to bacterial contamination. Infant mortality rates were among the highest in the world and other illnesses such as diarrhea, dysentery, typhoid, cholera, and hepatitis were common. In the 1970s, international agencies, including UNICEF, started a program of drilling wells to access abundant groundwater supplies at depths of 20 meters to 100 meters. Eventually over 8 million such wells were drilled. Infant mortality and illness rates dropped dramatically, but it was later discovered that the water from a high proportion of these wells has arsenic above safe levels (Figure 14.4.2).

Most of the wells in the affected areas are drilled into relatively recent sediments of the vast delta of the Ganges and Brahmaputra Rivers. While these sediments are not particularly enriched in arsenic, they have enough organic matter in them to use up any oxygen present. This leads to water with a naturally low oxidation potential (anoxic conditions); arsenic is highly soluble under these conditions, and so any arsenic present in the sediments easily gets dissolved into the groundwater. Arsenic poisoning leads to headaches, confusion, and diarrhea, and eventually to vomiting, stomach pain, and convulsions. If not treated, the final outcomes are heart disease, stroke, cancer, diabetes, coma, and death. There are ways to treat arsenic-rich groundwater, but it is a challenge in Bangladesh to implement the simple and effective technology that is available.

Anthropogenic Contamination of Groundwater

Groundwater can become contaminated by pollution at the surface (or at depth), and there are many different anthropogenic (human-caused) sources of contamination.

The vulnerability of aquifers to pollution depends on several factors, including the depth to the water table, the permeability of the material between the surface and the aquifer, the permeability of the aquifer, the slope of the surface, and the amount of precipitation. Confined aquifers tend to be much less vulnerable than unconfined ones, and deeper aquifers are less vulnerable than shallow ones. Steeper slopes mean that surface water tends to run off rather than infiltrate (and this can reduce the possibility of contamination). Contamination risk is also less in dry areas than in areas with heavy rainfall.

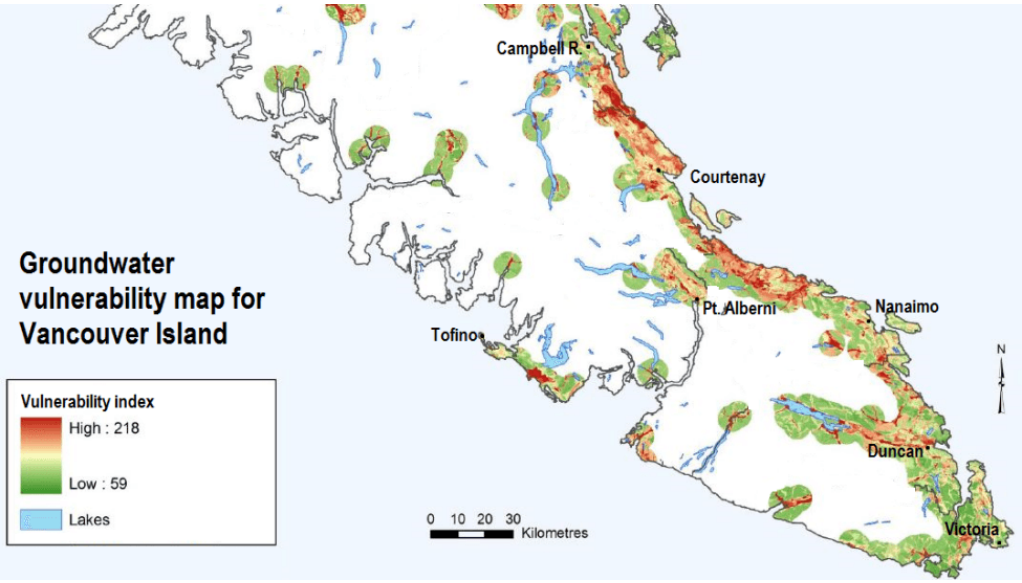


Figure 14.4.3 The vulnerability to anthropogenic contamination of aquifers on southern Vancouver Island. Much of the island is not mapped (shown as white) because of a lack of aquifer information in areas without wells.

Studies of groundwater vulnerability have been completed for various regions of British Columbia. A groundwater vulnerability map for southern Vancouver Island is shown in Figure 14.4.3. The yellow to red areas are considered to have high vulnerability to pollution from surface sources, and most of these are where the aquifers are unconfined in quite permeable unconsolidated sediments of either glacial or fluvial origin, where the water table is relatively shallow and the terrain is relatively flat.

The important sources of anthropogenic groundwater contamination include the following:

- Chemicals and animal waste related to agriculture, and chemicals applied to golf courses and domestic gardens
- Landfills
- Industrial operations
- Mines, quarries, and other rock excavations
- Leaking fuel storage tanks (especially those at gas stations)
- Septic systems
- Runoff from roads (e.g., winter salting) or chemical spills of materials being transported

Agriculture

Intensive agricultural operations and golf courses can have a significant impact on the environment, especially where chemicals and other materials are used to enhance growth or control pests. An example of agricultural contamination is in the Abbotsford area of the Fraser Valley, where nitrate levels above the 44 milligrams per litre maximum acceptable level (expressed as nitrate) in the Abbotsford-Sumas aquifer have been observed since the 1950s; however, the problem became much worse as agriculture intensity increased in the 1980s. By 2004, groundwater with nitrate levels in excess of 44 milligrams per litre was reported over an area of about 75 kilometers square around Abbotsford, and the problem extended across the border into the Sumas area of Washington State.

This region is intensively used for berry crops (especially raspberries and blueberries) and large poultry operations, as well as lesser amounts of grazing and forage crops. Chicken manure is typically stored in fields adjacent to chicken barns, and may release nitrogen to the environment from runoff water, and from releases of ammonia gas. Over decades, both chemical fertilizers and chicken manure and other manures have been applied to the berry crops to provide extra nitrogen to help maximize berry growth. If the fertilizer added is in excess of what the plants need, or is poorly timed compared to when it is needed, then the extra nitrogen may be leached into the groundwater below. Berry crops are irrigated over the summer to help the crops grow. Summer irrigation and winter rainfall may carry excess nitrate from the near surface to the aquifer below.

Since the 1990s, agricultural practices have been tightened up to reduce the rate of groundwater contamination, but it will take decades for nitrate levels to drop in the Abbotsford-Sumas aquifer. Agriculture and Agri-Food Canada and many others are conducting research on better irrigation and nitrate management techniques to reduce the amount of nitrogen that leaches to

groundwater.

Landfills

In the past, domestic and commercial refuse was commonly trucked to a “dump” (typically a hole in the ground), and when the hole was filled, it was covered with soil and forgotten. In situations like this, rain and melting snow can easily pass through the soil used to cover the refuse. This water passes into the waste itself, and the resulting landfill **leachate** that flows from the bottom of the landfill can seriously contaminate the surrounding groundwater and surface water. In the past few decades, regulations around refuse disposal have been significantly strengthened, and important steps have been taken to reduce the amount of landfill waste by diverting recyclable and compostable materials to other locations.

A modern engineered landfill has an impermeable liner (typically heavy plastic, although engineered clay liners or natural clay may be adequate in some cases), a plumbing system for draining leachate (the rainwater that flows through the refuse and becomes contaminated), and a network of monitoring wells both within and around the landfill (Figure 14.4.4). Once part or all of a landfill site is full, it is sealed over with a plastic cover, and a system is put in place to extract **landfill gas** (typically a mixture of carbon dioxide and methane). That gas can be sent to a nearby location where it is burned to create heat or used to generate electricity. The leachate must be treated, and that can be done in a normal sewage treatment plant.

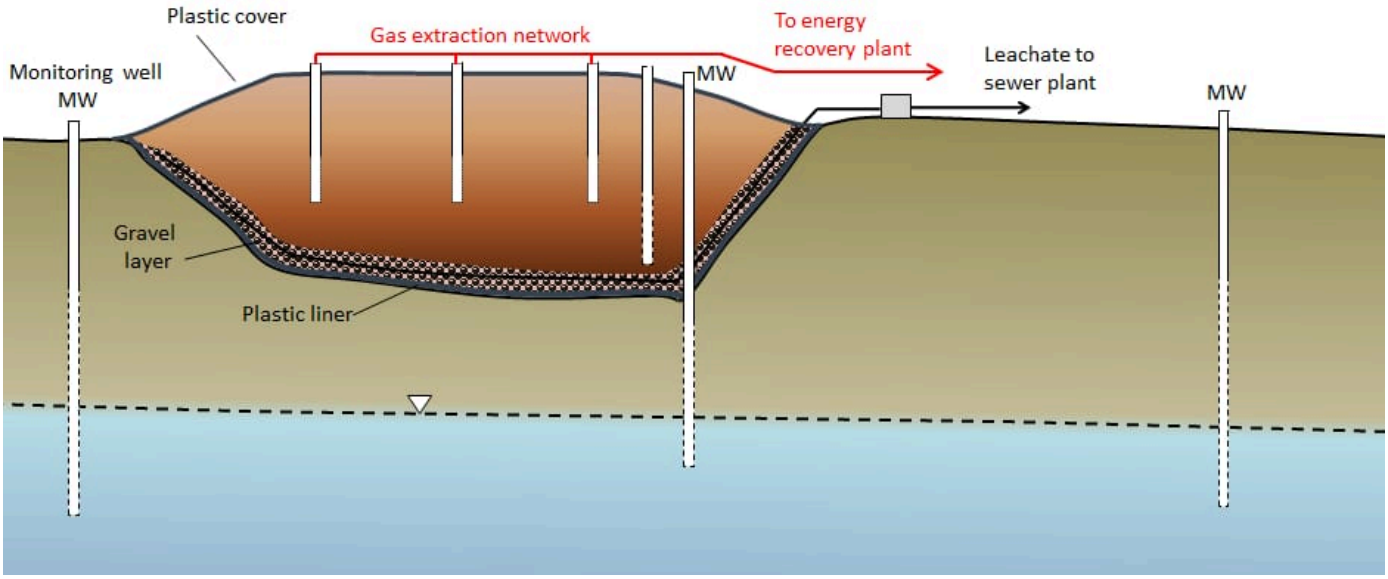


Figure 14.4.4 A cross-section of a typical modern landfill.

The monitoring wells are used to assess the level of the water table around the landfill and to collect groundwater samples so that any leakage can be detected. Because some leakage is almost inevitable, the ideal placement for landfills is in areas where the depth to the water table is significant (tens of meters if possible) and where the aquifer material is relatively impermeable. Landfills should also be situated far from streams, lakes, or wetlands so that contamination of aquatic habitats can be avoided.

Today there are hundreds of abandoned dumps scattered across the country; most have been left to contaminate groundwater that we might wish to use sometime in the future. In many cases, it’s unlikely that we’ll be able to do so.

Exercise 14.4 What goes on at your landfill?



Figure 14.4.5

Unless you live in a remote rural area, there's a good chance that the refuse you can't recycle is picked up at the curb and taken to a landfill. Most landfills are operated by cities or districts, and you should be able to find information about yours on the appropriate local government website. See if you can answer some the following questions:

1. Which government body operates your landfill?
2. Where is the landfill situated?
3. Is your waste all placed in a landfill, or are there other processes in use (e.g., incineration or composting)?
4. Are landfill gases captured, and, if so, what is done with them?
5. What could be changed to improve the waste disposal situation in your community (e.g., more recycling, compost collection, waste-to-energy technology)?

See Appendix 3 for Exercise 14.4 answers.

Industrial Operations

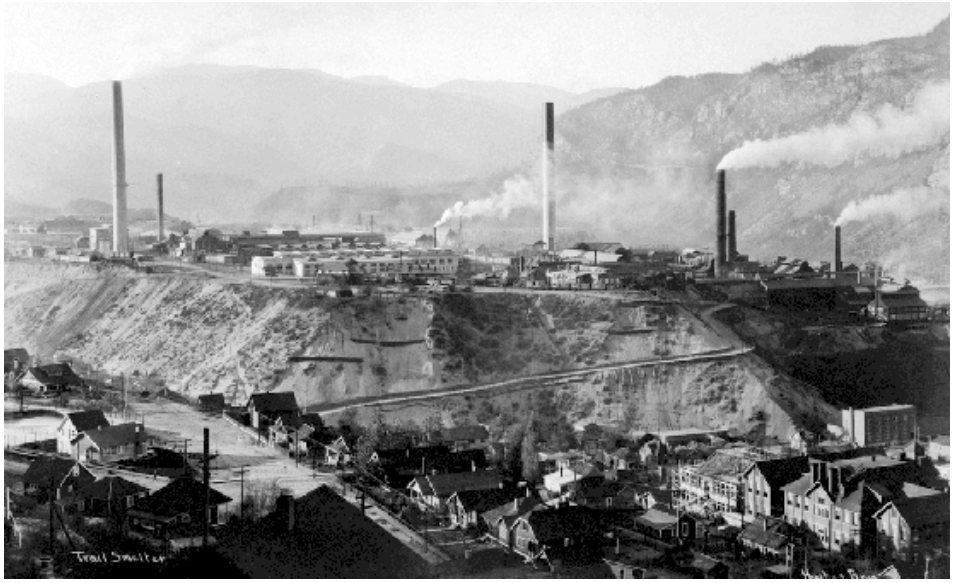


Figure 14.4.6 The Trail lead-zinc smelter in 1929.

Although western Canada doesn't have the same extent of industrial pollution as other parts of the country, there are still seriously contaminated sites in the west, most with the potential to contaminate groundwater. One example is the lead and zinc smelter at Trail, B.C. The largest in the world, it has been operating for over 100 years and has left a residue of metal contamination around the region (Figure 14.4.6). In some parts of Trail, the contamination is serious enough that existing soil has been removed

from residential properties and replaced with clean soil brought in from elsewhere. This contaminated soil has contributed to contamination of groundwater in the Trail area. Groundwater beneath the actual smelter site is contaminated, and the operator (Teck Resources) is currently working on plans to prevent that water from reaching the nearby Columbia River.

Mines, Quarries, and Rock Excavations

Mines and other operations that involve the excavation of large amounts of rock (e.g., highway construction) have the potential to create serious environmental damage. The exposure of rock that has previously not been exposed to air and water can lead to the oxidation of sulfide-bearing minerals, such a pyrite (FeS2), within the rock. The combination of pyrite, water, oxygen, and a special type of bacteria (*Acidithiobacillus ferrooxidans*) that thrives in acidic conditions leads to the generation of acidity, in some cases to pH less than 2. Water that acidic is hazardous by itself, but the low pH also has the property of increasing the solubility of certain heavy metals. The water that is generated by this process is known as acid rock drainage (ARD). ARD can occur naturally where sulfide-bearing rocks are near the surface.



Figure 14.4.7 Acidic runoff at the abandoned Mt. Washington Mine near Courtenay, B.CThe issue of ARD is a major environmental concern at both operating mines and abandoned mines (see Chapter 20). In streams around the Mt. Washington Mine on Vancouver Island (Figure 14.4.7), copper levels are high enough to be toxic to fish. Groundwater adjacent to the contaminated streams in the area is very likely contaminated as well.

Leaking Fuel Tanks



Figure 14.4.8 A closed and fenced gas station site in Nanaimo, B.C. The white pipes in the background are wells for monitoring groundwater contamination on the site.

Underground storage tanks (USTs) are used to store fuel at gas stations, industrial sites, airports, and anywhere that large volumes of fuel are used. They do not last forever, and eventually they start to leak their contents into the ground. This is a particular problem at older gas stations — although it may also become a future problem at newer gas stations. You may have noticed gas stations that have been closed and then surrounded by chain-link fence (Figure 14.4.8). In virtually all such cases the closure has been triggered by the discovery of leaking USTs and the requirement to cease operations and remediate the site.

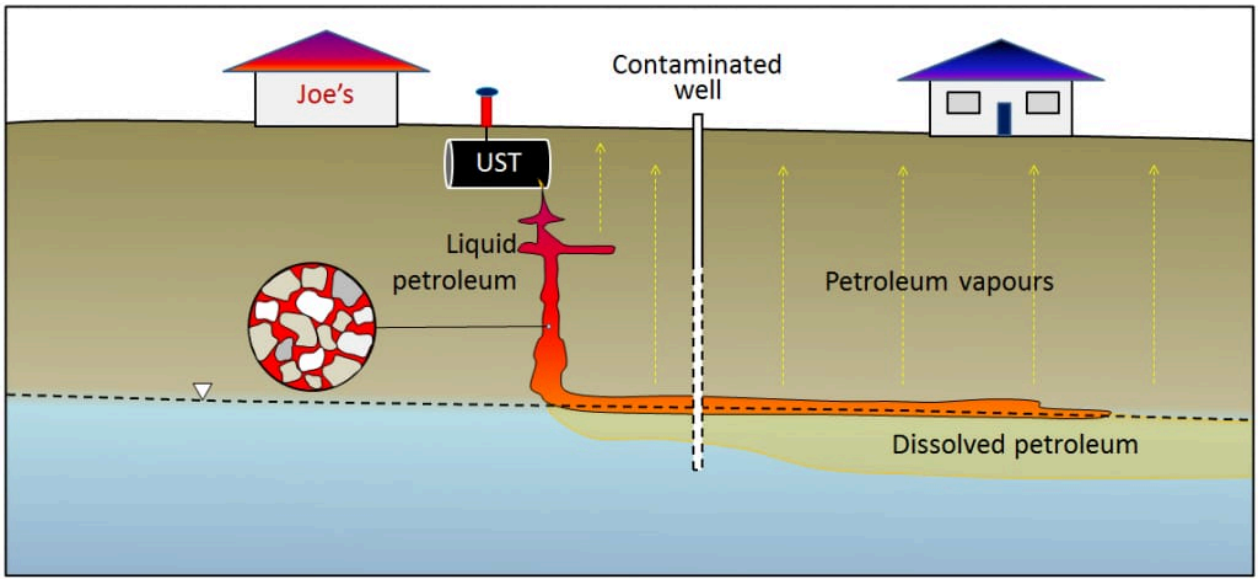


Figure 14.4.9 A depiction of the fate of different components of a petroleum spill from an underground storage tank.

Petroleum fuels are complex mixtures of hydrocarbon compounds and the properties of their components—such as density, viscosity, solubility in water, and volatility—tend to vary widely. As a result, a petroleum spill is like several spills for the price of one. The petroleum liquid slowly settles through the unsaturated zone and then tends to float on the surface of the groundwater (Figure 14.4.9). The more readily soluble components of the spill dissolve in the groundwater and are dispersed along with the normal groundwater flow, and the more volatile components of the spill rise toward the surface, potentially contaminating buildings.

Exercise 14.5 Find a leaking UST in your community

There is almost certainly a leaking UST at a former gas station near you. Look for an empty property that is surrounded by a chain-link fence with “No Trespassing” signs. You might see evidence of monitoring wells (like those shown in Figure 14.4.9, and there could be some petroleum barrels around that are being used to store contaminated water. Once you’ve identified one of these, you’ll probably start seeing them everywhere!

See Appendix 3 for Exercise 14.5 answers.

Septic Systems

In areas that are not served by sewage networks leading to a central sewage treatment plant, most homeowners rely on **septic systems** for disposal of sewage. There are two primary components to a simple septic system, the septic tank and the drainage field (Figure 14.4.10). A typical septic tank is constructed of either concrete or plastic and has a volume of 5,000 litres to 10,000 litres (5 m^3 to 10 m^3). This forms the first treatment and is designed to be **anaerobic** (without oxygen). That promotes the activity of certain bacteria that help break down the waste. As the waste is degraded, some portions tend to sink to form sludge at the base of the tank, and others float to the surface, forming a scum layer. A septic tank may be divided into two parts to keep the sludge at the bottom and the scum on the top from draining out. The water then moves to the drainage field, which provides the right conditions for a different set of bacteria that operate in **aerobic** conditions. The drainage field includes an array of plastic pipes that are perforated to allow the effluent to drain out over a large area and seep slowly into the ground. In order to install a drainage field, it is first necessary to test the soil below, as it must be sufficiently permeable to allow the effluent to percolate away, but not so permeable that it flows too quickly and the soil is not able to filter out the pathogenic bacteria.

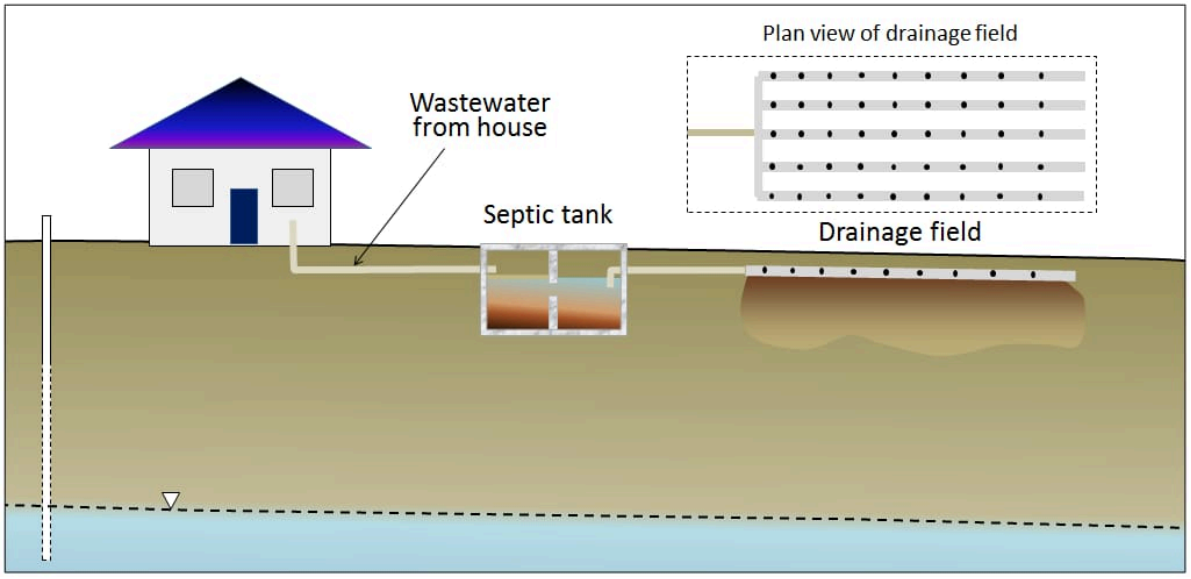


Figure 14.4.10 A typical septic system.

If they are properly installed and used, and if the sludge is periodically removed from the tank, a septic system should be effective in treating the sewage for decades. The anaerobic and aerobic bacteria should be able to break down the incoming waste and there should be little risk to the surface environment or groundwater. But many things can go wrong with a septic system, including the following:

- If inappropriate chemicals are added to the waste stream, they may interfere with the natural breakdown of the sewage.
- If the tank is not periodically pumped out, solids can get into the drainage field and compromise the drainage, resulting in the flow of effluent toward the surface.
- If the soil is either not sufficiently permeable or too permeable, the effluent will not drain away (and will start to pool at the surface) or it will drain too quickly.
- If the drainage field is constructed in an area where the water table is close to surface, some of the effluent is likely to flow into the groundwater without being treated.

Prevention and Mitigation of Groundwater Contamination

As illustrated in the landfill example above, there are two fairly simple ways to significantly reduce the chance and degree of groundwater contamination from surface sources. One is to prevent rainwater from infiltrating down to the water table and picking up contaminants; this can be achieved by simply capping or roofing over the landfill, mine tailings, or spill site. The second is to provide an impermeable barrier beneath the contaminant. Modern landfills and mine tailings impoundments are all built using some combination of clay and engineered plastic barriers. Both of these solutions — caps and liners — are subject to failure due to leaks.

Once contaminants are in the groundwater, the main form of remediation is to pump out the contaminated water and treat it at the surface. This can be a slow process, and preventing the contaminant from travelling significantly during this process can be accomplished by manipulating local groundwater flow through the extraction or injection of water at certain locations. Consider this in the exercise below.

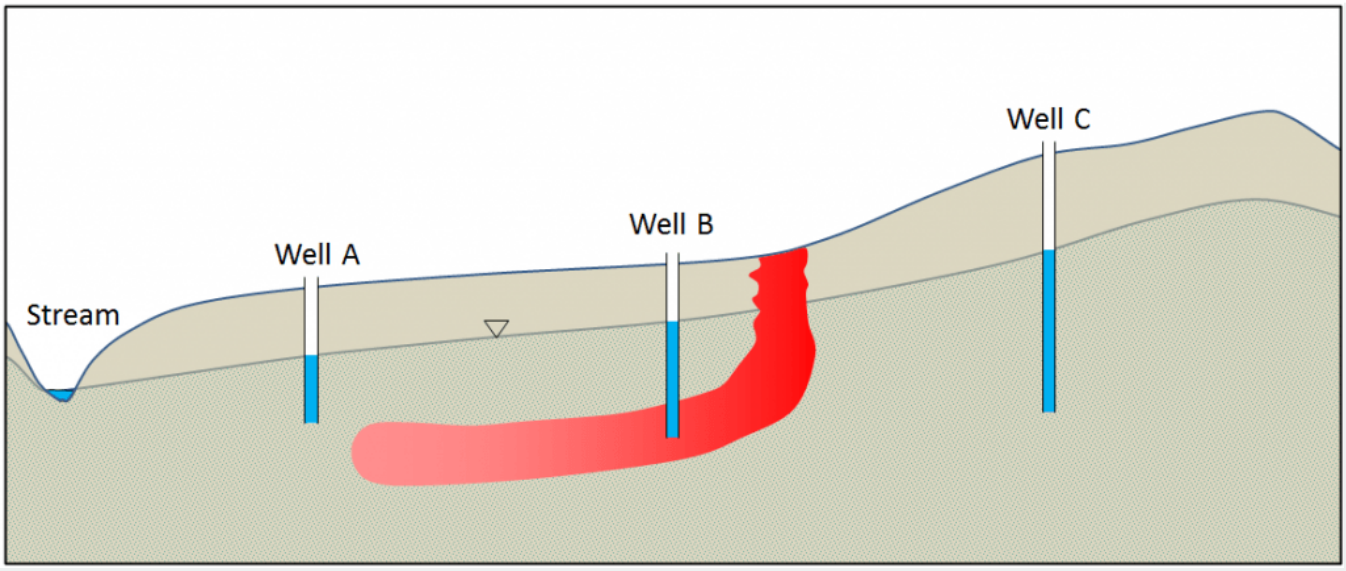


Figure 14.4.11 A plume of contamination (red) from a surface source

Figure 14.4.11 shows a groundwater contaminant plume in red. The source of the contamination has been removed but if the plume is not dealt with, it will eventually enter the stream and threaten the health of wildlife. Pumping the contaminant from well B for treatment will not be sufficient to prevent some of the contamination from making it to the stream.

What could you do at wells A and C to prevent this? Explain and use the diagram below to illustrate the expected changes to the water table and the movement of the plume.

See Appendix 3 for Exercise 14.6 answers.

Image Descriptions

Figure 14.4.11 image description: A cross section of a stream, three wells, and a contaminant plume. The stream marks the lowest point. As you move to the right, the ground and water table rises and you meet Well A, Well B, and Well C . The contaminant was introduced on the surface between Well B and Well C. It sunk down into the ground, entered the groundwater, and spread out to the left to surround the bottom part of Well B. It has continued to flow downhill towards Well A and the stream, but has yet to reach Well A. [Return to Figure 14.4.11]

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- Figure 14.4.3: Image from Newton, P. and Gilchrist, A. 2010. Technical summary of intrinsic vulnerability mapping methods of Vancouver Island, Vancouver Island Water Resources Vulnerability Mapping Project, Vancouver Island University, 45pp. Used with permission.
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1. "Adsorb" (with a "d") is not the same as "absorb" (with a "b"). Water can be absorbed by a sponge. Ions dissolved in water can be adsorbed onto—or desorbed from—the surfaces of clay minerals. ↵

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14.5: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
14.1 Groundwater and Aquifers	Porosity is the percentage of open space within a rock or unconsolidated sedimentary deposit, while permeability is the facility with which water can be transmitted through that material. An aquifer is a body of rock or sediment that has sufficient permeability for water to be extracted, while an aquitard is an impermeable body. An aquifer is described as confined if it is overlain by an impermeable layer (confining layer), or unconfined if it has no such confining layer.
14.2 Groundwater Flow	The water table is the upper surface of the saturated zone in an unconfined aquifer. A confined aquifer has a potentiometric surface (instead of a water table), which is defined as the level to which water would rise if a well were drilled into the confined aquifer. Change in groundwater head over distance is the hydraulic gradient. The theoretical velocity of flow in an aquifer is defined by Darcy's law as the hydraulic conductivity (a measure of permeability) times the hydraulic gradient ($V = K * i$). It is possible to predict groundwater flow paths if we can draw equipotential lines within an aquifer. In areas where limestone has solutional openings (e.g., caves), water flow is determined by gravity above the water table and by the hydraulic gradient below the water table.
14.3 Groundwater Extraction	Groundwater can be extracted at springs, but in most cases, wells are needed to ensure a steady supply. Pumping groundwater from a well lowers groundwater head near the well, creating flow toward the well. This creates a cone of depression around the well. Excessive pumping can lead to a well running dry or to a lack of water in nearby wells. During extended periods of dry weather, or if consistent over-pumping occurs, aquifers may be depleted. Observation wells are used to monitor short-term and long-term changes in water levels that can indicate changes in aquifer health.
14.4 Groundwater Quality	The quality of groundwater can be compromised by both natural and anthropogenic contamination. Natural contamination can be caused by particularly high levels of contaminants within the aquifer itself, but is more commonly a result of enhanced solubility of contaminants due to the aquifer chemistry. Some common sources of anthropogenic contamination include agriculture, industry, mining, landfills, and leaking underground storage tanks. We can assess the vulnerability of aquifers to contamination by mapping regional variations in parameters such as depth to the water table, permeability, slope, and precipitation.

Questions for Review

Answers to review questions can be found in Appendix 2.

1. What is the difference between porosity and permeability?
2. Both sand and clay deposits can have high porosity, but while most sand also has high permeability, clay does not. Why not?
3. Arrange the following types of rock in order of their likely permeability, as measured by the hydraulic conductivity (K): mudstone, fractured granite, limestone in a karst region, sandstone, and unfractured gneiss.
4. Sue, the owner of Joe's 24-Hour Gas, has a shallow well (15 meters deep) as illustrated in the diagram. The well can only produce 0.5 litres per minute, but that's enough for water to make coffee and supply a washroom that gets used several times a day. Frank, who operates a raspberry farm next door, uses up to 250,000 litres of water per day to irrigate his crop during summer. He gets water from a deeper well that can produce 250 litres per minute. See the diagram below.

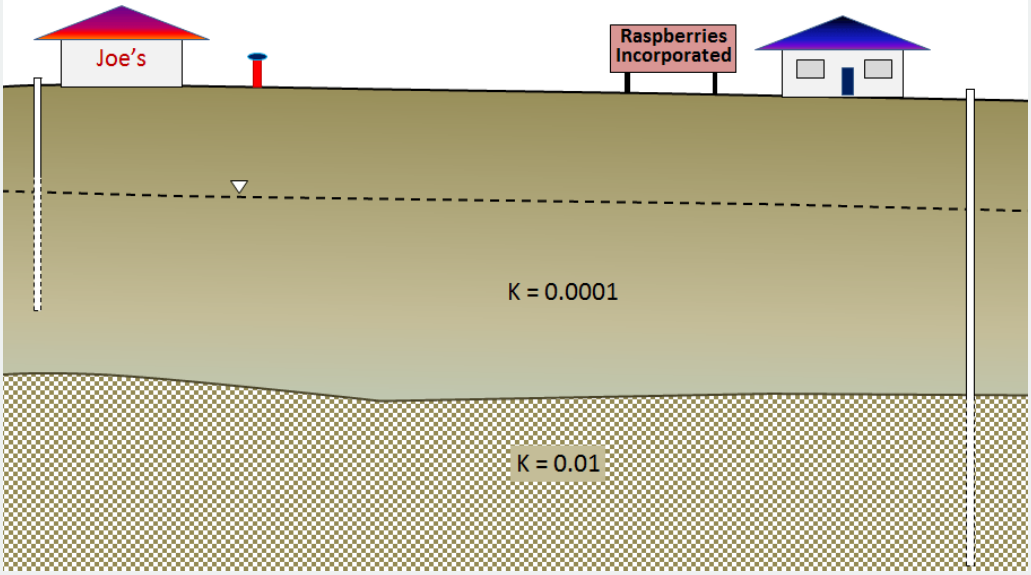


Figure A

- What type of aquifer does Sue use?
- What type of aquifer does Frank use?
- It seems that what Sue calls an aquifer is an aquitard (confining layer) from Frank's perspective. How is that possible?

- Two wells 70 meters apart have water levels of 77 meters and 83 meters above sea level respectively. The aquifer has a hydraulic conductivity of 0.0003 meters per second. What is the likely velocity of groundwater flow in the region between these two wells?
- The well in question 5—with a water level of 83 meters—is heavily used and after several months the water level has dropped by 9 meters. How will that affect the flow of groundwater in the area between the two wells?
- Explain why it is important for provincial governments to operate observation well networks.
- What is the main difference between natural and anthropogenic contamination of groundwater?
- Why is a highly permeable aquifer more vulnerable to anthropogenic contamination than a less permeable aquifer?
- How can a livestock operation lead to contamination of groundwater? What is the most likely contaminant?
- Which mineral in the rock of a mining operation is typically responsible for acid rock drainage?
- Why is it necessary to test the permeability of the soil before constructing a septic field?

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15: MASS WASTING

Mass wasting is the geomorphic process by which soil, sand, regolith, and rock move downslope typically as a mass, largely under the force of gravity, but frequently affected by water and water content as in submarine environments and mudflows.

15.1: Factors That Control Slope Stability

15.2: Classification of Mass Wasting

15.3: Preventing, Delaying, Monitoring, and Mitigating Mass Wasting

15.4: Summary

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15.1: Factors That Control Slope Stability

Mass wasting happens because tectonic processes have created uplift. Erosion, driven by gravity, is the inevitable response to that uplift, and various types of erosion, including mass wasting, have created slopes in the uplifted regions. Slope stability is ultimately determined by two factors: the angle of the slope and the strength of the materials on it.

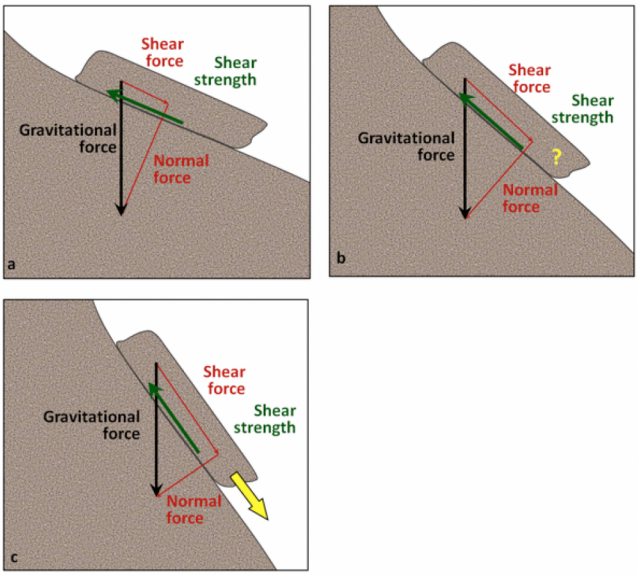


Figure 15.1.1 Differences in the shear and normal components of the gravitational force on slopes with differing steepness. The gravitational force is the same in all three cases. In (a) the shear force is substantially less than the shear strength, so the block should be stable. In (b) the shear force and shear strength are about equal, so the block may or may not move. In (c) the shear force is substantially greater than the shear strength, so the block is very likely to move.

Figure 15.2 shows a block of rock situated on a rock slope. The block is being pulled toward Earth's centre (vertically down) by gravity. We can split the vertical gravitational force into two components relative to the slope: one pushing the block down the slope (the **shear force**), and the other pushing into the slope (the **normal force**). The shear force, which wants to push the block down the slope, has to overcome the strength of the connection between the block and the slope, which may be quite weak if the block has split away from the main body of rock, or may be very strong if the block is still a part of the rock. This is the **shear strength**, and in Figure 15.1.1a, it is greater than the shear force, so the block should not move. In Figure 15.1.1b the slope is steeper and the shear force is approximately equal to the shear strength. The block may or may not move under these circumstances. In Figure 15.1.1c, the slope is steeper still, so the shear force is considerably greater than the shear strength, and the block will very likely move.

As already noted, slopes are created by uplift followed by erosion. In areas with relatively recent uplift (such as most of British Columbia and the western part of Alberta), slopes tend to be quite steep. This is especially true where glaciation has taken place because glaciers in mountainous terrain create steep-sided valleys. In areas without recent uplift (such as central Canada), slopes are less steep because hundreds of millions of years of erosion (including mass wasting) has made them that way. However, as we'll see, some mass wasting can happen even on relatively gentle slopes.

The strength of the materials on slopes can vary widely. Solid rocks tend to be strong, but there is a very wide range of rock strength. If we consider just the strength of the rocks, and ignore issues like fracturing and layering, then most crystalline rocks—like granite, basalt, or gneiss—are very strong, while some metamorphic rocks—like schist—are moderately strong. Sedimentary rocks have variable strength. Dolostone and some limestone are strong, most sandstone and conglomerate are moderately strong, and some sandstone and all mudstones are quite weak.

Fractures, metamorphic foliation, or bedding can significantly reduce the strength of a body of rock, and in the context of mass wasting, this is most critical if the planes of weakness are parallel to the slope and least critical if they are perpendicular to the slope. This is illustrated in Figure 15.1.2. At locations A and B the bedding is nearly perpendicular to the slope and the situation is relatively stable. At location D the bedding is nearly parallel to the slope and the situation is quite unstable. At location C the

bedding is nearly horizontal and the stability is intermediate between the other two extremes.

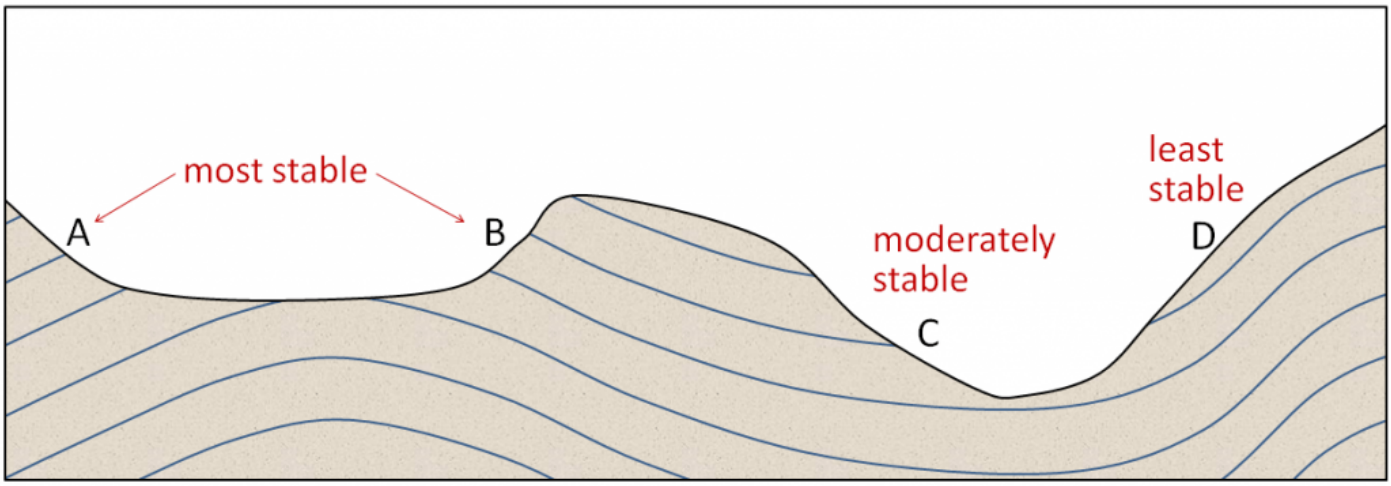


Figure 15.1.2 Relative stability of slopes as a function of the orientation of weaknesses (in this case bedding planes) relative to the slope orientations.

Internal variations in the composition and structure of rocks can significantly affect their strength. Schist, for example, may have layers that are rich in sheet silicates (mica or chlorite) and these will tend to be weaker than other layers. Some minerals tend to be more susceptible to weathering than others, and the weathered products are commonly quite weak (e.g., the clay formed from feldspar). The side of Johnson Peak that failed in 1965 (Hope Slide) is made up of chlorite schist (metamorphosed sea-floor basalt) that has feldspar-bearing sills within it (they are evident within the inset area of Figure 15.1.1). The foliation and the sills are parallel to the steep slope. The schist is relatively weak to begin with, and the feldspar in the sills, which has been altered to clay, makes it even weaker.

Unconsolidated sediments are generally weaker than sedimentary rocks because they are not cemented and, in most cases, have not been significantly compressed by overlying materials. This binding property of sediment is sometimes referred to as *cohesion*. Sand and silt tend to be particularly weak, clay is generally a little stronger, and sand mixed with clay can be stronger still. The deposits that make up the cliffs at Point Grey in Vancouver include sand, silt, and clay overlain by sand. As shown in Figure 15.1.3 (left) the finer deposits are relatively strong (they maintain a steep slope), while the overlying sand is relatively weak, and has a shallower slope that has recently failed. Glacial till—typically a mixture of clay, silt, sand, gravel, and larger clasts—forms and is compressed beneath tens to thousands of meters of glacial ice so it can be as strong as some sedimentary rock (Figure 15.1.3, right).



Figure 15.1.3 Left: Glacial outwash deposits at Point Grey, in Vancouver. The dark lower layer is made up of sand, silt, and clay. The light upper layer is well-sorted sand. Right: Glacial till on Quadra Island, B.C. The till is strong enough to have formed a near-vertical slope.

Apart from the type of material on a slope, the amount of water that the material contains is the most important factor controlling its strength. This is especially true for unconsolidated materials, like those shown in Figure 15.1.3, but it also applies to bodies of rock. Granular sediments, like the sand at Point Grey, have lots of spaces between the grains. Those spaces may be completely dry (filled only with air); or moist (often meaning that some spaces are water filled, some grains have a film of water around them, and small amounts of water are present where grains are touching each other); or completely saturated (Figure 15.1.4). Unconsolidated sediments tend to be strongest when they are moist because the small amounts of water at the grain boundaries hold the grains together with surface tension. Dry sediments are held together only by the friction between grains, and if they are well sorted or well rounded, or both, that cohesion is weak. Saturated sediments tend to be the weakest of all because the large amount of water actually pushes the grains apart, reducing the mount friction between grains. This is especially true if the water is under pressure.

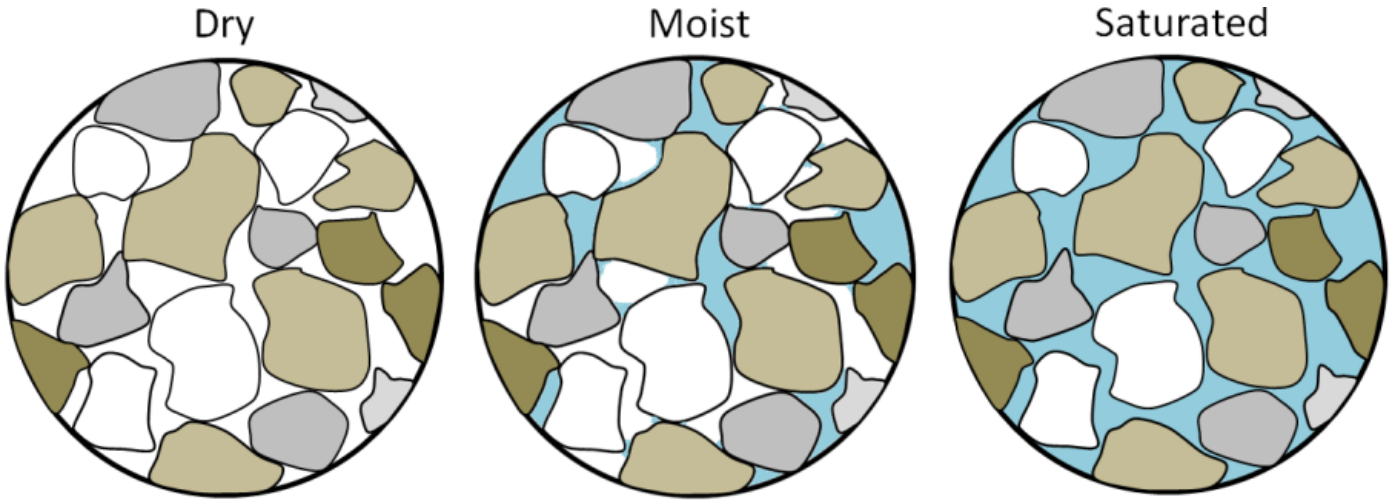


Figure 15.1.4 Depiction of dry, moist, and saturated sand.

Exercise 15.1 Sand and water



Figure 15.1.5

If you've ever been to the beach, you'll already know that sand behaves differently when it's dry than it does when it's wet, but it's worth taking a systematic look at the differences in its behaviour. Find about half a cup of clean, dry sand (or get some wet sand and dry it out), and then pour it from your hand onto a piece of paper. You should be able to make a cone-shaped pile that has a slope of around 30°. If you pour more sand on the pile, it will get bigger, but the slope should remain the same. Now add some water to the sand so that it is moist. An easy way to do this is to make it completely wet and then let the water drain away for a minute. You should be able to form this moist sand into a steep pile (with slopes of around 80°). Finally, put the same sand into a cup and fill the cup with water so the sand is just covered. Swirl it around so that the sand remains in suspension, and then

quickly tip it out onto a flat surface (best to do this outside). It should spread out over a wide area, forming a pile with a slope of only a few degrees.

Water will also reduce the strength of solid rock, especially if it has fractures, bedding planes, or clay-bearing zones. This effect is even more significant when the water is under pressure, which is why you'll often see holes drilled into rocks on road cuts to relieve this pressure. One of the hypotheses advanced to explain the 1965 Hope Slide is that the very cold conditions that winter caused small springs in the lower part of the slope to freeze over, preventing water from flowing out. It is possible that water pressure gradually built up within the slope, weakening the rock mass to the extent that the shear strength was no longer greater than the shear force.

Water also has a particular effect on clay-bearing materials. All clay minerals will absorb a little bit of water, and this reduces their strength. The **smectite** clays (such as the **bentonite** used in cat litter) can absorb a lot of water, and that water pushes the sheets apart at a molecular level and makes the mineral swell. Smectite that has expanded in this way has almost no strength; it is extremely slippery.

And finally, water can significantly increase the mass of the material on a slope, which increases the gravitational force pushing it down. A body of sediment that has 25% porosity and is saturated with water weighs approximately 13% more than it does when it is completely dry, so the gravitational shear force is also 13% higher. In the situation shown in Figure 15.1.1b, a 13% increase in the shear force could easily be enough to tip the balance between shear force and shear strength.

Mass-Wasting Triggers

In the previous section, we talked about the shear force and the shear strength of materials on slopes, and about factors that can reduce the shear strength. Shear force is primarily related to slope angle, and this does not change quickly. But shear strength can change quickly for a variety of reasons, and events that lead to a rapid reduction in shear strength are considered to be **triggers** for mass wasting.

An increase in water content is the most common mass-wasting trigger. This can result from rapid melting of snow or ice, heavy rain, or some type of event that changes the pattern of water flow on the surface. Rapid melting can be caused by a dramatic increase in temperature (e.g., in spring or early summer) or by a volcanic eruption. Heavy rains are typically related to major storms. Changes in water flow patterns can be caused by earthquakes, previous slope failures that dam up streams, or human structures that interfere with runoff (e.g., buildings, roads, or parking lots). An example of this is the deadly 2005 debris flow in North Vancouver (Figure 15.1.6). The 2005 failure took place in an area that had failed previously, and a report written in 1980 recommended that the municipal authorities and residents take steps to address surface and slope drainage issues. Little was done to improve the situation.



Figure 15.1.6 The debris flow in the Riverside Drive area of North Vancouver in January, 2005 happened during a rainy period, but was likely triggered by excess runoff related to the roads at the top of this slope and by landscape features, including a pool, in the area surrounding the house visible here.

In some cases, a *decrease* in water content can lead to failure. This is most common with clean sand deposits (e.g., the upper layer in Figure 15.1.3 (left)), which lose strength when there is no water to hold the grains together.

Freezing and thawing can also trigger some forms of mass wasting. More specifically, the thawing can release a block of rock that was attached to a slope by a film of ice.

One other process that can weaken a body of rock or sediment is shaking. The most obvious source of shaking is an earthquake, but shaking from highway traffic, construction, or mining will also do the job. Several deadly mass-wasting events (including snow avalanches) were triggered by the M7.8 earthquake in Nepal in April 2015.

Saturation with water and then seismic shaking led to the occurrence of thousands of slope failures in the Sapporo area of Hokkaido, Japan in September 2018, as shown on Figure 15.1.7. The area was drenched with rain from tropical storm Jebi on September 4th. On September 6th it was shaken by a M6.6 earthquake which triggered debris flows in the water-saturated volcanic materials on steep slopes. There were 41 deaths related to the slope failures.



Figure 15.1.7 Slope failures in the Sapporo area of Japan following a typhoon (Sept. 4th, 2018) and earthquake (Sept. 6th, 2018) (Before and after Landsat 8 images: left: July 2017, right: September 2018).

Media Attributions

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15.2: Classification of Mass Wasting

It's important to classify slope failures so that we can understand what causes them and learn how to mitigate their effects. The three criteria used to describe slope failures are:

- The type of material that failed (typically either bedrock or unconsolidated sediment)
- The mechanism of the failure (how the material moved)
- The rate at which it moved

The type of motion is the most important characteristic of a slope failure, and there are three different types of motion:

- If the material drops through the air, vertically or nearly vertically, it's known as a **fall**.
- If the material moves as a mass along a sloping surface (without internal motion within the mass), it's a **slide.**
- If the material has internal motion, like a fluid, it's a **flow**.

Unfortunately it's not normally that simple. Many slope failures involve two of these types of motion, some involve all three, and in many cases, it's not easy to tell how the material moved. The types of slope failure that we'll cover here are summarized in Table 15.1.

Table 15.1 Classification of slope failures based on type of material and type of motion

[Skip Table]

Failure Type	Type of Material	Type of Motion	Rate of Motion
Rock fall	Rock fragments	Vertical or near-vertical fall (plus bouncing in many cases)	Very fast (Greater than 10s of meters per second)
Rock slide	A large rock body	Motion as a unit along a planar surface (translational sliding)	Typically very slow (millimeters per year to centimeters per year), but some can be faster
Rock avalanche	A large rock body that slides and then breaks into small fragments	Flow (at high speeds the mass of rock fragments is suspended on a cushion of air)	Very fast (Greater than tens of meters per second)
Creep or solifluction	Soil or other overburden; in some cases, mixed with ice	Flow (although sliding motion may also occur)	Very slow (millimeters per year to centimeters per year)
Slump	Thick deposits (a metre to 10s of meters) of unconsolidated sediment	Motion as a unit along a curved surface (rotational sliding)	Slow (centimeters per year to meters per year)
Mudflow	Loose sediment with a significant component of silt and clay	Flow (a mixture of sediment and water moves down a channel)	Moderate to fast (centimeters per second to meters per second)
Debris flow	Sand, gravel, and larger fragments	Flow (similar to a mudflow, but typically faster)	Fast (meters per second)

Rock Fall

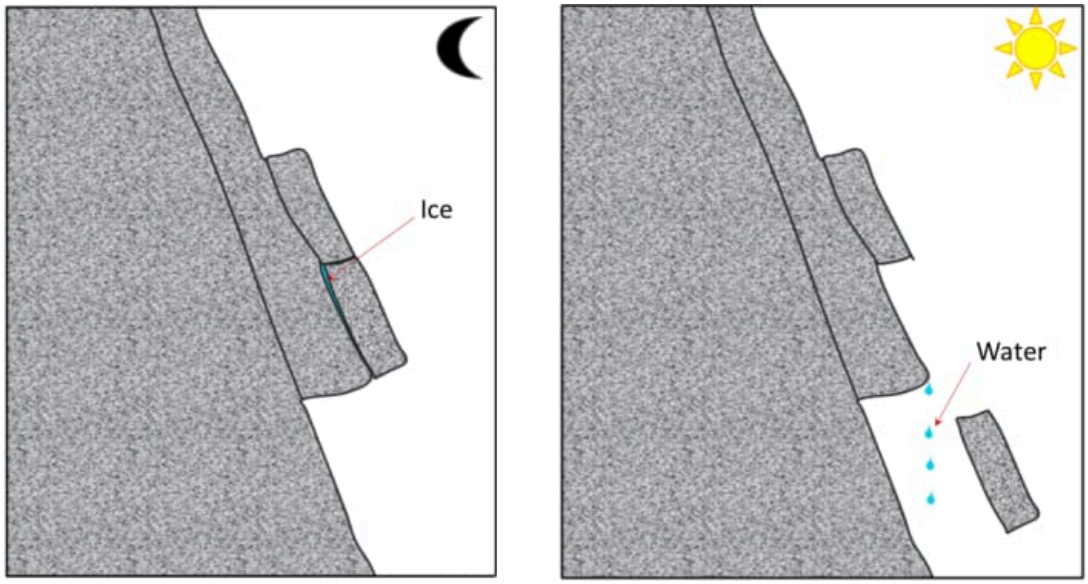


Figure 15.2.1 The contribution of freeze-thaw to rock fall.

Rock fragments can break off relatively easily from steep bedrock slopes, most commonly due to frost-wedging in areas where there are many freeze-thaw cycles per year. If you've ever hiked along a steep mountain trail on a cool morning, you might have heard the occasional fall of rock fragments onto a **talus slope**. This happens because the water between cracks freezes and expands overnight, and then when that same water thaws in the morning sun, the fragments that had been pushed beyond their limit by the ice fall to the slope below (Figure 15.2.1).

A typical talus slope, near Keremeos in southern B.C., is shown in Figure 15.2.2. In December 2014, a large block of rock split away from a cliff in this same area. It broke into smaller pieces that tumbled down the slope and crashed into the road, smashing the concrete barriers and gouging out large parts of the pavement. Luckily no one was hurt.



Figure 15.2.2 Left: A talus slope near Keremeos, B.C., formed by rock fall from the cliffs above. Right: The results of a rock fall onto a highway west of Keremeos in December 2014.

Rock Slide

A rock slide is the sliding motion of rock along a sloping surface. In most cases, the movement is parallel to a fracture, bedding, or metamorphic foliation plane, and it can range from very slow to moderately fast. The word **sackung** describes the very slow motion of a block of rock (millimeters per year to centimeters per year) on a slope. A good example is the Downie Slide north of

Revelstoke, B.C., which is shown in Figure 15.2.3. In this case, a massive body of rock is very slowly sliding down a steep slope along a plane of weakness that is approximately parallel to the slope. The Downie Slide, which was first recognized in the 1950s, prior to the construction of the Revelstoke Dam in the late 1970s, was moving very slowly at the time (a few centimeters per year). Geological engineers were concerned that the presence of water in the reservoir (visible in Figure 15.2.3) could further weaken the plane of failure, leading to an acceleration of the motion. The result would have been a catastrophic failure into the reservoir that would have sent a wall of water over the dam and into the community of Revelstoke. During the construction of the dam they tunneled into the rock at the base of the slide and drilled hundreds of drainage holes upward into the plane of failure. This allowed water to drain out so that the pressure was reduced, which reduced the rate of movement of the sliding block. BC Hydro monitors this site continuously; the slide block is currently moving more slowly than it was prior to the construction of the dam.



Figure 15.2.3 The Downie Slide, a sackung, on the shore of the Revelstoke Reservoir (above the Revelstoke Dam). The head scarp is visible at the top and a side-scarp along the left side.



Figure 15.2.4 Site of the 2008 rock slide at Porteau Cove. Notice the prominent fracture set parallel to the surface of the slope. The slope has been stabilized with rock bolts (visible near to the top of the photo) and holes have been drilled into the rock to improve drainage (one is visible in the lower right). Risk to passing vehicles from rock fall has been reduced by hanging mesh curtains (background).

In the summer of 2008, a large block of rock slid rapidly from a steep slope above Highway 99 near Porteau Cove (between Horseshoe Bay and Squamish). The block slammed into the highway and adjacent railway and broke into many pieces. The highway was closed for several days, and the slope was subsequently stabilized with rock bolts and drainage holes. As shown in Figure 15.2.4, the rock is fractured parallel to the slope, and this almost certainly contributed to the failure. However, it is not actually known what triggered this event as the weather was dry and warm during the preceding weeks, and there was no

significant earthquake in the region.

Rock Avalanche

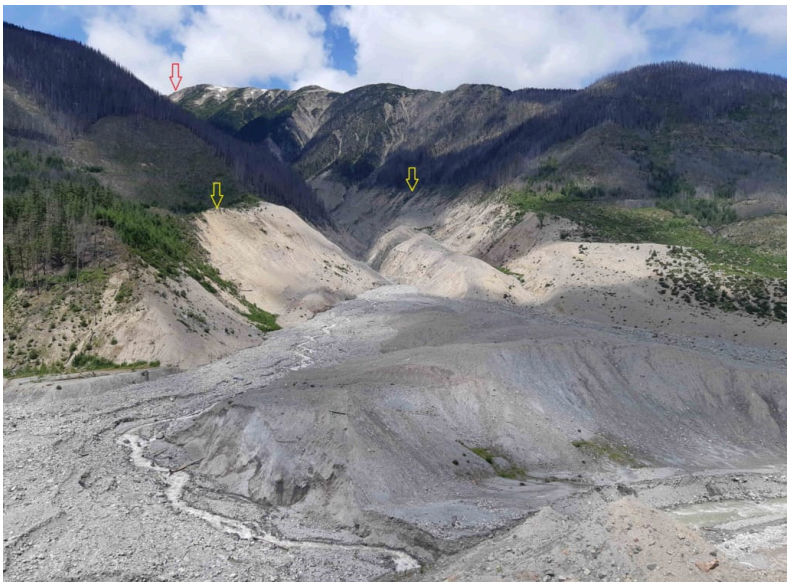


Figure 15.2.5 The August 2010 Mount Meager rock avalanche, showing where the slide originated (red arrow, 4 km upstream) and its path down a steep narrow valley. The yellow arrows show how far up the valley the avalanche extended.

If a rock slides and then starts moving quickly (meters per second), the rock is likely to break into many small pieces, and at that point it turns into a **rock avalanche**, in which the large and small fragments of rock move in a fluid manner supported by a cushion of air within and beneath the moving mass. The 1965 Hope Slide (Figure 15.2.1) was a rock avalanche, as was the famous 1903 Frank Slide in southwestern Alberta. The 2010 slide at Mount Meager (west of Lillooet) was also a rock avalanche, and rivals the Hope Slide as the largest slope failure in Canada during historical times (Figure 15.2.5).

Creep or Solifluction

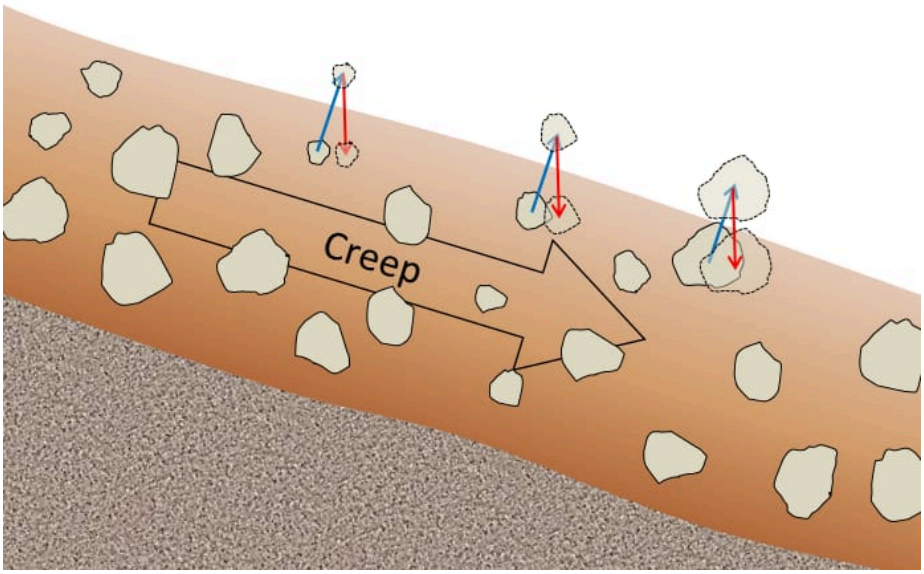


Figure 15.2.6 A depiction of the contribution of freeze-thaw to creep. The blue arrows represent uplift caused by freezing in the wet soil underneath, while the red arrows represent depression by gravity during thawing. The uplift is perpendicular to the slope, while the drop is vertical.

The very slow—millimeters per year to centimeters per year—movement of soil or other unconsolidated material on a slope is

known as creep. **Creep**, which normally only affects the upper several centimeters of loose material, is typically a type of very slow flow, but in some cases, sliding may take place. Creep can be facilitated by freezing and thawing because, as shown in Figure 15.2.6, particles are lifted perpendicular to the surface by the growth of ice crystals within the soil, and then let down vertically by gravity when the ice melts. The same effect can be produced by frequent wetting and drying of the soil. In cold environments, **solifluction** is a more intense form of freeze-thaw-triggered creep.

Creep is most noticeable on moderate-to-steep slopes where trees, fence posts, or grave markers are consistently leaning in a downhill direction. In the case of trees, they try to correct their lean by growing upright, and this leads to a curved lower trunk known as a “pistol butt.” An example is shown on Figure 15.2.7.



Figure 15.2.7 Pistol-butt shaped trees on a slope that is experiencing creep

Slump

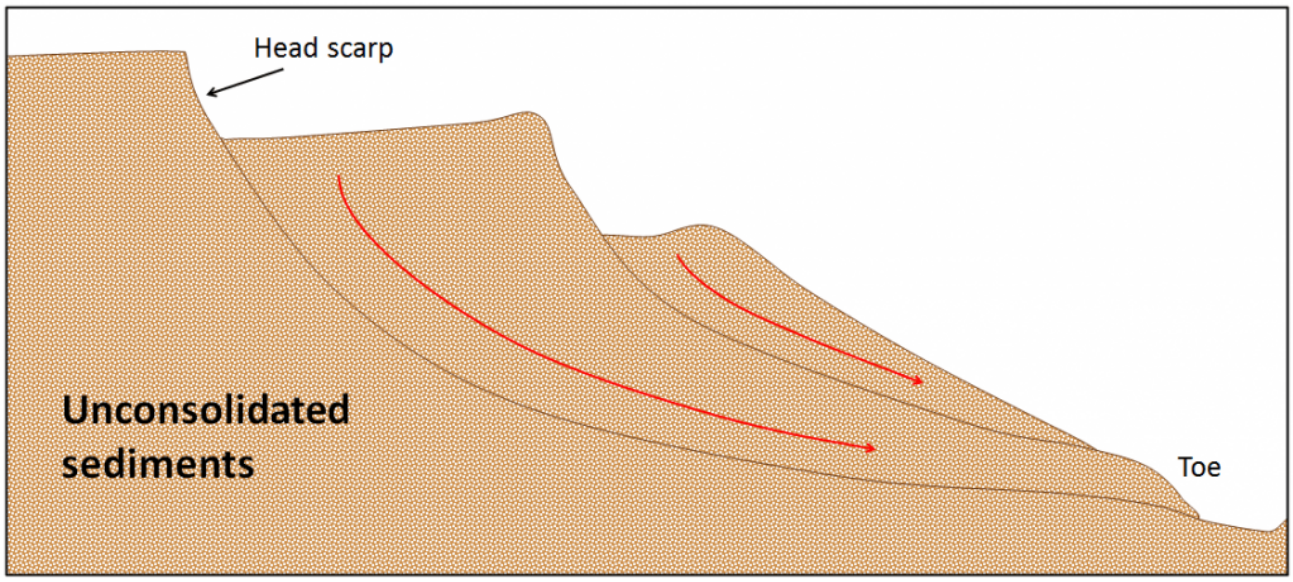


Figure 15.2.8 A depiction of the motion of unconsolidated sediments in an area of slumping.

Slump is a type of slide (movement as a mass) that takes place within thick unconsolidated deposits (typically thicker than 10 meters). Slumps involve movement along one or more curved failure surfaces, with downward motion near the top and outward motion toward the bottom (Figure 15.2.8). They are typically caused by an excess of water within these materials on a steep slope.



Figure 15.2.9 A slump along the banks of a small coulee near Lethbridge, Alberta. The main head-scarp is clearly visible at the top, and a second smaller one is visible about one-quarter of the way down. The toe of the slump is being eroded by the seasonal stream that created the coulee.

An example of a slump in the Lethbridge area of Alberta is shown in Figure 15.2.9. This feature has likely been active for many decades, and moves a little more whenever there are heavy spring rains and significant snowmelt runoff. The toe of the slump is failing because it has been eroded by the small stream at the bottom.

Mudflows and Debris Flows



Figure 15.2.10 A slump (left) and an associated mudflow (centre) at the same location as Figure 15.2.9, near Lethbridge, Alberta.

As you saw in Exercise 15.1, when a mass of sediment becomes completely saturated with water, the mass loses strength, to the extent that the grains are pushed apart, and it will flow, even on a gentle slope. This can happen during rapid spring snowmelt or heavy rains, and is also relatively common during volcanic eruptions because of the rapid melting of snow and ice. (A mudflow or debris flow on a volcano or during a volcanic eruption is a *lahar*.) If the material involved is primarily sand-sized or smaller, it is known as a mudflow, such as the one shown in Figure 15.2.10.

If the material involved is gravel sized or larger, it is known as a debris flow. Because it takes more gravitational energy to move larger particles, a debris flow typically forms in an area with steeper slopes and more water than does a mudflow. In many cases, a debris flow takes place within a steep stream channel, and is triggered by the collapse of bank material into the stream. This creates a temporary dam, and then a major flow of water and debris when the dam breaks. This is the situation that led to the fatal debris flow at Johnsons Landing, B.C., in 2012. A typical west-coast debris flow is shown in Figure 15.2.11. This event took place in November 2006 in response to very heavy rainfall. There was enough energy to move large boulders and to knock over large trees.



Figure 15.2.11 The lower part of debris flow within a steep stream channel near Buttle Lake, B.C., in November 2006.

Exercise 15.2 Classifying slope failures

These four photos show some of the different types of slope failures described above. Try to identify each types and provide some criteria to support your choice.



Figure 15.2.12a



Figure 15.2.12b



Figure 15.2.12c



Figure 15.2.12d

a:	b:
c:	d:

See Appendix 3 for Exercise 15.2 answers.

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15.3: Preventing, Delaying, Monitoring, and Mitigating Mass Wasting

As already noted, we cannot prevent mass wasting in the long term as it is a natural and ongoing process; however, in many situations there are actions that we can take to reduce or mitigate its damaging effects on people and infrastructure. Where we can neither delay nor mitigate mass wasting, we should consider moving out of the way.

Preventing and Delaying Mass Wasting

It is comforting to think that we can prevent some effects of mass wasting by mechanical means, such as the rock bolts in the road cut at Porteau Cove (Figure 15.3.4), or the drill holes used to drain water out of a slope, as was done at the Downie Slide (Figure 15.3.3), or the building of physical barriers, such as retaining walls. What we have to remember is that the works of humans are mostly insignificant compared to the works of nature. The rock bolts in the road cut at Porteau Cove will slowly start to corrode after a few years, and within a few decades, many of them will begin to lose their strength. Unless they are replaced, they will no longer support that slope. Likewise, drainage holes at the Downie Slide will eventually become plugged with sediment and chemical precipitates, and unless they are periodically unplugged, their effectiveness will decrease. Eventually, unless new holes are drilled, the drainage will be so compromised that the slide will start to move again. This is why careful long-term slope monitoring by geological and geotechnical engineers is important at these sites. The point here is that our efforts to "prevent" mass wasting are only as good as our resolve to maintain those preventive measures.

Delaying mass wasting is a worthy endeavour, of course, because during the time that the measures are still effective they can save lives and reduce damage to property and infrastructure. The other side of the coin is that we must be careful to avoid activities that could make mass wasting more likely. One of the most common anthropogenic causes of mass wasting is road construction, and this applies both to remote gravel roads built for forestry and mining and large urban and regional highways. Road construction is a potential problem for two reasons. First, creating a flat road surface on a slope inevitably involves creating a cut bank that is steeper than the original slope. This might also involve creating a filled bank that is both steeper and weaker than the original slope (Figure 15.3.12). Second, roadways typically cut across natural drainage features, and unless great care is taken to reroute the runoff water and prevent it from forming concentrated flows, oversaturating fill of materials can result. A specific example of the contribution of construction-related impeded drainage to slope instability was shown earlier in Figure 15.3.6.

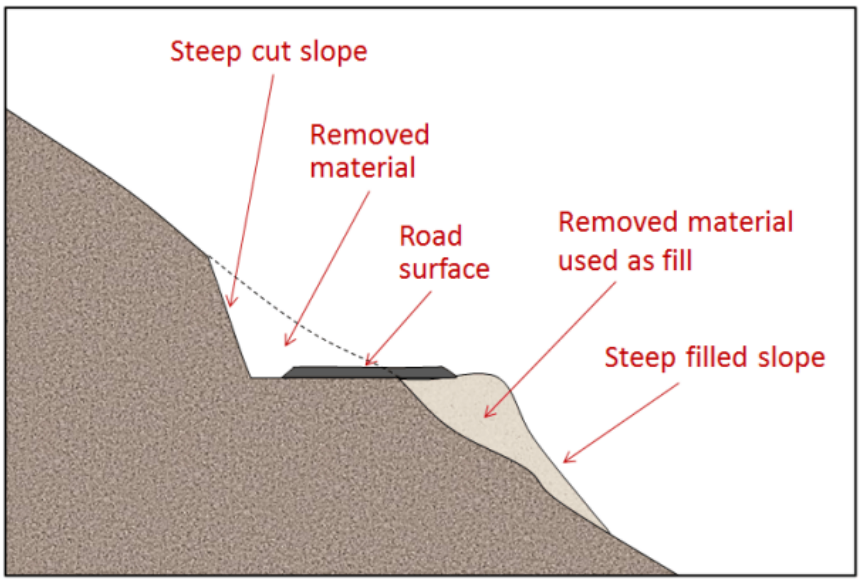


Figure 15.3.1 An example of a road constructed by cutting into a steep slope. The cut material was moved to outside part of the road to act as fill.

Apart from water issues, engineers building roads and other infrastructure on bedrock slopes have to be acutely aware of the

geology, and especially of any weaknesses or discontinuities in the rock related to bedding, fracturing, or foliation. If possible, situations like that at Porteau Cove (Figure 15.3.3) should be avoided — by building somewhere else — rather than trying to stitch the slope back together with rock bolts.

It is widely believed that construction of buildings on the tops of steep slopes can contribute to the instability of the slope. This is probably true, but in most cases that is not because of the weight of the building. As you'll see by completing Exercise 15.3, a typical house isn't usually heavier than the fill that was removed from the hole in the ground made to build it. A more likely contributor to instability of the slope around a building is the effect that it and the changes made to the surrounding area have on drainage.

Exercise 15.3 How much does a house weigh, and can it contribute to a slope failure?

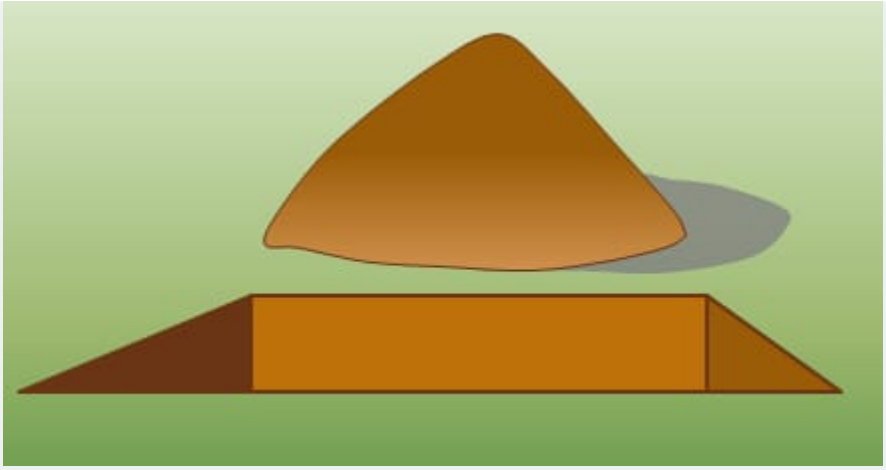


Figure 15.3.2

It is commonly believed that building a house (or some other building) at the top of a slope will add a lot of extra weight to the slope, which could contribute to slope failure. But what does a house actually weigh? A typical 150 square metre (approximately 1,600 square feet) wood-frame house with a basement and a concrete foundation weighs about 145 tonnes. But most houses are built on foundations that are excavated into the ground. This involves digging a hole and taking some material away, so we need to subtract what that excavated material weighs. Assuming our 150 square metre house required an excavation that was 15 meters by 11 meters by 1 metre deep, that's 165 cubic meters of "dirt," which typically has a density of about 1.6 tonnes per cubic metre.

Calculate the weight of the soil that was removed and compare that with the weight of the house and its foundation.

If you're thinking that building a bigger building is going to add more weight, consider that bigger buildings need bigger and deeper excavations, and in many cases the excavations will be into solid rock, which is much heavier than surficial materials.

Much more important than worrying about the weight of a building is to consider how a building might change the drainage on a slope. There are a number ways. Water can be collected by roofs, go into downspouts, and form concentrated flows that are directed onto or into the slope. Likewise drainage from nearby access roads, lawn irrigation, leaking pools, and septic systems can all alter the surface and groundwater flow in a slope.

See Appendix 3 for Exercise 15.3 answers.

Monitoring Mass Wasting



Figure 15.3.3 Part of a motion-monitoring device at the Checkerboard Slide near Revelstoke, B.C. The lower end of the cable is attached to a block of rock that is unstable. Any incremental motion of that block will move the cable, which will be detectable on this device.

In some areas, it is necessary to establish warning systems so that we know if conditions have changed at a known slide area, or if a rapid failure, such as a debris flow, is actually on its way downslope. The Downie Slide above the Revelstoke Resevoir is monitored 24/7 with a range of devices, such as inclinometers (slope-change detectors), bore-hole motion sensors, and GPS survey instruments. A simple mechanical device for monitoring the nearby Checkerboard Slide (which is also above the Revelstoke Reservoir) is shown in Figure 15.3.3. Both of these are very slow-moving rock slides, but it's critically important to be able to detect changes in their rates of motion because at both of these locations a rapid failure would result in large bodies of rock plunging into the reservoir and sending a wall of water over the Revelstoke Dam, potentially destroying the nearby town of Revelstoke.

Mount Rainier, a glacier-covered volcano in Washington State, has the potential to produce massive mudflows or debris flows (lahars) with or without a volcanic eruption. Over 100,000 people in the Tacoma, Puyallup, and Sumner areas are in harm's way because many of them currently reside on deposits from past lahars (Figure 15.3.3). In 1998, a network of acoustic monitors was established around Mount Rainier. The monitors are embedded in the ground adjacent to expected lahar paths. They are intended to provide warnings to emergency officials, and when a lahar is detected, the residents of the area will have anywhere from 40 minutes to three hours to get to safe ground.



Figure 15.3.4 Mount Rainier, Washington, from Tacoma.

Mitigating the Impacts of Mass Wasting

In situations where we can't predict, prevent, or delay mass-wasting, some effective measures can be taken to minimize the associated risk. For example, many highways in B.C. and western Alberta have avalanche shelters like that shown in Figure 15.3.5. In some parts of the world, similar features have been built to protect infrastructure from other types of mass wasting.



Figure 15.3.5 A snow avalanche shelter on the Coquihalla Highway. The expected path of the avalanche is the steep un-treed slope above.

Debris flows are inevitable, unpreventable, and unpredictable in many parts of B.C., but nowhere more so than along the Sea-to-Sky Highway between Horseshoe Bay and Squamish. The results have been deadly and expensive many times in the past. It would be very expensive to develop a new route in this region, so provincial authorities have taken steps to protect residents and traffic on the highway and the railway. Debris-flow defensive structures have been constructed in several drainage basins, as shown in Figure 15.3.6. One strategy is to allow the debris to flow quickly through to the ocean along a smooth channel. Another is to capture the debris within a constructed basin that allows the excess water to continue through, but catches the debris materials.



Figure 15.3.6 Two strategies for mitigating debris flows on the Sea-to-Sky Highway. Left: A concrete –lined channel on Alberta Creek allows debris to flow quickly through to the ocean. Right: A debris-flow catchment basin on Charles Creek. In 2010, a debris flow filled the basin to the level of the dotted white line.

Finally, in situations where we can't do anything to delay, predict, contain, or mitigate slope failures, we simply have to have the sense to stay away. There is a famous example of this in B.C. at a site known as Garibaldi, 25 kilometers south of Whistler. In the early 1980s the village of Garibaldi had a population of about 100, with construction underway on some new homes, and plans for many more. In the months that followed the deadly 1980 eruption of Mount St. Helens in Washington State, the B.C. Ministry of Transportation commissioned a geological study that revealed that a steep cliff known as The Barrier (Figure 15.3.7) had collapsed in 1855, leading to a large rock avalanche, and that it was likely to collapse again unpredictably, putting the village of Garibaldi at extreme risk. In an ensuing court case, it was ruled that the Garibaldi site was not a safe place for people to live. Those who already had homes there were compensated, and everyone else was ordered to leave.

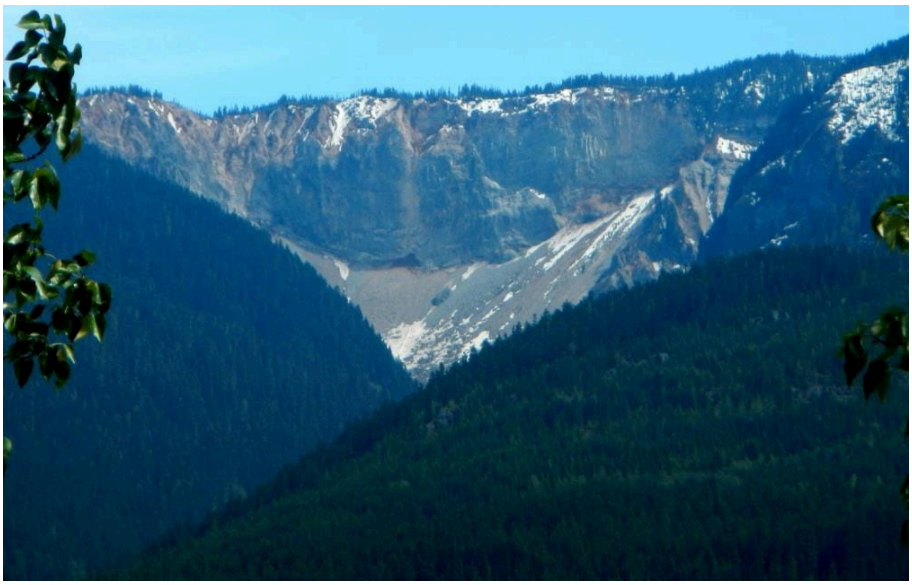


Figure 15.3.7 The Barrier, south of Whistler, B.C., was the site of a huge rock avalanche in 1855, which extended from the cliff visible here 4 kilometers down the valley and across the current location of the Sea-to-Sky Highway and the Cheakamus River.

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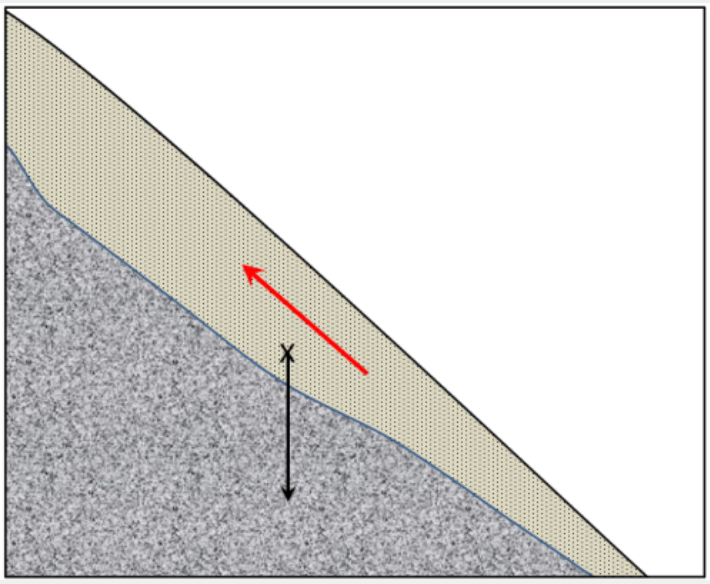
15.4: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
15.1 Factors That Control Slope Stability	Slope stability is controlled by the slope angle and the strength of the materials on the slope. Slopes are a product of tectonic uplift, and their strength is determined by the type of material on the slope and its water content. Rock strength varies widely and is determined by internal planes of weakness and their orientation with respect to the slope. In general, the more water, the greater the likelihood of failure. This is especially true for unconsolidated sediments, where excess water pushes the grains apart. Addition of water is the most common trigger of mass wasting, and can come from storms, rapid melting, or flooding.
15.2 Classification of Mass Wasting	The key criterion for classifying mass wasting is the nature of the movement that takes place. This may be a precipitous fall through the air, sliding as a solid mass along either a plane or a curved surface, or internal flow as a viscous fluid. The type of material that moves is also important—specifically whether it is solid rock or unconsolidated sediments. The important types of mass wasting are creep, slump, translational slide, rotational slide, fall, and debris flow or mudflow.
15.3 Preventing, Delaying, Monitoring, and Mitigating Mass Wasting	We cannot prevent mass wasting, but we can delay it through efforts to strengthen the materials on slopes. Strategies include adding mechanical devices such as rock bolts or ensuring that water can drain away. Such measures are never permanent, but may be effective for decades or even centuries. We can also avoid practices that make matters worse, such as cutting into steep slopes or impeding proper drainage. In some situations, the best approach is to mitigate the risks associated with mass wasting by constructing shelters or diversionary channels. And in other cases, where slope failure is inevitable, we should simply avoid building anything there.

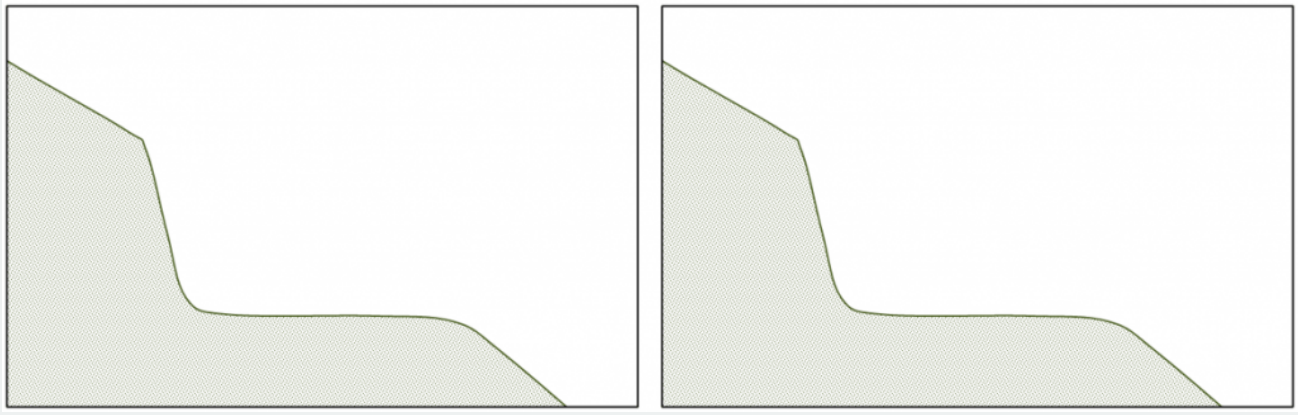
Questions for Review

Answers to Review Questions can be found in Appendix 2.



1. In the scenario shown here, the gravitational force on the unconsolidated sediment overlying the point marked with an X is depicted by the black arrow. Draw in the two arrows that show how this force can be resolved into the shear force (along the slope) and the normal force (into the slope).

2. The red arrow in the diagram depicts the shear strength of the sediment. Assuming that the relative lengths of the shear force arrow (which you drew in question 1), and the shear strength arrow are indicative of the likelihood of failure, predict whether this material is likely to fail or not.
3. After several days of steady rain, the sediment becomes saturated with water and its strength is reduced by 25%. What are the likely implications for the stability of this slope?
4. The diagrams below represent a cross-section of a road cut that has been constructed in sedimentary rock. In the first diagram, draw in the orientation of the bedding that would represent the greatest likelihood of slope failure. In the second diagram, show the orientation that would represent the least likelihood of slope failure.



5. Explain why moist sand is typically stronger than either dry sand or saturated sand.
6. In the context of mass wasting, how does a flow differ from a slide?
7. If a large rock slide starts moving at a rate of several meters per second, what is likely to happen to the rock, and what would the resulting failure be called?
8. In what ways does a debris flow differ from a typical mudflow?
9. In the situation described in the chapter regarding lahar warnings at Mount Rainier, the residents of the affected regions have to assume some responsibility and take precautions for their own safety. What sort of preparation should the residents make to ensure that they can respond appropriately when they hear lahar warnings?
10. What factors are likely to be important when considering the construction of a house near the crest of a slope that is underlain by glacial sediments?

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16: GLACIATION

16.1: Glacial Periods in Earth's History
16.2: How Glaciers Work
16.3: Glacial Erosion
16.4: Glacial Deposition
16.5: Summary

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16.1: Glacial Periods in Earth's History

We are currently in the middle of a **glacial period** (although it's less intense now than it was 20,000 years ago) but this is not the only period of glaciation in Earth's history; there have been many in the distant past, as illustrated in Figure 16.1.1. In general, however, Earth has been warm enough to be ice-free for much more of the time than it has been cold enough to be glaciated.

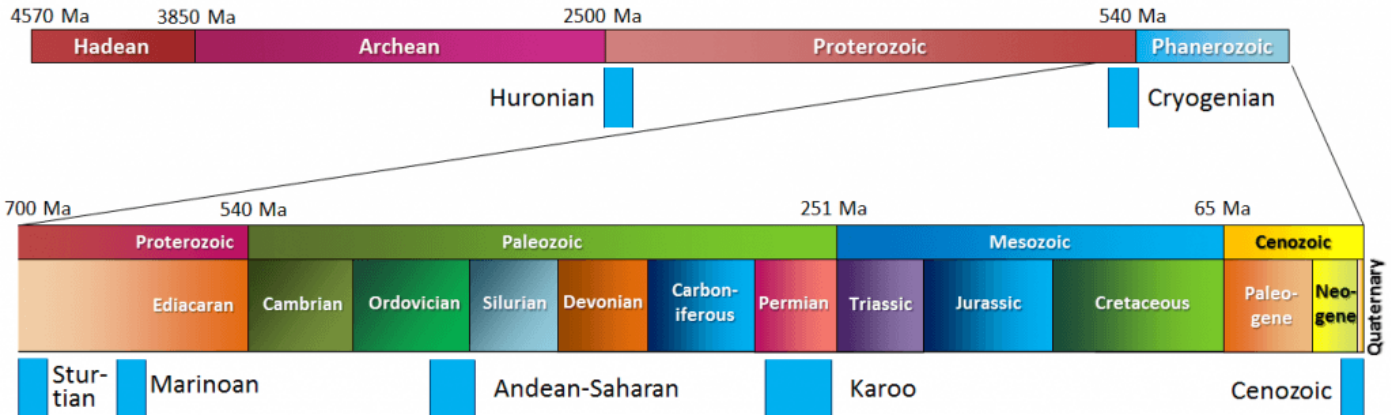


Figure 16.1.1 The record of major past glaciations during Earth's history.

The oldest known glacial period is the Huronian. Based on evidence of glacial deposits from the area around Lake Huron in Ontario and elsewhere, it is evident that the Huronian Glaciation lasted from approximately 2,400 to 2,100 Ma. Because rocks of that age are rare, we don't know much about the duration, intensity or global extent of this glaciation.

Late in the Proterozoic, for reasons that are not fully understood, the climate cooled dramatically and Earth was seized by what appears to be its most intense glaciation. The glaciations of the Cryogenian Period (*cryo* is Latin for icy cold) are also known as the "Snowball Earth" glaciations, because it is hypothesized that the entire planet was frozen—even in equatorial regions—with ice up to 1 km thick on the oceans. A visitor to our planet at that time might not have held out much hope for its inhabitability, although life still survived in the oceans. There were two main glacial periods within the Cryogenian, each lasting for about 20 million years: the Sturtian at around 700 Ma and the Marinoan at 650 Ma. There is also evidence of some shorter glaciations both before and after these. The end of the Cryogenian glaciations coincides with the evolution of relatively large and complex life forms on Earth. This started during the Ediacaran Period, and then continued with the so-called explosion of life forms in the Cambrian. Some geologists think that the changing environmental conditions of the Cryogenian are what actually triggered the evolution of large and complex life.

There have been three major glaciations during the Phanerozoic (the past 540 million years), including the Andean/Saharan (recorded in rocks of South America and Africa), the Karoo (named for rocks in southern Africa), and the Cenozoic glaciations. The Karoo was the longest of the Phanerozoic glaciations, persisting for much of the time that the supercontinent Gondwana was situated over the South Pole (~360 to 260 Ma). It covered large parts of Africa, South America, Australia, and Antarctica (see Figure 16.1.3). As you might recall from Chapter 10, this widespread glaciation, across continents that are now far apart, was an important component of Alfred Wegener's evidence for continental drift. Unlike the Cryogenian glaciations, the Andean/Saharan, Karoo, and Cenozoic glaciations only affected parts of the Earth. During Karoo times, for example, what is now North America was near the equator and remained unglaciated.

Earth was warm and essentially unglaciated throughout the Mesozoic. Although there may have been some alpine glaciation at this time, there is no longer any record of it. The dinosaurs, which dominated terrestrial habitats during the Mesozoic, did not have to endure icy conditions.

A warm climate persisted into the Cenozoic; in fact there is evidence that the Paleocene (~50 to 60 Ma) was the warmest part of the Phanerozoic since the Cambrian (Figure 16.1.2). A number of tectonic events during the Cenozoic contributed to persistent and significant planetary cooling since 50 Ma. For example, the collision of India with Asia and the formation of the Himalayan range and the Tibetan Plateau resulted in a dramatic increase in the rate of weathering and erosion. Higher than normal rates of weathering of rocks with silicate minerals, especially feldspar, consumes carbon dioxide from the atmosphere and therefore reduces the greenhouse effect, resulting in long-term cooling.

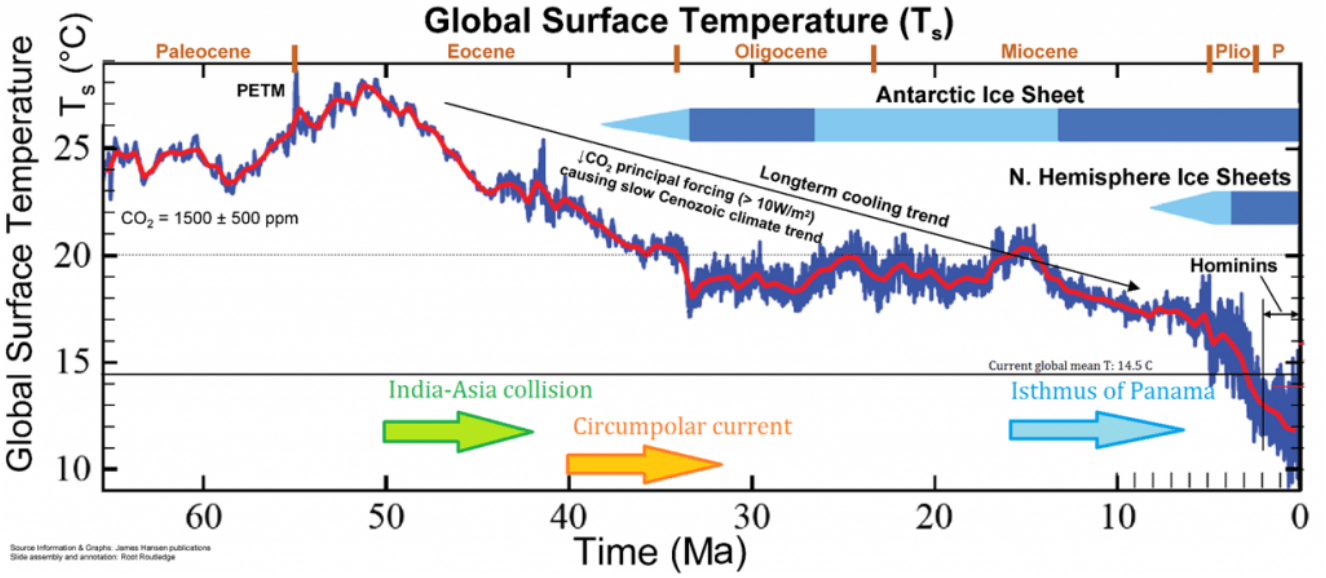


Figure 16.1.2 The global temperature trend over the past 65 Ma (the Cenozoic). From the end of the Paleocene to the height of the Pleistocene Glaciation, global average temperature dropped by about 14°C. (PETM is the Palecene-Eocene thermal maximum.)

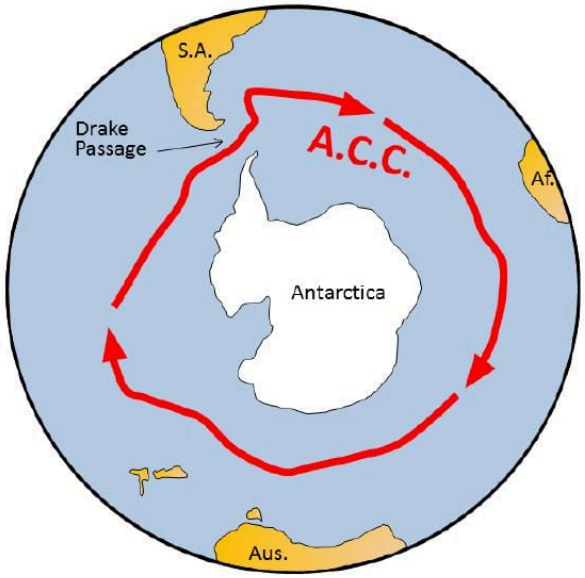


Figure 16.1.3 The Antarctic Circumpolar Current (red arrows) prevents warm water from the rest of Earth's oceans from getting close to Antarctica.

At 40 Ma, ongoing plate motion widened the narrow gap between South America and Antarctica, resulting in the opening of the Drake Passage. This allowed for the unrestricted west-to-east flow of water around Antarctica, the Antarctic Circumpolar Current (Figure 16.1.3), which effectively isolated the southern ocean from the warmer waters of the Pacific, Atlantic, and Indian Oceans. The region cooled significantly, and by 35 Ma (Oligocene) glaciers had started to form on Antarctica

Global temperatures remained relatively steady during the Oligocene and early Miocene, and the Antarctic glaciation waned during that time. At around 15 Ma, subduction-related volcanism between central and South America created the connection between North and South America, preventing water from flowing between the Pacific and Atlantic Oceans. This further restricted the transfer of heat from the tropics to the poles, leading to a rejuvenation of the Antarctic glaciation. The expansion of that ice sheet increased Earth's reflectivity enough to promote a positive feedback loop of further cooling: more reflective glacial ice, more cooling, more ice, etc. By the Pliocene (~5 Ma) ice sheets had started to grow in North America and northern Europe (Figure 16.1.4). The most intense part of the current glaciation — and the coldest climate — has been during the past million years (the last one-third of the Pleistocene), but if we count Antarctic glaciation, it really extends from the Oligocene to the Holocene, and will likely continue into the future.

The Pleistocene has been characterized by significant temperature variations (through a range of almost 10°C) on time scales of

40,000 to 100,000 years, and corresponding expansion and contraction of ice sheets. These variations are attributed to subtle changes in Earth's orbital parameters (Milankovitch cycles), which are explained in more detail in Chapter 21. Over the past million years, the glaciation cycles have been approximately 100,000 years; this variability is visible in Figure 16.1.4.

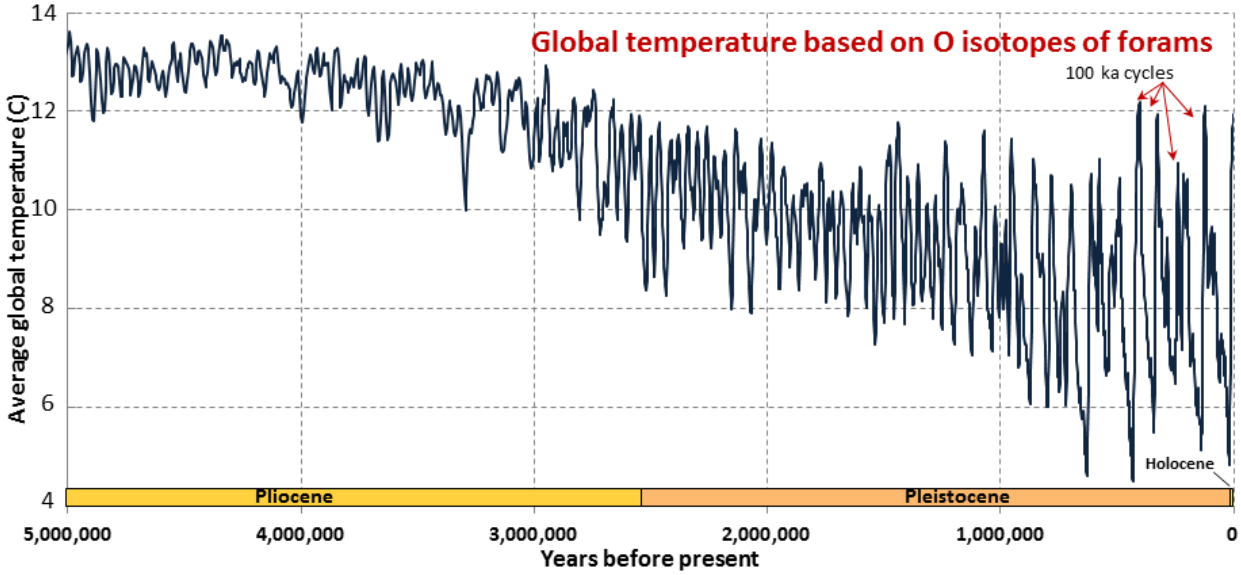


Figure 16.1.4 Foram oxygen isotope record for the past 5 million years based on O isotope data from sea-floor sediments shows a decrease in global temperture.

Exercise 16.1 Pleistocene glacials and interglacials

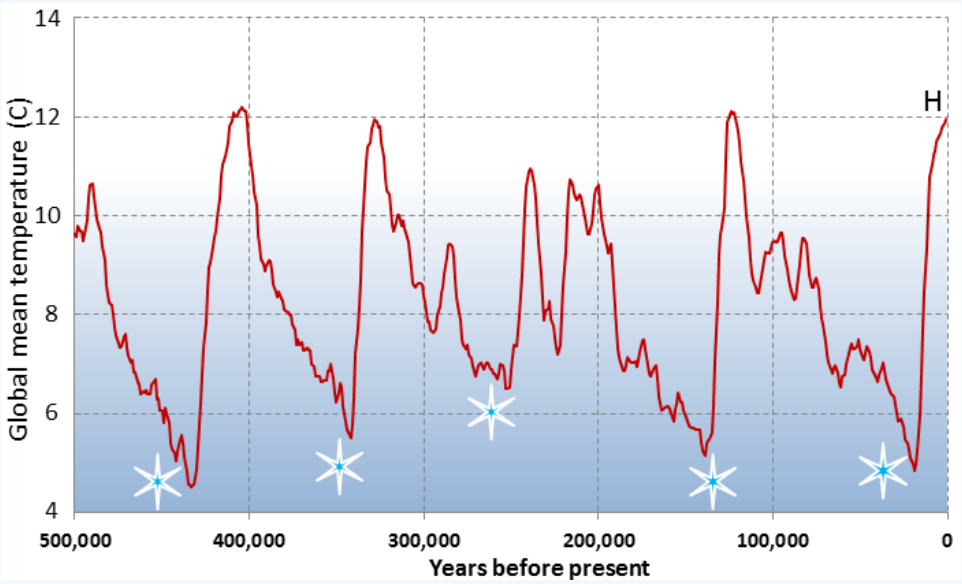


Figure 16.1.5

This diagram (Figure 16.1.5) shows the past 500,000 years of global temperature variations based on the same data used for Figure 16.1.4. The last five glacial periods are marked with snowflakes. The most recent one, which peaked at around 20 ka, is known as the Wisconsin Glaciation. Describe the nature of temperature change that followed each of these glacial periods.

The current interglacial (Holocene) is marked with an H. Point out the previous five interglacial periods.

See Appendix 3 for Exercise 16.1 answers.

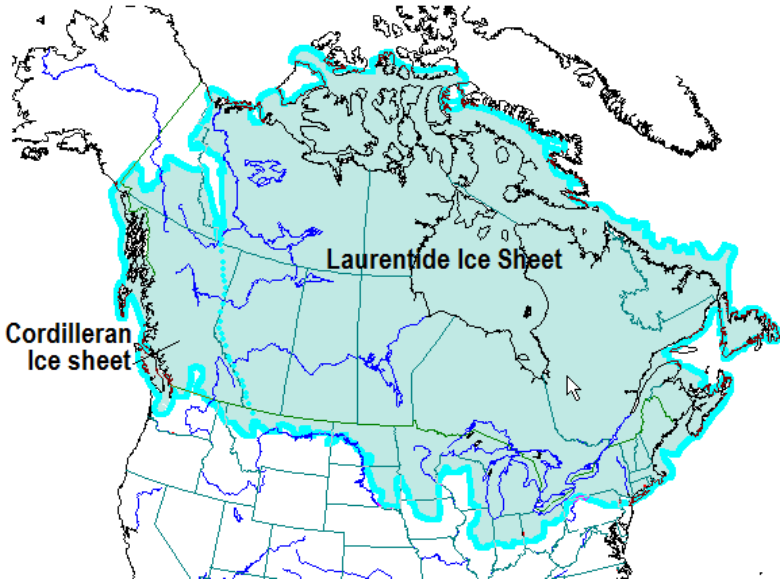


Figure 16.7 The extent of the Cordilleran and Laurentide Ice Sheets near the peak of the Wisconsin Glaciation, around 15 ka.

At the height of the last glaciation (**Wisconsin Glaciation**), massive ice sheets covered almost all of Canada and much of the northern United States (Figure 16.1.6). (In fact glacial ice extended well south of Wisconsin, into Illinois, Indiana and Ohio.) The massive **Laurentide Ice Sheet** covered most of eastern Canada (and adjacent USA), as far west as the Rockies, and the smaller **Cordilleran Ice Sheet** covered most of the western region. At various other glacial peaks during the Pleistocene and Pliocene, the ice extent was similar to this, and in some cases, even more extensive. The combined Laurentide and Cordilleran Ice Sheets were comparable in volume of ice to the current Antarctic Ice Sheet.

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16.2: How Glaciers Work



Figure 16.2.1 Part of the continental ice sheet in Greenland, with some outflow alpine glaciers in the foreground.

There are two main types of glaciers. **Continental glaciers** cover vast areas of land in extreme polar regions, including Antarctica and Greenland (Figure 16.2.1). **Alpine glaciers** (a.k.a. valley glaciers) originate on mountains, mostly in temperate and polar regions (Figure 16.2.1), but even in tropical regions if the mountains are high enough.

Earth's two great continental glaciers, on Antarctica and Greenland, comprise about 99% of all of the world's glacial ice, and approximately 68% of all of Earth's fresh water. As is evident from Figure 16.2.2, the Antarctic Ice Sheet is vastly bigger than the Greenland Ice Sheet; it contains about 17 times as much ice. If the entire Antarctic Ice Sheet were to melt, sea level would rise by about 80 m and most of Earth's major cities would be completely submerged.

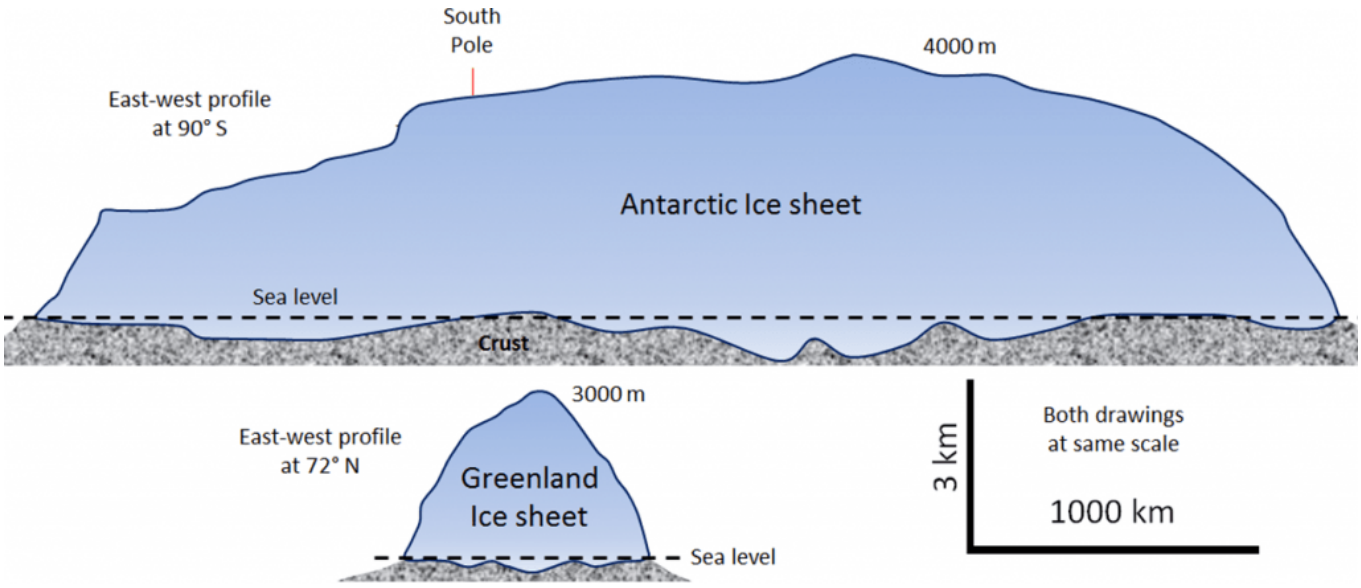


Figure 16.2.2 Simplified cross-sectional profiles the continental ice sheets in Greenland and Antarctica – both drawn to the same scale.

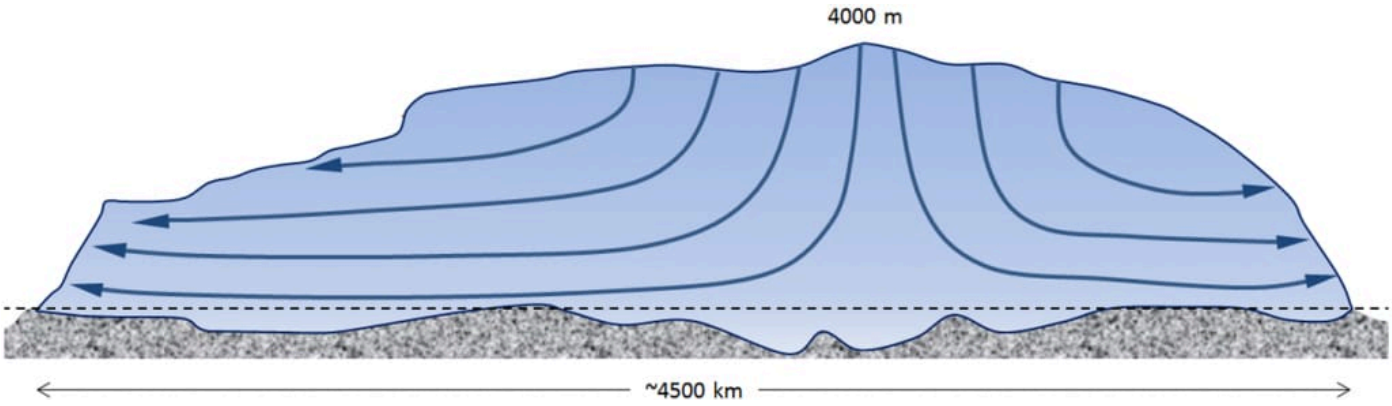


Figure 16.2.3 Schematic ice-flow diagram for the Antarctic Ice Sheet.

Continental glaciers do not flow “downhill” because the large areas that they cover are generally flat. Instead, ice flows from the region where it is thickest toward the edges where it is thinner, as shown in Figure 16.2.3. This means that in the central thickest parts, the ice flows almost vertically down toward the base, while in the peripheral parts, it flows out toward the margins. In continental glaciers like Antarctica and Greenland, the thickest parts (4,000 m and 3,000 m respectively) are the areas where the rate of snowfall and therefore the rate of ice accumulation are highest.

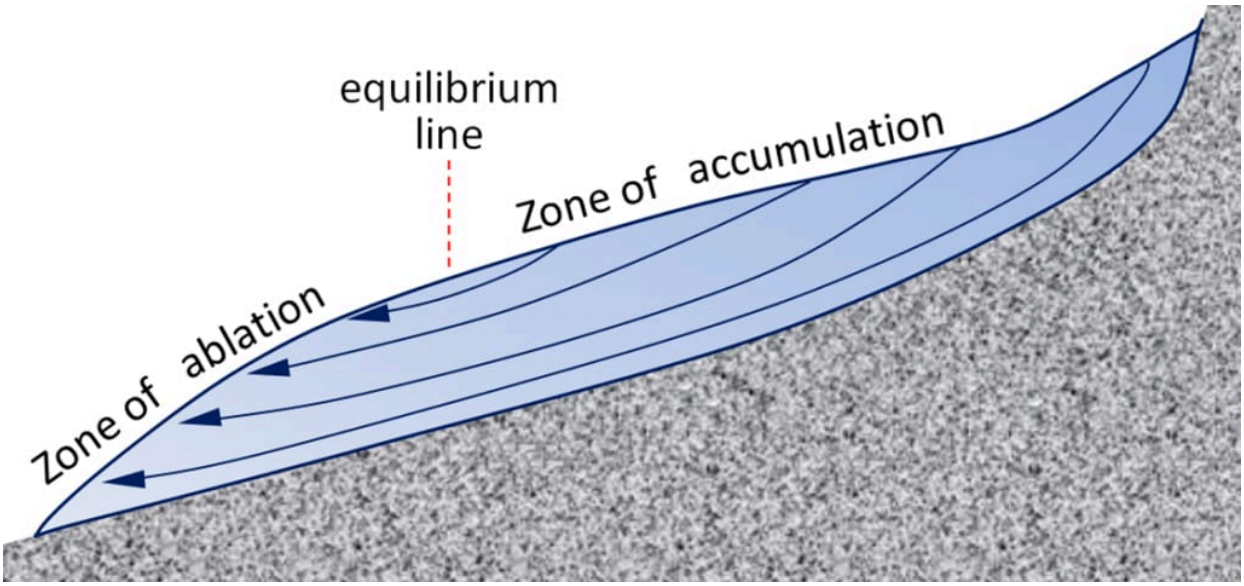


Figure 16.2.4 Schematic ice-flow diagram for an alpine glacier.

The flow of alpine glaciers is primarily controlled by the slope of the land beneath the ice (Figure 16.2.4). In the **zone of accumulation**, the rate of snowfall is greater than the rate of melting. In other words, not all of the snow that falls each winter melts during the following summer, and the ice surface is always covered with snow. In the **zone of ablation**, more ice melts than accumulates as snow. The **equilibrium line** marks the boundary between the zones of accumulation (above) and ablation (below).

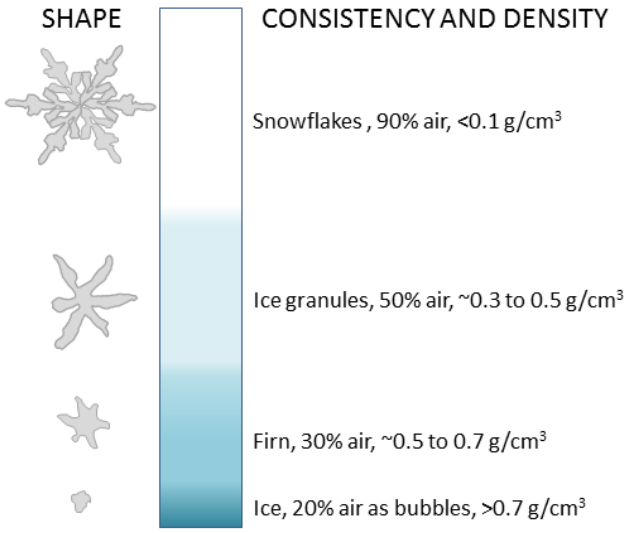


Figure 16.2.5 Steps in the process of formation of glacial ice from snow, granules, and firn.

Above the equilibrium line of a glacier, not all of the winter snow melts in the following summer, so snow gradually accumulates. The snow layer from each year is covered and compacted by subsequent snow, and it is gradually compressed and turned into **firn** within which the snowflakes lose their delicate shapes and become granules. With more compression, the granules are pushed together and air is squeezed out. Eventually the granules are "welded" together to create glacial ice (Figure 16.2.5). Downward percolation of water from melting taking place at the surface contributes to the process of ice formation.

The equilibrium line of a glacier near Whistler, B.C., is shown in Figure 16.2.6. Below that line, in the zone of ablation, bare ice is exposed because last winter's snow has all melted; above that line, the ice is still mostly covered with snow from last winter. The position of the equilibrium line changes from year to year as a function of the balance between snow accumulation in the winter and snowmelt during the summer. More winter snow and less summer melting obviously favours the advance of the equilibrium line (and of the glacier's leading edge), but of these two variables, it is the summer melt that matters most to a glacier's budget. Cool summers promote glacial advance and warm summers promote glacial retreat.



Figure 16.2.6 The approximate location of the equilibrium line (red) in September 2013 on the Overlord Glacier, near Whistler, B.C.

Glaciers move because the surface of the ice is sloped. This generates a stress on the ice, which is proportional to the slope and to the depth below the surface. As shown in Figure 16.2.6, the stresses are quite small near the ice surface but much larger at depth, and also greater in areas where the ice surface is relatively steep. Ice will deform, meaning that it will behave in a plastic manner, at stress levels of around 100 kilopascals; therefore, in the upper 50 m to 100 m of the ice (above the dashed red line),

flow is not plastic (the ice is rigid), while below that depth, ice is plastic and will flow.

When the lower ice of a glacier flows, it moves the upper ice along with it, so although it might seem from the stress patterns (red numbers and red arrows) shown in Figure 16.2.7 that the lower part moves the most, in fact while the lower part deforms (and flows) and the upper part doesn't deform at all, the upper part moves the fastest because it is pushed along by the lower ice.

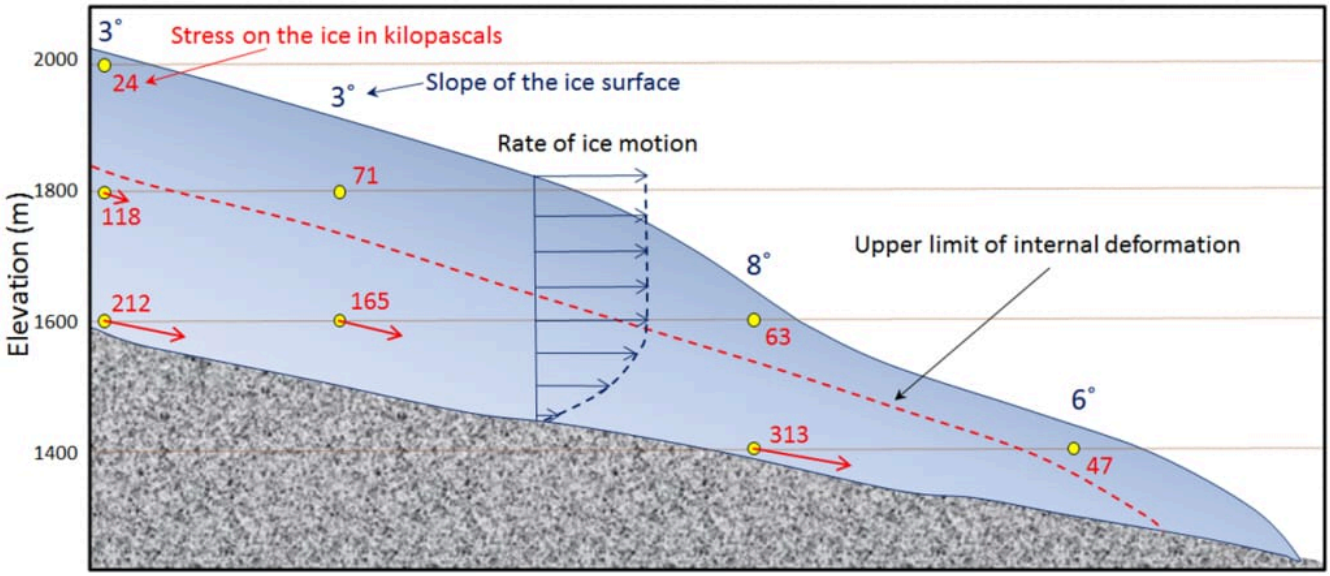


Figure 16.2.7 Stress within a valley glacier (red numbers) as determined based on the slope of the ice surface and the depth within the ice. The ice will deform and flow where the stress is greater than 100 kilopascals, and the relative extent of that deformation is depicted by the red arrows. Any deformation motion in the lower ice will be transmitted to the ice immediately above it, so although the red stress arrows get shorter toward the top, the ice velocity increases upward (blue arrows). The upper ice (above the red dashed line) does not flow, but it is pushed along with the lower ice.

The plastic lower ice of a glacier can flow like a very viscous fluid, and can therefore flow over irregularities in the base of the ice and around corners. However, the upper rigid ice cannot flow in this way, and because it is being carried along by the lower ice, it tends to crack where the lower ice has to flex. This leads to the development of **crevasses** in areas where the rate of flow of the plastic ice is changing. In the area shown in Figure 16.2.8, for example, the glacier is speeding up over the steep terrain, and the rigid surface ice has to crack to account for the change in velocity.



Figure 16.2.8 Crevasses on Overlord Glacier in the Whistler area, B.C.

The plastic lower ice of a glacier can flow like a very viscous fluid, and can therefore flow over irregularities in the base of the ice

and around corners. However, the upper rigid ice cannot flow in this way, and because it is being carried along by the lower ice, it tends to crack where the lower ice has to flex. This leads to the development of **crevasses** in areas where the rate of flow of the plastic ice is changing. In the area shown in Figure 16.2.8, for example, the glacier is speeding up over the steep terrain, and the rigid surface ice has to crack to account for the change in velocity

The base of a glacier can be cold (below the freezing point of water) or warm (above the freezing point). If it is warm, there will likely be a film of water between the ice and the material underneath, and the ice will be able to slide over that surface. This is known as **basal sliding** (Figure 16.2.9, left). If the base is cold, the ice will be frozen to the material underneath and it will be stuck—unable to slide along its base. In this case, all of the movement of the ice will be by internal flow.

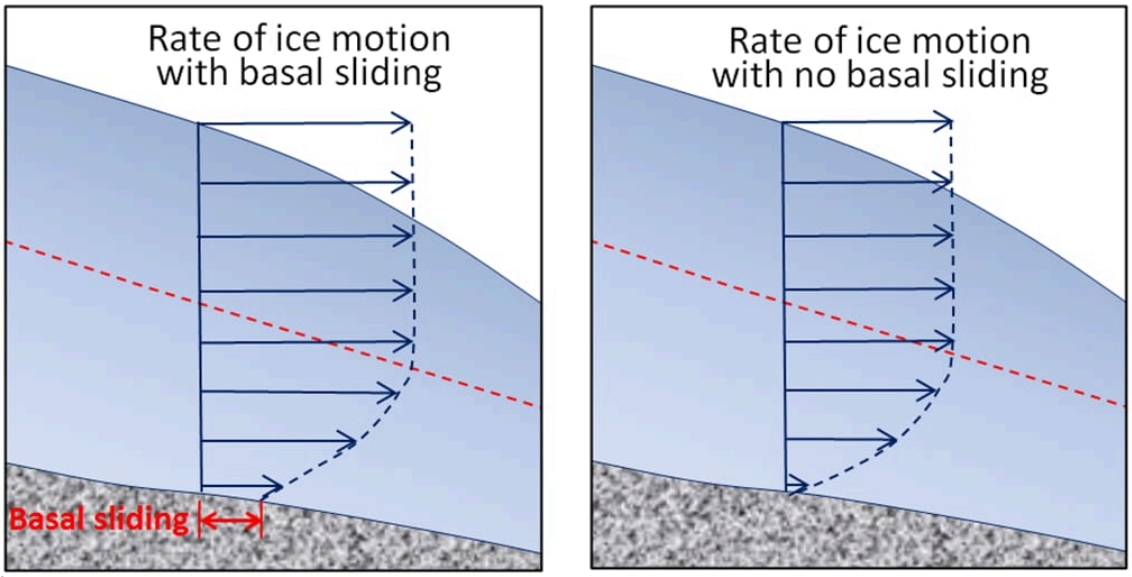


Figure 16.2.9 Differences in glacial ice motion with basal sliding (left) and without basal sliding (right). The dashed red line indicates the upper limit of plastic internal flow.

One of the factors that affects the temperature at the base of a glacier is the thickness of the ice. Ice is a good insulator. The slow transfer of heat from Earth's interior provides enough heat to warm up the base if the ice is thick, but not enough if it is thin and that heat can escape. It is typical for the leading edge of an alpine glacier to be relatively thin (see Figure 16.2.7), so it is common for that part to be frozen to its base while the rest of the glacier is still sliding. This is illustrated in Figure 16.2.10 for the Athabasca Glacier. Because the leading edge of the glacier is stuck to its frozen base, while the rest continues to slide, the ice coming from behind has pushed (or thrust) itself over top of the part that is stuck fast.

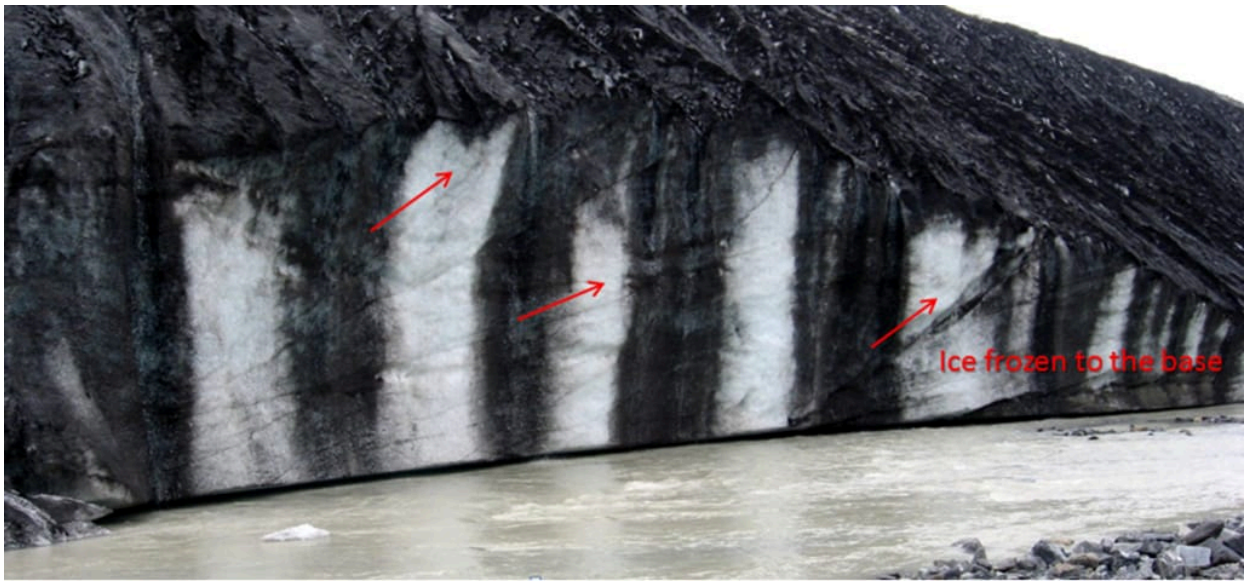


Figure 16.2.10 Thrust faults at the leading edge of the Athabasca Glacier, Alberta. The arrows show how the trailing ice has been thrust over the leading ice. (The dark vertical stripes are mud from sediments that have been washed off of the lateral moraine lying on the surface of the ice.)

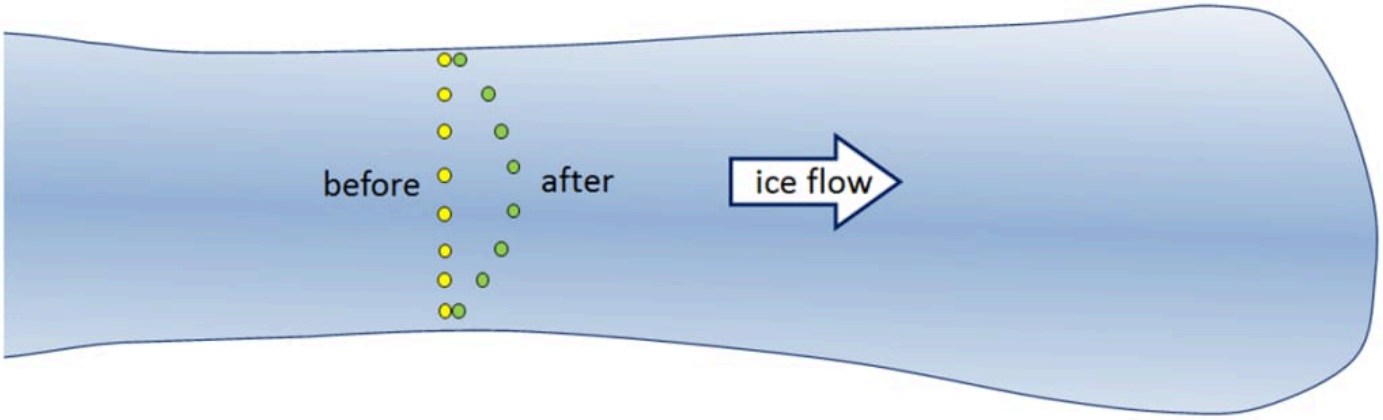


Figure 16.2.11 Markers on an alpine glacier move forward over a period of time with the ones in the middle moving faster than the ones on the edge.

Just as the base of a glacier moves more slowly than the surface, the edges, which are more affected by friction along the sides, move more slowly than the middle. If we were to place a series of markers across an alpine glacier and come back a year later, we would see that the ones in the middle had moved farther forward than the ones near the edges (Figure 16.2.11).

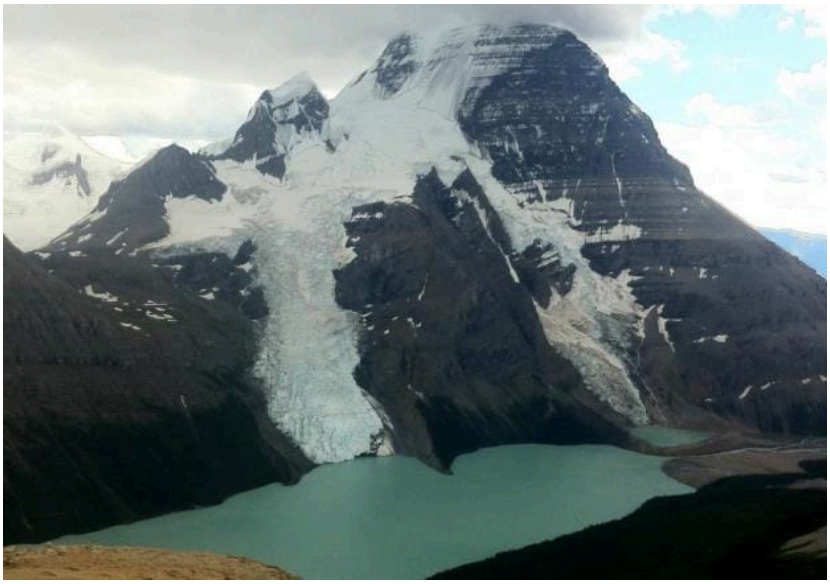


Figure 16.2.12 Mt. Robson, the tallest peak in the Canadian Rockies, Berg Glacier (centre), and Berg Lake. Although there were no icebergs visible when this photo was taken, the Berg Glacier loses mass by shedding icebergs into Berg Lake.

Glacial ice always moves downhill, in response to gravity, but the front edge of a glacier is always either melting or **calving** into water (shedding icebergs). If the rate of forward motion of the glacier is faster than the rate of **ablation** (melting), the leading edge of the glacier advances (moves forward). If the rate of forward motion is about the same as the rate of ablation, the leading edge remains stationary, and if the rate of forward motion is slower than the rate of ablation, the leading edge retreats (moves backward).

Calving of icebergs is an important process for glaciers that terminate in lakes or the ocean. An example of such a glacier is the Berg Glacier on Mt. Robson (Figure 16.2.12), which sheds small icebergs into Berg Lake. The Berg Glacier also loses mass by melting, especially at lower elevations.

Exercise 16.2 Ice advance and retreat

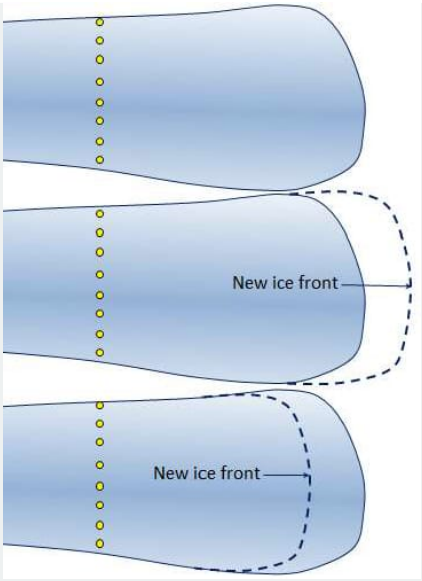


Figure 16.2.13

These diagrams represent a glacier with markers placed on its surface to determine the rate of ice motion over a one-year period. The ice is flowing from left to right.

1. In the middle diagram, the leading edge of the glacier has advanced. Draw in the current position of the markers.
2. In the lower diagram, the leading edge of the glacier has retreated. Draw in the current position of the markers.

See Appendix 3 for Exercise 16.2 answers.

Image Descriptions

Figure 16.2.2 image description: An east to west cross section of the ice on Antarctica is over 4000 km long and up to 4 km high in some places. An east to west cross section of the ice on Greenland is less than 1000 km long and up to 3 km high. [Return to Figure 16.2.2]

Figure 16.2.5 image description: The comparative density of snowflakes, ice granules, firn, and ice.

Shape	Amount of air	Density (grams per centimeters cubed)
Snowflake	90%	Less than 0.1
Ice Granules	50%	From around 0.3 to 0.5
Firn	30%	From around 0.5 to 0.7
Ice	20% (as bubbles)	Greater than 0.7

[Return to Figure 16.2.5]

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- Figure 16.2.8: “Crevasses at Overlord Glacier” by Isaac Earle. CC BY.
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16.3: Glacial Erosion

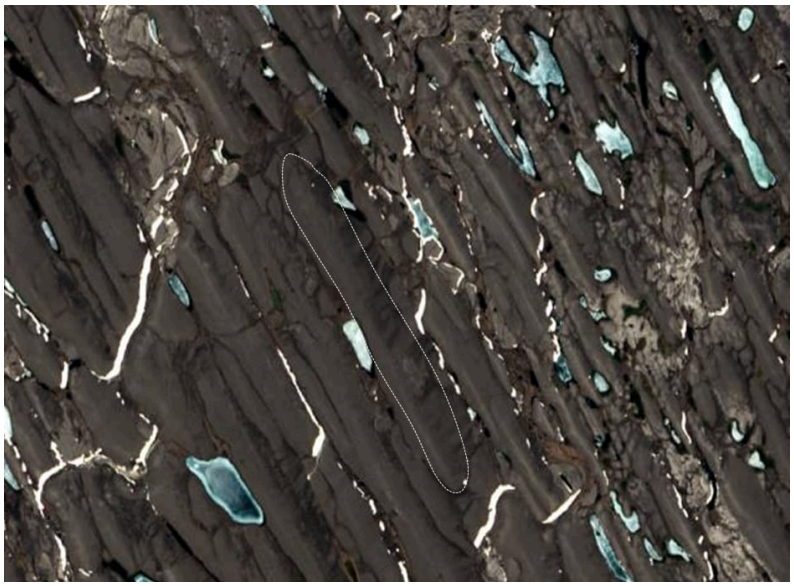


Figure 16.3.1 Drumlins — streamlined hills formed beneath a glacier, here made up of sediment — in the Amundsun Gulf region of Nunavut. The drumlins are tens of meters high, a few hundred meters across, and a few kilometers long. One of them is highlighted with a dashed white line.

Glaciers are effective agents of erosion, especially in situations where the ice is not frozen to its base and can therefore slide over the bedrock or other sediment. In fact the ice itself is not particularly effective at erosion because it is relatively soft (Mohs hardness 1.5 at 0°C); instead, it is the rock fragments embedded in the ice and pushed down onto the underlying surfaces that do most of the erosion. A useful analogy would be to compare the effect of a piece of paper being rubbed against a wooden surface, as opposed to a piece of sandpaper that has embedded angular fragments of garnet.

The results of glacial erosion are different in areas with continental glaciation versus alpine glaciation. Continental glaciation tends to produce relatively flat bedrock surfaces, especially where the rock beneath is uniform in strength. In areas where there are differences in the strength of rocks, a glacier obviously tends to erode the softer and weaker rock more effectively than the harder and stronger rock. Much of central and eastern Canada, which was completely covered by the huge Laurentide Ice Sheet at various times during the Pleistocene, has been eroded to a relatively flat surface. In many cases the existing relief is due the presence of glacial deposits—such as drumlins, eskers, and moraines (all discussed below)—rather than to differential erosion (Figure 16.3.1).

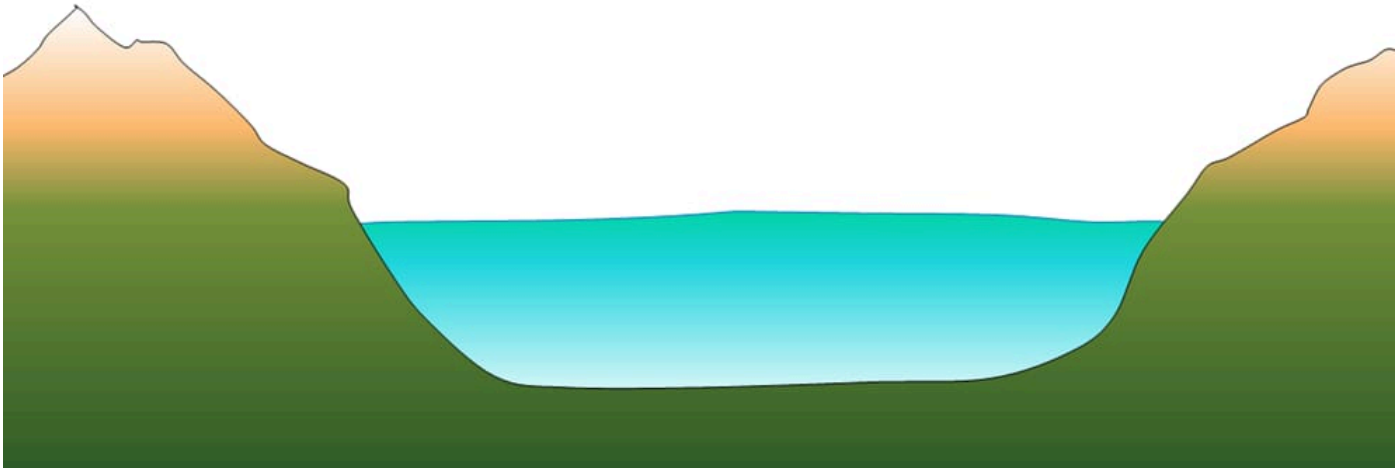


Figure 16.3.2 A depiction of a U-shaped valley occupied by a large glacier.

Alpine glaciers produce very different topography than continental glaciers, and much of the topographic variability of western Canada can be attributed to glacial erosion. In general, glaciers are much wider than rivers of similar length, and since they tend to erode more at their bases than their sides, they produce wide valleys with relatively flat bottoms and steep sides—known as **U-shaped valleys** (Figure 16.3.2). Howe Sound, north of Vancouver, was occupied by a large glacier that originated in the Squamish, Whistler, and Pemberton areas, and then joined the much larger glacier in the Strait of Georgia. Howe Sound and most of its tributary valleys have pronounced U-shaped profiles (Figure 16.3.3).

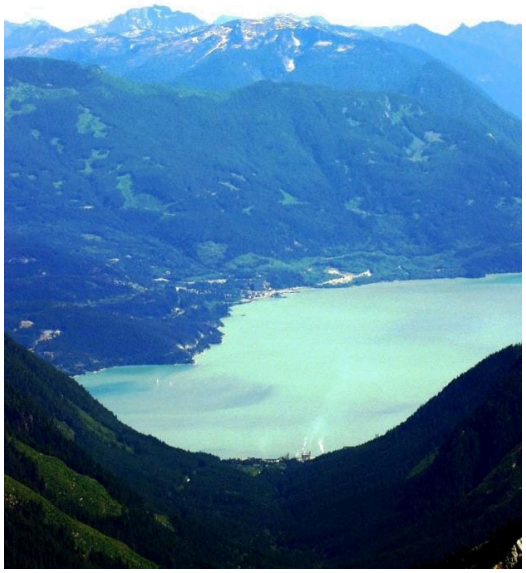


Figure 16.3.3 The view down the U-shaped valley of Mill Creek valley toward the U-shaped valley of Howe Sound (north of Vancouver BC), with the village of Britannia on the opposite side.

U-shaped valleys and their tributaries provide the basis for a wide range of alpine glacial topographic features, examples of which are visible on the International Space Station view of the Swiss Alps shown in Figure 16.3.4. This area was much more intensely glaciated during the past glacial maximum. At that time, the large U-shaped valley in the lower right was occupied by glacial ice, and all of the other glaciers shown here were longer and much thicker than they are now. But even at the peak of the Pleistocene Glaciation, some of the higher peaks and ridges would have been exposed and not directly affected by glacial erosion. A peak that extends above the surrounding glacier is called a **nunatuk**. In these areas, and in the areas above the glaciers today, most of the erosion is related to freeze-thaw effects.

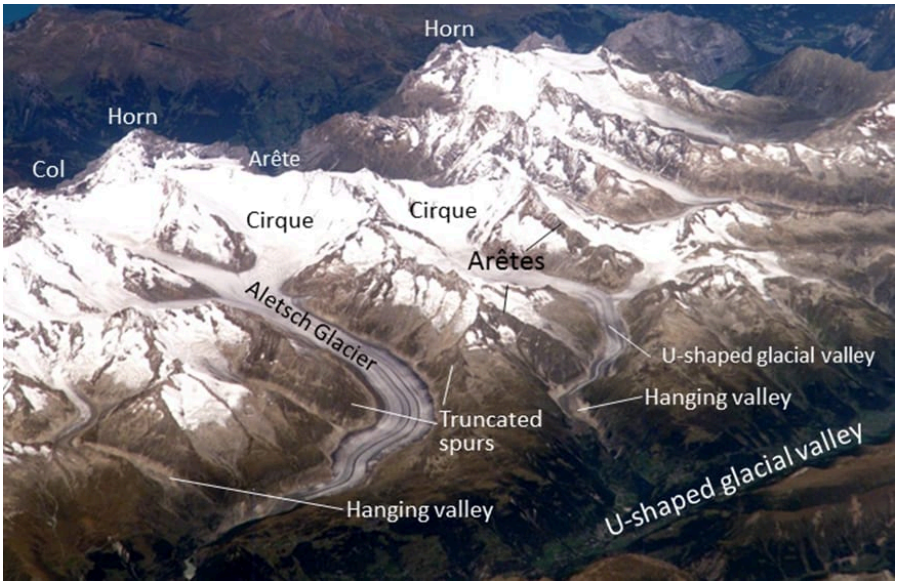


Figure 16.3.4 A view from the International Space Station of the Swiss Alps in the area of the Aletsch Glacier. The prominent peaks labelled "Horn" are the Eiger (left) and Wetterhorn (right). A variety of alpine glacial erosion features are labelled.

Figure 16.3.4 is a Space Station view of a glaciated terrain in the Swiss Alps. Some of the important features visible are **arêtes**: sharp ridges between U-shaped glacial valleys; **cols**: low points along arêtes that constitute passes between glacial valleys; **horns**: steep peaks that have been glacially and freeze-thaw eroded on three or more sides; **cirques**: bowl-shaped basins that form at the head of a glacial valley; **hanging valleys**: U-shaped valleys of tributary glaciers that hang above the main valley because the larger main-valley glacier eroded more deeply into the terrain; and **truncated spurs** (a.k.a. "spurs"): the ends of arêtes that have been eroded into steep triangle-shaped cliffs by the glacier in the corresponding main valley.

Some of these alpine-glaciation erosional features are also shown in Figure 16.3.5 in diagram form.

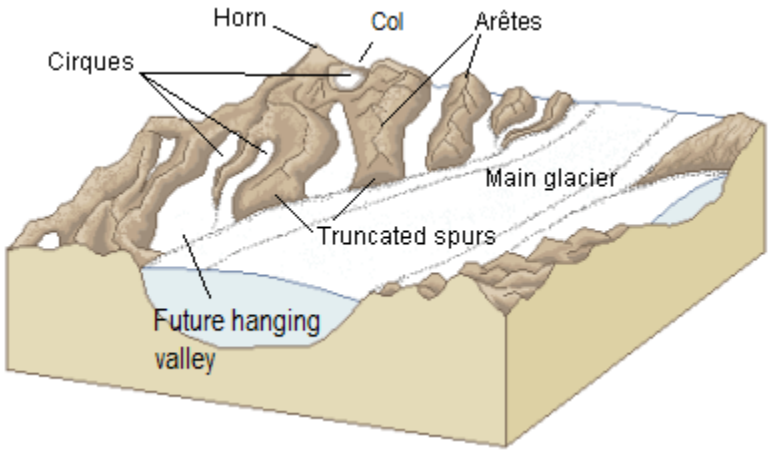


Figure 16.3.5 A diagram of some of the important alpine-glaciation erosion features.

Exercise 16.3 Identify alpine glacial erosion features

This is a photo of Mt. Assiniboine in the B.C. Rockies. What are the features at locations **a** through **e**? Look for one of each of the following: a horn, an arête, a truncated spur, a cirque, and a col. Try to identify some of the numerous other arêtes in this view, as well as another horn.

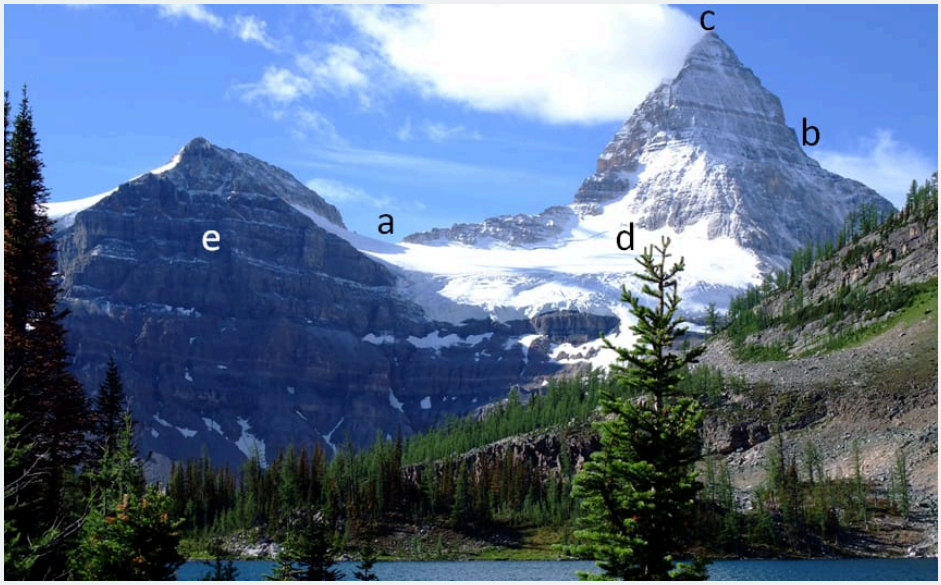


Figure 16.3.6

See Appendix 3 for Exercise 16.3 answers.

A number of other glacial erosion features exist at smaller scales. For example, a **drumlin** is an elongated feature that is streamlined at the down-ice end. The one shown in Figure 16.3.7 is larger than most, and is made up almost entirely of rock. Drumlins made up of glacial sediments are very common in some areas of continental glaciation (Figure 16.3.1).



Figure 16.3.7 Bowyer Island, a drumlin in Howe Sound, B.C. Ice flow was from right to left.

A **roche moutonée** is another type of elongated erosional feature that has a steep and sometimes jagged down-ice end (Figure 16.3.8, left). On a smaller scale still, **glacial grooves** (tens of centimeters to meters wide) and **glacial striation** (millimeters to centimeters wide) are created by fragments of rock embedded in the ice at the base of a glacier (Figure 16.3.8, left and right). Glacial striae are very common on rock surfaces eroded by both alpine and continental glaciers.



Figure 16.3.8 Left: Roches moutonées with glacial striae near Squamish, B.C. Right: Glacial striae at the same location near Squamish. Ice flow was from right to left in both cases. The brown rock in the right-hand photo is a mafic dyke about 40 cm wide intruding into granite.

Glacial lakes



Figure 16.3.9 Lower Thornton Lake, a tarn, in the Northern Cascades National Park, Washington.

Lakes are common features in glacial environments. A lake that is confined to a glacial cirque is known as a **tarn** (Figure 16.3.9). Tarns are common in areas of alpine glaciation because the ice that forms a cirque typically carves out a depression in bedrock that then fills with water. In some cases, a series of such basins will form, and the resulting lakes are called **rock basin lakes** or **paternoster lakes**.

A lake that occupies a glacial valley, but is not confined to a cirque, is known as a **finger lake**. In some cases, a finger lake is confined by a dam formed by an end moraine, in which case it may be called a **moraine lake** (Figure 16.3.10).



Figure 16.3.10 Peyto Lake in the Alberta Rockies, is both a finger lake and a moraine lake as it is dammed by an end moraine, on the right.

In areas of continental glaciation, the crust is depressed by the weight of glacial ice that is up to 4,000 m thick. Basins are formed along the edges of continental glaciers (except for those that cover entire continents like Antarctica and Greenland), and these basins fill with glacial meltwater. Many such lakes, some of them huge, existed at various times along the southern edge of the Laurentide Ice Sheet. One example is Glacial Lake Missoula, which formed within Idaho and Montana in northwestern United States. During the latter part of the last glaciation (30 ka to 15 ka), the ice holding back Lake Missoula retreated enough to allow some of the lake water to start flowing out, which escalated into a massive and rapid outflow (over days to weeks) during which much of the volume of the lake drained along the valley of the Columbia River to the Pacific Ocean. It is estimated that this type of flooding happened at least 25 times over that period, and in many cases, the rate of outflow was equivalent to the discharge

of all of Earth's current rivers combined. The record of these massive floods is preserved in the Channelled Scablands of Idaho, Washington, and Oregon (Figure 16.3.11).



Figure 16.3.11 Potholes Coulee near Wenatchee, Washington, one of many basins that received Lake Missoula floodwaters during the late Pleistocene. Here the water flowed from right to left, over the cliff and into this basin.

Another type of glacial lake is a **kettle lake**. These are discussed in section 16.4 in the context of glacial deposits.

Image Descriptions

Figure 16.3.6 image description:

1. A valley between two peaks.
2. A steep edge of a mountain.
3. A mountain peak.
4. An ice field in between two mountains.
5. A flat, rocky side of a mountain.

[Return to Figure 16.3.6]

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16.4: Glacial Deposition

Sediments transported and deposited during the Pleistocene glaciations are abundant throughout Canada and much of the northern USA. They are important sources of construction materials and are valuable as reservoirs for groundwater. Because they are almost all unconsolidated, they have significant implications for mass wasting.



Figure 16.4.1 Part of the Bering Glacier in southeast Alaska, the largest glacier in North America. It is about 14 km across in the centre of this view.

Figure 16.4.1 illustrates some of the ways that sediments are transported and deposited. The Bering Glacier is the largest in North America, and although most of it is in Alaska, it flows from an icefield that extends into southwestern Yukon. The surface of the ice is partially, or in some cases completely covered with rocky debris that has fallen from surrounding steep rock faces. There are muddy rivers issuing from the glacier in several locations, depositing sediment on land, into Vitus Lake, and directly into the ocean. There are dirty icebergs shedding their sediment into the lake. And, not visible in this view, there are sediments being moved along beneath the ice.

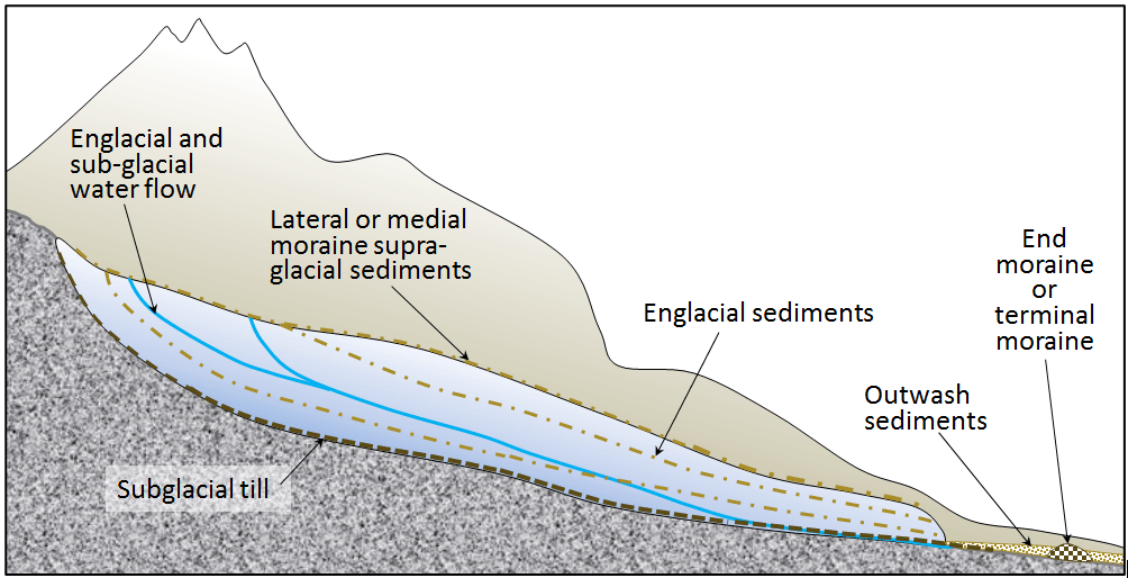


Figure 16.4.2 A depiction of the various types of sediments associated with glaciation. The glacier is shown in cross-section.

The formation and movement of sediments in glacial environments is shown diagrammatically in Figure 16.4.2. There are many types of glacial sediment generally classified by whether they are transported on, within, or beneath the glacial ice. The main types of sediment in a glacial environment are described below.

Supraglacial (on top of the ice) and **Supraglacial** (within the ice) sediments that slide off the melting front of a stationary glacier can form a ridge of unsorted sediments called an **englacial**. The end moraine that represents the farthest advance of the glacier is a **end moraine**. Supraglacial and englacial sediments can also be deposited when the ice melts. Sediments transported and deposited by glacial ice are known as **till**.

Subglacial sediment (e.g., **lodgement till**) is material that has been eroded from the underlying rock by the ice, and is moved by the ice. It has a wide range of grain sizes (in other words it is poorly sorted), including a relatively high proportion of silt and clay. The larger clasts (pebbles to boulders in size) tend to become partly rounded by abrasion. Lodgement till forms as a sheet of well-compacted sediment beneath a glacier, and ranges from several centimeters to many meters in thickness. Lodgement till is normally unbedded. An example is shown in Figure 16.4.3a.

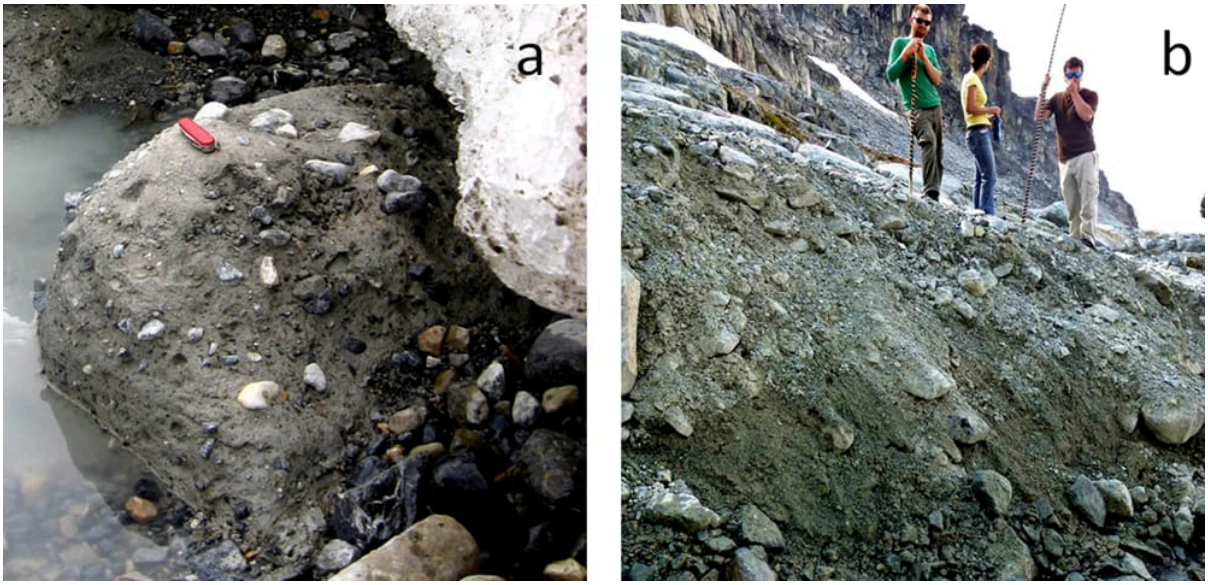


Figure 16.4.3 Examples of glacial till: a: lodgement till from the front of the Athabasca Glacier, Alberta; b: ablation till at the Horstman Glacier, Blackcomb Mountain, B.C.

Supraglacial sediments are primarily derived from freeze-thaw eroded material that has fallen onto the ice from rocky slopes above. These sediments form **lateral moraines** (Figure 16.4.1) and, where two glaciers meet, **medial moraines**. (Medial moraines are visible on the Aletsch Glacier in Figure 16.4.4.) Most of this material is deposited on the ground when the ice melts, and is therefore called **ablation till**, a mixture of fine and coarse angular rock fragments, with much less sand, silt, and clay than lodgement till. An example is shown in Figure 16.4.3b. When supraglacial sediments become incorporated into the body of the glacier, they are known as englacial sediments (Figure 16.4.2).

Massive amounts of water flow on the surface, within, and at the base of a glacier, even in cold areas and even when the glacier is advancing. Depending on its velocity, this water is able to move sediments of various sizes and most of that material is washed out of the lower end of the glacier and deposited as outwash sediments. These sediments accumulate in a wide range of environments in the **proglacial** region (the area in front of a glacier), most in fluvial environments, but some in lakes and the ocean. **Glaciofluvial sediments** are similar to sediments deposited in normal fluvial environments, and are dominated by silt, sand, and gravel. The grains tend to be moderately well rounded, and the sediments have similar sedimentary structures (e.g., bedding, cross-bedding, clast imbrication) to those formed by non-glacial streams (Figure 16.4.4a and 16.4.4b).

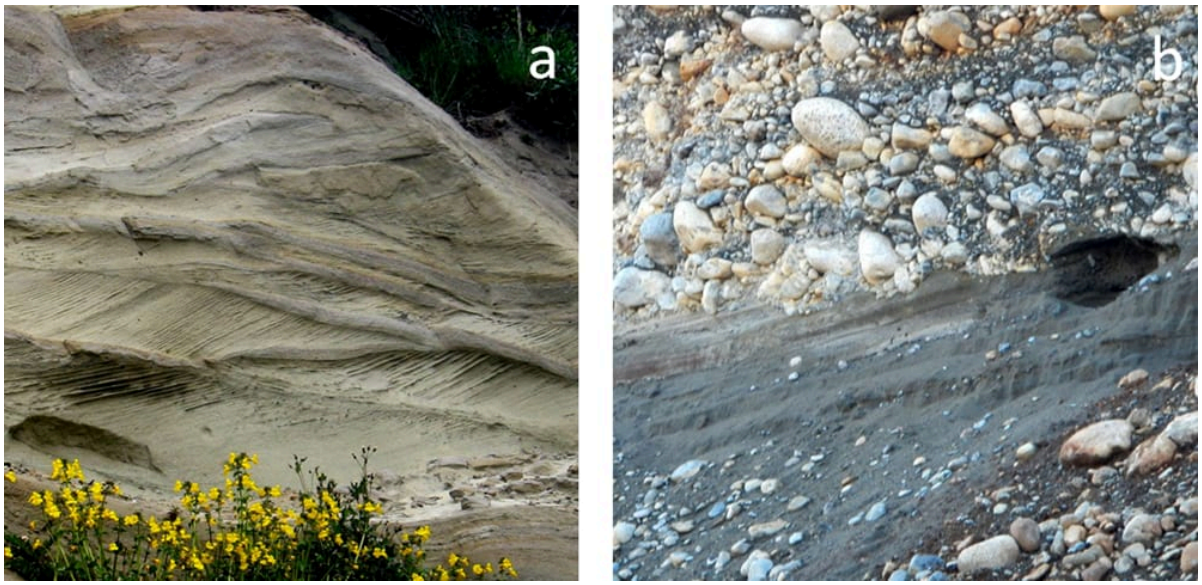


Figure 16.4.4 Examples of glaciofluvial sediments: a: glaciofluvial sand of the Quadra Sand Formation at Comox, B.C.; b: glaciofluvial gravel and sand, Nanaimo, B.C.

A large proglacial plain of sediment is called a **sandur** (a.k.a. an **outwash plain**)**,** and within that area, glaciofluvial deposits can be tens of meters thick (Figure 16.4.5). In situations where a glacier is receding, a block of ice might become separated from the main ice sheet and then get buried in glaciofluvial sediments. When the ice block eventually melts, a depression forms, known as a **kettle**, and if this fills with water, it is known as a **kettle lake** (Figure 16.4.6).



Figure 16.4.5 Part of a sandur in front of the Vatnajokull Glacier in Iceland,. The Sandur extends for many tens of km. Here is has been partially eroded by a stream (not visible).



Figure 16.4.6 A kettle lake amid vineyards and orchards in the Osoyoos area of B.C.



Figure 16.4.7 Part of an esker that formed beneath the Laurentide Ice Sheet in northern Canada.

A subglacial stream will create its own channel within the ice, and sediments that are being transported and deposited by the stream will build up within that channel. When the ice recedes, the sediment will remain to form a long sinuous ridge known as an **esker**. Eskers are most common in areas of continental glaciation. They can be several meters high, tens of meters wide, and tens of kilometers long (Figure 16.4.7).

Outwash streams commonly flow into proglacial lakes where **glaciolacustrine sediments** are deposited. These are dominated by silt- and clay-sized particles and are typically laminated on the millimetre scale. In some cases, **varves** develop; varves are series of beds with distinctive summer and winter layers: relatively coarse in the summer when melt discharge is high, and finer in the winter, when discharge is very low. Icebergs are common on proglacial lakes, and most of them contain englacial sediments of various sizes. As the bergs melt, the released clasts sink to the bottom and are incorporated into the glaciolacustrine layers as **drop stones** (Figure 16.4.8a).

The processes that occur in proglacial lakes can also take place where a glacier terminates at the ocean. The sediments deposited there are called **glaciomarine sediments** (Figure 16.4.8b).

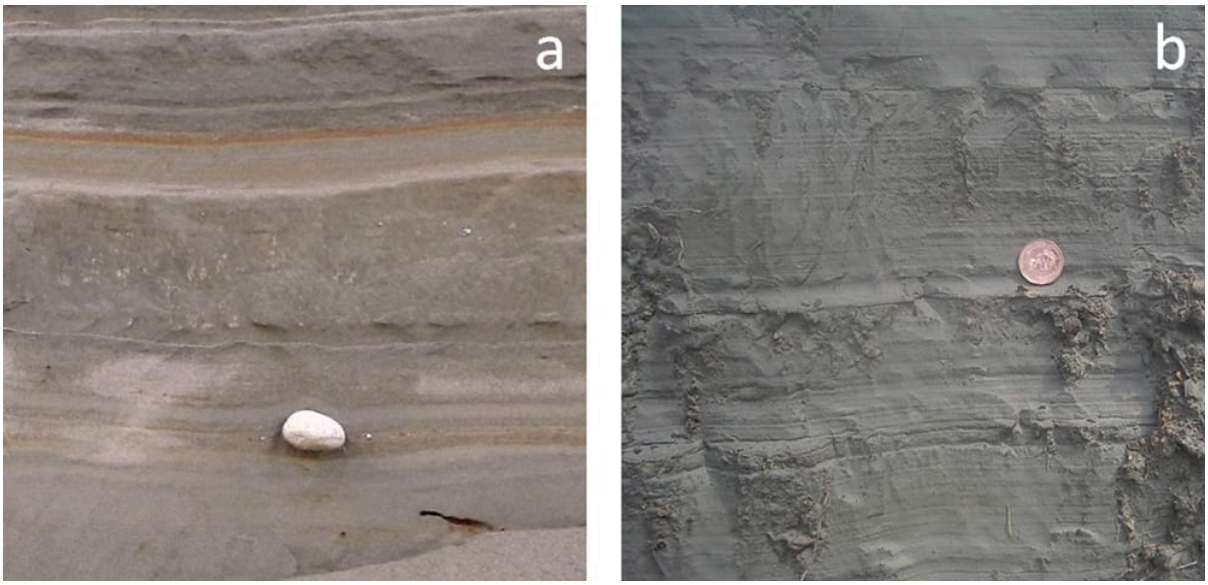


Figure 16.4.8 Examples of glacial sediments formed in quiet water: a: glaciolacustrine sediment with a drop stone, Nanaimo, B.C.; and b: a laminated glaciomarine sediment, Englishman River, B.C. Although not visible in this photo, the glaciomarine sediment has marine shell fossils.

Exercise 16.4 Identify glacial depositional environments



Figure 16.4.9

This photo shows the Bering Glacier in Alaska (same as Figure 16.4.1).

Glacial sediments of many different types are being deposited throughout this area. Identify where you would expect to find the following types of deposits:

1. Glaciofluvial sand
2. Lodgement till
3. Glaciolacustrine clay with drop stones
4. Ablation till
5. Glaciomarine silt and clay

See Appendix 3 for Exercise 16.4 answers.

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16.5: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
16.1 Glacial Periods in Earth's History	There have been many glaciations in Earth's distant past, the oldest known starting around 2,400 Ma. The late Proterozoic "Snowball Earth" glaciations were thought to be sufficiently intense to affect the entire planet. The current glacial period is known as the Pleistocene Glaciation, and while it was much more intense 20,000 years ago than it is now, we are still in the middle of it. The periodicity of the Pleistocene glaciations is related to subtle changes in Earth's orbital characteristics, which are exaggerated by a variety of positive feedback processes.
16.2 How Glaciers Work	The two main types of glaciers are continental glaciers, which cover large parts of continents, and alpine glaciers, which occupy mountainous regions. Ice accumulates at higher elevations—above the equilibrium line—where the snow that falls in winter does not all melt in summer. In continental glaciers, ice flows outward from where it is thickest. In alpine glaciers, ice flows downslope. At depth in the glacier ice, flow is by internal deformation, but glaciers that have liquid water at their base can also flow by basal sliding. Crevasses form in the rigid surface ice in places where the lower plastic ice is changing shape.
16.3 Glacial Erosion	Glaciers are important agents of erosion. Continental glaciers tend to erode the land surface into flat plains, while alpine glaciers create a wide variety of different forms. The key feature of alpine glacial erosion is the U-shaped valley. Arêtes are sharp ridges that form between two valleys, and horns form where a mountain is glacially eroded on at least three sides. Because tributary glaciers do not erode as deeply as main-valley glaciers, hanging valleys exist where the two meet. On a smaller scale, both types of glaciers form drumlins, roches moutonées, and glacial grooves or striae.
16.4 Glacial Deposition	Glacial deposits are quite varied, as materials are transported and deposited in a variety of different ways in a glacial environment. Sediments that are moved and deposited directly by ice are known as till. Glaciofluvial sediments are deposited by glacial streams, either forming eskers or large proglacial plains known as sandurs. Glaciolacustrine and glaciomarine sediments originate within glaciers and are deposited in lakes and the ocean respectively.

Questions for Review

See Appendix 2 for answers to Review Questions.

1. Why are the Cryogenian glaciations called Snowball Earth?
2. Earth cooled dramatically from the end of the Paleocene until the Holocene. Describe some of the geological events that contributed to that cooling.
3. When and where was the first glaciation of the Cenozoic?
4. Describe the extent of the Laurentide Ice Sheet during the height of the last Pleistocene glacial period.
5. In an alpine glacier, the ice flows down the slope of the underlying valley. Continental glaciers do not have a sloped surface to flow down. What feature of a continental glacier facilitates its flow?
6. What does the equilibrium line represent in a glacier? Explain.
7. Which of the following is more important to the growth of a glacier: very cold winters or relatively cool summers? Explain.
8. Describe the relative rates of ice flow within the following parts of a glacier: (a) the bottom versus the top and (b) the edges versus the middle. Explain.
9. What condition is necessary for basal sliding to take place?
10. Why do glaciers carve U-shaped valleys, and how does a hanging valley form?
11. A horn is typically surrounded by cirques. What is the minimum number of cirques you would expect to find around a horn?
12. A drumlin and a roche moutonée are both streamlined glacial erosion features. How do they differ in shape?
13. Four examples of glacial sediments are shown here. Describe the important characteristics (e.g., sorting, layering, grain-size range, grain shape, sedimentary structures) of each one and give each a name (choose from glaciofluvial, glaciolacustrine, lodgement till, ablation till, and glaciomarine).



Figure A

- What are drop stones, and under what circumstances are they likely to form?
- What types of glacial sediments are likely to be sufficiently permeable to make good aquifers?

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17: SHORELINES

17.1: Waves
17.2: Landforms and Coastal Erosion
17.3: Landforms and Coastal Deposition
17.4: Sea-Level Change
17.5: Human Interference with Shorelines
17.6: Summary

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17.1: Waves

Waves form on the ocean and on lakes because energy from the wind is transferred to the water. The stronger the wind, the longer it blows, and the larger the area of water over which it blows (the **fetch**), the larger the waves are likely to be.

The important parameters of a wave are the **wavelength** (the horizontal distance between two crests or two troughs), the **amplitude** (the vertical distance between a **trough** and a **crest**), and the wave velocity (the speed at which wave crests move across the water) (Figure 17.1.1).

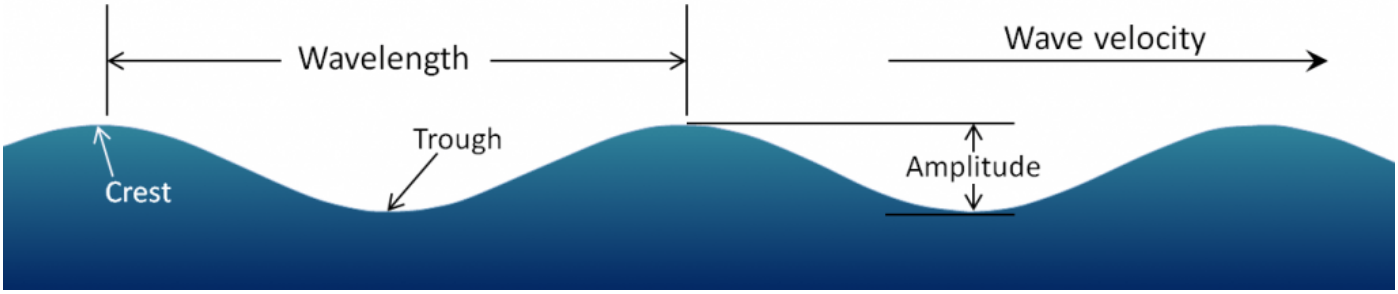


Figure 17.1.1 The parameters of water waves.

The typical sizes and speeds of waves in situations where they have had long enough to develop fully are summarized in Table 17.1. In a situation where the fetch is short (say 19 km on a lake) and the wind is only moderate (19 km/h), the waves will develop fully within 2 hours, but they will remain quite small (average amplitude about 27 cm, wavelength 8.5 m). On a large body of water (the ocean or a very large lake) with a fetch of 139 km and winds of 37 km/h, the waves will develop fully in 10 hours; the average amplitude will be around 1.5 m and average wavelength around 34 m. In the open ocean, with strong winds (92 km/h) that blow for at least 69 hours, the waves will average nearly 15 m high and their wavelengths will be over 200 m. Small waves (amplitudes under a metre) tend to have relatively shallow slopes (amplitude is 3% to 4% of wavelength), while larger waves (amplitudes over 10 m) have much steeper slopes (amplitude is 6% to 7% of wavelength). In other words, not only are large waves bigger than small ones, they are also generally more than twice as steep, and therefore many times more impressive—and potentially dangerous. It is important to recognize, however, that amplitudes decrease with distance from the area where the waves were generated. Waves on our coast that are generated by a storm near Japan will have similar wavelengths but lower amplitudes than those generated by a comparable storm just offshore.

Table 17.1 The parameters of wind waves in situations where the wind blows in roughly the same direction for long enough for the waves to develop fully. The duration times listed are the minimum required for the waves to develop fully.[1]

Wind Speed (kilometers per hour)	Fetch (kilometers)	Duration (hours)	Amplitude (meters)	Wavelength (meters)	Wave Period (seconds)	Wave Velocity (meters per second)	Wave Velocity (kilometers per hour)
19	19	2	0.27	8.5	3.0	2.8	10.2
37	139	10	1.5	33.8	5.7	5.9	19.5
56	518	23	4.1	76.5	8.6	8.9	32.0
74	1,313	42	8.5	136	11.4	11.9	42.9
92	2,627	69	14.8	212	14.3	14.8	53.4

Exercise 17.1 Wave height versus length

This table shows the typical amplitudes and wavelengths of waves generated under different wind conditions. The steepness of a wave can be determined from these numbers and is related to the ratio: amplitude/wavelength.

1. Calculate these ratios for the waves shown. The first one is done for you.

2. How would these ratios change with increasing distance from the wind that produced the waves?

Calculate the wave ratios.

Amplitude (meters)	Wavelength (meters)	Ratio (amplitude divided by wavelength)
0.27	8.5	0.03
1.5	33.8	
4.1	76.5	
8.5	136	
14.8	212	

See Appendix 3 for Exercise 17.1 answers.

Relatively small waves move at up to about 10 km/h and arrive on a shore about once every 3 seconds. Very large waves move about five times faster (over 50 km/h), but because their wavelengths are so much longer, they arrive less frequently—about once every 14 seconds.

As a wave moves across the surface of the water, the water itself mostly just moves up and down and only moves a small amount in the direction of wave motion. As this happens, a point on the water surface describes a circle with a diameter that is equal to the wave amplitude (Figure 17.1.2). This motion is also transmitted to the water underneath, and the water is disturbed by a wave to a depth of approximately one-half of the wavelength. Wave motion is illustrated quite clearly on the Wikipedia "Wind wave" site. If you look carefully at that animation, and focus on the small white dots in the water, you should be able to see how the amount that they move decreases with depth.

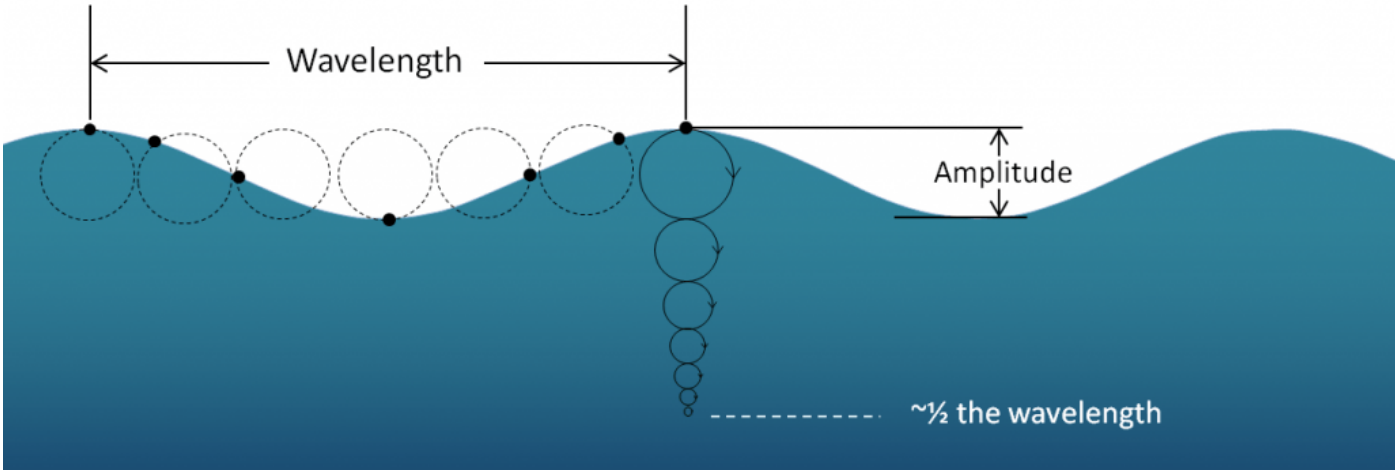


Figure 17.1.2 The orbital motion of a parcel of water (black dot) as a wave moves across the surface.

The one-half wavelength depth of disturbance of the water beneath a wave is known as the **wave base**. Since ocean waves rarely have wavelengths greater than 200 m, and the open ocean is several thousand meters deep, the wave base does not normally interact with the bottom of the ocean. However, as waves approach the much shallower water near the shore, they start to "feel" the bottom, and they are affected by that interaction (Figure 17.1.3). The wave "orbits" are both flattened and slowed by dragging, and the implications are that the wave amplitude (height) increases and the wavelength decreases (the waves become much steeper). The ultimate result of this is that the waves lean forward, and eventually break (Figure 17.1.4).

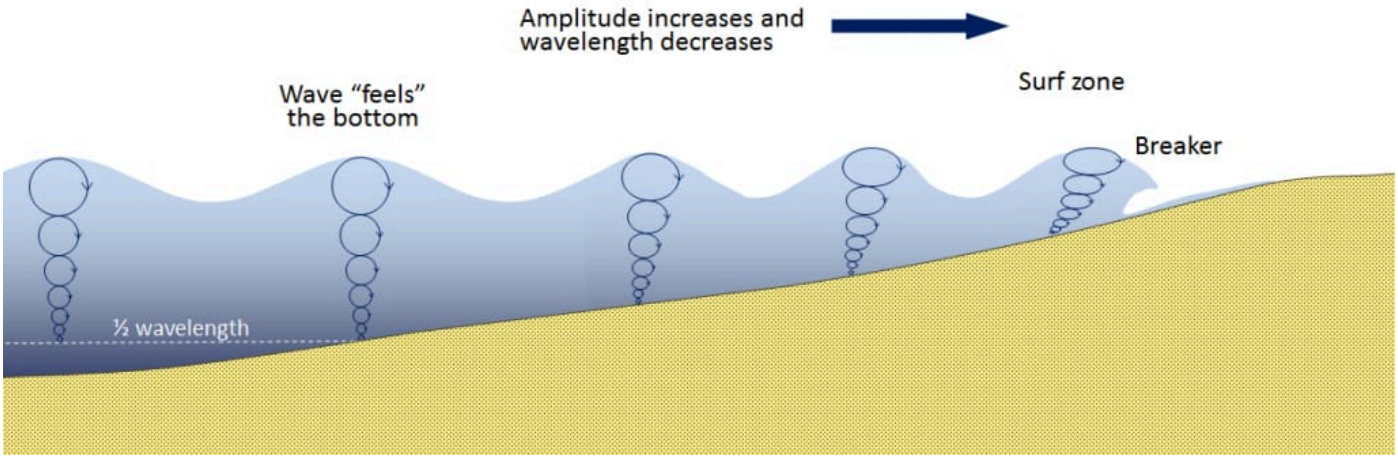


Figure 17.1.3 The effect of waves approaching a sandy shore.



Figure 17.1.4 Waves breaking on the shore at Greensand Beach, Hawaii (the sand is green because it is made up mostly of the mineral olivine eroded from the nearby volcanic rocks).

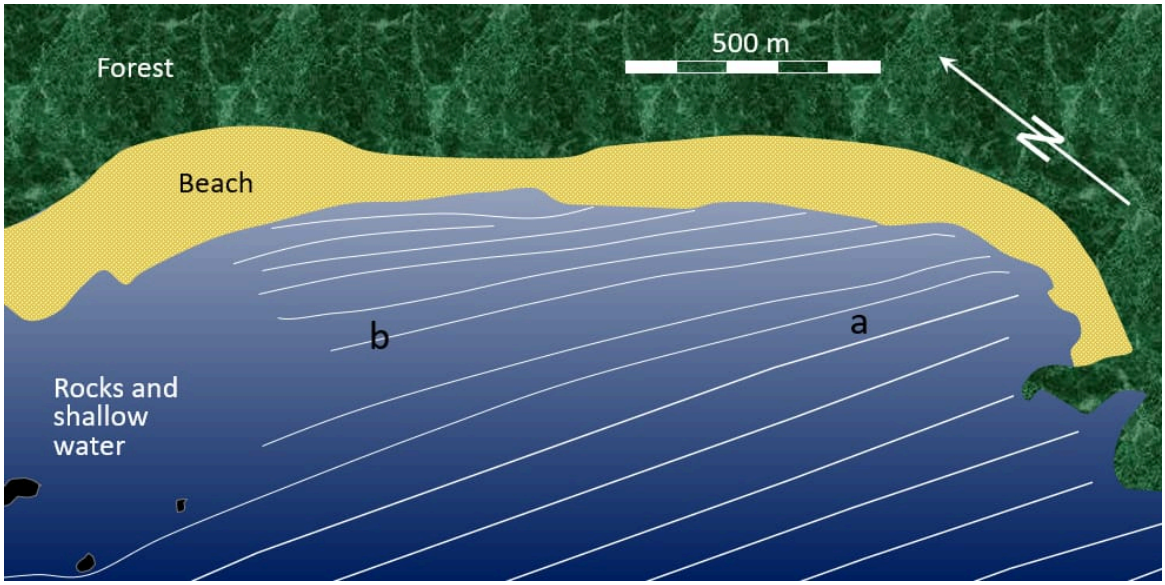


Figure 17.1.5 Waves approaching the shore of Long Beach in Pacific Rim National Park. As the waves (depicted by white lines) approach shore, they are refracted to become more parallel to the beach, and their wavelength decreases.

Waves normally approach the shore at an angle, and this means that one part of the wave feels the bottom sooner than the rest of it, so the part that feels the bottom first slows down first. This process is illustrated in Figure 17.1.5, which is based on an aerial photograph showing actual waves approaching Long Beach on Vancouver Island. When the photo was taken, the waves (with crests shown as white lines in the diagram) were approaching at an angle of about 20° to the beach. The waves first reached shore at the southern end ("a" on the image). As they moved into shallow water they were slowed, and since the parts of the waves still in deep water ("b" on the image) were not slowed they were able catch up, and thus the waves became more parallel to the beach.

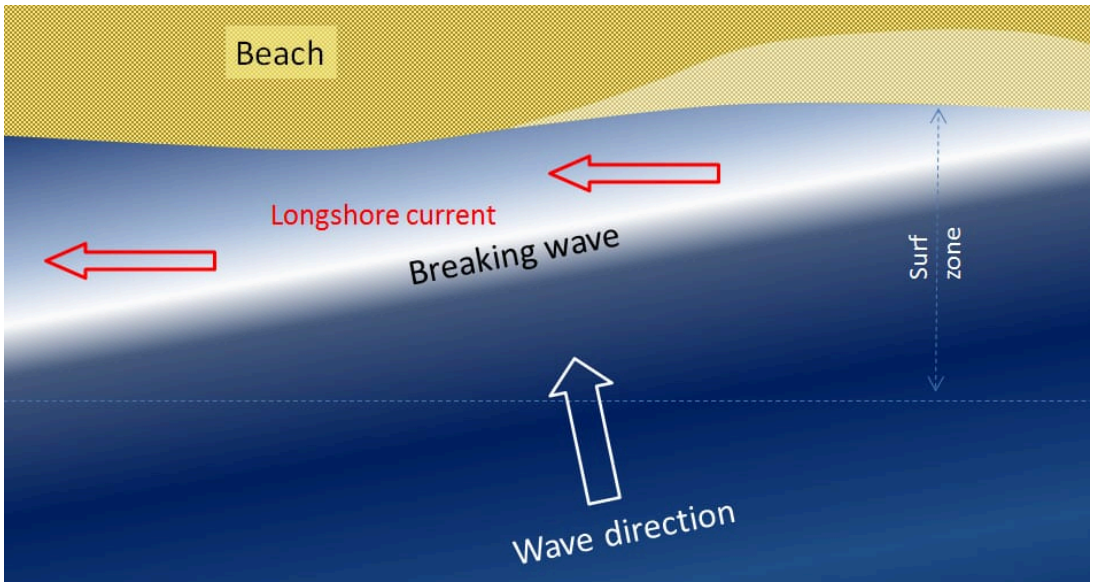


Figure 17.1.6 The generation of a longshore current by waves approaching the shore at an angle.

In open water, these waves had wavelengths close to 100 m. In the shallow water closer to shore, the wavelengths decreased to around 50 m, and in some cases, even less.

Even though they bend and become nearly parallel to shore, most waves still reach the shore at a small angle, and as each one arrives, it pushes water along the shore, creating what is known as a **longshore current** within the **surf zone** (the areas where waves are breaking) (Figure 17.1.6).

Exercise 17.2 Wave refraction

A series of waves (dashed lines) is approaching the coast on the map shown here. The location of the depth contour that is equivalent to 1/2 of the wavelength is shown as a red dashed line.

a. Draw in the next several waves, showing how their patterns will change as they approach shallow water and the shore.
b. Show, with arrows, the direction of the resulting longshore current.

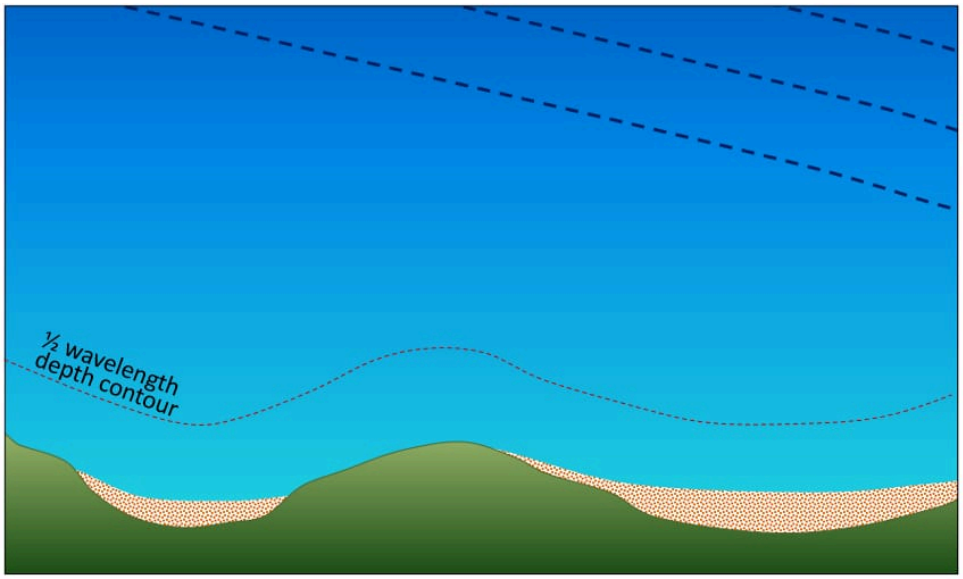


Figure 17.1.7

See Appendix 3 for Exercise 17.2 answers.

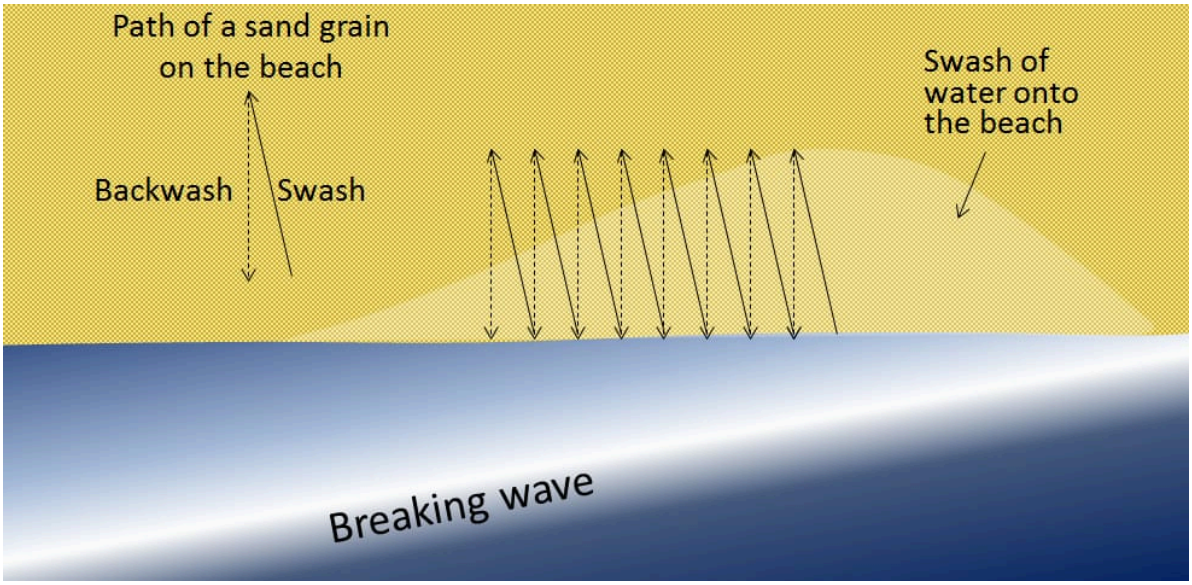


Figure 17.1.8 The movement of particles on a beach as a result of swash and backwash.

Another important effect of waves reaching the shore at an angle is that when they wash up onto the beach, they do so at an angle, but when that same wave water flows back down the beach, it moves straight down the slope of the beach (Figure 17.1.8). The upward-moving water, known as the **swash**, pushes sediment particles along the beach, while the downward-moving water, the **backwash**, brings them straight back. With every wave that washes up and then down the beach, particles of sediment are moved along the beach in a zigzag pattern.

The combined effects of sediment transport within the surf zone by the longshore current and sediment movement along the beach by swash and backwash is known as **longshore drift**. Longshore drift moves a tremendous amount of sediment along coasts (both oceans and large lakes) around the world, and it is responsible for creating a variety of depositional features that we'll discuss in section 17.3.

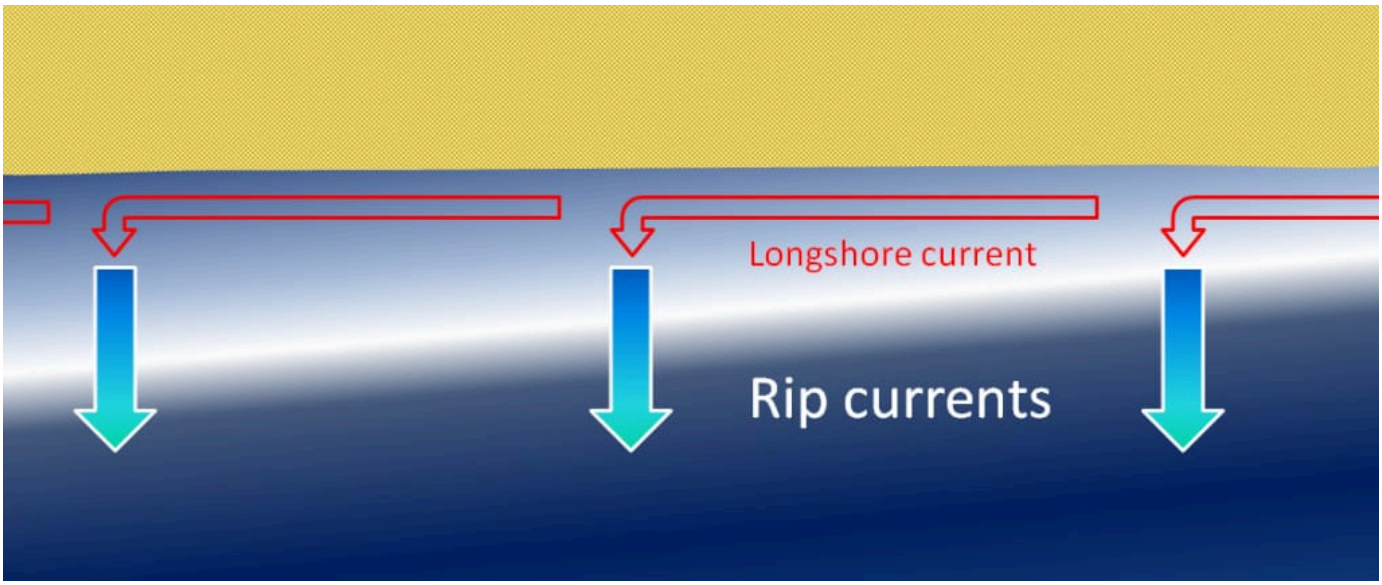


Figure 17.1.9 The formation of rip currents on a beach with strong surf.

A **rip current** is another type of current that develops in the nearshore area, and has the effect of returning water that has been pushed up to the shore by incoming waves. As shown in Figure 17.1.9, rip currents flow straight out from the shore and are fed by the longshore currents. They die out quickly just outside the surf zone, but can be dangerous to swimmers who get caught in them. If part of a beach does not have a strong unidirectional longshore current, the rip currents may be fed by longshore currents going in both directions.

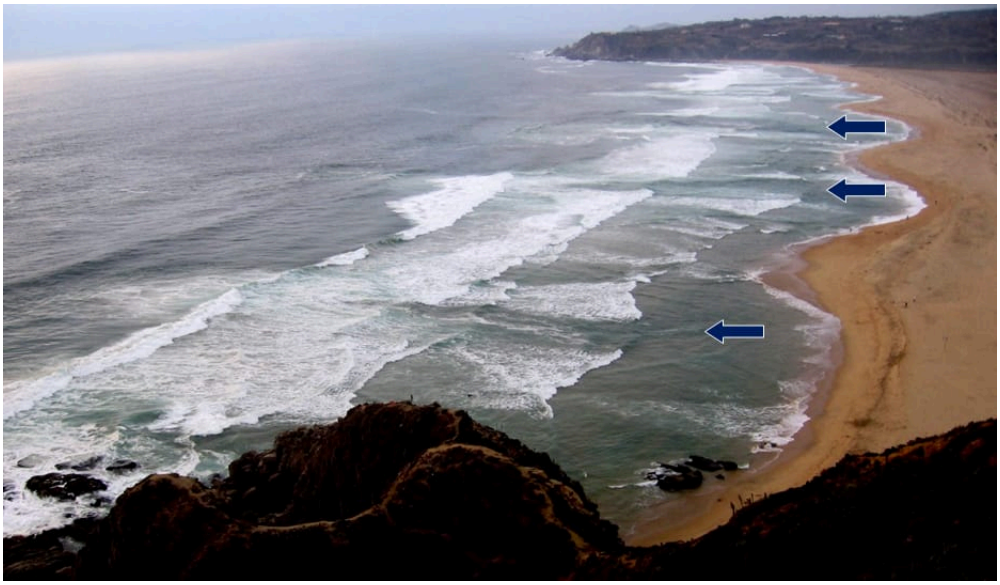


Figure 17.1.10 Rip currents on Tunquen Beach in central Chile.

Rip currents are visible in Figure 17.1.10, a beach at Tunquen in Chile near Valparaiso. As is evident from the photo, the rips correspond with embayments in the beach profile. Three of them are indicated with arrows, but it appears that there may be several others farther along the beach.

Tides are related to very long-wavelength but low-amplitude waves on the ocean surface (and to a much lesser extent on very large lakes) that are caused by variations in the gravitational effects of the Sun and Moon. Tide amplitudes in shoreline areas vary quite dramatically from place to place. On the west coast of Canada, the tidal range is relatively high, in some areas as much as 6 m, while on most of the east coast the range is lower, typically around 2 m. A major exception is the Bay of Fundy between Nova Scotia and New Brunswick, where the daily range can be as great as 16 m. Anomalous tides like that are related to the shape and size of bays and inlets, which can significantly enhance the amplitude of the tidal surge. The Bay of Fundy has a natural oscillation cycle of 12.5 hours, and that matches the frequency of the rise and fall of the tides in the adjacent Atlantic Ocean. Ungava Bay, on

Quebec's north coast, has a similarly high tidal range.

As the tides rise and fall they push and pull a large volume of water in and out of bays and inlets and around islands. They do not have as significant an impact on coastal erosion and deposition as wind waves do, but they have an important influence on the formation of features within the intertidal zone, as we'll see in the following sections.

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- Figure 17.1.10: "Tunquen Chile" by Cecilia and Randy Lascody. Public domain.

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1. Table by Steven Earle from data at: en.Wikipedia.org/wiki/Wind_wave] ↩

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17.2: Landforms and Coastal Erosion

Large waves crashing onto a shore bring a tremendous amount of energy that has a significant eroding effect. Several unique erosion features commonly form on rocky shores with strong waves.

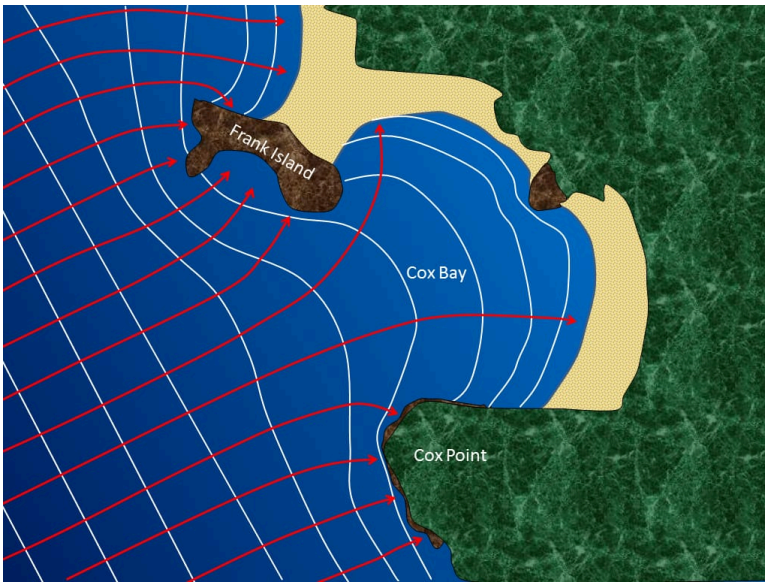


Figure 17.2.1 The approach of waves (white lines) in the Cox Bay area of Long Beach, Vancouver Island. The red arrows represent wave energy; most of that energy is focused on the headlands of Frank Island and Cox Point.

When waves approach an irregular shore, they are slowed down to varying degrees, depending on differences in the water depth, and as they slow, they are bent or refracted. In Figure 17.2.1, wave energy is represented by the red arrows. That energy is evenly spaced out in the deep water, but because of refraction, the energy of the waves—which moves perpendicular to the wave crests—is being focused on the **headlands** (Frank Island and Cox Point in this case). On irregular coasts, the headlands receive much more wave energy than the intervening bays, and thus they are more strongly eroded. The result of this is **coastal straightening**. An irregular coast, like the west coast of Vancouver Island, will eventually become straightened, although that process will take millions of years.

Wave erosion is greatest in the surf zone, where the wave base is impinging strongly on the sea floor and where the waves are breaking. The result is that the substrate in the surf zone is typically eroded to a flat surface known as a **wave-cut platform** (or wave-cut terrace) (Figure 17.2.2). A wave-cut platform typically extends across the intertidal zone.



Figure 17.2.2 A wave-cut platform in bedded sedimentary rock on Gabriola Island, B.C. The wave-eroded surface is submerged at high tide.



Figure 17.2.3 A stack on the Juan de Fuca Trail section of the southwestern shore of Vancouver Island. The rock surrounding the stack is part of a wave-cut platform.

Relatively resistant rock that does not get completely eroded during the formation of a wave-cut platform will remain behind to form a **stack**. An example from the Juan de Fuca Trail of southwestern Vancouver Island is shown in Figure 17.2.3. Here the different layers of the sedimentary rock have different resistance to erosion. The upper part of this stack is made up of rock that resisted erosion, and that rock has protected a small pedestal of underlying softer rock. The softer rock will eventually be eroded and the big rock will become just another boulder on the beach. Note that this is a somewhat unique situation. Most stacks do not show that nature of differential erosion.



Figure 17.2.4 Top: An arch in tilted sedimentary rock at the mouth of the Barachois River, Newfoundland, July 2012. Bottom: The same location in June 2013. The arch has collapsed and a small stack remains.

Arches and **sea caves** are related to stacks because they all form as a result of the erosion of relatively non-resistant rock. An arch in the Barachois River area of western Newfoundland is shown in Figure 17.2.4. This feature started out as a sea cave, and then, after being eroded from both sides, became an arch. During the winter of 2012/2013, the arch collapsed, leaving a small stack at the end of the point. If you look carefully at the upper photograph you can see that the hole that makes the arch developed within a layer of relatively soft and weak rock.

Figure 17.2.5 summarizes the process of transformation of an irregular coast, initially produced by tectonic uplift, into a straightened coast with **sea cliffs** (wave-eroded escarpments) and the remnants of stacks, arches, and wave-cut platforms. The next stages of this process would be the continued landward erosion of the sea cliffs and the complete erosion of the stacks and wave-cut platforms in favour of a continuous nearly straight coast. That straight coast might be a sea cliff, or—if there is a sufficient and ongoing source of sand—a beach

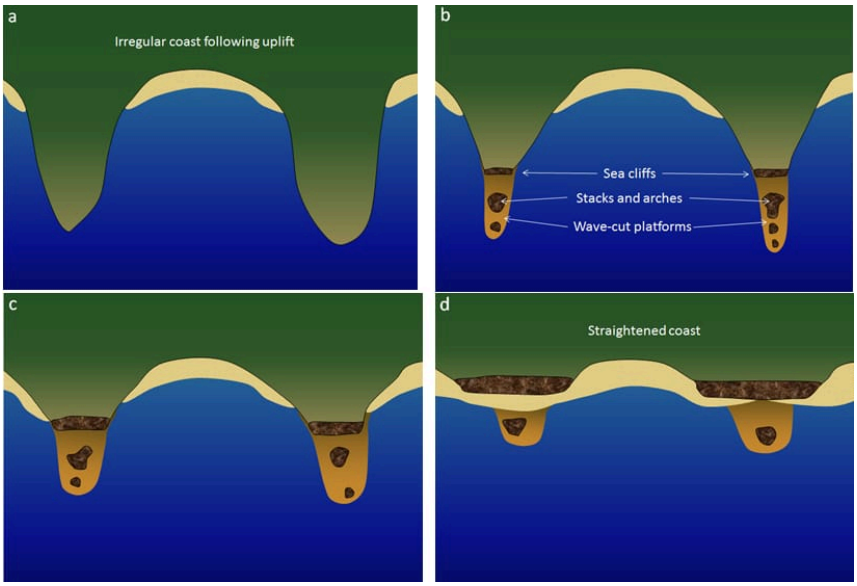


Figure 17.2.5 Evolution of a straightened coast through the erosion to stacks and arches, sea cliffs, and wave-cut platforms.

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- Figure 17.2.4: Image by Dr. David Murphy. Used with permission.

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17.3: Landforms and Coastal Deposition

Some coastal areas are dominated by erosion, an example being the Pacific coast of Canada and the United States, while others are dominated by deposition, examples being the Atlantic and Caribbean coasts of the United States. But on almost all coasts, both deposition and erosion are happening to varying degrees most of the time, although in different places. This is clearly evident in the Tofino area of Vancouver Island (Figure 17.3.1), where erosion is the predominant process on the rocky headlands, while depositional processes predominate within the bays. On deposition-dominant coasts, the coastal sediments are still being eroded from some areas and deposited in others.

A key factor in determining if a coast is dominated by erosion or deposition is its history of tectonic activity. A coast like that of British Columbia is tectonically active, and compression and uplift have been going on for tens of millions of years. This coast has also been uplifted during the past 15,000 years by isostatic rebound due to deglaciation. The coasts of the United States along the Atlantic and the Gulf of Mexico have not seen significant tectonic activity in a few hundred million years, and except in the northeast, have not experienced post-glacial uplift. These areas have relatively little topographic relief, and there is now minimal erosion of coastal bedrock. Another important factor is the supply of sediments. Unless there is a continuous supply of sandy and coarser sediment to a coast it will not be a depositional coast.

On coasts that are dominated by depositional processes, most of the sediment being deposited typically comes from large rivers. An obvious example is where the Mississippi River flows into the Gulf of Mexico at New Orleans; another is the Fraser River at Vancouver. There are no large rivers bringing sandy sediments to the west coast of Vancouver Island, but there are still long and wide sandy beaches there. In this area, most of the sand comes from glaciofluvial sand deposits situated along the shore behind the beach, and some comes from the erosion of the rocks on the headlands.

The components of a typical beach are shown in Figure 17.3.1. On a sandy marine beach, the **beach face** is the area between the low and high tide levels. A **berm** is a flatter region beyond the reach of high tides; this area stays dry except during large storms.

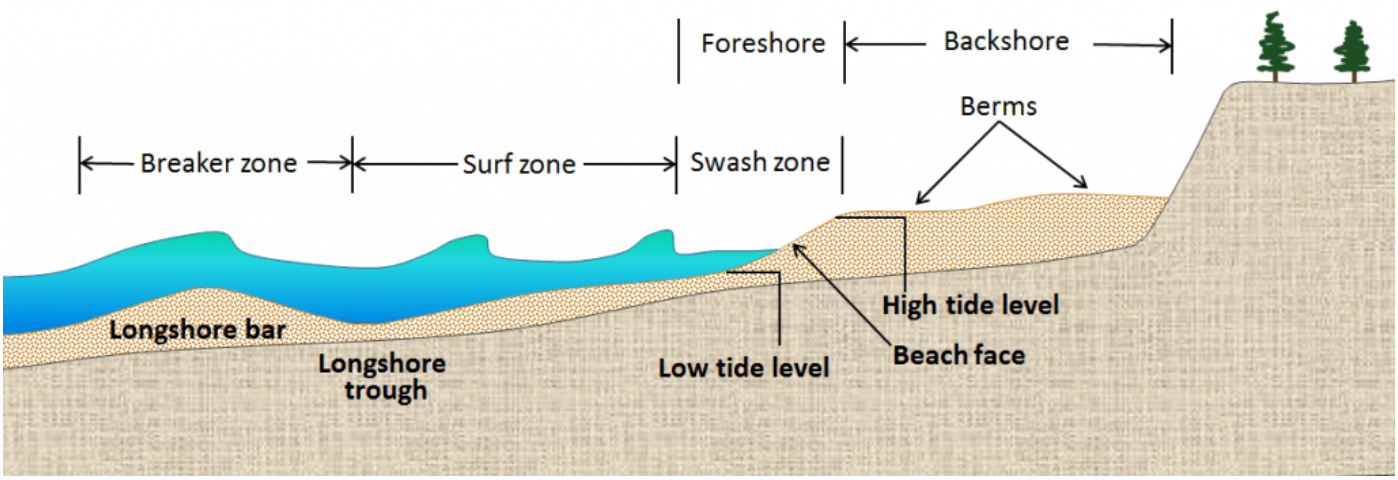


Figure 17.3.1 The components of a sandy marine beach.

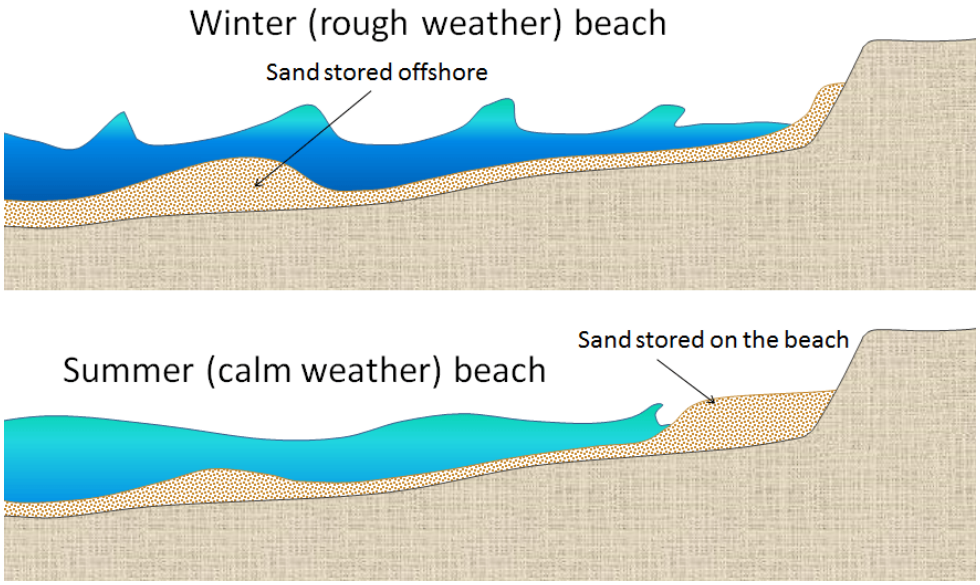


Figure 17.3.2 The differences between summer and winter on beaches in areas where the winter conditions are rougher and waves have a shorter wavelength but higher energy. In winter, sand from the beach is stored offshore.

Most beaches go through a seasonal cycle because conditions change from summer to winter. In summer, sea conditions are relatively calm with long-wavelength, low-amplitude waves generated by distant winds. Winter conditions are rougher, with shorter-wavelength, higher-amplitude waves caused by strong local winds. As shown in Figure 17.3.2, the heavy seas of winter gradually erode sand from beaches, moving it to an underwater sandbar offshore from the beach. The gentler waves of summer gradually push this sand back toward the shore, creating a wider and flatter beach.

The evolution of sandy depositional features on sea coasts is primarily influenced by waves and currents, especially longshore currents. As sediment is transported along a shore, either it is deposited on beaches, or it creates other depositional features. A **spit**, for example is an elongated sandy deposit that extends out into open water in the direction of a longshore current. A good example is Goose Spit at Comox on Vancouver Island (Figure 17.3.3). At this location, the longshore current typically flows toward the southwest, and the sand eroded from a 60 m high cliff of Pleistocene glaciofluvial Quadra Sand is pushed in that direction and then out into Comox Harbour.

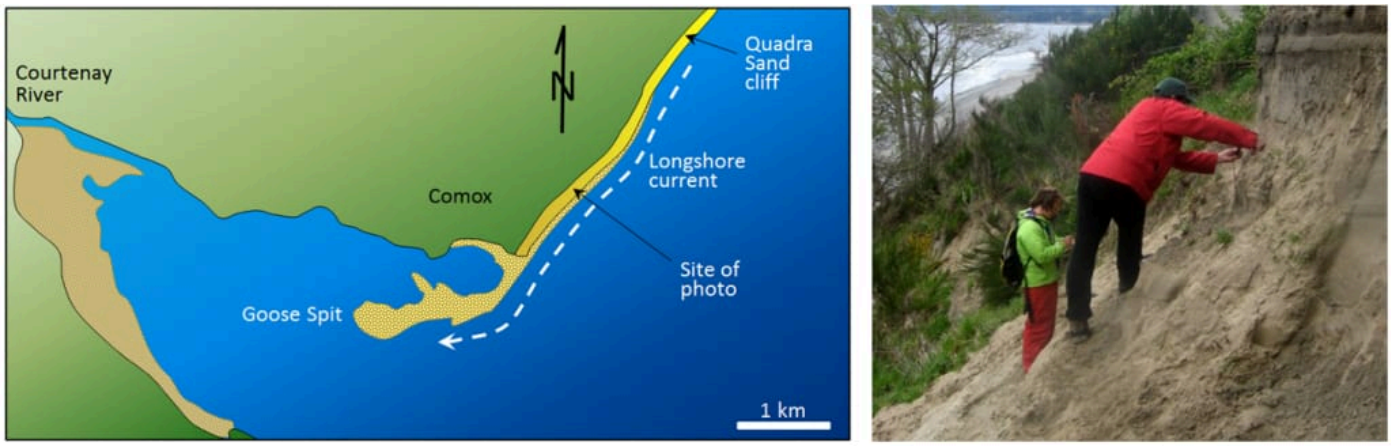


Figure 17.3.3 The formation of Goose Spit at Comox on Vancouver Island. The sand that makes up Goose Spit is derived from the erosion of Pleistocene Quadra Sand (a thick glaciofluvial sand deposit, as illustrated in the photo on the right).

The Quadra Sand at Comox is visible in Figure 17.3.4. There are numerous homes built at the top of the cliff, and the property owners have gone to considerable expense to reinforce the base of the cliff with large angular rocks (**rip-rap**) and concrete barriers so as to limit further erosion of their properties. One result of this will be to starve Goose Spit of sediments and eventually contribute to its erosion. Of course the rocks and concrete barriers are only temporary; they will be eroded by strong winter storms over the next few decades and the Quadra Sand will once again contribute to the maintenance of Goose Spit.



Figure 17.3.4 The Quadra Sand cliff at Comox, and the extensive concrete and rip-rap barrier that has been constructed to reduce erosion. Note that the waves (dashed lines) are approaching the shore at an angle, contributing to the longshore drift.

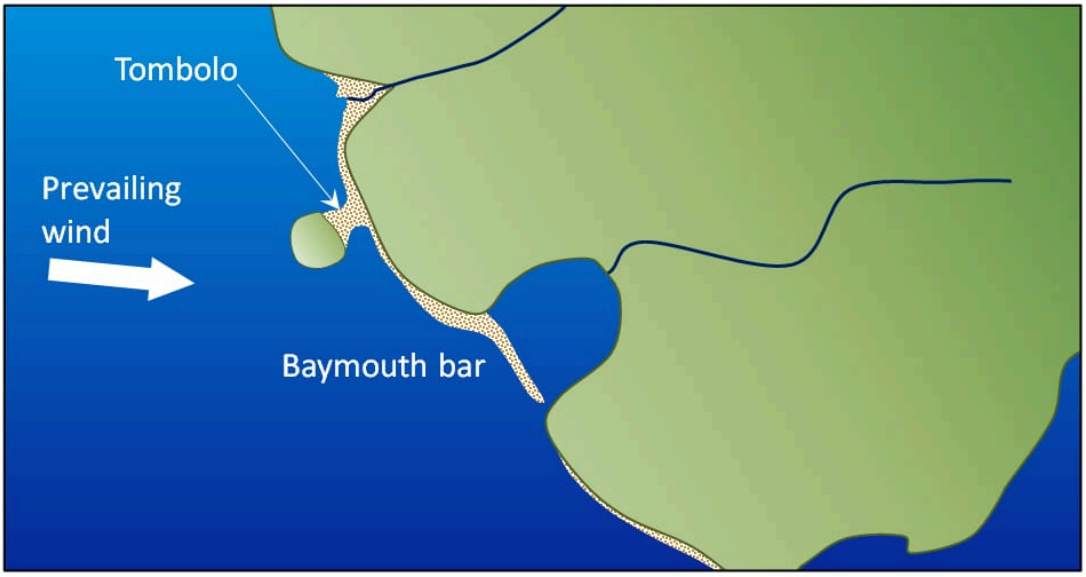


Figure 17.3.5 A depiction of a baymouth bar and a tombolo.

A spit that extends across a bay to the extent of closing, or almost closing it off, is known as a **baymouth bar**. Most bays have streams flowing into them, and since this water has to get out, it is rare that a baymouth bar will completely close the entrance to a bay. In areas where there is sufficient sediment being transported, and there are near-shore islands, a **tombolo** may form (Figure 17.3.5).

Tombolos are common around the southern part of the coast of British Columbia, where islands are abundant, and they typically form where there is a wave shadow behind a nearshore island (Figure 17.3.6). This becomes an area with reduced energy, and so the longshore current slows and sediments accumulate. Eventually enough sediments accumulate to connect the island to the mainland with a tombolo. There is a good example of a tombolo in Figure 17.3.1, and another in Figure 17.3.7.

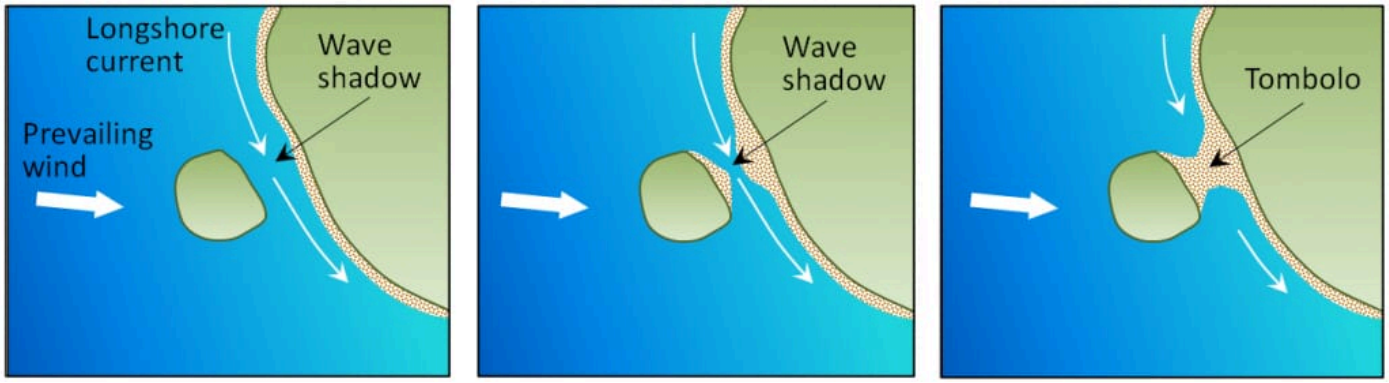


Figure 17.3.6 The process of formation of a tombolo in a wave shadow behind a nearshore island.



Figure 17.3.7 A stack (with a wave-cut platform) connected to the mainland by a tombolo, Gabriola Island, B.C.

In areas where coastal sediments are abundant and coastal relief is low (because there has been little or no recent coastal uplift), it is common for barrier islands to form. Barrier islands are elongated islands composed of sand that form a few kilometers away from the mainland. They are common along the U.S. Gulf Coast from Texas to Florida, and along the U.S. Atlantic Coast from Florida to Massachusetts (Figure 17.3.8). North of Boston, the coast becomes rocky, partly because that area has been affected by post-glacial crustal rebound.



Figure 17.3.8 Assateague Island on the Maryland coast, U.S. This barrier island is about 60 km long and only 1 km to 2 km wide. The open Atlantic Ocean is to the right and the lagoon is to the left. This part of Assateague Island has recently been eroded by a tropical storm, which pushed massive amounts of sand into the lagoon.

Exercise 17.3 Beach forms



Figure 17.3.9

On the map, sketch where you would expect the following to form:

- A spit
- A baymouth bar
- A tombolo

What conditions might lead to the formation of barrier islands in this area?

See Appendix 3 for Exercise 17.3 answers.

Some coasts in tropical regions (between 30° S and 30° N) are characterized by carbonate **reefs**. Reefs form in relatively shallow marine water within a few hundred to a few thousand meters of shore in areas where the water is clear because there is little or

no input of clastic sediments from streams, and marine organisms such as corals, algae, and shelled organisms can thrive. The associated biological processes are enhanced where upwelling currents bring chemical nutrients from deeper water (but not so deep that the water is cooler than about 25°C) (Figure 17.3.10). Sediments that form in the **back reef** (shore side) and **fore reef** (ocean side) are typically dominated by carbonate fragments eroded from the reef and from organisms that thrive in the back-reef area that is protected from wave energy by the reef.

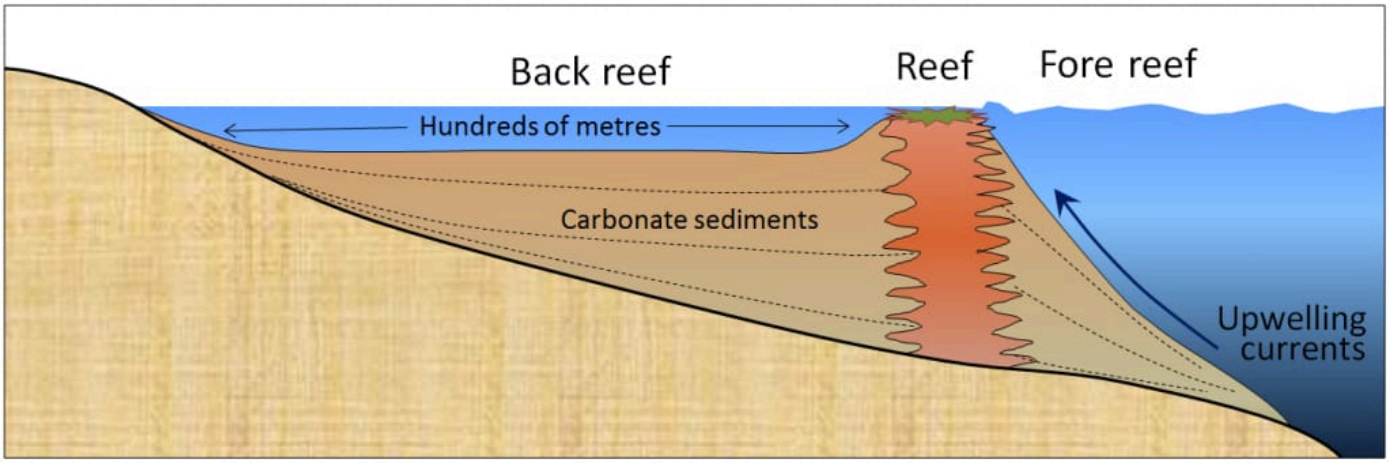


Figure 17.3.10 Cross-section through a typical barrier or fringing reef.

Image descriptions

Figure 17.3.1 image description: A berm, the part of a beach that is beyond the reach of high tide, is part of the backshore. The beach face, the part of the beach between low tide and high tide level, includes the swash zone and the foreshore. Beyond the swash zone is the surf zone and beyond that is the breaker zone. [Return to Figure 17.3.1]

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17.4: Sea-Level Change

Sea-level change has been a feature on Earth for as long as there have been oceans (billions of years), and it has important implications for coastal processes and both erosional and depositional features. There are three main mechanisms of sea-level change, as described below.

Eustatic sea-level changes are global sea-level changes related either to changes in the volume of glacial ice on land or to changes in the shape of the sea floor caused by plate tectonic processes. For example, changes in the rate of mid-ocean spreading will change the shape of the sea floor near the ridges, and this affects sea level.

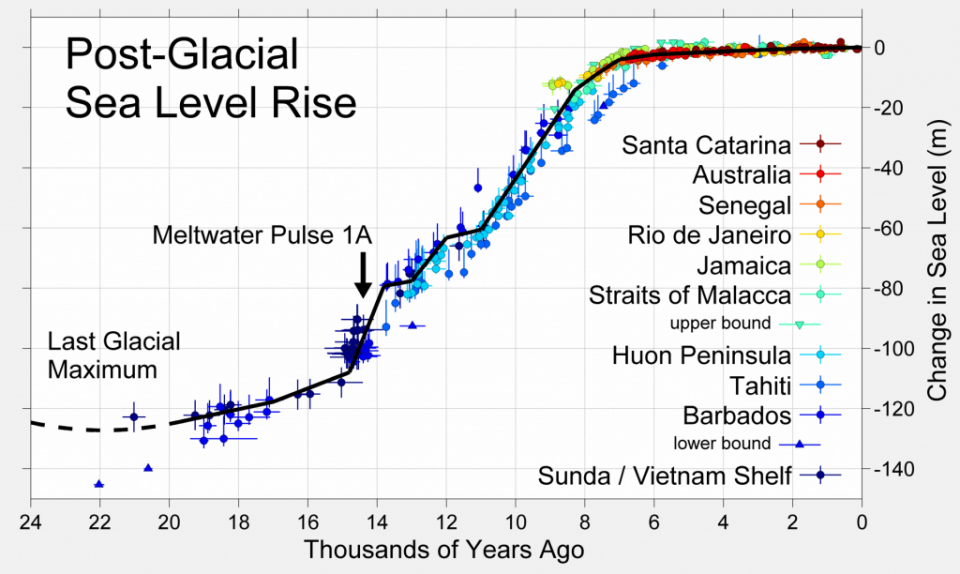


Figure 17.4.1 Eustatic sea-level curve for the past 24 ka (sea-level rise resulting from the melting of glacial ice). Sea-level rise is global; the locations listed in the caption are the places where data were acquired to create this diagram.

Over the past 20,000 years, there has been approximately 125 m of eustatic sea-level rise due to glacial melting. Most of that took place between 15,000 and 7,500 years ago during the major melting phase of the North American and Eurasian Ice Sheets (Figure 17.4.1). During that time the average rate of sea level rise was approximately 14 mm/y. At around 7,500 years ago, the rate of glacial melting and sea-level rise decreased dramatically. The average rate over the past 6000 years has been 0.5 mm/y. Anthropogenic climate change led to accelerating sea-level rise starting around 1870. Since then the average rate has been about 1.1 mm/y, but it has been gradually increasing. The current rate is over 3 mm/y.

Isostatic sea-level changes are local changes caused by subsidence or uplift of the crust related either to changes in the amount of ice on the land, or to growth or erosion of mountains.

Almost all of Canada and parts of the northern United States were covered in thick ice sheets at the peak of the last glaciation. Following the melting of this ice there has been an isostatic rebound of continental crust in many areas. This ranges from several hundred meters of rebound in the central part of the Laurentide Ice Sheet (around Hudson Bay) to 100 m to 200 m in the peripheral parts of the Laurentide and Cordilleran Ice Sheets—in places such as Vancouver Island and the mainland coast of B.C. In other words, although global sea level was about 130 m lower during the last glaciation, the glaciated regions were depressed at least that much in most places, and more than that in places where the ice was thickest.



Figure 17.4.2 This stream is on the southwest coast of Vancouver Island near Sooke. Like many other streams along this coast, it used to flow directly into the ocean, but the land has been uplifted by post-glacial isostatic rebound.

There is evidence of isostatic rebound along the southwest coast of Vancouver Island, where a number of streams enter the ocean as 5 m high waterfalls, as shown in Figure 17.4.2.

Tectonic sea-level changes are local changes caused by tectonic processes. The subduction of the Juan de Fuca Plate beneath British Columbia, Washington, Oregon and northern California is creating tectonic uplift (about 1 mm/year) along the western edge of the continent, although much of this uplift is likely to be reversed when the next large subduction-zone earthquake strikes.



Figure 17.4.3 Howe Sound, north of Vancouver, is a fjord filling a former glacial valley

Coastlines in areas where there has been net sea-level rise in the geologically recent past are commonly characterized by estuaries and fjords. Howe Sound, north of Vancouver, is an example of a fjord (Figure 17.4.3). This valley was filled with ice during the last glaciation, and there has been a net rise in sea level here since that time. Coastlines in areas where there has been net sea-level drop in the geologically recent past are characterized by uplifted wave-cut platforms (or stream valleys as shown in Figure 17.4.2). Uplifted beach lines are another product of relative sea-level drop, although these are difficult to recognize in areas with vigorous vegetation. They are relatively common in Canada's far north.

Exercise 17.4 A holocene uplifted shore

The blue-grey sediments in the photo contain marine fossils of early Holocene age (~12,500 years ago) situated at about 60 m above sea level on Gabriola Island, BC. Explain how eustatic and isostatic sea-level change processes might have contributed to the existence of these materials at this elevation.



Figure 17.4.4 Early Holocene marine sediments at 60 m elevation on Gabriola Island, B.C.

See Appendix 3 for Exercise 17.4 answers.

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17.5: Human Interference with Shorelines

There are various modifications that we make in an attempt to influence shoreline processes for our own purposes. Sometimes these changes are effective, and may appear to be beneficial, although in most cases there are unintended negative consequences that we don't recognize until much later.

An example is at the beach near Comox (described above), which has been armoured with rip-rap and concrete blocks in an attempt to limit the natural erosion that is threatening the properties at the top of the cliff (Figure 17.5.4). As already noted, the unintended effect of this installation will be to starve Goose Spit of sediment. As long as the armour remains in place, which might be several decades, there is a significant risk that the spit will start to erode, which will affect many of the organisms that use that area as their habitat, and many of the people who go there for recreation.



Figure 17.5.1 The seawall at Stanley Park, Vancouver.

Seawalls, like the one around Vancouver's Stanley Park (Figure 17.5.1), also help to limit erosion and can be very pleasant amenities for the public, but they have geological and ecological costs. When a shoreline is "hardened" in this way, important marine habitat is lost and sediment production is reduced, and that can affect beaches elsewhere. Seawalls also affect the behaviour of waves and longshore currents, sometimes with negative results.

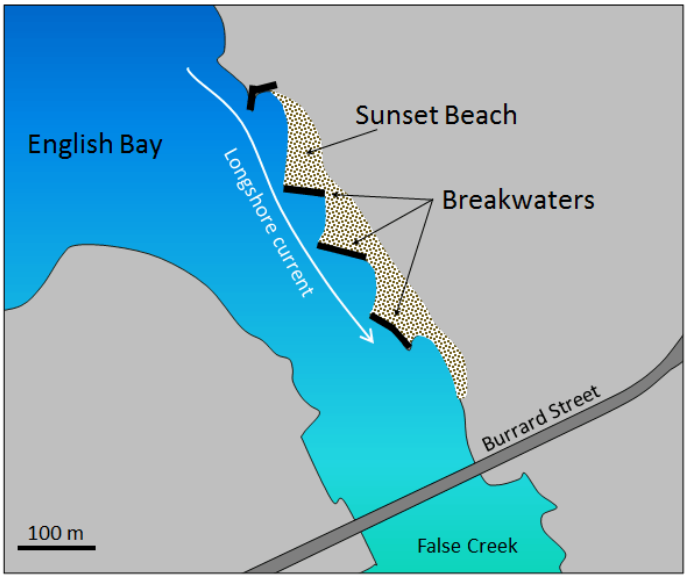


Figure 17.5.2 Map of the impact of breakwaters (or groynes) on beach formation at Sunset Beach, Vancouver.

Another example is at Sunset Beach in Vancouver. As shown in Figure 17.5.2, a series of rip-rap **breakwaters** (structures parallel to the shore) were built in the 1990s and sand has accumulated behind them to form the beach. The breakwaters have acted as islands and the sand has been deposited in the low-energy water behind them, in the same way that a tombolo forms. This can be seen from a photograph taken from the Burrard Bridge in 2015 (Figure 17.5.3). The two benefits of this project are that a pleasant beach has been created, and some of the sediment that previously would have been moved into False Creek, and could have blocked its entrance, has been trapped in English Bay. The negative impacts are probably not well understood, but have likely involved loss of marine animal habitat.



Figure 17.5.3 Photograph of the impact of breakwaters on beach development at Sunset Beach, Vancouver.

Groynes (or groins in the U.S.) have an effect that is similar to that of breakwaters, although groynes are constructed perpendicular to the beach (Figure 17.5.4), and they trap sediment by slowing the longshore current.



Figure 17.5.4 A groyne at Crescent Beach, Surrey, B.C.

Most of the sediment that forms beaches along our coasts comes from rivers, so if we want to take care of beaches, we have to take care of rivers. When a river is dammed, its sediment load is deposited in the resulting reservoir, and for the century or two while the reservoir is filling up, that sediment cannot get to the sea. During that time, beaches (including spits, baymouth bars, and tombolos) within tens of kilometers of the river's mouth (or more in some cases) are at risk of erosion.

Exercise 17.5 Crescent beach groynes

This diagram shows the same area illustrated in Figure 17.5.4 at Crescent Beach in Surrey, B.C. Based on information that you can find on the Internet about the function of groynes, determine which way the prevailing longshore current is moving at this location.

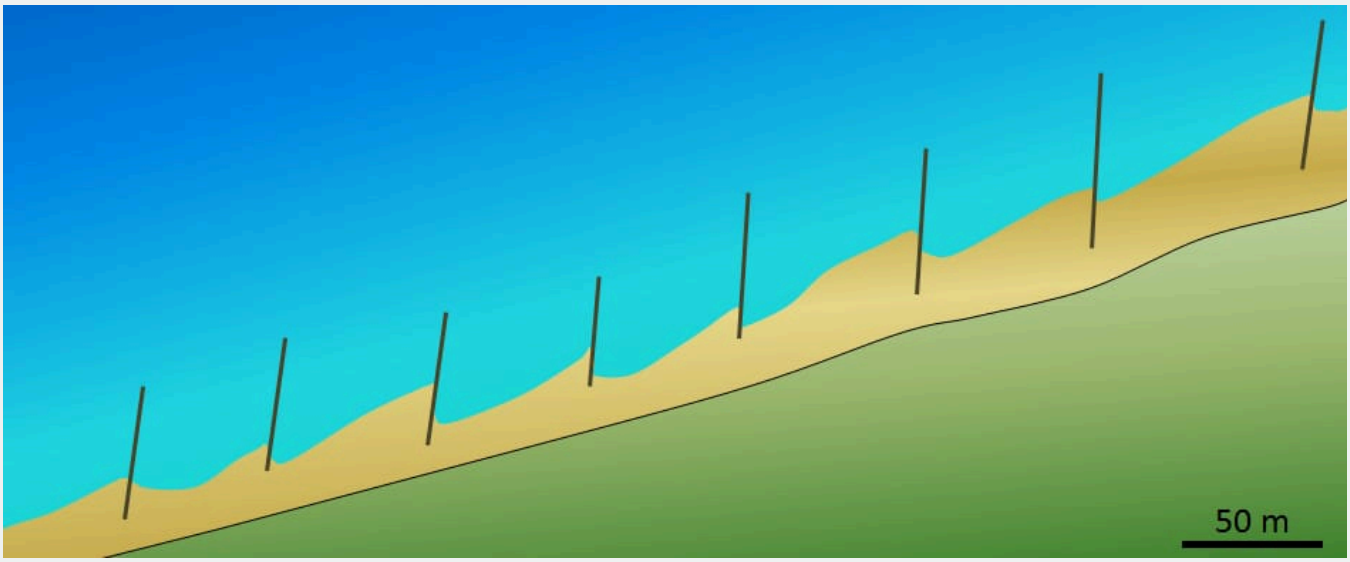


Figure 17.5.5 Beach and groyne locations at Crescent Beach, Surrey.

See Appendix 3 for Exercise 17.5 answers.

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17.6: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
17.1 Waves	Waves form when wind blows over water. The size of the waves depends on the wind speed, the area over which it is blowing, and time. The important parameters of a wave are its amplitude, wavelength, and speed. The water beneath a wave is disturbed to a depth of one-half the wavelength, and a wave is slowed when it approaches shallow water. A longshore current develops where waves approach the shore at an angle, and swash and backwash on a beach move sediment along the shore. The combined effect of these two processes is sediment transport by longshore drift.
17.2 Landforms and Coastal Erosion	Coasts that have experienced uplift within the past few million years tend to have irregular shapes and are dominated by erosional processes. Wave paths are bent where the coast is irregular and wave energy is focused on headlands. Rocky headlands are eroded into sea caves, arches, stacks, and sea cliffs, and the areas around these features are eroded into wave-cut platforms. Over the long term (millions of years), irregular coasts are straightened.
17.3 Landforms and Coastal Deposition	Coasts that have not been uplifted for tens of millions of years tend to be relatively straight, and are dominated by depositional features, although deposition is also important on irregular coasts. Waves and longshore drift are important in controlling the formation of beaches, as well as spits, tombolos, baymouth bars, and barrier islands. Beaches can be divided into zones, such as foreshore and backshore, and beach shapes typically change from season to season. Carbonate reefs and carbonate sediments form in tropical regions where there is little input of clastic sediments.
17.4 Sea-Level Change	The relative levels of the land and sea have significant implications for coastal processes and landforms, and they have been constantly changing over geological time. Eustatic sea-level changes are global in effect, and are typically related to glacial ice formation or melting. Isostatic sea-level changes are local effects caused by uplift or subsidence of continental crust, typically because of the gain or loss of glacial ice. Tectonic sea-level changes are related to plate interactions. Net sea-level rise leads to development of estuaries and fjords, while net sea-level drop creates uplifted marine terraces and beaches.
17.5 Human Interference with Shorelines	Humans have a strong urge to alter coasts for their own convenience by building seawalls, breakwaters, groynes, and other barriers. Although these types of features may have economic and other benefits, they can have both geological and ecological implications that must be considered.

Questions for Review

See Appendix 2 for answers to Review questions.

1. What factors control the size of waves?
2. Referring to Table 17.1, approximately what size of waves (amplitude and wavelength) would you expect with a 65 km/h wind blowing for 40 hours over 1,000 km of sea?
3. If the average wavelength of a series of waves is 100 m, at what depth of water will the waves start to feel the bottom, and how will that change their behaviour?
4. What is the difference between a longshore current and longshore drift?
5. On Figure A the waves (dashed blue lines) are approaching an irregular coast. The red arrows represent the energy of those waves, and one has been extended to show where that energy would hit the shore. Extend the other “energy lines” in a similar way, and comment on how this relates to erosion of this coastline.

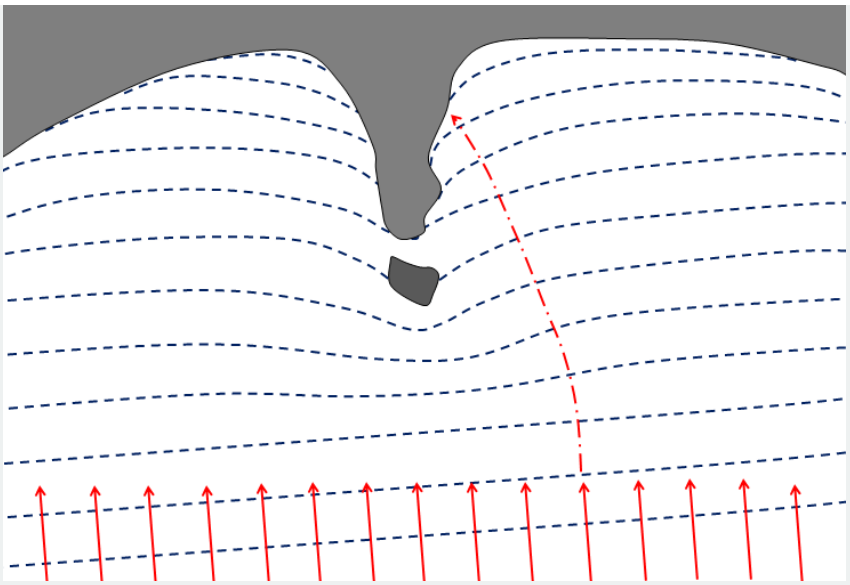


Figure A

- Explain the origins of a wave-cut platform.
- How do we define the limits of the beach face, and what are some other terms used to describe this zone?
- A spit is really just a beach that is only attached to the shore at one end. What conditions are necessary for the formation of a spit?
- Barrier islands are common along the Atlantic coast of the U.S. as far north as Massachusetts. Why are there almost none in the northeastern U.S. or along the coasts of New Brunswick, Nova Scotia, and Newfoundland?
- Figure B represents an island on the central coast of B.C. that has experienced 140 m of isostatic rebound since deglaciation, and has also been affected by the global eustatic sea-level rise over the same period. The dashed line marks sea level during glaciation. How much higher or lower should that line be now?

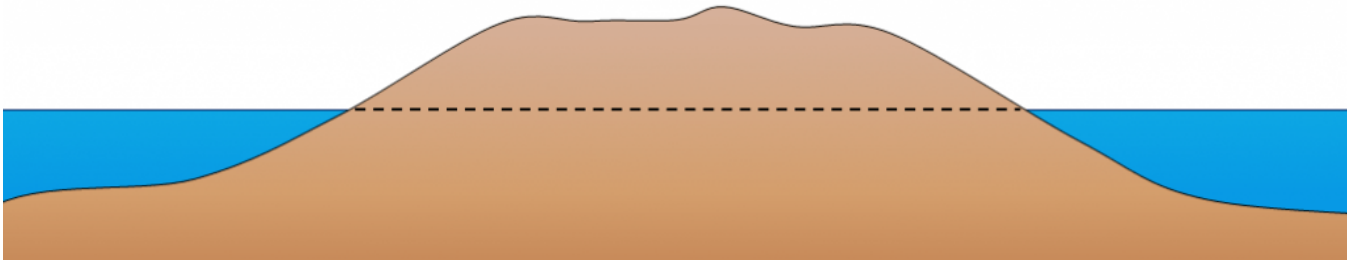


Figure B

- If a dam were to be built on the Fraser River near Hope, what would be the long-term implications for beaches in the Vancouver area? Explain why.

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18: GEOLOGY OF THE OCEANS

Marine geology or geological oceanography is the study of the history and structure of the ocean floor. It involves geophysical, geochemical, sedimentological and paleontological investigations of the ocean floor and coastal zone.

18.1: The Topography of the Sea Floor

18.2: The Geology of the Oceanic Crust

18.3: Sea-Floor Sediments

18.4: Ocean Water

18.5: Summary

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18.1: The Topography of the Sea Floor

We examined the topography of the sea floor from the perspective of plate tectonics in Chapter 10, but here we are going to take another look at bathymetry from an oceanographic perspective. The topography of the northern Atlantic Ocean is shown in Figure 18.1.1. The important features are the extensive **continental shelves** less than 250 meters deep (pink); the vast deep **ocean plains** between 4,000 and 6,000 meters deep (light and dark blue); the mid-Atlantic ridge, in many areas shallower than 3,000 meters; and the deep ocean trench north of Puerto Rico (8,600 meters).

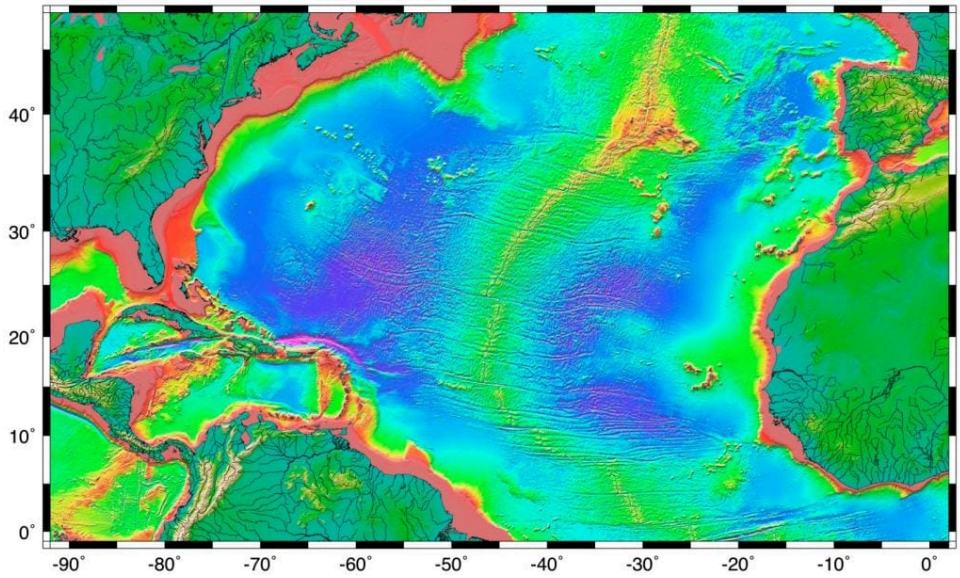


Figure 18.1.1 The topography of the Atlantic Ocean sea floor between 0° and 50° north. Red and yellow colors indicate less than a 2,000 metre depth; green less than 3,000 meters; blue 4,000 meters to 5,000 meters; and purple greater than 6,000 meters.

A topographic profile of the Pacific Ocean floor between Japan and British Columbia is shown in Figure 18.1.2. Be very careful when interpreting this diagram (and others like it), because in order to show the various features clearly the vertical axis is exaggerated, in this case by about 200 times. The floor of the Pacific, like those of the other oceans, is actually very flat, even in areas with seamounts or deep trenches. The vast sediment-covered **abyssal plains** of the oceans are much flatter than any similar-sized areas on the continents.

The main features of the Pacific Ocean floor are the continental slopes, which drop from about 200 meters to several thousand meters over a distance of a few hundred kilometers, the abyssal plains—exceedingly flat and from 4,000 meters to 6,000 meters deep, volcanic seamounts and islands, and trenches at subduction zones that are up to 11,000 meters deep.

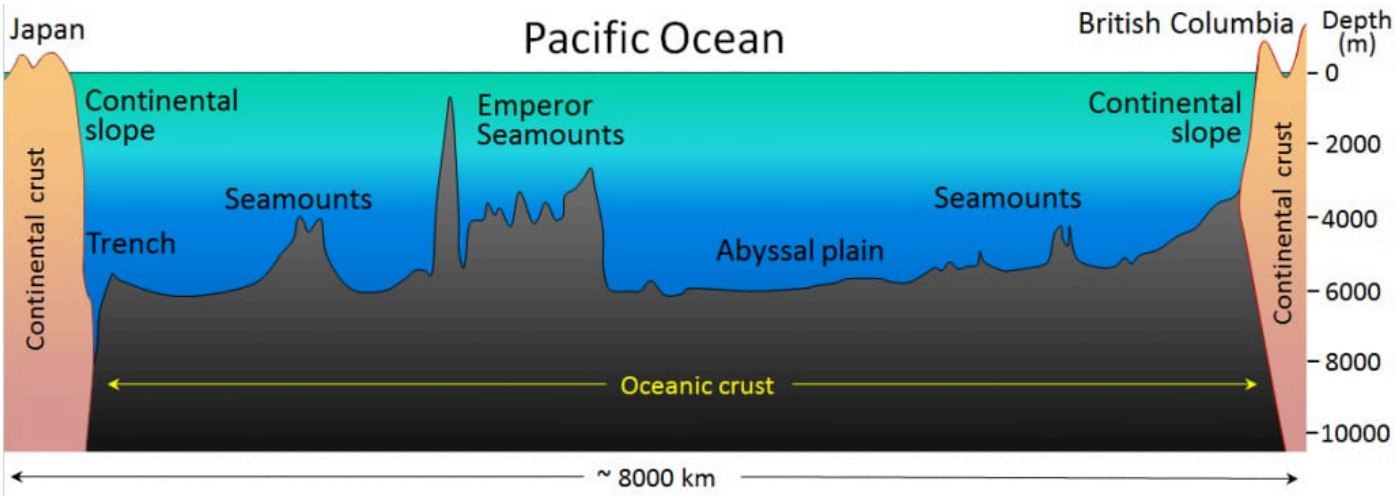


Figure 18.1.2 The generalized topography of the Pacific Ocean sea floor between Japan and British Columbia. The vertical exaggeration is approximately 200 times.

The ocean floor is almost entirely underlain by mafic oceanic crust (mostly basalt and gabbro, as described in more detail below), while the continental slopes are underlain by felsic continental crust (mostly granitic and sedimentary rocks). And, as you'll remember from Chapter 10, the heavier oceanic crust floats lower on the mantle than continental crust does, and that's why oceans are oceans.

The continental shelf and slope offshore from Nova Scotia is shown in Figure 18.1.3. In this passive-margin area (no subduction zone), the shelf is over 150 kilometers wide. On the Pacific coast of Canada, the shelf is less than half as wide. Continental shelves are typically less than 200 meters in depth; 200 meters is also the limit of the **photic zone**, the maximum depth to which sufficient light penetrates to allow photosynthesis to take place. As a result of that photosynthesis, the photic zone is oxygenated, and therefore suitable for animal life. Approximately 90% of marine life is restricted to the photic zone. The photic zone is also known as the **epipelagic zone**. The **mesopelagic zone** extends from 200 meters to 1,000 meters, the **bathypelagic zone** from 1,000 meters to 4,000 meters, and **abyssalpelagic zone** is deeper than 4,000 meters. (**Pelagic** refers to the open ocean, and thus excludes areas that are near to the shores or the ocean floor.)

Although the temperature of the ocean surface varies widely—from a few degrees either side of freezing in polar regions to over 25°C in the tropics—in most parts of the ocean, the water temperature is around 10°C at 1,000 meters depth and about 4°C from 2,000 meters depth all the way to the bottom.

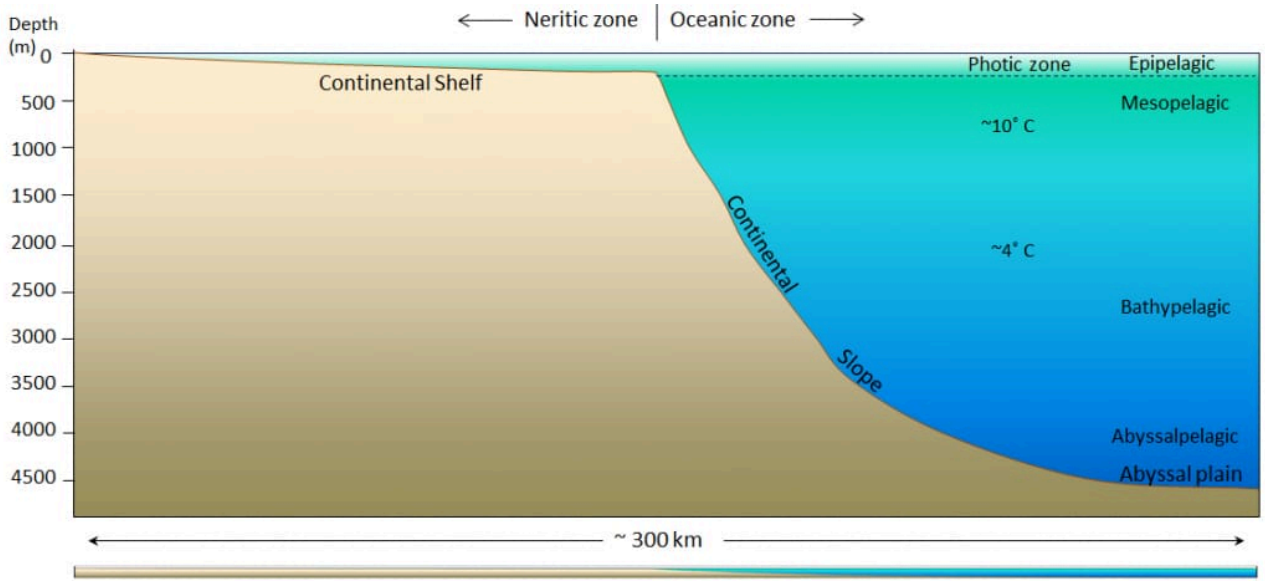


Figure 18.1.3 The generalized topography of the Atlantic Ocean floor within 300 kilometers of Nova Scotia. The vertical exaggeration is approximately 25 times. The panel at the bottom shows the same profile without vertical exaggeration.

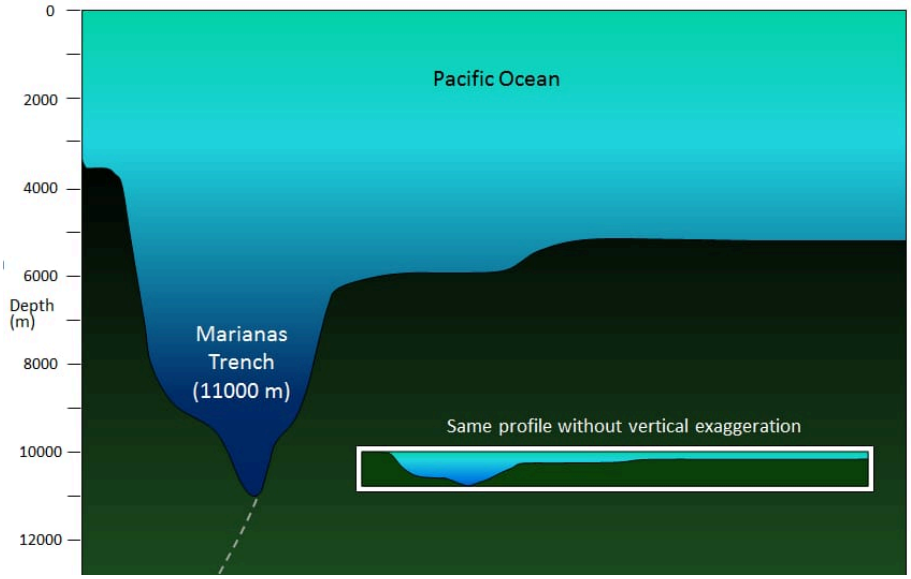


Figure 18.1.4 The generalized topography of the Pacific Ocean floor in the area of the Marianas Trench, near Guam. The dashed grey line represents the subduction of the Pacific Plate (to the right) beneath the Philippine Plate (to the left).

The deepest parts of the ocean are within the subduction trenches, and the deepest of these is the Marianas Trench in the southwestern Pacific (near Guam) at 11,000 meters (Figure 18.1.4). There are other trenches in the southwestern Pacific that are over 10,000 meters deep; the Japan Trench is over 9,000 meters deep; and the Puerto Rico and Chile-Peru Trenches are over 8,000 meters deep. Trenches that are relatively shallow tend to be that way because they have significant sediment infill. There is no recognizable trench along the subduction zone of the Juan de Fuca Plate because it has been filled with sediments from the Fraser and Columbia Rivers (or their ancient equivalents).

Exercise 18.1 Visualizing sea floor topography

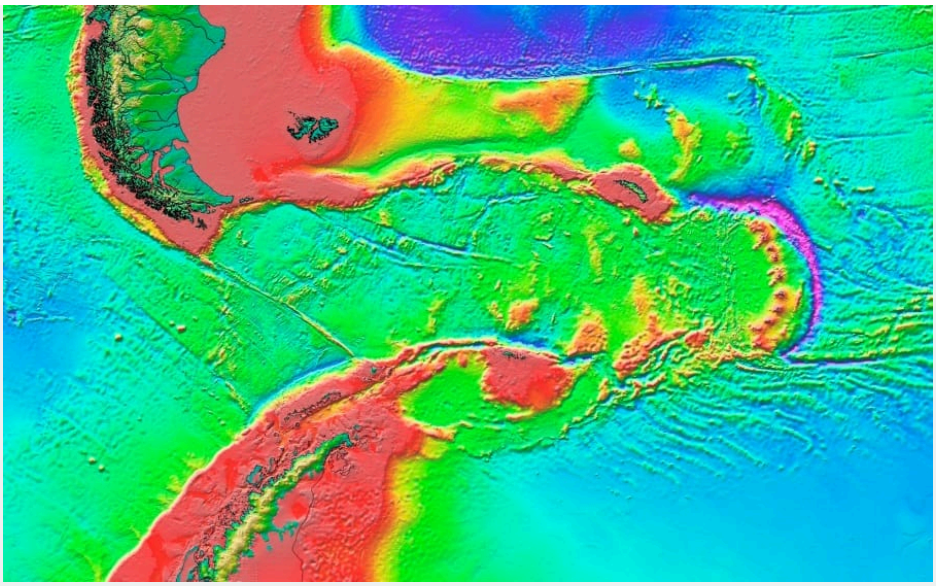


Figure 18.1.5

This map shows a part of the sea floor.

1. Identify the following features.
    1. a continental shelf
    2. a continental slope
    3. a spreading ridge

4. a subduction zone with a deep trench
5. an abyssal plain
6. some isolated seamounts

2. Where is this? (North is up.)

Image Descriptions

Figure 18.1.1 **image description:** Along the coast, the Atlantic Ocean is less than 2,000 meters deep. The depth increases farther from shore. In the middle, the Atlantic Ocean can be anywhere from 4000 to over 6,000 meters deep. A ridge stretches across the centre of the Atlantic Ocean in a northeast direction. The ridge causes the ocean depth to decreases to less than 2,000 meters. [Return to Figure 18.1.1]

Figure 18.1.3 **image description:** The continental shelf stretches out about 150 kilometers from the shore and the depth does not increase more than 300 meters. Once the continental slope begins, the depth continues to drop until it reaches 4500 meters at 300 kilometers from the shore. [Return to Figure 18.1.3]

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18.2: The Geology of the Oceanic Crust

As we discussed in Chapter 10, oceanic crust is formed at sea-floor spreading ridges from magma generated by decompression melting of hot upward-moving mantle rock (Figure 18.2.3). About 10% of the mantle rock melts under these conditions, producing mafic magma. This magma oozes out onto the sea floor to form pillow basalts (Figure 18.2.1), breccias (fragmented basaltic rock), and flows, interbedded in some cases with limestone or chert. Beneath the volcanic rock are layers with gabbroic sheeted dykes (which sometimes extend up into the pillow layer), gabbroic stocks, and finally layered peridotite (ultramafic rock) at the base. The ultramafic rock of the mantle lies below that. Over time, the igneous rock of the oceanic crust gets covered with layers of sediment, which eventually become sedimentary rock, including limestone, mudstone, chert, and turbidites. The lithologies of the layers of the oceanic crust are shown in Figure 18.2.1.

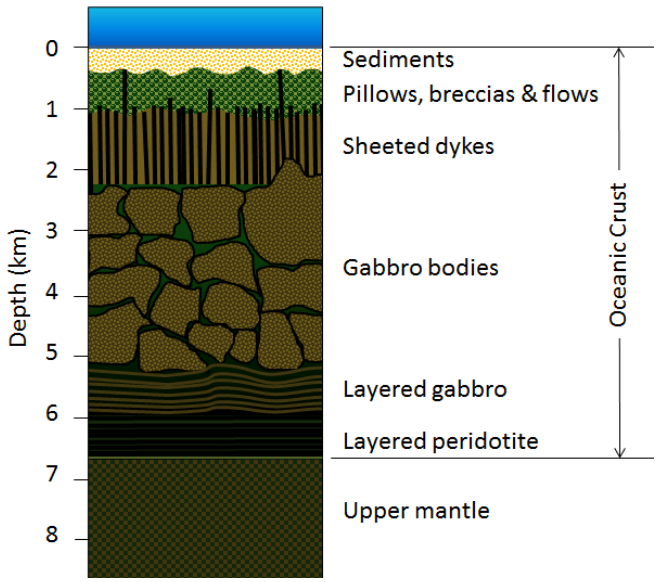


Figure 18.2.1 Schematic representation of the lithologic layers of typical oceanic crust.

The age of the oceanic crust has been determined by systematic mapping variations in the strength of the Earth's magnetic field across the sea floor and comparing the results with our understanding of the record of Earth's magnetic field reversal chronology for the past few hundred million years. The ages of different parts of the crust are shown in Figure 18.2.2. The oldest oceanic crust is around 280 Ma in the eastern Mediterranean, and the oldest parts of the open ocean are around 180 Ma on either side of the north Atlantic. It may be surprising, considering that parts of the continental crust are close to 4,000 Ma old, that the oldest sea floor is less than 300 Ma. Of course, the reason for this is that all sea floor older than that has been either subducted or pushed up to become part of the continental crust. For example, there are fragments of sea floor in British Columbia that date back to around 380 and 220 Ma, and there are similar rocks in the Canadian Shield that are older than 3 Ga.

As one would expect, the oceanic crust is very young near the spreading ridges (Figure 18.2.2), and there are obvious differences in the rate of sea-floor spreading along different ridges. The ridges in the Pacific and southeastern Indian Oceans have wide age bands, indicating rapid spreading (approaching 10 centimeters per year (cm/y) on each side in some areas), while those in the Atlantic and western Indian Oceans are spreading much more slowly (less than 2 cm/y on each side in some areas).

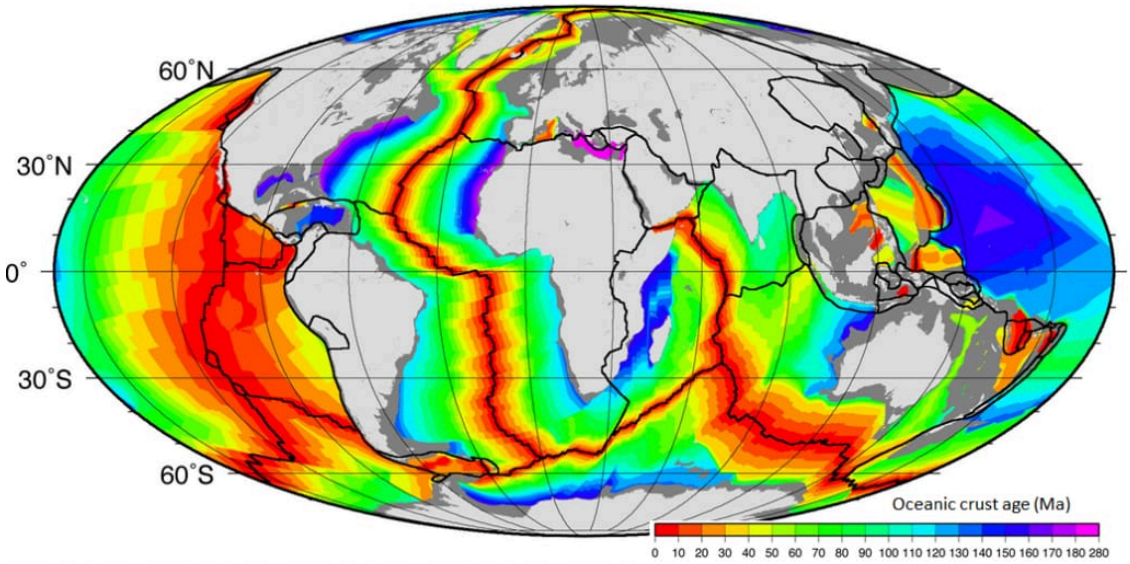


Figure 18.2.2 The age of the oceanic crust.

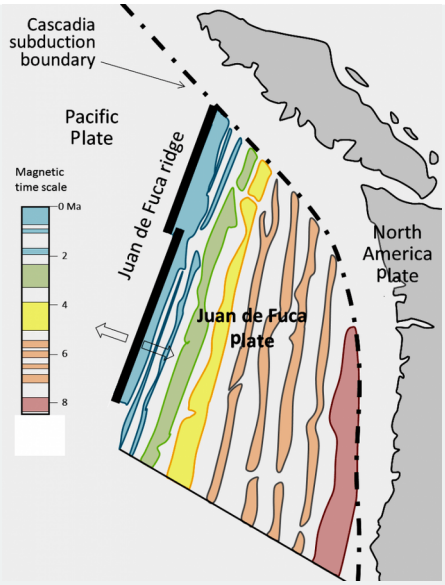


Figure 18.2.3

This map shows the magnetic patterns on the Juan de Fuca plate. The colored bands represent periods of normal magnetism, while the white bands represent reversed magnetism. A magnetic-reversal time scale is also shown.

1. How old is the oldest part of the Juan de Fuca Plate that is subducting along the Cascadia subduction boundary?
2. How old is the youngest part of the Juan de Fuca Plate that is subducting?

The magnetic patterns and chronology shown here have been color-coded to make them easy to interpret, but on most such maps the magnetic patterns are shown only as black and white stripes, making it much more difficult to interpret the ages of the sea floor. Magnetic-reversal patterns that have no context (such as the 0 age along the spreading ridge in this case) are very difficult to interpret.

As is evident from Figures 18.1.1 and 18.1.2, the sea floor is dotted with chains of seamounts, isolated seamounts, and ocean islands. Almost all of these features are volcanoes, and most are much younger than the oceanic crust on which they formed. Some seamounts and ocean islands are formed above mantle plumes, the best example being Hawaii. The oldest of the Hawaiian/Emperor seamounts is dated at around 80 Ma; it is situated on oceanic crust aged around 90 to 100 Ma. The youngest of the Hawaiian lavas—at Kilauea Volcano on the island of Hawaii—is now more than a year old (last eruption was April 30th 2018). The

island is surrounded by oceanic crust that is around 85 Ma old. All of the mantle-plume-derived volcanic islands are dominated by mafic rocks.

Many seamounts are related to subduction along ocean-ocean convergent boundaries. These include the Aleutians, extending from Alaska to Russia, and the Lesser Antilles in the eastern part of the Caribbean.

Some of the linear belts of volcanoes in the Pacific Ocean do not show age-distance relationships like the volcanoes of the Hawaii-Emperor chain or the Galapagos Islands. For example, the Line Islands, which spread out over more than 1,000 kilometers south of the Hawaiian chain, were all formed between 70 and 85 Ma and are interpreted to be related to rifting.

Most tropical islands have associated carbonate reefs, in some cases, as fringes right around the island, and in some cases, as barriers some distance away. In many cases, the reef is there, but the island that is assumed to have led to its formation is gone. The formation of **fringing reefs**, **barrier reefs**, and **atolls** is illustrated in Figure 18.2.4.

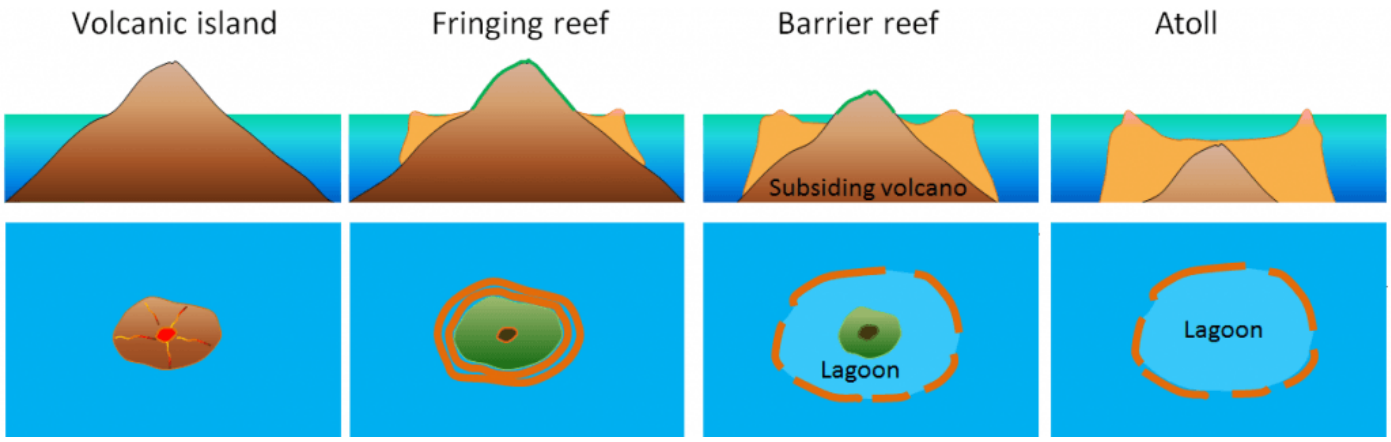


Figure 18.2.4 The formation of a fringing reef, a barrier reef, and an atoll around a subsiding tropical volcanic island.

The key factor in this process is sea-level change, either because of post-glacial sea-level rise, or because of subsidence of a volcano — as it is moved away from a spreading ridge — or both. If the rate of sea-level change is slow enough (e.g., less than 1 cm/year), a reef can keep up and maintain its position at sea level long after its parent volcanic island has disappeared beneath the waves.

Image Descriptions

Figure 18.2.1 image description: Layers in the oceanic crust.

Depth (kilometers)	Material
0 to 0.5	Sediments
0.5 to 1	Pillows, breccias, and flows
1 to 2	Sheeted dykes
2 to 5	Gabbro bodies
5 to 6	Layered gabbro
6 to 6.5	Layered peridotite
Greater than 6.5	Upper mantle

[Return to Figure 18.2.1]

Figure 18.2.3 image description: The Juan de Fuca plate lies between the Pacific Plate and the North America Plate along the west coast of Vancouver Island and Washington State. The Juan de Fuca Plate is subducting under the North America Plate along the Cascadia subduction boundary. The Juan de Fuca Plate is youngest along the Juan de Fuca ridge at the Pacific Plate and is older as it moves east. The magnetic time scales shows periods of magnetic reversal, and the ages of the parts of the Juan de

Fuca plate that are subducting range from just over 0 Ma in the northwest corner of the plate to over 8 Ma in the southeast corner of the plate. [Return to Figure 18.2.3]

Figure 18.2.4 image description: A volcanic island forms and a fringing reef develops around it in the water. It becomes a barrier reef as the volcanic island subsides and water is able to pool between the island and the reef to form a lagoon. An atoll is formed when the volcano subsides enough that it no longer breaches the ocean surface but the reef remains to form a pool. [Return to Figure 18.2.4]

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18.3: Sea-Floor Sediments

Except within a few kilometers of a ridge crest, where the volcanic rock is still relatively young, most parts of the sea floor are covered in sediments. This material comes from several different sources and is highly variable in composition, depending on proximity to a continent, water depth, ocean currents, biological activity, and climate. Sea-floor sediments (and sedimentary rocks) can range in thickness from a few millimeters to several tens of kilometers. Near the surface, the sea-floor sediments remain unconsolidated, but at depths of hundreds to thousands of meters (depending on the type of sediment and other factors) the sediment becomes lithified.

The various sources of sea-floor sediment can be summarized as follows:

- **Terrigenous** sediment is derived from continental sources transported by rivers, wind, ocean currents, and glaciers. It is dominated by quartz, feldspar, clay minerals, iron oxides, and terrestrial organic matter.
- Pelagic carbonate sediment is derived from organisms (e.g., **foraminifera**) living in the ocean water (at various depths, but mostly near surface) that make their shells (a.k.a. **tests**) out of carbonate minerals such as calcite.
- Pelagic silica sediment is derived from marine organisms (e.g., **diatoms** and **radiolaria**) that make their tests out of silica (microcrystalline quartz).
- Volcanic ash and other volcanic materials are derived from both terrestrial and submarine eruptions.
- Iron and manganese nodules form as direct precipitates from ocean-bottom water.

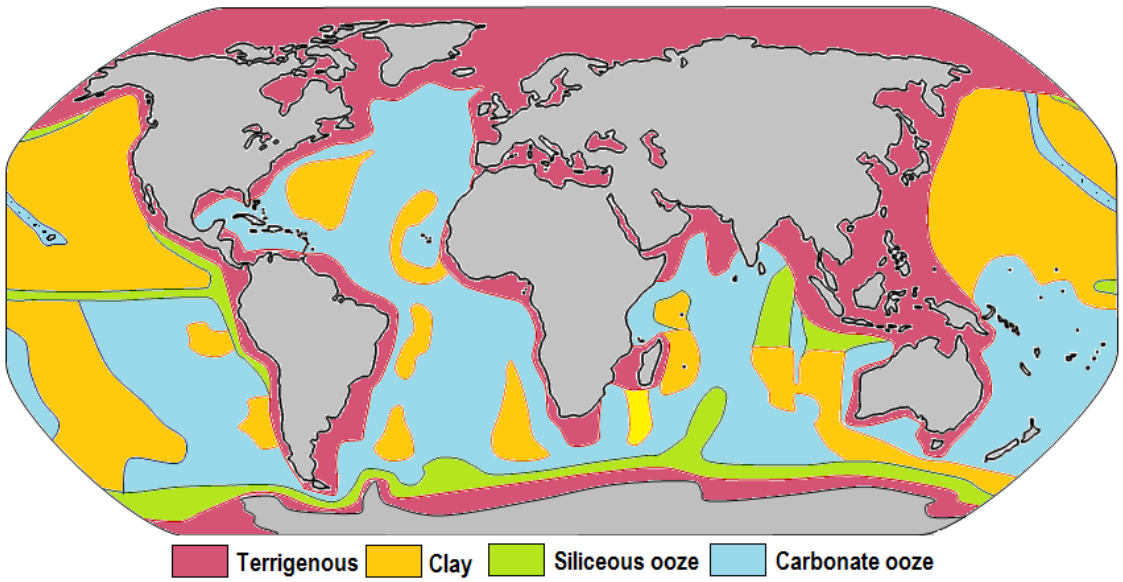


Figure 18.3.1 The distribution of sediment types on the sea floor. Within each colored area, the type of material shown is what dominates, although other materials are also likely to be present.

The distributions of some of these materials around the seas are shown in Figure 18.3.1. Terrigenous sediments predominate near the continents and within inland seas and large lakes. These sediments tend to be relatively coarse, typically containing sand and silt, but in some cases even pebbles and cobbles. Clay settles slowly in nearshore environments, but much of the clay is dispersed far from its source areas by ocean currents. Clay minerals are predominant over wide areas in the deepest parts of the ocean, and most of this clay is terrestrial in origin. Siliceous oozes (derived from radiolaria and diatoms) are common in the south polar region, along the equator in the Pacific, south of the Aleutian Islands, and within large parts of the Indian Ocean. Carbonate oozes are widely distributed in all of the oceans within equatorial and mid-latitude regions. In fact, clay settles everywhere in the oceans, but in areas where silica- and carbonate-producing organisms are prolific, they produce enough silica or carbonate sediment to dominate over clay.

Carbonate sediments are derived from a wide range of near-surface pelagic organisms that make their shells out of carbonate (Figure 18.3.2). These tiny shells, and the even tinier fragments that form when they break into pieces, settle slowly through the water column, but they don't necessarily make it to the bottom. While calcite is insoluble in surface water, its solubility increases with depth (and pressure) and at around 4,000 meters, the carbonate fragments dissolve. This depth, which varies with latitude

and water temperature, is known as the **carbonate compensation depth**, or CCD. As a result, carbonate oozes are absent from the deepest parts of the ocean (deeper than 4,000 meters), but they are common in shallower areas such as the mid-Atlantic ridge, the East Pacific Rise (west of South America), along the trend of the Hawaiian/Emperor Seamounts (in the northern Pacific), and on the tops of many isolated seamounts.



Figure 18.3.2 Foraminifera from the Ambergris Caye area of Belize. Most of the shells are about 1 millimetre across.

Exercise 18.3 What type of sediment?

The diagram shows the sea floor in an area where there is abundant pelagic carbonate sediment. There is a continent within 100 kilometers of this area, to the right. What type of sediment (coarse terrigenous, clay, siliceous ooze, or carbonate ooze) would you expect at find at locations a, b, c, and d?

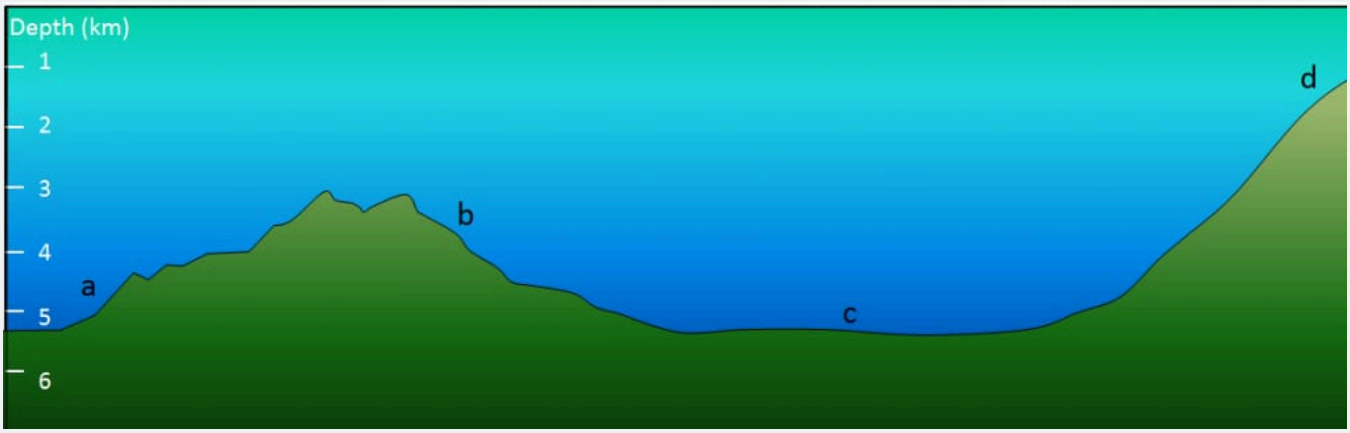


Figure 18.3.3

See Appendix 3 for Exercise 18.3 answers.

All terrestrial erosion products include a small proportion of organic matter derived mostly from terrestrial plants. Tiny fragments of this material plus other organic matter from marine plants and animals accumulate in terrigenous sediments, especially within a few hundred kilometers of shore. As the sediments pile up, the deeper parts start to warm up (from geothermal heat), and bacteria get to work breaking down the contained organic matter. Because this is happening in the absence of oxygen (a.k.a. **anaerobic** conditions), the by-product of this metabolism is the gas methane (CH_4). Methane released by the bacteria slowly bubbles upward through the sediment toward the sea floor.

At water depths of 500 meters to 1,000 meters, and at the low temperatures typical of the sea floor (close to 4°C), water and

methane combine to create a substance known as **methane hydrate**. Within a few meters to hundreds of meters of the sea floor, the temperature is low enough for methane hydrate to be stable and hydrates accumulate within the sediment (Figure 18.3.4). Methane hydrate is flammable because when it is heated, the methane is released as a gas (Figure 18.3.4). The methane within sea-floor sediments represents an enormous reservoir of fossil fuel energy. Although energy corporations and governments are anxious to develop ways to produce and sell this methane, anyone that understands the climate-change implications of its extraction and use can see that this would be folly. As we'll see in the discussion of climate change in Chapter 19, sea-floor methane hydrates have had significant impacts on the climate in the distant past.



Figure 18.3.4 Left: Methane hydrate within muddy sea-floor sediment from an area offshore from Oregon. Right: Methane hydrate on fire.

Image Descriptions

Figure 18.3.3 image description: A. is farthest from the continent. D is closest to the continent.

1. A depth of 4.5 kilometers.
2. A depth of 3.5 kilometers.
3. A depth of 5 kilometers.
4. A depth of 1 kilometre, close to the edge of a continent.

[Return to Figure 18.3.3]

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- Figure 18.3.4 (Right): "Burning Gas Hydrates" by J. Pinkston and L. Stern (USGS). Public domain.

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18.4: Ocean Water

As everyone knows, seawater is salty. It is that way because the river water that flows into the oceans contains small amounts of dissolved ions, and for the most part, the water that comes out of the oceans is the pure water that evaporates from the surface. Billions of years of a small amount of salt going into the ocean—and none coming out (most of the time)—has made the water salty. The salts of the ocean (dominated by sodium, chlorine, and sulfur) (Figure 18.4.1) are there because they are very soluble and they aren't consumed by biological processes (most of the calcium, for example, is used by organisms to make carbonate minerals). If salts are always going into the ocean, and never coming out, one might assume that the oceans have been continuously getting saltier over geological time. In fact this appears not to be the case. There is geological evidence that Earth's oceans became salty early during the Archaean, and that at times in the past, they have been at least half again as salty as they are now. This implies that there must be a mechanism to remove salt from the oceans, and that mechanism is the isolation of some parts of the ocean into seas (such as the Mediterranean) and the eventual evaporation of those seas to create salt beds that become part of the crust. The Middle Devonian Prairie Evaporite Formation of Saskatchewan and Manitoba is a good example of this.

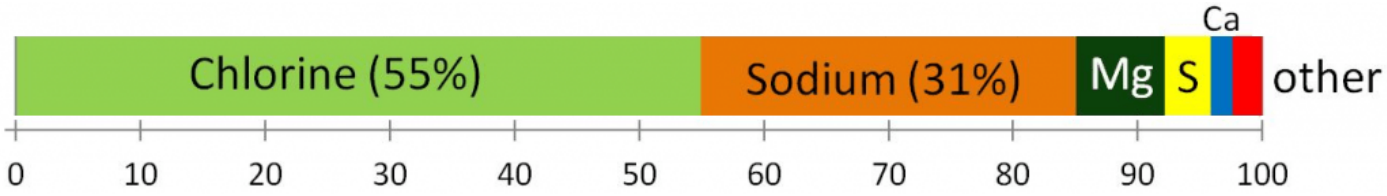


Figure 18.4.1 The proportions (by weight) of the major dissolved elements in ocean water.

The average salinity of the oceans is 35 g of salt per litre of water, but there are significant regional variations in this value, as shown in Figure 18.4.2. Ocean water is least salty (around 31 g/L) in the Arctic, and also in several places where large rivers flow in (e.g., the Ganges/Brahmaputra and Mekong Rivers in southeast Asia, and the Yellow and Yangtze Rivers in China). Ocean water is most salty (over 37 g/L) in some restricted seas in hot dry regions, such as the Mediterranean and Red Seas. You might be surprised to know that, in spite of some massive rivers flowing into it (such as the Nile and the Danube), water does not flow out of the Mediterranean Sea into the Atlantic. There is so much evaporation happening in the Mediterranean basin that water flows into it from the Atlantic, through the Strait of Gibraltar.

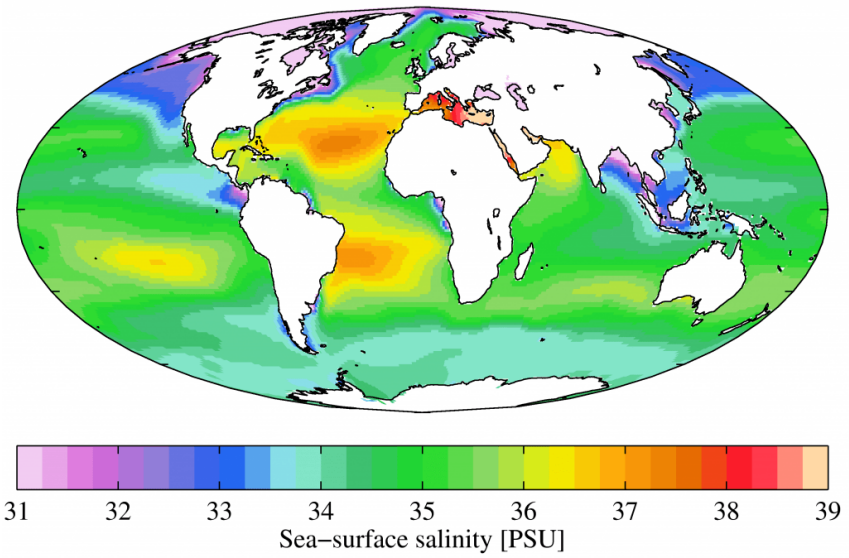


Figure 18.4.2 The distribution of salinity in Earth's oceans and major seas.

In the open ocean, salinities are elevated at lower latitudes because this is where most evaporation takes place. The highest salinities are in the subtropical parts of the

Atlantic, especially north of the equator. The northern Atlantic is much more saline than the north Pacific because the Gulf Stream current brings a massive amount of salty water from the tropical Atlantic and the Caribbean to the region around Britain, Iceland, and Scandinavia. The salinity in the Norwegian Sea (between Norway and Iceland) is substantially higher than that in other polar areas.

Exercise 18.4 Salt chunk



Figure 18.4.3

How salty is the sea? If you've ever had a swim in the ocean, you've probably tasted it. To understand how salty the sea is, start with 250 mL of water (1 cup). There is 35 g of salt in 1 L of seawater so in 250 mL (1/4 litre) there is 35/4 = 8.75 or ~9 g of salt. This is just short of 2 teaspoons, so it would be close enough to add 2 level teaspoons of salt to the cup of water. Then stir until it's dissolved. Have a taste!

Of course, if you used normal refined table salt, then what you added was almost pure NaCl. To get the real taste of seawater you would want to use some evaporated seawater salt (a.k.a. sea salt), which has a few percent of magnesium, sulfur, and calcium plus some trace elements.

See Appendix 3 for Exercise 18.4 answers.

Not unexpectedly, the oceans are warmest near the equator—typically 25° to 30°C—and coldest near the poles—around 0°C (Figure 18.4.4). (Sea water will remain unfrozen down to about -2°C.) At southern Canadian latitudes, average annual water temperatures are in the 10° to 15°C range on the west coast and in the 5° to 10°C range on the east coast. Variations in sea-surface temperatures (SST) are related to redistribution of water by ocean currents, as we'll see below. A good example of that is the plume of warm Gulf Stream water that extends across the northern Atlantic. St. John's, Newfoundland, and Brittany in France are at about the same latitude (47.5° N), but the average SST in St. John's is a frigid 3°C, while that in Brittany is a reasonably comfortable 15°C.

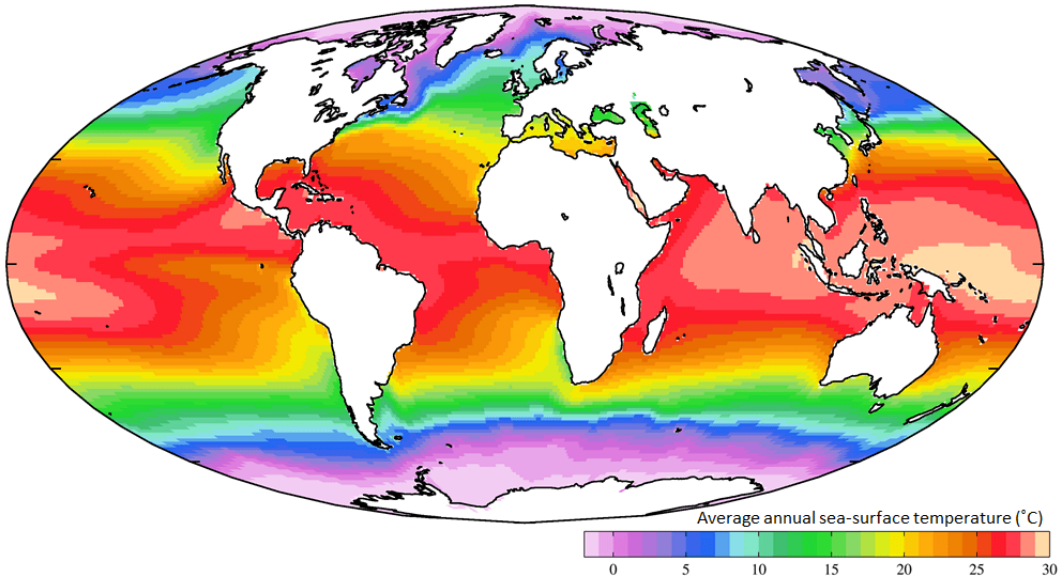


Figure 18.4.4 The global distribution of average annual sea-surface temperatures.

Currents in the open ocean are created by wind moving across the water and by density differences related to temperature and salinity. An overview of the main ocean currents is shown in Figure 18.4.5. As you can see, the northern hemisphere currents form circular patterns (**gyres**) that rotate clockwise, while the southern hemisphere gyres are counter-clockwise. This happens for the same reason that the water in your northern hemisphere sink rotates in a clockwise direction as it flows down the drain; this is caused by the **Coriolis effect**.

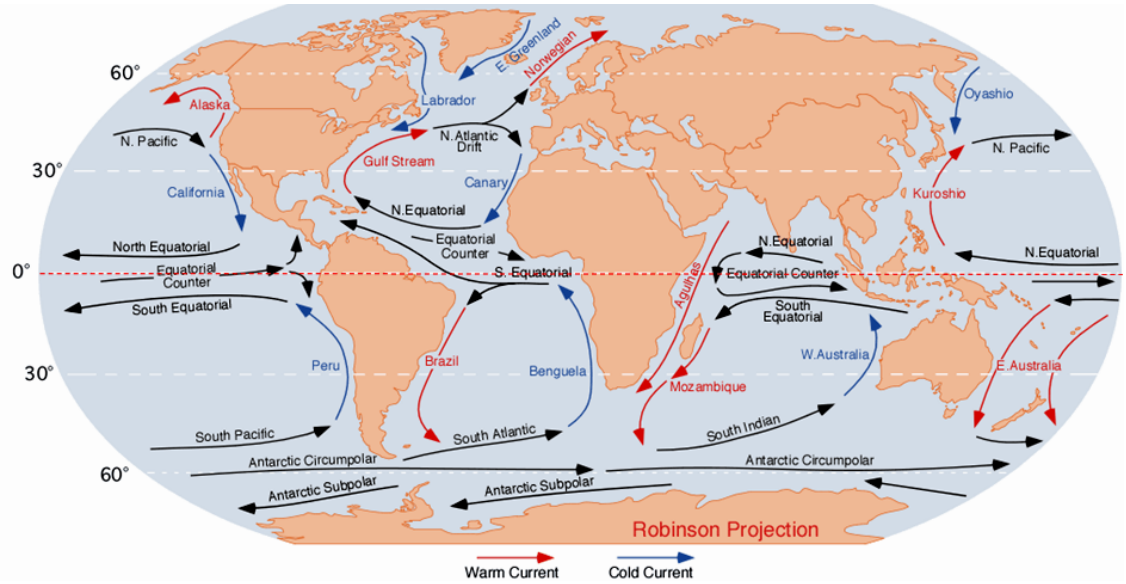


Figure 18.4.5 Overview of the main open-ocean currents. Red arrows represent warm water moving toward colder regions. Blue arrows represent cold water moving toward warmer regions. Black arrows represent currents that don't involve significant temperature changes.

Because the ocean basins aren't like bathroom basins, not all ocean currents behave the way we would expect. In the North Pacific, for example, the main current flows clockwise, but there is a secondary current in the area adjacent to our coast—the Alaska Current—that flows counter-clockwise, bringing relatively warm water from California, past Oregon, Washington, and B.C. to Alaska. On Canada's eastern coast, the cold Labrador Current flows south past Newfoundland, bringing a stream of icebergs past the harbour at St. John's (Figure 18.4.7). This current helps to deflect the Gulf Stream toward the northeast, ensuring that Newfoundland stays cool, and western Europe stays warm.



Figure 18.4.6 An iceberg floating past Exploits Island on the Labrador Current

Exercise 18.5: Understanding the Coriolis effect

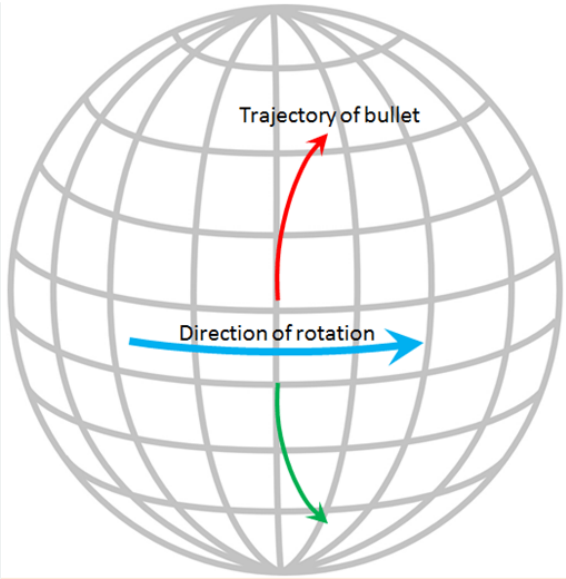


Figure 18.4.7

The Coriolis effect has to do with objects that are moving in relation to other objects that are rotating. An ocean current is moving across the rotating Earth, and its motion is controlled by the Coriolis effect.

Imagine that you are standing on the equator looking straight north and you fire a gun in that direction. The bullet in the gun starts out going straight north, but it also has a component of motion toward the east that it gets from Earth's rotation, which is 1,670 kilometers per hour at the equator. Because of the spherical shape of Earth, the speed of rotation away from the equator is not as fast as it is at the equator (in fact, the Earth's rotational speed is 0 kilometers per hour at the poles) so the bullet actually traces a clockwise curved path across Earth's surface, as shown by the red arrow on the diagram. In the southern hemisphere the Coriolis effect is counterclockwise (green arrow).

The Coriolis effect is imparted to the rotations of ocean currents and tropical storms. If Earth were a rotating cylinder, instead of a sphere, there would be no Coriolis effect.

See Appendix 3 for Exercise 18.5 answers.

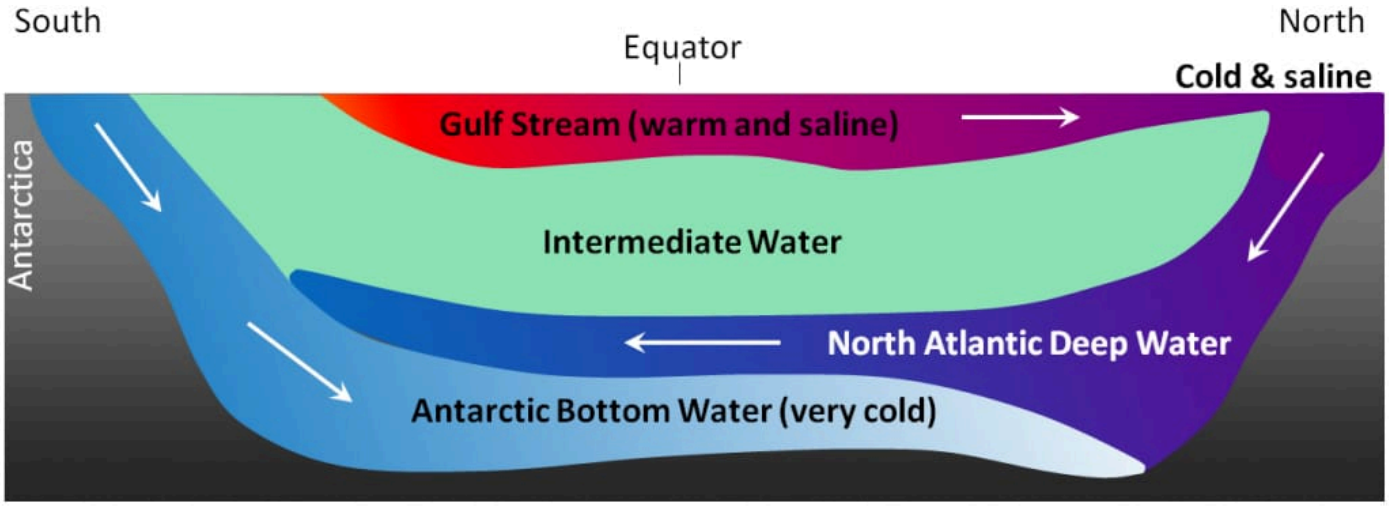


Figure 18.4.8 A depiction of the vertical movement of water along a north-south cross-section through the Atlantic basin.

The currents shown in Figure 18.4.5 are all surface currents, and they only involve the upper few hundred meters of the oceans. But there is much more going on underneath. The Gulf Stream, for example, which is warm and saline, flows past Britain and Iceland into the Norwegian Sea (where it becomes the Norwegian Current). As it cools down, it becomes denser, and because of its high salinity, which also contributes to its density, it starts to sink beneath the surrounding water (Figure 18.4.8). At this point, it is known as **North Atlantic Deep Water** (NADW), and it flows to significant depth in the Atlantic as it heads back south. Meanwhile, at the southern extreme of the Atlantic, very cold water adjacent to Antarctica also sinks to the bottom to become **Antarctic Bottom Water** (AABW) which flows to the north, underneath the NADW.

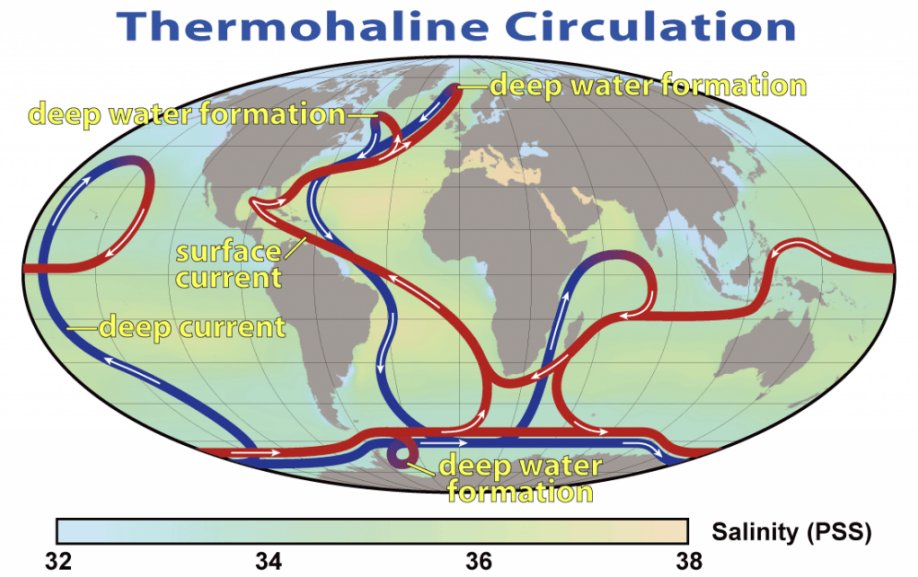


Figure 18.4.9 The thermohaline circulation system, also known as the Global Ocean Conveyor.

The descent of the dense NADW is just one part of a global system of seawater circulation, both at surface and at depth, as illustrated in Figure 18.4.9. The water that sinks in the areas of deep water formation in the Norwegian Sea and adjacent to Antarctica moves very slowly at depth. It eventually resurfaces in the Indian Ocean between Africa and India, and in the Pacific Ocean, north of the equator.

The thermohaline circulation is critically important to the transfer of heat on Earth. It brings warm water from the tropics to the poles, and cold water from the poles to the tropics, thus keeping polar regions from getting too cold and tropical regions from getting too hot. A reduction in the rate of thermohaline circulation would lead to colder conditions and enhanced formation of sea ice at the poles. This would start a positive feedback process that could result in significant global cooling. There is compelling evidence to indicate that there were major changes in thermohaline circulation, corresponding with climate changes, during the Pleistocene Glaciation.

Image Descriptions

Figure 18.4.5 **image description:** The currents of the world's oceans work together to form a number of general patterns. Currents flow into each other to form larger currents. Groups of currents in the northern hemisphere flow clockwise. This includes groups of currents in the North Pacific Ocean and the North Atlantic Ocean. Currents in the southern hemisphere flow counter-clockwise. This includes groups of currents in the South Pacific Ocean, the South Atlantic Ocean, and the Indian Ocean. Currents flowing towards the equator are colder than the surrounding water. Currents flowing away from the equator are warmer than the surrounding water. Currents below 60° South flow from east to west (or west to east) around Antarctica. Currents along the Equator also flows east to west (or west to east). Currents flowing from east to west (or west to east) are the same temperature as the surrounding water. For a more detailed description of specific currents, refer to the following table, which describes 26 major currents, including their location, direction of flow, and relationship to surrounding currents. They are arranged in alphabetical order. Or, you can [Return to Figure 18.4.5].

Figure 18.4.6 *table: Description of the main currents of the ocean and their direction of flow*

Name of Current	Temperature of current compared to the surrounding water	Direction of flow	Relationship to nearby currents
Agulhas	A warm current	The Agulhas current flows south from the Arabian peninsula down the east coast of Africa.	The Agulhas current joins with the Mozambique current, which also flows south.
Alaska	A warm current	The Alaska current flows north up the west coast of the United States and Canada before circling to the west once it reaches Alaska	The Alaska current flows into the Oyashio current
Antarctic Circumpolar	No temperature difference	The Antarctic Circumpolar current flows east to circle around Antarctica.	The Antarctic Circumpolar flows east above the Antarctic Subpolar, which flows west.
Antarctic Subpolar	No temperature difference	The Antarctic Subpolar current flows west along the coast of Antarctica.	The Antarctic Subpolar current flows west below the Antarctic Circumpolar current, which flows east.
Benguela	A cold current	The Benguela current flows north along the south west coast of Africa.	The Benguela current flows into the South Equatorial current and is fed by the South Atlantic current.
Brazil	A warm current	The Brazil current flows south along the east coast of South America.	The Brazil current flows into the South Atlantic current and is fed by the south branch of the South Equatorial current.
California	A cold current	The California current flows south from the southwest coast of the United States down along the west coast of Mexico.	The California current flows into the North Equatorial current and is fed by the North Pacific current.
Canary	A cold current	The Canary current flows from south along the north west coast of Africa from Morocco to Sengal.	The Canary current flows into the North Equatorial current and is fed by the North Atlantic Drift.

Name of Current	Temperature of current compared to the surrounding water	Direction of flow	Relationship to nearby currents
East Australian	A warm current	The East Australian currents flow from the equator and south, past the east coast of Australia. One flows between New Zealand and Australia, and the other flows past the east side of New Zealand.	The East Australian currents flow into the South Pacific current and they are fed by the South Equatorial current.
East Greenland	A cold current	The East Greenland current flows south along the east coast of Greenland.	The East Greenland current flows into the Labrador current.
Equatorial Counter	No temperature difference.	The Equatorial Counter current flows east along the equator. It is broken up into three sections: One in the Pacific Ocean, one in the Atlantic Ocean, and one in the Indian Ocean.	The Equatorial Counter current flows east between the North Equatorial and the South Equatorial currents, which both flow west.
Gulf Stream	A warm current	The Gulf Stream flows north from the Caribbean along the east coast of the United States.	The Gulf Stream flows into the North Atlantic Drift current and is fed by the North Equatorial current.
Kuroshio	A warm current	The Kuroshio current flows north along the east coast of the Philippines and Japan.	The Kuroshio current flows into the North Pacific current and is fed by the North Equatorial current.
Labrador	A cold current	The Labrador current flows south along the eastern coast of Canada to the northern United States.	The Labrador current is partially fed by the East Greenland current. Once it reaches the northern United States, it flows past the Gulf Stream.
Mozambique	A warm current	The Mozambique current flows south along the east coast of Madagascar and into the Southern Ocean.	The Mozambique current flows into the South Indian current and is fed by the South Equatorial current.
North Atlantic Drift	No temperature difference	The North Atlantic Drift current flows east across the Atlantic Ocean from the north coast of the United States to the south coast of Spain.	The North Atlantic Drift current splits to flow north into the Norwegian current and to flow south into the Canary current. It is fed by the Gulf Stream.
North Equatorial	No temperature difference.	The North Equatorial current flows west just above the equator. It is broken up into three sections: One in the Pacific Ocean, one in the Atlantic Ocean, and one in the Indian Ocean.	The North Equatorial current in the Pacific Ocean flows into the Kuroshio current and is fed by the California current. The North Equatorial current in the Atlantic Ocean flows into the Gulf Stream and is fed by the Canary current. The North Equatorial current in the Indian Ocean turns at Africa to join the Equatorial Counter current.

Name of Current	Temperature of current compared to the surrounding water	Direction of flow	Relationship to nearby currents
North Pacific	No temperature difference	The North Pacific current flows west across the Pacific Ocean from Japan to the south coast of the United States.	The North Pacific current flows into the California current and is fed by the Kuroshio current.
Norwegian	A warm current	The Norwegian current flows north from the north coast of the United Kingdom to along the coast of Norway.	This current is fed by the northern branch of the North Atlantic Drift current and flows into the Arctic Ocean.
Oyashio	A cold current	The Oyashio current flows south along the east coast of Russia.	The Oyashio current clashes with the Kuroshio current, which flows north into the North Pacific current.
Peru	A cold current	The Peru current flows north along the central west coast of South America.	The Peru current flows into the South Equatorial current and is fed by the South Pacific current.
South Atlantic	No temperature difference	The South Atlantic current flows from the south tip of South America to towards the south tip of Africa.	The South Atlantic current flows into the Benguela current and is fed by the Brazil current.
South Equatorial	No temperature difference	The South Equatorial current flows west just below the equator. It is broken up into three sections: One in the Pacific Ocean, one in the Atlantic Ocean, and one in the Indian Ocean.	The South Equatorial current in the Pacific Ocean flows into the East Australian currents and is fed by the Peru current. The South Equatorial current in the Atlantic Ocean flows north and south: North along the north east coast of South America and south into the Brazil current. It is fed by the Benguela current. The South Equatorial current in the Indian Ocean flows into the Mozambique current and is fed by the West Australian current.
South Indian	No temperature difference	The South Indian current flows from the southern part of the Indian ocean towards the south west coast of Australia.	The South Indian current flows into the West Australian current and is fed by the Mozambique current.
South Pacific	No temperature difference	The South Pacific current flows east from the south east coast of Australia to the south west coast of South America.	The South Pacific current flows into the Peru current and is fed by the East Australian current.
West Australian current	A cold current	The West Australian current flows north along the west coast of Australia.	The West Australian current flows into the South Equatorial current and is fed by the South Indian current.

[Return to Figure 18.4.5]

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18.5: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
18.1 The Topography of the Sea Floor	The oceans are about 4,000 meters deep on average, but they also have a wide range of topographical features, including shallow continental shelves, continental slopes, continuous ridges related to plate divergence, numerous isolated seamounts, and deep submarine canyons at subduction zones.
18.2 The Geology of the Oceanic Crust	Most oceanic crust forms during sea-floor spreading and is characterized by pillow basalts, sheeted dykes, gabbro bodies, layered gabbro, and layered ultramafic rock. The oldest parts of the sea floor are older than 200 Ma, but most of the sea floor is younger than 100 Ma. Seamounts are common and almost all are volcanoes, related to mantle plumes, subduction, or other processes. In tropical regions, ocean islands tend to be surrounded by carbonate reefs.
18.3 Sea-Floor Sediments	Almost all of the sea floor is covered by young sediments and sedimentary rocks, derived either from erosion of continents or from marine biological processes. Clastic sediments, some quite coarse, predominate on shelves and slopes. Terrigenous clays are distributed across the sea floor, but in areas where either carbonate- or silica-forming organisms thrive, the sediments are likely to be dominated by carbonate or silica oozes. Methane hydrates, derived from bacterial decomposition of organic matter, form within sediments on shelves and slopes.
18.4 Ocean Water	Average ocean water has about 35 g/L of salt, mostly made up of chlorine and sodium, but also including magnesium, sulfur, and calcium. Salinity levels are highest in the tropics where evaporation is greatest. Sea-surface temperatures range from less than 0°C at the poles to over 25°C in equatorial regions. Open-ocean currents, which generally rotate clockwise in the northern hemisphere and counter-clockwise in the south, are critically important in redistributing heat on Earth. Deep-ocean currents, driven by density differences, are another key part of the heat redistribution system. Changes to current patterns or intensity have significant implications for global climate.

Questions for Review

See Appendix 2 for answers to review questions.

1. What is the origin of the sediments that make up continental shelves? Why are the shelves on the eastern coast of North America so much wider than those along the west coast?
2. The ocean trenches at some subduction zones are relatively shallow. What is one explanation for this?
3. What are the main lithological components of oceanic crust, and how does this rock form?
4. Referring to Figure 18.8, determine the age of the oldest sea floor in the Indian Ocean.
5. Explain why relatively coarse terrigenous sediments (e.g., sand) tend to accumulate close to the continents, while terrigenous clay is dispersed all across the ocean floor.
6. Although clay is widely dispersed in the oceans, in some areas, deep-sea sediments are dominated by clay, while in others they are dominated by carbonate or silica ooze. Why do these differences exist?

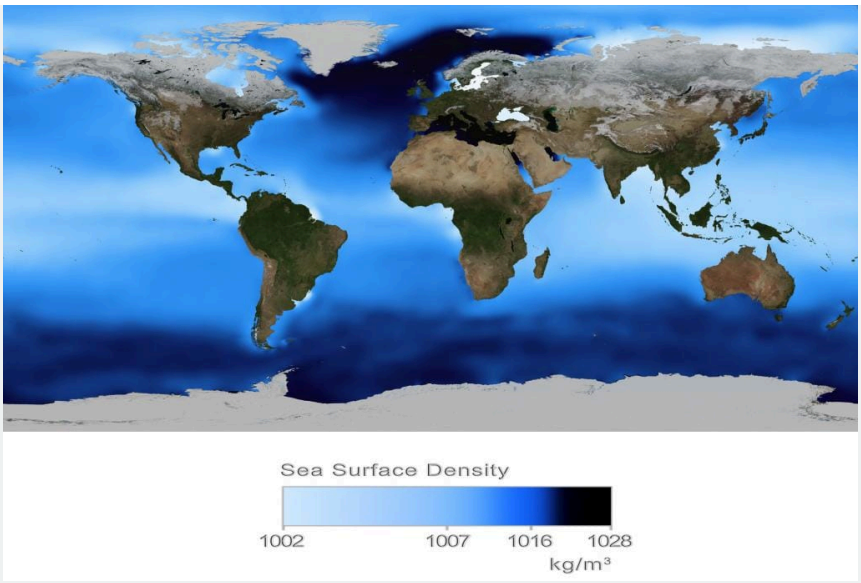


Figure A

- Explain why carbonate sediments are absent from the deepest parts of the oceans.
- What is the source of the carbon that is present in sea-floor methane hydrate deposits?
- Where are the saltiest parts of the oceans? Why?
- Explain why sea-surface water with the greatest density is found in the north Atlantic, as shown on Figure A.
- What type of ocean currents result from the relatively dense water in the north Atlantic?
- How do the open-ocean currents affect the overall climate patterns on Earth?

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19: CLIMATE CHANGE

19.1: What Makes the Climate Change?
19.2: Anthropogenic Climate Change
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19.1: What Makes the Climate Change?

There are two parts to climate change, the first one is known as **climate forcing**, which is when conditions change to give the climate a little nudge in one direction or the other. The second part of climate change, and the one that typically does most of the work, is what we call a **feedback**. When a climate forcing changes the climate a little, a whole series of environmental changes take place, many of which either exaggerate the initial change (**positive feedbacks**), or suppress the change (**negative feedbacks**).

An example of a climate-forcing mechanism is the increase in the amount of carbon dioxide (CO_2) in the atmosphere that results from our use of fossil fuels. CO_2 traps heat in the atmosphere and leads to climate warming. Warming changes vegetation patterns; contributes to the melting of snow, ice, and **permafrost**; causes sea level to rise; reduces the solubility of CO_2 in sea water; and has a number of other minor effects. Most of these changes contribute to more warming. Melting of permafrost, for example, is a strong positive feedback because frozen soil contains trapped organic matter that is converted to CO_2 and methane (CH_4) when the soil thaws. Both these gases accumulate in the atmosphere and add to the warming effect. On the other hand, if warming causes more vegetation growth, that vegetation should absorb CO_2, thus reducing the warming effect, which would be a negative feedback. Under our current conditions—a planet that still has lots of glacial ice and permafrost—most of the feedbacks that result from a warming climate are positive feedbacks and so the climate changes that we cause get naturally amplified by natural processes.

What is a greenhouse gas?

Throughout this chapter we'll be talking about the role of **greenhouse gases** (GHGs) in controlling the climate, so it's important to understand what greenhouse gases are and how they work. As you know, the dominant gases of the atmosphere are nitrogen (as N_2) and oxygen (as O_2). These gas molecules have only two atoms each and are not GHGs. Some of the other important gases of the atmosphere are water vapor (H_2O), carbon dioxide (CO_2), and methane (CH_4). All of these have more than two atoms, and they are GHGs.

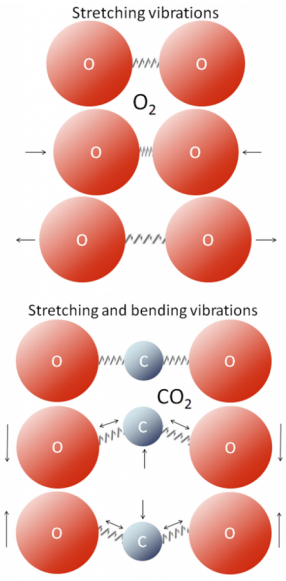


Figure 19.1.1 Stretching versus bending vibrations in atmospheric gases

All molecules vibrate at various frequencies and in various ways, and some of those vibrations take place at frequencies within the range of the infrared (IR) radiation that is emitted by Earth's surface. Gases with two atoms, such as O_2, can only vibrate by stretching (back and forth), and those vibrations are much faster than the IR radiation (Figure 19.1.1). Gases with three or more atoms (such as CO_2) vibrate by stretching as well, but they can also vibrate in other ways, such as by bending. Those vibrations are slower and match IR radiation frequencies.

When IR radiation interacts with CO_2 or with one of the other GHGs, the molecular vibrations are enhanced because there is a match between the wavelength of the IR light and the vibrational frequency of the molecule. This makes the molecule vibrate more vigorously, heating the surrounding air in the process. These molecules also emit IR radiation in all directions, some of which reaches Earth's surface. The heating caused by the more vigorous vibrations of GHGs is the **greenhouse effect**.

Natural Climate Forcing

Natural climate forcing has been going on throughout geological time. A wide range of processes has been operating at widely different time scales, from a few years to billions of years.

The longest-term natural forcing variation is related to the evolution of the Sun. Like most other stars of a similar mass, our Sun is evolving. For the past 4.57 billion years, its rate of nuclear fusion has been increasing, and it is now emitting about 40% more energy (as light) than it did at the beginning of geological time (Figure 19.1.2). A difference of 40% is big, so it's a little surprising that the temperature on Earth has remained at a reasonable and habitable temperature for all of this time. The mechanism for that relative climate stability has been the evolution of our atmosphere from one that was dominated by CO_2, and also had significant levels of CH_4—both GHGs—to one with only a few hundred parts per million of CO_2 and just under 1 part per million of CH_4. Those changes to our atmosphere have been no accident; over geological time, life and its metabolic processes have evolved and changed the atmosphere to conditions that remained cool enough to be habitable. A scientific explanation for how this could happen is known as the **Gaia hypothesis**.

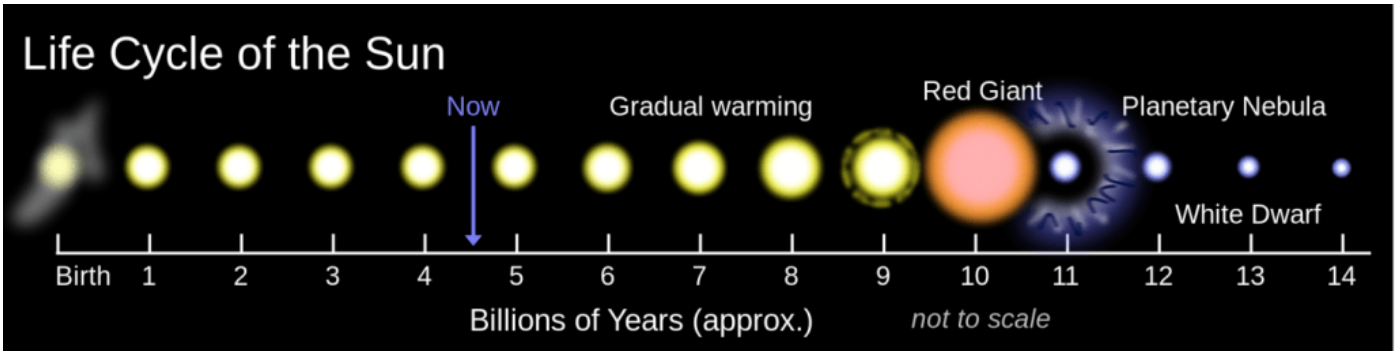


Figure 19.1.2 The life cycle of our Sun and of other similar stars.

The Gaia hypothesis



Figure 19.1.3 You can read more about Lovelock, Gaia, and Daisyworld by searching the Internet using any one of those terms.

The Gaia hypothesis, developed by British scientist and environmentalist James Lovelock in the 1960s, is the theory that

organisms evolve in ways that contribute to ensuring that their environment remains habitable. It does not include any sort of coordination of effort among organisms or any consciousness of a need to make changes. Gaia is not a superorganism. A way of understanding Gaia is through Lovelock's simple Daisyworld model. A planet with a warming star is populated only by two types of daisies, white ones and black ones (Figure 19.1.3). The black ones contribute to warming because they absorb solar energy, while the white ones reflect light and contribute to cooling. As the star's luminosity gradually increases, the white daisies have better outcomes because their reflectivity cools their local environment, while the black daisies, suffering from the heat, do not reproduce as well. Over time white daisies gradually dominate the population, but eventually the star becomes so bright that even white daisies cannot compensate, and all of the daisies perish. Obviously Earth is not Daisyworld, but similar processes—such as the evolution of photosynthetic bacteria that consume CO_2—have taken place that influence the atmosphere and moderate the climate.

Plate tectonic processes contribute to climate forcing in several different ways, and on time scales ranging from tens of millions to hundreds of millions of years. One mechanism is related to continental position. For example, we know that Gondwana (South America + Africa + Antarctica + Australia) was positioned over the South Pole between about 450 and 250 Ma, during which time there were two major glaciations (Andean-Saharan and Karoo) affecting the South polar regions (Figure 19.1.1) and cooling the rest of the planet at the same time. Another mechanism is related to continental collisions. As described in Chapter 16, the collision between India and Asia, which started at around 50 Ma, resulted in massive tectonic uplift. The consequent accelerated weathering of this rugged terrain consumed CO_2 from the atmosphere and contributed to gradual cooling over the remainder of the Cenozoic. Also, as described in Chapter 16, the opening of the Drake Passage — due to plate-tectonic separation of South America from Antarctica — led to the development of the Antarctic Circumpolar Current, which isolated Antarctica from the warmer water in the rest of the ocean and thus contributed to Antarctic glaciation starting at around 35 Ma.

As we discussed in Chapter 4, volcanic eruptions don't just involve lava flows and exploding rock fragments; various particulates and gases are also released, the important ones being sulfur dioxide and CO_2. Sulphur dioxide is an aerosol that reflects incoming solar radiation and has a net cooling effect that is short lived (a few years in most cases, as the particulates settle out of the atmosphere within a couple of years), and doesn't typically contribute to longer-term climate change. Volcanic CO_2 emissions can contribute to climate warming but only if a greater-than-average level of volcanism is sustained over a long time (at least tens of thousands of years). It is widely believed that the catastrophic end-Permian extinction (at 250 Ma) resulted from warming initiated by the eruption of the massive Siberian Traps over a period of at least a million years.

Exercise 19.1 Climate change at the K-Pg boundary

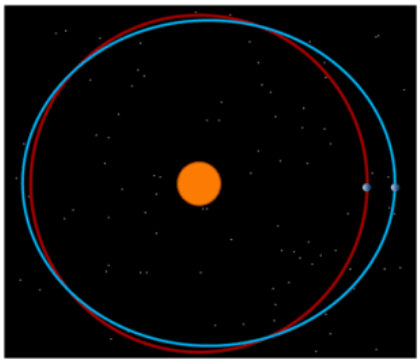


Figure 19.1.4

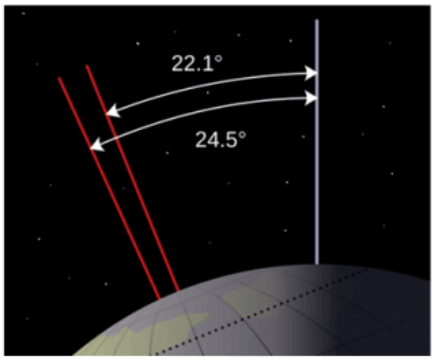
The large extraterrestrial impact at the end of the Cretaceous (Cretaceous-Paleogene or K-Pg boundary, a.k.a. K-T boundary) is thought to have produced a massive amount of dust, which may have remained in the atmosphere for several years, and also a great deal of CO_2. What do you think would have been the short-term and longer-term climate-forcing implications of these two factors?

See Appendix 3 for Exercise 19.1 answers.

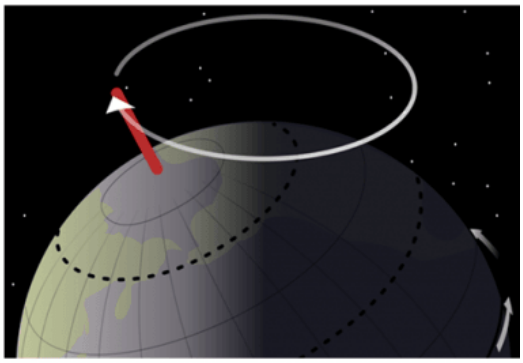
Earth's orbit around the Sun is nearly circular, but like all physical systems, it has natural oscillations. First, the shape of the orbit changes on a regular time scale — close to 100,000 years — from being close to circular to being very slightly elliptical. But the circularity of the orbit is not what matters; it is the fact that as the orbit becomes more elliptical, the position of the Sun within that ellipse becomes less central or more eccentric (Figure 19.1.5a). **Eccentricity** is important because when it is high, the Earth-Sun distance varies more from season to season than it does when eccentricity is low.



(a) The 100,000 year cycle of eccentricity of the Sun within the Earth's orbit. The blue orbit is more elliptical than the red one, and the Sun is offset from the centre of the ellipse



(b) The 41,000 year cycle of the angle of tilt (obliquity) of the Earth's axis (red lines) compared to a line perpendicular to the plane of the Earth's orbit (light blue line).



(c) the 20,000 year cycle of precession of the Earth's rotational axis

Figure 19.1.5 The cycles of Earth's orbit and rotation.

Second, Earth rotates around an axis through the North and South Poles, and that axis is at an angle to the plane of Earth's orbit around the Sun (Figure 19.1.5b). The angle of tilt (also known as **obliquity**) varies on a time scale of 41,000 years. When the angle is at its maximum (24.5°), Earth's seasonal differences are accentuated. When the angle is at its minimum (22.1°), seasonal differences are minimized. The current hypothesis is that glaciation is favoured at low seasonal differences as summers would be cooler and snow would be less likely to melt and more likely to accumulate from year to year.

Third, the direction in which Earth's rotational axis points also varies, on a time scale of about 20,000 years (Figure 19.1.5c). This variation, known as **precession**, means that although the North Pole is presently pointing to the star Polaris (the pole star), in 10,000 years it will point to the star Vega.

The importance of eccentricity, tilt, and precession to Earth's climate cycles (now known as **Milankovitch Cycles**) was first pointed out by Yugoslavian engineer and mathematician Milutin Milankovitch in the early 1900s. Milankovitch recognized that although the variations in the orbital cycles did not affect the total amount of **insolation** (light energy from the Sun) that Earth received, it did affect where on Earth that energy was strongest. Glaciations are most sensitive to the insolation received at latitudes of around 65°, and with the current configuration of continents, it would have to be 65° north (because there is almost no land at 65° south).

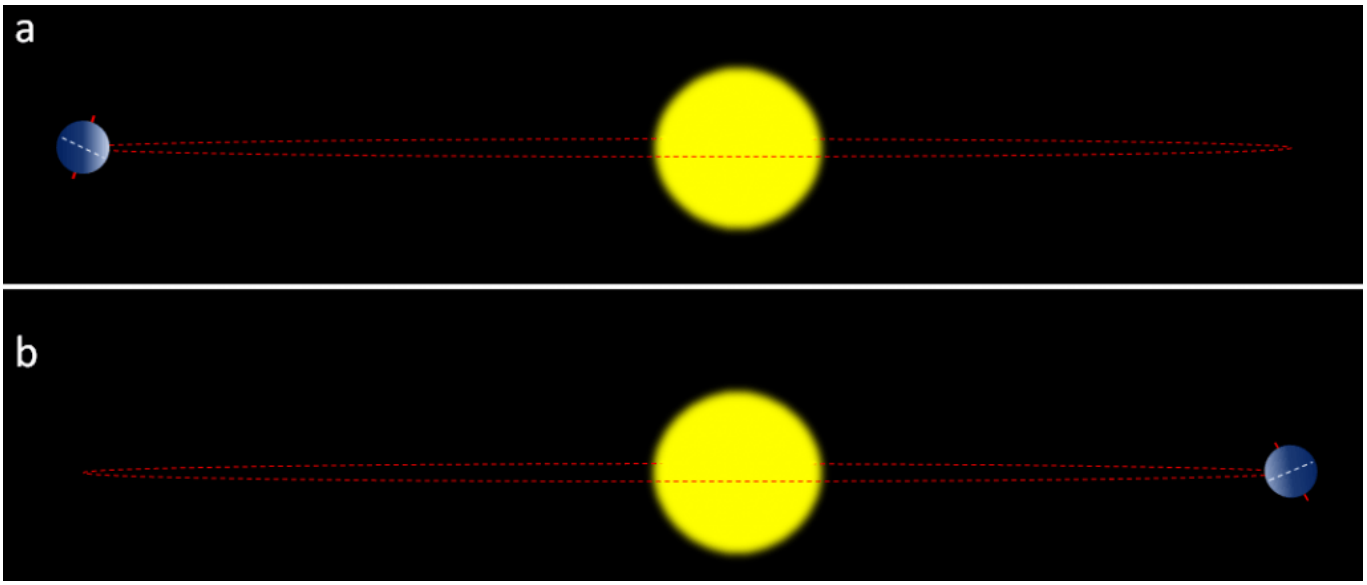


Figure 19.1.6 The effect of precession on insolation in the northern hemisphere summers. In (a) the northern hemisphere summer takes place at greatest Earth-Sun distance, so summers are cooler. In (b) (10,000 years or one-half precession cycle later) the opposite is the case, so summers are hotter. The red dashed line represents Earth's path around the Sun.

The most important aspects are whether the northern hemisphere is pointing toward the Sun at its closest or farthest approach, and how eccentric the Sun's position is in Earth's orbit. Two opposing situations are illustrated in Figure 19.1.6. In the upper panel, the northern hemisphere is at it farthest distance from the Sun during summer, which means cooler summers. In the lower panel, the northern hemisphere is at its closest distance to the Sun during summer, which means hotter summers. Cool summers—as opposed to cold winters—are the key factor in the accumulation of glacial ice, so the upper scenario in Figure 19.1.6 is the one that promotes glaciation. This factor is greatest when eccentricity is high.

Data for tilt, eccentricity, and precession over the past 400,000 years have been used to determine the insolation levels at 65° north, as shown in Figure 19.1.7. Also shown in Figure 19.1.7 are Antarctic ice-core temperatures from the same time period. The correlation between the two is clear, and it shows up in the Antarctic record because when insolation changes lead to growth of glaciers in the northern hemisphere, southern-hemisphere temperatures are also affected.

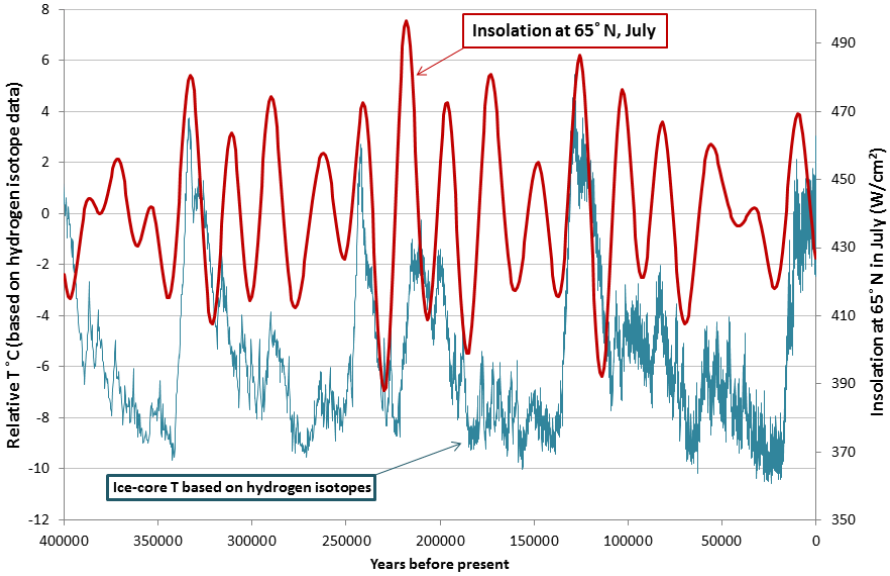


Figure 19.1.7 Insolation at 65° N in July compared with Antarctic ice-core temperatures.

Ocean currents are important to climate, and currents also have a tendency to oscillate. Glacial ice cores show clear evidence of changes in the Gulf Stream (and other parts of the thermohaline circulation system) that affected global climate on a time scale of about 1,500 years during the last glaciation. The east-west changes in sea-surface temperature and surface pressure in the

equatorial Pacific Ocean—known as the El Niño Southern Oscillation or ENSO—varies on a much shorter time scale of between two and seven years. These variations tend to garner the attention of the public because they have significant climate implications in many parts of the world. The past 65 years of ENSO index values are shown in Figure 19.1.8. The strongest **El Niños** in recent decades were in 1983 and 1998, and those were both very warm years from a global perspective. During a strong El Niño, the equatorial Pacific sea-surface temperatures are warmer than normal and heat the atmosphere above the ocean, which leads to warmer-than-average global temperatures.

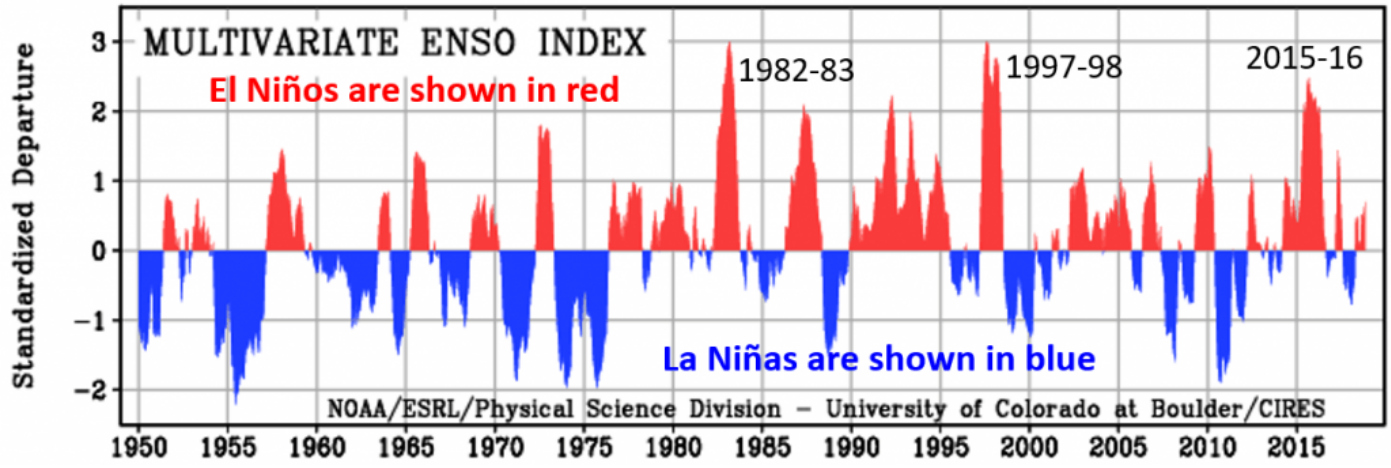


Figure 19.1.8 Variations in the ENSO index from 1950 to early-2019.

Climate Feedbacks

As already stated, **climate feedbacks** are critically important in amplifying weak climate forcings into full-blown climate changes. When Milankovitch published his theory in 1924, it was widely ignored, partly because it was evident to climate scientists that the forcing produced by the orbital variations was not strong enough to drive the significant climate changes of the glacial cycles. Those scientists did not recognize the power of positive feedbacks. It wasn't until 1973, 15 years after Milankovitch's death, that sufficiently high-resolution data were available to show that the Pleistocene glaciations were indeed driven by the orbital cycles, and it became evident that the orbital cycles were just the forcing that initiated a range of feedback mechanisms that made the climate change.

Since Earth still has a very large volume of ice — mostly in the continental ice sheets of Antarctica and Greenland, but also in alpine glaciers and permafrost — melting is one of the key feedback mechanisms. Melting of ice and snow leads to several different types of feedbacks, an important one being a change in **albedo**. Albedo is a measure of the reflectivity of a surface. Earth's various surfaces have widely differing albedos, expressed as the percentage of light that reflects off a given material. This is important because most solar energy that hits a very reflective surface is not absorbed and therefore does little to warm Earth. Water in the oceans or on a lake is one of the darkest surfaces, reflecting less than 10% of the incident light, while clouds and snow or ice are among the brightest surfaces, reflecting 70% to 90% of the incident light (Figure 19.1.9).

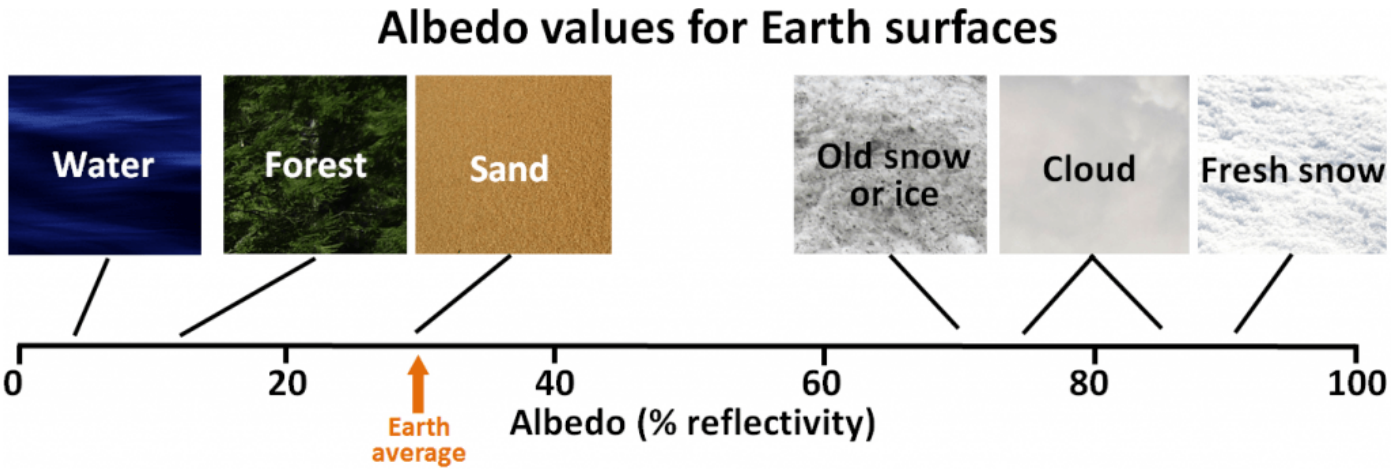


Figure 19.1.9 Typical albedo values for Earth surfaces.

Exercise 19.2 Albedo implications of forest harvesting



Figure 19.1.10

When a forest is harvested, there are significant changes to the rate of biological uptake of CO_2, but there is also a change in albedo. Using Figure 19.1.9, and assuming that a clear-cut has an albedo similar to sand, what is the albedo-only implication of clear-cutting for climate change? Be aware that this implication applies only to the albedo change caused by clear-cutting. Clear-cutting reduces the capacity of the area to absorb CO_2, and also has numerous negative ecological and geological implications.

See Appendix 3 for Exercise 19.2 answers.

When sea ice melts, as it has done in the Arctic Ocean at a disturbing rate over the past decade, the albedo of the area affected changes dramatically, from around 80% down to less than 10%. This is a positive feedback because much more solar energy is absorbed by the water than by the pre-existing ice, and the temperature increase is amplified. The same applies to ice and snow on land, but the difference in albedo is not as great.

When ice and snow on land melt, sea level rises. (Sea level is also rising because the oceans are warming and that increases their volume.) A higher sea level means a larger proportion of the planet is covered with water, and since water has a lower albedo than land, more heat is absorbed and the temperature goes up a little more. Since the last glaciation, sea-level rise has been about 125 m; a huge area that used to be land is now flooded by heat-absorbent seawater. During the current period of anthropogenic climate change, sea level has risen only about 20 cm, and although that doesn't make a big change to albedo, sea-level rise is accelerating.



Figure 19.1.11 A degrading permafrost site on the north coast of Alaska.

Most of northern Canada has a layer of permafrost that ranges from a few centimeters to hundreds of meters in thickness; the same applies in Alaska, Russia, and Scandinavia. Permafrost is a mixture of soil and ice (Figure 19.1.11), and it also contains a significant amount of trapped organic carbon that is released as CO_2 and CH_4 when the permafrost breaks down. Because the amount of carbon stored in permafrost is in the same order of magnitude as the amount released by burning fossil fuels, this is a feedback mechanism that has the potential to equal or surpass the forcing that has unleashed it.

In some polar regions, including northern Canada, permafrost includes methane hydrate (see section 18.3), a highly concentrated form of CH_4 trapped in solid form. Breakdown of permafrost releases this CH_4. Even larger reserves of methane hydrate exist on the sea floor, and while it would take significant warming of ocean water down to a depth of hundreds of meters, this too is likely to happen in the future if we don't limit our impact on the climate. There is strong isotopic evidence that the Paleocene-Eocene thermal maximum (see Figure 19.1.1) was caused, at least in part, by a massive release of sea-floor methane hydrate.

There is about 45 times as much carbon in the ocean (as dissolved bicarbonate ions, HCO_3^-) as there is in the atmosphere (as CO_2), and there is a steady exchange of carbon between the two reservoirs. But the solubility of CO_2 in water decreases as the temperature goes up. In other words, the warmer it gets, the more of that oceanic bicarbonate gets transferred to the atmosphere as CO_2. That makes CO_2 solubility another positive feedback mechanism.

Vegetation growth responds positively to both increased temperatures and elevated CO_2 levels, and so in general, it represents a negative feedback to climate change because the more the vegetation grows, the more CO_2 is taken from the atmosphere. But it's not quite that simple because when trees grow bigger and more vigorously, forests become darker (they have lower albedo) so they absorb more heat. Furthermore, climate warming isn't necessarily good for vegetation growth; some areas have become too hot, too dry, or even too wet to support the plant community that was growing there, and it might take centuries for something to replace it successfully.

All of these positive (and negative) feedbacks work both ways. For example, during climate cooling, growth of glaciers leads to higher albedos, and formation of permafrost results in storage of carbon that would otherwise have returned quickly to the atmosphere.

Image Descriptions

Figure 19.1.2 image description: Our sun is roughly 4.5 billion years old. Over the next 5 billion years, it will continue getting warmer until it becomes a red giant. Six billion years from now, it will become a planetary nebula and then a white dwarf. [Return to Figure 19.1.2]

Figure 19.1.5 image description:

1. The eccentricity of the Sun within the Earth's orbit goes through a 100,000 year cycle. The Earth's orbit around the sun can

be very circular, but when it becomes more oval-shaped, the Sun is offset from the centre of the ellipse. This means that sometimes the Earth will be farther from the sun than normal and sometimes it will be closer to the sun than normal.
2. The angle of tilt (or obliquity) or the Earth's axis goes through a 41,000 year cycle. When comparing the North Pole to a line perpendicular to the plane of the earth's orbit, the obliquity can range from 22.1° to 24.5°.
3. The Earth's axial precession operates on a 20,000 year cycle.

[Return to Figure 19.1.5]

Figure 19.1.9 **image description:** The albedo of Earth's surfaces: Water=around 5%; Forest=10% to 15%; Sand=30%; Old snow and ice=70%; Cloud=75% to 85%. Fresh snow=90%. Earth's average albedo is 30%. [Return to Figure 19.1.9]

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- Figure 19.1.8: "Multivariate ENSO Index (MEI)" by NOAA. Adapted by Steven Earle. Public domain.
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19.2: Anthropogenic Climate Change

When we talk about anthropogenic climate change, we are generally thinking of the industrial era, which really got going when we started using fossil fuels (coal to begin with) to drive machinery and trains. That was around the middle of the 18th century. The issue with fossil fuels is that they involve burning carbon that was naturally stored in the crust over hundreds of millions of years as part of Earth's process of counteracting the warming Sun.

Some climate scientists argue that anthropogenic climate change actually goes back much further than the industrial era, and that humans began to impact the climate by clearing land to grow grains in Europe and the Middle East around 8,000 years BCE and by creating wetlands to grow rice in Asia around 5,000 years BCE. Clearing forests for crops is a type of climate-forcing because the CO_2 storage capacity of the crops is generally lower than that of the trees they replace, and creating wetlands is a type of climate forcing because the anaerobic bacterial decay of organic matter within wetlands produces CH_4.

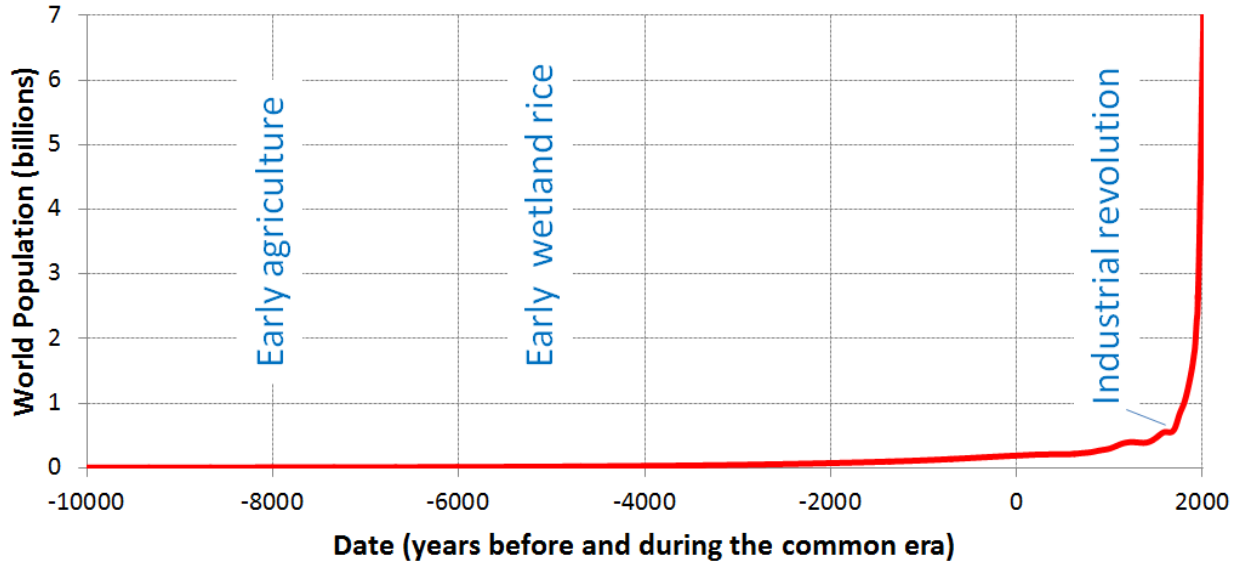


Figure 19.2.1 World population growth over the past 12,000 years.

In fact, whether anthropogenic climate change started with the agricultural revolution or the industrial revolution is not important, because the really significant climate changes didn't start until the early part of the 20th century, and although our activities are a major part of the problem, our increasing numbers are a big issue as well. Figure 19.2.1 shows the growth of the world population from around 5 million, when we first started growing crops, to about 18 million when wetland rice cultivation began, to over 800 million at the start of the industrial revolution, to over 7,700 million in 2019. A big part of the incredible growth in our population is related to the availability of the cheap and abundant energy embodied in fossil fuels, which we use for transportation, heating and cooling, industry, and food production. It will be hard to support a population of this size without fossil fuels, but we have to find a way to do it.

A rapidly rising population, the escalating level of industrialization and mechanization of our lives, and an increasing dependence on fossil fuels for transportation and energy generation have driven the anthropogenic climate change of the past century. The trend of mean global temperatures since 1880 is shown in Figure 19.2.2. For approximately the past 55 years, the temperature has increased at a relatively steady and disturbingly rapid rate, especially compared to past changes. The average temperature now is approximately 1.0°C higher than before industrialization, and two-thirds of this warming has occurred since 1975. One of the driving factors of the recent increase in the rate of climate change has been the migration of North Americans from city centres to the suburbs, and the resulting need for virtually every household to own at least one car, when previously they were able to get around on foot or public transit.

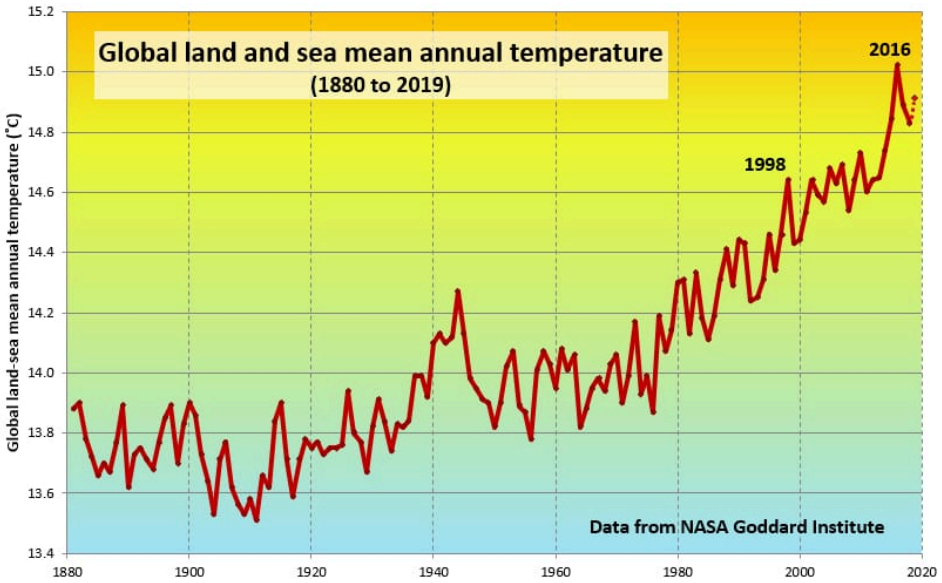


Figure 19.2.2 Global mean annual temperatures for the period from 1880 to 2019. The value for 2019 is projected, based on only 6 months of data. 1998 and 2016 were strong El Niño years.

The **Intergovernmental Panel on Climate Change** (IPCC)—established by the United Nations in 1988—is responsible for reviewing the scientific literature on climate change and issuing periodic reports on several topics, including the scientific basis for understanding climate change, our vulnerability to observed and predicted climate changes, and what we can do to limit climate change and minimize its impacts. Figure 19.2.3, which is based on data from the fifth assessment report of the IPCC, issued in 2014, shows the relative contributions of various long-lasting anthropogenic GHGs to current climate forcing, based on the changes from levels that existed in 1750.

The biggest anthropogenic contributor to warming is CO_2, which accounts for 56% of positive forcing. CH_4 accounts for 32%, and the halocarbon gases (mostly leaked from older air-conditioning appliances that still contain CFCs) and nitrous oxide (N_2O) (from burning fossils fuels) account for 6% each.

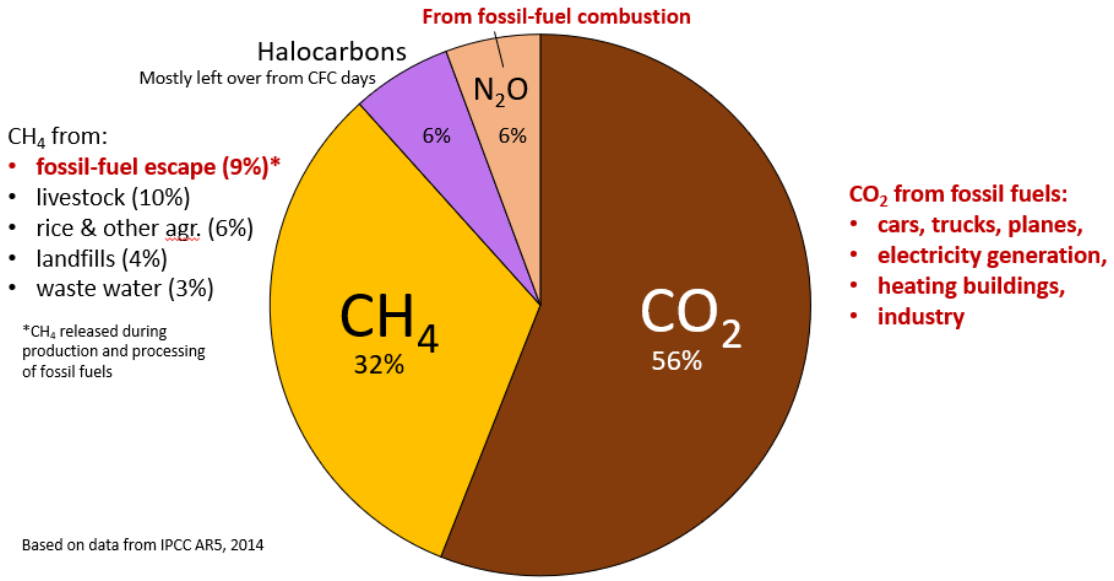


Figure 19.2.3 The relative climate-warming effects of the various greenhouse gases. (Effects listed in red are all related to fossil fuels.)

CO_2 emissions come mostly from coal- and gas-fired power stations, motorized vehicles (cars, trucks, and aircraft), and industrial operations (e.g., smelting). CH_4 emissions come from production of fossil fuels (escape from coal mining and from gas and oil production and processing), livestock farming (mostly beef), landfills, waste water, and wetland rice farming. N_2O is derived almost entirely from the combustion of fossil fuels. In summary, most (by far) of our current GHG emissions come from fossil fuel

production and use.

Exercise 19.3 What are the primary sources of climate forcing?

Figure 19.2.3 shows the proportions of climate forcing related to the 4 main greenhouse gases, and also includes lists of where those gases come from. The ones that are primarily derived from our use of fossil fuels are shown in red text.

Using the numbers provided in Figure 19.2.3 determine the percentage proportions of GHGs from the 4 main sources listed in the table below. Make sure that your percentages add to 100.

Source	Percentage
Fossil fuel production and use	
Food production	
Solid and liquid wastes	
Halocarbons	

See Appendix 3 for Exercise 19.3 answers.

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19.3: Implications of Climate Change

Although we've all experienced the effects of climate change over the past decade it's not straightforward for climatologists to make the connection between a warming climate and specific weather events, and most are justifiably reluctant to ascribe any specific event to climate change. In this respect, the best measures of climate change are those that we can detect over several decades, such as the temperature changes shown in Figure 19.3.2, or the sea-level rise shown in Figure 19.3.1. As already stated, sea level has risen approximately 20 cm since 1750, and that rise is attributed to both warming (and therefore expanding) seawater and melting glaciers and other land-based snow and ice (melting of sea ice does not contribute directly to sea-level rise as it is already floating in the ocean).

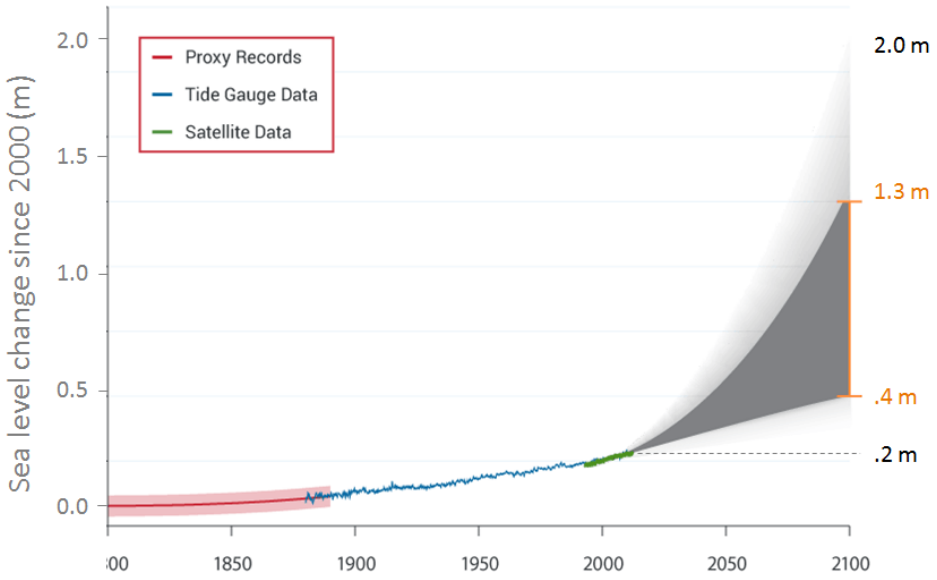


Figure 19.3.1 Projected sea-level increases to 2100, showing likely range (grey) and possible maximum.

Projections for sea-level rise to the end of this century vary widely. This is in large part because we do not know which of the possible climate change scenarios we will most closely follow, but many are in the range from 0.5 m to 2.0 m. One of the problems in predicting sea-level rise is that we do not have a strong understanding of how large ice sheets—such as Greenland and Antarctica—will respond to future warming. Another issue is that the oceans don't respond immediately to warming. For example, with the current amount of warming, we are already committed to a future sea-level rise of between 1.3 m and 1.9 m, even if we could stop climate change today. This is because it takes decades to centuries for the existing warming of the atmosphere to be transmitted to depth within the oceans and to exert its full impact on large glaciers. Most of that committed rise would take place over the next century, but some would be delayed longer. And for every decade that the current rates of climate change continue, that number increases by another 0.3 m. In other words, if we don't make changes quickly, by the end of this century we'll be locked into about 3 m of future sea-level rise.

In a 2008 report, the Organisation for Economic Co-operation and Development (OECD) estimated that by 2070 approximately 150 million people living in coastal areas could be at risk of flooding due to the combined effects of sea-level rise, increased storm intensity, and land subsidence. The assets at risk (buildings, roads, bridges, ports, etc.) are in the order of $35 trillion ($35,000,000,000,000). Countries with the greatest exposure of population to flooding are China, India, Bangladesh, Vietnam, U.S.A., Japan, and Thailand. Some of the major cities at risk include Shanghai, Guangzhou, Mumbai, Kolkata, Dhaka, Ho Chi Minh City, Tokyo, Miami, and New York.



Figure 19.3.2 Damage to the Casino Pier, Seaside Heights, New Jersey, from Hurricane Sandy, November 2012.

One of the other risks for coastal populations, besides sea-level rise, is that climate warming is also associated with an increase in the intensity of tropical storms (e.g., hurricanes or typhoons), which almost always bring serious flooding from intense rain and storm surges. Some recent examples are New Orleans in 2005 with Hurricane Katrina, and New Jersey and New York in 2012 with Hurricane Sandy (Figure 19.3.2).

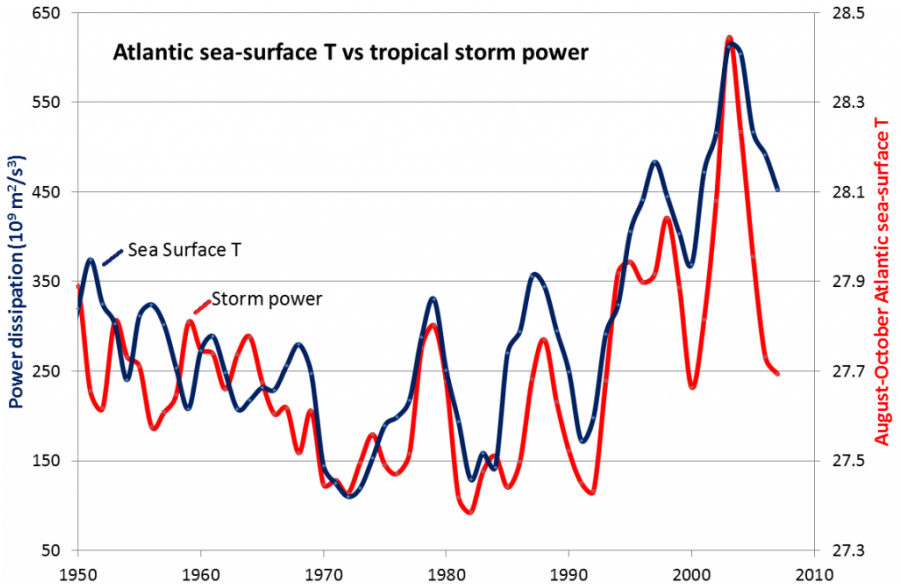


Figure 19.3.3 Relationship between Atlantic tropical storm cumulative annual intensity and Atlantic sea-surface temperatures.

Tropical storms get their energy from the evaporation of warm seawater in tropical regions. In the Atlantic Ocean, this takes place between 8° and 20° N in the summer. Figure 19.3.3 shows the variations in the sea-surface temperature (SST) of the tropical Atlantic Ocean (in blue) versus the amount of power represented by Atlantic hurricanes between 1950 and 2008 (in red). Not only has the overall intensity of Atlantic hurricanes increased with the warming since 1975, but the correlation between hurricanes and sea-surface temperatures is very strong over that time period.

Because warm air is able to hold more water than cold air, the general global trend over the past century has been one of increasing precipitation (Figure 19.3.4).

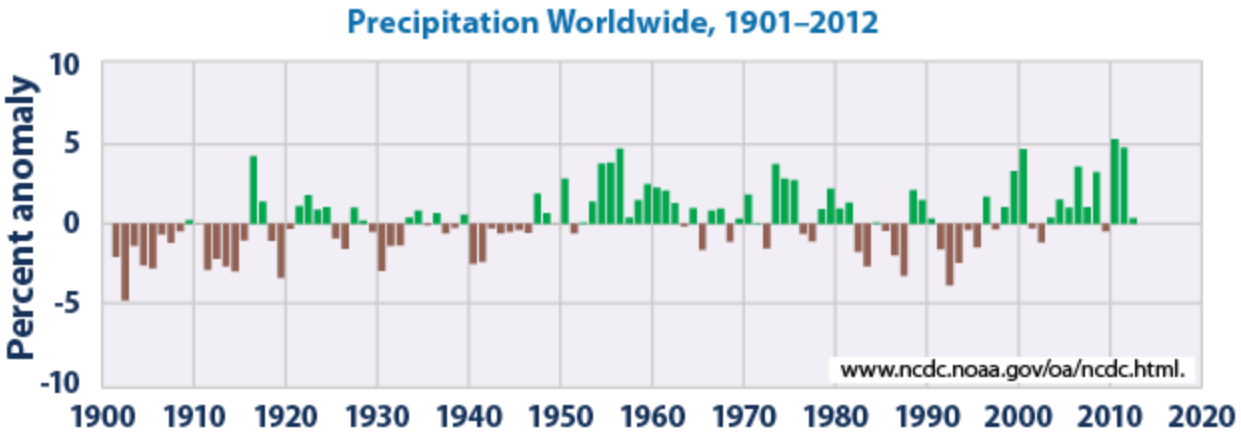


Figure 19.3.4 Global precipitation anomalies compared with the average over the period from 1901 to 2012.

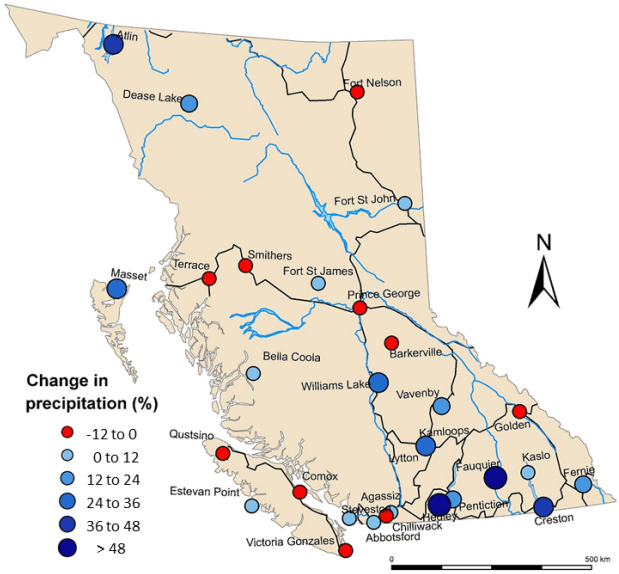


Figure 19.3.5 Change in precipitation amounts over the period 1945 to 2005 for 29 stations in British Columbia.

A similar trend is evident for British Columbia based on weather data from 1945 to 2005 for 29 stations distributed around the province (Figure 19.3.5). Of those stations, 19 show an increase in precipitation and 10 show a decrease. While the decreases are all less than 12%, some of the increases are greater than 48%. Based on the data from these stations, it is estimated that approximately 60 mm/year more precipitation fell on British Columbia in 2005 compared with 1945. That is equivalent to about six months of the average flow of the Fraser River.

While the overall amount of precipitation (total volume of rain plus snow) increased at 19 out of 29 stations between 1945 and 2005, the amount of snowfall decreased at every single station. This is a disturbing trend for many of us, including: operators and users of winter resorts and hydroelectric dams, the Wildfire Management Branch, people who drink water from reservoirs that are replenished by snow, and people who eat food that is grown across western Canada and is irrigated with water derived from melting snow.

Exercise 19.4 Rainfall and ENSO

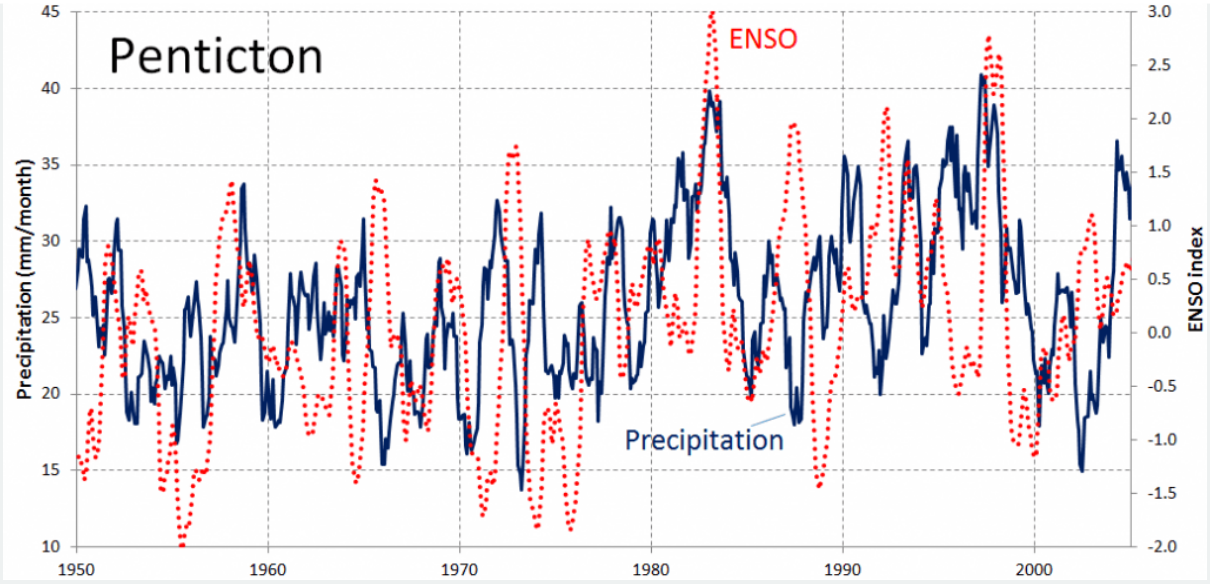


Figure 19.3.6 Monthly precipitation in Pentiction and ENSO index from 1950 to 2005.

Figure 19.3.6 shows the monthly precipitation data for Penticton, BC from 1950 to 2005 (solid line) along with the ENSO (El Niño Southern Oscillation) index values (dotted line). High ENSO index values correspond to strong El Niño events, such as 1983 and 1998. Describe the relationship between ENSO and precipitation in B.C.'s southern interior.

It's not necessarily a consistent relationship.

See Appendix 3 for Exercise 19.4 answers.



Figure 19.3.7 Mountain pine beetle damage in Manning Park, British Columbia.

The geographical ranges of diseases and pests, especially those caused or transmitted by insects, have been shown to extend toward temperate regions because of climate change. West Nile virus and Lyme disease are two examples that already directly affect Canadians, while dengue fever could be an issue in the future. Canadians are also indirectly affected by the increase in populations of pests such as the mountain pine beetle (Figure 19.3.7).

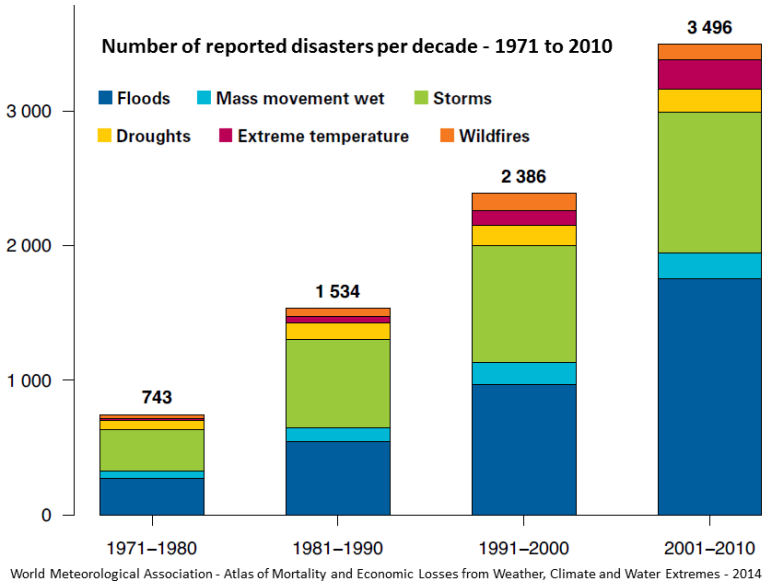


Figure 19.3.8 Numbers of various types of disasters between 1971 and 2010.

A summary of the impacts of climate change on natural disasters is given in Figure 19.3.8. The major types of disasters related to climate are floods and storms, but the health implications of extreme temperatures are also becoming a great concern. In the decade 1971 to 1980, extreme temperatures were the fifth most common natural disasters; by 2001 to 2010, they were the third most common.

For several days in both June and July of 2019, many parts of Europe experienced massive heat waves with all-time national record temperatures set in several countries (Belgium, Finland, France, Germany, Luxembourg, Netherlands, and United Kingdom) (Figure 19.3.9). At the time of writing (August 2019) the death toll from these events is not known. A similar event in Russia in 2010 is estimated to have resulted in over 55,000 deaths.

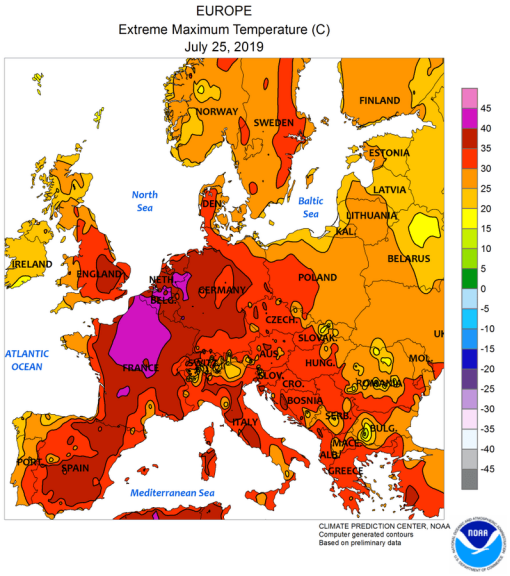


Figure 19.3.9 Maximum temperatures across Europe on July 25th 2019

Exercise 19.5 How can you reduce your impact on the climate?

If you look back to Figure 19.3.3 and the related text, you can easily see what aspects of our way of life are the most responsible for climate change. Think about how you could make changes to your own lifestyle to reduce your impact on the climate. It may depend on where you live, and the degree to which fossil fuels are used to generate the electricity that you use, but it's most likely to include how, how far, how fast, and how frequently you move around.

If you hold the opinion that there isn't much point in making changes to your lifestyle because others won't or because your contribution is only a tiny fraction of the problem, bear in mind that all of us have the opportunity to set an example that others can follow. And remember the words of the American anthropologist Margaret Mead:

Never doubt that a small group of thoughtful, committed citizens can change the world. Indeed, it is the only thing that ever has.

Image Descriptions

Figure 19.3.5 image description: The percent change in precipitation by city in British Columbia from 1945 to 2005

Negative 12% to 0%	0% to 12%	12% to 24%	24% to 36%	26% to 48%	Greater than 48%

- [Return to Figure 19.3.5]
  Figure 19.3.8 image description: From the 1970s, the number of reported disasters by decade has grown steadily. In the 1970s, 743 disasters were reported; in the 1980s, 1,534 were reported; in the 1990s, 2,386 were reported, and in the 2000s, 3,496 were reported. Together, floods and storms make up about three quarters of the reported disasters each decade, followed by mass movement wet, droughts, extreme temperature, and wildfires. [Return to Figure 19.3.8]

Media Attributions

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19.4: Summary

The main topics of this chapter can be summarized as follows:

Section	Summary
19.1 What Makes the Climate Change?	The two components of climate change are forcings and feedbacks. Natural climate forcings, which have operated throughout geological time, include solar evolution and cycles, continental drift, continental collisions and mountain building, volcanism, orbital variations, and ocean current cycles. Feedbacks include melting of ice, snow, and permafrost (changing albedo and releasing GHGs); temperature-related changes to solubility of CO_2; and vegetation growth.
19.2 Anthropogenic Climate Change	The key contributors to anthropogenic climate change are our use of fossil fuels, what we eat and how we produce it, and how much waste we produce and what we do with it.
19.3 Implications of Climate Change	The most reliable indicators of climate change are those that we can detect by looking at records going back for decades. These include temperature and other climate parameters, of course, but also sea-level rise and the incidence of major storms. Some of the implications of climate change include changes to the distribution of disease vectors and pests, and an increase in the incidence and severity of heat waves.

Questions for Review

See Appendix 2 for answers to Review Questions.

1. What property of greenhouse gases allows them to absorb infrared radiation and thus trap heat within the atmosphere?
2. Explain why the emission of CO_2 from fossil fuel use is a climate forcing, while the solubility of CO_2 in seawater is a climate feedback.
3. Explain how the positioning of Gondwana at the South Pole contributed to glaciation during the Paleozoic.
4. Most volcanic eruptions lead to short-term cooling, but long-term sustained volcanism can lead to warming. Describe the mechanisms for these two different consequences.
5. Using the orbital information on eccentricity, tilt, and precession, we could calculate variations in insolation for any latitude on Earth and for any month of the year. Why is it useful to choose the latitude of 65° as opposed to something like 30°? Why north instead of south? Why July instead of January?
6. If the major currents in the oceans were to slow down or stop, how would that affect the distribution of heat on Earth, and what effect might that have on glaciation?
7. Explain the climate implications of the melting and breakdown of permafrost.
8. Much of the warming of the Paleocene-Eocene thermal maximum is thought to have been caused by the release of CH_4 from sea-floor methane hydrates. Describe what must have happened before this could take place.
9. Burning fossil fuels emits CO_2 to the atmosphere via reactions like this one: $CH_4 + O_2 \longrightarrow CO_2 + 2H_2O$. Describe some of the other ways that our extraction, transportation, and use of fossil fuels impact the climate.
10. Explain why, even if we could stop our impact on the climate tomorrow, we would still be facing between 1 and 2 m of additional sea-level rise.
11. Use the Internet to research West Nile virus, and explain why its spread into Canada from the United States is related to climate change.

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20: GEOLOGICAL RESOURCES

20.1: Metal Deposits
20.2: Industrial Materials
20.3: Fossil Fuels
20.4: Diamonds
20.5: Summary

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20.1: Metal Deposits

Mining has always been a major part of Canada's economy. Canada has some of the largest mining districts and deposits in the world, and for the past 150 years, we have been one of the world's most important suppliers of metals. Extraction of Earth's resources goes back a long way in Canada. For example, the First Nations of British Columbia extracted obsidian from volcanic regions for tools and traded it up and down the coast. In the 1850s, gold was discovered in central British Columbia, and in the 1890s, even more gold was discovered in the Klondike area of Yukon. These two events were critical to the early development of British Columbia, Yukon, and Alaska.

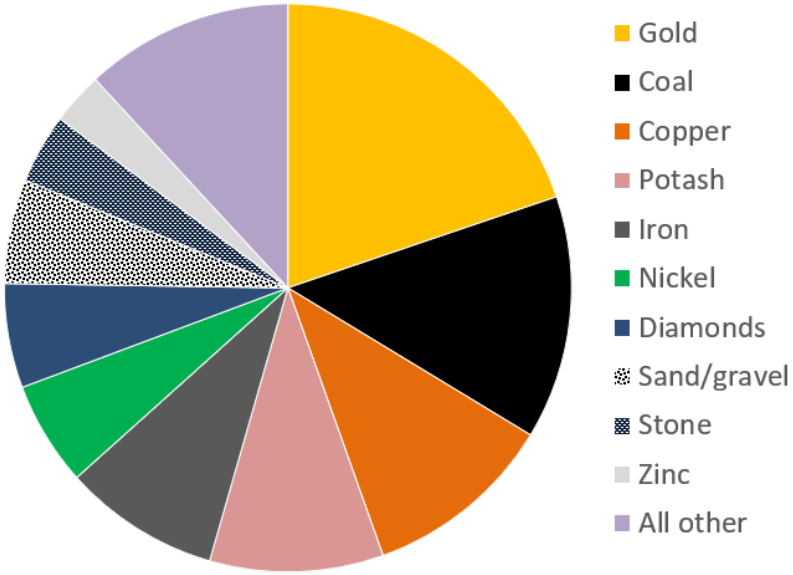


Figure 20.1.1 The value of various Canadian mining sectors in 2017. The total value of these products was about $44 billion in 2017.

Canada's mining sector had revenues in the order of $44 billion in 2017 (Figure 20.1.1). The 4 most valuable commodities were gold, coal, copper and potash, with important amounts from iron, nickel, diamonds, sand and gravel aggregates, stone and zinc. Revenues from the petroleum sector are significantly higher, at over $100 billion per year.

A metal deposit is a body of rock in which one or more metals have been concentrated to the point of being economically viable for recovery. Some **background** levels of important metals in average rocks are shown on Table 20.1, along with the typical grades necessary to make a viable deposit, and the corresponding concentration factors. Looking at copper, for example, we can see that while average rock has around 40 ppm (parts per million) of copper, a grade of around 10,000 ppm or 1% is necessary to make a viable copper deposit. In other words, copper ore has about 250 times as much copper as typical rock. For the other elements in the list, the concentration factors are much higher. For gold, it's 2,000 times and for silver it's around 10,000 times.

Table 20.1 Typical background and ore levels of some important metals.[1]

Metal	Typical Background Level	Typical Economic Grade*	Concentration Factor
Copper	40 ppm	10,000 ppm (1%)	250 times
Gold	0.003 ppm	6 ppm (0.006%)	2,000 times
Lead	10 ppm	50,000 ppm (5%)	5,000 times
Molybdenum	1 ppm	1,000 ppm (0.1%)	1,000 times
Nickel	25 ppm	20,000 ppm (2%)	800 times
Silver	0.1 ppm	1,000 ppm (0.1%)	10,000 times
Uranium	2 ppm	10,000 ppm (1%)	5,000 times
Zinc	50 ppm	50,000 ppm (5%)	1,000 times

Metal	Typical Background Level	Typical Economic Grade*	Concentration Factor
*It's important to note that the economic viability of any deposit depends on a wide range of factors including its grade, size, shape, depth below the surface, and proximity to infrastructure, the current price of the metal, the labour and environmental regulations in the area, and many other factors.			

It is clear that some very significant concentration must take place to form a mineable deposit. This concentration may occur during the formation of the host rock, or after the rock forms, through a number of different types of processes. There is a very wide variety of ore-forming processes, and there are hundreds of types of mineral deposits. The origins of a few of them are described below.

Magmatic Nickel Deposits



Figure 20.1.2 The nickel smelter at Thompson, Manitoba.

A magmatic deposit is one in which the metal concentration takes place primarily at the same time as the formation and emplacement of the magma. Most of the nickel mined in Canada comes from magmatic deposits such as those in Sudbury (Ontario), Thompson (Manitoba) (Figure 20.1.2), and Voisey's Bay (Labrador). The magmas from which these deposits form are of mafic or ultramafic composition (derived from the mantle), and therefore they had relatively high nickel and copper contents to begin with (as much as 100 times more than normal rocks in the case of nickel). These elements may be further concentrated within the magma as a result of the addition of sulfur from partial melting of the surrounding rocks. The heavy nickel and copper sulfide minerals are then concentrated further still by gravity segregation (i.e., crystals settling toward the bottom of the magma chamber). In some cases, there are also significant concentrations of platinum-bearing minerals in magmatic deposits.

Most of these types of deposits around the world are Precambrian in age; the mantle was significantly hotter at that time, and the necessary mafic and ultramafic magmas were more likely to be emplaced in the continental crust.

Volcanogenic Massive Sulphide Deposits

Much of the copper, zinc, lead, silver, and gold mined in Canada is mined from **volcanic-hosted massive sulfide** (VHMS) deposits associated with submarine volcanism (VMS deposits). Examples are the deposits at Kidd Creek, Ontario, Flin Flon on the Manitoba-Saskatchewan border, Britannia on Howe Sound, and Myra Falls (within Strathcona Park) on Vancouver Island.

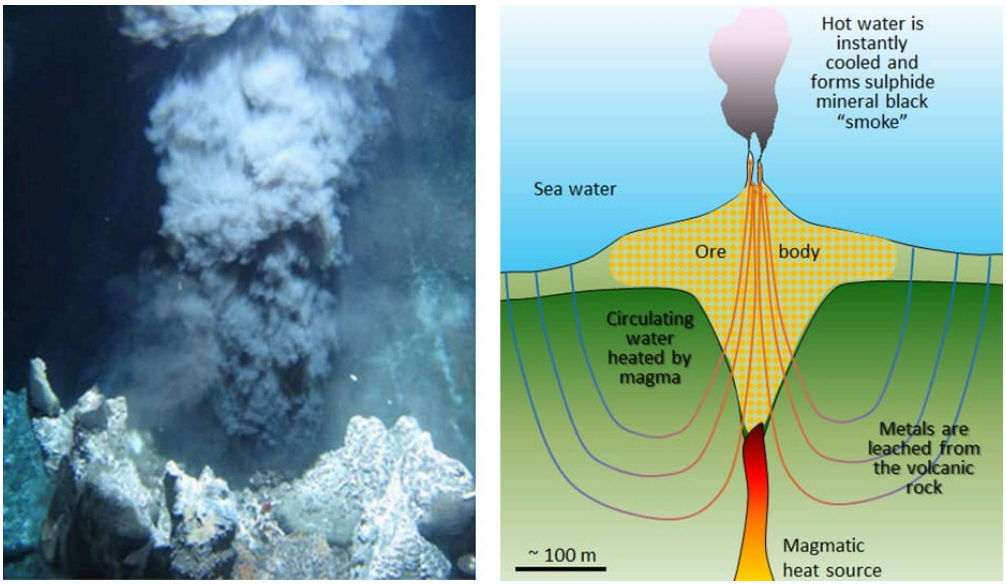


Figure 20.1.3 Left: A black smoker on the Juan de Fuca Ridge off the west coast of Vancouver Island. Right: A model of the formation of a volcanogenic massive sulfide deposit on the sea floor.

VMS deposits are formed from the water discharged at high temperature (250° to 300°C) at ocean-floor hydrothermal vents, primarily in areas of subduction-zone volcanism. The environment is comparable to that of modern-day black smokers (Figure 20.1.3), which form where hot metal- and sulfide-rich water issues from the sea floor. They are called massive sulfide deposits because the sulfide minerals (including pyrite (FeS2) , sphalerite (ZnS), chalcopyrite ($CuFeS_2$), and galena (PbS)) are generally present in very high concentrations (making up the majority of the rock in some cases). The metals and the sulfur are leached out of the sea-floor rocks by convecting groundwater driven by the volcanic heat, and then quickly precipitated where that hot water enters the cold seawater, causing it to cool suddenly and change chemically. The volcanic rock that hosts the deposits is formed in the same area and at the same general time as the accumulation of the ore minerals.

Porphyry Deposits

Porphyry deposits are the most important source of copper and molybdenum in British Columbia, the western United States, and Central and South America. Most porphyry deposits also host some gold, and in rare cases gold is the primary commodity. B.C. examples include several large deposits within the Highland Valley mine (Figure 20.1.1) and numerous other deposits scattered around the central part of the province.

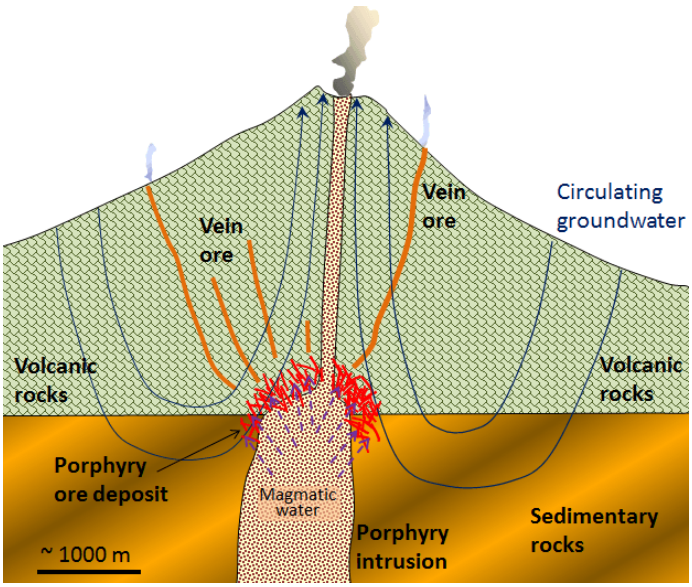


Figure 20.1.4 A model for the formation of a porphyry deposit around an upper-crustal porphyritic stock and associated vein deposits.

A porphyry deposit forms around a cooling felsic stock in the upper part of the crust. They are called “porphyry” because upper crustal stocks are typically porphyritic in texture, the result of a two-stage cooling process. Metal enrichment results in part from convection of groundwater related to the heat of the stock, and also from metal-rich hot water expelled by the cooling magma (Figure 20.1.4). The host rocks, which commonly include the stock itself and the surrounding country rocks, are normally highly fractured and brecciated. During the ore-forming process, some of the original minerals in these rocks are altered to potassium feldspar, biotite, epidote, and various clay minerals. The important ore minerals include chalcopyrite ($CuFeS_2$), bornite (Cu_5FeS_4), and pyrite in copper porphyry deposits, or molybdenite (MoS_2) and pyrite in molybdenum porphyry deposits. Gold is present as minute flakes of native gold.

This type of environment (i.e., around and above an intrusive body) is also favourable for the formation of other types of deposits—particularly vein-type gold deposits (a.k.a. **epithermal deposits**). Some of the gold deposits of British Columbia (such as in the Eskay Creek area adjacent to the Alaska panhandle), and many of the other gold deposits situated along the western edge of both South and North America are of the vein type shown in Figure 20.1.4, and are related to nearby magma bodies.

Banded Iron Formation



Figure 20.1.5 Banded iron formation from an unknown location in North America on display at a museum in Germany. The rock is about 2 m across. The dark grey layers are magnetite and the red layers are hematite. Chert is also present.

Most of the world’s major iron deposits are of the **banded iron formation** type, and most of these formed during the initial oxygenation of Earth’s atmosphere between 2,400 and 1,800 Ma. At that time, iron that was present in dissolved form in the ocean (as Fe2+) became oxidized to its insoluble form (Fe3+) and accumulated on the sea floor, mostly as hematite interbedded with chert (Figure 20.1.5). Unlike many other metals, which are economically viable at grades of around 1% or even much less, iron deposits are only viable if the grades are in the order of 50% iron and if they are very large.

Unconformity-Type Uranium Deposits

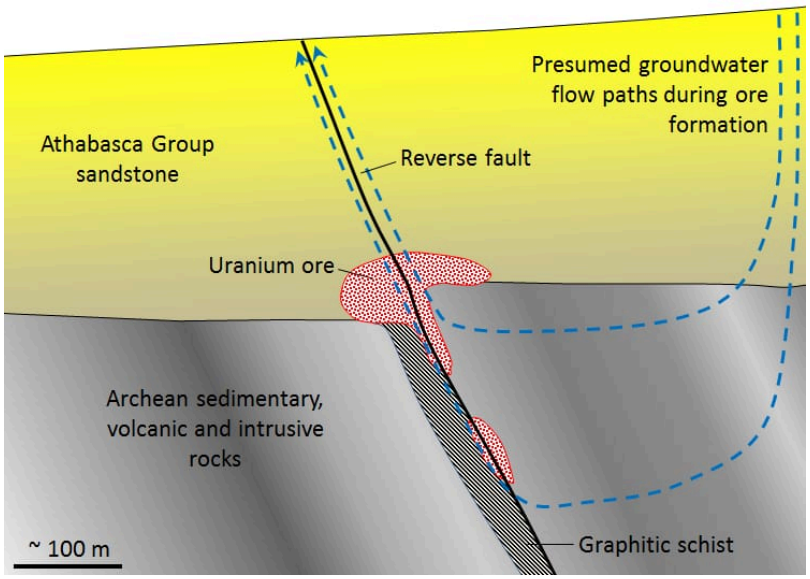


Figure 20.1.6 Model of the formation of unconformity-type uranium deposits of the Athabasca Basin, Saskatchewan.

There are several different types of uranium deposits, but some of the largest and richest are those within the Athabasca Basin of northern Saskatchewan. These are called **unconformity-type uranium deposits** because they are all situated very close to the unconformity between the Proterozoic Athabasca Group sandstone and the much older Archean sedimentary, volcanic, and intrusive igneous rock (Figure 20.1.6). The origin of unconformity-type U deposits is not perfectly understood, but it is thought that two particular features are important: (1) the relative permeability of the Athabasca Group sandstone, and (2) the presence of graphitic schist within the underlying Archean rocks. The permeability of the sandstone allowed groundwater to flow through it and leach out small amounts of U, which stayed in solution in the oxidized form U^{6+}. The graphite (C) created a reducing (non-oxidizing) environment that converted the uranium from U^{6+} to insoluble U^{4+}, at which point it was precipitated as the mineral uraninite (UO_2).

Exercise 20.2 The importance of heat and heat engines

For a variety of reasons, thermal energy (heat) from within Earth is critical in the formation of many types of ore deposits. Look back through the deposit-type descriptions above and complete the following table, describing which of those deposit types depend on a source of heat for their formation, and why.

Which deposit types depend on heat for formation?

Deposit Type	Is heat a factor?	If so, what is the role of the heat?
Magmatic		
Volcanogenic massive sulfide		
Porphyry		
Banded iron formation		
Unconformity-type uranium		

See Appendix 3 for Exercise 20.2 answers.

Mining and Mineral Processing



Figure 20.1.7 Underground at the Myra Falls Mine, Vancouver Island.

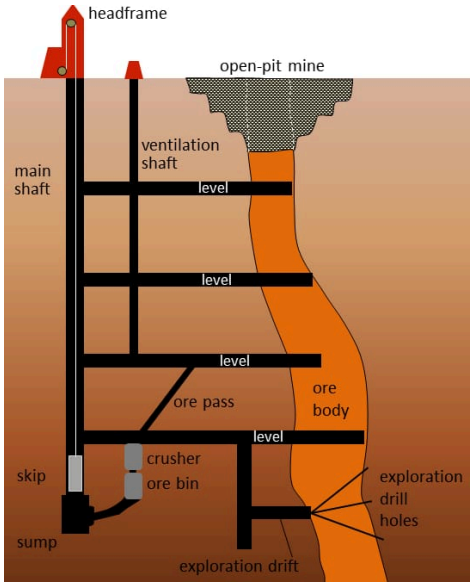


Figure 20.1.8 Schematic cross-section of a typical underground mine.

Metal deposits are mined in a variety of different ways depending on their depth, shape, size, and grade. Relatively large deposits that are quite close to the surface and somewhat regular in shape are mined using **open-pit mine** methods (Figure 20.1.1). Creating a giant hole in the ground is generally cheaper than making an underground mine, but it is also less precise, so it is necessary to mine a lot of waste rock along with the ore. Relatively deep deposits or those with elongated or irregular shapes are typically mined from underground with deep vertical **shafts**, **declines** (sloped tunnels), and **levels** (horizontal tunnels) (Figures 20.1.7 and 20.1.9). In this way, it is possible to focus the mining on the ore body itself. However, with relatively large ore bodies, it may be necessary to leave some pillars to hold up the roof.

In many cases, the near-surface part of an ore body is mined with an open pit, while the deeper parts are mined underground (Figures 20.1.8 and 20.1.9).



Figure 20.1.9 Entrance to an exploratory decline (white arrow) for the New Afton Mine situated in the side of the open pit of the old Afton Mine, near Kamloops, B.C.

A typical metal deposit might contain a few percent of ore minerals (e.g., chalcopyrite or sphalerite), mixed with the minerals of the original rock (e.g., quartz or feldspar). Other sulfide minerals are commonly present within the ore, especially pyrite.

When ore is processed (typically very close to the mine), it is ground to a fine powder and the ore minerals are physically separated from the rest of the rock to make a **concentrate**. At a molybdenum mine, for example, this concentrate may be almost pure molybdenite (MoS_2). The rest of the rock is known as **tailings**. It comes out of the concentrator as a wet slurry and must be stored near the mine, in most cases, in a tailings pond.

The tailings pond at the Myra Falls Mine on Vancouver Island and the settling ponds for waste water from the concentrator are shown in Figure 20.1.10. The tailings are contained by an embankment. Also visible in the foreground is a pile of waste rock, which is non-ore rock that was mined in order to access the ore. Although this waste rock contains little or no ore minerals, at many mines it contains up to a few percent pyrite. The tailings and the waste rock at most mines are an environmental liability because they contain pyrite plus small amounts of ore minerals. When pyrite is exposed to oxygen and water, it generates sulfuric acid—also known as **acid rock drainage** (ARD). Acidity itself is a problem to the environment, but because the ore elements, such as copper or lead, are more soluble in acidic water than neutral water, ARD is also typically quite rich in metals, many of which are toxic.



Figure 20.1.10 The tailings pond at the Myra Falls Mine on Vancouver Island. The dry rock in the middle of the image is waste rock. Myra Creek flows between the tailings pond and the headframe. The settling ponds (right) are used for processing water from the concentrator.

Tailings ponds and waste-rock storage piles must be carefully maintained to ensure their integrity and monitored to ensure that

acidic and metal-rich water is not leaking out. In August 2014, the tailings pond at the Mt. Polley Mine in central B.C. failed and 10 million cubic meters of waste water along with 4.5 million cubic meters of tailings slurry was released into Polley Lake, Hazeltine Creek, and Quesnel Lake (Figure 20.1.11, a and b). As of 2019, the environmental implications of this event are still not fully understood.



Figure 20.1.11 The Mt. Polley Mine area before and after to the dam breach of August 2014. The tailings were stored in the area labelled "retention basin." The retention dam failed and water and tailings released flowed into Polley Lake, Hazeltine Creek, and Quesnel Lakes.

Most mines have concentrators on site because it is relatively simple to separate ore minerals from non-ore minerals and this significantly reduces the costs and other implications of transportation. But separation of ore minerals is only the preliminary stage of metal refinement, for most metals the second stage involves separating the actual elements within the ore minerals. For example, the most common ore of copper is chalcopyrite ($CuFeS_2$). The copper needs to be separated from the iron and sulfur to make copper metal and that involves complicated and very energy-intensive processes that are done at **smelters** or other types of refineries. Because of their cost and the economies of scale, there are far fewer refineries than there are mines.

There are several metal refineries (including smelters) in Canada; some examples are the aluminum refinery in Kitimat, B.C. (which uses ore from overseas); the lead-zinc smelter in Trail, B.C.; the nickel smelter at Thompson, Manitoba; numerous steel smelters in Ontario, along with several other refining operations for nickel, copper, zinc, and uranium; aluminum refineries in Quebec; and a lead smelter in New Brunswick.

Image Descriptions

Figure 20.1.1 image description: The value of the different mining sectors in Canada.

Mining Sector	Value (in billions)
Total value	$36.7
Potash	$6.1
Gold	$5.9
Iron	$5.3
Copper	$4.6
Nickel	$3.5
Diamonds	$2.0
Sand and gravel	$1.7
Cement	$1.6
Stone	$1.5
Zinc	$0.8
Uranium	$0.7

Mining Sector	Value (in billions)
Salt	$0.6
Other metals	$2.4

[Return to Figure 20.1.1]

Figure 20.1.8 image description: An open-pit mine is dug to access the ore that is near the surface. For ore farther down, an underground mine will be constructed to access the ore. This diagram shows the main shaft (a large vertical tunnel) with four levels (horizontal tunnels) connected to it. The levels run from the main shaft into the ore body. A ventilation shaft runs up through the four levels in between the main shaft and the ore for air circulation.

[Return to Figure 20.1.8]

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20.2: Industrial Materials

Metals are critical for our technological age, but there are a lot of other not-so-shiny materials that are needed to facilitate our way of life. For everything made out of concrete or asphalt, we need sand and gravel. To make the cement that holds concrete together, we also need limestone. For the glass in our computer screens and for glass-sided buildings, we need silica sand plus sodium oxide (Na_2O), sodium carbonate (Na_2CO_3), and calcium oxide (CaO). Potassium is an essential nutrient for farming in many areas, and for a wide range of applications (e.g., ceramics and many industrial processes), we also need various types of clay.

The best types of **aggregate** (sand and gravel) resources are those that have been sorted by streams, and in Canada the most abundant and accessible fluvial deposits are associated with glaciation. That doesn't include till of course, because it has too much silt and clay, but it does include glaciofluvial outwash, which is present in thick deposits in many parts of the country, similar to the one shown in Figure 20.2.2. In a typical gravel pit, these materials are graded on-site according to size and then used in a wide range of applications from constructing huge concrete dams to filling children's sandboxes. Sand is also used to make glass, but for most types of glass, it has to be at least 95% quartz (which the sandy layers shown in Figure 20.2.1 are definitely not), and for high-purity glass and the silicon wafers used for electronics, the source sand has to be over 98% quartz.



Figure 20.2.1 Sand and gravel in an aggregate pit near Nanaimo, BC.

Approximately 80 million tonnes of concrete are used in Canada each year—a little over 2 tonnes per person. The cement used for concrete is made from approximately 80% calcite ($CaCO_3$) and 20% clay. This mixture is heated to 1450°C to produce the required calcium silicate compounds (e.g., Ca_2SiO_4). The calcite typically comes from limestone quarries like the one on Texada Island, B.C. (Figure 20.2.2). Limestone is also used as the source material for many other products that require calcium compounds, including steel and glass, pulp and paper, and plaster products for construction.



Figure 20.2.2 Triassic Quatsino Formation limestone being quarried on Texada Island, B.C.

Sodium is required for a wide range of industrial processes, and the most convenient source is sodium chloride (rock salt), which is mined from evaporite beds in various parts of Canada. The largest salt mine in the world is at Goderich, Ontario, where salt is recovered from the 100 m thick Silurian Salina Formation. The same formation is mined in the Windsor area. Rock salt is also used as a source of sodium and chlorine in the chemical industry to melt ice on roads, as part of the process of softening water, and as a seasoning. Under certain conditions, the mineral sylvite (KCl) accumulates in evaporite beds, and this rock is called potash. This happened across the Canadian prairies during the Devonian, creating the Prairie evaporite formation (Figure 20.2.8). Potassium is used as a crop fertilizer, and Canada is the world's leading supplier, with most of that production coming from Saskatchewan.



Figure 20.2.3 Slate used as a facing material on a concrete building column in Vancouver.

Another evaporite mineral, gypsum ($CaSO_4.2H_2O$), is the main component of plasterboard (drywall) that is widely used in the construction industry. One of the main mining areas for gypsum in Canada is in the Milford Station area of Nova Scotia, site of the world's largest gypsum mine.

Rocks are quarried or mined for many different uses, such as building facades (Figure 20.2.3), countertops, stone floors, and headstones. In most of these cases, the favoured rock types are granitic rocks, slate, and marble. Quarried rock is also used in some applications where rounded gravel isn't suitable, such as the ballast (road bed) for railways, where crushed angular rock is needed.

Exercise 20.3 Sources of important lighter metals

When we think of the manufacture of consumer products, plastics and the heavy metals (copper, iron, lead, zinc) easily come to mind, but we often forget about some of the lighter metals and non-metals that are important. Consider the following elements and determine their sources. Answers for all of these except magnesium are given above. See if you can figure out a likely mineral source of magnesium.

What are the sources of the following elements?

Element	Source(s)
Silicon	
Calcium	
Sodium	
Potassium	
Magnesium	

See Appendix 3 for Exercise 20.3 answers.

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20.3: Fossil Fuels

There are numerous types of fossil fuels, but all of them involve the storage of organic matter in sediments or sedimentary rocks. Fossil fuels are rich in carbon and almost all of that carbon ultimately originates from CO_2 taken out of the atmosphere during photosynthesis. That process, driven by solar energy, involves reduction (the opposite of oxidation) of the carbon, resulting in it being combined with hydrogen instead of oxygen. The resulting organic matter is made up of complex and varied carbohydrate molecules.

Most organic matter is oxidized back to CO_2 relatively quickly (within weeks or years in most cases), but any of it that gets isolated from the oxygen of the atmosphere (for example, deep in the ocean or in a stagnant bog) may last long enough to be buried by sediments and, if so, may be preserved for tens to hundreds of millions of years. Under natural conditions, that means it will be stored until those rocks are eventually exposed at the surface and weathered.

In this section, we'll discuss the origins and extraction of the important fossils fuels, including coal, oil, and gas. Coal, the first fossil fuel to be widely used, forms mostly on land in swampy areas adjacent to rivers and deltas in areas with humid tropical to temperate climates. The vigorous growth of vegetation leads to an abundance of organic matter that accumulates within stagnant water, and thus does not decay and oxidize. This situation, where the dead organic matter is submerged in oxygen-poor water, must be maintained for centuries to millennia in order for enough material to accumulate to form a thick layer (Figure 20.3.1a). At some point, the swamp deposit is covered with more sediment — typically because a river changes its course or sea level rises (Figure 20.3.1b). As more sediments are added, the organic matter starts to become compressed and heated. Low-grade **lignite** coal forms at depths between a few 100 m and 1,500 m and temperatures up to about 50°C (Figure 20.3.1c). At between 1,000 m to 5,000 m depth and temperatures up to 150°C m, **bituminous** coal forms (Figure 20.3.1d). At depths beyond 5,000 m and temperatures over 150°C, **anthracite** coal forms.

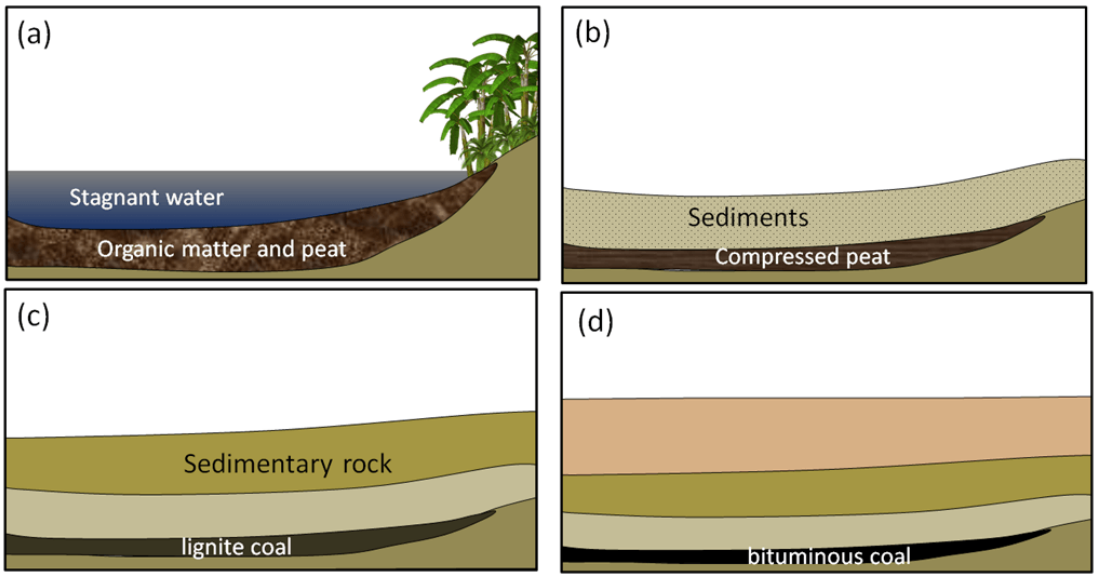


Figure 20.3.1 Formation of coal: (a) accumulation of organic matter within a swampy area; (b) the organic matter is covered and compressed by deposition of a new layer of clastic sediments; (c) with greater burial, lignite coal forms; and (d) at even greater depths, bituminous and eventually anthracite coal form.

There are significant coal deposits in many parts of Canada, including the Maritimes, Ontario, Saskatchewan, Alberta, and British Columbia. In Alberta and Saskatchewan, much of the coal is used for electricity generation. Coal from the Highvale Mine (Figure 20.3.2), Canada's largest, is used to feed the Sundance and Keephills power stations west of Edmonton. Almost all of the coal mined in British Columbia is exported for use in manufacturing steel.



Figure 20.3.2 The Highvale Mine (background) and the Sundance (right) and Keephills (left) generating stations on the southern shore of Wabamun Lake, Alberta.

While almost all coal forms on land from terrestrial vegetation, most oil and gas is derived primarily from marine micro-organisms that accumulate within sea-floor sediments. In areas where marine productivity is high, dead organic matter is delivered to the sea floor fast enough that some of it escapes oxidation. This material accumulates in the muddy sediments, which become buried to significant depth beneath other sediments.

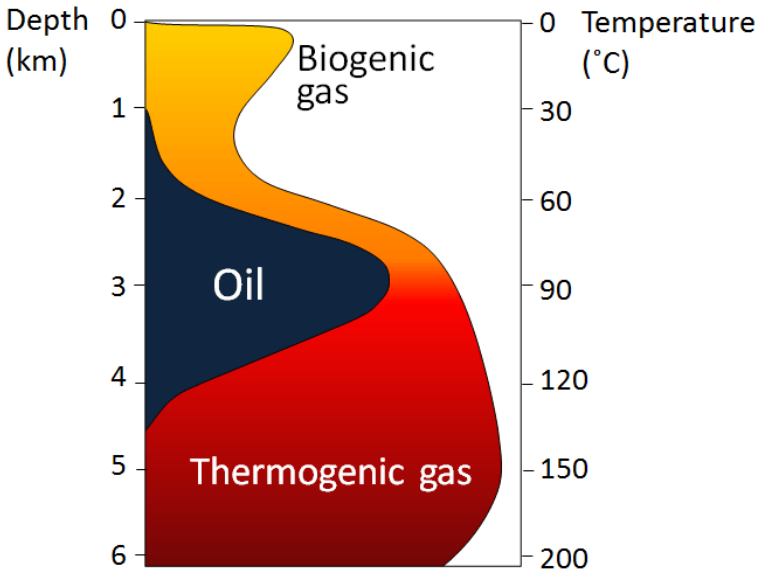


Figure 20.3.3 The depth and temperature limits for biogenic gas, oil, and thermogenic gas.

As the depth of burial increases, so does the temperature—due to the geothermal gradient—and gradually the organic matter within the sediments is converted to hydrocarbons (Figure 20.3.3). The first stage is the biological production (involving anaerobic bacteria) of methane. Most of this escapes back to the surface, but some is trapped in methane hydrates near the sea floor. At depths beyond about 2 km, and at temperatures ranging from 60° to 120°C, the organic matter is converted by chemical processes to oil. This depth and temperature range is known as the **oil window**. Beyond 120°C most of the organic matter is chemically converted to methane.

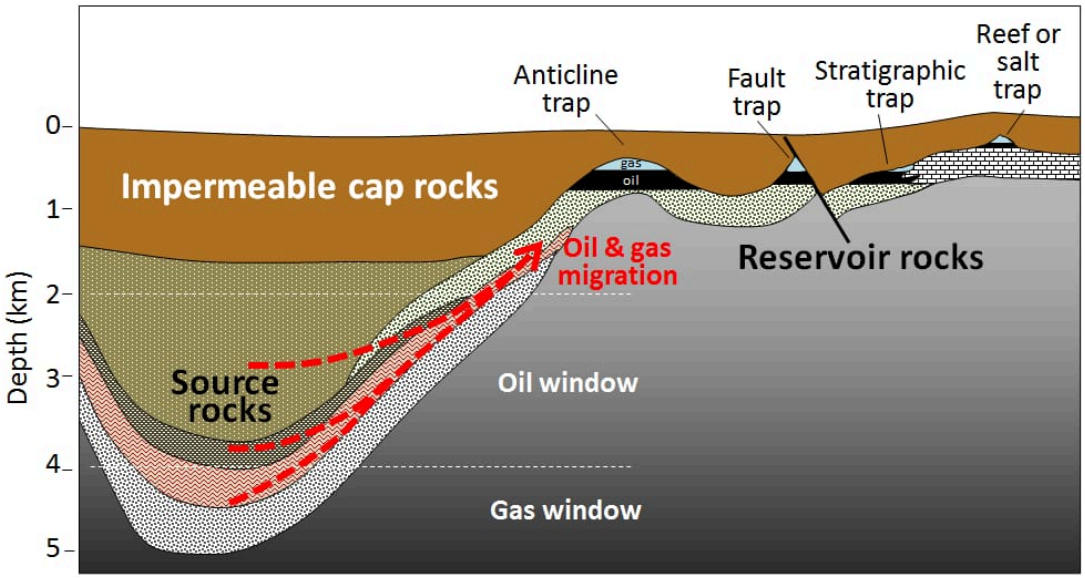


Figure 20.3.4 Migration of oil and gas from source rocks into traps in reservoir rocks.

The organic matter-bearing rock within which the formation of gas and oil takes place is known to petroleum geologists as the **source rock**. Both liquid oil and gaseous methane are lighter than water, so as liquids and gases form, they tend to move slowly toward the surface, out of the source rock and into **reservoir rocks**. Reservoir rocks are typically relatively permeable because that allows migration of the fluids from the source rocks, and also facilitates recovery of the oil or gas. In some cases, the liquids and gases make it all the way to the surface, where they are oxidized, and the carbon is returned to the atmosphere. But in other cases, they are contained by overlying impermeable rocks (e.g., mudrock) in situations where anticlines, faults, stratigraphy changes, and reefs or salt domes create traps (Figure 20.3.4).

The liquids and gases that are trapped within reservoirs become separated into layers based on their density, with gas rising to the top, oil below it, and water underneath the oil. The proportions of oil and gas depend primarily on the temperature in the source rocks. Some petroleum fields, such as many of those in Alberta, are dominated by oil, while others, notably those in northeastern B.C., are dominated by gas.

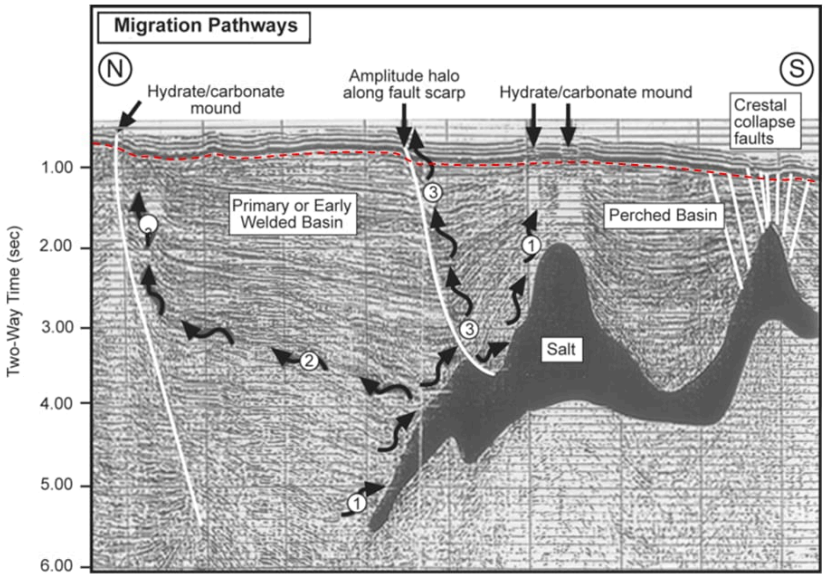


Figure 20.3.5 Seismic section through the East Breaks Field in the Gulf of Mexico. The dashed red line marks the approximate boundary between deformed rocks and younger undeformed rocks. The wiggly arrows are interpreted migration paths. The total thickness of this section is approximately 5 km.

In general, petroleum fields are not visible from the surface, and their discovery involves the search for structures in the subsurface that have the potential to form traps. Seismic surveys are the most commonly used tool for early-stage petroleum

exploration, as they can reveal important information about the stratigraphy and structural geology of subsurface sedimentary rocks. An example from the Gulf of Mexico south of Texas is shown in Figure 20.3.5. In this area, a thick evaporite deposit ("salt") has formed domes because salt is lighter than other sediments and tends to rise slowly toward the surface; this has created traps. The sequence of deformed rocks is capped with a layer of undeformed rock.

Exercise 20.4 Interpreting a Seismic Profile

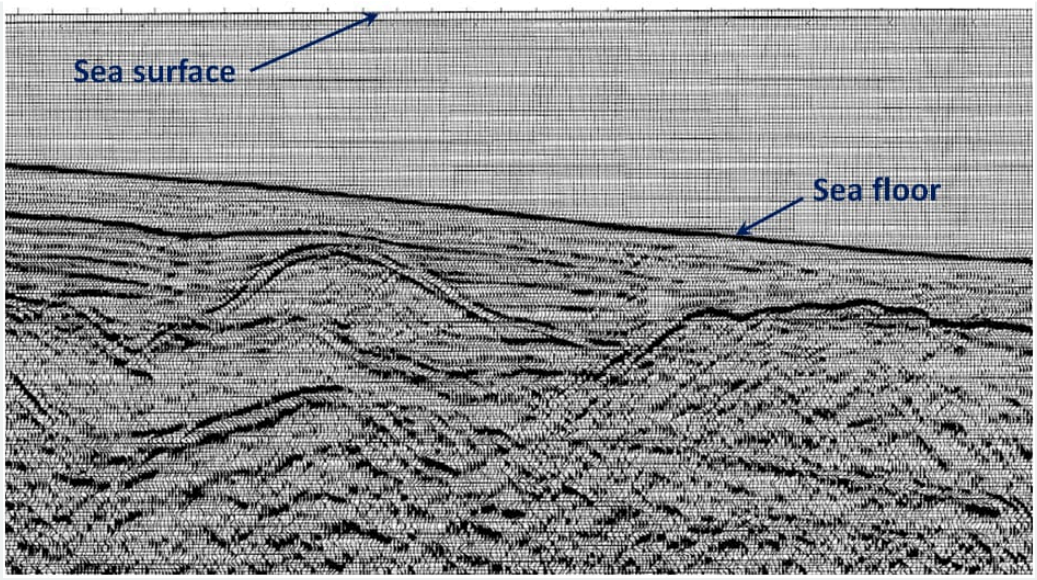


Figure 20.3.6

The cross-section shown in Figure 20.3.6 is from a ship-borne seismic survey in the Bering Sea off the west coast of Alaska. As a petroleum geologist, it's your job to pick two separate locations to drill for oil or gas. Which locations would you choose?

See Appendix 3 for Exercise 20.4 answers.

The type of oil and gas reservoirs illustrated in Figures 20.3.4 and 20.3.5 are described as conventional reserves. Some unconventional types of oil and gas include oil sands, shale gas, and coal-bed methane.

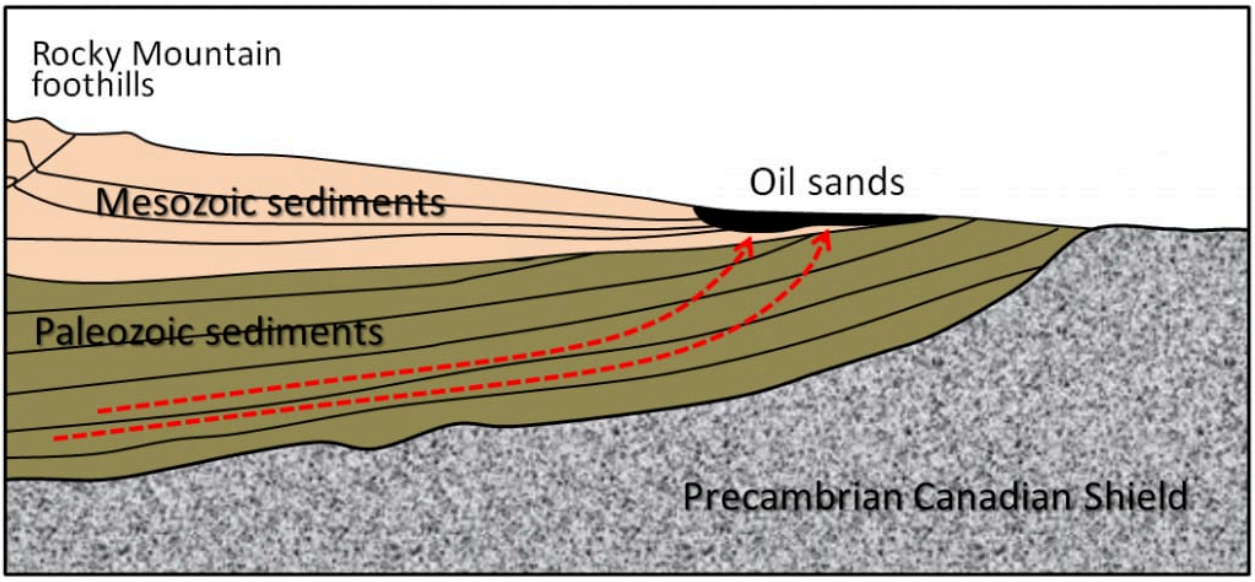


Figure 20.3.7 Schematic cross-section of northern Alberta showing the source rocks and location of the Athabasca Oil Sands.

Oil sands are important because the reserves in Alberta are extremely large (the largest single reserve of oil in the world), but

they are very controversial from an environmental and social perspective. They are “unconventional” because the oil is exposed near the surface and is highly viscous because of microbial changes that have taken place at the surface. The hydrocarbons that form this reserve originated in deeply buried Paleozoic rocks adjacent to the Rocky Mountains and migrated up and toward the east (Figure 20.3.7).

The oil sands are controversial primarily because of the environmental cost of their extraction. Since the oil is so viscous, it requires heat to make it sufficiently liquid to process. This energy comes from gas; approximately 25 m^3 of gas is used to produce 0.16 m^3 (one barrel) of oil. (That’s not quite as bad as it sounds, as the energy equivalent of the required gas is about 20% of the energy embodied in the produced oil.) The other environmental cost of oil sands production is the devastation of vast areas of land where strip-mining is taking place and tailings ponds are constructed, and the unavoidable release of contaminants into the groundwater and rivers of the region.

At present, most oil recovery from oil sands is achieved by mining the sand and processing it on site. Exploitation of oil sand that is not exposed at the surface depends on in situ processes, an example being the injection of steam into the oil-sand layer to reduce the viscosity of the oil so that it can be pumped to the surface.

Shale gas is gas that is trapped within rock that is too impermeable for the gas to escape under normal conditions, and it can only be extracted by fracturing the reservoir rock using water and chemicals under extremely high pressure. This procedure is known as **hydraulic fracturing** or “**fracking**.” Fracking is controversial because of the volume of water used, and because, in some jurisdictions, the fracking companies are not required to disclose the nature of the chemicals used. Although fracking is typically done at significant depths, there is always the risk that overlying water-supply aquifers could be contaminated (Figure 20.3.8). Fracking also induces low-level seismicity.

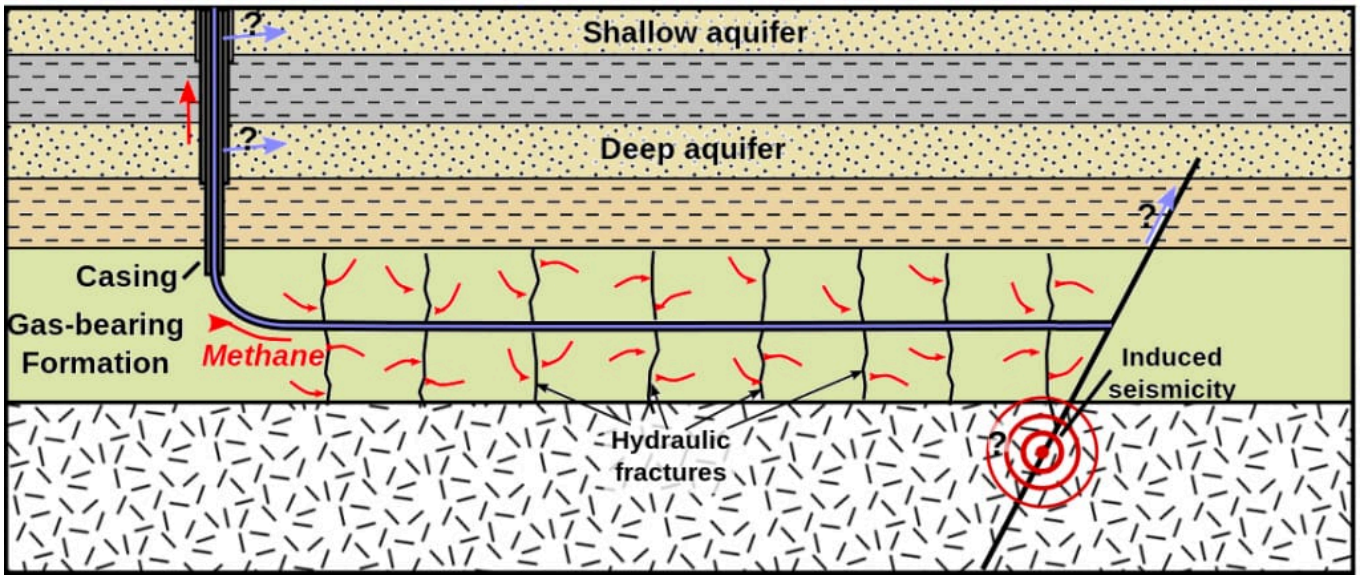


Figure 20.3.8 Depiction of the process of directional drilling and fracking to recover gas from impermeable rocks. The light blue arrows represent the potential for release of fracking chemicals to aquifers.

During the process that converts organic matter to coal, some methane is produced, which is stored within the pores of the coal. When coal is mined, methane is released into the mine where it can become a serious explosion hazard. Modern coal-mining machines have methane detectors on them and actually stop operating if the methane levels are dangerous. It is possible to extract the methane from coal beds without mining the coal; gas recovered this way is known as **coal-bed methane**.

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20.4: Diamonds

Although Canada's diamond mining industry didn't get started until 1998, diamonds are currently the seventh most valuable product mined in the country (Figure 20.4.1), and Canada ranks third in the world in diamond production. Diamonds form deep in the mantle (approximately 200 km to 250 km depth) under very specific pressure and temperature conditions, from carbon that is naturally present in mantle rock (not from coal!). The diamond-bearing rock is brought to the surface coincidentally via a type of volcanism that is extremely rare (the most recent kimberlite eruption is thought to have taken place 10,000 years ago and prior to that at around 30 Ma). There is more on the volcanology of kimberlites in section 4.3. All of the world's kimberlite diamond deposits are situated within ancient shield areas (**cratons**) in Africa, Australia, Russia, South America, and North America.

It has long been known that diamonds could exist within the Canadian Shield, but up until 1991, exploration efforts had been unsuccessful. In 1980 two geologists, Chuck Fipke and Stu Blusson, started searching in the Northwest Territories by sampling glacial sediments looking for some of the minerals that are normally quite abundant within **kimberlites**: chromium-bearing garnet, chromium-bearing pyroxene, chromite (Cr_2O_3), and ilmenite ($FeTiO_3$). These distinctive minerals are used for this type of exploration because they are many times more abundant in kimberlite than diamond is. After more than a decade of exploration, Fipke and Blusson finally focused their search on an area 250 km northeast of Yellowknife, and, in 1991, they announced the discovery of a diamond-bearing kimberlite body at Lac de Gras. That discovery is now the Diavik Mine. The Ekati mine is situated 25 km to the northwest (Figure 20.4.1). There are two separate mines at Diavik accessing three different kimberlite bodies, and there are five at Ekati. See Figure 20.4.16 for a close-up view of the Ekati Mine. As of 2019 there are seven operating diamond mines in Canada, four in the Northwest Territories (including Diavik and Ekati), and one each in Nunavut, Ontario and Quebec.

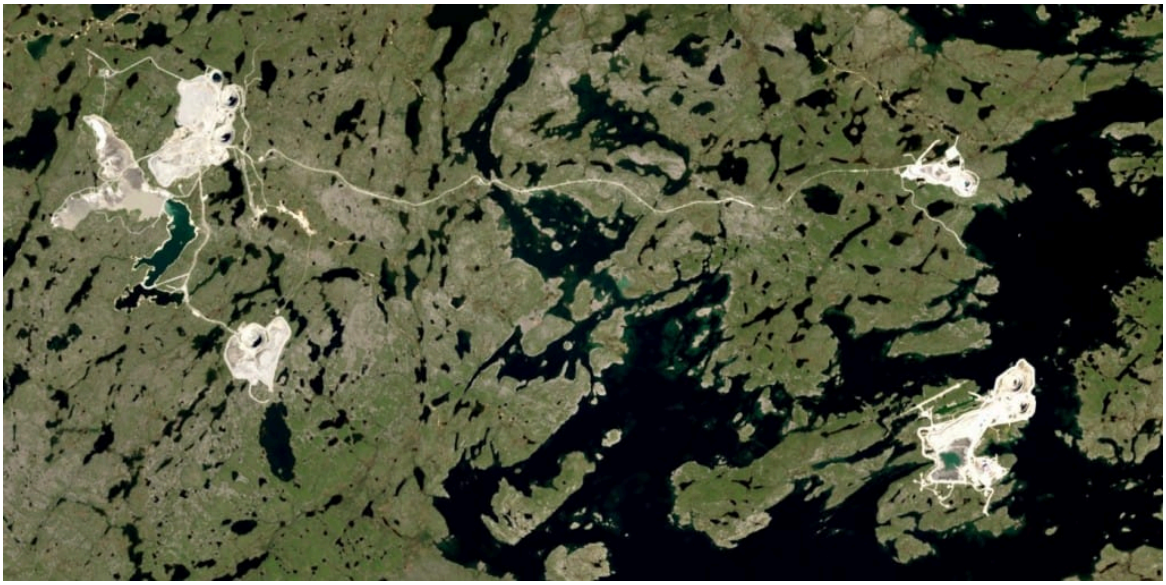


Figure 20.4.1 Diamond mines in the Lac de Gras region, Nunavut. The twin pits of the Diavik Mine are visible in the lower right on an island within Lac de Gras. The five pits of the Ekati mine are also visible, on the left and the upper right. The two main mine centres are 25 km apart.

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20.5: Summary

The main topics of this chapter can be summarized as follows:

Section	Summary
20.1 Metal Deposits	Geological resources are critical to our way of life and important to the Canadian economy. Gold, coal, iron, copper, nickel, and potash are Canada's most valuable mined commodities. The concentrations of metals in mineral deposits are typically several thousand times higher than those in average rocks, and such concentrations only form through specific geological processes. Some deposits form within a magma chamber, others during volcanism or adjacent to a pluton, and some are related to sedimentary processes. Mining involves both surface and underground methods, but in either case, rock is brought to surface that can react with water and oxygen to produce acid rock drainage and metal contamination.
20.2 Industrial Materials	Non-metallic materials are very important to infrastructure and agriculture. Some of the major industrial minerals include sand and gravel, limestone for cement and agriculture, salt for a range of applications, potash fertilizer, and decorative stone.
20.3 Fossil Fuels	The main fossil fuels are coal, oil, and gas. Coal forms on land in wet environments where organic matter can remain submerged and isolated from oxygen for millennia before it is buried by more sediments. The depth of that burial influences the grade of coal produced. Oil and gas originate from organisms living in marine environments, and again, fairly rapid burial is required to preserve the organic matter on the sea floor. At moderate burial depth (2 km to 4 km), oil is produced, and at greater depth, gas is produced. Both oil and gas migrate toward the surface and can be trapped beneath impermeable rock layers in structural features, such as anticlines or faults. Some unconventional fossil fuel resources include oil sands, shale gas, and coal-bed methane.
20.4 Diamonds	Diamonds originate in the mantle and are only brought to the surface by the very rare eruption of kimberlitic volcanoes. The relatively recent discovery of diamonds in Canada was based on the exhaustive search for diamond indicator minerals in glacial sediments. There are now six diamond mines in Canada.

Questions for Review

See Appendix 2 for answers to Review Questions.

1. What are some of Earth's resources that are needed to make a compact fluorescent light bulb? The image on the right of Figure A shows the contents of the ballast.

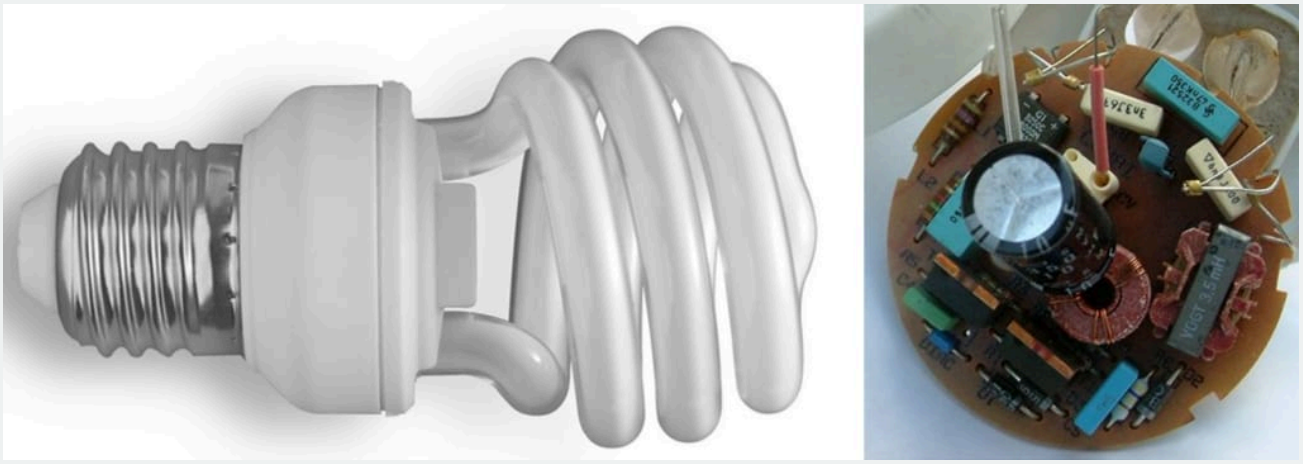


Figure A

- Explain why nickel deposits are associated only with mafic magma and not with intermediate or felsic magma?
- What is the composition of the black smoke in a black smoker, and how does that relate to a volcanogenic massive sulfide deposit?

- How might an epigenetic gold deposit be related to a porphyry deposit?
- Oxidation and reduction processes are important to both banded iron formation deposits and unconformity-type uranium deposits. Explain the role in each case.
- A typical kimberlite in northern Canada may look something like Figure B. In this case, the diameter at the surface is around 500 m, and the total depth is about 2,500 m. Bearing in mind that an open pit cannot typically be any deeper than it is wide, what mining method(s) might be most applicable to a deposit of this type?

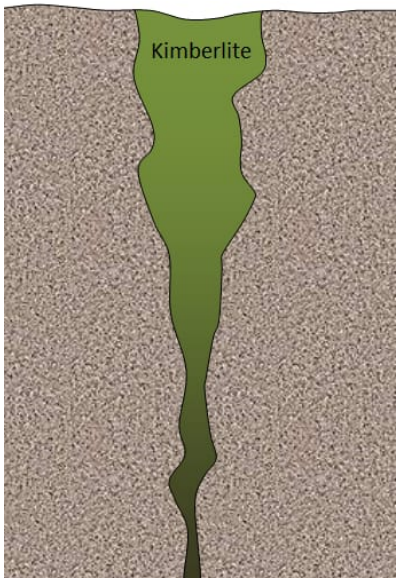


Figure B

- What mineral is typically responsible for acid rock drainage around mine sites, and why is this mineral so common in this setting?
- Explain why glaciofluvial gravel is more suitable than till as a source for aggregate.
- The raw material for making cement is lime (CaO), and this is typically produced by heating limestone (mostly $CaCO_3$) to about 1,000°C. Why is this an environmental issue?
- Name some important industrial minerals that form in an evaporite setting.
- If organic matter accumulates at an average rate of 1 mm per year, and if 10 m of organic matter is required to make 1 m of coal, how long must a swampy environment remain stable and wet in order to form a 1.5 m coal seam?
- What are the ideal characteristics of petroleum source rocks and petroleum reservoir rocks?
- How deep must the source rocks be buried to produce oil?
- Why is shale gas an unconventional reserve, and how is it recovered? What are some of the environmental issues associated with that process?
- If you were sampling glacial deposits to search for kimberlites, why would you be advised to look for kimberlite indicator minerals rather than diamonds?

Media Attributions

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21: GEOLOGICAL HISTORY OF WESTERN CANADA

21.1: Geological History of Canada
21.2: Western Canada during the Precambrian
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21.1: Geological History of Canada

Laurentia, which makes up the core of North America, is the largest and arguably the oldest of Earth's **cratons** (regions of stable ancient crust). Some of the rocks are over 4 billion years old, and Laurentia has been together in its present form for the last billion years. Over the past 650 million years, Laurentia has moved along a zigzag path from deep in the southern hemisphere to close to the North Pole (Figure 21.1.1). During that time, it collided several times with other continents and was temporarily part of two supercontinents (Pannotia and Pangea).

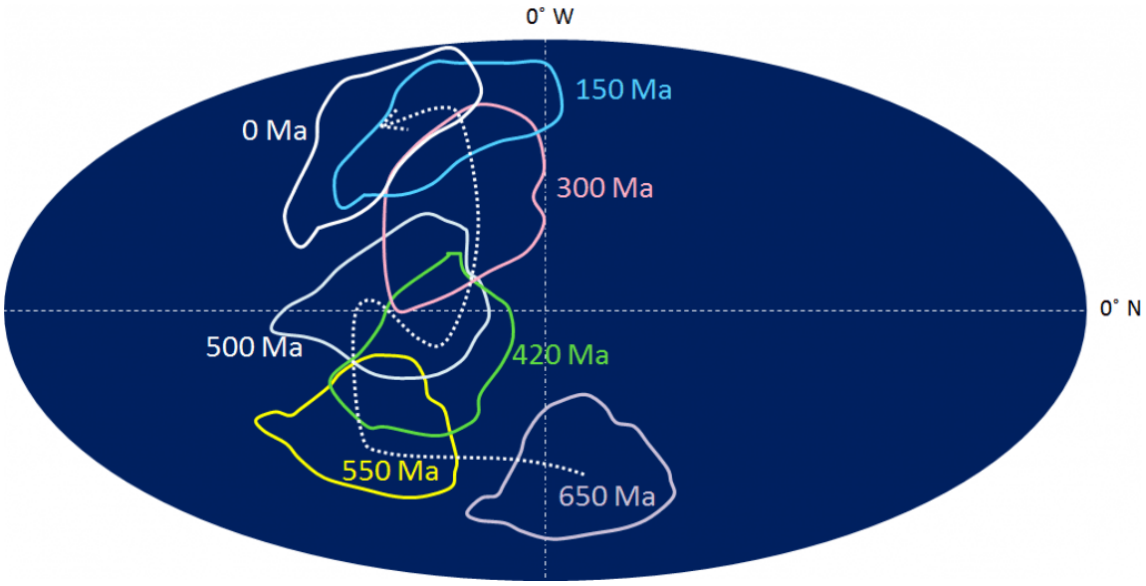


Figure 21.1.1 The path of Laurentia over the past 650 Ma.

Bodies of rock tend to be eroded and recycled through the processes of plate tectonics, including uplift leading to erosion and burial leading to melting, and thus there are very few areas of truly ancient rocks on Earth. The oldest undisputed rocks are those of the Acasta Gneiss from north of Yellowknife, Northwest Territories, aged 4.03 Ga. But there are some rocks that could be even older within the Nuvvuagittuq greenstone belt on the east coast of Hudson Bay, in Quebec. These have been isotopically dated at 4.28 Ga, although the reliability of that date has been questioned. Based on other data, it is acknowledged that the Nuvvuagittuq rocks are at least as old as 3.75 Ga. The Acasta and Nuvvuagittuq rocks are situated within the Slave and Superior Cratons respectively, the oldest parts of Laurentia (Figure 21.1.2). Although not all of the rocks in these ancient cratons are that old, they are generally older than 3 Ga, as is part of the Wyoming Craton. The Hearne and Rae Cratons are older than 2 Ga, while most of the other parts of Laurentia are aged between 1 Ga and 2 Ga. The various provinces of Laurentia were assembled by plate-tectonic processes between 1 Ga and 3 Ga.

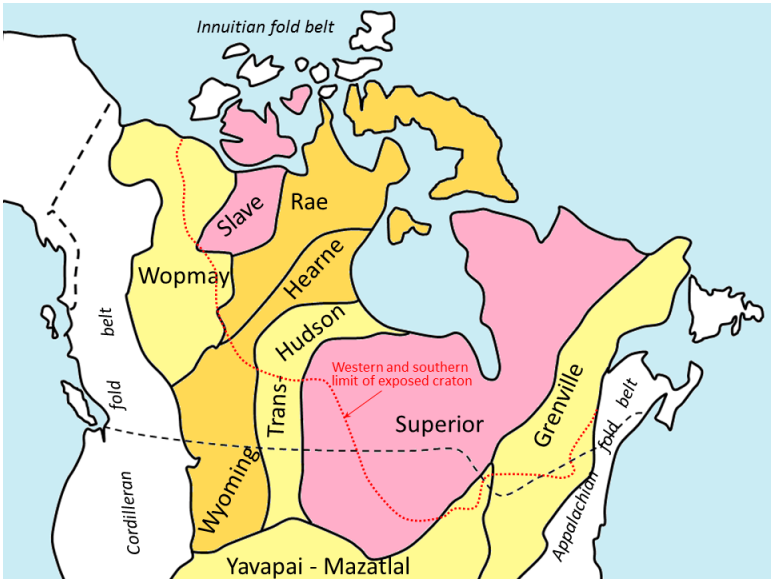


Figure 21.1.2 The main provinces of Laurentia. The pink areas are the oldest; light yellow are the youngest. All of the areas south and west of the dotted red line are now covered with younger rocks. The white areas represent rocks that were added to North America since 700 Ma.

The areas of Figure 21.1.2 that are left uncolored—the Appalachian, Innuitian, and Cordilleran fold belts—are geological regions that have been added to North America since 500 Ma. These are at least partly made up of sedimentary rocks that were deposited along the coasts and then folded, faulted, and uplifted during continental collisions.

The term *Laurentia* is geologically equivalent to the term **Canadian Shield**, although the latter is generally considered to be the area where the ancient Laurentian rocks are exposed at the surface and not covered with younger rocks. That applies to most of the region to the north and east of the red dotted line in Figure 21.1.2.

Laurentia was part of the supercontinent Rodinia during the period between 1,100 Ma and 700 Ma. As Rodinia started to break up after 700 Ma, sediments derived from the erosion of the interior of the continent began to accumulate along its coasts, initially along the west coast, then the east coast at around 600 Ma, and finally on the north coast by around 550 Ma. This process continued for several hundred million years. By around 450 Ma, large areas of the interior of Laurentia were depressed below sea level—probably because of the downward pull of an underlying subducting plate—and marine sediments were deposited over parts of Quebec, Ontario, Manitoba, Saskatchewan, Alberta, and the Northwest Territories during the Ordovician, Silurian, and Devonian Periods (450 Ma to 350 Ma). These sediments are colored various shades of blue on the geological map of Canada (Figure 21.1.3).

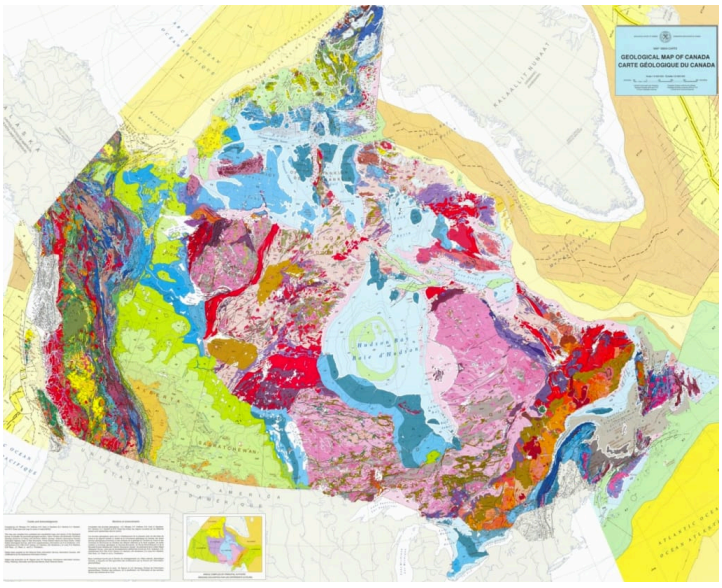


Figure 21.1.3 Geological map of Canada from the Geological Survey of Canada.

At approximately 350 Ma, the part of Gondwana that is now Africa collided with the eastern coast of North America, thrusting volcanic islands and sedimentary layers far inland to become the Appalachian fold belt. The Appalachian Mountains would likely have rivalled the Himalayas in extent and height during the Devonian. At about the same time, a smaller continent, Pearya, collided with the north coast, creating the Innuitian fold belt.

At around 200 Ma, small continents that now make up the interior of B.C. and part of Yukon collided with the west coast of North America, starting the process of thrusting the sedimentary rocks inland and upward to form the Rocky Mountains.

The west-central part of North America subsided once again at around 150 Ma, due to an underlying subducting plate, and this led to the deposition of more marine rocks across Manitoba, Saskatchewan, and Alberta, and north into the Northwest Territories and Yukon (the green areas in Figure 21.1.3).

Finally, at around 90 Ma, more small continents, which now comprise Vancouver Island and Haida Gwaii, collided with the west coast, leading to further uplift of the Rocky Mountains.

Exercise 21.1 Find the geological provinces of Canada

Figure 21.1.3 shows the geology of Canada in some detail, with the colors representing the lithologies and ages of the rocks. Identifying in Figure 21.1.3 some of the major features shown in Figure 21.1.2 will help you understand the distribution of Canada's geological features. Start by outlining the extent of the exposed Canadian Shield (the dotted red line); you might also be able to identify some of the cratons within the Shield. Then look for the limits of the Appalachian and Innuitian fold belts. Finally pick out the extent of the Cordilleran fold belt.

The best way to do this would be to print out a copy of Figure 21.1.3 and draw the boundaries from Figure 21.1.2 on it with a pencil.

If you're interested, you can get your own high resolution copy of the geological map of Canada: http://geoscan.nrcan.gc.ca/starweb/g...arch1=R=208175.

See Appendix 3 for Exercise 21.1 answers.

Image Descriptions

Figure 21.1.2 image description: Oldest areas: From Northern Quebec to southern Ontario. Also some land on northern part of the border between and Northwest Territories and Nunavut, including Victoria Island. The next oldest areas include most of central and eastern Nunavut, down through Alberta and Saskatchewan and into the United States. The youngest areas include northeastern British Columbia and up along the Yukon-Northwest Territory border. It also comes down through Manitoba and into the United States. The other youngest area stretches from Newfoundland and Labrador, down through eastern Quebec and into the United States before turning west through the central United States. The eastern and western coastal areas of the United States and Canada have been added in the last 700 Ma. [Return to Figure 21.1.2]

Media Attributions

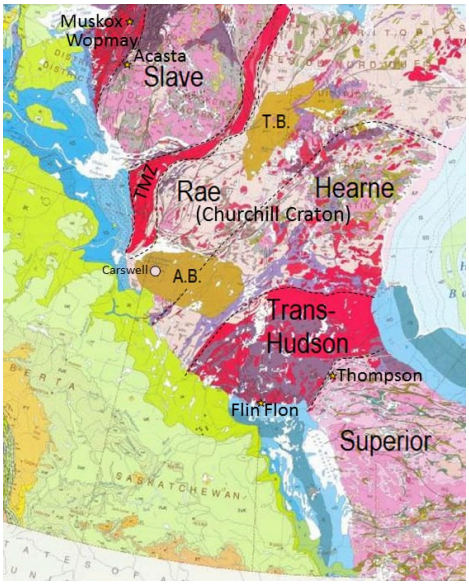
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21.2: Western Canada during the Precambrian

Laurentia extends as far west as eastern B.C. (Figure 21.2.2), but the ancient rocks of the craton are almost completely covered by younger rocks in B.C., Yukon, and most of Alberta except the far northeast corner. Laurentia is well represented in northern Saskatchewan and across large parts of Manitoba, the Northwest Territories, and Nunavut (Figure 21.2.1). Where they are exposed, the rocks of the Canadian Shield are highly varied lithologically, typically strongly metamorphosed due to their deep burial at some time in the past, and in some cases, quite different from what could be expected to occur on Earth today.

Starting from the south, in eastern Manitoba and adjacent Ontario, we have the ancient rocks of the Superior Province. On the map the Superior Province, rocks are mostly pink, representing granitic and gneissic rocks, with strips and blotches of green, representing metamorphosed sea-floor basalt and sediments, also known as greenstone belts. These rocks are widely interpreted to have deep crustal origins, and include large areas of granulite facies metamorphic rock formed at high temperatures and moderate to high pressures (see Figure 21.2.7). Superior Province greenstone belts in Ontario and Quebec host some of the world's largest volcanogenic massive sulfide deposits. As described in Chapter 20, the Superior Province in northern Manitoba is host to important nickel deposits at Thompson. These formed from mantle-derived mafic magma that interacted with sulfur-bearing crustal rocks, and within which heavy-metal sulfide minerals formed.



21.2.1 The Trans-Hudson Orogen (THO), as its name implies, extends through Saskatchewan and Manitoba and over to the eastern side of Hudson Bay. It represents the continent-continent collision zone between the Superior Craton to the south and the Churchill Craton (including the Wyoming, Hearne, and Rae Cratons) to the north; thus it's a remnant of the initial formation of Laurentia at around 1.9 Ga. At the time of the collision, the THO would have been a major mountain range, and the rocks that we see there now — which evolved deep beneath those mountains — are highly metamorphosed sedimentary and volcanic rocks intruded by large granitic bodies. The important volcanogenic massive sulfide deposits around Flin Flon are within the THO.

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The Churchill Craton is lithologically similar to the Superior Craton, although not generally as old. It includes two important sedimentary basins: the Athabasca Basin in Saskatchewan and the Thelon Basin in Nunavut, both filled with rocks aged around 1.7 Ga. These consist primarily of sandstones and minor mudstones that are only weakly metamorphosed and essentially undeformed

(not folded) because they are situated within a stable craton and so have not been subjected to significant tectonic forces. The Athabasca Basin is economically important for its large and rich unconformity-type uranium deposits (see Chapter 20). At its western end, there is the remnant of a large extraterrestrial impact, the 40 km diameter Carswell Crater. When the meteor struck at this location, at around 115 Ma, the impact and subsequent rebound of the crust was enough to bring metamorphic rock up to surface from beneath about 2,000 m of Athabasca Group sandstone. There is no connection between the Carswell Crater and the much older (~1.2 Ga) uranium deposits.

The Taltson Magmatic Zone (TMZ), which forms the boundary between the Churchill and Slave Cratons, consists primarily of granitic rock. One interpretation is that the TMZ formed along a convergent boundary, although this is not universally accepted.



Figure 21.2.2 A sample of the Acasta Gneiss on display at the Natural History Museum in Vienna.

The Slave Craton is dominated by granitic rocks and metamorphosed clastic sedimentary rocks. On its western edge, there is a large area of very old gneissic rock that includes the Acasta Gneiss, dated at 4.03 Ga, which—for the time being at least—is the oldest rock in the world (Figure 21.2.2).

The Wopmay Orogen, interpreted as the site of another ancient continent-continent collision, lies to the west of the Slave Craton. Although mostly composed of felsic igneous rocks and gneisses, the Wopmay Orogen includes a body of mafic and ultramafic igneous rock called the Muskox Intrusion. Derived from a mantle plume and dated at about 1.1 Ga, the Muskox is comparable to a handful of other mafic and ultramafic intrusions around the world in that it has distinctive repetitive layering caused by settling of heavy metal-rich minerals within the low-viscosity magma. Muskox has high levels of nickel, copper, and chromium, and has the potential to have platinum and palladium like a similar body in South Africa. Ultramafic intrusions like Muskox do not take place on Earth today because the mantle is no longer hot enough.

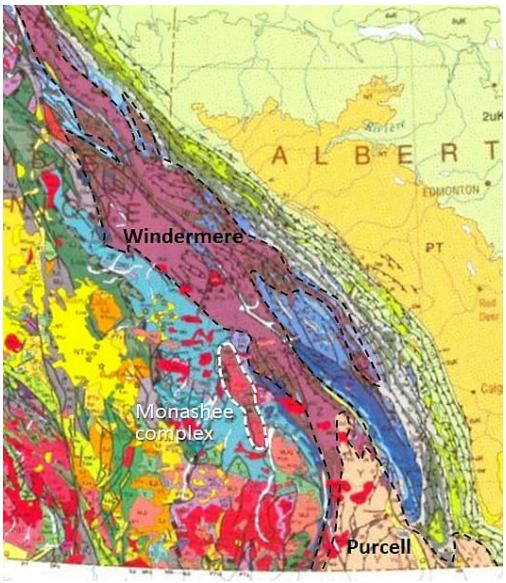


Figure 21.2.3 Precambrian rocks in southern B.C. and Alberta.

The oldest rocks in British Columbia are the strongly metamorphosed sedimentary, volcanic, and intrusive rocks of the Monashee Complex, situated to the west of the Columbia River near Revelstoke (Figure 21.2.3). Aged around 2 Ga, these may actually be part of Laurentia.

There are much more extensive Precambrian rocks within the Columbia and Rocky Mountains of southeastern B.C. and the southwestern corner of Alberta. The rocks of the Purcell Supergroup (a **supergroup** comprises more than one group) are present in the extreme southeastern corner of B.C. and adjacent Alberta, and extend well into the United States (as the Belt Supergroup). These are mostly unmetamorphosed clastic rocks deposited in rivers and lakes during the middle Proterozoic, at around 1,400 Ma, while Laurentia was still part of the supercontinent Columbia. When Columbia rifted apart, the division happened within the area of the Purcell/Belt rocks. Similar rocks of the same age are present in Tasmania and Siberia, and it is postulated that they were once part of the same depositional basin.

Exercise 21.2 Purcell rocks down under?

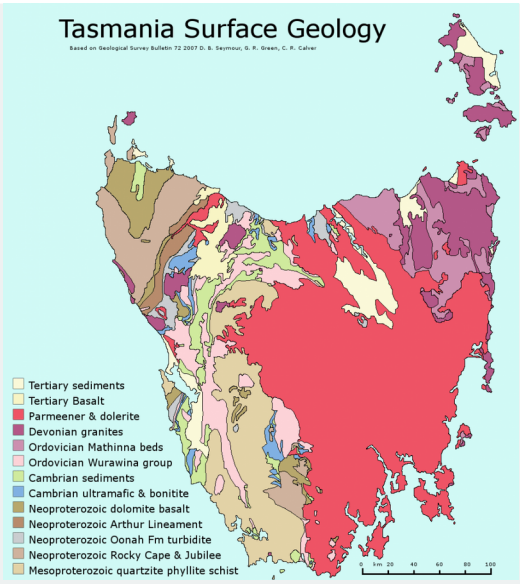


Figure 21.2.4

This map shows the geology of the Australian state of Tasmania. Identify which rocks might be comparable to the Purcell rocks of B.C. and Alberta.

In what way are these rocks different from those in Canada?

See Appendix 3 for Exercise 21.2 answers.

The Windermere Group rocks—also mostly clastic sedimentary—were deposited in the ocean along the western edge of Laurentia (Figure 21.2.3) in the late Proterozoic (around 700 Ma) after the breakup of Columbia. In fact, sedimentary rocks of this age extend all along the western side of the Rocky Mountains, well into Yukon.



Figure 21.2.5 Late Proterozoic Toby Formation mudstone with glacial dropstones, south of Cranbrook, B.C.

Deposition in this area was taking place during the late Proterozoic Snowball Earth glaciations, as can be seen in Windermere Group rocks of the Toby Formation from the area south of Cranbrook, B.C. (Figure 21.2.5). The Toby Formation is a fine-grained marine rock (mudstone) with numerous large angular clasts of limestone and quartz. The mud was deposited in the quiet water of a continental slope environment, and the large clasts were dropped from floating ice derived from glaciers on Laurentia. The Toby Formation is unique in this area; most of the rest of the late Proterozoic clastic sedimentary rocks in this region do not have glacial dropstones.

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21.3: Western Canada during the Paleozoic

At the beginning of the Paleozoic (542 Ma), Laurentia was near the equator (Figure 21.3.1) and sedimentation was continuing on all of Laurentia's marine margins, including the passive margin (not tectonically active) on what is now the west coast. The clastic sediments of the Windermere Group are succeeded by mostly limestone beds (represented by the blue areas in Figure 21.3.3) interbedded in some areas with mudstone and sandstone. The most famous Cambrian rocks in the Rockies are those around Field, B.C., within Yoho and Kootenay National Parks. The Burgess Shale of the Stephen Formation is considered by some to be the most important fossil bed in the world because of its spectacular preservation of detail in a wide array of organisms that are ancestors to many of today's organisms and are not present in earlier rocks. The Walcott Quarry, on the pass between Mt. Field and Wapta Mountain has been known and studied for over 100 years (Figure 21.3.1). In 2012 a new Burgess Shale discovery was made at Marble Canyon, about 30 km to the southeast, by a team led by the Royal Ontario Museum (ROM). Fossils with similar levels of preservation are present, and several previously unknown organisms have been found. The ROM continues to work in the Marble Canyon area and some of their discoveries are described and illustrated on their website, such as a discovery of a large Burgess Shale fossil site in 2014. The Paleozoic strata of the Rockies also include Ordovician, Devonian, Carboniferous, and Permian sedimentary rocks. For example, Carboniferous limestone makes up most of the upper part of Crowsnest Mountain in southern Alberta (Figure 21.3.1).



Figure 21.3.1 The Cambrian Burgess Shale at the Walcott Quarry, Yoho Park, B.C., with Wapta Mountain in the upper right.

While clastic and carbonate sediments were accumulating along the western edge of Laurentia, much of the interior of the continent was submerged under inland seas that were connected to ocean most of the time. This region is known as the Western Canada Sedimentary Basin (WCSB). The Paleozoic sediments that accumulated within this basin show up as the blue areas in Figures 21.1.3 and 21.2.1; however, their extent is much wider than that because Paleozoic sedimentary rocks also underly the Mesozoic rocks within most of the areas that are light green on those maps. By way of example, a schematic cross-section through the Paleozoic and Mesozoic rocks of southern Manitoba is given in Figure 21.3.2. The section extends from the Saskatchewan-Manitoba border on the left to just east of Winnipeg on the right, and shows the Paleozoic rocks overlain on the rocks of the Precambrian Superior Craton.

Fifteen different Paleozoic formations, ranging in age from Ordovician to Carboniferous, are shown in Figure 21.3.2. Of these, 11 are dominated by carbonate rocks (limestone or dolomite) that very likely formed in an ocean-connected marine environment. The non-carbonate formations are the lowermost one (resting on Precambrian rocks), which is sandstone of marine origin; the Devonian Prairie Evaporite Formation (in red)—the same formation from which potash is mined in Saskatchewan; and the upper two Devonian formations (in yellow), which are shale. When the Prairie Evaporite formed, the basin was isolated from the open ocean, and the rate of evaporation was greater than the rate of input from precipitation and river inflow. During that time, probably at least several million years, there were numerous changes in sea level or land level that allowed additional ocean water—and therefore additional salt—into the basin.

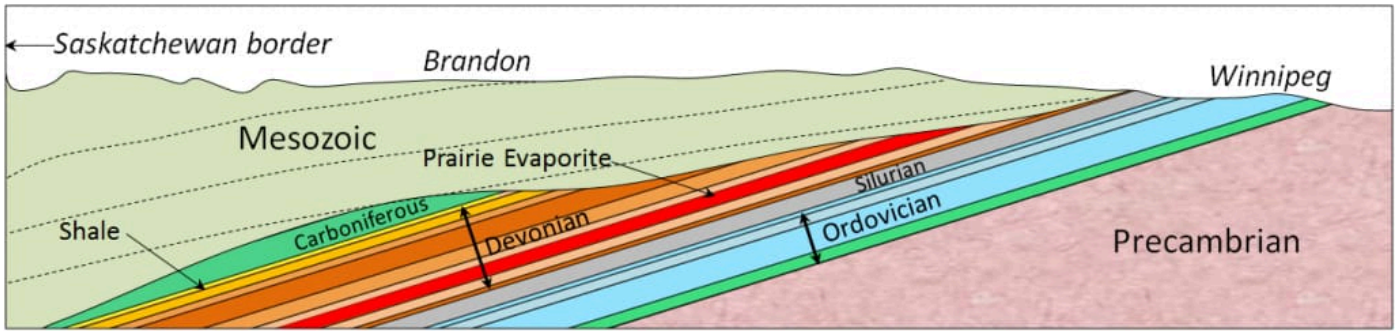


Figure 21.3.2 The Paleozoic sedimentary rocks of southern Manitoba along a section extending from the Saskatchewan border on the left to the Winnipeg area on the right. The section is 400 km wide and 1,800 m high, and the vertical exaggeration is about 100 times. The dip of the beds is also exaggerated by 100 times; their original and current attitudes are close to horizontal.

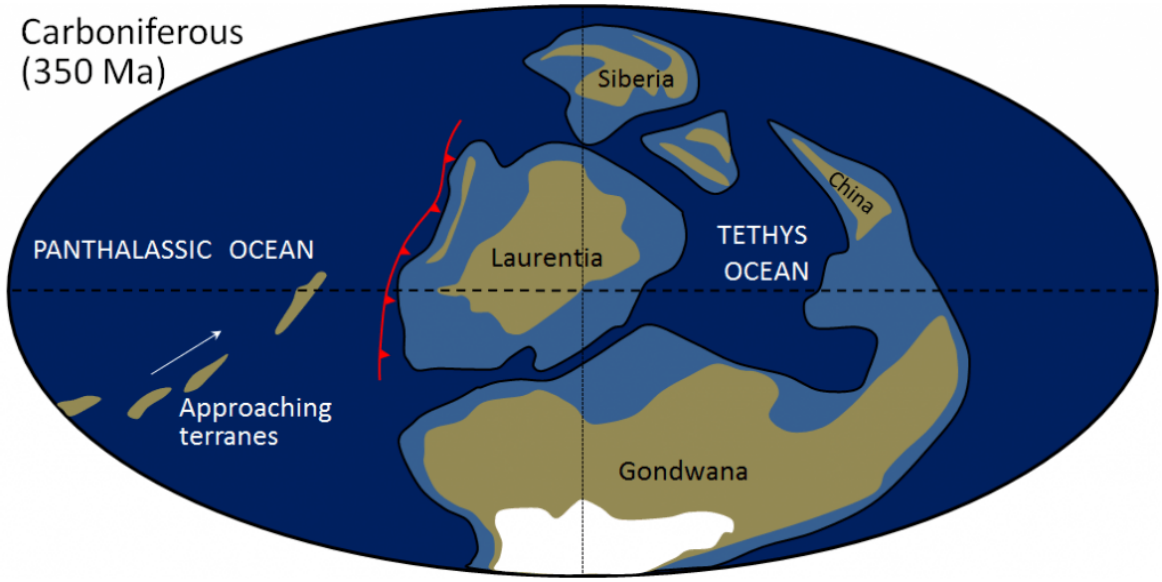


Figure 21.3.3 The distribution of continents in the early Carboniferous, showing the terranes that later became attached to the west coast of North America. The light blue areas are continental shelves, the white is ice of the Karoo Glaciation, and the red line shows subduction of oceanic crust beneath Laurentia. Panthalassic is the name for the huge ocean that preceded the Pacific Ocean.

There are Paleozoic rocks in the central and western parts of British Columbia and Yukon, but they formed far away and did not become part of North America until the Mesozoic. Subduction started along the western edge of Laurentia by the middle Paleozoic. That meant that oceanic crust was moving toward the continent, bringing small segments of exotic continental crust with it (Figure 21.3.3). These crustal blocks along western North America are called **terranes**, indicating that they are sections of the continent that have an exotic origin (Figure 21.3.4). Most of British Columbia is made up of terranes that include sedimentary rocks with fossils that imply an origin south of the equator, or volcanic rocks with magnetic orientations that indicate a southern-hemisphere origin.



Figure 21.3.4 The Carboniferous Mt. Mark Formation on Vancouver Island is part of the Wrangellia Terrane, which arrived on the edge of North America during the Cretaceous.

Exercise 21.3 What is Vancouver Island made of?

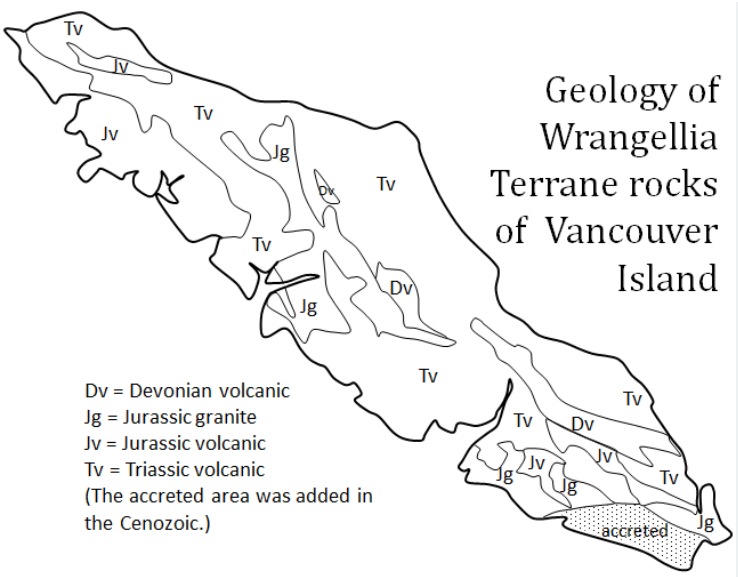


Figure 21.3.5 Rocks of Vancouver Island: Triassic volcanic, Devonian volcanic, Jurassic granite, and Jurassic volcanic.

This map shows the main geological features of the Wrangellia Terrane rocks that were present when Vancouver Island arrived on the coast of North America.

1. Roughly what percentage of Vancouver Island is underlain by Paleozoic rock?
2. What is the most common age and type of rock on Vancouver Island?

To answer these questions, you might find it useful to fill in each rock type area using colored pencils.

See Appendix 3 for Exercise 21.3 answers.

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21.4: Western Canada during the Mesozoic

The Mesozoic extends over 187 million years from the beginning of the Triassic (252 Ma) to the end of the Cretaceous (65.5 Ma). It was a particularly important period for the geology of western Canada. During this time, several continental collisions occurred along the west coast, resulting in the formation of the Rocky Mountains and the **accretion** (addition) of much of the land mass of British Columbia, and continuing deposition within the WCSB.

Terrane Accretion in British Columbia and Yukon

Continued subduction along the western edge of North America carried a number of continental terranes toward the coast, with the first collisions taking place in the early part of the Triassic, as the Quesnel, Cache Creek, and Stikine Terranes combined to form the Intermontane Superterrane, so named because it forms the interior plateau of British Columbia, between the Rockies to the east and the Coast Range to the west (Figure 21.4.1).

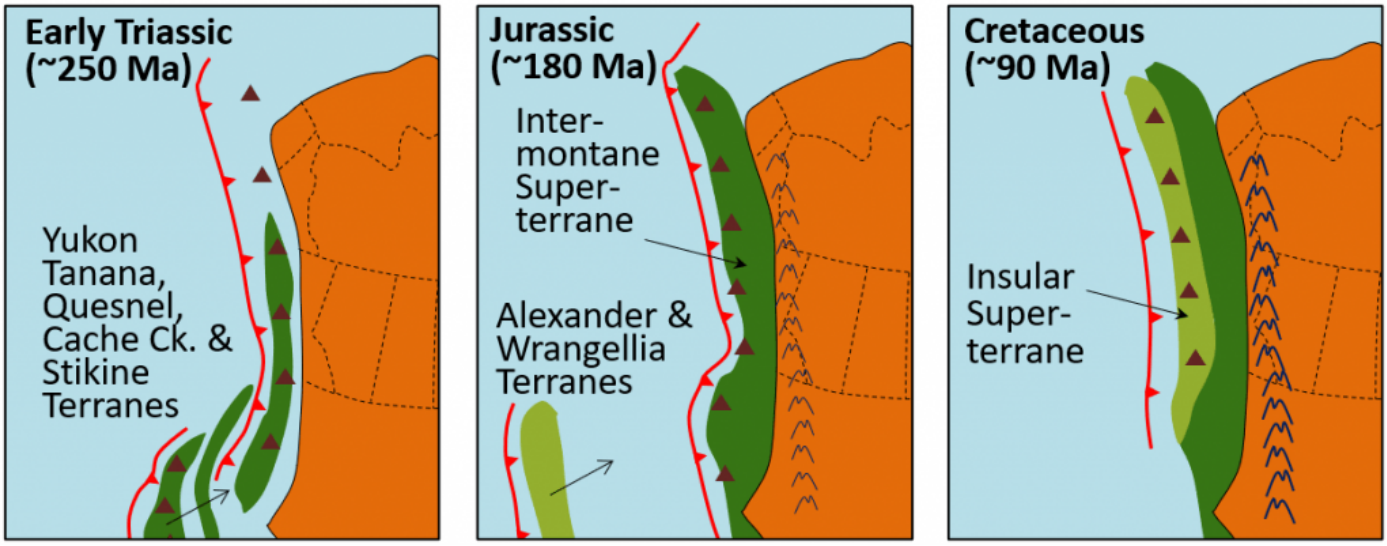


Figure 21.4.1 Model of the accretion of the Intermontane and Insular Superterranes to the west coast of North America during the Mesozoic. Subduction zones are the red-toothed lines. The dark-red triangles represent volcanoes.

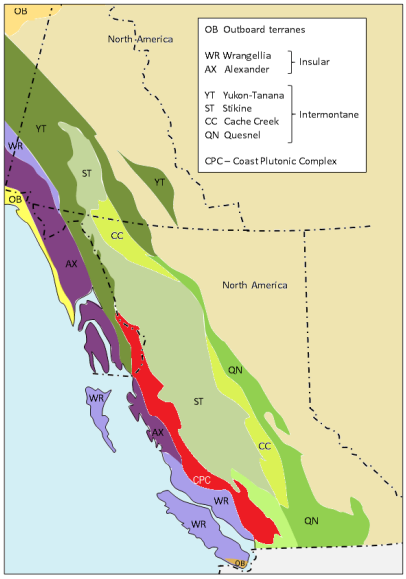


Figure 21.4.2 A generalized overview of the accreted terranes of B.C., Yukon, and Alaska. The Intermontane terranes are in green, the Insular terranes in purple, and the Outboard terranes in yellow. The Coast Plutonic Complex (CPC) formed in situ and is not a terrane.

Approximately 100 million years later, another pair of terranes—Alexander and Wrangellia—collided to form most of Vancouver Island and Haida Gwaii, plus a significant part of Alaska. During the Cenozoic, additional terranes (the Outboard terranes) were added to the western edge of North America. An overview of the accreted terranes of B.C., Yukon, and Alaska is given in Figure 21.4.2.

During the Jurassic, the Intermontane Superterrane acted like a giant bulldozer, pushing, folding, and thrusting the existing Proterozoic and Paleozoic west coast sediments eastward and upward to form the Rocky Mountains (Figure 21.4.3). The same process continued into the Cretaceous as the Insular Superterrane collided with North America and pushed the Intermontane Superterrane farther east. Folding in Rocky Mountain rocks, like that shown in Figures 21.4.4 and 21.4.5, is one of the results of this process.

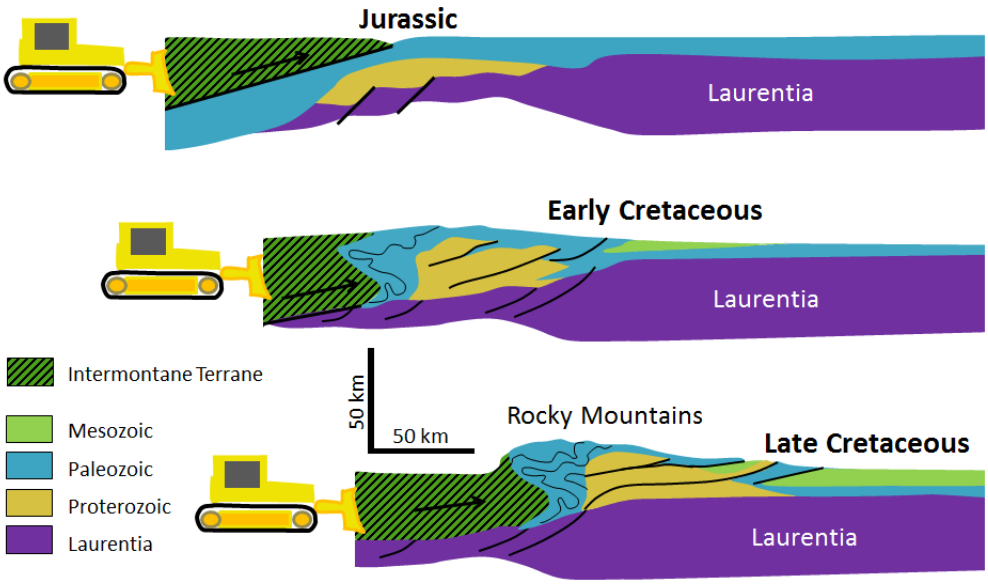


Figure 21.4.3 Cross-section of the accretion of the Intermontane Superterrane to the west coast of North America and the resulting compression, folding, and thrusting of North American sedimentary rocks. In the Late Cretaceous, it was the accretion of the Insular Superterrane pushing against the Intermontane Superterrane that did most of the work.



Figure 21.4.4 Tight folds in sedimentary rocks of the Rocky Mountains near Field, B.C.



Figure 21.4.5 Folding of sedimentary rocks at Mt. Rae, Alberta.

Thrusting is another important process in the formation of fold-belt mountains, as described in Chapter 12. During plate convergence, entire sheets of sedimentary rock are slowly pushed over top of other sheets, resulting in situations where older rocks lie on top of younger ones. One of the best known examples of this is at Mt. Yamnuska, near Exshaw, Alberta (Figure 21.4.6), where the older Cambrian rocks were pushed east by a total of 40 km, over top of younger rocks.

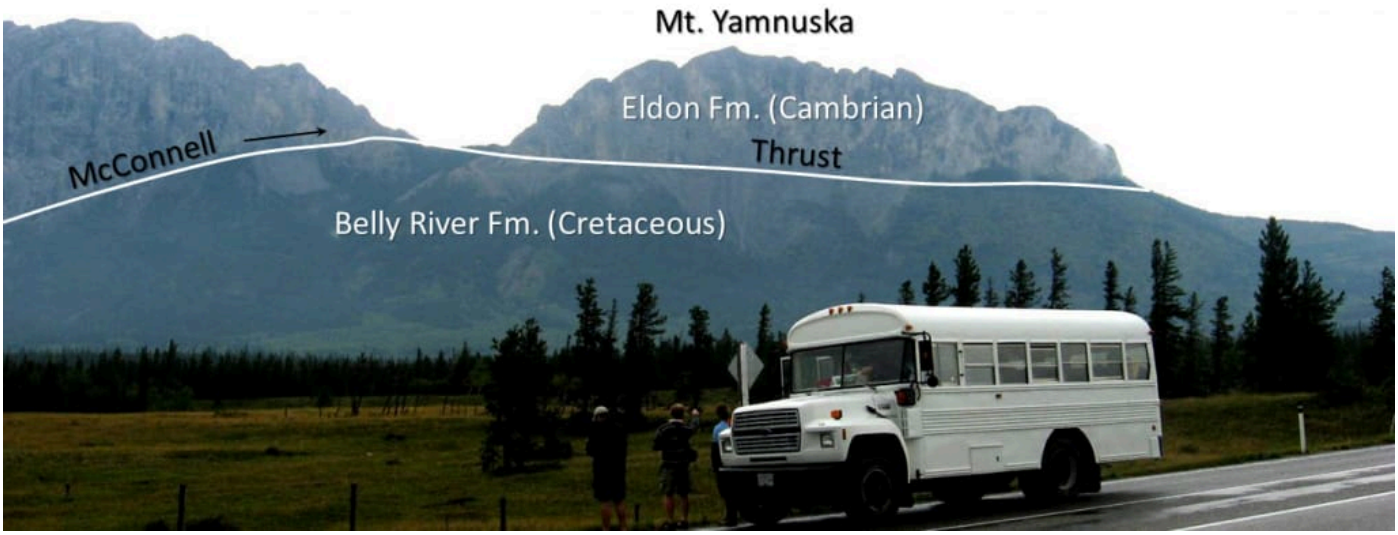


Figure 21.4.6 The McConnell Thrust at Mt. Yamnuska near Exshaw, Alberta. Carbonate rocks of Cambrian age have been thrust over top of Cretaceous mudstone.

In the area near the U.S. border, within B.C., Alberta, and Montana, a sheet of Paleozoic rocks has been thrust about 80 km east over top of Cretaceous rocks along the Lewis Fault (Figure 21.4.7).

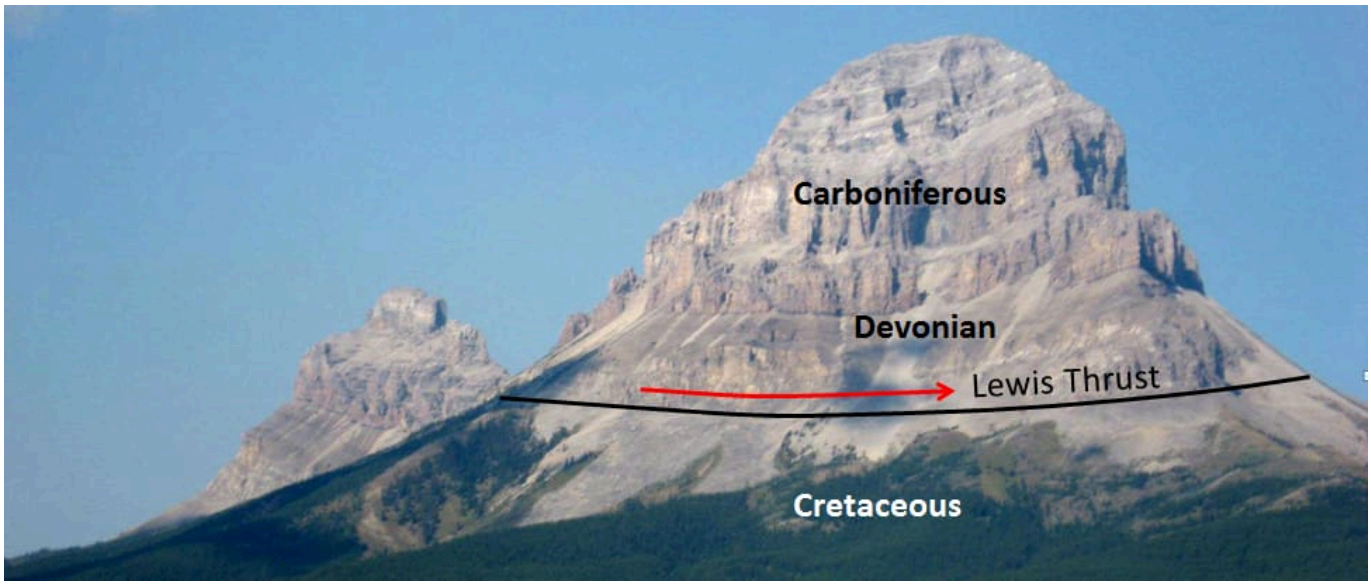


Figure 21.4.7 The Lewis Thrust at Crowsnest Mountain near Frank, Alberta. Carbonate rocks of Devonian and Carboniferous age have been pushed 80 km to the east and thrust over top of Cretaceous mudstone.

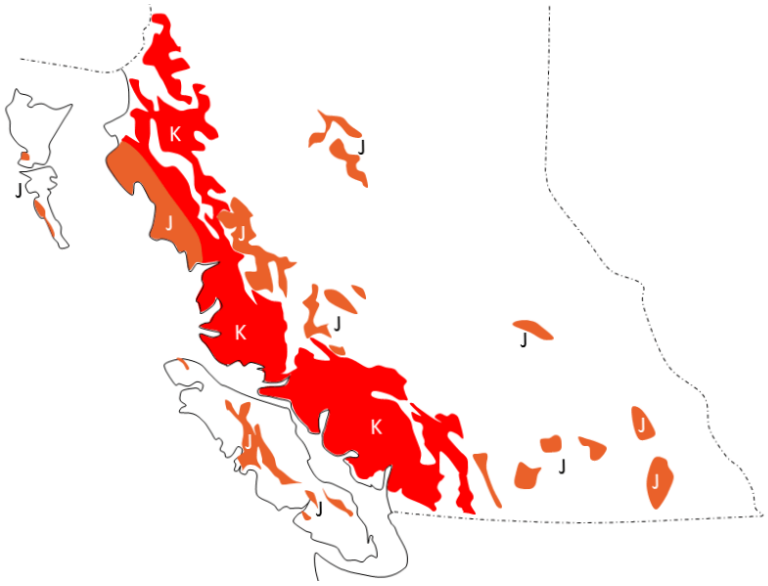


Figure 21.4.8 Jurassic (J) and Cretaceous (K) granitic bodies in south-central B.C.

Not only did the subduction of oceanic crust beneath North America during the Mesozoic deliver geologically exotic terranes to the western edge of the continent, it also resulted in massive amounts of volcanism along the boundary (Figure 21.4.1). The upper-crustal magma chambers that fed those now-eroded volcanoes slowly cooled into granitic and dioritic stocks, and those stocks gradually coalesced into batholiths that extend from the southwest corner of B.C. all the way into Yukon and Alaska (Figures 21.4.8 and 21.4.9). Most of the granitic rocks of this region fall into two main age ranges: many are middle-Jurassic to early Cretaceous in age (~ 170 Ma to 140 Ma), while others are late Cretaceous to Paleogene (~50 Ma to 90 Ma). Many of the older bodies intruded into the terranes they are on before they arrived on the North America coast. This applies to those on Vancouver Island and Haida Gwaii. Some of the older ones formed in situ when the subduction zone was farther east (see Figure 21.4.1). Most of the younger bodies formed in situ when the subduction zone was close to where it is now (west of Vancouver Island) or slightly to the east.



Figure 21.4.9 Jurassic-Cretaceous granite adjacent to the Coquihalla Highway, north of Hope, BC. (The dark streaks are water from melting snow.)

Although these intrusive igneous rocks cooled at depth in the upper crust, they now form some of the highest peaks in Canada, many of them hundreds of meters higher than those of the Rocky Mountains (including Mt. Waddington, the highest peak entirely within British Columbia at 4,091 m). It is estimated that over the past 100 million years some of these igneous bodies have been uplifted in the order of 8,000 m. Much of that uplift is a result of the relative low density of the granitic rocks compared with the surrounding rocks.

The Western Canada Sedimentary Basin during the Mesozoic

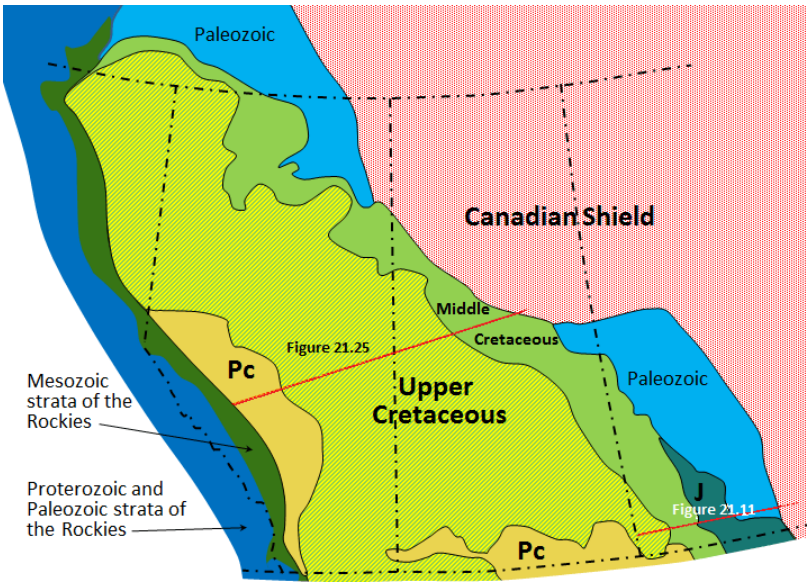


Figure 21.4.10 The distribution of Mesozoic sedimentary rocks in the Western Canada Sedimentary Basin. J stands for Jurassic and Pc for Paleocene. The lines of the cross-sections of Figures 21.3.2 and 21.4.11 are shown.

The construction of the Rocky Mountains during the Jurassic and Cretaceous—and their ensuing erosion—created a significant new source of sediments for the WCSB. Based on the ages, distributions, and thicknesses of the sedimentary layers (Figure 21.4.11), it is evident that the greatest volumes of sediment were produced in the Upper Cretaceous (100 Ma to 65 Ma) and into the Paleocene (65 Ma to 55 Ma). The sediments accumulated in a basin that is thought to have been at least partly formed by the presence of a subducting slab of oceanic lithosphere underneath this part of North America. However, the ongoing uplift of the Rockies through this time period also led to isostatic depression of the crust. The western edge of the basin, which has about 4,500 m of Mesozoic rock alone, is a foreland basin. (See Chapter 6 for more on the origins of basins.)

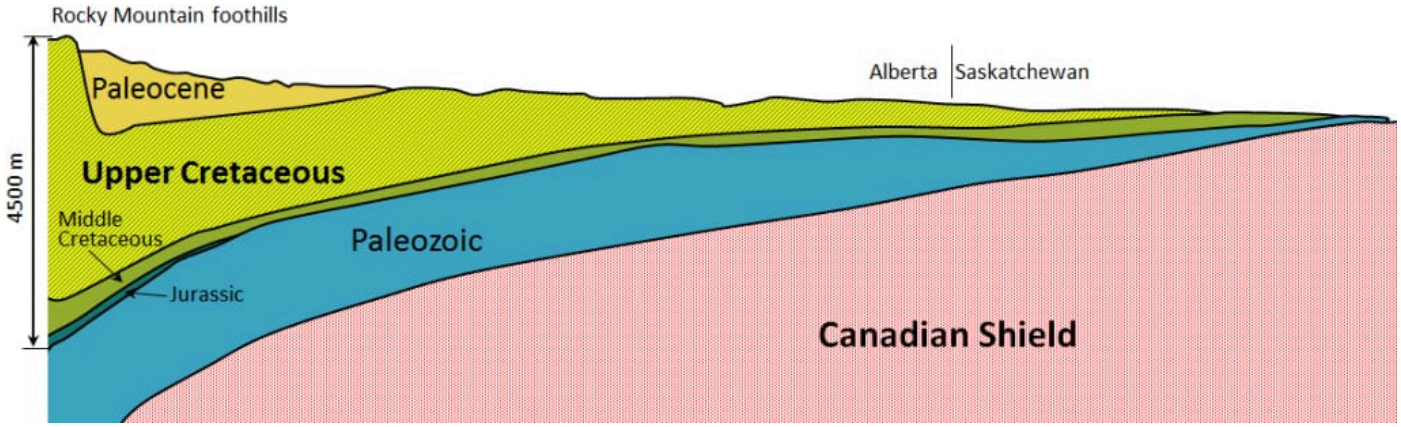


Figure 21.4.11 A cross-section showing the Mesozoic sedimentary rocks in the WCSB from the Rocky Mountain foothills to north-central Saskatchewan.

From time to time during the Mesozoic, the WCSB was filled to varying degrees with marine water. The Jurassic rocks at the base of the sequence are marine in origin, although the Jurassic sequence in Manitoba also includes evaporite layers. Most of the Middle Cretaceous rocks across the basin are marine, but the majority of the Upper Cretaceous rocks are of terrestrial origin, deposited within the flood plains and deltas of rivers. Some of these terrestrial sediments include coal layers; as described in Chapter 20, there are significant coal deposits in central Alberta.

The Paleozoic sediments within the WCSB were buried deeply beneath the Mesozoic sediments and were heated enough to form both oil and gas. There are large petroleum resources in reservoir rocks of various ages extending from northeastern B.C. to southwestern Manitoba.



Figure 21.4.12 The Horseshoe Canyon Formation near Drumheller, Alberta.

Several of the terrestrial Cretaceous formations in the WCSB are host to important dinosaur fossils. Some are within B.C. and Saskatchewan, but the most famous are in Alberta, including the Dinosaur Park Formation (Figure 21.4.1), the Scollard Formation, and the Horseshoe Canyon Formation (Figure 21.4.12). The Dinosaur Park Formation has one of the greatest concentrations of dinosaur fossils of any rock on Earth, with at least 50 genera of dinosaurs represented, ranging from tiny *Hesperonychus* to giant *Albertosaurus*. The Hilda Bone Bed, situated about 80 km to the east of Dinosaur Park, is estimated to have the remains of approximately 1500 ceratopsians, all of which are interpreted to have died in a flood related to a tropical storm. A few of the larger herbivorous dinosaurs found at Dinosaur Park are illustrated in Figure 21.4.13.



Figure 21.4.13 Depiction of some of the large herbivorous dinosaurs from the upper part of the Dinosaur Park Formation. Left to right: the ceratopsian Pentaceratops , the hadrosaur Lambeosaurus eating from a tall tree, the ceratopsian Styracosaurus, the ankylosuar Scolosaurus, the hadrosaur Prosaurolophus (in the distance), the ankylosaur Panoplosaurus, and a herd of Styracosaurs in the background.

Exercise 21.4 Dinosaur country?

The photo is from the Cretaceous Dinosaur Park Formation in Alberta. What characteristic of this rock tells you that it might be suitable for the discovery of the fossils of dinosaurs or other terrestrial organisms?



Figure 21.4.14

See Appendix 3 for Exercise 21.4 answers.

Several important depositional basins existed in British Columbia during the Mesozoic, including the large Jurassic-aged Bowser Basin north of Terrace, and the smaller late-Cretaceous Nanaimo Basin between Vancouver Island and the mainland. In both cases, the rocks are mostly clastic, with both terrestrial and marine deposition.

Image Descriptions

Figure 21.4.1 **image description:** During the early Triassic period (around 250 Ma), subduction along the west coast of North America caused the Quesnel, Cache Creek, and Stikine Terranes to collide with the continent. By the Jurassic persion (180 Ma), the collision had caused the formation of mountains along the coast and the creation of the Intermontane Superterrane, the present interior plateau of British Columbia. The Alexander and Wrangellia Terranes were also pulled towards the coast and by the Cretaceous period (around 90 Ma), they had formed the Insular Superterrane. [Return to Figure 21.4.1]

Figure 21.4.2 **image description:** This image shows two groups of terranes, the Intermontane terranes and the Insular terranes. The Intermontane terranes make up the interior portions of present-day British Columbia and the Yukon. The Yukon-Tanana terrane in the Yukon and the Quesnel terrane in BC are the farthest inland. As you move closer to the coast, there is the Cache Creek terrane and the Stikine terrane. Then comes the Insular terrane group. This includes the Alexander terrane and the Wrangellia terrane, which make up the coast and islands of British Columbia. [Return to Figure 21.4.2]

Figure 21.4.8 **image description:** Cretaceous rocks cover a lot of the southwest coast of BC and a bit in the north. Jurassic rocks can be found on the northwest coast of BC, scattered sparsely through the interior, and a bit on Haida Gwaii, and Vancouver Island. [Return to Figure 21.4.8]

Figure 21.4.10 *image description: The age and location of sedimentary rocks in the Western Canada Sedimentary Basin*

Period/ Era/ Epoch	Age (Ma)	Location of sedimentary rocks
Paleozoic Era	540 to 251	• [Return to Figure 21.4.10] **Figure** 21.4.11 **image description:** At the Rocky Mountian foothills, the Paleozoic sedimentary rocks are buried over 4500 meters under Mesozoic sedimentary rocks. At the Alberta-Saskatchewan border, the Mesozoic layers are much thinner, so the Paleozoic sedimentary rocks are much nearer to the surface. [Return to Figure 21.4.11]

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21.5: Western Canada During the Cenozoic

Two additional relatively small terranes collided with North America early in the Cenozoic. At around 55 Ma metamorphosed sedimentary and volcanic rocks of the Pacific Rim Terrane were forced a few tens of kilometers underneath the west coast of Vancouver Island (Figure 21.5.1).

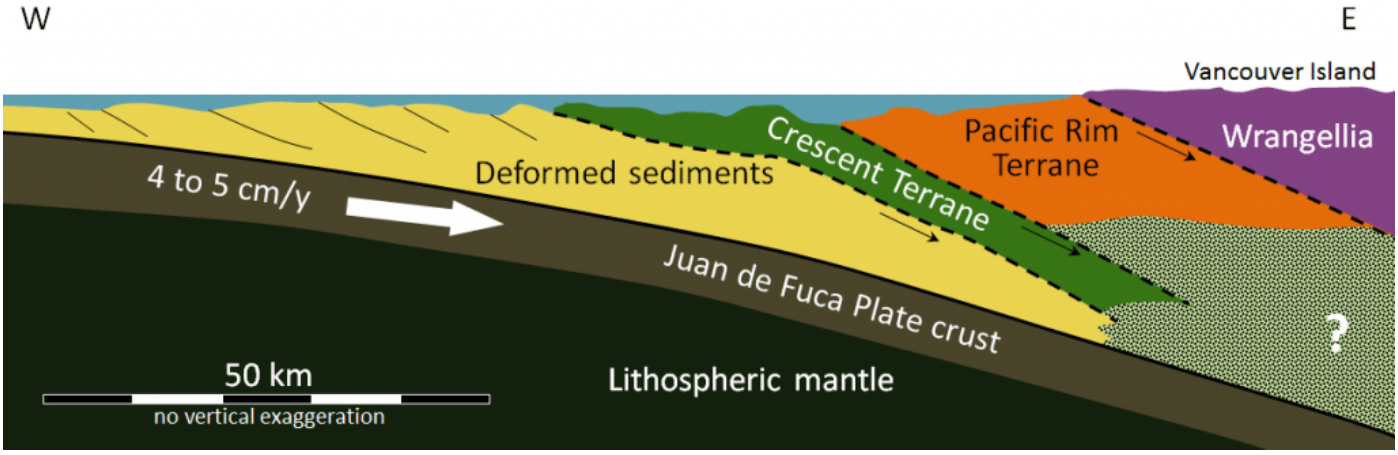


Figure 21.5.1 East-west cross-section showing the accretion of the Pacific Rim and Crescent Terranes beneath Vancouver Island, and the ongoing subduction of the Juan de Fuca Plate. The dashed lines are inactive faults.

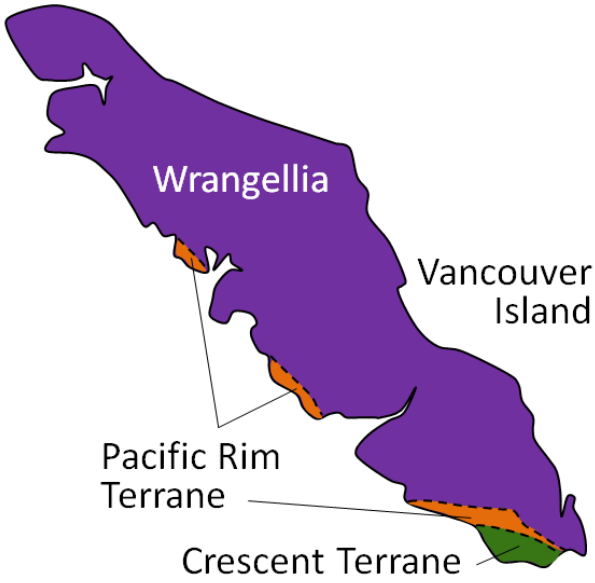


Figure 21.5.2 The distribution of Pacific Rim and Crescent Terrane rocks on Vancouver Island.

These rocks are distributed along the west coast of the island and in the area around Victoria (Figure 21.5.2). At around 42 Ma, sea-floor pillow basalt and gabbro of the Crescent Terrane was accreted to the southern margin of Vancouver Island and also to the adjacent part of Washington State (Figuer 21.30). These two terranes are shown as “Outboard terranes” in Figure 21.5.2.

The accretion of the Pacific Rim and Crescent Terranes had the effect of pushing Vancouver Island closer to the North American mainland, resulting in the uplift of the sediments deposited within the Nanaimo Basin to form islands in the Strait of Georgia (Figure 21.5.3) and mountains on Vancouver Island.



Figure 21.5.3 The Geoffrey Formation of the Nanaimo Group on Ruxton Island, B.C.



Figure 21.5.4 The current plate situation along the western edge of northern North America. Blue lines are divergent boundaries, red lines are transform boundaries, and black lines with teeth are subduction boundaries. The dark red triangles are volcanoes.

Following these events, the subduction of the Juan de Fuca Plate, which is a remnant of the former, much larger, Farallon Plate, was re-established at its current location farther to the west of Vancouver Island. This subduction—and that of the North America Plate beneath Alaska—has produced recently active volcanoes in Alaska, and all along the west coast from southwestern B.C. to northern California (Figure 21.5.4). In southwestern B.C. there are several dormant volcanoes of Pleistocene age (including Garibaldi and Meager) that trend along a line that also passes through Mt. Baker in Washington State. About 40 km to the east is a trend of slightly older igneous complexes (Pliocene to Oligocene). The displacement between these belts could be explained by a westward shift in the position of the subduction zone over that time period. The subduction and transform boundaries along this coast also generate relatively frequent earthquakes throughout this region, as illustrated in Exercise 11.1.

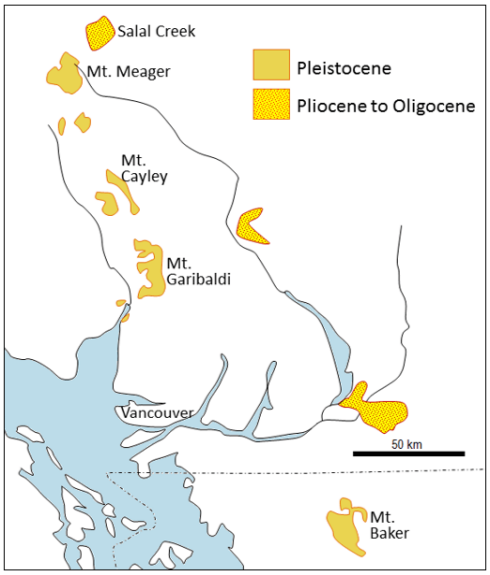


Figure 21.5.5 Oligocene to Pleistocene igneous complexes and volcanoes in southwestern B.C. and adjacent Washington.

Sedimentation in the Western Canada Sedimentary Basin continued into the Cenozoic (Figure 21.5.10) with deposition of the Paskapoo Formation adjacent to the Rockies in Alberta (Figure 21.5.6), the Ravenscrag Formation in the Cypress Hills of southern Alberta and Saskatchewan, and the Turtle Hills Formation in southern Manitoba. All of these strata were deposited in terrestrial fluvial and deltaic environments, and all of them include coal deposits. Numerous mammalian (and other) fossils have been found in these rocks in Alberta and Saskatchewan. The mammals include primitive ungulates (ancestors to the deer and their relatives), a type of pangolin, a colugo (a gliding mammal that was possibly a primate ancestor), and some true primates in the suborder Plesiadapiformes, which became extinct and are not ancestors to any modern primates.



Figure 21.5.6 The Paskapoo Formation exposed on the banks of the Red Deer River, Alberta.

Exercise 21.5 The volume of the Paskapoo Formation

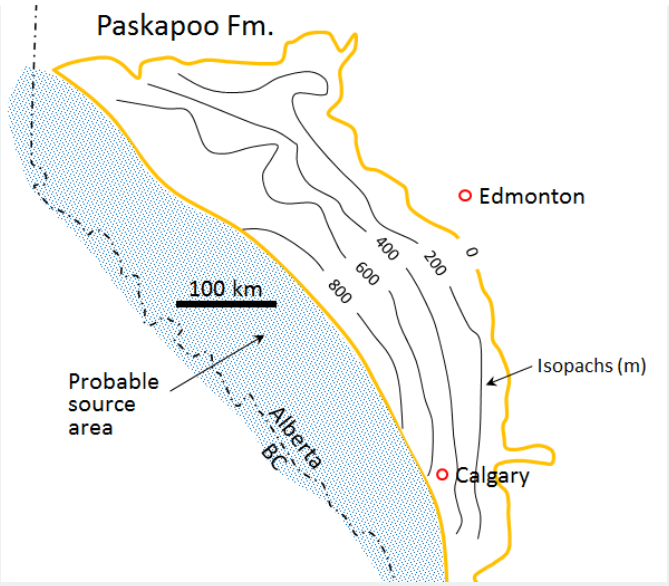


Figure 21.5.7

The area underlain by the Paleocene Paskapoo Formation is outlined in yellow on the map shown here. The Paskapoo ranges up to about 1,000 m thick, and contours of its thickness (known as isopachs) are shown. The average thickness is about 500 m, and the area covered by the formation is about 90,000 km^2. This means that the rock has a volume of about 45,000 km^3. The sediments of the Paskapoo were derived from the Paleocene Rocky Mountains, within the area shown in blue, which is about 60,000 km^2.

1. What average depth of erosion would have been required within the source area to produce the 45,000 km^3 of sediment, assuming that all of the eroded sediment ended up in the Paskapoo Formation?
2. The Paskapoo was deposited over a period of 4 million years from 62.5 Ma to 58.5 Ma. Assuming an average thickness of 500 m, what was the average rate of deposition (in mm/year) over that period?

See Appendix 3 for Exercise 21.5 answers.



Figure 21.5.8 The Tranquille Formation at the McAbee fossil site west of Kamloops, B.C.

Rocks younger than Paleocene (i.e., younger than 55 Ma) are relatively rare across the prairies, but there are widespread Eocene-aged volcanic and sedimentary rocks in central and southern B.C. The Kamloops Group includes the Tranquille Formation, made up of lacustrine (lake deposited) sediments, overlain by the Dewdrop Flats Formation of basaltic and andesitic volcanic flows and breccias. The Tranquille Formation includes the McAbee Beds and a number of other important sites with Eocene fossils (Figure 21.5.8).

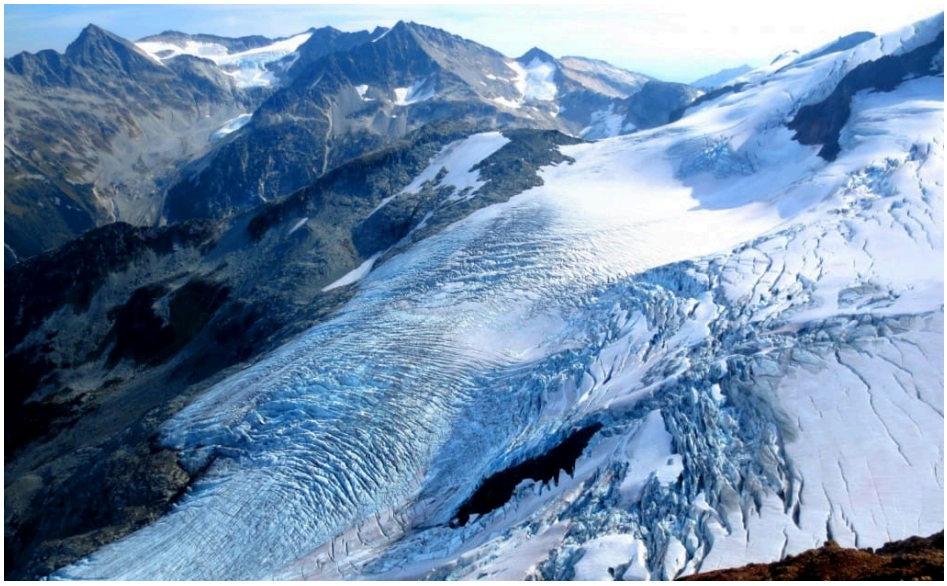


Figure 21.5.9 Various glacial erosion features (top left) and glaciation at the Overlord Glacier, Coast Mountains, B.C.



Figure 21.5.10 A drumlin field with an esker (centre) in the Cree Lake area of northern Saskatchewan.

The earliest Pleistocene glaciation in Canada started at about 2.64 Ma (late Pliocene) in the Klondike area of Yukon. This was part of the Cordilleran Ice Sheet. The Laurentide Ice Sheet started to form shortly afterward, and within 200,000 years had covered a large part of Canada and extended well into the United States. The Pleistocene glaciations had a major impact on the topography and geology of western Canada, creating extraordinary glacial erosion features in the mountainous regions of the west (Figure 21.5.9), and leaving enormous volumes of glacial sediment and glacial depositional features throughout the region (Figure 21.5.10).

Image Descriptions

Figure 21.5.4 image description: The Juan de Fuca Plate lies between the Pacific and the North America plates along the south

coast of British Columbia and the northwest coast of the United States. It is subducting beneath the North America Plate to form mountains along the west coast of North America. The Pacific Plate and the Juan de Fuca Plate are pulling apart to create divergent and transform boundaries. The Pacific Plate and the North America Plate have a transform boundary along the north coast of British Columbia, but the Pacific Plate is subducting beneath the North America Plate along the Alaskan coast. [Return to Figure 21.5.4]

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21.6: Summary

The main topics of this chapter can be summarized as follows:

Section	Summary
21.1 Geological History of Canada	The continent Laurentia, which includes what is now the Canadian Shield, was formed through the assembly of a number of smaller continents over the period from around 4 Ga to 1 Ga. Over the past 650 Ma, Laurentia moved north from deep within the southern hemisphere. During that time, a number of important plate-tectonic events have taken place, including formation of the Appalachian and Innuitian fold belts, sedimentation within the interior of the continent, and terrane accretion and mountain formation along the west coast.
21.2 Western Canada during the Precambrian	The two oldest parts of the Canadian Shield in western Canada are the Slave and Superior Cratons, and both include some of Earth's oldest rocks. During the formation of Laurentia, these cratons were combined with the Rae and Hearn Cratons, with the collision zones now represented by the Trans-Hudson Orogen and the Taltson Magmatic Zone. Continental sediments accumulated in Saskatchewan and Nunavut at around 1,700 Ma and in the area of the southern B.C.-Alberta border at around 1,400 Ma. This region was rifted apart during the breakup of the supercontinent Columbia, after which sedimentation continued on the western margin of Laurentia with the deposition of the Windermere Group.
21.3 Western Canada during the Paleozoic	Sedimentation on the west coast of North America continued into the early Paleozoic, and by the Ordovician, the Western Canada Sedimentary Basin (WCSB) had developed, extending from southern Manitoba to the northern Northwest Territories. Most of the Paleozoic rocks in this basin, which range in age up to the Carboniferous, have marine affinities, although there are important evaporites as well. Sedimentation continued on the west coast through this time, but by the late Paleozoic, a subduction boundary had developed along the coast and small continents were moving toward North America.
21.4 Western Canada during the Mesozoic	The various parts of the Intermontane Superterrane began colliding with the west coast of North America during the Jurassic (~180 Ma). This started the building of the Rocky Mountains by the thrusting of existing sedimentary rocks toward the east. The arrival of the Insular Superterrane during the Cretaceous (~90 Ma) contributed to further thrusting and uplift of the Rockies, creating a significant source of sediments for the WCSB. The greatest volume of Mesozoic rocks in the basin are of Upper Cretaceous age, and that likely coincides with the period of maximum collision-related uplift of the Rockies. Subduction of oceanic crust at various locations along the west coast and within the accreted terranes prior to their arrival has produced massive volumes of intrusive igneous rocks within the Coast Range.
21.5 Western Canada During the Cenozoic	The Pacific Rim and Crescent Terranes were added to the western edge of Vancouver Island during the Paleogene, pushing the island closer to the mainland and forcing recently deposited Nanaimo Group rocks onto the island. Continuing subduction along the coast has generated ongoing volcanism and earthquake activity in southwestern B.C. Sedimentation continued in the WCSB into the Cenozoic, especially in the Paleocene with deposition of the terrestrial Paskapoo Formation in Alberta and similar rocks in southern Saskatchewan.

Questions for Review

See Appendix 2 for answers to Review Questions.

1. What are the oldest parts of Laurentia?
2. The five main geological regions of Canada are shown on Figure A. Name the regions A through E.

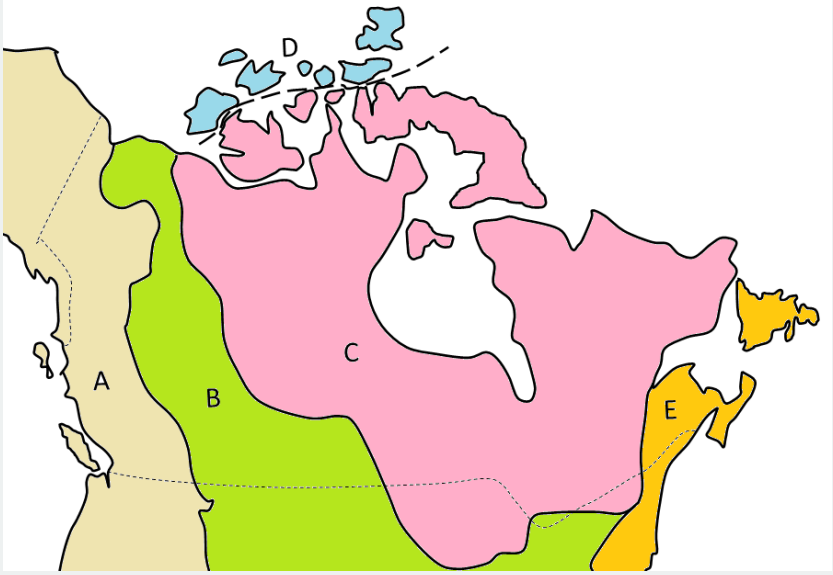


Figure A

- Which ancient continent collided with North America to form the Innuitian fold belt, and when did that take place?
- Explain why the ancient sedimentary rocks of the Athabasca and Thelon Basins are generally unmetamorphosed and undeformed.
- Explain why ultramafic intrusions, like those of the Muskox Intrusion, are relatively common in Archean rocks, but rare in Phanerozoic rocks.
- Use the Internet to find out why Cambrian marine organisms are so well preserved in the rocks of the Burgess Shale of British Columbia.
- The Prairie Evaporite Formation overlies marine carbonate rocks of the Winnipegosis Formation and is overlain in turn by marine carbonate rocks of the Dawson Bay Formation. What type of changes might have led to the accumulation of evaporites during this period of marine deposition?
- What features of the Intermontane Superterrane have been used to indicate that these rocks formed south of the equator?
- What is the connection between terrane accretion on the west coast and the relatively rapid accumulation of sediments within the WCSB?
- Why is the WCSB considered to be a foreland basin during the Mesozoic?
- The four main terranes of the Intermontane Superterrane are Cache Creek, Quesnel, Stikine, and Yukon-Tanana. Referring to Figure 21.6.2, determine the order in which these terranes are likely to have reached North America.
- The presence of Nanaimo Group sedimentary rocks far inland and at relatively high elevations on Vancouver Island is attributed to the accretion of the Pacific Rim and Crescent Terranes. What is the likely connection?
- Referring to the diagram in Exercise 21.5, explain why the Paskapoo Formation gets thinner toward the northeast.

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22: THE ORIGIN OF EARTH AND THE SOLAR SYSTEM

22.1: Starting with a Big Bang
22.2: Forming Planets from the Remnants of Exploding Stars
22.3: How to Build a Solar System
22.4: Earth's First 2 Billion Years
22.5: Are There Other Earths?
22.6: Summary

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22.1: Starting with a Big Bang

According to the **big bang theory**, the universe blinked violently into existence 13.77 billion years ago (Figure 22.1.1). The big bang is often described as an explosion, but imagining it as an enormous fireball isn't accurate. The big bang involved a sudden expansion of matter, energy, and space from a single point. The kind of Hollywood explosion that might come to mind involves expansion of matter and energy *within* space, but during the big bang, space *itself* was created.

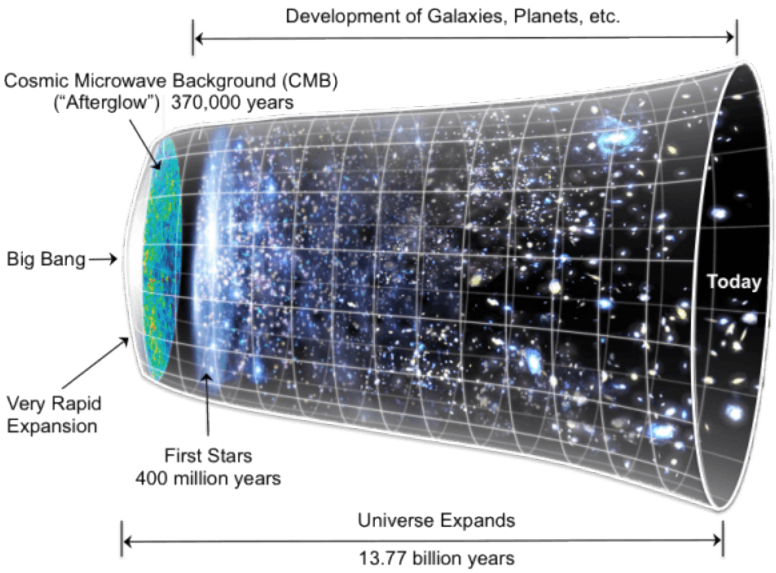


Figure 22.1.1 The big bang. The universe began 13.77 billion years ago with a sudden expansion of space, matter, and energy, and it continues to expand today.

At the start of the big bang, the universe was too hot and dense to be anything but a sizzle of particles smaller than atoms, but as it expanded, it also cooled. Eventually some of the particles collided and stuck together. Those collisions produced hydrogen and helium, the most common elements in the universe, along with a small amount of lithium.

You may wonder how a universe can be created out of nothing, or how we can know that the big bang happened at all. Creating a universe out of nothing is mostly beyond the scope of this chapter, but there is a way to think about it. The particles that make up the universe have opposites that cancel each other out, similar to the way that we can add the numbers 1 and –1 to get zero (also known as "nothing"). As far as the math goes, having zero is exactly the same as having a 1 and a –1. It is also exactly the same as having a 2 and a –2, a 3 and a –3, two –1s and a 2, and so on. In other words, *nothing* is really the potential for *something* if you divide it into its opposite parts. As for how we can know that the big bang happened at all, there are very good reasons to accept that it is indeed how our universe came to be.

Looking back to the early stages of the big bang

The notion of seeing the past is often used metaphorically when we talk about ancient events, but in this case it is meant literally. In our everyday experience, when we watch an event take place, we perceive that we are watching it as it unfolds in real time. In fact, this isn't true. To see the event, light from that event must travel to our eyes. Light travels very rapidly, but it does not travel instantly. If we were watching a digital clock 1 m away from us change from 11:59 a.m. to 12:00 p.m., we would actually see it turn to 12:00 p.m. three billionths of a second after it happened. This isn't enough of a delay to cause us to be late for an appointment, but the universe is a very big place, and the "digital clock" in question is often much, much farther away. In fact, the universe is so big that it is convenient to describe distances in terms of **light years**, or the distance light travels in one year. What this means is that light from distant objects takes so long to get to us that we see those objects as they were at some considerable time in the past. For example, the star Proxima Centauri is 4.24 light years from the sun. If you viewed Proxima Centauri from Earth on January 1, 2015, you would actually see it as it appeared in early October 2010.

We now have tools that are powerful enough to look deep into space and see the arrival of light from early in the universe's history.

Astronomers can detect light from approximately 375,000 years after the big bang is thought to have occurred. Physicists tell us that if the big bang happened, then particles within the universe would still be very close together at this time. They would be so close that light wouldn't be able to travel far without bumping into another particle and getting scattered in another direction. The effect would be to fill the sky with glowing fog, the "afterglow" from the formation of the universe (Figure 22.1.1).

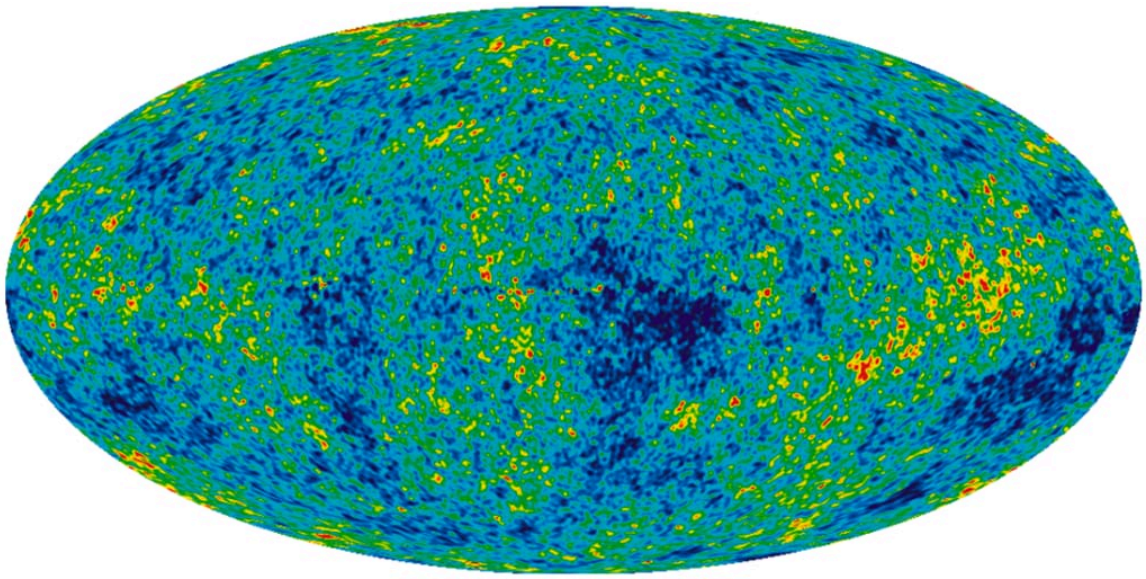


Figure 22.1.2 Cosmic microwave background (CMB) map of the sky, a baby picture of the universe. The CMB is light from 375,000 years after the big bang. The colors reveal variations in density. Red patches have the highest density and blue patches have the lowest density. Regions of higher density eventually formed the stars, planets, and other objects we see in space today.

In fact, this is exactly what we see when we look at light from 375,000 years after the big bang. The fog is referred to as the **cosmic microwave background** (or CMB), and it has been carefully mapped throughout the sky (Figure 22.1.2). The map displays the cosmic microwave background as temperature variations, but these variations translate to differences in the density of matter in the early universe. The red patches are the highest density regions and the blue patches are the lowest density. Higher density regions represent the eventual beginnings of stars and planets. The map in Figure 22.1.2 has been likened to a baby picture of the universe.

The big bang is still happening, and we can see the universe expanding

The expansion that started with the big bang never stopped. It continues today, and we can see it happening by observing that large clusters of billions of stars, called **galaxies**, are moving away from us. (The exception is the Andromeda galaxy with which we are on a collision course.) The astronomer Edwin Hubble came to this conclusion when he observed that the light from other galaxies was red-shifted. The **red shift** is a consequence of the Doppler effect. This refers to how we see waves when the object that is creating the waves is moving toward us or away from us.

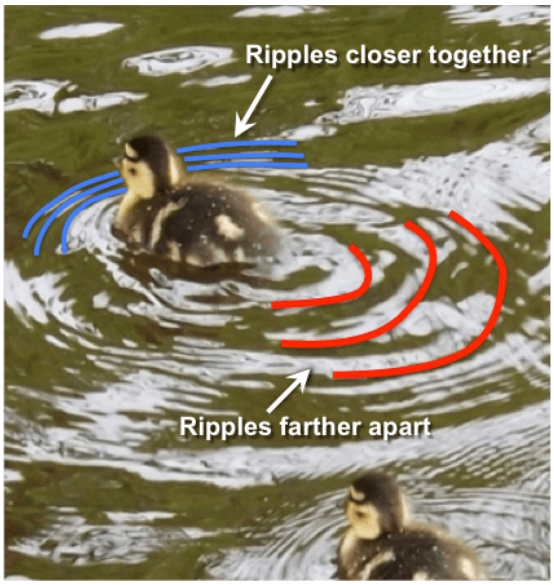


Figure 22.1.3 Duckling illustrates the Doppler effect in water. The ripples made in the direction the duckling is moving (blue lines) are closer together than the ripples behind the duckling (red lines).

Before we get to the Doppler effect as it pertains to the red shift, let's see how it works on something more tangible. The duckling swimming in Figure 22.1.3 is generating waves as it moves through the water. It is generating waves that move forward as well as back, but notice that the ripples ahead of the duckling are closer to each other than the ripples behind the duckling. The distance from one ripple to the next is called the **wavelength**. The wavelength is shorter in the direction that the duckling is moving, and longer as the duckling moves away.

When waves are in air as sound waves rather than in water as ripples, the different wavelengths manifest as sounds with different pitches—the short wavelengths have a higher pitch, and the long wavelengths have a lower pitch. This is why the pitch of a car's engine changes as the car races past you.

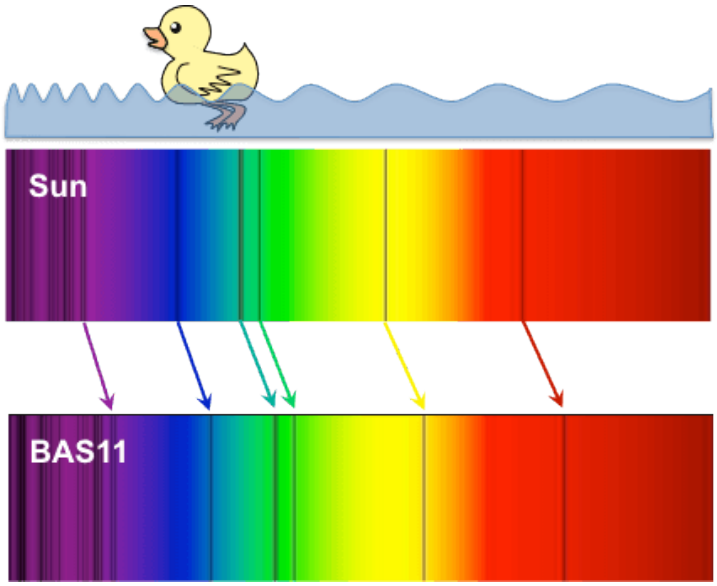


Figure 22.1.4 Red shift in light from the supercluster BAS11 compared to the sun's light. Black lines represent wavelengths absorbed by atoms (mostly H and He). For BAS11 the black lines are shifted toward the red end of the spectrum compared to the Sun.

For light waves, wavelength translates to color (Figure 22.1.4). In the spectrum of light that we can see, shorter wavelengths are on the blue end of the spectrum, and longer wavelengths are on the red end of the spectrum. Does this mean that galaxies look red because they are moving away from us? No, but the color we see is shifted toward the red end of the spectrum and longer wavelengths.

Notice that the sun's spectrum in the upper part of Figure 22.1.4 has some black lines in it. The black lines are there because some colors are missing in the light we get from the Sun. Different elements absorb light of specific wavelengths, and many of the black lines in Figure 22.1.4 represent colors that are absorbed by hydrogen and helium within the Sun. This means the black lines are like a bar code that can tell us what a star is made of. The lower spectrum in Figure 22.1.4 is the light coming from BAS11, an enormous cluster of approximately 10,000 galaxies located 1 billion light years away. The black lines represent the same elements as in the Sun's spectrum, but they are shifted to the right toward the red end of the spectrum because BAS11 is moving away from us as the universe continues to expand. So to summarize, because almost all of the galaxies we can see have light that is red-shifted, it means they are all moving away from us. In fact, the farther away they are, the faster they are going. This is evidence that the universe is still expanding.

Media Attributions

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22.2: Forming Planets from the Remnants of Exploding Stars

If we were to take an inventory of the elements that make up Earth, we would find that 95% of Earth's mass comes from only four elements: oxygen, magnesium, silicon, and iron. Most of the remaining 5% comes from aluminum, calcium, nickel, hydrogen, and sulfur. We know that the big bang made hydrogen, helium, and lithium, but where did the rest of the elements come from?

The answer is that the other elements were made by stars. Sometimes stars are said to "burn" their fuel, but burning is not at all what is going on within stars. The burning that happens when wood in a campfire is turned to ash and smoke is a chemical reaction—heat causes the atoms that were in the wood and in the surrounding atmosphere to exchange partners. Atoms group in different ways, but the atoms themselves do not change. What stars do is change the atoms. The heat and pressure within stars cause smaller atoms to smash together and fuse into new, larger atoms. For example, when hydrogen atoms smash together and fuse, helium is formed. Large amounts of energy are released when some atoms fuse and that energy is what causes stars to shine.

It takes larger stars to make elements as heavy as iron and nickel. Our Sun is an average star; after it uses up its hydrogen fuel to make helium, and then some of that helium is fused to make small amounts of beryllium, carbon, nitrogen, oxygen, and fluorine, it will be at the end of its life. It will stop making atoms and will cool down and bloat until its middle reaches the orbit of Mars. In contrast, large stars end their lives in spectacular fashion, exploding as supernovae and casting off newly formed atoms—including the elements heavier than iron—into space. It took many generations of stars creating heavier elements and casting them into space before heavier elements were abundant enough to form planets like Earth.

Until recently, astronomers have only been able to see stars that already contain heavier elements in small amounts, but not the first-generation stars that started out before any of the heavier elements were produced. That changed in June of 2015 when it was announced that a distant galaxy called CR7 had been found that contained stars made only of hydrogen and helium. The galaxy is so far away that it shows us a view of the universe from only 800 million years after the big bang.[1]

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1. Read more about the discovery of CR7 here: "Astronomers spot first-generation stars, made from big bang," (http://bit.ly/CR7galaxy) ↩

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22.3: How to Build a Solar System

A **solar system** consists of a collection of objects orbiting one or more central stars. All solar systems start out the same way. They begin in a cloud of gas and dust called a **nebula**. Nebulae are some of the most beautiful objects that have been photographed in space, with vibrant colors from the gases and dust they contain, and brilliant twinkling from the many stars that have formed within them (Figure 22.3.1). The gas consists largely of hydrogen and helium, and the dust consists of tiny mineral grains, ice crystals, and organic particles.



Figure 22.3.1 Photograph of a nebula. The Pillars of Creation within the Eagle Nebula viewed in visible light (left) and near infrared light (right). Near infrared light captures heat from stars and allows us to view stars that would otherwise be hidden by dust. This is why the picture on the right appears to have more stars than the picture on the left.

Step 1: Collapse a nebula

A solar system begins to form when a small patch within a nebula (small by the standards of the universe, that is) begins to collapse upon itself. Exactly how this starts isn't clear, although it might be triggered by the violent behaviour of nearby stars as they progress through their life cycles. Energy and matter released by these stars might compress the gas and dust in nearby neighbourhoods within the nebula.

Once it is triggered, the collapse of gas and dust within that patch continues for two reasons. One of those reasons is that gravitational force pulls gas molecules and dust particles together. But early in the process, those particles are very small, so the gravitational force between them isn't strong. So how do they come together? The answer is that dust first accumulates in loose clumps for the same reason dust bunnies form under your bed: static electricity. Given the role of dust bunnies in the early history of the solar system, one might speculate that an accumulation of dust bunnies poses a substantial risk to one's home (Figure 22.3.2). In practice, however, this is rarely the case.

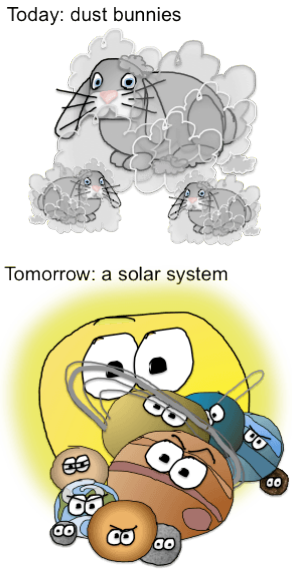


Figure 22.3.2 Public service announcement. If you don't think housekeeping is important, then you don't understand the gravity of the situation.

Step 2: Make a disk and put a star at its centre

As the small patch within a nebula condenses, a star begins to form from material drawn into the centre of the patch, and the remaining dust and gas settle into a disk that rotates around the star. The disk is where planets eventually form, so it's called a **protoplanetary disk**. In Figure 22.3.3 the image in the upper left shows an artist's impression of a protoplanetary disk, and the image in the upper right shows an actual protoplanetary disk surrounding the star HL Tauri. Notice the dark rings in the protoplanetary disk. These are gaps where planets are beginning to form. The rings are there because incipient planets are beginning to collect the dust and gas in their orbits. There is an analogy for this in our own solar system, because the dark rings are akin to the gaps in the rings of Saturn (Figure 22.3.3, lower left), where moons can be found (Figure 22.3.3, lower right).

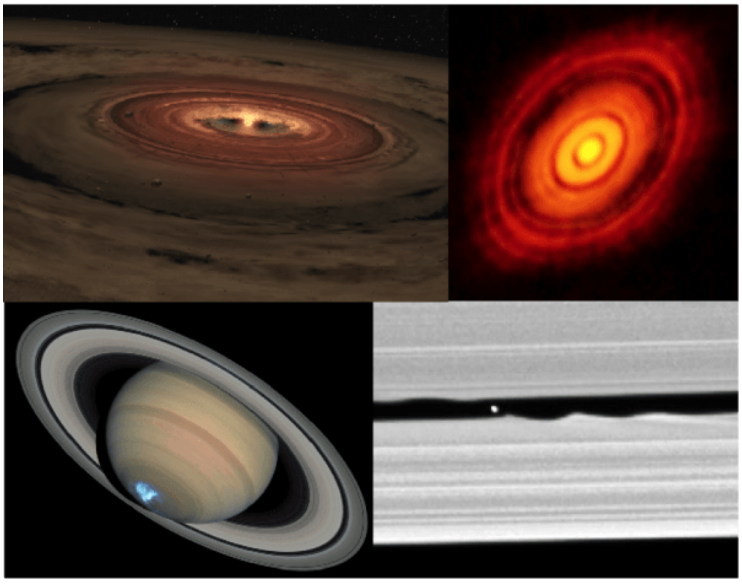


Figure 22.3.3 Protoplanetary disks and Saturn's rings. Upper left: An artists impression of a protoplanetary disk containing gas and dust, surrounding a new star. Upper right: A photograph of the protoplanetary disk surrounding HL Tauri. The dark rings within the disk are thought to be gaps where newly forming planets are sweeping up dust and gas. Lower left: A photograph of Saturn showing similar gaps within its rings. The bright spot at the bottom is an aurora, similar to the northern lights on Earth. Lower right: a close-up view of a gap in Saturn's rings showing a small moon as a white dot.

Step 3: Build some planets

In general, planets can be classified into three categories based on what they are made of (Figure 22.3.4). **Terrestrial planets** are those planets like Earth, Mercury, Venus, and Mars that have a core of metal surrounded by rock. **Jovian planets** (also called **gas giants**) are those planets like Jupiter and Saturn that consist predominantly of hydrogen and helium. **Ice giants** are planets such as Uranus and Neptune that consist largely of water ice, methane (CH_4) ice, and ammonia (NH_3) ice, and have rocky cores. Often, the ice giant planets Uranus and Neptune are grouped with Jupiter and Saturn as gas giants; however, Uranus and Neptune are very different from Jupiter and Saturn.

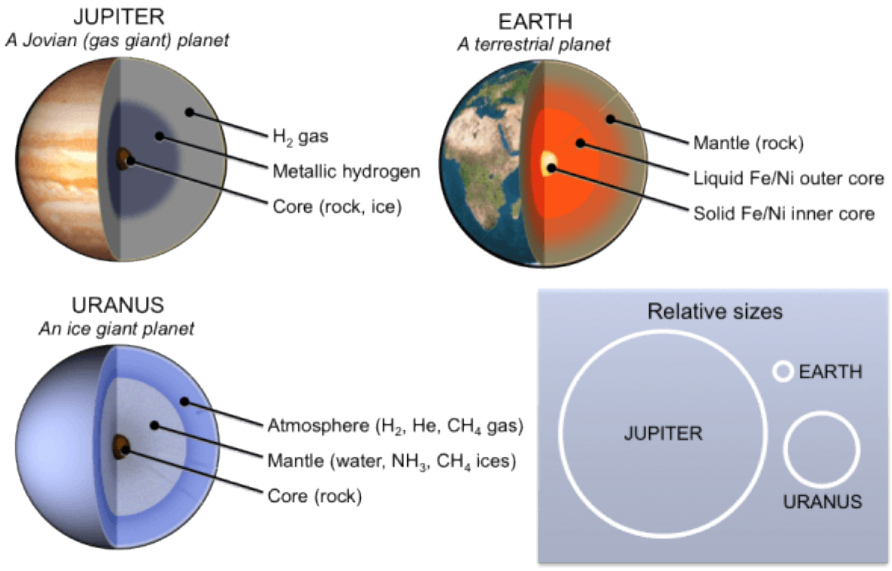


Figure 22.3.4 Three types of planets. Jovian (or gas giant) planets, such as Jupiter, consist mostly of hydrogen and helium. They are the largest of the three types. Ice giant planets, such as Uranus, are the next largest. They contain water, ammonia, and methane ice. Terrestrial planets such as Earth are the smallest, and they have metal cores covered by rocky mantles.

These three types of planets are not mixed together randomly within our solar system. Instead they occur in a systematic way, with terrestrial planets closest to the sun, followed by the Jovian planets and then the ice giants (Figure 22.3.5). Smaller **solar system** objects follow this arrangement as well. The **asteroid belt** contains bodies of rock and metal. Bodies ranging from meters to hundreds of meters in diameter are classified as **asteroids**, and smaller bodies are referred to as **meteoroids**. In contrast, the **Kuiper belt** (*Kuiper* rhymes with *piper*), and the **Oort cloud** (*Oort* rhymes with *sort*), which are at the outer edge of the solar system, contain bodies composed of large amounts of ice in addition to rocky fragments and dust. (We will talk more about smaller solar system objects in a moment.)

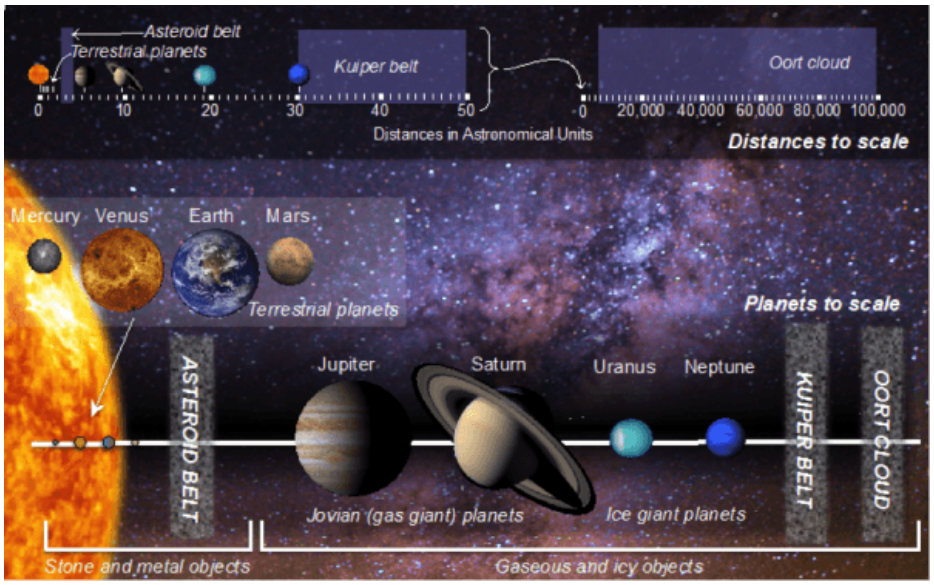


Figure 22.3.5 Our solar system. Top: The solar system shown with distances to scale. Distances are in astronomical units (AU), where 1 AU is the average distance from Earth to the Sun. The edge of the Kuiper belt extends to 50 AU (7.5 billion km), but this distance is minuscule compared to the size of the solar system as a whole, which extends to the edge of the Oort cloud, thought to be 15 trillion km away. Bottom: Solar system with the Sun and planets to scale. The gas giants are the largest planets, followed by the ice giants, and then the terrestrial planets. Note that the planets in this diagram likely do not reflect the entire population of planets in our solar system because evidence suggests that large planets are present beyond the Kuiper belt.

Part of the reason for this arrangement is the **frost line** (also referred to as the **snow line**). The frost line separated the inner part of the protoplanetary disk closer to the sun, where it was too hot to permit anything but silicate minerals and metal to crystalize, from the outer part of the disk farther from the Sun, where it was cool enough to allow ice to form. As a result, the objects that formed in the inner part of the protoplanetary disk consist largely of rock and metal, while the objects that formed in the outer part consist largely of gas and ice. The young sun blasted the solar system with raging **solar winds** (winds made up of energetic particles), which helped to drive lighter molecules toward the outer part of the protoplanetary disk.

The objects in our solar system formed by **accretion**. Early in this process, particles collected in fluffy clumps because of static electricity. As the clumps grew larger, gravity became more important and collected clumps into solid masses, and solid masses into larger and larger bodies. If you were one of these bodies in the early solar system, and participating in the accretion game with the goal of becoming a planet, you would have to follow some key rules:

- Keep your velocity just right. If you move too fast and collide with another body, you both smash up and have to start again. If you move slowly enough, gravity will keep you from bouncing off each other and you can grow larger.
- Your distance from the Sun will determine how big you can get. If you are closer, there is less material for you to collect than if you are farther away.
- To begin with, you can only collect mineral and rock particles. You have to grow above a certain mass before your gravity is strong enough to hang onto gas molecules, because gas molecules are very light.
- As your mass increases, your gravity becomes stronger and you can grab material from farther away. The bigger you are, the faster you grow.

You would also have to watch out for some dangers:

- In the early stages of the game, the protoplanetary disk is turbulent, and you and other objects can get thrown into different orbits or at each other. This might be a good thing, or it might not, depending on how the rules above apply to you.
- If the game progresses to the point where there is no more material within your reach and you are not yet a planet, then it's game over.
- If you slow down too much (e.g., from bumping into other objects), you could spiral into the Sun (game over).
- If another planet gets big enough, it can:
  - Rip you apart and then swing the pieces around so fast that for the rest of the game you collide too hard with other pieces to grow any bigger (game over)
  - Fling you out of the solar system (game over)
  - Grab you for itself (game over)

- Trap you in an orbit around it, turning you into a moon (game over, and incredibly humiliating)

The outcome of the game is evident in Figure 22.3.5. Today eight official winners are recognized, with Jupiter taking the grand prize, followed closely by Saturn. Both planets have trophy cases with more than 60 moons each, and each has a moon that is larger than Mercury. Prior to 2006, Pluto was also counted a winner, but in 2006 a controversial decision revoked Pluto's planet status. The reason was a newly formalized definition of a planet, which stated that an object can only be considered a planet if it is massive enough to have swept its orbit clean of other bodies. Pluto is situated within the icy clutter of the Kuiper belt, so it does not fit this definition. Pluto's supporters have argued that Pluto should have been grandfathered in, given that the definition came after Pluto was declared a planet, but to no avail. Pluto has not given up, and on July 13, 2015, it launched an emotional plea with the help of the NASA's New Horizons probe. New Horizons sent back images of Pluto's heart (Figure 22.3.6). On closer inspection, Pluto's heart was discovered to be broken.

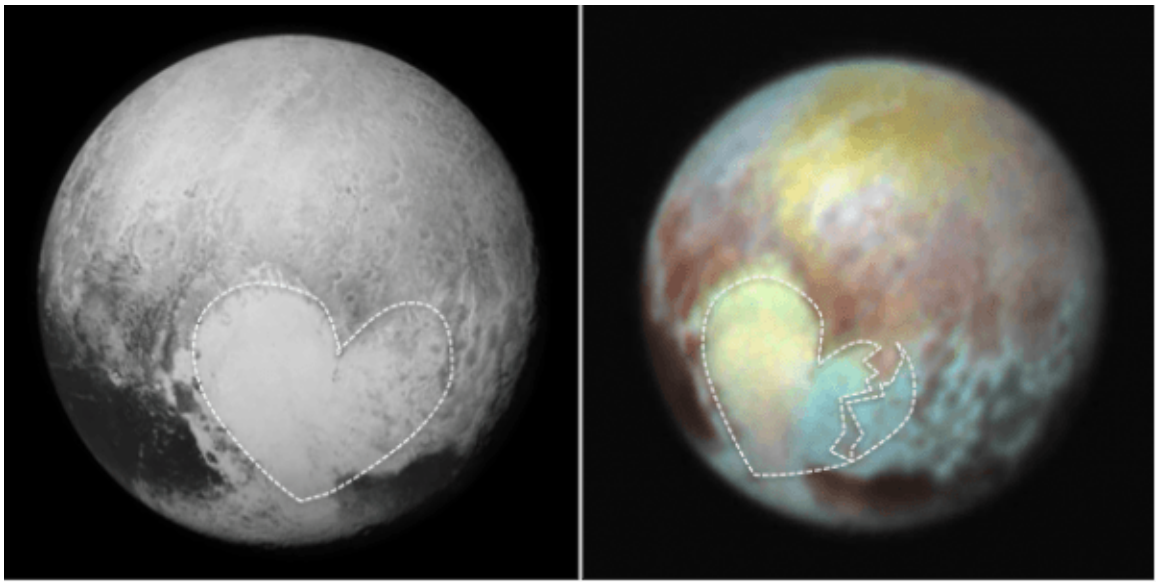


Figure 22.3.6 Photographs of Pluto. Left: The heart-shaped region called Tombaugh Regio is outlined. This region is named after Pluto's discoverer, Clyde Tombaugh.

The rules and dangers of the planet-forming game help to explain many features of our solar system today.

- Proximity to the Sun explains why the terrestrial planets are so much smaller than the gas giant and ice giant planets.
- Mars is smaller than it should be, given the rule that distance from the Sun determines how much material a body can accumulate, and this can be explained by its proximity to Jupiter. Jupiter's immense gravity interfered with Mars' ability to accrete. Further evidence of Jupiter's interference is the debris field that forms the asteroid belt. From time to time, Jupiter still flings objects from the asteroid belt out into other parts of the solar system, some of which have collided with Earth to catastrophic effect.
- The Kuiper belt is an icy version of the asteroid belt, consisting of fragments left over from the early solar system. The material in the Kuiper belt is scattered because of Neptune's gravity. From time to time, Jupiter interferes here as well, flinging Kuiper belt objects toward the Sun and into orbit. As these objects approach the Sun, the Sun causes dust and gas to be blasted from their surface, forming tails. We know these objects as comets.
- Comets may also come from the Oort cloud where gravitational forces from outside of the solar system can hurl objects from the Oort cloud toward the Sun.

Exercise 22.1 How do we know what other planets are like inside?

The densities of planets give us important clues about the planets' compositions. For example, in our solar system, Earth (a terrestrial planet) has a density of 5.51 grams per cubic centimetre (g/cm^3), but Jupiter (a gas giant) has a density of 1.33 g/cm^3. We can also use density to determine something about the interior structures of planets. In this exercise, you will determine how much of each terrestrial planet is made up of core, and translate that result to a diagram for easy comparison.

It is useful to approximate the structure of a terrestrial planet as having two parts: a metal core and a rocky mantle. If we know the density of the planet as a whole, and the densities of the materials making up the rocky mantle and the core, we can find out how much of the planet is core and how much is rocky. The density of the planet is the sum of the percent having the density of the core and the percent having the density of rock. This can be written as follows:

planet density = % core/100 x core density + (1– %core ÷ 100) × rock density

Rearranging the equation gives us:

% core = (planet density – rock density)/ (core density – rock density) × 100

Step 1.

Find the percent core for each of the terrestrial planets using the data in Tables 22.1 and 22.2. For our calculations, the planet density will be the **uncompressed density** of the planet. Uncompressed density is the density after removing the effects of gravity squeezing the planet together. (Notice that the density we mentioned for Earth is 5.51 g/cm^3, but Earth's uncompressed density is only 4.05 g/cm^3.) The first one is done for you.

Table 22.1 Core and mantle density from meteorites

Description	Density (g/cm^3)	Source	Why?
Core density	8.00	iron meteorites	Iron meteorites come from the cores of broken up asteroids and planets and approximate what the density of Earth's core would be without gravitational squeezing.
Rocky mantle density	3.25	HED* stony meteorites	HED (Howardites, Eucrites, and Diogenites) meteorites come from the rocky mantles of asteroids and planets that have separated into mantle and core, and then broken up. These approximate what the density of Earth's mantle would be without gravitational squeezing.

*HED stands for the names of three types of meteorites: howardites, eucrites, and diogenites.

Table 22.2 Finding the fraction of volume that is core

Description	Earth	Mars	Venus	Mercury
Planet density (uncompressed) in g/cm^3	4.05	3.74	4.00	5.30
Percent core ((planet density – 3.25 g/cm^3) ÷ 4.75 g/cm^3) × 100	**16.8%**			

Step 2.

Once we have the percent of core, we can use it to find the volume of the core for each planet. The core volume is the percent of core times the volume of the planet. Use the planet volumes in Table 22.3 to calculate the core volume. Record your answers.

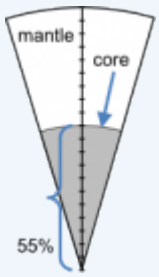
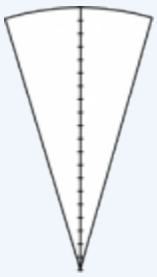
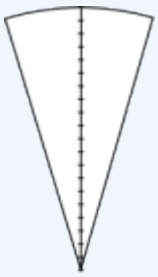
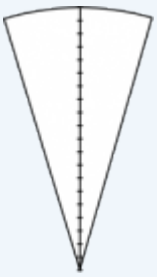
Table 22.3 Finding the volume of the core for each planet

Description	Earth	Mars	Venus	Mercury
Planet volume* in km^3	1.47 × 1012	1.72 × 1011	1.22 × 1012	6.23 × 1010
Core volume in in km^3 (% core ÷ 100) × planet volume	**2.48 × 1011**			

*Unsqueezed" values

Step 3. We can get the radius of the core from its volume by using the formula for the volume of a sphere (volume = 4 ÷ 3pr^3, where *r* is the radius). This calculation is done for you in Table 22.4. From these values, express each radius as a percentage of the total radius. To do this, divide the core radius by the planet radius and multiply by 100. Using your results, fill in the diagrams at the bottom of Table 22.4 by drawing in the boundary between the core and mantle.

Table 22.4 Finding the percent of each planet's radius that is core

Description	Earth	Mars	Venus	Mercury
Core radius* in km	3,900	1,617	3,581	1,858
Planet radius* in km	7,059	3,447	6,623	2,458
Percent of radius that is core: (core radius ÷ planet radius) × 100	**55%**			
Planet diagram Diagrams represent a wedge of the planet from surface to centre. The distance between each tick mark is 5% of the radius.				

One of the terrestrial planets is thought to have been involved in collisions that resulted in the permanent loss of a substantial amount of its mantle. You might be able to guess which one it is from the uncompressed densities of the planets. It should also be clear from your diagrams. Which planet is it?

See Appendix 3 for Exercise 22.1 answers.

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22.4: Earth's First 2 Billion Years

If you were to get into a time machine and visit Earth shortly after it formed (around 4.5 billion years ago), you would probably regret it. Large patches of Earth's surface would still be molten, which would make landing your time machine very dangerous indeed. If you happened to have one of the newer time-machine models with hovering capabilities and heat shields, you would still face the inconvenience of having nothing to breathe but a tenuous wisp of hydrogen and helium gas, and depending on how much volcanic activity was going on, volcanic gases such as water vapor and carbon dioxide. Some ammonia and methane might be thrown in just to make it interesting, but there would be no oxygen. Assuming you had the foresight to purchase the artificial atmosphere upgrade for your time machine, it would all be for naught if you materialized just in time to see an asteroid, or worse yet another planet, bearing down on your position. The moral of the story is that early Earth was a nasty place, and a time machine purchase is not something to take lightly.

Why was early Earth so nasty?

The early Earth was hot

Chapter 9 explains that Earth's heat comes from the decay of radioactive elements within Earth, as well as from processes associated with Earth's formation. Let's look more closely at how those formation processes heated up Earth:

- Heat came from the thermal energy already contained within the objects that accreted to form the Earth.
- Heat came from collisions. When objects hit Earth, some of the energy from their motion went into deforming Earth, and some of it was transformed into heat. Clap your hands vigorously to experience this on a much smaller (and safer!) scale.
- As Earth became larger, its gravitational force became stronger. This increased Earth's ability to draw objects to it, but it also caused the material making Earth to be compressed, rather like Earth giving itself a giant gravitational hug. Compression causes materials to heat up.

Heating had a very important consequence for Earth's structure. As Earth grew, it collected a mixture of silicate mineral grains as well as iron and nickel. These materials were scattered throughout Earth. That changed when Earth began to heat up: it got so hot that both the silicate minerals and the metals melted. The metal melt was much denser than the silicate mineral melt, so the metal melt sank to Earth's centre to become its core, and the silicate melt rose upward to become Earth's crust and mantle. In other words, Earth unmixed itself. The separation of silicate minerals and metals into a rocky outer layer and a metallic core, respectively, is called **differentiation**. The movement of silicate and metal melts within Earth caused it to heat up even more.

Earth's high temperature early in its history also means that early tectonic processes were accelerated compared to today, and Earth's surface was more geologically active.

Earth was heavily bombarded by objects from space

Although Earth had swept up a substantial amount of the material in its orbit as it was accreting, unrest within the solar system caused by changes in the orbits of Saturn and Jupiter was still sending many large objects on cataclysmic collision courses with Earth. The energy from these collisions repeatedly melted and even vaporized minerals in the crust, and blasted gases out of Earth's atmosphere. Very old scars from these collisions are still detectable, although we have to look carefully to see them. For example, the oldest impact site discovered is the 3 billion year old Maniitsoq "crater" in west Greenland, although there is no crater to see. What is visible are rocks that were 20 km to 25 km below Earth's surface at the time of the impact, but which nevertheless display evidence of deformation that could only be produced by intense, sudden shock.

The evidence of the very worst collision that Earth experienced is not subtle at all. In fact, you have probably looked directly at it hundreds of times already, perhaps without realizing what it is. That collision was with a planet named Theia, which was approximately the size of Mars (Figure 22.4.1). Not long after Earth formed, Theia struck Earth. When Theia slammed into Earth, Theia's metal core merged with Earth's core, and debris from the outer silicate layers was cast into space, forming a ring of rubble around Earth. The material within the ring coalesced into a new body in orbit around Earth, giving us our moon. Remarkably, the debris may have coalesced in 10 years or fewer! This scenario for the formation of the moon is called the **giant impact hypothesis.**

Earth's atmosphere as we know it took a long time to develop

Earth's first experiment with having an atmosphere didn't go well. It started out with a thin veil of hydrogen and helium gases that came with the material it accreted. However, hydrogen and helium are very light gases, and they bled off into space.

Earth's second experiment with having an atmosphere went much better. Volcanic eruptions built up the atmosphere by releasing gases. The most common volcanic gases are water vapor and carbon dioxide (CO_2), but volcanoes release a wide variety of gases. Other important contributions include sulfur dioxide (SO_2), carbon monoxide (CO), hydrogen sulfide (H_2S), hydrogen gas, and methane (CH_4). Meteorites and comets also brought substantial amounts of water and nitrogen to Earth. It is not clear what the exact composition of the atmosphere was after Earth's second experiment, but carbon dioxide, water vapor, and nitrogen were likely the three most abundant components.

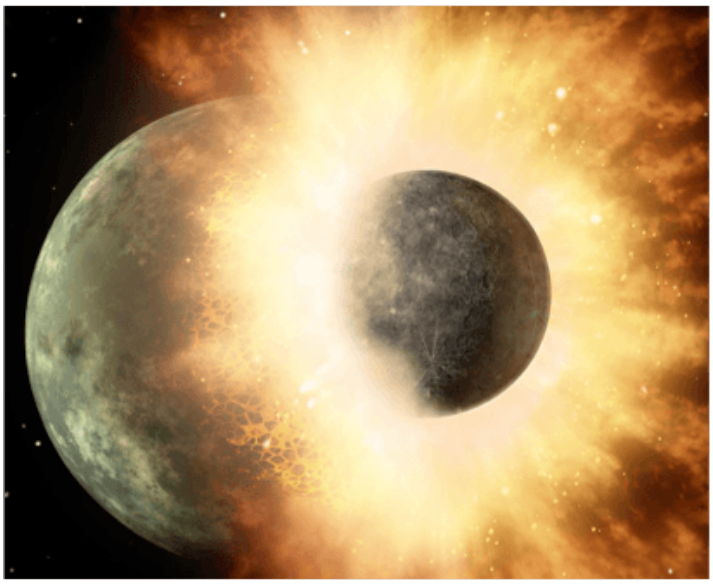


Figure 22.4.1 Artist's impression of a collision between planets. A similar collision between Earth and the planet Theia might have given us our Moon. Fortunately for us, the collision that gave us the moon was a glancing blow rather than the direct hit shown here. Earth might not have survived a direct hit.

One thing we can say for sure about Earth's second experiment is that there was effectively no free oxygen (O_2, the form of oxygen that we breathe) in the atmosphere. We know this in part because prior to 2 billion years ago, there were no sedimentary beds stained red from oxidized iron minerals. Iron minerals were present, but not in oxidized form. At that time, O_2 was produced in the atmosphere when the Sun's ultraviolet rays split water molecules apart; however, chemical reactions removed the oxygen as quickly as it was produced.

It wasn't until well into Earth's third experiment—life—that the atmosphere began to become oxygenated. Photosynthetic organisms used the abundant CO_2 in the atmosphere to manufacture their food, and released O_2 as a by-product. At first all of the oxygen was consumed by chemical reactions, but eventually the organisms released so much O_2 that it overwhelmed the chemical reactions and oxygen began to accumulate in the atmosphere, although present levels of 21% oxygen didn't occur until about 350 Ma. Today the part of our atmosphere that isn't oxygen consists largely of nitrogen (78%).

The oxygen-rich atmosphere on our planet is life's signature. If geologic process were the only processes controlling our atmosphere, it would consist mostly of carbon dioxide, like the atmosphere of Venus. It is an interesting notion (or a disconcerting one, depending on your point of view) that for the last 2 billion years the light reflected from our planet has been beaming a bar code out to the universe, similar to the ones in Figure 22.4.4, except ours says "oxygen." For 2 billion years, our planet has been sending out a signal that could cause an observer from another world to say, "That's odd... I wonder what's going on over there."

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22.5: Are There Other Earths?

If by that you mean, are there other planets where we could walk out of a spaceship with no equipment other than a picnic basket, and enjoy a pleasant afternoon on a grassy slope near a stream, then that remains to be seen. On the other hand, if you are asking if other planets exist that are rocky worlds approximately Earth's size, and orbiting within their star's **habitable zone** (the zone in which liquid water, and potentially life, can exist), then many planet hunters are cautiously optimistic that we have found at least 12 such worlds so far.

As of July 2015, NASA's Kepler mission has detected a total of 4,696 possible exoplanets. The Kepler spacecraft has an instrument to measure the brightness of stars, and looks for tiny variations in brightness that could be caused by a planet passing between the star it orbits and the instrument observing the star. Potential candidates are then examined in more detail to see whether they are in fact planets or not. So far 1,030 of those candidates have been confirmed as planets.[1] Of those, 12 satisfy the criteria of being one to two times the size of Earth, and orbiting their star within the habitable zone.[2]

The uncertainty about the 12 possible Earth-like worlds is related to their composition. We don't yet know their composition; however, it is tempting to conclude that they are rocky because they are similar in size to Earth. Remember the rules of the accretion game: you can only begin to collect gas once you are a certain size, and how much matter you collect depends on how far away from the Sun you are. Given how large our gas giant and ice giant planets are compared to Earth, and how far away they are from the Sun, we would expect that a planet similar in size to Earth, and a similar distance from its star, should be rocky.

It isn't quite as simple as that, however. We are finding that the rules to the accretion game can result in planetary systems very different from our own, leading some people to wonder whether those planetary systems are strange, or ours is, and if ours is strange, how strange is it?

Consider that in the Kepler mission's observations thus far, it is very common to find planetary systems with planets larger than Earth orbiting closer to their star than Mercury does to the Sun. It is rare for planetary systems to have planets as large as Jupiter, and where large planets do exist, they are much closer to their star than Jupiter is to the Sun. To summarize, we need to be cautious about drawing conclusions from our own solar system, just in case we are basing those conclusions on something truly unusual.

On the other hand, the seemingly unique features of our solar system would make planetary systems like ours difficult to spot. Small planets are harder to detect because they block less of a star's light than larger planets. Larger planets farther from a star are difficult to spot because they don't go past the star as frequently. For example, Jupiter goes around the Sun once every 12 years, which means that if someone were observing our solar system, they might have to watch for 12 years to see Jupiter go past the Sun once. For Saturn, they might have to watch for 30 years.

So let's say the habitable-zone exoplanets are terrestrial. Does that mean we could live there?

The operational definition of "other Earths," which involves a terrestrial composition, a size constraint of one to two times that of Earth, and location within a star's habitable zone, does not preclude worlds incapable of supporting life as we know it. By those criteria, Venus is an "other Earth," albeit right on the edge of the habitable zone for our Sun. Venus is much too hot for us, with a constant surface temperature of 465°C (lead melts at 327°C). Its atmosphere is almost entirely carbon dioxide, and the atmospheric pressure at its surface is 92 times higher than on Earth. Any liquid water on its surface boiled off long ago. Yet the characteristics that make Venus a terrible picnic destination aren't entirely things we could predict from its distance from the sun. They depend in part on the geochemical evolution of Venus, and at one time Venus might have been a lot more like a youthful Earth. These are the kinds of things we won't know about until we can look carefully at the atmospheres and compositions of habitable-zone exoplanets.

Exercise 22.2 How well do we know the size of exoplanets?

One of the techniques for finding exoplanets is to measure changes in the brightness of a host star as the planet crosses in front of it and blocks some of its light. This diagram shows how the brightness changes over time. The dip in brightness reflects a planet crossing between the star and the instrument observing the star.

Often the planet itself is too small to see directly. If all we know is how the planet affects the brightness of the star, and we can't even see the planet, then how do we know how big the planet is? The answer is that the two are related. We can write an equation for this relationship using the radius of the planet and the radius of the star.

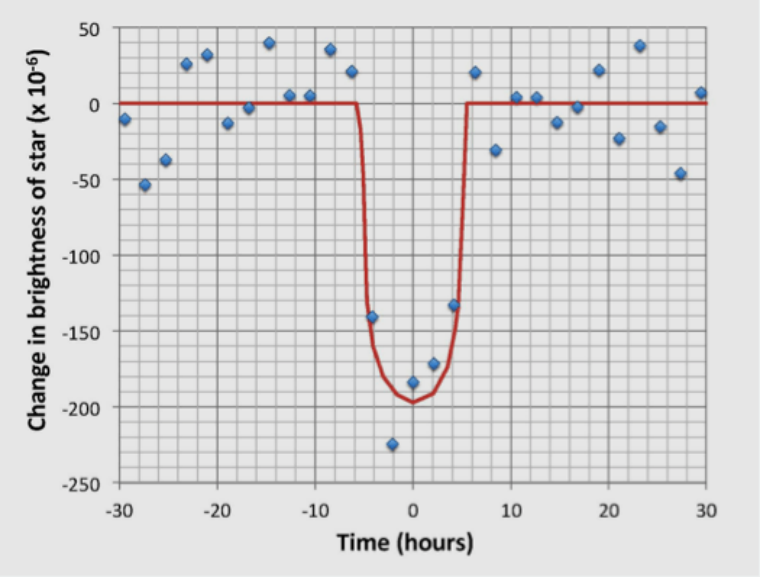


Figure 22.5.1 Plot showing how the star Kepler-452 dims as the planet Kepler-452b moves in front of it.

$$decrease\ in\ brightness = \frac{planet\ radius^2}{star\ radius^2}$$

Equation 1: Calculate decrease in brightness

Let's try this out for the Earth-like exoplanet called Kepler-452b. The first thing we need to know is the size of the host star Kepler-452. We can get that information by comparing its surface temperature and brightness to that of the sun. Start by calculating the ratios of the sun's temperature to the star's temperature, and the star's luminosity to the sun's luminosity using the data in Table 22.5. Record your answers in the table. Then find the star's radius using the following equation, and record your result:

$$star\ radius = sun\ radius \times \left(\frac{sun\ temperature}{star\ temperature}\right)^2 \times \sqrt{\frac{star\ luminosity}{sun\ luminosity}}$$

Equation 2: Calculate star radius.

Table 22.5 Calculating the radius of star Kepler-452

Description	Sun	Kepler-452	Ratio
Temperature (degrees Kelvin)	5,778	5,757	
Luminosity (× 1026 watts)	3.846	4.615	
Radius (km)	696,300		

The second thing we need to know is how the brightness of Kepler-452 changes as planet Kepler-452b moves in front of it. Use the plot shown in this exercise box to find this information. Find the value on the y-axis where the red curve shows the most dimming from the planet and record your result in Table 22.6.

Table 22.6 Calculating the radius of planet Kepler-452b

Decrease in brightness*	Earth radius in km	Kepler-452b radius in km	Kepler-452b radius/Earth radius
x 10^{-6}	6,378		

* Because we know this is a decrease, you don't need to keep the negative sign.

Use the following equation to find the radius of Kepler-452b:

$$planet\ radius = \sqrt{star\ radius^2 \times decrease\ in\ brightness}$$

Equation 3: Calculate planet radius.

To put the size of Kepler-452b in perspective, divide its radius by that of Earth and record your answer.

See Appendix 3 for Exercise 22.2 answers.

Image Descriptions

Equation 2 image description: Star radius equals sun radius times begin fraction sun temperature over star temperature end fraction squared times begin square root begin fraction star luminosity over sun luminosity end fraction end square root. [Return to Equation 2]

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- Figure 22.5.1: © Karla Panchuk. CC BY. Based on data from Jenkins, J. et al, 2015, Discovery and validation of Kepler-452b: a 1.6REarth super Earth exoplanet in the habitable zone of a G2 star, Astronomical Journal, V 150, DOI 10.1088/0004-6256/150/2/56.

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1. You can access a catalogue of confirmed exoplanets found by NASA and other planet-hunting organizations at http://exoplanet.eu/catalog/↵
2. Read more about habitable-zone planets discovered so far at http://www.nasa.gov/jpl/finding-another-earth↵

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22.6: Summary

The topics covered in this chapter can be summarized as follows:

Section	Summary
22.1 Starting with a Big Bang	The universe began 13.77 billion years ago when energy, matter, and space expanded from a single point. Evidence for the big bang is the cosmic “afterglow” from when the universe was still very dense, and red-shifted light from distant galaxies, which tell us the universe is still expanding.
22.2 Forming Planets from the Remnants of Exploding Stars	The big bang produced hydrogen, helium, and lithium, but heavier elements come from nuclear fusion reactions in stars. Large stars make elements such as silicon, iron, and magnesium, which are important in forming terrestrial planets. Large stars explode as supernovae and scatter the elements into space.
22.3 How to Build a Solar System	Solar systems begin with the collapse of a cloud of gas and dust. Material drawn to the centre forms a star, and the remainder forms a disk around the star. Material within the disk clumps together to form planets. In our solar system, rocky planets are closer to the Sun, and ice and gas giants are farther away. This is because temperatures near the Sun were too high for ice to form, but silicate minerals and metals could solidify.
22.4 Earth’s First 2 Billion Years	Early Earth was heated by radioactive decay, collisions with bodies from space, and gravitational compression. Heating melted Earth, causing molten metal to sink to Earth’s centre and form a core, and silicate melt to float to the surface and form the mantle and crust. A collision with a planet the size of Mars knocked debris into orbit around Earth, and the debris coalesced into the moon. Earth’s atmosphere is the result of volcanic degassing, contributions by comets and meteorites, and photosynthesis.
22.5 Are There Other Earths?	The search for exoplanets has identified 12 planets that are similar in size to Earth and within the habitable zone of their stars. These are thought to be rocky worlds like Earth, but the compositions of these planets are not known for certain.

Questions for Review

Answers to review questions can be found in Appendix 2.

1. How can astronomers view events that happened in the universe’s distant past?
2. In this image of three spectra, one is from the Sun, and the other two are from galaxies. One of the galaxies is the Andromeda galaxy. Which spectrum is from Andromeda?

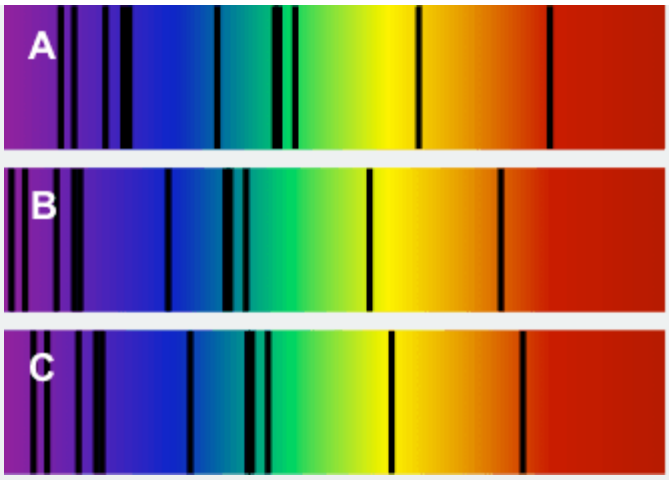


Figure A Spectra for the sun and two galaxies.

- Astronomers looking for some of the earliest stars in the universe were surprised to find a planetary system called HIP 11952, which existed 12.8 billion years ago. This was very early in the universe's history, when stars still consisted largely of hydrogen and helium. Do you think there were terrestrial planets in this system? Why or why not?
- Summarize the trends in size and composition of objects in the solar system.
- What is the frost line, and what does it help to explain?
- This cartoon shows three of the same type of solar system object. One goes on an adventure and comes back the worse for wear. What are the objects, and where might they be located?

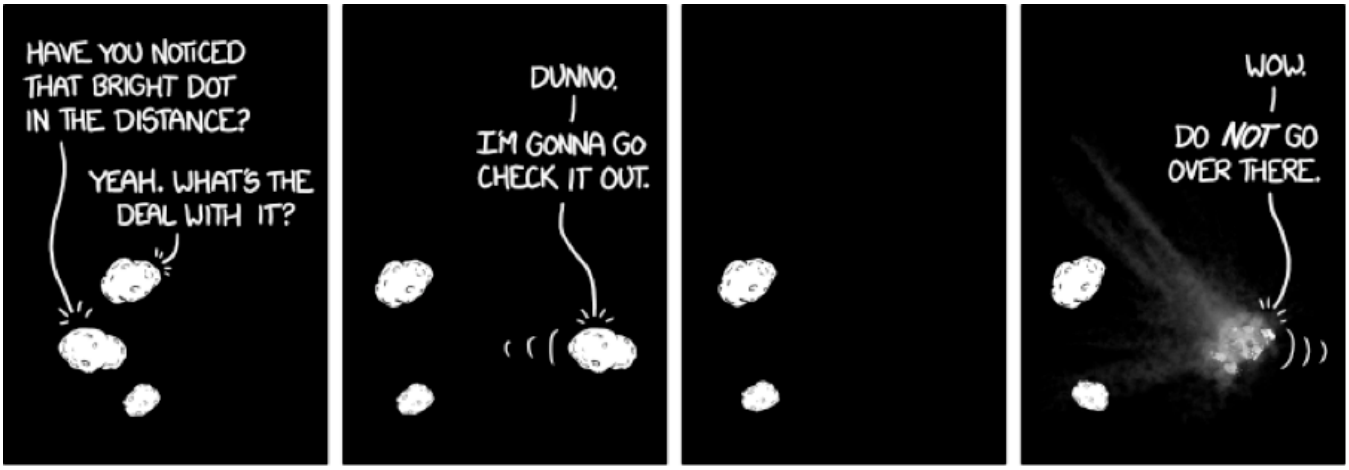


Figure B Denizens of the solar system.

- Why is Pluto not considered a planet?
- What is differentiation, and what must happen to a planet or asteroid for differentiation to occur?
- The exoplanet Kepler-452b is within the habitable zone of its star. In our solar system, planets a similar distance from the Sun are terrestrial planets. Why can we not say for certain that Kepler-452b's distance from its star means it is a terrestrial planet?
- Of the planetary systems discovered thus far, none are exactly like our solar system. Does this mean our solar system is unique in the universe?

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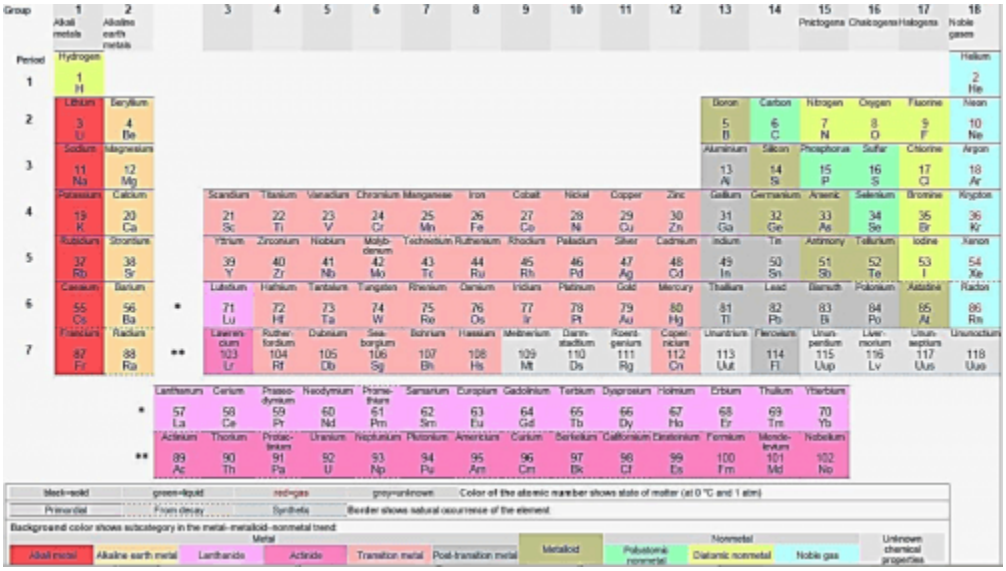
23.3: Appendix 1 - List of Geologically Important elements and the Periodic Table

The following table includes 36 of the geologically important elements, listed alphabetically by their element name, along with their atomic number and the atomic mass of their most stable isotope. The geologically most important elements are bolded, and the eight main elements of silicate minerals are identified with an asterisk (*).

Symbol	Name	Atomic No.	Atomic Mass
Al*	Aluminum	13	27
As	Arsenic	33	75
Ba	Barium	56	137
Be	Beryllium	4	9
B	Boron	5	11
Cd	Cadmium	48	112
Ca*	Calcium	20	40
C	**Carbon**	**6**	**12**
Cl	**Chlorine**	**17**	**35**
Cr	Chromium	24	52
Co	Cobalt	27	59
Cu	**Copper**	**29**	**64**
F	**Flourine**	**9**	**19**
Au	**Gold**	**79**	**197**
He	Helium	2	4
H	**Hydrogen**	**1**	**1**
Fe*	Iron	26	56
Pb	**Lead**	**82**	**207**
Mg*	Magnesium	12	24
Mn	**Manganese**	**25**	**55**
Mo	Molybdenum	42	96
Ne	Neon	10	20
Ni	**Nickel**	**28**	**59**
N	Nitrogren	7	14
O*	Oxygen	8	16
P	**Phosphorus**	**15**	**31**
Pt	Platinum	78	195
K*	Potassium	19	39
Si*	Silicon	14	28
Ag	Silver	47	108

Symbol	Name	Atomic No.	Atomic Mass
Na*	Sodium	11	23
Sr	Strontium	38	88
S	**Sulfur**	**16**	**32**
Ti	**Titanium**	**22**	**48**
U	Uranium	92	238
Zn	**Zinc**	**30**	**65**

The periodic table is a list of all of the elements arranged in groups according to their atomic configuration. In this table the elements are color-coded according to their chemical and physical properties.



For an accessible version of the periodic table please see Syngenta Period Table of Elements (http://www.syngentaperiodictable.co.uk/periodic-table.php?keyStage=5)

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23.4: Appendix 2 - Answers to Review Questions

Answers to the chapter-end review questions are provided below. (Answers to the embedded exercise questions are provided in Appendix 3.)

Chapter 1

1. Geology involves integration of various different sciences (chemistry, physics, and biology for example), but also requires an understanding of the importance of billions of years of geological time.
2. Paleontology is an important aspect of geology and requires an understanding of biology, including evolution, the physiology of animals and plants and ecological relationships.
3. Geologists provide information to reduce the risk of harm from hazards such as earthquakes, volcanoes, and slope failures; they play a critical role in the discovery of important resources; they contribute to our understanding of life and its evolution through paleontological studies; and they play a leading role in the investigation of climate change, past and present and its implications.
4. Halite is composed of sodium (Na) and chlorine (Cl) with the Na+ and Cl- ions alternating with one another in all three directions within a cubic structure.
5. A mineral has a specific chemical composition and lattice structure. Rocks are made out of minerals, and most rocks contain several different types of minerals.
6. The main component of Earth's core is iron (Fe).
7. Transfer of heat from the core to the mantle leads to heating of lower mantle rock. When heated, the rock expands and its density is reduced. Because the mantle is plastic, this lower-density material tends to rise toward the surface, and cooler denser mantle material moves in to take its place.
8. Mantle convection creates the traction that can force plates to move around on the surface.
9. Hot mantle rock moving toward the surface partially melts because the pressure is reduced. The magma produced moves upward into cracks in the crust and is extruded onto the sea floor.
10. 215 – 65 = 150 Ma. Since the age of the Earth is 4570 Ma, this represents 150/4,570 = 0.033 or 3.3% of geological time.
11. At 1 mm/y 30,000,000 mm would accumulate over that 30 million years. This is equivalent to 30,000 m or 30 km. Few sequences of sedimentary rock are even close to that thickness because most sediments accumulate at much lower rates, more like 0.1 mm/y.

Chapter 2

1. Charges: proton: +1, neutron: 0, electron: -1, Masses: proton: 1, neutron: 1, electron: almost 0.
2. The element's atomic number will determine the extent to which its outer layers are populated with electrons. If the outer shell is not quite full, the atom may gain electrons to fill them and become an anion (negative charge). If the outer shell has only a few electrons, it may lose them and become a cation (positive charge). Cations and anions attract each other to form molecules with ionic bonding.
3. Helium and neon (and the other noble gases) have complete outer shells and therefore no tendency to form ionic bonds.
4. Electrons are transferred from one atom to another to form an ionic bond. Electrons are shared between atoms to form a covalent bond.
5. An anion has a negative charge and a cation has a positive charge.
6. Minerals are classified into groups based on their anion or anion group.
7. Name the mineral group for the following minerals:

calcite	$CaCO_3$ carbonate	biotite	silicate	pyrite	FeS_2 sulfide
gypsum	$CaSO_4$ sulphate	galena	PbS sulfide	orthoclase	$KAlSi_3O_8$ silicate
hematite	Fe_2O_3 oxide	graphite	C native	magnetite	Fe_3O_4 oxide
quartz	SiO_2 silicate	fluorite	CaF_2 halide	olivine	$MgSiO_4$ silicate

8. An unbonded silica tetrahedron has one Si ion (+4 charge) and 4 oxygens (-2 charge each) so the overall charge is 4 – 8 =

-4 for SiO_4^{-4}

9. Magnesium can substitute freely for iron in olivine and several other minerals because they have similar charges (+2) and similar ionic radii.
10. Pyroxene is made up of single chains of tetrahedra while amphibole is made up of double chains.
11. Biotite includes iron and/or magnesium in its formula, while muscovite does not.
12. The two end-members of the plagioclase series are Albite ($NaAlSi_3O_8$) and Anorthite ($CaAl_2Si_2O_8$)
13. In quartz each silica tetrahedron is bonded to four other tetrahedra, and since oxygens are shared at each bond the overall ratio is silicon (+4) to two oxygens (2 x -2 = -4), which is balanced.
14. Some minerals have distinctive colors, but many have a wide range of colors due to differing impurities.
15. Glass has a Mohs hardness of about 5.5 while porcelain is close to 6.5. The mineral is between these two, so it must be close to 6.

Chapter 3

1. The rock must be exposed at surface so in many cases uplift and removal of overlying sediments is required. Then chemical and/or physical weathering can take place, which reduces the rock to smaller loose fragments. These fragments are sediments that can be eroded and then transported by a variety of maechanisms.
2. Sediments are buried beneath other sediments where, because of the increased pressure, they become compacted and dewatered. With additional burial they are warmed to the point where cementing minerals can form between the grains (less than 200˚C).
3. Rock is buried within the crust and heated because of the geothermal gradient. At temperatures over 200˚C some of the existing minerals may become unstable and will be converted to new minerals, or recrystallized into larger crystals.
4. As the temperature decreases minerals that formed early (e.g., olivine) may react with the remaining magma to form new minerals (e.g., pyroxene).
5. Calcium-rich plagioclase forms early on in the cooling process of a magma, but as the temperature drops, a more sodium-rich variety forms around the existing crystals.
6. Early-forming minerals, which are typically quite dense (e.g., olivine) may sink to the bottom of the magma chamber (if the magma is not too viscous) and thus become separated from the rest of the magma, resulting in a change to the composition of the remaining magma (it becomes more felsic).
7. If the texture is aphanitic the crystals are too small to see without a microscope. In rocks with phaneritic textures the minerals are large enough to see and distinguish from each other with the naked eye. The dividing line is somewhere between 0.1 and 1 mm, depending on the minerals.
8. In porphyritic rocks there are two distinct crystal sizes that are indicative of two stages of cooling (slow then fast). The fine material can range from glass to several mm, as long as the coarse crystals are distinctively larger. In pegmatitic rocks the crystals are consistently coarser than 1 cm, and can be much larger. Pegmatites form the slow cooling of water-rich magmas.

9.

(a) An extrusive rock with 40% Ca-rich plagioclase and 60% pyroxene	basalt
(b) An intrusive rock with 65% plagioclase, 25% amphibole, and 10% pyroxene	diorite
c) An intrusive rock with 25% quartz, 20% orthoclase, 50% plagioclase, and minor amounts of biotite	granite

10. A concordant body (a sill) is parallel to any pre-existing layering (bedding or foliation) in the country rock is. A discordant body (a dyke) cuts across any pre-existing layering or is situated at any angle in country rock that has no layering (e.g., granite).
11. A rock has to crack in order for a dyke to intrude into it, and it has to be cool to crack. When the hot magma intrudes into the cold country rock its margins cool quickly (forming small crystals), while its centre cools more slowly (forming larger crystals).
12. A batholith has an exposed area of greater than 100 $km^{2;}$ a stock has an exposed area less than that.
13. Batholiths (or stocks) intrude into existing rock by (a) melting through the country rock, or (b) causing the country rock to break and fall into the magma (stoping), or (c) pushing the country rock aside.
14. Compositional layering forms when early-crystallizing mineral sink toward the bottom of a magma chamber. This can only happen in non-viscous magma, and mafic magma is typically much less viscous than felsic magma.

Chapter 4

1. The three main tectonic settings for volcanism are (1) subduction zones at convergent plate boundaries, (2) divergent plate boundaries, and (3) mantle plumes (a.k.a. hot spots).
2. The primary mechanism for partial melting at a convergent plate boundary is the addition of water to hot mantle rock. The water reduces the melting temperature of the rock (flux melting).
3. The explosiveness of a volcanic eruption depends on the pressure of the magma. Gases create that pressure, and if the magma is viscous those gases cannot escape easily. Felsic and intermediate magmas tend to have more gas than mafic magmas, and are also more viscous, trapping the gas in.
4. When magma is deep within the crust the pressure is too high for the gases to bubble out of solution.
5. Pillow lavas form where mafic lava erupts in water. When the magma oozes out into the water the outside cools first forming a hard skin that maintains the pillow shape.
6. Composite volcanoes can produce rocks with a wide range of textures, including (1) aphanitic or porphyritic rock from lava flows, (2) pyroclastic rock (with textures ranging from fine ash to coarse fragments) from explosive eruptions, and (3) sedimentary rock from lahars.
7. A lahar is a mud flow or debris flow on a volcano. Lahars are common on composite volcanoes because they are steeper than shield volcanoes, they typically have ice and snow, and they are not as strong as shield volcanoes.
8. Some lahars form during an eruption when snow and ice melt quickly, while others may form from heavy rain.
9. he magma at shield volcanoes is typically non-viscous. It can flow easily and also tends to form lava tubes, and thus is able to extend a long way from the vent, forming a low broad shield.
10. Shield volcanoes tend to have much longer lives than composite volcanoes. Most of the Hawaiian shields, for example lasted 1 million years, while most composite volcanoes are younger than 100,000 years.
11. Weak seismic activity is associated with all stages of a volcanic eruption. In the early stages magma is moving at depth and pushing rock aside, creating small earthquakes. The flow of magma can also produce special type of seismic response known as harmonic tremor.
12. GPS technology is used to determine if there is any slow deformation of the flanks of a volcano related to movement of magma toward the surface.
13. The Mt. St. Helens columnar basalts were formed by a flow of mafic lava.
14. The Nazko Cone is thought to be related to a mantle plume.
15. No one is certain why there a lower rate of volcanism in B.C. than in adjacent Washington and Oregon, but one theory is that the northern part of the Juan de Fuca Plate (the Explorer Plate) is not subducting as quickly as the rest of the plate.
16. It is likely that carbon dioxide released during the eruption flowed downhill from the volcano to the village on the shore of the Nass River.

Chapter 5

1. Before a rock can be exposed at surface it has to be uplifted from where it formed deep in the crust, and the material on top has to be eroded.
2. Frost wedging is most effective at times when the weather swings between freezing at night and thawing during the day. In cold parts of B.C. that only happens consistently in spring and fall. In warmer regions it only happens consistently during the winter.
3. Under conditions of strong chemical weathering, feldspar albite ($NaAlSi_3O_8$) will be converted to a clay (such as kaolinite) and sodium ions in solution. Where mechanical weathering is predominant, albite will be broken into small pieces.
4. Acid rock drainage (ARD) creates acidic stream runoff and also enhances the solubility of a wide range of metals, some of which can be toxic.
5. Feldspar-rich sand is formed in areas where granitic rocks are being weathered and where mechanical weathering is strongly predominant over chemical weathering.
6. Most of the clay that forms during hydrolysis of silicate minerals ends up in rivers and is washed out to the oceans. There it eventually settles to the sea floor.
7. The mineral composition of the parent rock or sediment will influence the composition of the resulting soil. Slope is important because it will affect the degree to which materials will be eroded.
8. Clay minerals and iron move downward to produce the B horizon of a soil.
9. Removal of vegetation leaves soil exposed to erosion by water, and wind are the main processes of soil erosion in Canada.
10. Chernozemic soils are common in the southern prairies and parts of the BC southern interior, in areas that experience water deficits during the summer.
11. Luvisolic soils are found in central B.C., mostly over sedimentary rocks.
12. The weathering feldspar to clay involves the conversion of atmospheric carbon dioxide to dissolved bicarbonate, which ends

up in the ocean.

Chapter 6

1. Sand grains range in size from 1/16 mm to 2 mm.
2. Both silt and clay feel smooth between your fingers, but only clay feels smooth in your mouth.
3. The key factor is particle size (not density). Settling velocity is controlled by the friction around the grain holding it up and the gravitational force pushing it down. The gravitational force is proportional to the grain volume and the friction is proportional to the surface area.
4. Conglomerate cannot be deposited by a slow-flowing river because clasts larger than 2 mm are not transported by slow-moving water.
5. Sediments are buried beneath other sediments where, because of the increased pressure, they become compacted and dewatered. With additional burial they are warmed to the point where cementing minerals can form between the grains (less than 200˚C).
6. Lithic arenite has less than 15% silt- and clay-sized particles, while a lithic wacke has more than 15%. Both have more than 10% rock fragments and more rock fragments than feldspar.
7. Feldspathic arenite has more than 10% feldspar and more feldspar than rock fragments. Quartz arenite has less than 10% feldspar and less than 10% rock fragments. Both have less than 15% silt and clay.
8. Source area lithology: rock that contains quartz (such as granite or sandstone), strong weathering to remove feldspar, long fluvial transportation to round the grains.
9. The carbon within carbonate deposits such as limestone originally comes from the atmosphere.
10. Most of Earth's banded iron formations formed during the initial oxygenation of the atmosphere between 2.4 and 1.8 Ga because iron that had been soluble in the anoxic oceans became insoluble in the oxidized oceans.
11. Terrestrial depositional environments: rivers, lakes, deltas, deserts, glaciers. Marine depositional environments: continental shelves, continental slopes, deep ocean.
12. A foreland basin forms in the vicinity of a large range of mountains where the weight of the mountains depresses the crust on either side. A forearc basin lies between a subduction zone and the related volcanic arc.
13. (a) Bedding forms where there is an interruption or change in the depositional process, or a change in the composition of the material being deposited. (b) Cross-bedding forms in fluvial or aeolian environments where sand-sized sediments are being moved and ripples or dunes are present. (c) Graded bedding, and (d) mud cracks form where fine-grained (silt or clay) sediments are allowed to dry because the level of a lake decreases.
14. Reverse graded bedding forms during gravity flows, such as debris flows.
15. A formation is a series of beds that is distinct from other beds above and below it, and is thick enough to be shown on the geological maps that are widely used within the area in question.
16. The Nanaimo Group was actively mined for coal for many decades. During that time the names were given to members and individual beds that were important to the coal miners.

Chapter 7

1. Heat and pressure are the main agents of metamorphism. Heat leads to mineralogical changes in the rock. Pressure also influences those mineralogical changes, while directed pressure (greater pressure in one direction) leads to foliation.
2. At very low, low, medium, and high metamorphic grades, mudrock will be transformed into slate, phyllite, schist, and gneiss.
3. Granite remains largely unchanged at lower metamorphic grades because its minerals are still stable at those lower temperatures.
4. Foliation exists because as new minerals are forming in a situation of directed pressure they are forced to grow with their long axes perpendicular to the main pressure direction.
5. At a spreading ridge the heat from volcanism leads to the development of a groundwater convection system in the oceanic crustal rock. Heated water rises in the hot regions and is expelled into the ocean, while cold ocean water is drawn into the crust to replace it. The heated water leads to the conversion of ferromagnesian minerals (e.g., olivine and pyroxene) into chlorite and serpentine.
6. The geothermal gradient varies as a function of tectonic setting, being greatest in volcanic regions and lowest along subduction zones. As a result the depth at which specific metamorphic grades is achieved will vary (greater depth where the gradient is least).
7. The geothermal gradient is low within subduction zones (because the cold subducting oceanic crust takes a long time to heat up), so while pressure increases at the normal rate the temperature does not.
8. In order of increasing metamorphic grade: chlorite biotite, garnet, sillimanite.
9. The rocks at significant depth in the crust are already hot and subject to regional metamorphism, so the additional heat from

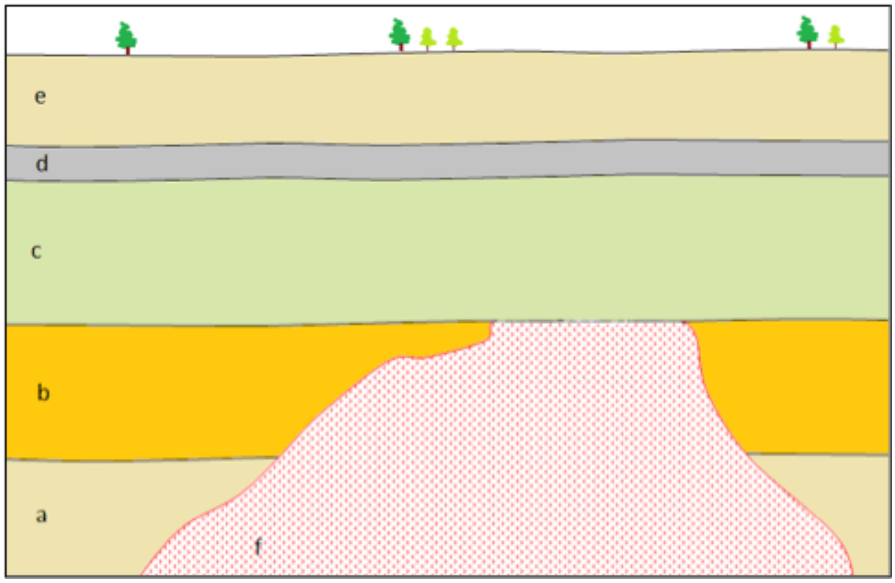
a pluton doesn't make a large difference.

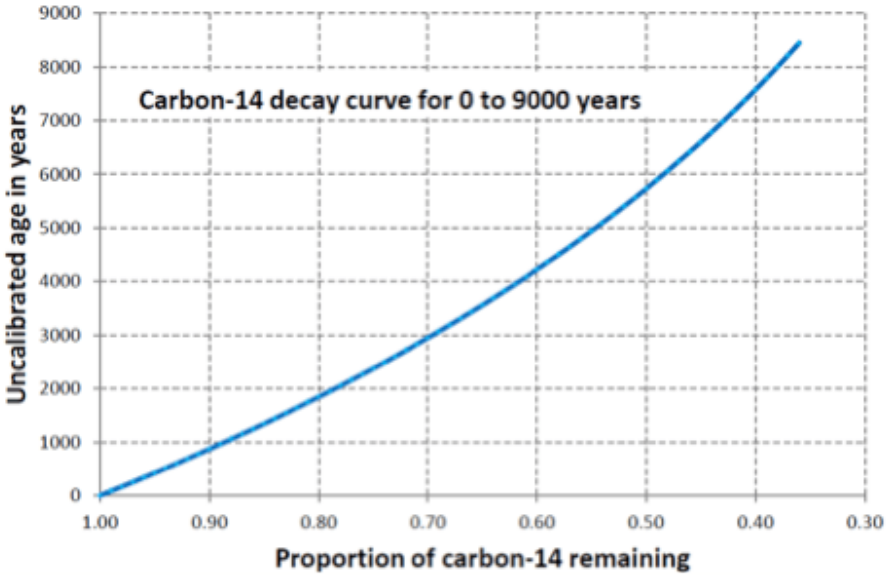
10. Water from any source facilitates metamorphism. Magmatic fluids typically contain dissolved ions at higher levels than in regular groundwater (especially copper, zinc, silver, gold, lithium, beryllium, boron and fluorine) so can lead to formation of a unique set of minerals.
11. Metasomatism involves fluids from magmatic or groundwater sources that play an important role in transporting ions and leading to the formation of new minerals.
12. A hot pluton heats the surrounding water and this contributes to the development of a convection system in the groundwater, which can result a great deal of water, in some cases with elevated levels of specific ions, passing through the rock. Magmatic water also contributes to metasomatism.
13. Limestone must be present in the country rock to produce a skarn.
14. Two things that a geologist first considers when looking at a metamorphic rock are what the parent rock might have been, and what type of metamorphism has taken place.

Metamorphic Rock	**Likely Parent Rock**	**Grade and/or Type of Metamorphism**
Chlorite schist	A rock enriched in ferromagnesian minerals, such as basalt.	Low-grade regional metamorphism.
Slate	Mudrock (shale, mudstone)	Very low grade regional metamorphism
Mica-garnet schist	A rock that is rich in aluminum, which includes most clay-bearing rocks.	Medium-grade regional metamorphism
Amphibolite	A rock enriched in ferromagnesian minerals, such as basalt.	Medium- to high-grade regional metamorphism.
Marble	Limestone or dolomite	Regional or contact metamorphism.

Chapter 8

1. Xenoliths of basalt within a granite must be older than the granite according to the principle of inclusions.
2. (a) At both disconformities and paraconformities the beds above and below are parallel, but at a disconformity there is clear evidence of an erosion surface (the lower layers have been eroded). (b) A nonconformity is a boundary between sedimentary rocks above and non-sedimentary rocks below while an angular unconformity is a boundary between sedimentary rocks above and tilted and eroded and sedimentary layers below.
3. A useful index fossil must have survived for a relatively short period (e.g., around a million years), and also should have a wide distribution so that it can be used to correlate rocks from different regions.
4. The granitic rock "f" has been dated to 175 Ma. The wood in layer "d" is approximately 5,000 years old, so we can assume that layer "d" is no older than that, although it could be as much as a few hundred years younger if the wood was already old when it got incorporated into the rock.

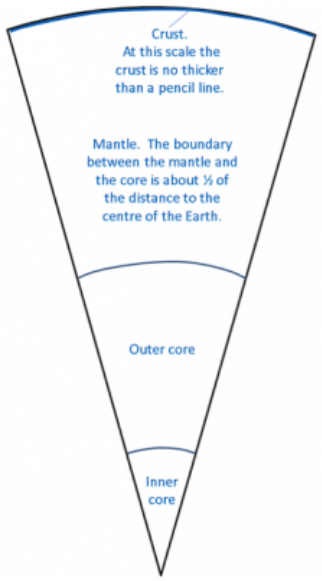




5. Layer “c” must be between 5,000 y and 275 Ma.
6. The unconformity between layer “c” and rock “f” is a nonconformity.
7. The granite (f) was eroded prior to deposition of “c”, so it’s likely that layer “b” was also eroded at the same time. If so, that makes the boundary between “c” and “b” a disconformity.
8. The last magnetic reversal was 780,000 years ago, so all rock formed since that time is normally magnetized and it isn’t possible to distinguish older rock from younger rock within that time period using magnetic data.
9. William Smith was familiar with the different diagnostic fossils of the rocks of England and Wales and was able to use them to identify rocks of different ages.
10. The last age of the Cretaceous is the Maastrichtian (70.6 to 65.5 Ma) and the first age of the Paleogene is the Danian (65.5 to 61.7 Ma).

Chapter 9

1. Typical stony meteorites are similar in composition to the Earth’s mantle, while typical iron meteorites are similar to the core.
2. The crust/mantle, mantle/core, outer core/inner core are shown on the diagram below:



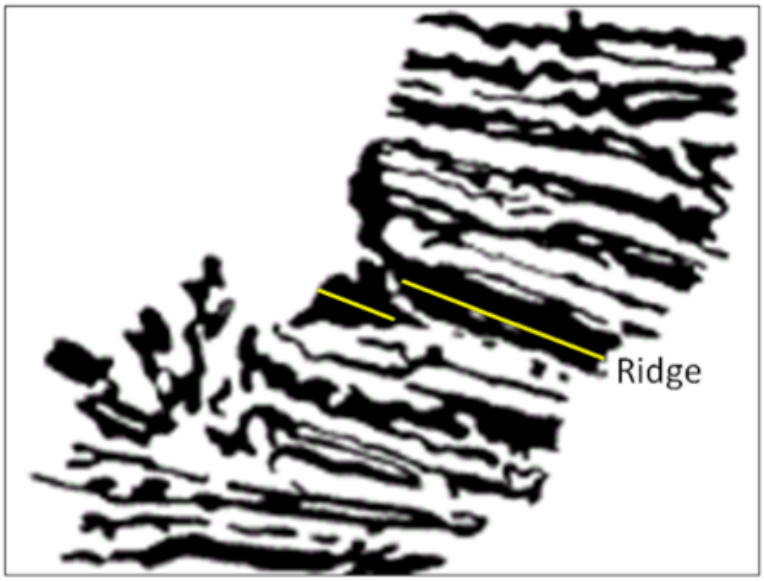
3. P-waves can pass through a liquid and travel approximately twice as fast as S-waves (which cannot pass through a liquid).
4. P-wave velocity decreases at the core-mantle boundary because the outer core is liquid.
5. The mantle gets increasingly dense and strong with depth because of the increasing pressure. This difference affects both P-wave and S-wave velocities, and they are refracted toward the lower density mantle material (meaning they are bent out toward Earth's surface).
6. The key evidence for mantle convection is that the rate of temperature increase within the mantle is less than expected and this can only be explained by a mantle that is mixing by convection. The mechanism for convection is the transfer of heat from the core to the mantle.
7. Earth's magnetic field is generated within the liquid outer part of the core by the motion of the metallic core material.
8. The last two reversals of Earth's magnetic field were at the beginning of the present Brunhes normal chron (0.78 Ma), and at the end of the Jaramillo normal subchron (0.90 Ma).
9. The isostatic relationship between the crust and the mantle is dependent on the plastic nature of the mantle.
10. In the area of the Rocky Mountains the crust is thickened and pushed down into the mantle. In Saskatchewan the crust is thinner and does not extend as far into the mantle.
11. During the Pleistocene glaciation British Columbia was pushed down by glacial ice and mantle rock flowed slowly out beneath the ocean floor. Now that the land area is rebounding, that mantle rock is flowing back and the offshore areas are subsiding.

Chapter 10

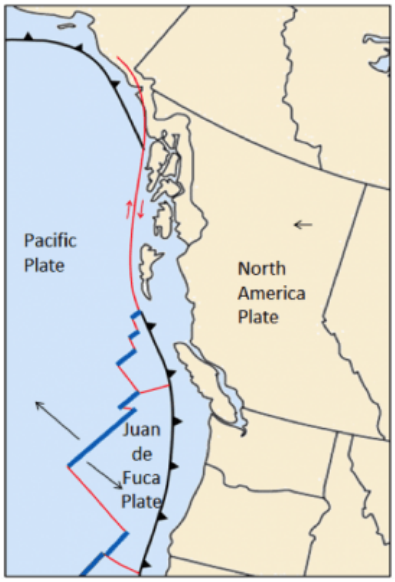
1. The evidence used by Wegener to support his idea of moving continents included matching continental shapes and geological features on either side of the Atlantic; common terrestrial fossils in South America, Africa, Australia, and India; and data on the rate of separation between Greenland and Europe.
2. The primary technical weakness of Wegener's theory was that he had no realistic mechanism for making continents move.
3. (a) Contractionists assumed that mountains formed because as the Earth contracted the crust wrinkled into mountains. (b) Permanentists assumed that mountains formed by the geosynclinal process.
4. In the late 19th century the trans-Atlantic paleontological matchups were explained by assuming that there must have been land bridges between the continents at some time in the past, or that terrestrial organisms had floated across the ocean on logs.
5. Continental crust is lighter than oceanic crust and cannot sink low enough into the mantle to become an ocean (although this can happen over limited areas, and commonly does happen along coastal areas of continental plates).
6. Prior to 1920, ocean depths were measured by dropping a weighted line over the side of ship. Echo sounding techniques were developed at around that time and that greatly facilitated the measurement of ocean depths.
7. Temperature increases quite rapidly with depth in the crust, but much less so in the mantle, and this implies mantle convection.
8. Paleomagnetic studies showed that old rocks on the continents had different pole positions than they do today, and also that they were progressively more different with time past. This implied either that the poles had moved or that the continents had moved. It was also found that the apparent polar wandering paths for different continents were different, and this supported

the concept that the continents had moved.

9. The trenches associated with subductions zones are the deepest parts of the oceans.
10. The ocean ridge areas are the youngest parts of the sea floor and thus there hasn't been time for much sediment to accumulate.
11. It was (and still is) assumed that high heat flow exists where mantle convection cells are moving hot rock from the lower mantle toward the surface, and that low heat flow exists where there is downward movement of mantle rock.
12. Earthquakes are consistently shallow and relatively small at ocean ridges. At ocean trenches earthquakes get increasingly deep in the direction that the subducting plate is moving. The earthquakes near to surface can be very large, while those at depth tend to be small.
13. In the Hess model new crust was formed at ocean ridges and then was consumed back into the mantle at the trenches.
14. Hess's theory did not include the concept of tectonic plates.
15. The spreading ridge is shown as a yellow line.



16. A mantle plume is a column of hot rock (not magma) that ascends toward the surface from the lower mantle. It is hypothesized that mantle plumes ascend as much as 10 times faster than the rate of mantle convection.
17. (a) Between the ridge segments there is movement in opposite directions along a transform fault. (b) Outside of the ridge segments the two plates are moving in the same direction and likely at about the same rate. In this case there is no faulting, and it is known as a fracture zone.
18. Tectonic plates are made up of crust and the lithospheric (rigid) part of the underlying mantle. The mantle part ensures that the very different oceanic and continental crust sections of a plate can act as one unit.
19. A mantle plume beneath a continent can cause the crust to form a dome which might eventually split open. Several mantle plumes along a line within a continent could lead to rifting.
20. Subduction does not take place at a continent-continent convergent zone because neither plate is dense enough to sink into the mantle.
21. The divergent boundaries are blue, the convergent boundaries are black with teeth on them, and the transform boundaries are red.
22. The motion directions are shown with black arrows (see map for names of plates).
23. The sense of motion on the Queen Charlotte Fault is shown with red arrows.

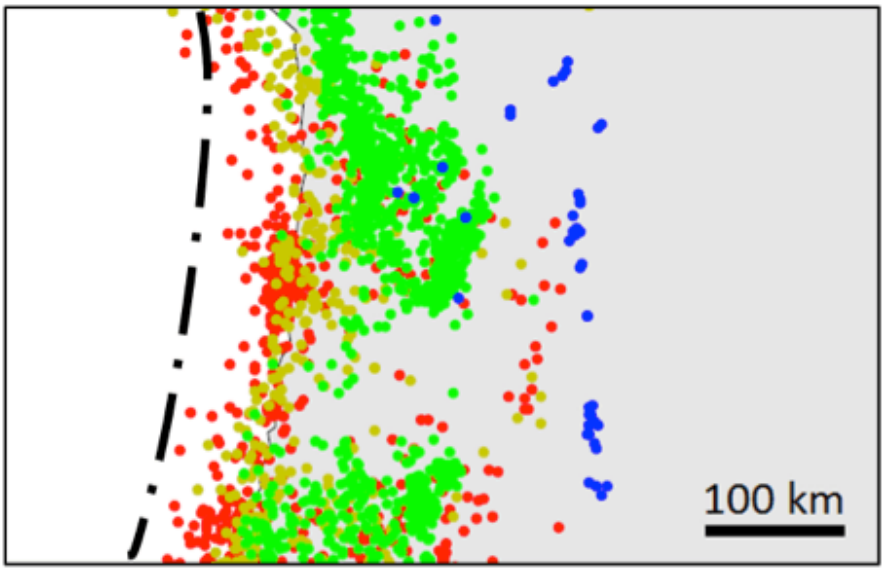


24. Continental rifting is taking place along the East Africa Rift, and sea floor has recently been created in the Red Sea and also in the Gulf of California.
25. Over the next 50 million years California is likely to split away from the rest of North America along the San Andreas Fault and then move north toward Alaska.
26. The accumulation of sediment at a passive ocean-continent boundary will lead to the depression of the lithosphere and could eventually result in the separation of the oceanic and continental parts of the plate and the beginning of subduction.

Chapter 11

1. An earthquake is the shaking caused by the release of energy that takes place when rocks under stress within Earth break and then the two sides slide past each other.
2. Rocks under stress will deform elastically until they reach the point where the stored elastic energy exceeds the rock strength. At that point the rock breaks and an earthquake is produced.
3. The rupture surface is the surface over which there is displacement of rock during an earthquake. The magnitude of an earthquake is proportional to the area of the rupture surface and the average amount of displacement over that surface.
4. An aftershock is any earthquake that is considered to have been caused by a previous earthquake as a result of the transfer of stress from the original earthquake.
5. Episodic slip on the middle part of the Cascadia subduction zone decreases stress within that area, but some of that stress is transferred to the locked zone up dip along the plate boundary, there increasing the level of stress on the locked part.
6. Magnitude is the amount of energy released by an earthquake. Each earthquake has only one magnitude, although there are different ways of measuring it, and they may give slightly different results. Intensity is a measure of the amount of damage done or what people felt. Intensity varies depending on the distance to the epicentre and the type of rock or sediment underlying an area.
7. An M7.3 earthquake releases 1,024 times as much energy as an M5.3 earthquake.
8. The map shows a subduction boundary. The depth of earthquakes increases inshore (to the east) from the location of the subduction zone.

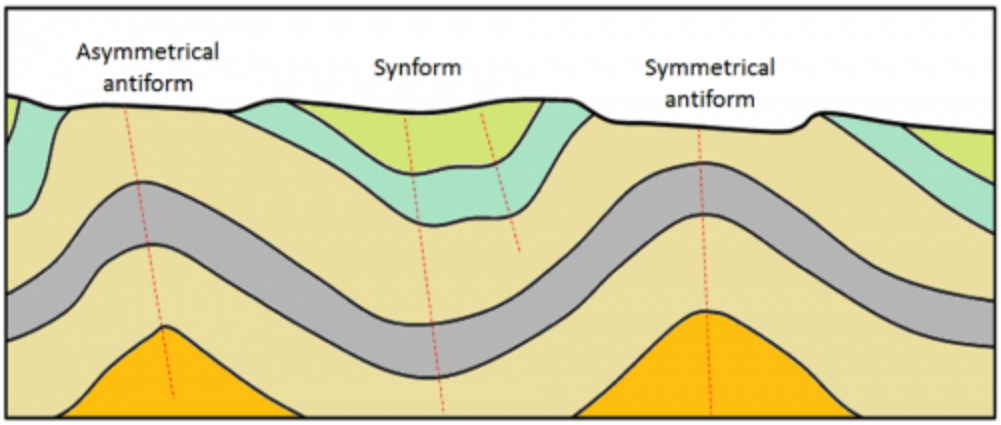
9. The dash-dot line shows approximately where the plate boundary is situated.



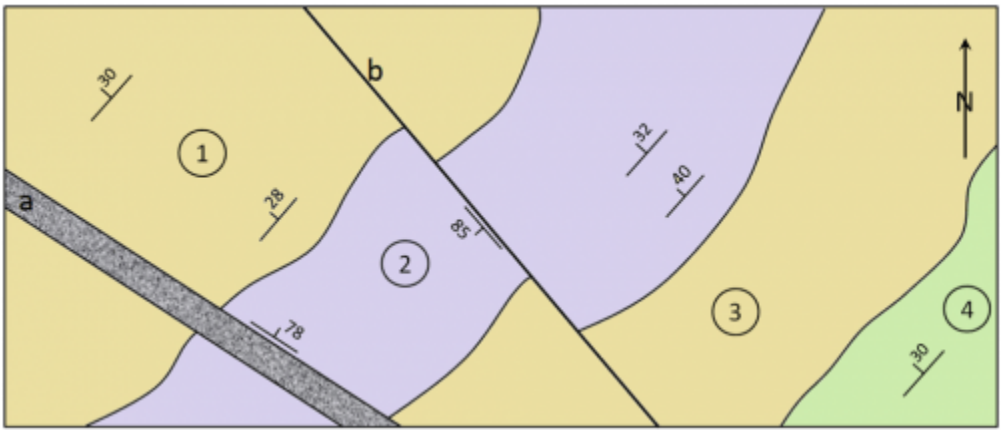
10. The plate on the left (Nazca Plate) is moving east and the one on the right (South America Plate) is moving west. This is the eastern coast of South America around Peru and Chile.
11. Both divergent and transform boundaries are associated with mid-ocean ridges. Most earthquakes take place on the transform boundaries.
12. The northward motion of the Pacific Plate relative to the North America Plate takes place along the San Andreas Fault in California and along the Queen Charlotte Fault off the coast of British Columbia and southern Alaska.
13. Unconsolidated sediments, especially if they are saturated with water, can lose strength when subjected to earthquake shaking. This can cause buildings to subside or tilt. Unconsolidated sediments can also amplify the vibrations of an earthquake.
14. Gas lines and electrical transmission wires are typically damaged during an earthquake, and this can lead to serious fires.
15. A large subduction earthquake (greater than M7.5) can generate a tsunami because they typically result in vertical displacement of the sea floor.
16. The 2004 Parkfield earthquake showed that we cannot rely on foreshocks to predict earthquakes, or on any of the many other parameters that were being carefully measured around Parkfield in the years leading up to the quake.
17. We should know about the history of past large earthquakes, the typical locations of small earthquakes, the types of geological materials beneath the surface (especially soft water-saturated sediments), the types of infrastructure that is present, and the various ways that people can be evacuated from an area or assistance can be brought in.
18. Forecasting involves estimating the risk of an earthquake happening in a region within a period of time (usually expressed in decades). Prediction involves stating that an earthquake is likely to happen at a certain location on a specific day or month or year in the future. With our current state of knowledge of earthquakes, prediction is not possible.

Chapter 12

1. Convergent plate boundaries are the most likely to contribute to compression, divergent boundaries to extension, and transform boundaries to shearing, however all of these stress regimes can exist at any one of these boundaries.
2. When elastic strain takes place the rock can rebound to its original shape. When there is plastic strain the rock will be permanently deformed.
3. Stronger rocks are more likely than weaker ones to deform elastically. Rock that is hot is more likely to deform plastically. Clay-bearing rocks are more likely to deform plastically when they are wet. If stress is applied quickly, the rock is more likely to break than if it is applied slowly.
4. The axial planes are shown with dashed red lines.



5. Volcanic rocks cool quickly at surface and the resulting reduction in volume can easily lead to fracturing.
6. In a normal fault the rock above the fault moves down with respect to the lower rock. This normally indicates extension. In a reverse fault the rock above the fault is pushed up, which indicates compression.
7. Most faults near transform boundaries are strike-slip faults, meaning that there is horizontal motion along the fault.
8. (a) The beds are dipping at about 30˚ to the northwest. (b) If it is possible to show that the beds are not overturned then we can say that bed 4 is the oldest. (c) “a” is a dyke and it is dipping steeply to the northeast. (d) “b” is a fault and it is dipping steeply to the southeast. (e) The motion on fault “b” appears to be left lateral. There may also be some vertical motion on “b” (or in fact the motion may be entirely vertical), but we cannot determine that from the information provided.



Chapter 13

1. Approximately 1% of the Earth’s water is liquid fresh water.
2. Approxmately 30% of the Earth’s fresh water is groundwater.
3. A trellis drainage pattern typically forms on sedimentary rock that has been tilted and eroded
4. Many of the streams in the southwestern part of Vancouver Island flow to the ocean as waterfalls because the land has been uplifted relative to sea level over the past several thousand years.
5. The fastest water flow on a straight stretch of a stream will be in the middle of the stream near the surface.
6. 1 mm sand grains will be eroded if the velocity if over 20 cm/s and will be kept in suspension as long as the velocity is over 10 cm/s.
7. If the flow velocity is 1 cm/s particles less than 0.1 mm (fine sand or finer) can be transported, while those larger than 0.1 mm cannot. At this velocity no particles can be eroded.
8. A braided stream can develop where there is more sediment available than can be carried in the amount of water present at the rate at which that water is flowing. This may happen where the gradient drops suddenly, or where there is a dramatic increase in the amount of sediment available (e.g., following an explosive volcanic eruption).
9. If a meander is cut off it reduces the length of a stream so it increases the gradient.
10. The average gradient of the Fraser River between Hope and the Pacific Ocean is 0.28 m/km (or 28 cm/km).
11. In coastal regions of B.C. the highest levels of precipitation are in the winter, and large parts of most drainage basins are not

frozen solid. As a result stream discharges tend to be greatest in the winter.

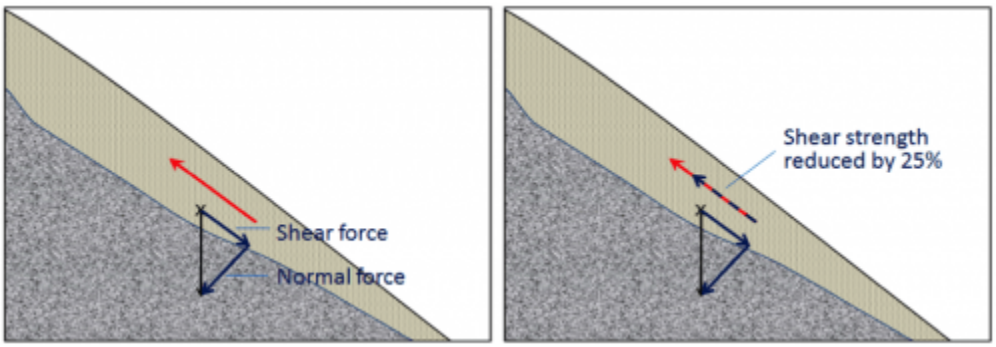
12. In most parts of Canada winter precipitation is locked up in snow until the melt season begins, and depending on the year and the location that happens in late spring or early summer. If the thaw is delayed because of a cold spring, and then happens very quickly, flooding is likely. Some regions also receive heavy rainfall during this period of the year.
13. Ri = (n+1)/r (where n is the length of the record) and r is the rank of the flood in question. In the Ashnola River case Ri = (65+1)/2 = 33. The probability of such a flood next year is 1/Ri, or 1/33 which is 0.03 or 3%.

Chapter 14

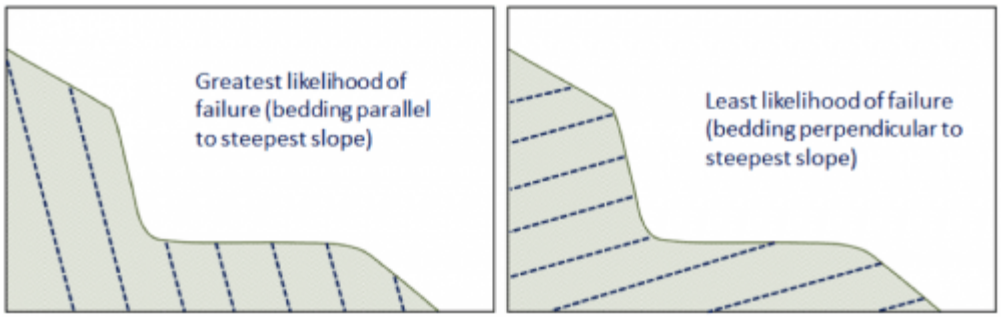
1. Porosity is the proportion of open space (space that can be filled with water), within a rock or unconsolidated sediment. Permeability is an expression of the ease with which water will flow through that material.
2. Clay deposits have low permeability because of the small size of the clay fragments. Water is tightly held to the grains by surface tension, and in the very small spaces between grains in clay there is virtually no water that is not able to flow.
3. From least to most permeable: unfractured gneiss, mudstone, sandstone, fractured granite, limestone in a karst region.
4. (a) Sue's well accesses an unconfined aquifer with low permeability. (b) Frank's well accesses a confined aquifer with high permeability. (c) Sue's low capacity aquifer acts as a (leaky) confining layer to Frank's high capacity aquifer.
5. V = *K*i, i = the gradient which is the elevation difference (83-77 = 6 m) over the distance (70 m) = 0.09, therefore V = 0.003 * 0.09 = 0.00027 m/s
6. After a drop of 9 m (from 83 to 74 m), and assuming that the other well did not drop at all, the gradient direction will have changed and the groundwater should flow toward the well that now has a level of 74 m.
7. Governments have the responsibility to protect our resources and to do their best to make sure that individuals and industry can access the groundwater that they need. Without observation well networks governments will have no independent information on how water levels are changing, and will be unable to make decisions on what might need to be done to ensure an adequate water supply for all.
8. Natural groundwater contamination originates from the natural reactions between the groundwater and the aquifer minerals. Anthropogenic groundwater contamination typically comes from human-sourced chemicals at or near to surface that are allowed to leak into the aquifer.
9. Water travels faster through a highly permeable aquifer and thus can spread the contamination further than in a less permeable one.
10. Livestock wastes are rich in nitrogen compounds, and these most commonly lead to nitrate contamination within the groundwater. Livestock wastes may also contain pharmaceuticals, which could contaminate groundwater.
11. The mineral pyrite is most likely to be responsible for acid rock drainage.
12. The waste water in a septic field needs to percolate slowly through the ground in order for natural processes to break down the contaminants. If the permeability is too low the waste water could come to surface. If the permeability is too high it could contaminate groundwater.

Chapter 15

1. The shear force and normal force vectors are shown on the left-hand diagram:



2. Based on the relative lengths of the arrows it appears that this material is stable, and unlikely to fail.
3. If the shear strength was reduced by 25% (right-hand diagram) the material would be much closer to failure, but the strength (based on the length of the arrows) still appears to be greater than the shear force.

4. 

5. In moist sand the grains are each surrounded by an envelope of water, and the water envelopes overlap. The attractive surface tension of the water holds the grains together.
6. In a the material moves like a fluid (individual particles move independently). In a the mass moves as an intact unit, with little or no relative motion between grains or clasts.
7. If a large rock slide starts moving at a rate of several meters per second, the rock is very likely to break into smaller pieces. If the pieces are small and numerous enough that the material can flow, then it becomes a rock avalanche.
8. A debris flow is composed mostly of sand-sized and larger clasts, while a mudflow is composed mostly of sand-sized and smaller clasts.
9. Residents at risk from Mt. Rainier lahars need to know what the warnings mean and roughly how much time they have between receiving a warning and being in actual danger. They need to create a plan to exit their residence quickly, and they need to know which way to go to get to safety as efficiently as possible.
10. Some of the important factors include:

    - The steepness of the slope
    - Any existing erosion processes happening at the base of the slope (e.g., wave or stream erosion)
    - The nature of surface or shallow sub-surface drainage in the upper part of the slope, and any effects that the construction might have on the drainage
    - The weight of the building (unless it is to be constructed in an excavation that represents more mass than the building itself)

Chapter 16

1. The Cryogenian glaciations are called Snowball Earth because it is thought that freezing conditions affected the entire planet and that the oceans were frozen over, even at the equator.
2. The cooling from the end of the Paleocene until the Holocene was related to the formation of mountains including the Himalayas, the Rockies, and the Andes; the opening of the Drake Passage; the development of Antarctic Circumpolar Current; and the closing of the Isthmus of Panama.
3. The first glaciation of the Cenozoic took place in Antarctica during the Oligocene (around 30 Ma).
4. At the height of the last glaciation, the Laurentide Ice Sheet covered almost all of Canada and extended south into the United States as far as Wisconsin.
5. Continental glaciers flow from the areas where the ice is thickest (and therefore at the highest elevation) toward areas (at the margins) where the ice is thinnest. Ice thickness tends to be related to the rate of ice accumulation.
6. The equilibrium line represents the boundary between the area where ice is accumulating (typically at high elevations), and where it is being depleted (mostly by melting). Above the equilibrium line more snow accumulates in winter than can melt in summer so the glacier is always covered in snow. Below the equilibrium line the snow cover is lost by the end of summer.
7. Relatively cool summers are more important because that controls how much snow will melt in the summer. In many situations very cold winters are associated with less snow accumulation than just cold winters.
8. (a) The ice at the bottom of a glacier flows more slowly than that at the top. In fact if the glacier is frozen to its base the lowermost ice might not be moving at all. (b) The edges also flow more slowly than the middle because there is more friction there between the ice and the valley walls.
9. Basal sliding will take place when the bed of the glacier is warm enough for water to be liquid. The water will act as a lubricant to allow the ice to flow.
10. Glaciers carve U-shaped valleys because they are relatively wide (compared with rivers) and most of the erosion takes place at the base rather than the sides. A hanging valley forms where a tributary glacier joins a larger glacier and where the larger glacier has eroded a deeper valley.
11. There must be at least three cirques to form a horn. In most cases there wouldn't be room for more than four.

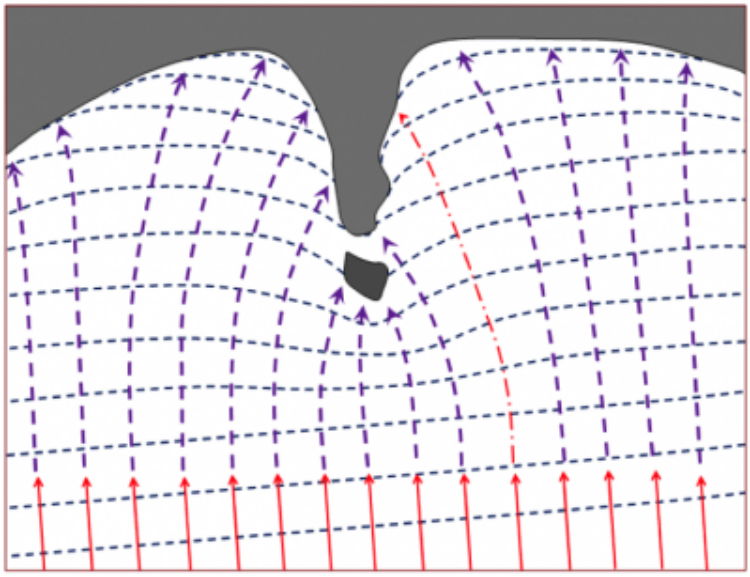
12. A drumlin is relatively steep at the up-ice end and streamlined at the down-ice end. A roche moutonée is streamlined at the up-ice end and jagged at the down-ice end where plucking has taken place.

Poorly sorted, clay to boulders, no layering, large clasts not well rounded, no structures.	Bedded, some fine beds (sand) and some much coarser. Clasts are relatively well rounded	Poorly sorted, pebbles to boulders, no layering, large angular clasts, no apparent structures.	Sand- and silt-sized. Well sorted. Bedded and cross bedded.
			
Lodgement till	Glaciofluvial sand & gravel	Ablation till	Glaciofluvial sand

13.
14. Drop stones are large clasts that are present with lacustrine or marine glacial sediments. They form when coarse material drops from melting icebergs.
15. Glaciofluvial sediments (sand or sand and gravel) are likely to be sufficiently permeable to make good aquifers.

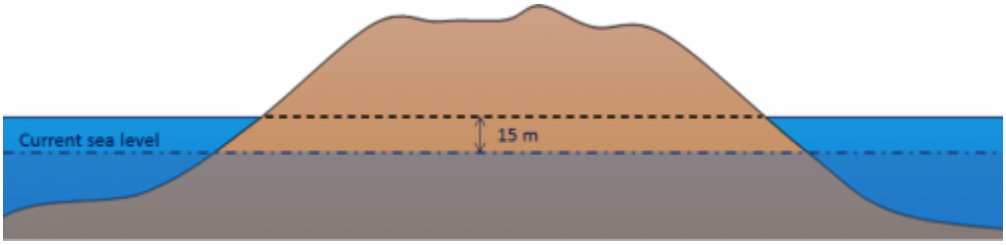
Chapter 17

1. The size of waves are determined by the wind velocity, the length of time the wind blows in the approximately the same direction, and the area of water over which it blows.
2. Table 17.1 provides data for 56 and 74 km/h winds, and 65 km/h is half way between these two values. The listed values for duration and fetch are high enough for the sea to fully develop, so the simple answer would be that the wave amplitude and wavelength would also be approximately half way between the listed values: amplitude ~6 m, wavelength ~106 m.
3. Waves will start to feel" the bottom at around 50% of the wavelength, so at 50 m depth in this case. This will slow the waves down and also cause their amplitude to increase.
4. A longshore current is the movement of water parallel to the shore in the surface zone caused by waves approaching at an angle. Longshore drift is the movement of sediment parallel to the shoreline, caused partly by the longshore current and also by swash and backwash on the beach.
5. Wave energy is focused on the headland, with more wave energy vectors per length of coast than in the bays, and thus the headland is eroding faster than the bays on either side, leading to coastal straightening.



6. Rocky coasts are eroded by waves and that erosion is greatest within the surface zone. As stacks and arches are eventually eroded away, a wave-cut platform is left.

7. The beach face is the relatively steep area of the beach between the low and high tide levels. This is also known as the foreshore or swash zone.
8. A spit can form where there is longshore drift and the geometry of the shoreline is such that a sand bar extends away from the shore.
9. The area of the Atlantic coast north of Massachusetts (including New Hampshire, Maine, New Brunswick, Nova Scotia, and Newfoundland plus all of the area inland) was glaciated during the Pleistocene and has since rebounded isostatically. This is all now relatively young rocky coast that is being actively eroded.
10. There has been approximately 125 m of eustatic sea-level rise since the last deglaciation, so the current sea level should be approximately 140 – 125 = 15 m lower than it was during glaciation. The dash-dot line marks present-day sea level. An uplifted coast like this should have uplifted wave-cut terraces and coastal waterfalls.



11. Sediments would be trapped in the reservoir behind such a dam, and the water flowing through the dam would be sediment-free. Although there would be erosion of new sediments downstream from the dam, the water reaching the ocean at Richmond would have less sediment than it does now. This is likely to result in the beaches around Vancouver being starved of sediment, and they would gradually get smaller.

Chapter 18

1. Most of the sediments on continental shelves originate from clastic sediments derived from erosion on the continents. The shelves on the eastern coast of North America are wider than those along the west coast because there has been relatively recent (Cenozoic) tectonic activity on the west coast, while the east coast has been passive for about 180 million years.
2. Subduction zone trenches may be partly filled in areas where there is significant sediment input from rivers.
3. From bottom to top, oceanic crust is composed of gabbro, sheeted mafic dykes, and pillow basalts. In most areas it is also covered with varying amounts of sea-floor sediments and sedimentary rocks.
4. The oldest sea floor in the Indian Ocean is in the order of 150 Ma. There is oceanic crust of this age along the western margin of Africa, and adjacent to the northwestern part of Australia.
5. Coarse terrigenous sediments accumulate mostly where major rivers enter the sea, but they are only washed a few kilometers out to sea (at most) because there isn't enough river velocity left to move them farther. Some of those sediments are moved many kilometers farther out to sea during flows of turbidity currents. Clay, on the other hand, can stay in suspension for centuries, and during that time can be dispersed well out into the ocean.
6. Carbonate sediments will accumulate on the sea floor wherever there is significant abundance of carbonate-shelled organisms near to surface, and where the ocean is shallower than the depth at which carbonate becomes soluble (the carbonate compensation depth). In these areas there is typically much more carbonate than clay present, so the sediments look carbonate-rich, even though there is clay there.
7. Carbonate sediments are absent from the deepest parts of the oceans because carbonate minerals are soluble below about 4,000 m depth, so carbonate fragments that settle to that depth dissolve back into the water.
8. The carbon in sea-floor methane hydrates is derived from the bacterial breakdown of organic matter at greater depth in the sediment pile.
9. The tropical parts of the oceans are saltiest because the rate of evaporation is highest. The Mediterranean and Red Seas are saltier than the open ocean.
10. Salty water is transported north by the Gulf Stream and gradually cools. As it cools it remains relatively salty and this cool salty is denser than either cold very fresh water or warm very salty water.
11. The relatively dense water in the north Atlantic sinks to become North Atlantic Deep Water (NADW), and gradually moves back towards the south.
12. The open-ocean currents have the effect of moderating Earth's surface temperature because warm tropical water is moved toward the poles, and cold polar water is moved toward the tropics.

Chapter 19

1. The greenhouse gases (GHG) vibrate at frequencies that are similar to those of infrared (IR) radiation. When IR radiation impinges on a GHG molecule, the molecule's vibrational energy is enhanced and the radiation energy is converted into heat, which is trapped within the atmosphere.
2. The combustion of fossil fuels releases CO_2 that was previously stored in the crust. The resulting increase in atmospheric CO_2 leads to a temperature increase. As the temperature increases, the solubility of CO_2 in the ocean decreases and additional CO_2 is released by the ocean, resulting in even higher atmospheric CO_2 levels and higher temperatures.
3. Gondwana was situated over the South Pole for much of the Paleozoic and became glaciated during the Ordovician (Andean-Saharan Glaciation) and again during the Permian (Karoo Glaciation). These glaciations cooled the entire planet during these periods.
4. From a climate perspective, the two important volcanic gases are SO_2 and CO_2. SO_2 is converted to sulphate aerosols which block sunlight and can lead to short-term cooling (years). CO_2 can lead to warming, but only in situations where there is an elevated level of volcanism over at least thousands of years.
5. We use 65° for estimating the glaciation potential of orbital variations because glaciers are most likely to form at high latitudes. We use 65° N rather than 65° S because for more than 50 million years the continents have been concentrated in the northern hemisphere. We use July instead of January because for glaciers to grow it's more important to have cool summers than cold winters.
6. If the major currents in the oceans were to slow down or stop, the tropics would get hotter and the high-latitude areas would get colder, leading to expansion of glaciers and sea ice. The various feedbacks (e.g., higher albedo because of increased ice cover) would result in an overall cooler climate.
7. The main climate implication of the melting and breakdown of permafrost is that carbon that was trapped in the frozen ground will be released and then converted to CO_2 and CH_4, leading to more warming.
8. Sea-floor methane hydrates are stable because the deep ocean water is cold. In order for the hydrates to become unstable, warmth from the upper layers of the ocean has to be transferred to depth.
9. A significant part of our GHG emissions take place during because of the (formerly) intentional and (presently) unavoidable release of natural gas (CH_4) during the extraction of oil and gas. Some is lost during transportation — for example when pipelines leak — and some is consumed during transportation — for example to pressurize pipelines or to power tanker trains. GHGs are also leaked to the atmosphere during the routine refuelling and operation of motor vehicles.
10. The rise of sea level results from a combination of melting glaciers and thermal expansion of the ocean water. Both of these large systems are slow to respond to the warming climate. For example it takes a long time for warm surface water to be transferred to depth in the ocean or for heat to be transferred to depth in a glacier. Even if we stabilized the GHG levels in the atmosphere today, the climate would continue to warm for approximately another 100 years, and sea level would continue to rise for much longer than that.
11. West Nile virus is carried by birds and is transmitted to humans by certain species of mosquitoes. The range and abundance of those mosquitoes is partly controlled by climate change, especially by warm winters. Sufficiently warm winters are increasingly common in the northern United States and southern Canada.

Chapter 20

1. Some of the components of a compact fluorescent lightbulb (and the resources used to make one) are as follows:

    - Steel (iron, carbon from coal plus some manganese, nickel, chromium, molybdenum)
    - Plastic housing (petroleum)
    - Glass coil (silica from sand, plus minor amounts of sodium, calcium, and magnesium)
    - Copper conductors, lead solder, and basal contact
    - Silica (sand), plastics (petroleum), ceramics (clay), aluminum, gold, copper, etc. in the electronics
    - Mercury inside the tube (less than 5 mg)

2. Nickel deposits form within mafic and ultramafic igneous bodies because the original magma have relatively high nickel levels to begin with, while intermediate or felsic magma have low levels.
3. The "smoke" in a black smoker is composed of tiny crystals of sulfide minerals. If those include significant quantities of ore minerals like chalcopyrite ($CuFeS_2$), sphalerite (ZnS), and galena (PbS), a VMS deposit could form during this process.
4. A porphyry deposit is situated in the rock around an igneous pluton that has intruded to a relatively high level in the crust (and hence is porphyritic), and they form at least in part from fluids released by the magma. Epigenetic gold deposits may be formed from the same or similar fluids, but are situated at a greater distance from the pluton/

5. Ferrous iron (Fe^{2+}) is soluble in water with a low oxidation potential, and gets converted to insoluble ferric iron (Fe^{3+}) when the water becomes oxidized. The opposite situation happens with uranium. Uranyl uranium (U^{6+}) is soluble under oxidizing conditions, but when the water in which it is dissolved encounters reducing conditions the uranium is converted to the insoluble uranous ion (U^{4+}).
6. It is common for the upper part of a kimberlite to be mined using an open pit (in this case around 500 m wide and up to 500 m deep), and for the lower part to be mined underground.
7. Pyrite (FeS_2) is typically responsible for acid rock drainage around mine sites, and it is very common for pyrite to form within the rock at the same time that other metal sulfides (e.g., chalcopyrite) are forming.
8. Glaciofluvial gravels are typically relatively well sorted, and may include clasts ranging in size from coarse sand to pebbles. Till, on the other hand, tends to be poorly sorted and may have clasts ranging from clay to boulders. More processing would be needed to separate the required size ranges, and because till tends to be relatively hard and strong, this would require a lot of effort.
9. During the manufacture of CaO limestone is heated and CO_2 is released to the atmosphere, adding to the greenhouse effect. The energy required for this process typically comes from fossil fuels (e.g., natural gas) and the combustion also releases CO_2.
10. Some important evaporite minerals include halite (NaCl), sylvite (KCl), and gypsum ($CaSO_4.2H_2O$).
11. The 15 m of organic matter required to make 1.5 m of coal, is equivalent to 15,000 mm, and if the organic matter accumulates at 1 mm/y that would require 15,000 years. That organic matter would have to remain submerged in oxygen-poor water for at least that length of time.
12. Petroleum source rocks must have a significant component of organic matter, and then need to be buried to at least 2,500 m depth so that the organic matter can be converted to oil or gas. Reservoir rocks must be both porous and permeable, so that the petroleum liquids can be extracted, and should also take the form of a trap (e.g., an anticline) and capped with impermeable rock.
13. The optimum depth for the generation of oil from buried organic matter is 2,500 to 3,500 m.
14. Shale gas is an unconventional reserve because shale is not permeable enough to allow the gas to be extracted. The rock has to be fractured (fracked) to allow recovery. Fracking involves the use of vast amounts of water, and there is the potential that the fracking fluids can contaminate freshwater aquifers.
15. Kimberlite indicator minerals are much more abundant than diamonds within kimberlites, and so they can typically be detected further away from the kimberlite source, and over a much wider area.

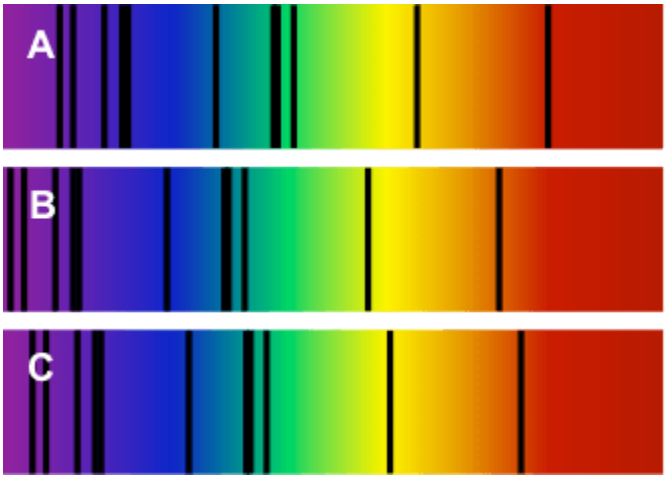
Chapter 21

1. The oldest parts of Laurentia are the Slave and Superior Provinces. Both have rocks that are in the order of 4 Ga.
2. The regions A through E are A-the Cordilleran Fold Belt, B-the Western Canada Sedimentary Basin, C-the Canadian Shield, D-the Innuitian Fold Belt, and E-the Appalachian Fold Belt.
3. Pearya collided with North America to form the Innuitian fold belt during the Devonian.
4. The ancient sedimentary rocks of the Athabasca and Thelon Basins were deposited on the stable Canadian Shield and were never involved in tectonic processes; nor were they buried deeply enough to be metamorphosed.
5. Ultramafic magma has to be very hot to be liquid, and while Earth's interior was hot enough during the Archean, it is no longer hot enough.
6. There are several reasons why the preservation is so good in the Burgess Shale: the rock is very fine grained so details are well defined; the dead organisms accumulated in a lifeless anoxic basin so they were not oxidized, scavenged, or broken down by bacteria while they were being fossilized; although some of the surrounding rocks are weakly metamorphosed, the Burgess Shale was protected from squeezing by adjacent strong limestone.
7. The Western Canada Sedimentary Basin was filled with marine water during pre-Prairie Evaporite times and Winnipegosis carbonate was deposited. It slowly dried out to produce the evaporite beds, but was later re-filled, leading to the deposition of Dawson Bay carbonate. The isolation of the basin during Prairie Evaporite times might have been due to a drop in sea level or tectonic uplift. A change to a dryer climate may also have been a factor.
8. The rocks of the Intermontane Superterrane have fossils that are indicative of southern hemisphere deposition, and also have magnetic inclinations that imply an origin south of the equator.
9. Terrane accretion on the west coast led to formation of the Rocky Mountains. The rapid erosion of these mountains provided a source for accumulation of sediments within the WCSB.
10. The western edge of the WCSB was pushed down by the mass of the Rocky Mountains toward the end of the Mesozoic, and thus can be thought of as a foreland basin.
11. The likely order is Yukon-Tanana, Quesnel, Cache Creek, and Stikine, although it is also possible that these terranes were

assembled as one unit prior to reaching North America.
12. Nanaimo Group sedimentary rocks were forced inland and up to relatively high elevations on Vancouver Island when the accretion of the Pacific Rim and Crescent Terranes pushed Vancouver Island closer to the mainland.
13. The Paskapoo Formation becomes thinner toward the northeast because the foreland basin gets shallower in that direction, and also because the source of the sediments is the Rocky Mountains, situated along the southeastern edge of the basin.

Chapter 22

1. To see an event, light from that event must reach our eyes. Light travels very quickly (about 300,000,000 m/s), but the universe is very, very large. Depending on how far away the event was, it could take billions of years for light to travel from the event to our eyes so we can see it. Astronomers take advantage of this fact to view the universe's past.
2. B is the spectrum from the Andromeda galaxy. We know that one spectrum represents the Sun, which is not moving toward or away from us. (Our orbit is not perfectly circular, but the small eccentricity is not a factor in this comparison.) We know that the Andromeda galaxy is on a collision course with us, so it is the exception to the rule that galaxies are moving away from us, and their light is red-shifted. That means the spectrum B which is shifted furthest to the left (blue-shifted) is Andromeda, and spectrum A which is furthest to the right (red-shifted) is a galaxy moving away from us. That means C is the Sun.



Spectra for the sun and two galaxies. [KP]

3. The planetary system consisted of two Jupiter-sized gas giant planets. Gas giant planets contain large amounts of hydrogen, and hydrogen was plentiful in the early universe. In contrast, terrestrial planets have heavier elements, especially silica, iron, magnesium, and nickel, that had yet to be manufactured by stars. Those elements were not present in sufficient abundance to form terrestrial planets until much later.
4. Closest to the sun we find the small, rocky, terrestrial planets with metal cores. Further out are the gas giant planets, which are the largest in the solar system. They consist mostly of hydrogen, and have cores of rock and ice. Beyond the gas giant planets are the ice giant planets, which are next largest. They have a mantle of ice (not just water ice but ammonia and methane ice), and a rocky core. Smaller objects in the solar system include rocky bodies within the asteroid belt between Mars and Jupiter, and bodies of ice and dust in the Kuiper belt and Oort cloud beyond Neptune.
5. The frost line marks the distance from the Sun beyond which temperatures were cool enough to allow ice to form. This helps to explain why the terrestrial planets are closer to the Sun, and the Jovian and ice giant planets farther away. Mineral grains could solidify and begin to accrete closer to the Sun, forming terrestrial planets, because they have higher melting points. In contrast, water vapor, methane, and ammonia had to be farther from the Sun before they could freeze and begin to accrete.
6. The objects are comets, and two places to find large numbers of comets in the solar system are the Oort cloud and the Kuiper belt. The bright dot the comets have noticed is the sun, and the adventurous comet returns displaying the consequences of the Sun's energy blasting gases and dust from its surface.
7. Planets are defined as having cleared their orbits of debris. Pluto is located within the Kuiper belt, so it shares its orbit with other objects. There are two other criteria in the definition of a planet: planets in our solar system must orbit the Sun, and they must have a spherical shape. Pluto satisfies both these criteria, but sadly the people deciding whether or not Pluto should be a planet are not amenable to a "best two out of three" compromise.
8. Differentiation is the separation of materials within a planet such that dense materials sink to the core, and lighter materials float upward. In Earth's case, the denser materials are iron and nickel, and the lighter materials are silicate minerals. In order

for differentiation to happen, the entire planet must be melted.

9. Thus far it appears that our solar system is unique compared to other planetary systems we have observed. In particular, some other planetary systems have gas giant planets very close to their star. The fact that we have terrestrial planets close to the Sun makes sense in terms of the frost line, but it does not seem to be a hard-and-fast rule in other planetary systems. Therefore, we can't conclude from Kepler-452b's position alone that it is a terrestrial planet.
10. The rules to the accretion game mean that there are many complex interactions, so even a small difference in the starting conditions or in how the game goes in the beginning could have major implications in the end. For that reason, we shouldn't expect to find a planetary system that matches ours in every minute detail. However, just because we haven't found a similar planetary system does not mean one does not exist. Our planet-finding methods are biased toward discovering large planets orbiting close to their stars, whereas our solar system has small planets close to the Sun and larger ones farther away. That doesn't mean our methods won't eventually turn up a system like ours, just that they are more likely to turn up systems that are different.

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